PHYSICAL PROPERTIES OF CHEMICAL COMPOUNDS-II

A systematic tabular presentation of accurate data on the physical properties of 476 organic straight-chain compounds compiled by R. R. Dreisbach of the Dow Chemical Co. These comprehensive and basic data were determined for specially prepared high purity compounds. In addition to the precisely measured properties, the author has calculated new values for many constants based upon his new experimental values.



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Introduction

This is a continuation of Mr. Dreisbach's compilation of physical properties of organic compounds. Data on 511 cyclic compounds were published in 1955 as Number 15 of the Advances in Chemistry Series under the title of "Physical Properties of Chemical Compounds." The present volume includes 476 acyclic compounds.

As in the earlier volume, this compilation contains many data not hitherto published. It also includes parameters which can be used for interpolating and extrapolating the determined data for practically all of the compounds listed.

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Physical Properties of Chemical Compounds

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Definition of the Symbols and Parameters Used, with the Methods of Calculating the Parameters

Mol. % Pur.: Mole % purity by weight.

F.P.: Freezing point, °C.

F.P. 100%: Freezing point curve extrapolated to 100% purity.

B.P. 760 mm., 100 mm., etc.: Boiling points at these pressures, °C.

P₂₅: Pressures at 25°C., in mm.

 P_{ϵ} : Pressure corresponding to temperature t_{ϵ} in mm.

d²⁰, etc.: Density at 20°C., etc., g./ml.

a. b: Constants of Law of Rectilinear Diameters, $d_v + d_L = a + bt$. $d_v = density of the vapor, g./ml.; d_L = density of the liquid, g./ml.$

n_D²⁰, etc.: Refractive index for the sodium line at 20°C., etc.

C: Constant of the Eykman equation, $(n_D^2 - 1)/(n_D + 0.4) \times 1/d = C$

MR (obsd.): Molal refraction (obsd.) = $(n_D^2 - 1)/(n_D^2 + 2) \times M/d = MR$ at 20 °C. (M = mol. wt.)

MR (calcd.): Molal refraction calculated from atomic refractive indices. See page 8.

- $(n_D d/2)$: Refractivity intercept equals refractive index minus one half the density, both at the same temperature, 20°C.
- D: Dielectric constant determined at a frequency of 10⁵ (cycles/sec.) and at 25°C. unless otherwise noted. When reported as data of The Dow Chemical Co., error about ±0.005. Where Reference 5 is noted it was obtained by squaring the refractive index at 20°C.
- A, B, C: Constants of the Antoine vapor pressure equation for the liquid state, giving P (pressure) in mm. and t (temperature) in °C. This is in the range between the temperatures as indicated. These temperatures in general are the boiling point at 30 mm. to a T_R of 0.75 to 0.80. See method of obtaining A, B, C on page 6.

Antoine equation: $\log P = A - B/(t + C)$.

 A^* , B^* , K, c, t_k , t_x : Constants of the saturated vapor density equation

$$\log d_v(g./ml.) = A^* - B^*/(t+C)$$
 to the temperature t_k .

$$\log d_v(g./ml.) = A^* - B^*/(t+C) + K/(1.1 T_c - 273.2 - t) + c$$

from temperature t_k to a reduced temperature, T_R , of 0.92

 t_k = Temperature at which it is necessary to change from the simple vapor density equation to the corrected vapor density equation in the higher ranges, °C.

$$t_k = t_x + K/c$$
 and $t_x = (1.1 T_c - 273.2)$ °C.

 A^* and B^* where the latent heat at the atmospheric boiling point is available.

$$V_q - V_L = (31381.7 \times \Delta Hv \times dt/dp)/T$$

Where the latent heat is not available use

$$M(\Delta Hv)/T_B = 21.0$$
 and from this $\Delta Hv = (T_B \times 21.0)/M$

The value 21.0 (or any other value as 21.4 say) is obtained from the nearest related compound which has a latent heat available. Then proceed as in case where latent heat is available for Vg value at B.P.

Since $d_v = 1/Vg$

$$\log d_{v760} = A^* - B^*/(t_B + C)$$
 at 760 mm.

$$\log d_{v30} = A^* - B^*/(t_{30} + C)$$
 at 30 mm.

Solve for A^* , B^* , since t and d_v at 760 mm. and 30 mm. and C are known.

- A', B', C': Constants of the Antoine vapor pressure equation below 30 mm. pressure, covering the temperature range as indicated. See method of obtaining the constants on page 6.
- A'^* , B'^* : Constants of the vapor density equation below 30 mm. These two values are obtained by using the boiling point at 30 mm. and the pressure at 25 °C. (obtained from the values A', B', C') and assuming that at 25 °C. the relationship PV/RT = 1. Then we have Vg at 25 °C. = RT/MP = 62, $361 \times (25 + 273.2)/MP$.

Then dv = 1/Vg. Inserting these values of vapor density we then solve the two equations for the values of A'^* and B'^* as in the case of A^* and B^* .

- Ac, Bc, Cc: Constants of the Antoine vapor pressure equation for the liquid state from $T_R = 0.75$ (or a higher T_R as indicated) to the critical temperature. See method of obtaining the constants on page 7.
- Cryoscopic Constants, A°, B°: Cryoscopic constants for calculating mole % purity. See J. Research Natl. Bur. Standards, 35 (1945); RP 1676.

t, °C.: Temperature at which a mole of the vapor occupies 22.414 liters and the vapor is in equilibrium with the liquid, in °C.

$$te = \frac{B^*}{(A^* - \log dv_e)} - C$$

dt/dp: Rate of change of boiling point with pressure, given by equation $dt/dp = B/(2.3026 \times P \times (A - \log P)^2)$ °C./mm. Also $dt/dp = (t + C)^2/2.3 PB$

 ΔHm : Latent heat of fusion in cal./g.

 ΔHv : Latent heat of vaporization at the temperature designated, cal./g.

 $t_r(d, e)$: The latent heat of vaporization at temperature t_e as given by the equation $\Delta H_v = d - et$, indicates the accuracy of this equation at temperature t_e .

 $\Delta H_v/T_e$: Molal latent heat of vaporization at t_e divided by T_e . (Equal to the mola entropy of vaporization at t_e .)

- d, e; d', e': These are parameters of the latent heat of vaporization equation. $\Delta Hv(\text{cal./} g.) = d et$. This is valid between the temperatures indicated. It has been found that the latent heat between the boiling point at 30 mm, and the boiling point at 760 mm, is almost a linear function of the temperature. As seen in most cases, this equation holds almost to temperature t_e . Above and below this the latent heat is not linear with temperature except for short intervals.
- d_c : Critical density, g./ml.
- v_e : Critical volume, ml./g.
- t_c: Critical temperature, °C. See also page 7.
- P_c mm.: Critical pressure in mm. Where this was not obtained from the literature it is calculated as follows (The Thomson method, private communication from George W. Thomson): The critical temperature is inserted in the Antoine equation, using the A, B, and C values to calculate the critical pressure.

This value is too low. This is then multiplied by 1.07 and is assumed to be the critical pressure. In the great majority of cases, this will agree with determined values to within $\pm 3\%$. For high boiling compounds this value must be decreased, since in most cases there is somewhat irregular drift with increasing temperature, so this should be continually lowered as the boiling point becomes increasingly higher.

PV/RT: Compressibility at the temperature designated.

$$z = PV/RT$$

where P = pressure in mm., V = volume in ml./mole, and R = 62361.

ΔHc: Heat of combustion, kcal./mole, gas at constant pressure, 298.16 °K. or 25 °C.

ΔHf: Heat of formation, kcal./mole, liquid at 298.16°K. or 25°C.

ΔFf: Free energy of formation, kcal./mole, liquid at 298.16°K. or 25°C.

η: Kinematic viscosity in centistokes, at temperature designated. The kinematic viscosity is given by the equation

$$\log \eta = A^{v} + B^{v}/T$$

between the temperatures indicated to an accuracy of 1% or better.

B.P. °C., 30 mm.; dt/dp; ΔHv ; PV/RT: These values at 30 mm. are calculated from the Antoine equation using A, B, and C. It has been found that at 30 mm. in almost all Cox chart families the ratio PV/RT is negligibly different from one. This, then, has been taken as one point (the other point being the B.P. at 760 mm.) from which to calculate A^* and B^* , always assuming the compressibility as 1.0000 at 30 mm.

- c_p: Specific heat at constant pressure at temperature designated, cal./g. °K.
- c_v: Specific heat at constant volume at temperature designated, cal./g. °K.
- f,g, h, f', g', h': Parameters of the heat capacity equation for the liquid for the temperature ranges designated, °K. $c_p(\text{liquid}) = f + gT + hT^2$
- m, n, o, m', n', o': Parameters of the heat capacity equation for the vapor for the temperature ranges designated, °K. $c_n(\text{vapor}) = m + nT + oT^2$
- γ: Surface tension in dynes/cm., at temperature designated.
- [P]: Parachor at the temperature designated: $M(\gamma)^{1/4}/(d_L - d_v) = [P]$
- [P] Sugd.: Parachor from atomic and structural values as given by Sugden. See table. The parachor value for oxygen as hydroxyl (alcohols) in these tables is taken as 15. Sugden gives the values of 20 for oxygen and 30 for oxygen in esters, which does not seem to work for alcohols and phenols.
- Exp. L.l.; Exp. L.u.: Explosion limits lower and upper range, % by wt.

Dispersion: Specific dispersion, $10^4(n_F - n_c)/d$, ml./g. at 25°C.

 n_F , n_c = refractive index for F and C lines.

d = density, g./ml.

Flash and Fire Points, °C.: Cleveland open cup (ASTM D 92-46) if not otherwise designated. Closed cup (ASTM D 56-36) will be designated as such.

M Spec.: Mass Spectrograph.

Ultra V.: Ultraviolet.

X-Ray Dif.: X-Ray Diffraction.

Infrared: Infrared Spectrograph.

Solubility at 25°C., in solvents as designated.

Explanation of the methods used for calculating the various parameters in the foregoing:

A, B, C: The A, B, and C constants, except where given by the API reports, are calculated by means of the Thomson method [Chem. Revs. 38, 1-39 (1946)] using the determined boiling points at three different pressures. The three formulas for this are as follows:

$$(y_3-y_2)/(y_2-y_1)\cdot(t_2-t_1)/(t_3-t_2)=1-(t_3-t_1)/(t_3+C)$$
 $B=(y_3-y_1)/(t_3-t_1)\cdot(t_1+C)(t_3+C)$ and $A=y_1+B/(t_1+C)$

where y_1 , y_2 , and y_3 are equal to $\log P_1$, $\log P_2$, and $\log P_3$ at temperatures t_1 , t_2 , and t_3 . Unless the data for the three points are *very* accurate, the C value can be considerably in error. As a check on this method an empirical formula developed by Thomson (private communication from George W. Thomson) will give a much better value of C if the data are much in error. This formula is $C = 239 - 0.19t_B$. The A and B values can then be readily determined from the two points given, since they are much less critical.

A', B', C' (for pressures below 30 mm.): Applicable when molar heats of vaporization are available at 25°C. and the Antoine equation can be used to obtain the boiling point at 30 mm. Let A, B, C be the constants of the usual Antoine equation valid above 30 mm. and let A', B', C' be the constants of the Antoine equation sought for below 30 mm. These two equations are taken to give the same value of the pressure-temperature slope at 30 mm.

$$\log 30 = A - B/(t_1 + \tilde{C}) = A' - B'/(t_1 + C')$$

$$B/(t_1 + C)^2 = B'/(t_1 + C')^2$$

Since PV/RT may be assumed to be 1.0000 at t_1 , the temperature corresponding to 30 mm., and is also 1.0000 at 25 °C., the molar heat of vaporization at 25 °C., $M\Delta Hv_2$ is given by

 $M\Delta H v_2 = 2.3026~RB'~[(t_2 + 273.2)/(t_2 + C')]^2$ where $t_2 = 25$ °C. To solve for A', B', C' let $g_2 = M\Delta H v_2/2.3026~R(t_2 + 273.2)^2 = M\Delta H v_2/406883$ if $t_2 = 25$ °C.

Since t_1 , t_2 , and all values on left-hand side of equations above are known, then B' and C' are readily obtained as follows:

$$[B'/(t_2 + C')^2][(t_1 + C')^2B'] = g_2(t_1 + C')^2/B' = \text{say}, h^2$$

Then $C' = (t_1 - ht_2)/(h - 1)$ and $B' = g_2(t_2 + C')^2$
Also $B' = B[(t_1 + C')/(t_1 + C)]^2$
 $A' = \log 30 + B'/(t_1 + C')$ since $P_1 = 30$ mm.

These formulas were developed with the aid of George Thomson.

When heats of vaporization at 25°C. are not known:

In this case the C' value is estimated and A' and B' are calculated from known data. It was noticed that C' has a value approximately 18 higher than C when latent heats at 25°C. are known. By adding this increment to C we have C', then B' from the relation for the first case

$$B' = B[(t_{30} + C')/(t_{30} + C)]^2$$
 and then A' as in first case.

In the case of the alkenes and alkynes the A', B', C' and A'^* , B'^* were not calculated by the above method, since the data for these compounds is much less reliable than in the case of the alkanes.

Ac, Bc, Cc: This method was developed by George Thomson [Chem. Revs. 38, No. 1, 23 (1946)] and is similar to the one for obtaining A', B', C'. It is assumed that parameters A, B, C of the Antoine equation are good to a T_R 0.75 or a higher reduced temperature, and this temperature corresponds to the 25°C. in the case of A', B', C', and the critical point corresponds to the 30-mm. point.

$$B/(t_1 + C)^2 \times (t_c - t_1)/(y_c - y_1) = 1 + (t_c - t_1)/(t_1 + Cc)$$

and $Bc = (y_c - y_1)/(t_c - t_1) \times (t_1 + Cc)(t_c + Cc)(t_c + C_c)$; Ac = $B/t_c + C_c + y_c$
where t_1 °C. = T_R 0.75, t_c °C. = critical temperature
 $y_1 = \log P$ at t_1 , $y_c = \log P_c$

The first equation is used to evaluate Cc, the second, Bc, and the third, Ac.

Association: The association in the vapor phase of organic acids seems to vary inversely as the temperature for some acids, at least for part of the range. In part of the range, and also apparently for some acids over the whole range, the association is fairly constant. The association is given in these sheets by the formula $M_x = p - rt$. For instance, for acetic acid this formula would be $M_x = 2.225 - 0.004085 t$ from 0° to 100° C. From 100° C. to a T_R of 0.92, $M_x = 1.85$. That is to say, the vapor density as calculated by the A^* , B^* formula would have to be multiplied by this correction factor to take care of the association. Further, if the reciprocal of the density is used as calculated to give vapor volume, it would be necessary to divide by 1.85 to get the actual vapor volume.

t_c: Where the critical temperature has not been determined, it is calculated by Watson's equation:

 $T_e/T_c = 0.283 \ (M/d_s)^{0.18}$

where $d_s = \text{liquid density, g./ml.}$ at the boiling point, and M = molecular weight. This is used for all hydrocarbons and halohydrocarbons.

f, g, h, m, n, o, etc.: For a short temperature range the equation $C_p = f + gT + hT^2$ reproduces almost exactly determined data. The parameters were set up on the IBM machines using eight determined values where that many or more were available.

The IBM machines were used to set up the Antoine constants from determined data. A preliminary C value was obtained from the equation $C = 239. -0.19t_B$. A and B were then obtained and new C values either side of the first C used and new A and B values found. In each case above, the boiling points at the experimental pressures were calculated and compared with the determined boiling points.

Actually the value of C was generally obtained from $C = 239. - 0.19t_B$, since the determined values must be very very accurate to give better values of C.

Cox Chart Families

1. Alkanes

2. Haloalkanes

3. Alkenes

4. Haloalkenes

5. Diolefins

6. Alkynes

Atomic Refractive Indices Used for Computing Molecular Refractive Index

All values are for the sodium line

All	values are for th	e sodium line.	
Carbon singly bound and alone Carbon singly bound Carbon double bond Carbon triple bond Carbon conjugated Hydrogen Oxygen—hydroxyl Oxygen—ethereal Oxygen—as ester Sulfur—as SH Sulfur—as RSR Sulfur—as RCNS Sulfur—as RSSR Nitrogen As aliphatic primary amine As aromatic primary amine As aromatic secondary amine As aromatic secondary amine	2.592 2.418 1.733 2.398 1.27 1.100 1.525 1.643 2.211 1.64 7.69 7.97 7.91 8.11 2.45 3.21 2.65 3.59 3.00	NO as nitrites NO as nitrosoamine NO ₂ as alkyl nitrite NO ₂ as alkyl nitrate NO ₂ as nitroparaffin NO ₂ as nitro aromatic NO ₂ as nitramine Fluorine Chlorine Bromine Iodine	5.91 5.37 7.44 7.59 6.72 7.30 7.51 0.95* 5.967 8.865 13.900
As aliphatic tertiary amine As aromatic tertiary amine As hydroxylamine As hydrazine	$egin{array}{c} 3.00 \\ 4.36 \\ 2.48 \\ 2.47 \end{array}$		
As aliphatic cyanide As aromatic cyanide As aliphatic oxime	$3.05 \\ 3.79 \\ 3.93$		
As primary amide As secondary amide As tertiary amide	$egin{array}{c} 2.65 \ 2.27 \ 2.71 \end{array}$		

^{*} This value for one fluorine atom attached to carbon. The value 1.1 is to be used for each fluorine atom in polyfluorides.

Atomic and Structural Constants for Calculation of Parachor

	Sugden		Sugden
CH ₂	. 39.0	Br	. 68.0
C		I	
H		Single bond	
0		Double bond	. 23 . 2
O (Alcohol)		Triple bond	. 46.6
O ₂ (Ester)		3-membered ring	. 16.7
N		4-membered ring	. 11.6
N (nitrile)		5-membered ring	. 8.5
S		6-membered ring	. 6.1
F		7-membered ring	
Cl		Aliphatic alcohol subtrac	t = 6.0

								No. 1	
NAME	Methane					ST	RUCTURAL	FORMUL.	A
							СН ₄		
Mole % Pur.		ecul		Molecular Veight 16.04	12		4		
	•	Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-182.48	2	dt/dP °C/mm			f g	to °K		
B. P. °C 760 mm 100 30 10	-161.49 -181.45 -190. -195.51	2 2 4 2	-140°C BP t _e 30 mm	0.00537 0.0160 0.3567 0.2141	5 2 5 4	f' g'	to*K		
1	-207.0	5	ΔHm cal/g	14.025	-	m	300 to	0.1259	4
Pressure mm=140°(3289.7 272.1	5 5	AHv cal/g -140°C 30 mm BP	115.67 133.34 121.87	3 5 2	n o	_60 <u>0</u> _•K	0.0015 0.0 ₆ 66	4
Density g/ml-1600 dt -150 4 -140	0.4222 0.4075 0.3916	3 3 3	t _e t _e (d, e) ΔHv/T _e	125.30 126.36 19.99	5 5 5	m' n' o'	1000 •K	0.1408 0.0013 -0.0 ₆ 34	4 4
a -160°C	0.4242 -0.00126	3 4	d -190 to e -160 °C d' to	56.90 0.4023	5		face tension es/cm180°C -170	15.8	2
Ref. Index n _D 20°C 25 30			e' °C d g/ml vc ml/g t °C	0.162 6.17 -82.5	2 2 2	Par	-160 achor [P] 20°C 30	13.7/4.	2
"C"			t _c °C P _c mm	34808.	2		40 Sugd.	73.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	6.818	5	PV/RT -140°C 30 mm	0.9208 1.0000	4 5	_	L.1.%/wt. u. persion	13.2	,
Dielectric			BP	0.9628 0.9726	4 5		sh Point C		
A -180 to B	6.61184 389.93 266.00	2 2 2	te tc ΔHc kcal/m	0.290	2	M.	Spec.		
A*L165 to B*L130 °C K	0.74077 362.78 7.79 -0.15125	4 4 4	ΔHf ΔFf Viscosity centistokes			X-F Infr	ra V. Ray Dif. ared ability in +	Yes	2
t _k [-130 to t _x -100 °C	-63.0	5	7 -180 °C -175 -170	0.418 0.364 0.325	2 2 2	Ca Be	etone rbon tet. nzene her		
B'°C C' A'* to			BV -160 to AV -160 °C	0. 295 101. 76 Z . 52937	4 4	n-: Eti Wa	Heptane hanol iter		
B¹* °C			(B ^V) to (A ^V) °C			Wa	iter in		
Bc t _c °C	0.0130	,	c _p liq.300 °K 400	0.53310 0.60691	2 2				
Cryos. A° consts. B°	0.0138 0.0057	2	c _p vap. °K						
t _e °C	-172.65	4	c _v vap.						
≠ solid			# at saturation				ams/100 gran		t
	ES: 1-Dow		PI 3-Lit. 4-0	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		AP							
PURIFICAT	ION: RE REFEREI	AP							
MIERTO.									

						No. 2
NAME	Ethane					STRUCTURAL FORMULA
						CD CD
Mole	Ref. Mo	lecul	ar C ₂ H ₆	Molecular		сн ₃ сн ₃
% Pur.	Fo			Weight 30.06		
	-183. 27 [≠]	Ref.		r	Ref.	Re
F.P. °C F.P. 100%		2	dt/dP *C/mm			f to to
B. P. °C	 	 	25°C	0.00213	5	g <u>*K</u>
760 mm	-88.63	2 2	BP t	0.0244 0.0351	2 5	f' to
10 0 3 0	-119.33 -132.74	4	t _e 30 mm	0,3350	5	g' ' <u>°</u> K_
10 1	-142.88 -159.51	5	ΔHm cal/g	22.728	2	h'
Pressure	-137.31	۲	ΔHv cal/g			m to
mm 25°C	29290.	5	25°C	76.86	5 5	n •K
t _e	480.9	5	30 mm BP	129.71 116.97	2	<u> </u>
Density g/ml-60°C	0.509	3	t _e ,	119.45	5 5	m' to K
at 20	0.363	3	te (d, e)	119.33	5	0'
⁴ 4 30	0.30	3	ΔHv/T _e	20.36	5	Surface tension
a b	0.364 -0.0363	4	e i -90 °C		5	dynes/cm110°C 19.57 2
Ref. Index	-0.0303	÷	d' to		1 1	-100 17.93 2 -90 16.31 2
n _D 20°C				0,203	2	Parachor [P]
25 30			d g/ml vc ml/g tc °C	4.92	2	20°C
"C"			t _c °C	32.27	2	40
MR (Obs.)		†	P _c mm	36632.	2	Sugd. 112.2 5
MR (Calc.)	4		PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.
(nD-d/2) Dielectric			30 mm	1.0000	5	Dispersion
A -132 to	6,80266	2	BP t _e	0.9622 0.9735	5 5	Flash Point °C
B <u>-44 °C</u>	656.40	2	tc	0.288	2	Fire Point
С	256.00	2	ΔHc kcal/m ΔHf	341.26	2	M Spec. Ultra V.
A* -132 to B* -90 °C	0.98156	5	ΔFf			X-Ray Dif. Infrared
K		Ĭ	Viscosity			Solubility in +
t _k	-		centistokes 7 -120 °C	0,438	2	Acetone
t _X C			-110	0.387	2	Carbon tet. Benzene
A' to			-100 -90	0.348 0.314	2 2	Ether
	•		B ^V -130 to	135.71	4	n-Heptane Ethanol
A'* to			A ^V -80 °C	2.75615		Water
B'* °C			(B ^V) to	1		Water in
Ac -44 to Bc t _c °C	7.67290 1096.9	5	(A ^V) °C			
Cc	320.54	5	c _p liq. •K	ļ		
Cryos. A°	0.04256	2	c _p vap.300°K	0.41225	2	
consts. B°	0.0095	2	c _v vap.	0.52148	2	
t _e °C	-96.7	5	<u> </u>			
T _R = 0.7			# at saturation			grams/100 grams solvent
SOURCE:	1-DOW	Z-AI		aic. irom det	. dat	ta 5-Calc. by formula
PURIFICAT	'ION:	A1				
	RE REFERE					
	NUFEREI	-CES	3 Young			

								No. 3	
NAME	Propane					ST	RUCTURAL	FORMUL	A
L			Т				CH ₃ -CH ₂	-CH ₃	
Mole % Pur.	Ref. Mo	ecul		Molecula <i>r</i> Weight 44.09	4				
		Ref.			Ref.			***************************************	Ref
F.P. °C	-187.69	2	dt/dP			f	to		
F.P. 100%	<u> </u>		°C/mm 25°C	0.00561	4	g	' ° <u>K</u>		
B. P. °C 760 mm	-42.07	2	BP	0.0298	2	h			-
100 30	-79.63 -96.07	2	t _e	0.0355	5	f' g'	to °K		
10	-108.5	4	30 mm	0.4109	2	h'			
1	-129.	5	ΔHv cal/g	19.10	+	m	300 to	0.0169	
Pressure mm 25°C	7095.	4	25°C	81.76	4	n o	600_•K	0.0014 -0.0 ₆ 43	4
t _e	608.5	4	30 mm BP	114.70 101.76	5 2		1		
Density g/ml 20°C	0.5005	2	t _e ,	102.98	5	m' n'	700 to	0.0960 0.0012	4
t 25 d4 30	0.4928≠	2	te (d, e)	102.94	5	0'	1	-0.0640	4
⁴ 4 30	0.4861	4	ΔHv/T _e	20.07	5	Sur	face tension		
a b	0.5375	4	_e _		5	dyn X	es/cm70°C	19.2 17.85	2
Ref. Index	 	i	d' to				-60 - 50	16.49	2
n _D 20°C	1.2898	5	d _c g/ml	0, 220	2	Par	achor [P]		
25 30			v ml/g	4. 358	2		20°C		
"C"	0.766	5		96.8	2 2		40		_
MR (Obs.)			P _c mm	31928.			o. L.1.%/wt.	151.2	5
MR (Calc. (nD-d/2)) 16.054 1.04	5	25°C	0.8461	4	-	u.		İ
Dielectric	1.66	5	30 mm BP	1.0000 0.9612	5 4		persion		ļ
A -130 to	6.82973	2	t _e	0.9669 0.278	5 2		sh Point C e Point		
B (_ <u>5°C</u>	_ 813.200 248.00	2 2	tc ΔHc kcal/m	488,53	2	M.	Spec.		ļ
A* -100 to	1.05579	5	ΔHf	-28.643	2		ra V. Ray Dif.		
B* <u>40</u> °C K	756.21	5 4	ΔFf	 	\vdash		ared	Yes	2
c	17.62	4	Viscosity centistokes				ubility in +		
t _k -40 to t 65 °C		5	η -80 °C -70	0.524 0.470	2 2		etone rbon tet.		
A' to	151.0	Ť	-60	0.425	2		nzene he <i>r</i>		
B'°C	_		-50	0.387	2	n-	Heptane		
A¹* to			B ^V -85 to A ^V -30 °C	190.47 Z.73347	4 4		hanol ater		
B'* °C			(BV) -140 to	184.09	4		ter in		
Acl 5 to		4	(A ^V) -85 °C	₹. 76598	4		cosity		
Bc tc °C	1090.0 287.8	4	c _p liq. °K			ι cen	tistokes -130°C	1.126	2
Cryos. A°	0. 05802	2	c vap300°K	0,40051	2	•	-90 -40	0.590 0.355	2
consts. B°	0,0073	2	400	0.51118	2		- 20	0.333	-
t _e °C	-46.98	5	c _v vap.	1	Ll	<u> </u>			L
T _R = 0.7			t at saturation				rams/100 gra		t
	CES: 1-Dow		PI 3-Lit. 4- PI	Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:	TION:		PI	-					
	RE REFERE								
			••						

								No. 4	
NAME	n-Butane				\Box	STR	UCTURAL 1	FORMULA	
							сн,сн,сн	LCH.	
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 58,12			301201	.20113	
		Ref.			Ref	i i			Ref.
F.P. °C	-138.350	2	dt/dP			ſ	to		
F.P. 100%			°C/mm			g	<u>•</u> K_		
B. P. *C			25°C BP	0.01768 0.03465	2	h	L		
760 mm 100	-0.50 -44.17	2	t _e	0.03597	5	f'	to		
30	-63.30	4	30 mm	0.4778	4	g'	. ' <u>*</u> K_		
10 1	-77.76 -101.5	4 5	ΔHm cal/g	19.167	2	h'	<u> </u>		
Pressure	1	-	ΔHv cal/g	- / /-		m n	300 to	0.0351 0.0013	4
mm 25°C	1823.	4	25°C 30 mm	86.63 105.08	2	o		-0.0640	4
te	724.6	5	BP	92.09	2	m'	700 to		4
Density g/ml 20°C	0.5788	2	te t (d, e)	92.35 92.35	5	n'	11000 •K	0.0974 0.0012	4
dt 25	0.5730	2	te (d, e) ΔHv/Te		5	٥'	i – – –	-0.0 ₆ 40	4
	0.5671	4	d -60 to	19.77 91.99	5		ace tension		
a b	0.6039 -0.0 ₃ 99	4	_ 0 ℃	0, 2069	5	dyne	s/cm30°C	18.43	2
Ref. Index	10.0377	<u> </u>	a' 0 to	91.98	4		-20 -10	17.22 16.02	2
n _D 20°C	1.3326	2	e' 30 °C	0, 2228	4	Para	chor [P]		
45	1.3292	2	d g/ml vc ml/g	0, 228 4, 387	2		20°C		
30	1.3252	5	tc °C	152.01	2		30 40		
"C"	0.7730	4	P _c mm	28477.	2			190.2	4
MR (Obs.) MR (Calc.	20.63 [‡] 20.772 ₄	2 5	PV/RT	İ		Exp.	L.1.%/wt.		
(nD-d/2)	1.0432	2	25°C 30 mm	0.9286 1.0000	4 5	Di	u.		
Dielectric	1.776	5	BP	0.9582	4		ersion h Point °C	-60.	3
A -60 to		2	t _e	0.9595 0.274	5		Point	-60.	3
B <u>45</u> °C	945.9 240.0	2 2	t _c AHc kcal/m	635, 05	2	M S	ec.		
A* -60 to		5	ΔHf	-35.29	2	Ultr			
B* _ 30 °C		5	ΔFf			Infra	ay Dif. ared	Yes	2
K — — —	1		Viscosity			Solu	bility in +		
t _k to	-		centistokes 7 -40 °C	0.491	2	Ace	tone		
t			30	0.446	2		rbon tet. Azene		
A' to			-20 -10	0.407 0.375	2	Eth	er	ļ	
B', ∟ _ °	-1		BV -40 to	234.16	4		leptane anol		
A¹* to			AV 10 °C	Z. 68698	4	Wa	ter		
B'* °C			(BV) -100 to	227. 4 2	4		ter in		<u> </u>
Ac 45 to		4	(A ^V) -40 °C	₹.71750	4		osity istokes		
Bc tc_°C	1299. - 289.1	4	c _p liq. °K			η	-90 °C	0.91	2
Cryos. A°	0, 03085	2	c _p vap.300°K	0.40261	2	∥ `	-50	0.545	2
consts. B°	0.0048	2	400	0.50929	2		0	0.350	۲ ا
t _e °C	-1.75	5	c _v vap.						L
			$T_R = 0.75 T_c$				ms/100 gran		
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-C	alc. from det	. da				
SOURCE:		AP	1						
PURIFICAT		AP		_					
LITERATU	RE REFEREI	NCES	: 3 NFPA 325						

No. 5 2-Methylpropane NAME STRUCTURAL FORMULA Isobutane CH3CH-CH3 Molecular C4H10 ċн₃ Mole Ref. Molecular % Pur. Weight 58.120 Ref. Ref. -159.600 F.P. °C F.P. 100% 2 dt/dP f to °C/mm °K_ g 25°C 0.0132 B. P. °C h BP 0.0337 2 760 mm -11.730 2 5 0.0361 f to 100 -54.07 2 g¹ -72,52 <u>°К</u> 30 0.4600 4 30 mm 4 10 -86.4 4 h! ∆Hm cal/g 18.668 2 1 -109.2 5 300 to m -0.0058 AHv cal/g Pressure n 600 °K 0.0015 25°C 78.63 2 mm 25°C 2611. o -0.0658 4 30 mm 99.79 5 $^{t_{\underline{e}}}$ 697. 5 ВP 87,56 2 m' 700 to 0.0920 4 Density 87.99 te te (d, e) n' 0.0013 1000 °K g/ml 20°C 0.5572 5 88.00 -0.0641 o' 4 0.5510[#] 25 2 d_4^t ΔHv/T_e 19.72 5 30 0.5450 4 Surface tension d -75 85, 20 to 5 0.5846 -0.0₃98 4 4 dynes/cm.-40°C 17.68 2 °C 0.2011 b ă - 10 -30 16.48 2 84.71 to 4 15.28 2 Ref. Index -20 e' 0.24314 25 ⁿD 20°C 5 1.3169 [P] Parachor d_c g/ml 0.221 2 25 20°C vc ml/g t_°C 2 4.525 30 30 ŧ, 134.98 2 40 "C" 0.7675 5 2 P_c mm 27360. Sugd. 190. 2 5 MR (Obs.) PV/RT Exp. L.1.%/wt. MR (Calc.) 20.772 5 25°C 0.9083 (nD-d/2)30 mm 1.0000 5 Dispersion Dielectric 1.734 5 BP 0.9648 4 5 Flash Point C -87 0.9668 5 7-75 to 6.74808 te t 2 Fire Point 2 0.283 В 1_30 °C 882.80 2 M. Spec. С 240.00 2 AHc kcal/m 633.05 2 Ultra V. ΔHf A* -75 to -37.87 1.04221 5 X-Ray Dif. ΔFf B*|_ 0_°C 820.37 Yes 2 Infrared ĸ Viscosity Solubility in c centistokes Acetone to -50 °C 0.619 Carbon tet. °C -40 0.549 2 Benzene 2 -30 0.491 A'ı to Ether -20 0.443 B' °C n-Heptane B^V | -90 to A^V | -55 °C C' 26<u>5</u>. 10 Ethanol A'* 2,6035 Water to Water in B'* (BV) -55 °C to 4 273.73 Acl 30 to 7.42067 (A^V)| Viscosity °C 2.5653 4 Bc tc °C 1288.1 centistokes cp liq. Cc 296.7 4 η -80°C 0.946 2 -60 0.703 2 0.04234 c_p vap 300°K Cryos. Aº 0.40003 2 -10 0.403 consts. B° 0.0057 400 0.51222 c_v vap. te °C -13.90 $= 0.75 T_{c}$ # at saturation pressure T_{R} grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

										No. 6	
NAME		n-Per	itane	•				STR	UCTURAL	FORMULA	4
									CII (CII)	CII	
Mole % Pur. 99	9.6	Ref.		lecul		Molecular Weight 72.14			сн ₃ (сн ₂) ₃	3 ^{CH} 3	
70 Pur. 77	. 00	الثا	FO	Ref	12 1	Weight 72.14	Ref				Ref
F. P. *C	1.1	29.72	,	2	11.715	T	Kei	<u> </u>		Γ	101
F.P. 1009		27.12	•	╁╌┤	dt/dP *C/mm	į.		f g	to °K		i
B. P. *C	+-			\vdash	25°C	0: 05257	4	h		}	1
760 mm		36. 07		2	BP t	0. 03856 0. 03631	2 5	f'	to		_
100 30		12.59 3 3 .93		2 4	30 mm	0.5334	4	g'			l
10	-	50.1		4	ΔHm cal/g	27, 805	2	h'	l]
1	 -	76.63		5	ΔHv cal/g		-	m	300 to	0.0310	
Pressure mm 25°C	5	12.5		4	25°C	87.54	2	n o	_600 •K_	0.0014 -0.0 ₆ 42	
t _e		25.9		5	30 mm BP	98.53 85.38	5 2	<u> </u>		-0.0642	<u> </u>
Density					t.	84.92	5	m' n'	700 to	0.0976	
g/ml 20°C	: [0.62 0.62		2 2	e (", ",	84.92	5	o'	11000 •K	0.0012 -0.0 ₆ 41	
d ₄ 25 30		0.61		4	AHv/T _e	19.65	5		<u> </u>	63.	<u> </u>
	+	0.64	604	4	d -35 to		5		ace tension s/cm, 20°C	16.00	2
b		- 0 .03	904	4	-å,- -40_ : €		5	3,	30	14.95	2
Ref. Index		1 25	740	1.1	e' •C			<u> </u>	40	13.8≠	2
n _D 20°C	1	1.35 1.35		2 2	d _c g/ml	0.232	2	Para	chor [P] 20°C	231.0	4
30		1.35		4	v_ mi/g	4.311 196.62	2		30	231.0	4
"C"	Γ^{-}	0.76	64	4	`	25316.	2		40 Suad	230.5 229.2	5
MR (Obs.)		25. 26		2	P _c mm PV/RT	23310.	<u> </u>	Fyn	L. 1. %/wt.	3,6	3
MR (Calc. (nD-d/2)	"	25.29 1.04		5 2	25°C	0.9662	5	Exp.	u.	18.6	3
Dielectric	+-	1.84		5	30 mm BP	1.0000 0.9547	5 4	Disp	ersion	98.0	2
A -35 to		6, 85		2	t _e	0.9523	5		h Point °C	-40.0	3
B _ 80_°		64. 63		2	tc	0, 268	2		Point		-
_c	2	32.0		2	ΔHc kcal/m ΔHf	782.04	2	M S _l		ļ	1
A* -35 to		1.18		5	ΔFf	-41.36 -2,25	2	X-R	ay Dif.		١.
B* ∟ 60 °C		95.37 24.0		4	Viscosity	 		Infra		727.	1
¢ ,	┨.	-0.16	0 75	4	centistokes				bility in +		1
t _k 60 to t _x 160 °C	2	85. 43. 6		4 5	7 10 °C	0.432 0.401	2	Car	bon tet.	œ	
A' I to				\vdash	20	0.375	2	Ber Eth	nzene	80 80	
B' L_ '	2			1 1	B ^v -10 to	0.351	2	n-F	leptane	œ	
C'	+-				B' -10 to A' 40 °C	251.11 2.71711	4	Eth Wa	anol	o o	Ì
A'* to B'* *(261.39	4		ter in		L
Ac 80 to	-	7, 37	001	5	(B') -70 to (A ^V) -10 °C	Z, 67863	4	Visc	osity		
Bcit (14	11.3		5	c _p liq. °K	2.01003	-	1 -	istokes -60°C	0.803	2
Ce	12	79.1		5	1 -			1	-20	0.514	2
Cryos, A° consts, B°	1	0. 04 0.0 0		2 2	c _p vap.300°K 400	0.40016 0.50633	2		-10	0.469	2
te °C	†	38.54		5	c _v vap.	0.50033	•]	1
$T_R = 0.7$				1 ⁻ 1	at saturation	pressure		+ ~==	ma/100 ~~	no oct	<u> </u>
			ow .		PI 3-Lit. 4-0		de:	ta 5-	ms/100 graz	nula mula	<u> </u>
SOURCE:					PI				by 1011		
PURIFICA	TION	 I:			PI						
			ERE		: 3 NFPA 32						
				- 							

	2. Mathu	lhutana						No. 7	
NAME	2-Methy Isopenta				\dashv	ST	RUCTURAL	FORMUL	,A
Mole % Pur. 99	Ref.	Molecula Formula		Molecular Weight 72,146	6		сн ₃ снсн сн ₃	2 ^{CH} 3	
		Ref.		T	Ref.				Ref
F.P. °C F.P. 1009	-159.900		dt/dP °C/mm			f g	to °K		
B. P. °C 760 mm	27.852	2	25°C BP	0.0412 0.03815	4 2	h			L
100 30	-20.14 -41.10	2 4	t _e 30 mm	0.0369	5 4	f' g'	to * <u>K</u>		l
10 1	-56.9 -82.8	4 5	ΔHm cal/g	17.063	2	h'			_
Pressure mm 25°C	688.1	4	ΔHv cal/g 25°C	81.47	2	m n	300 to	0.0043 0.0015	4
t _e	793.8	5	30 mm BP	94.54 80.97	5 2	0	1	-0.0651	+
Density g/ml 20°0	0,619	67 2	te te (d, e)	80.73 80.72	5 5	m' n'	700 to	0.1052 0.0012	
dt 25	0.614	62 2	ΔHv/T _e	19.27	5	٥'		-0.0 ₆ 39	
a b	0.640	50 4	d -40 to e 30 °C	86.45 0.1968	5		face tension es/cm. 20°C 30	15.00 13.93#	2 2
Ref. Index			d' to ℃				40	12.73	5
ⁿ D 20°0 25 30	1.353 1.350 1.347	88 2	d g/ml vc ml/g t °C	0.234 4.269	2 2	Par	achor [P] 20°C 30	230.0 229.9	4 4
"C"	0.766		tc°C P _c mm	187.8 25004.	2 2		40	229. 2	5
MR (Obs.) MR (Calc.			PV/RT	 	$\overline{\Box}$	Exp	. L.1.%/wt.	/:-	†
(nD-d/2)	1.043		25°C 30 mm	0.9478 1.0000	5	Dis	u. persion	98.6	2
Dielectric	1.843 6.789		BP t _e	0.9452 0.9438	5		sh Point °C	-60.	5
B 1_75 ℃		2	te tc AHc kcal/m	780.12	2	M.	Spec.		\vdash
A* -45 to B* 40 °C		47 5	∆Hf ∆Ff	-42.85 -3.59	2	X-F	ra V. lay Dif. ared	Yes	2
K ——	_		Viscosity centistokes			Solu	bility in +	res	+-
t _k to			η -10 °C	0.479 0.434	2 2		etone rbon tet.	60	
A' to		-	10	0.396	2		nzene her	∞ ∞	
B' °	-		B ^v -10 to	0.364 305.9	2		Heptane hanol	80 80	
A'* to B'* °C			$\begin{array}{c c} \mathbf{A^{V}} & 30 & \mathbf{C} \\ \hline (\mathbf{B^{V}}) & -40 & \mathbf{to} \end{array}$	2 .51761	4	Wa	iter iter in	1	
Acl 75 to	7. 256		(A ^V) -10 °C	326.2 Z.55042	4 4		cosity		T
Bc tc °C	1325.2 275.8	5 5	c _p liq. °K				istokes -50°C	0.790	2
Cryos. Acconsts. B			c _p vap.300°K	0.39559	2 2	(-20 30	0.535 0.337	2 2
te °C	29.12	5	c _v vap.	0.50578					
$T_R = 0.7$	5 T _c # 2	20°C	#at saturation				ams/100 gra		ıt
*	CES: 1-Do			Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			PI						
PURIFICA			PI						
	OL NEFE	NEI40E3	: 3 NBS Circ	., 714					

								No. 8	
NAME	2,2-D	imethylp	propane			STR	UCTURAL		Ł
	Neope	ntane					СН ₃ С С		
Mole % Pur.	Ref.	Molecu Formu		Molecular Weight 72.14	6		CH ₃	·-3	
A Fui.		Ref		weight	Ref				Ref
F. P. *C	-16.550		dt/dP	1	-	f	1 1	<u> </u>	
F.P. 100%			*C/mm	1		g	to K		
B. P. *C	1		25°C BP	0. 0244 0. 03652	4	h			
760 mm 100	9.503	3 2	t.	0.0370	5	f'	to		
30	-56.27	4	30 mm	0.4973	4	g'			
10 1	-71.3 -95.9	4 5	AHm cal/g	10.786	2	h'	<u> </u>		
Pressure	1 - /3. /	-+-	ΔHv cal/g			m	300 to	-0.0159	4
mm 25°C	1285.2	4	25°C 30 mm	72.15	2	n o	_600 °K_	0.0016 -0.0 ₆ 62	
t _e	750.8	5	BP	86.88 75.37	5 2		!		-
Density g/ml 20°C	0 501	10 1 2	to (d a)	75.43	5	m' n'	700 to	0.1199 0.0012	4
dt 25	0.591 0.585	51 2	te (d, e)	75.43	5	01		-0.0642	4
⁴ 4 30	0.579	92 4	ΔHv/T _e	19.27	5	Surf	ace tension		\vdash
a b	0.616		d -60 to		5		s/cm. 20°C	12.05	5
Ref. Index	-0.001		d' 50			•	30 40	10.98 9.94	5
n _D 20°C		2 2		 	-	Par	chor [P]		
25 30	1.342	, 2	v ^c ml/g	0.238 4.200	2 2		20°C		
"C"	0.777		tc °C	160.60	2		30 40		
	+		P _c mm	2399 3.	2			229.2	5
MR (Obs.) MR (Calc.		5 2 0 5	PV/RT	0.0221	-	Exp	L.1.%/wt.		
(nD-d/2)	1.046	5 [*] 2	25°C 30 mm	0.9 32 1 1.0 0 00	5	Diar	u. ersion	98. *	2
Dielectric	1.801	1 5	BP	0.9554	4 5		h Point °C	-75.	5
A -60 to			te t _c	0.9557 0.269	2		Point	-13.	
B L 55_•C	950.84	2 2	ΔHc kcal/m	777, 37	2	M S		Yes	2
A* -60 to	1,108	331 5	AHf	-44.98	2	Ultr X-R	a V. ay Dif.		1
B* ∟20 °C	886. 37	5	ΔFf (gas)	-3, 66	2	Infr		Yes	2
С		- 1	Viscosity centistokes				bility in +		
tk to			7 -10 °C	0.626	2		tone rbon tet.	e0 e0	l
t _x ; °C			-5	0.576 0.534	2 2		nzene	80	
B' •C		ĺ	5	0.498	2	Eth n-H	er Ieptane	e0 e0	
C'		_	B ^V -10 to	484.9	4	Eth	anol	œ0	
A'* to B'* *C			H-75	3.95415	4	Wa Wa	ter ter in		
Acl 55 to	+	572 5	(B ^V) to						\Box
Bc t °C	1261.3	5	<u>'</u>	 					
Cc	280.8	5	c _p liq. °K						
Cryos. A° consts. B°	0, 005	595 2	c _p vap.300°K 400	0.40487 0.52047	2 2				
t _e °C	9.167	7 5	c, vap.						L
$T_{\mathbf{R}} = 0.79$			≠ at saturation				ms/100 gran	ns solven	t
REFERENC	ES: 1-Do			Calc, from det	t. da	ta 5-	Calc. by for	mula	
SOURCE:			API						
PURIFICAT			API						
LITERATU	RE REFE	RENCE	S:						

								No. 9	
NAME	n-He	xa ne				ST	RUCTURAL	. FORMUL	A
		,					CH ₃ (CH ₂) ₄	СН	
Mole	Ref.	Molecula		Molecular			3, 3, 2, 4	3	
% Pur. 99.	99 2	Formula	6.14	Weight 86.17		ı .			Ref
F, P, *C	-95, 348	Ref.	14/17		Ref.			7	Kei.
F.P. 100%			dt/dP °C/mm		١	f g	to °1		
B. P. °C	/0.74		25°C BP	0.1524 0.04191	2	h			
760 mm 100	68.740 15.81	2 2	t	0.0363	5	ſ'	t	5	
30	-7.44	4 4	30 mm	0.5815	4	g'	•1	K	
10 1	-25,1 -54.0	5	∆Hm cal/g	36.137	2	h'			ļ
Pressure			ΔHv cal/g 25°C	87.50	2	m n	300 to 600 °1		
mm 25°C	151.3 914.5	4 5	30 mm	93.37	5	•	l 1	-0.0643	
t _e Density	714.3		BP	80.03 79.01	5	m'	700 to		
g/ml 20°C	0.659	37 2	t _e (d, e)	78.98	5	n'	1000 •1		
dt 25 d4 30	0,654		ΔHv/T _e	19.57	5	0'		-0.0640	1
A 30	0,677		d -10 to	92.06	5		face tension		2
b	-0.038		e 75 °C to	0.1751	5	الا الا	es/cm. 20°C 30	17.38	2
Ref. Index		10/ 2	• · · · · · · · · · · · · · · · · · · ·			ļ	40	16.36	2
ⁿ D 20°C	1.374		d _c g/ml	0.234	2	Par	achor [P] 20°C	271.0	4
30	1.369	38 4	l v ml/g	4.271 234.7	2		30	270.9	4
"C"	0.760	7 4	t _c *C	22739.	2		40 Suga	270.9 1. 268.2	5
MR (Obs.)	29.907		PV/RT	 	 	Ext	L. 1. %/wt		31
MR (Calc. (nD-d/2)	29.908 1.045		25°C	0.9847	5	1	u.	18.0	31
Dielectric	1.890) 3	30 mm BP	1.0000 0.9503	4	J	persion	98.0	2
A -10 to	6.877	76 2	te	0.9447 0.264	5 2		sh Point °C e Point	-26.0	3'
B (110 °C	1171.53 224.366	2 2	tc ΔHc kcal/m	928, 93	2	М.	Spec.	Yes	1
A* -10 to	1,257		ΔHf	-47.52	2		ra V. Ray Dif.		
B*[100 °C		5	ΔFf	-1.03	2		ared	60.	1
K	-0.131		Viscosity centistokes		l		ubility in +		
tk 100 to	117.	4 5	ິ ກ 30 °C	0.4389	2		etone rbon tet.	90	
195 °C	285.6	- 3	· 40 50	0.4085 0.3817	2 2	Be	nzene	•	
B' C			60	0.3577	2		her Heptane	80	
C'			B ^V 15 to A ^V 60 °C	297.4 2.66156	4	Et	hanol	•	
A'* to B'* °C	ļ		(BV)1-30 to	309.4	4		ater ater in		
Ac 110 to	7,319	38 4	(A ^V) 15 °C	2.61885	4	Visc	osity	T	
Bc tc C	1483.1	4	c _p liq. °K	1 2.51535	١ ٠	11 .	tistokes -20°C	0 401	,
Ce— —	265.9	4			1.	א	20	0.691 0.4741	2
Cryos, A° consts, B°	0.049		c _p vap.300K 400	0.39885 0.50446	2 2		70	0.336	2
t _e °C	74.76	5	c vap.	0.30440	-				
$T_{R} = 0.75$			1		_	" + g	rams/100 gr	ams solver	ıt.
REFEREN		2-A	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:		API							
PURIFICA'	TION:	API							
LITERATU	RE REFE	RENCES	5: 3 NBS 514;	3' NFPA 325					

								No. 10	
NAME	2-Methylp	entar	ne			STR	UCTURAL 1	FORMULA	
	Isohexane						сн ₃ снсн ₂ с	н.сн.	
Mole	2.6			Malandan			CH ₃	23	
		lecul rmul		Molecular Weight 86, l	72		3		
		Ref.	l .		Ref.				Ref.
F.P. *C	-153.670	2	dt/dP			f	to		
F.P. 100	6		°C/mm	0 1142		g	<u>*</u> K		
B. P. °C 760 mm	(0.37)		25°C BP	0.1143 0.04141	2	h			
100	60.271 8.06	2 2	-t	0.0366	5	f'	to		
30 10	-14.82	4	30 mm	0.5717	4	g'	K_		
1	-32.1 -60.55	5	∆Hm cal/g	17.407	2	h'	1 200 4	0.0150	-
Pressure			ΔHv cal/g	03.03	١, ١	m n	300 to	0.0150 0.0014	4
mm 25°C		4	25°C 30 mm	82.83 89.76	2 5	0		-0.0650	4
t _e Density	891.5	5	BP	77.09	2	m'	700 to	0.1533	4
g/ml 20°	0.65315	2	te te (d, e)	76. 25 76. 23	5	n'	11000 • K	0 .0011	4
dt 25	0.6485€	2	ΔHv/T _e	19.41	5	0'		-0.0635	4
	0.64386		d -15 to	87.26	5		ace tension		
a b	0.67177 -0.0 ₃ 890	4	_e _i _65•C	0.1688	5	dyne	s/cm. 20°C 30	17.38 16.37	2 2
Ref. Index			d' to				40	15.36	2
n _D 20°		2	d _c g/ml	0.235	2	Para	chor [P]		
25 30	1.36873	2 4	l v mi/g	4.259	2	1	20°C 30	269.7 269.7	4
"C"	0,7613	4	t _c ·C	224.9	2		40	269.6	4
MR (Obs.		2	P _c mm	22762.	2			268.2	5
MR (Calc.	29.908	5	PV/RT 25°C	0.9793	5	Exp.	L. 1. %/wt. u.		
(nD-d/2)	1.04488	2	30 mm	1.000 0	5	Disp	ersion	98.6	2
Dielectric		5	BP t _e	0.9512 0.9463	4 5		h Point °C	-14.	5
A -15 t B 100 °		2 2	tc	0.269	2	ļ	Point		
c	226.572	2	ΔHc kcal/m	927.23	2	M Sp Ultra			
A* -15 to	1.22708	5	ΔHf ΔFf	-48.82 -1.97	2	X-R	ay Dif.		
B* 175 °C	<u>C</u> 1063.37	5	Viscosity		- -	Infra		Yes	2
°			centistokes				oility in +		
t _k t			7 0 ℃	0.5525 0.4746	3	Car	bon tet.	œ	
A' t	, 	-	30	0.4349	3	Ber Eth	zene	00 00	
B' *	<u>s</u>		B ^V -10 to	 	-	n-H	leptane	x	
C' t	_	-	B' -10 to A' 40 °C	286.99 T.69184	4	Eth. Wat	anol	oc	
A'* t			(BV) to	1	ا		er in		
Ac 100 to	7.31635	4	(A ^V) °C	1					
Bc tc_	C 1468.9 271.5	4	c _p liq. °K	†	_	1			
Cryos. A		2	(I -	0.40:07	١, ١				
consts. B		2	c _p vap.300°K 400	0.40187 0.51061	2				
t _e °C	65.38	5	c _v vap. 388.56°K	0.366	3	<u> </u>			
$T_R = 0$						T gra	ms/100 gran	ns solvent	t
	CES: 1-Dow			Calc. from det	da:	ta 5-	Calc. by for	mula	
SOURCE:	TION		PI PI						
PURIFICA									
LITERAT	KE REFERE	NCES	3 ASTM 109	•					

No. 11 3-Methylpentane NAME STRUCTURAL FORMULA CH3CH2CH CH2CH3 ĊH₃ Ref. Molecular Molecular Mole $C_{6}H_{14}$ % Pur. 99.8 Weight 86.172 3 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 0.1262 B.P. °C h ВP 0.04182 2 760 mm 63, 282 2 t_e 0.0366 5 ſ١ to 100 10.53 2 g' <u>°К</u> 30 -12.61 4 30 mm 0.5781 4 10 -30.1 4 h' ∆Hm cal/g -58.87 5 300 to 0.0804 4 m AHv cal/g Pressure n _6<u>0</u>0_°K 0.0014 4 25°C 83.96 mm 25°C 189.8 4 o 0.0643 4 30 mm 90.29 5 te 902.1 5 RP 77.88 2 0.0648 m' 1 700 to Density 4 76.99 5 te (d, e) n' 0.0013 1000 °K g/ml 20°C 4 0.66431 2 76.97 5 ٥' -0.0₆46 4 25 0.65976 $\mathbf{d_{4}^{t}}$ 2 AHv/Te 5 19.40 30 0.65519 4 Surface tension -15 to 88.23 5 4 0.68259 dynes/cm. 20°C 18.12 2 <u>70</u> 0.1636 5 ь -0.0₃88 4 ă۳ 17.08 to 40 16.03 2 e' Ref. Index 20°C ⁿD 1.37652 2 [P] Parachor d_c g/ml 0.235 2 1.37386 2 25 20°C 267.9 4 vc ml/g tc °C 4.259 2 4 30 1.37106 30 4 267.8 t_c 231.2 2 40 267.6 4 "C" 0.7582 4 P_c mm 2 23431. Sugd 268.2 5 MR (Obs.) 29.802 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 29.908 5 25°C 0.9822 5 (nD-d/2)1.04436 2 30 mm 5 1.0000 Dispersion 97.1 2 Dielectric 1.895 5 BP 0.9531 4 Flash Point °C -35. 5 A -15 to te tc 0.9480 5 6.84887 2 Fire Point 0.273 2 B (105 °C 1152.368 2 M. Spec. C 227, 129 2 AHc kcal/m 927.87 2 Ultra V. ΔHf -48.28 2 A* -15 to 1.22845 5 X-Ray Dif. ΔFf Z 1078.38 -1.34 B*| 80 °C Infrared 2 Yes ĸ Viscosity Solubility in centistokes Acetone œ to 0 °C t_k 0.5774 31 Carbon tet. °C œ $t_{\mathbf{x}}$ 15 0.5068 31 Benzene œ 25 31 0.4653 A'ı to Ether œ °C 40 0.4156B' n-Heptane œ B^V A ċ١ -10 to 305.47 4 Ethanol 00 50 °C **Z.** 64336 4 Water A1* to B'* Water in °C (B^V) to Acl 105 to (A^V) 7.36387 4 °C Bc tc °C 1520.0 cp liq. ۰ĸ Cc 276.46 4 c_p vap.300°K Cryos. Aº 0.39885 2 consts. Bo 400 0.50446 2 te °C c, vap. 68.84 5 $T_{\mathbf{R}} = 0.75 \, \mathbf{T}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES: 3 ACS 74, 1951 (1952); 3' Timmermans

							No. 12	
NAME	2, 2-Dimet	nylbu	tane			STRUCTURAL	FORMULA	L
	Neohexane					CH3		
24.1.	7, 1,			M-11		сн3с сн	2 ^{CH} 3	
Mole % Pur. 99.		lecul rmul		Molecular Weight 86.17	2	CH ₃		
		Ref.			Ref.			Ref.
F.P. *C	-99.870	2	dt/dP			f to		
F.P. 100%	<u>'</u>		*C/mm 25*C	0.0814	4	g <u> </u>	-	1
B. P. °C 760 mm	49.741	2	BP	0.04117	2	h		<u> </u>
100	-2.0	2	t _e	0.0372	5	f' to		1
30 10	-24.49 -41.5	4	30 mm	0.5619	4	g' ' <u>*</u> K	1	1
ī	-69.3	5	ΔHm cal/g	1.607	2	1 222	0,0046	4
Pressure			ΔHv cal/g 25°C	76, 79	2	m 300 to	0.0014	4
mm 25°C	319.1 864.9	5	30 mm	84.62	5	•	7 -0.0 ₆ 47	4
Density	1 001.7	<u> </u>	BP	72.96 72.32	2 5	m' 700 to	-0.0210	
g/ml 20°C		2	t _e (d, e)	72.32	5	n' 11000 °K		
dt 25	0.64446 0.63972	2	AHv/Te	19.06	5		-0.0 ₆ 64	1.
. 50	0.66820	4	d -25 to		5	Surface tension dynes/cm, 20°C	16.30	2
ь	-0.03889	4	 å,¦90- : S		5	30	15.31	2
Ref. Index			i ic			40	14.33	2
ⁿ D 20°C	1.36876	2	d _c g/ml	0.240	2	Parachor [P] 20°C	267.2	4
30	1.36312	4	vc ml/g tc °C	4.166 216.2	2 2	30	267.2	4
"C"	0.7608	4	P _c mm	23309.	2	40 Sugal	267.1 . 268.2	5
MR (Obs.)		2	PV/RT	23307.	1	Exp. L. 1. %/wt.	1200.2	+-
MR (Calc. (nD-d/2)) 29.908 1.04418	5 2	25°C	0.9754	5	u.		
Dielectric	1.873	5	30 mm BP	1.0000 0.9544	5 4	Dispersion	99.8	2
A -25 to		2	t _e	0.9505	5	Flash Point °C Fire Point	-47.	5
B L95_°C	1081.176	2 2	t _c	0.274	2	M Spec.	 	\vdash
A* -25 to	1.14940	5	ΔHc kcal/m ΔHf	924.53 -51.00	2 2	Ultra V.	ļ	
B* _ 65 °C		5	ΔFf	-2.90	2	X-Ray Dif. Infrared	Yes	2
K			Viscosity centistokes			Solubility in +	1	\vdash
the Tto		1 1	7 0 °C	0.7144	3	Acetone Carbon tet.	•	
* 1			20	0.5777	3	Benzene	80	ĺ
A' to		1	25 30	0.5446 0.5158	3	Ether n-Heptane	•	İ
c, =			B ^V -10 to	39 <u>0</u> .6	4	Ethanol	80	1
A'* to B'* *C			A ^V 40 °C	7.42425	4	Water Water in	ł	
B'* °C			(B ^V) to	}			 	\vdash
Bc t °C	1494.7	4						
Cc	286.2	4	c _p liq. ∘K	1				
Cryos. A° consts. B°	0.002321 0.000	2 2	c _p vap.300°K 400	0.39560 0.50712	2 2			
t _e °C	53.84	5	c _v vap.					L
$T_R = 0.7$						⁺ grams/100 gra		t
	CES: 1-Dow	2-AI		Calc, from det	t. da	ta 5-Calc. by for	mula	
SOURCE:	FION	AP						
PURIFICAT		AP	3 Timmerm			··		
	NO NOT DIVE	NO 20	. 3 limmern	ans				
L								

23 No. 13 2, 3-Dimethylbutane NAME STRUCTURAL FORMULA сн3сн сн сн3 ċн,ċн, Molecular Ref. Molecular C6H14 % Pur. 99.99 Formula Weight 86.172 Ref Ref. Ref. °C -128.538 2 dt/dP f to F.P. 100% °C/mm g °K 25°C 0.1059 B.P. °C h ВP 0.04173 2 760 mm 57.988 2 0.0370 5 ſ١ ŧ, 100 5.45 2 to g¹ -17.53 <u> °K</u> 30 4 30 mm 0.5738 4 10 - 34. 9 4 h' 2.251 ∆Hm cal/g 2 -63.37 5 300 to -0,0058 m AHv cal/g Pressure 600 °K 0.0015 n 4 25°C 80.77 mm 25°C 234.6 o -0.0₆54 4 30 mm 87.57 5 887.5 5 t_e BP 75.65 2 m' 700 to 0.0686 4 Density 74.87 te (d, e) 5 5 n' g/ml 20°C 1000 •K 0.0013 -0.0₆46 0.66164 44 2 74.86 ٥' 0.65702 2 d_4^t 25 AHv/T 19.19 5 30 0.65237 4 Surface tension d -20 to 84.80 0.68026 5 4 dynes/cm. 20°C 17.37 2 2 °C 0.1578 _65_ ь -0.03885 4 30 ăח 16.37 to 40 15.37 2 Ref. Index e' ⁿD 20°C 1.37495 2 Parachor [P] d_c g/ml 0.241 2 25 1.37231 2 20°C 266.3 4 vc ml/g t_°C 4.154 30 1.36939 4 30 t_c 266.2 227.1 2 40 "C" 266.1 4 0.7583 5 P_c mm 23552. 2 Sugd. 5 268. **2** MR (Obs.) 29.810 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 29.908 25°C 0.9800 (nD-d/2)2 u. 1.04413 30 mm 1.0000 5 Dispersion 98.3 2 Dielectric 1.890 5 BP 0.9535 4 Flash Point C -37. 5 0.9489 A -20 to 5 6.80983 2 Fire Point 0.270 2 1700 °C 1127.187 2 M. Spec. С ∆Hc kcal/m 228.900 2 926.40 2 Ultra V. A# -20 to ΔHf -49.48 1.19420 5 X-Ray Dif. ΔFf -1.69 B* 75 °C 1053, 55 Infrared Yes 2 Viscosity Solubility in c centistokes Acetone to °C œ Carbon tet. t_x °C œ Benzene œ A' to Ether B' °C B_v | n-Heptane œ C' to Ethanol œ ۰c A'* Water to Water in B'* (BV) °C to Ac | 100 to 7.30917 (A^V)| °C Bc tc °C 1481.8 c_p liq. ۰ĸ 277.4 4 Cryos. A° consts. B° 0.00467 2 cp vap.300°K 0.39177 0.50248 2 c, vap. te °C 62.99 5 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

											No. 14	
NAME		n-He	ptan	e				STR	UCTURA	L :	FORMULA	
									CH IC	ч ,	CH	
Mole		Ref.	Ma	lecul		Molecular			СН ₃ (С	′°°2′	5 ^{C11} 3	
	. 94	2		rmul	C-H , I	Weight 100 1	98					
				Ref.			Ref.					Ref.
F.P. C		90.61	0	2	dt/dP			ſ		0 to	0.0373	5
F.P. 100%	<u>'</u>			\vdash	*C/mm 25*C	0.4376	4	g	1 60	0 . ₹	0.0013	1
B. P. °C 760 mm	1	98.42	27	2	BP	0.04479	5	h	 		-0.0 ₆ 39	5
100		41.77		2	t _e	0.0365	5	f' g'	ii	to °K		
30 10		16.80 -2.1	,	4 2	30 mm	0.6237	4	h'	;	_*`-	1	
1		33.2		4	ΔHm cal/g	33.474	2	m	30	0 to	0,0272	4
Pressure		45 01			ΔHv cal/g 25°C	87.18	2	n		·K	0.0014	4
mm 25°C		45.81 88.01		4 5	30 mm	89.14	4	٥	i		-0.0 ₆ 45	4
Density	+			Н	BP t	75.60 73.96	2 5	m'	700		0.1062	4
g/ml 20°C	:	0.68		2	t _e (d, e)	74.08	5	n' o'	1 1000) •K	0.0012 -0.0 ₆ 40	4
dt 25 4 30		0.67 0.67		2 4	ΔHv/T _e	19.46	5		<u> </u>		-0.0610	_
	+	0, 70		4	d 15 to	91.93	4		ace tensi		19.29	5
ь		-0.03		4	-a,, -1°5 €	0, 1659	4	3,	3	0	18.34	5
Ref. Index					e' ' °C	,		<u> </u>	4		17.41	5
ⁿ D 20°C	1	1.38 1.38		2 2	d _c g/ml	0.235	2	Para	chor [P	0°C		
30	1	1.38		4	v _c ml/g t _c °C	4. 252 267, 01	2 2		3	0		
"C"		0.75	72	4	P _c mm	20528.	2		4 S		307.2	5
MR (Obs.)		34. 55		2	PV/RT		\vdash	Exp	L.1.%/		3, 4	31
MR (Calc. (nD-d/2)	ή :	34.52 1.04		5 2	25°C	1.0000	5		u.		18.0	31
Dielectric	\top	1.92		3	30 mm BP	1.0000 0.9401	5 4		ersion	_	97.7	2
A 15 to	,	6.90		2	t _e	0.9326	5		h Point * Point	C	-3.89	3'
B 1730.0		68.11		2	tc	0, 260	2	M S				
A* 15 to	_	16.90		5	ΔHc kcal/m ΔHf	1075.85 -53.63	2 2	Ultr	a V.		1	
B* 155 °C	111	94.60		5	ΔFf	0.27	2	Infra	ay Dif. ared		593.	1
K	7 7	24. -0.14	584	5 4	Viscosity	l		Solu	bility in	+		_
t 155 to	1 !	5 7 .	704	4	r centistokes	0.5586	2	Ace	tone		80	
t 220 °C		21.		5	50	0.4752	2		rbon tet. nsene		8 0	
A' to					7 0 90	0.4118 0.3624	2	Eth			80	ļ
č, – – <u>-</u>	-1			ll	B ^V 20 to	344.88	4		leptane anol		8 0	
A'* to					AV 100 °C	Z . 60962	4	Wa				
B'* *(-			\vdash	(B ^V) -60 to	3 91.3	4	W &	ter in			
Ac 130 to	151	7. 32 81.7	70	5 5	(AV) 20 °C	Z. 44706	4	ł				
Cc C	- 2	57.6		5	c _p liq.300 °K	0.39780	5					
Cryos. A		0.05		2	c _p vap.300°K	0.50182 0.39781	2					
consts. B°	+-	0.00		2	400	0.50320	2					
t _e °C T _R = 0.7		07.58		5	c _v vap.	L	لــــا	L			L	L
				2 1-	NT 2 T/:						ms solvent	<u> </u>
SOURCE:	·ES:	1-10	OW.	Z-AF AF		alc. from det	t. da	ta 5-	Calc. by	for	mula	
PURIFICA'	74017			AF								
			rD Er		: 3 NBS Circ.	£14. 21 N						
TILERAIU	KE,	KEF 1	ur ei	√LS	: 3 NBS Circ.	514; 3' NFF	^A 32	5				

No. 15 2-Methylhexane NAME STRUCTURAL FORMULA CH3CHCH2(CH2)2CH3 Molecular C7H16 сн3 Mole Ref. Molecular % Pur. Weight 100.198 Formula Ref. Ref. F.P. °C F.P. 10 -118,276 2 dt/dP f to 100% °C/mm °<u>K</u> g 25°C 0.3189 4 B. P. *C h BP 0.04431 2 760 mm 90.052 2 t_e 0.0366 5 ſ١ 100 34.093 to g' <u>°K</u> 30 9.52 4 30 mm 0.6145 4 -9.09 10 4 h! ∆Hm cal/g 21.907 2 1 -39.7 5 300 to 0.0272 AHv cal/g Pressure n 600 °K 0.0014 25°C 83.02 2 mm 25°C 65.88 -0.0₆45 o 4 30 mm 85.98 5 t_e 971.1 5 ВP 73.14 2 m 700 to 0.1062 4 Density 71.85 5 te (d, e) 1000 °K 0.0012 4 g/ml 20°C 0.67859 5 2 71.79 ٥' -0.0640 4 25 0.67439 $\mathbf{d_{4}^{t}}$ AHv/T 19.37 5 30 0.67022 4 Surface tension 10 to 87,50 5 0.69538 a dynes/cm, 20°C 19.29 2 100 °C 0.1595 5 å⊢ -0.03825 Ъ 4 18.32 30 2 to 40 2 Ref. Index 17.36 e' ⁿD 20°C 1.38485 [P] Parachor dc g/ml 0.234 2 25 1.38227 2 20°C 309.6 4 vc ml/g t °C 2 4.272 30 1.37979 4 30 309.5 4 t_c 257.9 2 40 309.4 4 "C" 0.7576 4 P_c mm 20672. 2 Sugd 307.2 5 MR (Obs.) 34.591 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.526 25°C 0.9897 5 (nD-d/2) 1.04556 2 30 mm 1.0000 Dispersion 98.5 2 Dielectric 1.919 3 BP 0.9467 4 Flash Point C -14. 5 0.9389 5 A 10 to B 125 °C 6.87318 1236.026 t_e 2 Fire Point 2 0.267 M. Spec. С 219.545 2 AHc kcal/m 1074.14 2 Ultra V. 2 1.30078 ΔHf A* 10 to -54.93 5 X-Ray Dif. ΔFf -0.68 2 B*[110 °C 1161.12 Infrared Yes 2 ĸ Viscosity Solubility in centistokes Acetone to 20 °C 0.5570 31 Carbon tet. °C Benzene œ ٦ī to Ether an B١ ۰c n-Heptane œ B^V A C' to Ethanol œ °C Water A'* to (BV) Water in B'* ۰c to Ac | 125 to 7.31001 (A^V)| °C Bc tc °C 1555.4 c_p liq 19. 38°C 0.5570 31 Cc 261.5 4 Cryos. A° consts. B° 0.04605 2 vap.300°K 0.39781 2 0.0036 2 0.50320 2 400 c_v vap. te °C 98.53 $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES: 3 NBS 514; 3' Timmermans

								No. 16	
NAME	3-Meth	ylhexane	•			STR	UCTURAL 1	FORMULA	1
						,	יש כש כשוכי	ם / כם	
	1 1		Т		\dashv		H ₃ CH ₂ CH(C	¹ 2'2 ^C 13	
Mole	Ref.	Molecul		Molecular			CH ₃		
% Pur.		Formul		Weight 100.1	_				_
		Ref.		·	Ref				Ref
F.P. °C			dt/dP			f	to		
F.P. 100%			*C/mm 25*C	0. 3391	4	g	K		
B. P. °C 760 mm	91.85	0 2	BP	0. 044 59	2	h	ļ 		L_
100	35.56	1 2	t _e	0.0369	5	f'	to	1	
3 0 10	10.85	4	30 mm	0,6179	4	g'		ł	
10	-7.9 - 38 .6	2 5	ΔHm cal/g	_		h'	<u> </u>		<u> </u>
Pressure	30.0	-	ΔHv cal/g			m n	300 to	0.0272	4
mm 25°C	61.59	4	25°C 30 mm	83.68 86.33	2	"		0.0014 -0.0 ₆ 45	
te	976.	5	BP	73.43	2			<u> </u>	-
Density			t.	72, 11	5	m'	700 to 1000 K	0.1062	4
g/ml 20°C	0.68 0.68		t (d, e)	72.04	5	0'	! ' <u>-</u> , -, -, -, -, -, -, -, -, -, -, -, -, -,	-0.0640	
dt 25 4 30	0.67		AHv/T	19.33	5	-		-	├
8	0.70		d 10 to		5		s/cm. 20°C	19.79	2
Ъ	-0.03		급,ᆜᅩᅆᇎ		5	3,	30	18.81	2
Ref. Index			• •				40	17.85	2
n _D 20°C	1.38		d g/ml v ml/g	0.240	2	Para	chor [P]		١.
30	1.38			4.172	2		20°C 30	307.7 307.6	4
"C"	0.75		-	262.4	2		40	307.5	4
MR (Obs.)	34, 46		P _c mm	21356.	2			307.2	5
MR (Calc.)	34.52		PV/RT 25°C	0, 9915	[] ا	Exp.	L.1.%/wt.	l	1
(nD-d/2)	1.04		30 mm	1,0000	5 5	Dian	u. ersion	97.4	2
Dielectric	20° 1.93	3	BP	0.9468	4		h Point °C	-14.	5
A 10 to	6.86		t _e	0.9389 0.267	5 2		Point	**.	
B [_130 °C	1240.19 219.22		t _c ΔHc kcal/m	1074.78	2	M S	ec.		
A* 10 to	1, 29		AHC KCal/m	-54.35	2	Ultr	a V.		
B* 110 °C		5	ΔFf	-0.39	2	Infra	ay Dif. red	Yes	2
ĸ	1		Viscosity				bility in +	 	┢
£	ł		centistokes 7 °C			Ace	tone	œ	1
ty C	1		7 ·c		(bon tet.	œ	1
A' to		-	ŀ	1		Eth	nzene er	00 00	
B' 'C	ł			 	$\vdash \vdash \vdash$	n-H	leptane	∞	
C'	 		B ^V to			Eth Wat	anol	∞	
A'* to B!* °C	1			-			ter in		
Ac 130 to	7, 39	633 4		1					
Bc t tc C	1635.1	4							
Cc	270.8	4	c _p liq. °K		[
Cryos, A° consts, B°			c _p vap.300°K	0.39781	2				
t _e °C	100,58	5	c _v vap.	0.50320	2				
$T_{R} = 0.75$			L	1		+ gra	ms/100 gran	ns solven	<u>. </u> t
REFERENC		w 2-AI	PI 3-Lit. 4-(Calc, from det	da1		Calc. by for		
SOURCE:		AI		· · · · · · · · · · · · · · · · · ·		<u> </u>	, -,		
PURIFICAT	ION:	AI	PI						
LITERATUE	E REFE		3: 3 NBS 514						

No. 17 3-Ethylpentane STRUCTURAL FORMULA NAME Triethylmethane CH3CH2CHCH2CH3 Molecular C7H16 C_2H_5 Mole Molecular Weight 100.198 % Pur. Ref. Ref F. P. ℃ -118.604 2 dt/dP f to F.P. 100% °C/mm 25°C °K g 0.3584 B. P. ℃ h BP 0.04482 2 760 mm 93, 475 2 0.0367 5 ſ١ t_e to 100 36.864 2 g' °K 30 12.00 4 30 mm 0.6218 10 -6.8 4 h! ∆Hm cal/g 22,775 2 1 -37.8 5 m 300 to 0.0272 ∆Hv cal/g Pressure n 1 600 °K 0.0014 25°C 84,02 2 mm 25°C 58.05 o -0.0645 4 4 30 mm 86.47 983.0 5 t_e BP 73,83 2 m' 700 to 0.1062 4 Density 72.81 5 te te (d, e) 'n 0.0012 1000 °K g/ml 20°C 0.69816 5 2 72.43 o' -0.0640 4 d_4^t 25 0.69395 2 AHv/Te 19.42 5 30 0.68982 4 Surface tension 10 to 88.34 5 0.71499 dynes/cm. 20°C 20.44 2 100 °C 0.1552 5 ь -0.03829 4 30 19.45 2 ٦ī to 18.47 40 2 Ref. Index e' ^{n}D 20°C 1.39339 2 [P] Parachor d_c g/ml 0.241 2 25 1.39084 2 305.3 20°C 4 vc ml/g t_°C 4.152 2 30 1.38840 4 30 305.2 4 t_c 267.6 2 40 305.2 4 "C" 0.7519 4 P_c mm 21736. 2 Sugd. 307. 2 5 MR (Obs.) 34.283 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.526 5 25°C 0.9917 5 (nD-d/2)1.04431 2 30 mm 1.0000 5 Dispersion 95.7 2 Dielectric 1.939 3 BP 0.9485 4 Flash Point C -14.0 5 0.9404 5 A 10 to 6.87564 2 Fire Point 0.268 2 B 1130 °C 1251.827 2 M. Spec. C AHc kcal/m 2 219.887 2 1075.40 Ultra V. ΔHf -53.77 2 A* 10 to 1.29505 5 X-Ray Dif. ΔFf 2 1.04 B* 115 °C 1174.98 2 Infrared Yes ĸ Viscosity Solubility in centistokes Acetone to œ $\mathbf{t_{k}^{t}}$ Carbon tet. œ °C Benzene 00 A'I to Ether œ В' °C n-Heptane B_v | œ C' to Ethanol œ °C Water A'* to Water in B'* °C (B^V)| to Acl 130 to 7.29098 4 (AV) °C Bc tc °C 1561.6 c_p liq. °K Cc 261.0 4 cp vap.300°K Cryos. A° 0.04807 2 0.39781 2 consts. B° 0.0039 0.50320 400 c_v vap. te °C 102.49 $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES: 3 NBS Circ. 514

								No . 18	
NAME	2, 2-Di	nethylpe	ntane		\exists	STRUCT	JRAL CH ₃	FORMULA	
L							CH ₂ C	н,сн,	
Mole % Pur. 99	. 99 Ref.	Molecul Formul		Molecular Weight 100.1	98		CH ₃	2 3	
		Ref.			Ref				Ref.
F. P. *C	-123, 811		dt/dP	Τ		f	•-		
F.P. 100%			*C/mm			g _	to °K		
B. P. *C	1		25°C	0.2138	4	h			
760 mm	79.197		BP	0.04394 0.0372	5	f' + T	to		
100 30	23.858	2 4	t _e 30 mm	0.6047	4	g' 1	°K		l
10	-18.6	4			2	h'			
1	-48,68	5	ΔHm cal/g	13.892	۲	m I	300 to	0,0272	4
Pressure	1,05 22		ΔHv cal/g 25°C	77.36	2	n	600°K	0.0014	4
mm 25°C	105.23 942.7	4 5	30 mm	81.39	5	•		-0.0645	4
Density	+		BP	69.55	2	m' I	700 to	0.1062	4
g/ml 20°C	0.673	85 2	te te (d, e)	68.44 68.45	5 5	n' 1_1	000 °K	0.0012	4
at 25	0.669	53 2	ΔHv/T	1	5	°'		-0.0640	4
4 30	0.665		d 0 to	19.06		Surface te	ension	<u> </u>	
a	0.691		90.00		5 5	dynes/cm	. 20°C	18.02	2
ь	-0.038	42 4	d' to	1		1	30 40	17.08	2 2
Ref. Index		15 2	e' j •c	ļ	\sqcup	Bana aban		10.14	-
ⁿ D 20°C	1.379		d _c g/ml	0.248	2	Parachor		306.6	4
30	1.376	94 4	v _c ml/g t _c °C	4.032 247.7	2 2		30	306.5	4
"C"	0.758	0 4		21584.	2		40	306.4	4
MR (Obs.)	34.617	2	P _c mm PV/RT	21304.	-	<u> </u>		307. 2	5
MR (Calc.			25°C	0.9878	5	Exp. L.1. u.			
(nD-d/2)	1.045		30 mm	1.0000	5	Dispersio		99.3	2
Dielectric	1.912	3	BP	0.9486	4 5	Flash Poi	nt °C	-25.	5
A 0 to			te t _c	0.9418 0.268	2	Fire Poin]
B LIFE.	223.303		ΔHc kcal/m	1071.45	2	M Spec.			
A* 0 to	$\overline{}$	$\overline{}$	ΔHf	-57.05	2	Ultra V.	•		
B* 95 ℃		5	ΔFf	-1.15	2	X-Ray Dif	١.	Yes	2
к — —			Viscosity		1	Solubility	in +	†	
ξ — — to	-1		centistokes 7 20 °C	0.5773	31	Acetone		∞	İ
t _k to			7 20 °C	0.5775	'	Carbon t	et.	∞	
A' to		-		İ		Benzene Ether		ο ο	
B' °	2		B ^V l to		\vdash	n-Heptan	e	œ	
	+	\dashv	B ^V to C			Ethanol Water		∞	
A'* to B'* °C			<u> </u>	-		Water in			İ
Ac 115 to		52 4	1						
Bc t °C		2 4		 					
Cc'	290.5	4	c _p liq. °K						
Cryos. A° consts. B°	0.031 0.003		c _p vap.300°K 400	0.39781 0.50320	2 2				
te °C	86.57	5	c _v vap.						
$T_R = 0.7$	5 T _C					+ grams/1	00 oras	ms solven	<u></u>
REFERENC		w 2-AF	PI 3-Lit. 4-0	Calc. from det	t. dat				
SOURCE:		AF							
PURIFICAT	TION:	AF							
			: 3 NBS 514;	31 Ind Fra	Char	20 220 4	19441		
			. 3 NBG 314,	J ma. Dag.	on o n	<u>50,</u> 530 (, , , , ,		

								No. 19			
NAME	2, 3-Dime	thylp	entane			STRUCTURAL FORMULA					
							сн сн сн (-н сн			
Mole % Pur.	Ref. Mo	lecul mula	ar C ₇ H ₁₆	Molecular Weight 100.19	8		ch ₃ ch ch c	20113			
		Ref.			Ref.				Ref.		
F. P. ℃			dt/dP			f	to				
F.P. 1009	6		°C/mm 25°C	0.3103	4	g	' <u>*K</u>				
B. P. ℃ 760 mm	89.784	2	B P	0.04482	2	h			_		
100	33.237	2	t _e	0.0369	5	f' g'	to *K				
30 10	8.43	4	30 mm	0.6200	4-	h'					
1	-41.19	5	ΔHm cal/g	 		m	300 to	0.0272	4		
Pressure mm 25°C	40.07		∆Hv cal/g 25°C	81,68	2	n	_6 <u>0</u> 0_•K	0.0014	4		
t _e	68.87	5	30 mm	84.58	5	°		-0.0 ₆ 45	4		
Density			BP te	72.48 71.23	2 5	m'	700 to	0.1062	4		
g/ml 20°0		2	t _e (d, e)	71.19	5	n'	[1000 •K	0.0012 -0.0 ₆ 40	4		
d ₄ 25 30	0.69091 0.68673	2 4	ΔHv/T _e	19.20	5			. 6			
a	0.71175	4	d 5 to		5		ace tension s/cm. 20°C	19.96	2		
Ъ	-0.03819	4	_e100 °C		5	8	30	18.98	2		
Ref. Index		١, ١	e' °C	:		<u> </u>	40	18.02	2		
ⁿ D 20°0	1.39196	2 2	d _c g/ml	0.247	2	Par	achor [P] 20°C	304.8	4		
30	1.38696	4	v _c ml/g t _c °C	4.042 264.6	2 2		30	304.7	4		
"C"	0.7527	4	P _c mm	22192.	2		40 Sugd.	304.6 307.2	4 5		
MR (Obs. MR (Calc.		2 5	PV/RT			Exp	. L.1.%/wt.				
(nD-d/2)	1.04442	2	25°C 30 mm	0.9903 1.0000	5	_	u.	04.3	2		
Dielectric	1.939	3	BP	0.9501	4		sh Point C	96.2 -16.	5		
A 5 to		2	t _e	0.9424 0.268	5 2		Point	-10.	,		
B (130 °C	221.823	2 2	t _c AHc kcal/m	1073.12	2	М.	Spec.				
A* 5 to		5	ΔHf	-55.81	2		a V. ay Dif.				
B*[110 °C		5	ΔFf	-1.27	2		ared	Yes	2		
K			Viscosity centistokes				bility in +				
t _k \ \ \ to		1	າ 15 °C	0.6233	31		etone rbon tet.	8 0			
x			30	0.6000	3'	Be	nzene	∞			
A' to B' °C			ļ	_	1		ner Heptane	∞ ∞			
C'			B ^V to A ^V C			Etl	nanol	øo			
A'* to B'* °C			⊩. <u>-</u> v. -	-			ter ter in				
Acl 130 to	+	4	(B') to								
Bc tc °C	1652.5	4	·		\vdash						
Ce	276.5	4	P								
Cryos, Accounts, B			c vap.300°K	0.39781 0.50320	2 2						
t _e °C	98.49	5	c _v vap.								
$T_{\mathbf{R}} = 0.7$			μ		لــــــــــــــــــــــــــــــــــــــ	+ gr	ams/100 gra	ms solven	t		
	CES: 1-Dow	2-A	PI 3-Lit. 4-	-Calc. from de	t. da						
SOURCE:		A	PI								
PURIFICA	TION:	A	Ρĭ								
LITERATU	JRE REFERE	NCE	S: 3 NBS 514;	3' Timmern	nans						
L											

									No. 20	
NAME	2,4-1	Dimet	hylpe	entane			STR	UCTURAL	FORMULA	4
Ī								сн.сн.сн.	ен сн.	
	1,	T.,				一		CH ₃ CH CH ₂ CH ₃		
Mole % Pur.	Ref		lecul		Molecular Weight 100.1	98		3	3	
			Ref.	T		Ref				Ref
F, P. °C	-119.2	42	2	dt/dP			ſ	l to		
F.P. 100%				°C/mm	0.3353	١. ا	g	<u>*</u> K		ŀ
B. P. *C 760 mm		00	١, ١	25°C BP	0.2252 0.04376	2	h	L		
100	80.5 25.3		2 2	t.	0.0369	5	f'.	to		l
30 10	1.2		4	30 mm	0.6031	4	g'	'•K_		1
1	-47.0		5	AHm cal/g	16.318	2	h'	/ 20045	0, 0272	-
Pressure				ΔHv cal/g 25°C	79.44	2	m n	300 to	0.0212	
mm 25°C	98.3 946.4		4 5	30 mm	78, 44 82, 54	5	٥		-0.0 ₆ 45	4
t _e Density	740.4		13	BP	70.36	2 5	m'	700 to	0.1062	4
g/ml 20°C	0.6	7270	2	t _e (d, e)	69.26	5	n'	11000 •K		4
dt 25 4 30		6832	2	AHv/T	19.21	5	٥'	<u> </u>	-0.0 ₆ 40	-
a 30	+	6393	+ -	d l 0 to	82,73	5		ace tension	10 15	١,
ь		9023 3 ⁸⁵⁵	4	-a ,90 €		5	yne	ss/cm. 20°C 30	18.15 17.17	2 2
Ref. Index				d' to				40	16.23	2
ⁿ D 20°C		8145 7882	2 2	d _c g/ml	0,239	2	Par	achor [P] 20°C	207 (١.
30		7617	4	II v mil/g	4.192 247.1	2 2	ľ	30	3 0 7.6 307.5	4
"C"	0.7	580	4	,, -	20824.	2		40	307.5	4
MR (Obs.)			2	P _c mm	20024.	-	F	L.1.%/wt.	307.2	5
MR (Calc. (nD-d/2)	34.5		5 2	25°C	0.9865	5	Exp	u.		
Dielectric	1.9		3	30 mm BP	1.000 0 0.9487	5		ersion	98.6	2
A 0 to	+	2621	2	t.	0.9418	5		h Point °C Point	-24.	5
B 1_115 °C	1192.0	41	2	t _c	0.270	2	M S			├─
C	221.6		2	ΔHc kcal/m ΔHf	1072.44 -56.17	2 2	Ultr	a V.		ì
A* 0 to B* 100 °C		6088 6	5	ΔFf	-0.49	2		ay Dif. ared		ł
K	-	•		Viscosity			!	bility in +	 	├─
t _k	. 			centistokes 7 15 °C	0.5538	3,	Ac	etone	œ	
t _x °C	;			30	0.5351	31		rbon tet. nzene	ο ο	Į
A' to					ĺ		Eth	er	œ e	Į.
B', L <u>*</u>	-		1	B ^V to	 			leptane anol	80 80	l
A'* to	,†		\vdash	A ^V °C	Ì		Wa	ter	~	ļ
B'* *($oldsymbol{ol}}}}}}}}}}}}}}}}}$	(B ^V) to	1		Wa	ter in	ļ	├
Ac 115 to	7.3	3247	4	(A ^V) °C					[ļ
Bc tc_C	270.5		4	c _p liq. °K					1	
Cryos. A°	0.0	3473	2	c _p vap300°K	0.39781	2			1	
consts. B°	0.0		2	400	0.50320	2				
te °C	87.9	B	5	c _v vap.	<u> </u>				L	<u> </u>
$T_R = 0.7$								ms/100 grai		t
REFEREN	_ES: 1-1	Jow	2-AI		Calc. from de	t. da	ta 5-	Calc. by for	mula	
SOURCE:	FION		AF							
PURIFICA			AF							
LITERATU	KE REF	ERE	NCES	3 NBS 514;	3' Timmerm	ans				

								N o. 21	
NAME _	3, 3-Dime	thylp	entan e			ST	RUCTURAL CH ₃		A
Mole % Pur. 99.		ecularmula		Molecular Weight 100,19	98		сн ₃ сн ₂ с сн ₃	сн ₂ сн ₃	
		Ref.	T	T	Ref.	T T			Ref.
F. P. °C	-134.46	2	dt/dP			ſ	to		
F.P. 100%			°C/mm			g			
B.P. °C			25°C B P	0.2674	2	h			
760 mm 100	86.064 29.241	2 2	t _e	0.04307	5	f'	to		
30	4, 36	4	30 mm	0.6215	4	g'	<u>*K</u>	1	
10	-14.4	2	ΔHm cal/g	16, 857	2	h'			1
1	-45.3	4	ΔHv cal/g	10.051	1	m	300 to	0.0272	4
Pressure mm 25°C	82,84	4	25°C	78. 76	2	n	_6 <u>0</u> 0_•K	0.0014	4
t _e	966.5	5	30 mm	81.94	4	٥	İ	-0.0 ₆ 45	4
Density	 	\vdash	BP	70.71 69.56	2 5	m'	700 to	0.1062	4
g/ml 20°C	0.69327	2	t _e (d, e)	69.55	5	n' o'	[10 <u>00 •</u> K		4
d ₄ 25	0.68908	2 4	AHv/Te	18.95	5			-0.0640	*
	0.68488	-	d 0 to	82.54	4		face tension		
a b	0.71004 -0.0 ₃ 820	4 4	_e_ _1 <u>00</u> °C	0.1375	4	dyn	es/cm. 20°C 30	19.59	2 2
Ref. Index	1.03020	1	d' to	1			40	17.67	2
n _D 20°C	1.39092	2		0, 239	31	Par	achor [P]		
- 2 5	1.38842	2	d g/ml	4. 183	3'		20°C	304.2	4
30	1.38595	4	vc ml/g tc °C	263.0	2		30 40	304.1 304.1	4
"C"	0.7527	4	P _c mm	22800.	2			307.2	5
MR (Obs.) MR (Calc.)	34.332 34.526	2 5	PV/RT	 	\vdash	Ext	L.1.%/wt.		
(nD-d/2)	1.04428	2	25°C	0.9898	5	-	u.		
Dielectric	1.937	3	30 mm BP	1.0000 0.9520	5 4		persion	97.1	2
A 5 to	6, 82667	2	te	0.9447	5		sh Point C e Point	-18.	5
B 130 °C_	1228.663	2	t _c				Spec.		├
С	225, 316	2	ΔHc kcal/m ΔHf	1072.57 -56.07	2 2		ra V.		
A* 5 to B* 105 °C	1.24360	5	ΔFf	-0.69	2		Ray Dif.		١,
K 105 0	1149. 27	"	Viscosity				ared	Yes	2
:			centistokes				ubility in [†] etone	o o	
t _k to			η ° ⊂				rbon tet.	x	
A'I to	<u> </u>	\vdash					nzene	00	
B'i °C			<u> </u>		1		her Heptane	ος ος	
C'			B ^V to C			Et	hanol	∞	
A'* to				-			ater ater in		
B'* °C	7.550/5		(B ^V) to						
Ac 130 to Bc t _c °C	7.55060	4	(A ^V) °C	-	\vdash				
Cc	299.3	4	c _p liq. °K					1	
Cryos. A° consts. B°	0.04418 0.0040	2 2	c _p vap.300°K 400	0.39781 0.50320	2 2			i i	
t _e ℃	94.53	5	c _v vap.	1					
$T_{\mathbf{R}} = 0.75$	T _c					+ g:	rams/100 gra	ms solven	t
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		AP	I						
PURIFICAT	ION:	AP	I						
LITERATUR	RE REFERE	NCES	S: 3 NBS 514;	3' ASTM 109)				
			. =						

									No. 22	
NAME	2, 2, 3	-Trin	nethy	lbutane				CH ₃		١
Mole % Pur.	Ref.		lecul rmul		Molecular Weight 100.1	98	c	сн ₃ с сно сн ₃ сн ₃	:H ₃	
			Ref.			Ref				Ref
F.P. °C	-24.91	2	2	dt/dP	1	\vdash	1	to		
F.P. 100%				°C/mm	ĺ		g	<u>*</u> K		l
B. P. °C				25°C BP	0.2227 0.04484	2	h ;			1
760 mm 100	80.88 24.46		2 2	t,	0.0377	5	f'	to		
30	-0.18		4	30 mm	0.6152	4	g'	•K_		
10 1	-18.8 -49.3		4 5	AHm cal/g	5.393	2	h'			_
Pressure	-47.3		٦	ΔHv cal/g			m	300 to	-0.0081 0.0 0 15	
mm 25°C	102.36	,	4	25°C	76.42	2	n	_600 °K	-0.0654	4
t _e	951.3		5	30 mm BP	80.10 69. 04	5 2				
Density				t _e	67.98	5	m'	700 to 1000 °K	0.1121 0. 0012	4
g/ml 20°C	0.69		2 2	1 te (u, e)	67.97	5	ö' !	12020 212	-0.0642	
dt 25 4 30	0.68		4	ΔHv/T _e	18.82	5	Sumfa			├
	0.70	705	4	d 0 to		5		ce tension /cm, 20°C	18.76	2
b	-0.03	825	4	├ ⋴ ,-¦ _% 🕏		5	*	30	17.77	2
Ref. Index	1.38	944	2	e' i •c				40	16.81	2
ⁿ D 20°C	1.38		2	d _c g/ml	0.254	2	Para	chor [P] 20°C	302.4	4
30	1.38	433	4	vc ml/g tc °C	3.9 32 258.3	2 2		30	302.1	4
"C"	0.75	35	4	P _c mm	22610.	2		40 Sugd	301.9 307.2	4 5
MR (Obs.)	34. 37		2	PV/RT	 	-	Fyn	L.1.%/wt.	301.2	
MR (Calc.) (nD-d/2)	34.52		5 2	25°C	0.9888	5	DAP.	u.		1
Dielectric	1.93		5	30 mm BP	1.000 0 0.9516	5 4		rsion	98.3	2
A 0 to	+		2	te.	0.9447	5	Flash Fire	Point °C	-22.	5
B 1125 °C	,		2	t _c	0.269	2				-
C	226.05	0	2	ΔHc kcal/m ΔHf	1071.78	2 2	M Spe		Yes	2
A* 0 to	1.21		5	ΔFf	-56.63 -0.17	2	X-Ra			
K L100 °C	1122.41			Viscosity	†		Infra		Yes	2
·	4			centistokes	i		Acet	,		
t _x to	ł		1	7 ℃	}		Cart	on tet.	œ	
A' to	 		\vdash		ľ		Ben: Ethe		80 80	l
B' ∟ °C	.}				 	├	n-He	ptane		}
C'			\vdash	B ^V to A ^V °C			Etha Wate		∞	1
A¹* to B¹* °C	[-j		Wate			İ
Ac 125 to	7.40	100	4							
Bc t C	1666.5		4		 	-				
Cc — —	288.7		4	c _p liq. ∘K	}	1				İ
Cryos, A° consts, B°	0.00 0.00		2	c _p vap.300°K 400	0.39781 0.50320	2 2	i			
t _e °C	88.73		5	c _v vap.			l			
$T_{\mathbf{R}} = 0.79$	T _c						+ gran	ms/100 gran	ns solven	ŧ
REFERENC	ES: 1-D	ow.	2-AF	PI 3-Lit, 4-(Calc. from det	t. da				
SOURCE:			AF							
PURIFICAT	ION:		AF	PI						
LITERATUI	RE REF	ERE	NCES	:						

								No. 23			
NAME	n-Octane					STRUCTURAL FORMULA					
							CH3(CH2)6	сн			
Mole % Pur. 99.	Ref. Mo	lecul rmula		Molecular Weight 114.2	224		3,0112,6	3			
, 1 u1.	7- 1- 110.	Ref.	0 18	weight 114.2	Ref.	ī			Ref		
F, P, °C	-56,795	2	dt/dP			f	to				
F.P. 100%			°C/mm			g	°K	,			
B. P. *C	125 //5		25°C BP	1.2698 0.04738	5 2	h	l				
760 mm 100	125.665 65.704	2 2	te	0.0364	5	_f,_	to				
30	39.28	4	30 mm	0.6613	4	g'	°K				
10 1	19.2 -14.9	2 4	∆Hm cal/g	43.397	2	h'					
Pressure	†		ΔHv cal/g	24.00		m	300 to	0.0265 0.0014			
mm 25°C	14.036	4	25°C 30 mm	86.80 85.63	2 5	0	, 000 K	-0.0646	4		
t _e	1064.5	5	BP	71.91	2	m'	700 to	0,1079	├		
Density g/ml 20°C	0.70252	2	t _e (d, e)	70.06 70.05	5 5	n'	1000 °K	0.0012			
dt 25	0.69849	2	ΔHv/T	19.45	5	٥'	İ	-0.0640	4		
	0.69445	4	d 20 to	90,50	4		face tension		_		
a b	0.71862 -0. 038 02	4	_e _ _140 °C	0.1479	4	dyn	es/cm. 20°C 30	21.76 20.77	2		
Ref. Index			d' to				40	19.80	2		
ⁿ D 20°C	1.39743	2 2	d _c g/ml	0,235	2	Par	achor [P]				
30	1.39269	4	V_mi/g	4.255 296.2	2		30°C	351.2 351.2	4		
"C"	0.7547	4		18726.	2		40	351.2	4 5		
MR (Obs.)	39.192	2	P _c mm	18720.	-		Sugd.	346.2	32		
MR (Calc.) (nD-d/2)	39.144 1.04617	5 2	25°C	1.0000	5	EXI	u. L.1.%/wt.	3.2 11.3	32		
Dielectric	1.948	3	30 mm BP	1.0000 0.9411	5 5	Dis	persion	98.0	2		
A 40 to	6, 92377	2	t _e	0.9299	5		sh Point °C	15.56	3²		
B [155 ℃	1355.126	2	tc	0.256	2		e Point	Yes	ī		
С	209.517	2	ΔHc kcal/m	1222.77 -59.74	2 2		Spec. ra V.	les	1		
A* 40 to	1.38276 1278.24	4	ΔFf	1.58	2		Ray Dif. ared	Yes	1		
к ———	24.	5	Viscosity				ubility in +	163	÷		
t _k [150 to	0.14684	4	centistokes り 60°C	0,5353	2	Ac	etone	∞ ∞			
tx 250 °C	353.	5	80	0.4602	2		rbon tet. enzene	× ×	ļ		
A' 15 to	7.47176	4	100 120	0.4022 0.3560	2 2	Et	her	- œ			
B' _ 40°C	1641.52 234.5	4	BV 20 to	392,25	4		Heptane hanol	oc oc	1		
A'* 15 to	1.87375	4	A ^V 75 °C	Z. 55168	4	W:	ater	~	ļ		
B'* 40 °C	1541.29	4	(B ^V) 75	387.11	4	ı	ater in		-		
Acl 155 to	7.30712 1648.0	4	(A ^V) 130	2.56694	4		cosity htistokes				
Bc t _c C	246.9	4	c _p liq. 20 °C.	0.52	31		30°C	0.7005	2		
Cryos. A°	0.05329	2	c vap.300°K	0.39703	2		70	0.4950	2		
consts. B°	0.0031	2	100	0.50217	2	l .					
t _e °C	138.25	5	c _v vap.	<u> </u>		L_	40	<u> </u>	<u></u>		
T _R = 0.757		2 4	DI 2 1 4 4	Cala (man 3	- د ه		Cala by for		it		
	ES: 1-Dow	AP:		Calc. from de	i. Q8	ia 5	-Caic. by for	IIIUIA			
SOURCE:	TON:	AP									
			S: 3 NBS Circ	514: 21 T'			. 22 NEDA 2	35			
	NOF ERE	.10E	. 3 NDS CITC	. 514; 5' Tin	nmer	mans	; 5" NFPA 3.	25			
<u>-</u>	···-						·				

F								No. 24	
NAME	2-Methylhe	ptan	e			STR	UCTURAL 1	FORMULA	
							כם כם וכם	١ ٢ ٢	
							CH ₃ CH (CH	2/40113	
Mole		lecul		Molecular Weight 114.2	,,		CH ₃		
% Pur.	F0	rmul		Weight 114.2	-				Ref.
	1 100 040	Ref.		T	Ref.				Ker.
F.P. *C F.P. 1007	-109.040	-	dt/dP *C/mm	į .		f g	to *K		
B, P, *C	-	\vdash	25°C	0.903	4	h			İ
760 mm	117.647	2	BP	0.04691 0.0364	2 5	f'	1		-
100 30	58.297 32.15	2	t _e 30 mm	0.6542	4	g¹	to to		
10	12.3	2		0.0342	-	h'			
1	-21.7	5	ΔHm cal/g	ļ	-	m	300 to	0.0265	4
Pressure	20 (27		ΔHv cal/g 25°C	83.02	2	n	_600 •K	0.0014	4
mm 25°C	20.637 1052.	5	30 mm	82.65	5	0		-0.0 ₆ 46	4
Density		-	BP	70.30 68.65	2 5	m'	700 to	0.1079	4
g/ml 20°0	0.69792	2	t _e (d, e)	68.56	5	n'	1 <u>1</u> 000°K	0.0012	4
dt 25 4 30	0.69392	2	ΔHv/Te	19.46	5	0'		-0.0640	4
	0.68991	4	d 25 to		5		ce tension		
a b	0.71390 -0.0 ₃ 794	4	e <u>l 130</u> °C		5	dyne	s/cm. 20°C 30	20.60 19.68	2
Ref. Index	 	╁	d' to				40	18.77	2
n _D 20°0		2			2	Para	chor [P]		
25	1.39257	2	d _c g/ml v _c ml/g	0.234 4.273	2	1	20°C	348.7	4
"C"	1.39020	4	tc °C	288.	2		30 40	348.8 348.8	4
	0.7551	4	P _c mm	18848.	2	l	Sugd.		5
MR (Obs.) MR (Calc.		2 5	PV/RT			Exp.	L. 1. %/wt.		
(nD-d/2)	1.04598	2	25°C 30 mm	1.0000 1.0000	5	Dien	u. ersion	98.5	2
Dielectric	1.946	5	BP	0.9487	4		h Point °C	70.5	ــــــــــــــــــــــــــــــــــــــ
A 35 t		2	t _e	0.9386 0.263	5 2		Point		1
B (_150°C	2 1337.468 213.693	2	t _c	<u> </u>	2	M S	ec.		
A* 35 to		5	ΔHc kcal/m ΔHf	1221.08 -60.98	2	Ultra	v.		
B* 140 °C		5	ΔFf	0.92	2	Infra	y Dif. red	Yes	2
к — —	_	1	Viscosity				oility in +		
1 tk	<u>-</u>		centistokes 7 °C			Ace	tone	oc	
t _x •(,				bon tet.	ος ος	
A' O to	7.66293	5		1		Eth		œ	
B' _ 35 °C	2 1729.19 247.4	5	B ^V to	 	-		eptane	00	l
A'* 10 to		5	Av i c	ì		Wat	anol er	∞	
B'* 35 °C		5	(BV) to	1			er in		<u></u>
Ac 150 to	7.24113	4	(A ^V) °C						
Bc tc_'	C 1581.7	4	c _p liq. °K		_				1
Cc —	245.3	4	11 -						
Cryos, A'consts, B'		2	c _p vap.300°K 400	0.39703 0.50217	2 2				
t _e °C	129.66	5	c, vap.	0.30211	-				
$T_R = 0.7$			L	L	L	+ ~~~	ms/100 gran	ne selve-4	<u> </u>
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	t. de	ta 5	Calc. by for	mula mula	
SOURCE:		AI					by 1011		
PURIFICA	TION:	AI	 						
<u> </u>	RE REFEREI								
	NEFEREI	-CEC	••						
1									
1									
1									
]									

No. 25 3-Methylheptane NAME STRUCTURAL FORMULA CH_3CH_2CH $(CH_2)_3CH_3$ CH₂ Mole Ref. Molecular Molecular $C_{8}H_{18}$ Weight 114, 224 % Pur. 99.99 2 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -120.500 2 dt/dP f to °C/mm g °K 25°C 0.9478 5 B. P. °C h BP 0.04712 2 760 mm 118.925 2 0.0361 5 ſ١ to 100 59.358 2 g' <u>°К</u> 30 33.15 4 30 mm 0.6556 4 10 13.3 2 h! ∆Hm cal/g 23,813 2 5 -21.0 0.0265 300 to m ∆Hv cal/g Pressure _6<u>0</u>0_**°K** 0.0014 n 4 25°C 83.35. 2 mm 25°C 19.582 4 o ·0.0₆46 4 30 mm 83.01 te 1070.6 5 BP 71.30 2 m' 700 to 0.1079 4 Density 5 t_e t_e (d, e) 69.60 n' 1000 °K 0.0012 g/ml 20°C 0.70582 2 69.56 5 o' -0.0640 4 d₄ 25 0.70175 2 ΔHv/T_e 5 19.64 30 0.69767 4 Surface tension Т 25 87.54 5 0.72208 to 4 dynes/cm. 20°C 21.17 2 e di 130 °C 0.1366 5 ь -0.03809 4 30 20.24 2 to 19.32 40 2 Ref. Index e¹ ⁿD 20°C 1.39848 [P] Parachor dc g/ml 0.239 2 25 1.39610 2 20°C 347.2 4 vc ml/g t_°C 4.185 2 30 1.39374 4 30 347.3 4 292. 2 t_c 40 347.4 4 "C" 0.7531 4 P_c mm 19456. 2 Sugd 346.2 5 MR (Obs.) 39.100 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 39.144 25°C 1.0000 5 (nD-d/2)u. 2 1.04557 30 mm 1,0000 5 97.5 Dispersion 2 Dielectric 1.956 5 BP 0.9601 4 Flash Point C 5 6. 0.9504 30 to 5 6.89944 2 Fire Point 0.264 2 В 1_150 °C 1331.530 2 M. Spec. С AHc kcal/m 2 1221.76 212.414 2 Ultra V. ΔHf -60.34 2 A* 30 to 1.33027 5 X-Ray Dif. ΔFf 2 1,12 B* 140 °C 1246.6 2 Infrared Yes ĸ Viscosity Solubility in centistokes Acetone to œ Carbon tet. °C 00 t_x_ Benzene 00 0 to 7,77698 Ether œ В' 30 °C 1797.39 5 n-Heptane B_v | C' 252.16 5 to Ethanol °C Water 10 to 2.15798 5 Water in B'* 30 °C 1684.52 (B^V)| to Ac | 150 to (A^V) 7.30319 5 °C Bc tc °C 1640.9 c_p liq. ۰ĸ Cc 252.4 5 c_p vap.300°K Cryos. A° 0.0587 2 0.39703 2 consts. B° 0.50217 400 c vap. t_e ℃ 5 131.67 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

Pressure mm 25°C 20.55 4 25°C 83.01 2 n 600°K 0.0014	······								····	No. 26	
Mole Mole	NAME	4-Met	hylhe	p tan e	e			STRUC	TURAL 1	FORMULA	Ł
Mole Molecular Communication Molecular Communication Molecular M	ſ							CH. (CH_)_CH(CH_)_CH_	
Ref.					T			011311			
Ref.		Ref.					224		03		
F. P. *C	/ Fui. //.	// <u> -</u>	10.			Weight 111.	_				Ref
F. P. 100% S. P. ° C T60 mm	E.D. 46	1 120 0		$\overline{}$	1: /15		1.01		T	1	-
B.P. C 760 mm 117, 709 2 100 58, 347 2 30 mm 0.04695 2 10 12, 4 2 2 22.0 4 2 30 mm 25°C 20.55 4 25°C 20.55 4 25°C 20.55 4 20.6696 4 25°C 20.70055 2 4 30 0.69646 4 25°C 2			,,,					, ,			l
1760 mm 117, 709 58, 347 2 30 mm 0.6534 4 10 12.4 2 1 -22.0 4 AHm cal/g 22.675 2 1 -28. AHm cal/g 22.675 2 1 -28. AHm cal/g 22.675 2 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 AHm cal/g 22.675 2 Amm cal/g 22.675 2 23.07 2 2 2 2 2 2 2 2 2		†			25°C			,			
30 32, 23 4 10 12, 4 2 1 -22, 0 4 AHm cal/g 22,675 2 7 -60,005 4 4 6 -22, 0 4 -22, 0 4 -22, 0 4 -22, 0 4 -22, 0 4 -22, 0 4 -22, 0 4 -22, 0 5 -22, 0 -									to	<u> </u>	\vdash
10]	ļ
Pressure mm 25°C						+	2	h'			l
The content of the		-22.0		4		+		m		0.0265	
The color of the		20.59	,	4	25°C				_600 •K		
Density g/ml 20°C 0.70463 2 t (d,e) 69.18 5 0.0012								ــــــــــــــــــــــــــــــــــــــ		-0.0610	<u> </u>
A 30	Density	†								0.1079	
A 30					te (d, e)	69.18	5		11000 F		
a	d ₄ 30				ΔHv/T _e	19.61	5	1		1.16.1	<u> </u>
Ref. Index n.D 20°C 1.39792 2 1.39553 2 2 1.39553 2 2 290. 2 290. 2 2 30 347. 2 30 30 347. 3 30 30 347. 3 30 30 347. 3 30 3	a	0.72	093	4						21.00	2
Ref. Index n	Ъ	-0.0	3810	4			5	*	30	20.07	2
25					e' °C			<u> </u>		19.15	2
30 1.39316 4		1			d _c g/ml			Parach		347 1	4
MR (Obs.) 39, 117 2 MR (Calc.) 39, 144 5	30				I v mi/g				30	347.2	4
MR (Obs.) 39, 147 5	"C"	0.7	533	4	P mm	1	1 1				4 5
Mark Calc. 39.144 5 2 25°C 1.0000 5 1.0000 5 2 30 mm 1.0000 5 4 4 25 to 6.90065 2 4 2 4 25 to 6.90065 2 4 2 4 2 2 2 2 2 2						+	 -	Evn I.		340.2	
Diselectric 1.954 5					25°C			Exp. D		}	1
A 25 to 6.90065 2 te 0.9477 5 Fire Point C 6.0 B 150 °C 1327.661 2 2 2 2 2 2 2 2 2								Disper	sion	97.5	2
B 150 °C 1327.661 2 1212.568 2 2 AHc kcal/m 1221.89 2 AHf Color Colo				-	_	0.9477	5			6.0	5
A* 25 to 1.33746 5 AHf AFf -60.17 2 X-Ray Dif. Infrared Yes A* 25 to 1.33746 5 AHf AFf 1.86 2 B* 140 °C 1244.16 5	B 150°C	1327.60	51	2						 	├
A		+		-						l	ĺ
Viscosity centistokes Visc											
Centistokes A		-	^		Viscosity					Yes	2
A' 0 to 7.8970 5 1860.28 5 5 257.54 5 8 4 0 to A' 0 to 2.2715 5 8 4 0 to A' 0 to A' 0 to A' 0 to A' 0 to 0		-				. [· y ·		1
A' 0 to 7.8970 5 1860.28 5					"	` !				1	J
B' 35 °C 1860.28 5 B' to A' 10 to 2.2715 5 B' to Water Water in Ac 150 to 7.31607 4 (A') °C Co Co Co Co Co Co To T	A' 0 to	7.89	70	5		1			ne	1	i
At* 10 to B* 35 °C 1743.38 5		1860.28			PV I	+	\vdash	n-Hep		∞	
B* 35 °C 1743. 38 5 (BV) to Water in Ac 150 to 7. 31607 4 (AV) °C Cc 253.5 4 c _p liq. °K Cryos. A° 0.0563 2 c _p vap.300°K 0.39703 2 t _e °C 130.15 5 c _v vap. T _R = 0.75 T _C T _c vap. 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API		+			A ^V C				01	, ao	
Ac 150 to 7.31607 4 (A ^V) °C C _p liq. °K C _p vap.300°K 0.39703 2 C _v vap.						-			in		<u> </u>
Bc tc °C 1645.2 4 cp liq. °K Cryos. A° 0.0563 2 cp vap.300°K 0.39703 2 te °C 130.15 5 cv vap. TR = 0.75 Tc + grams/100 grams solven REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API		7.3			1	1	1				
Cryos. A° consts. B° 0.0563 2 c _p vap.300°K 0.39703 2 400 0.50217 2 t _e °C 130.15 5 c _v vap. ** TR = 0.75 T _C ** REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API	Bc tc_°C	1645.2			<u> </u>						
consts. B°			. (2		l -						
T _R = 0.75 T _C + grams/100 grams solven REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API	consts. B°	ļ			400	0.39703 0.50217					
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API	te °C		5	5	vap.	J		L		L	
SOURCE: API PURIFICATION: API											<u> </u>
PURIFICATION: API		بين: I-D	OW .			Calc. from de	t. da	ta 5-Cal	c. by for	mula	
		TON									
ATTERATURE REFERENCES:			ED ES								
		KE KEF	ek en	CES							

No. 27 3-Ethylhexane NAME STRUCTURAL FORMULA CH3CH2CH (CH2)2CH3 Ċ₂H₅ Mole Molecular Molecular $C_8 H_{18}$ Weight 114.224 % Pur. Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g °Κ 0.9283 B. P. °C h ВP 0.04719 2 760 mm 118.534 2 0.0359 5 f¹ te tο 100 58,902 2 °K g' 4 0.6559 4 30 32.68 30 mm 10 12.8 2 h١ AHm cal/g -20.47 5 300 to 0.0265 m AHv cal/g Pressure n 1 600 °K 0.0014 25°C 83.19 mm 25°C 20.03 o -0.0646 4 5 30 mm 82.70 1080.9 5 t_e ВP 71,70 2 m 700 to 0.1079 Density 5 70.52 t_e (d, e) 0.0012 n' 1000 °K g/ml 20°C 5 0.71358 2 70.01 ٥' 4 -0.0640 dt 4 25 0.70948 5 AHv/Te 19.88 30 0.70537 Surface tension Т d 30 86, 88 5 to 0.72996 4 dynes/cm. 20°C 21.51 0.1281 ᇷᅴ 1<u>30 °C</u> 5 ь -0.0381 4 20.58 30 2 40 19.64 2 Ref. Index e¹ 20°C 1.40162 ^{n}D [P] Parachor d_c g/ml 0.245 2 25 1.39919 2 20°C 345.2 ,ς ... 4 ml/g 4.080 2 30 1.39644 5 30 345.1 4 ^tc 294. 5 40 344.9 4 "C" 0.7498 5 P_c mm 19914. 5 Sugd. 346.2 5 MR (Obs.) 38.944 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 39.144 25°C 1.0000 (nD-d/2)1.04483 2 u. 30 mm 1.0000 5 Dispersion 96.4 2 0.9687 Dielectric 1.964 5 BP 4 Flash Point C -4. 5 0.9596 5 A 25 to 6.89098 2 Fire Point 0.264 2 B (150 °C 1327.884 2 M. Spec. С 212,595 2 2 ∆Hc kcal/m 1222, 19 Ultra V. ΔHf -59.88 A* 25 to 1.30712 5 X-Ray Dif. ΔFf 1.80 B*[140 °C 1239.30 Infra red Yes 2 ĸ Viscosity Solubility in centistokes Acetone to t_k Carbon tet. °C 00 Benzene œ ۸' 0 to 7.30017 Ether œ B' 35 °C 1529.6 5 $\overline{B_{\mathbf{v}}^{\mathbf{v}}}$ n-Heptane œ C' 230. 5 Ethanol œ °C Water 1.72722 A!* 10 to 5 Water in B'* 35 °C 5 (B^V)| 1437.6 to Acl 150 to 7.29602 5 (AV) °C Bc tc °C 1639.8 5 c_p liq. °K Cc' 253,2 5 c_p vap.300K Cryos. A 0.39703 2 consts. B° 400 0.50217 2 c vap. t_e °C 131,67 5 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API **PURIFICATION:** LITERATURE REFERENCES:

								No. 28	
NAME	2, 2-Di	methylh	exane			STR	UCTURAL I	FORMULA	1
Mole	Ref.	Molecu	lar	Molecular	\dashv		сн ₃ с - (сн	₂) ₃ CH ₃	
% Pur. 99.		Formu		Weight 114.2	24			····	
		Ref			Ref.				Ref
F.P. C F.P. 100%	-121, 18	2	dt/dP *C/mm			f g	300 to 600°K	0.0445 0.0013	
B. P. *C			25°C BP	0.5772 0.04650	2	h		-0.0637	5
760 mm 100	106.84 48.22		t.	0.0366	5	f¹	to		
30	22.54	4	30 mm	0.6417	4	g'		1	1
10 1	3.1 -28.8	2	ΔHm cal/g			h'	l	L	
	-20.0	-+-	ΔHv cal/g			m	300 to	0.0265	
Pressure mm 25°C	34.05	4	25°C	78.02	2	n o	_600 °K	0.0014 -0.0 ₆ 46	
t _e	1033.1	5	30 mm BP	79.03 67.7	5 2	ļ	<u> </u>		├
Density	T .] t_	66.23	5	m' n'	700 to 1000 °K	0.1079 0.0012	
g/ml 20°C	0.69		l e (4, 6)	66. 19	5	0'	1 1 1 0 0 0 IL	-0.0640	
d ^t 25 4 30	0.68		ΔHv/T _e	19.33	5	<u> </u>	<u> </u>	- 67	<u> </u>
	0,71	190 4	d 20 to		4		ace tension s/cm. 20°C	19.60	2
Ъ	-0.03		<u> </u>		4	3,	30	18.69	2
Ref. Index] • · · · · · · · · · · · · · · · · · ·				40	17.80	2
ⁿ D 20°C	1.39		d _c g/ml	0.245	2	Par	chor [P]		į
30	1.38		II A - 1111/6	4.088	2		20°C 30	{	
"C"	0,75		7) ** ~	279.	2		40		
MR (Obs.)	39.25		P _c mm	19456.	2			346.2	5
MR (Calc.	39.14		PV/RT	0.00147	_	Exp	L.1.%/wt.	1	
(nD-d/2)	1.04		25°C 30 mm	0.99147 1.0000	5	Diar	u. ersion	99.7	2
Dielectric	1.94	2 5	BP	0.9580	4		h Point °C	-3.	5
A 20 to		715 2	t _e	0.9490 0.264	5 2		Point	-3.	,
B 1740.€			t _c			M S	Dec.		
	215.07		ΔHc kcal/m ΔHf	1218.88	2 2	Ultr	a V.	ŀ	l
A* 20 to B* 130 °C			ΔFf	0.72	2	Infra	ay Dif.	Yes	2
к — — —	-		Viscosity				bility in +	100	-
t.	-		7 °C	.]			tone	∞	1
tk to			7 °C	` 			bon tet.	∞ ∞	
A' to	 		1	ł	l	Eth	nzene er	- so	1
B' L _ *C	<u>:</u>		-v	 	├	n-F	leptane	∞ ∞	
C1	 		B ^V to			Eth Wa	anol	∞	
A'* to B'* *C		1	H-v	-			ter in		
Ac 140 to		434 5	1 v. '	1					
Bc t C	1580.0	5			-	ľ		1	1
Cc	256.0	5	c _p liq. 300°K 400	0. 39891 0. 50224	5				
Cryos. A° consts. B°	0.03	54 2	c _p vap.300°K 400		2 2				
t _e °C	118.07	5	c _v vap.		<u> </u>	L			
$T_R = 0.7$	5 Т _с					+ gra	ms/100 grai	ns solven	t
REFERENC	ES: 1-De	ow 2-A	PI 3-Lit. 4-	Calc. from det	t. da	ta 5-	Calc. by for	mula	
SOURCE:		A	PI						
PURIFICAT	TION:	A	PI						
LITERATU	RE REFE	ERENCE	S:						

No. 29 2, 3-Dimethylhexane STRUCTURAL FORMULA NAME CH3CH CH (CH2)2CH3 ĊH₃ĊH₃ Mole Ref. Molecular Molecular Weight 114.224 C8H18 % Pur Formula Ref. Ref. Ref. F. P. °C dt/dP f 300 to 0.0445 5 F.P. 100% °C/mm _6<u>0</u>0_**°K** 0.0013 5 g 25°C 0.8131 4 B. P. °C 5 h -0.0637 ВP 0.04724 2 115.607 760 mm 2 0.0363 5 ſ١ 100 55.965 2 g' °<u>K</u> 30 29.77 4 30 mm 0.6548 4 10 9.9 2 h' ∆Hm cal/g -25.45 4 1 300 to 0.0265 m AHv cal/g Pressure | 600 °K n 0.0014 25°C 81.17 2 mm 25°C 23, 44 4 o -0.0646 4 30 mm 81.29 4 te 1066.2 5 BP 70.20 2 0.1079 m' 700 to 4 Density 68.59 5 te (d, e) n' g/ml 20°C 1000 •K 0.0012 0.71214 2 68.57 5 ۰, 4 -0.0640 ď4 25 0.70809 2 ΔHv/Te 5 19,52 30 0.70403 4 Surface tension 30 85.14 d 4 to 0.72832 4 dynes/cm. 20°C 20.99 2 130 0.1293 4 ь -0.03804 4 30 20.07 2 d٦ to 40 19.16 2 Ref. Index e' °C 20°C 1.40113 [P] nD d g/ml vc ml/g t °C Parachor 0.248 2 1.39880 25 2 Z0°C 4.036 2 30 1.39625 4 30 t_c 293. 2 40 "C" 0.7509 4 P_c mm 20216. 2 Sugd. 346. 2 5 MR (Obs.) 38.981 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 39. 144 5 25°C 1.0000 (nD-d/2)5 u. 1.04506 2 30 mm 1,0000 5 Dispersion 97.0 2 Dielectric 1.963 5 BP 0.9639 4 Flash Point C 0.9546 6.87004 A | 25 to 2 Fire Point 0.264 2 2 B (150 °C 1315.503 M. Spec. c 2 214.157 AHc kcal/m 1221,45 2 Ultra V ΔHf -60.40 A* 25 to 5 2 1.29661 X-Ray Dif. ΔFf B* 140 °C 1228.9 2.17 Infrared Yes 2 ĸ Viscosity Solubility in centistokes Acetone œ to t_k | t_x | Carbon tet. œ °C Benzene œ 0 to 8.51379 Ether œ В' 30 °C 2239.64 4 n-Heptane œ B_v C 288.51 4 to Ethanol 00 °C A'* 10 to B'* 30 °C 2.84821 Water 4 Water in (B^V) 2097.36 4 to Acl 150 to 7.27817 5 (AV) °C 1630.7 Bc Ltc_ °C c_p liq. 300°K 0.39891 5 255.6 5 400 0.50224 Cryos. A° cp vap.300°K 0.39703 2 consts. B° 400 0.50217 2 c, vap. t_e °C 128.23 5 TR $= 0.75 T_{\rm c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 5-Calc. by formula 4-Calc. from det. data API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

						No	. 30
NAME	2,4-Dime	ethyll	nexane			STRUCTURAL FORM	IU LA
Γ						CH ₃ CH ₃	
	$\overline{}$				\neg	CH ₃ CH CH ₂ CH CH ₂	CH ₃
Mole	Ref. Mo	lecul		Molecular Weight 114.2	24		
% Pur.	1 1 1 50	_	<u>a 0 10 </u>	weight 114.2	Ref	 	Ref.
E D 46		Ref.		Γ	Kei		
F.P. °C F.P. 100%	 	+	dt/dP *C/mm	İ			0445 5 0013 5
B. P. *C	 	+-	25°C	0.6388	4	"	0637 5
760 mm	109.429	2	BP	0.04664 0.0365	5	f' to	63.13
100 30	50.585 24.77	2 4	t _e 30 mm	0.6451	4	g' 'K	
10	5.2	2	ΔHm cal/g	0.033	Ė	h'	
1	-26.9	4	ΔHv cal/g		\vdash		0265 4
Pressure mm 25°C	30, 35	4	25°C	79.02	2		0014 4 0,46 4
t _e	1042.0	5	30 mm BP	79.82 68.5	4 2	-0.	0646 4
Density	1		t_	66.99	5		1079 4 0012 4
g/ml 20°C	0.70036 0.69620	2	e (4,0)	66. 95	5		0012 4 0 ₆ 40 4
dt 25 4 30	0.69203	4	ΔHv/T _e	19.41	5		<u> </u>
a	0.71698	4	d 25 to	83.13 0.1337	4 4	Surface tension dynes/cm. 20°C 20.	05 2
ь	-0.03824	4	-a, - 120 ° €	0.1337	*	30 19.	13 2
Ref. Index		1.	e' i °C			40 18.	22 2
n _D 20°C	1.39534 1.39291	2 2	d _c g/ml	0.245	2	Parachor [P]	1
30	1.39050	4	v _c ml/g t _c °C	4.080 282.	2 2	30	
"C"	0.7533	4	P _c mm	19608.	2	40 Sugd. 346.	2 5
MR (Obs.)		2	PV/RT	7,000.	-	Exp. L.1.%/wt.	- + -
MR (Calc.) (nD-d/2)	39.144 1.04516	5 2	25°C	0.9907	5	u.	
Dielectric	1.947	5	30 mm BP	1.0000 0.9591	5 4	Dispersion 97.	8 2
A 15 to		2	te	0.9501	5	Flash Point °C -2. Fire Point	5
B <u>145</u> °C	1287.876	2	tc	0, 264	2	M Spec.	
С	214.790	2	ΔHc kcal/m ΔHf	1220, 15 -61, 47	2 2	Ultra V.	
A* 15 to B* 130 °C		5	ΔFf	0.89	2	X-Ray Dif. Infrared Ye	s 2
к — — —	-		Viscosity			l	8 2
t to	-		centistokes 7°C	İ		Solubility in + 0	1
t _x t _o	:		7 °⊂			Carbon tet. ∞	
A' to		t		r		Benzene ∞ Ether ∞	1
B', ∟ °			B" to		-	n-Heptane ∞	
A¹* to	+		B to C			Ethanol ∞ Water	ĺ
B'* °C			(B ^V) to	1		Water in	
Ac 145 to	7.29625	5	(A ^V) °C				
Bc t _c °C	1624.8 - 258.9	5	c _p liq. 300°K	0. 39891	5	1	
Cryos. A°	230.7	+	li - 400	0.50224	5		
consts. B°			c _p vap 300°K	0.39703 0.50217	2 2		1
t _e °C	121.02	5	c _w vap.				
$T_R = 0.7$	5 T _c			L		grams/100 grams so	lvent
	ES: 1-Dow	2-AI	PI 3-Lit. 4-C	alc. from det	t. dat		_ / Vasb
SOURCE:			PI				
PURIFICAT	ION:	P	PI				
LITERATU	RE REFERE	NCES	5:				

								No. 31	
NAME	2,5-Dime	thylh	nexane			ST	RUCTURAL	FORMUL	A
							СН3	сн3	
Mole % Pur. 99		lecul mul		Molecular Weight 114,22	24	c	:н ₃ сн (сн ₂) ₂		
		Ref.			Ref.				Ref.
F. P. °C	-91.200	2	dt/dP			f	300 to	0.0445	5
F. P. 1009	<u> </u>	├-	*C/mm 25*C	0.6359	4	g	'_600_•K	0.0013	5
B. P. °C 760 mm	109.103	2	BP	0.04646	2	h	 	-0.0 ₆ 37	5
100 30	50.468	2	t _e	0.0364	5 4	f' g'	to *K		
10	24.74 5.3	2	30 mm	0,6432	 	h'			
1	-26.76	4	ΔHm cal/g	+	\vdash	m	300 to	0.0265	4
Pressure mm 25°C	30, 41	4	ΔHv cal/g 25°C	79.21	2	n o	_6 <u>0</u> 0_•K	0.0014	4
t _e	1040.3	5	30 mm BP	80.03 68.60	2			-0.0646	*
Density			t.	67.08	5	m'	700 to	0.1079 0.0012	4
g/ml 20°0	0.69354	2	le (a, e)	67.04	5	٥'	<u> </u>	-0.0640	4
d ₄ 25	0.68513	4	ΔHv/T _e	19.46	5	Sur	face tension		
a b	0.71032	4	d 25 to e 120 °C	83.39 0.1355	4 4		es/cm. 20°C	19.73	2
Ref. Index	-0.03831	•	d' to			,	30 4 0	18.82 17.92	2
n _D 20°0	1.39246	2		0,239	2	Par	achor [P]		
25	1.39004	2	d g/ml vc ml/g t °C	4.185	2		20°C 30	346.8 347.6	4
"C"	0,7552	4	`	279.	2		40		
MR (Obs.	39.260	2	P _c mm	19000.	2	<u> </u>		346.2	5
MR (Calc. (nD-d/2)		5 2	25°C	0.9904	5	Exp	u. L.1.%/wt.		
Dielectric	1.04569	5	30 mm BP	1.0000 0.9585	5 4		persion	99.0	2
A 20 to		2	ll t	0.9495	5		sh Point C e Point	1.	5
B (140 °C		2 2	t _c	0.264	2		Spec.		
A* 20 to		5	ΔHc kcal/m ΔHf	1219.37 -62.26	2 2	Ult	ra V.		
B*[130 °C		5	ΔFf	0.59	2		Ray Dif. ared	Yes	2
K c			Viscosity centistokes				ability in +		
t _k to			່າ 15 °C	0.7340	3		etone rbon tet.	ος ος	
t _x °C		-	30	0,6305	3	Be	nzene	∞	
B'			B _v to				her Heptane	οο οο	
C'		-	A to				hanol ater	oc	
A'* to B'* °C			(B ^V) - to	-			ter in		
Ac 140 to		5	(A ^V) °C						
Bc tc °C	1570.2 251.5	5 5	c _p liq. 300°K	0.39891	5	1			
Cryos, A		2	H - 400	0.50224 0.39703	5 2				
consts. B		Ĺ	400	0.50217	2				
t _e °C	120.58	5	c _v vap.			<u>L</u>		l	
$T_R = 0.7$		<u> </u>	DI 2 II.				ams/100 gra		t
	CES: 1-Dow	2-A A	PI 3-Lit. 4- PI	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE: PURIFICA	TION.		PI						
			S: 3 Timmerm	nans			· · · · · · · · · · · · · · · · · · ·		
1									
!									

				· · · · · · · · · · · · · · · · · · ·			No. 32	
NAME	3, 3-Dimet	hylhe	xane			STRUCTURAL	FORMU LA	L
						CH ₃		
Mole	Ref. Mo	lecul		Molecular	ļ	сн ₃ сн ₂ с - (с	2 ^H 2 ^J 2 ^{CH} 3	
% Pur.		rmul		Weight 114.2	24	ĊH ₃		
		Ref.			Ref			Re
F.P. ℃	-126.10	2	dt/dP		П	f to		
F.P. 100%			*C/mm		.4	g <u> •K</u> _		
B. P. *C	111 060	2	25°C BP	0.6825 0.04741	2	h		L
760 mm 100	111.969 52.16	2	te	0.03679	5	f' to		
30	25.93 6.1	4 2	30 mm	0.65557	4	g' '°K_	4	
10 1	-27.	5	ΔHm cal/g			h'		L
Pressure		\vdash	ΔHv cal/g			m 300to n 600°K	0.0265	4
mm 25°C	28.611	5	25°C 30 mm	79.20 79.15	5	0 -000 11-	-0.0646	
t _e	1053.406	5	BP	68.5	2	m' 700 to	0.1079	4
Density g/ml 20°C	0 .71000	2	te te (d, e)	66.89	5 5	n' 1000°K	0.0012	4
at 25	0.70596	2	ΔHv/T _e	19.26	5	o'	-0.0 ₆ 40	4
4 30	0.70192	4	d 25 to	82,36	4	Surface tension		_
a b	0+72614 -0.0 ₃ 80	4	e i 125 °C		4	dynes/cm. 20°C	20.63	2
Ref. Index	-0.0300	+-	d' 10 to	80.61	5	30 40	19.72 18.81	2 2
n _D 20°C	1.40009	2		 	5	Parachor [P]		
25 30	1.39782 1.39526	2 4	d g/ml v ml/g	0.254 3.940	2	20°C	342.9	4
"C"	0.7513	+	tc •C	277.9	5	30 40	343.0 343.0	4
MR (Obs.)		2	P _c mm	17388.	5	Sugd.	346.2	5
MR (Calc.)	39.009 39.144	5	PV/RT	1 0000	4	Exp. L.1.%/wt.		
(nD-d/2)	1.04509	2	25°C 30 mm	1.0000	5	u. Dispersion	97.3	2
Dielectric			BP	0.96198	4 5	Flash Point °C	/	一
A 25 to B 140 °C	6.85121 1307.882	2	te t _C	0.95282 0.264	2	Fire Point		
ç 🗀 🚾 C	217.439	2	ΔHc kcal/m	1219.97	2	M Spec.		
A* 25 to	1.28231	5	ΔHf	-61.58	2 2	Ultra V. X-Ray Dif.		
B* ∟135 °C	1221.27	5	ΔFf	1,23	-	Infrared		
c			Viscosity centistokes	ł	ll	Solubility in +		
¹k 100			η ∘c			Acetone Carbon tet.		
t _x i		1		ļ		Benzene		j
A' 0 to B' <u>35 °C</u>	7.2390 1503.4	5				Ether n-Heptane	,	
<u>c' </u>	235.	5	B ^v l to			Ethanol		
A'* 10 to	1.64796	5	A ^V i C			Water Water in		
B'* 35 °C		5	(B ^V) to	l		Water an	 	\vdash
Ac 140 to Bc t _c °C	7.26174 1617.9	5	(A ^V) °C	ļ	\sqcup			
Cc'	257.6	5	c _p liq. °K					
Cryos. A° consts. B°	0.04	2	c _p vap.300°K 400	0.39703 0.50217	2 2			
te °C	124.168	5	c _w vap.		1 1			1
$T_{R} = 0.75$						grams/100 gram	ms solveni	_
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from de	t. dat			
SOURCE:		Al						
PURIFICAT	ION:	Al	PI					
LITERATUR	E REFERE	NCES	:					

										No. 33	·
NAME		3, 4-D	imet	hylhe	exane			ST	RUCTURAL	FORMUL	A
									СН ₃ СН	[₂	
		Т			T		\neg	C	н,сн,сн сн		
Mole		Ref.		ecul		Molecular	.	-	301201	23	
% Pur.		لـــــــــــــــــــــــــــــــــــــ	For	mula		Weight 114.2					-
	_			Ref.			Ref.	<u> </u>		T	Ref.
F.P. ℃ F.P. 100%	+			\vdash	dt/dP °C/mm	1		f	300 to	0.0445	
B. P. °C	+			\vdash	25°C	0.8713	4	g	l ' <u>_60</u> 0_° <u>K</u> _	0.0013	ł
760 mm	1	17.72	5	2	BP	0.04752 0.03643	5	h		-0.0 ₆ 37	13
1 0 0 30		57.69 31.31	3	2 4	t _e 30 mm	0.6596	4	f' g'	to °K		
10		11.3		2		0.0370	+ -	h'			
1	٠	-22.33		4	ΔHm cal/g	 		m	300 to	0.0265	4
Pressure		21 /4		١. ا	ΔHv cal/g 25°C	81.65	2	n	<u>_600 °K</u>	0.0014	4
mm 25°C	10	21.64)66.3		4 5	30 mm	81.52	5	٥	! 	-0.0 ₆ 46	4
Density	+			\vdash	BP	70.20 68.72	2 5	m'	700 to	0.1079	4
g/ml 20°C	;	0.71		2	te te (d, e)	68.55	5	n' o'	[10 <u>00 •</u> K	0.0012	
d ₄ 25		0.71 0.71		2 4	AHv/Te	20.08	5	i	L	-0.0640	1
a	+	0.73		4	d 30 to	85.63	5		face tension	21 64	2
ь		-0.03		4	_e130 °C		5	gyn	es/cm. 20°C 30	21.64 20.71	2
Ref. Index					e' °C				40	19.79	2
ⁿ D 20°C	7	1.40		2 2	d _c g/ml	0.253	2	Par	achor [P]	343.3	4
30		1.39		4	V _C m1/g	3.957	2		20°C 30	343.2	4
"C"	T	0.74	86	4	1 _	298. 20824.	2		40 Sugard	342.6	4
MR (Obs.)		38.84		2	P _c mm PV/RT	20024.	-			346.2	5
MR (Calc. (nD-d/2))	39.14 1.04		5 2	25°C	1.0002	5	Exp	u. L.1.%/wt.		
Dielectric	+	1.97		5	30 mm BP	1.0000	5 4	Dis	persion	96.6	2
A 30 to		6, 87		2	t	0.9593 0.9496	5		sh Point C	4.	5
B [155 °C	_ 13			2	tc	0.264	2		e Point		
	-	214.86		2	ΔHc kcal/m ΔHf	1221.68 -60.23	2 2		Spec. ra V.		
A* 30 to B* 140 °C		1,31 244,36		5 5	ΔFf	2.03	2		Ray Dif. ared	Yes	2
K C = =	-				Viscosity					165	-
t _{1.} [-to	-				centistokes n °C				ubility in Tetone	oc	i
t _k to					7 °C				rbon tet.	00	ļ
A' 0 to		7, 27		5					her	&	
B' _ 30 °C		533. 24 233.		5	B _w to				Heptane	00	
A'* 10 to	_	1.68	408	5	A C	Ì			hanol ater	• • • • • • • • • • • • • • • • • • •	
B'* 30 °C		134.84	170	5	(B ^V) to	-		W	ater in		l
Acl 155 to		7.33	062	5	(A ^V) °C						
Bc tc °C		684.6 261.3		5 5	c liq. 300°K	0.39891	5			İ	
Cryos. A°	+			 	1 100	0.00227	5 2				
consts. B°					c _p vap.300°K 400	0.39703 0.50217	2				
t _e °C		30.42	4	5	c _w vap.						
$T_{\mathbf{R}} = 0.7$	5 T	:						+ g	rams/100 gra	ms solver	ıt
REFEREN	CES	: 1-D	ow	2-A	PI 3-Lit. 4	Calc. from de	t. da				
SOURCE:				AP	·I						
PURIFICA	TIO	N:		AP	ľ						
LITERATU	RE	REF	ERE	NCES	S:						

				_						No. 34	
NAME		3-Ethy	1-2-	meth	ylpentane			STR	UCTURAL	FORMULA	
									CH3C2H5		
	-	\top			T			(сн,сн сн		
Mole		Ref.	Мо	lecul		Molecular	.		•	- ,	
% Pur. 99	. 99!	5 2	Fo	rmul	-8-18	Weight 114.2	_				-
	_			Ref.			Ref		 		Ref.
F.P. *C F.P. 1009		114.96	<u> </u>	2	dt/dP *C/mm			f	300 to	0.0445	
B. P. *C	+			H	25°C	0.8023	4	g	600•K	0.0013	l
760 mm		115.65	0	2	BP	0.04748	2	h	 	-0.0637	5
100 30		55.71		2	t _e	0.0366	5	f' g'	to*K_		l
10	ı	29.40 9.5		4 2	30 mm	0.6577	4	h'	i	1	
1		-26.1		4	ΔHm cal/g	ļ	\vdash	<u></u>	1 300 to	0, 0265	4
Pressure	T				ΔHv cal/g 25°C	80,60	2	n n	_600 °K_	0.0014	
mm 25°C		23.92 063.7		4 5	30 mm	80.74	5	۰	i	-0.0 ₆ 46	4
Density	+	003.7		-	BP	69.70	2 5	m'	700 to	0.1079	4
g/ml 20°0	:	0.71	932	2	te te (d, e) AHv/T	68.11	5	n'	1 <u>1</u> 000 • K	0.0012	
dt 25 4 30		0.71		2	ΔHv/T	19.38	5	0'	İ	-0.0 ₆ 40	4
	+	0.71		-	d 25 to	 	4		ace tension		
a b	-	0.73 -0.0 ₃		4 4	<u>• 130 °C</u>	0.1280	4	dyne	s/cm, 20°C 30	21.52 20.58	2
Ref. Index					d' i to				40	19.65	2
n _D 20°0		1.40		2	 _	0,254	2	Para	chor [P]		
25 30		1.40		2 4	d g/ml v ml/g	3.940	2		20°C 30		
"C"	+	0,74		4	t _c ·C	295.	2		40		
MR (Obs.	+	38.83		-	P _c mm	20824.	2		Sugd.	346.2	5
MR (Calc.		39.14		2 5	PV/RT	1 0000		Exp.	L.1.%/wt.		
(nD-d/2)	\perp	1.04	435	2	25°C 30 mm	1.0000 1.0000	5 5	Dist	u. ersion	96.1	2
Dielectric		1.97	1	5	BP	0.9618	4		h Point °C	3.	5
A 25 t		6.86		2	t _e	0.9524 0.264	5 2		Point		
B 1_150 °		215.30		2 2	t _c	1222.11	2	M S	pec.		
A* 25 to	_	1, 29		5	ΔHf	-59.69	2	Ultr.	a V. ay Dif.		
B* 140°		231.73		5	ΔFf	3.03	2	Infra		Yes	2
K C					Viscosity centistokes			Solu	bility in +		
the Te					7 °C	ľ			etone	60	
t _x					•		1 1		rbon tet. nzene	80 80	
A' 0 to		8.55		4				Eth		60	[
C 30 -		278, 23 292, 31		4	B ^V to				Heptane anol	80 80]
A'* 10 to	_	2.88		4	A ^V I °C			Wa	ter		
B'* 30 °	C 2	132.76		4	(B ^V) to	1		Wa	ter in		-
Ac 150 to	2 .	7.40	735	5	(A ^V) °C	<u> </u>					
Bc Ltc_	- 1	749.0 271.2		5	c _p liq. 300°K	0.39891	5				l
Cryos, A'	•	0.05	44	2	c _p vap.300°K	0.50224					
t _e °C	+	128 25		5	c _v vap.	0.50217	2				
$T_R = 0.7$		128, 25		لــًـا	L	L	$\sqcup \sqcup$	L	40.00	<u> </u>	L
REFEREN				2 4 -	OT 2 124 4 4	2-1- 6			ms/100 gran		<u> </u>
SOURCE:	Ų EX	s. 1-D	∨₩	Z-AF		Calc. from det	. dat	a 5-	Calc. by for	mula	
		N.		AF							
PURIFICA			- n	AI							
21121411	, KL	KEFI		1CES	•						

TABLE C. ALKANES

No. 35

45

			-							No. 35	
NAME		3-Eth	yl-3-	met	hylpentane			ST	RUCTURAL	FORMUL	A
1 1									сн3		
		т							CII CII C	CII CII	
Mole		Ref.	Mole	ecul		Molecula <i>r</i>			CH ₃ CH ₂ C		
% Pur. 99	. 995	5 2	For	mula	8118	Weight 114.22	24		с ₂ н ₅		
				Ref.			Ref.				Ref.
F, P. ℃		90.87	0	2	dt/dP			f	300 to	0.0445	5
F.P. 1009	6				°C/mm			g	_600_°K		
B.P. ℃	-			- 1	25°C BP	0.8459 0.04844	2	h	l	-0.0637	5
760 mm	۱ ا	.18, 25 57, 10	9	2	te	0.0370	5	f'	to		
30		30.25	٦	4	30 mm	0.6713	4	g'	° <u>K</u>		
10		9.9		2	ΔHm cal/g			h'			
11	+	25.9		4	ΔHv cal/g		一	m	300 to	0.0265	4
Pressure mm 25°C	.	23.01	- 1	5	25°C	79.49	2	n	_6 <u>0</u> 0_•K	0.0014	
t _e		72.7	- 1	5	30 mm	79.55	4	٥	İ	-0.0 ₆ 46	4
Density	+-	-			BP	69.30 67.75	2 5	m¹	700 to	0.1079	4
g/m1 20°0	c	0.72		2	te te (d, e)	67.76	5	n' o'	[10 <u>00 •K</u>		4
dt 25		0.72		2	AHv/Te	19.12	5			-0.0 ₆ 40	-
	+-	0.71		4	d 30 to	83.08	5		face tension		
a b		0.74 -0.0 ₃		4	_e_ _130 °C	0.1165	5	dyn	es/cm. 20°C	21.99	2
Ref. Index		•3		\dashv	d' to				30 4 0	20.15	2
n _D 20°0		1.40	775	2			<u> </u>	Par	achor [P]		
_ 25	ı	1.40	549	2	d _c g/ml	0, 263 3, 808	2 2		20°C		
30	\bot	1.40		4	v _c ml/g t _c °C	305.	2		30		
"C"		0.74	66	4	P _c mm	21964.	2		40 Sugd.	346.2	5
MR (Obs.		38.71		2	PV/RT	 	 	Evr	L.1.%/wt.		<u> </u>
MR (Calc. (nD-d/2)	7	39.14 1.04		5 2	25°C	1.0000	5	2,	u.		
Dielectric	+-		\rightarrow	5	30 mm BP	1.0000	5 4	Dis	persion	95.8	2
A 30 to		1.98	-		t _e	0.9625 0.9528	5		sh Point C	6.	5
B 1160 °C		6, 86° 47, 20°		2 2	tc	0.265	2		e Point		
c '		19.68		2	ΔHc kcal/m	1221.20	2		Spec. ra V.		
A* 30 to	. [1.28	717	5	ΔHf ΔFf	-60.46 2.69	2 2		Ray Dif.		
B*[140 °C	12	57.01		5		2.69	-		ared	Yes	2
c c	ŀ		- 1		Viscosity centistokes		İ		ubility in +		
t _k T to			İ		η °c				etone rbon tet.	∞	
t _x °C					Ť				enzene	00 00	
A' 0 to		8. 17	016	5			1		her	oc	
B' 1_30 °C		77.18 80.09	۱ ۵	5	B ^V to				Heptane hanol	oc oc	
A'* 10 to		2.51	-	5	B ^V		ĺ		ater	"	
B'* 30 °C	- 1	42.04	-40	5	(B ^V) to	1		W.	ater in		
Acl 160 to		7.49	210	5	(A ^V) °C						
Bc tc °C	18	64.1		5	c liq. 300°K	0.39891	5				
Cc		86.7		5	1 400	0.50224	5				
Cryos, Accounts, B		0.03	92	2	c _p vap.300°K	0.39703	2	l			
		21 44			c, vap.	0.50217	2				
t _e °C		31.44		5	V	L		L		l	<u> </u>
$T_R = 0.7$									rams/100 gra		t
REFEREN	CES	1-D	ow .			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:					PI						
PURIFICA	TIOI	۷:		A	PI						
LITERATU	JRE	REFE	REN	CES	3:						
											

								No. 36	
NAME	2, 2, 3-Tri	meth	vlpentane			STRI	JCTURAL I		`
NAME -			, - p		\dashv	01111	сн ₃ сн ₃	. 01(1110 -1	•
							:н ₃ с - сн	сн.сн.	
Mole	Ref. Mo	lecul		Molecular		`	- •	20113	
% Pur. 99.		rmul		Weight 114.22	24		CH ₃		
		Ref.			Ref				Ref.
F.P. *C	-112,27	2	dt/dP			f	300to	0.0445	5
F.P. 100%	1-11-11-1	-	*C/mm	}]]	g ¦	600°K	0.0013	
B, P. °C	<u> </u>	\vdash	25°C	0.6199	4	h !		-0.0637	
760 mm	109.841	2	BP	0.04755 0.0372	5	<u>f'</u> +	to	-	
100 30	49.922 23.69	2	t _e 30 mm	0,6553	4	g'	•K]
10	3.9	2		0.0333	+	h'			
11	-28.7	4	ΔHm cal/g		\vdash	m l	300 to	0.0265	4
Pressure	22.00	١.,	ΔHv cal/g 25°C	77.24	2	n ;	_600°K_	0.0014	4
mm 25°C	32.06 1043.0	5	30 mm	78.00	4	°i		-0.0646	4
Density		Ť	BP	67.30 65.84	2 5	m'	700 to	0.1079	4
g/ml 20°C	0.71602	2	te te (d, e)	65.83	5	n'	1 <u>1</u> 0 <u>0</u> 0°K	0.0012	4
dt 25 4 30	0.71207	2	ΔHv/T	19.05	5	0'		-0.0 ₆ 40	4
	0.70811	4	d 25 to	80.94	4		ce tension		
a b	0.73180 -0.0 ₃ 781	4	e 120 °C		4	dyne	30°C 30°C	20.67 19.77	5
Ref. Index	0.03.02	H	G' 6	ĺ		1	40	18.87	5
n _D 20°C	1.40295	2		0.2(1		Para	chor [P]		
- 25	1.40066	2	d g/ml v ml/g	0.261 3.826	2 2		20°C		
30	1.39811	4	tc °C	294.	Ž		30 40		İ
"C"	0.7500	4	P _c mm	21432.	2		Sugd.	346.2	5
MR (Obs.) MR (Calc.)	38.925	5	PV/RT		Н	Exp.	L.1.%/wt.		
(nD-d/2)	39.144 1.04494	2	25°C	0.9925	5	_	u.		
Dielectric	1.968	5	30 mm BP	1.0000 0.9585	5 4		rsion	97.2	2
A 20 to		2	te	0.9493	5		Point °C		
B 1150 ℃	1294.875	2	t _c	0.265	2	M Sp			├
С	218.420	2	ΔHc kcal/m ΔHf	1219.98 -61.44	2 2	Ultra			ı
A* 20 to	1.26422	5	ΔFf	2, 22	2		y Dif.		
B* ∟130 °C	1209. 78	5	Viscosity		\vdash	Infra		Yes	2
¢	_		centistokes	ļ		Solub Ace	ility in +		
tk 10°C	i i		η ∘c				on tet.	&	ĺ
A' to	 	-		İ		Ben		œ	
B'						Ethe n-He	er eptane	80	
C'			B ^V to	ļ		Etha	nol	∞	l
A'* to	1		AV C	1		Wate	er in		
B'* °C	 		(B ^V) to	1					
Ac 150 to Bc t _c °C	7.29189	5	(A ^V) °C						
Cc Cc	267.9	5	c _p liq. 300°K 400	0.39891 0.50224	5 5				
Cryos. A°	0.0401	2	c _p vap.300°K	0.39703	2				
consts. B°			400	0.50217	2				
t _e °C	121.69	5	c _v vap.						
$T_{R} = 0.75$	Tc					† gran	ms/100 gran	ns solven	t
	ES: 1-Dow	2-AF		alc, from det	dat	ta 5-0	Calc. by form	nula	
SOURCE:		AF	PI						
PURIFICAT		AI							
LITERATU	RE REFERE	NCES	3:						

TABLE C. ALKANES

No. 37

47

										No. 37	
NAME		2,2,4	-Tri	methy	ylpentane			ST	RUCTURAL	FORMUL	A
		Isooct	ane							CH ₃	
							\neg		CH ₃ C CH ₂ C	сн сн3	
Mole		Ref.	Mo	lecula		Molecular	.		Ċн ₃	-	
% Pur.	_		FOI			Veight 114.22	+				D - C
	1	107.00		Ref.			Ref.		1		Ref.
F.P. °C F.P. 100°		-107.38	50	2	dt/dP °C/mm		1	f	to K		
B. P. °C	+			\vdash	25°C	0.4226	4	g h			
760 mm		99.23	8	2	BP	0.04651 0.03732	2 5	f'			
100 30		40.66 15.05		2 4	t _e 30 mm	0.6398	4	g	to °K		
10		-4.3	,	2		0.0378	┝┸	h'			
1	_	-36.1		4	ΔHm cal/g		-	m	300 to	0.0265	4
Pressure mm 25°C	.	40.04		١. ١	ΔHv cal/g 25°C	73.50	2	n	_6 <u>0</u> 0_•K	0.0014	
t _e		49.34 1009.4	ŀ	4 5	30 mm	75.31	4	0		-0.0 ₆ 46	4
Density	\dashv			H	BP t _e	64.87 63.60	2 5	m'	700 to	0.1079	4
g/ml 20°	С	0.69		2	te (d, e)	63.59	5	n' o'	1000 •K	0.0012 -0.0 ₆ 40	4
dt 25	1	0.68 0.68		2 4	ΔHv/T _e	18.98	5			-0.0610	L.
a	+	0.70		4	d 15 to	77.17	5		face tension	18.77	2
ь		-0.03		4		0.1240	5	gyn	es/cm. 20°C 30	17.88	2
Ref. Inde:					e' c				40	16.99	2
ⁿ D ^{20°}	С	1.39		2 2	d _c g/ml	0.237	2	Par	achor [P]		
30		1.38		5	V _C m1/g	4.220	2 2		20°C 30		
"C"	7	0.75	552	4	t° ℃	271.15	I i		40		_
MR (Obs.)	39, 26	2	2	P _c mm	19380.	2			346. 2	5
MR (Calc	.)	39.14		5	PV/RT 25°C	0.9910	4	Ехр	u. L.1.%/wt.		
(nD-d/2)	+	1.04		2	30 mm	1.0000	5	Dis	persion	100.5	2
Dielectric		1.93		5	BP t _e	0.9562 0.9478	4 5		sh Point C	-12.	3
B (135 °C		6.81 1257.84		2 2	t _c e	0, 275	2		e Point		ļ
с		220,73		2	∆Hc kcal/m	1219.01	2		Spec. raV.	Yes	1
A* 15 to		1.26		5	ΔHf ΔFf	-61.97 1.65	2 2	X-F	Ray Dif.		
B* 120 °C	-	1175.54	ŀ	5	Viscosity		╽▔	 	ared	1772.	1
·	_				centistokes				ubility in ^T	∞	
t _k to					η 20 °C	0.7259 0.5958	1 1		rbon tet.	×	
t _x °C	1			Н	60	0.4999	1		nzene her	- so	l
B' °					80	0.4270	1		ner Heptane	80	
C'	4				B ^V 30 to A ^V 90 °C	400.09 2.49766	4		hanol	∞	
A'* to B'* °C						2.49700	*		ater ater in		
Acl 135 to	-	7.27	7905	5		1					
Bc tc °		1612.2	, 55	5		†	\vdash	1			
Cc —	\rightrightarrows	267.7		5	P	ļ					
Cryos, A consts, B		0.04		2	c _p vap.300°K	0.39703	2				
t _e °C	\dashv	0, 00		5	c _v vap.	0.50217	2				
$T_R = 0$.	75			12	L	L		+ -		L	<u></u>
REFEREN)ov.	2. 4	PI 3-Lit. 4-0	Calc from 3			ams/100 gra		ıt
SOURCE:	U.E	.∪. 1-L	, U W		PI 3-Lit. 4-0	Calc. Irom de	aa	. ua. 5	-Caic. by for	u.a	
PURIFICA	TI	ON:			PI						
			EDE			-					
LILERATI	UK.	E REF	eke.	NUES	5: 3 NFPA 325	•					
1											
L											

										No. 38	
NAME	2	., 3, 3-	Tri	methy	lpentane			STR	UCTURAL	FORMULA	1
									СН ₃ СН ₃		
l									сн _з сн с	H_CH_	
Mole		Ref.	Мо	lecul	ar c	Molecular			CH ₃	23	
% Pur. 99.	87	2		rmul		Weight 114.2	24		CH ₃		
				Ref.			Ref				Ref.
F. P. *C	- 1	00.70		2	dt/dP			f	300 to	0.0445	5
F.P. 100%					°C/mm	1	1.1	g	_6 <u>0</u> 0 <u>•</u> K	0.0013	5
B. P. °C	Τ.		_		25°C BP	0.7357 0.04833	2	h	,	-0.0637	5
760 mm 100		14.76 53.81		2 2	t	0.0373	5	f'	to		
30		27.10		4	30 mm	0.6676	4	g'			ł
10 1		6.9 33.15		2 4	ΔHm cal/g			h'	1		
	╀	33.15		+	ΔHv cal/g		1	m	300 to	0.0265	4
Pressure mm 25°C	1	27.00		4	25°C	77.87	2	n	_600 K	0.0014	
t _e		60.3		5	30 mm BP	78.34 68.10	2		<u></u>	-0.0 ₆ 46	-
Density	\top				t.	66.61	5	m'	700 to	0.1079	4
g/ml 20°C		0.72		2	e (4, 5)	66.62	5	n' o'	1000 °K	0.0012	4
dt 25 4 30		0.72 0.71		2 4	AHv/Te	18.99	5	<u> </u>	1	-0.0640	1
<u>a</u>	+-	0.74		14	d 25 to	81.50	5		ace tension	31.57	١.
ъ	1	-0.03		4	-a,- -125 °C		5	gyne	ss/cm, 20°C 30	21.56 20.65	2 2
Ref. Index					d' to				40	19.75	2
n _D 20°C		1.40		2	d g/ml	0, 264	2	Par	achor [P]		
25 30		1.40 1.40		2 4	v ml/g	3.791	2		20°C		
"C"	+	0.74		4	1 °C	303.	2	İ	40		
MR (Obs.)	+-	38.76		2	P _c mm	22040.	2		Sugd.	346. 2	5
MR (Calc.		39.14		5	PV/RT	1 0003	,	Exp	. L.1.%/wt.		
(nD-d/2)		1.04	440	2	25°C 30 mm	1.0002 1.0000	5	Dier	u. ersion	96.1	2
Dielectric		1.98	1	5	BP	0.9607	4		h Point °C	70.1	-
A 25 to		6.84		2	t _e	0.9512 0.266	5 2		Point		
B [_160 •C		28.04 20.37		2 2	t _c ΔHc kcal/m	1220.86	2	M S	pec.		
A* 25 to	+	1.27		5	ΔHf	-60.63	2	Ultr			1
B* 140 °C		39.27		5	ΔFf	2,54	2		ay Dif. ared	Yes	2
к — — —	1			1 1	Viscosity	j			bility in +		
¢	-				centistokes 7°C			Ace	etone	60	
t _x + °C					'	1			rbon tet. nzene	60	
A' to						į		Eth		e0 e0	
B' •	-			1	B ^V to	 	-		leptane	•	1
A'* to	+			+	B to	ĺ		Wa	anol ter	∞	
B'* °C					(B ^V) to	i		Wa	ter in		
Ac 160 to	+	7.38	917	5	(A ^V) °C						
Bc t °C	17	74.2		5		0, 39891	-				
Cc — —	- 2	79.5		5	c _p liq. 300°K 400	0.39891	-5 5				1
Cryos, A° consts, B°	-	0.00	62	2	c _p vap.300°K	0.3970 3 0.50217	2 2				
t _e °C	Щ.	27.46		5	c _v vap.						
$T_{R} = 0.7$								*+ gra	ams/100 gran	ns solven	t
REFERENC	ES:	1-D	ow.			Calc. from det	t. da	ta 5-	Calc. by for	mula	
SOURCE:				AF	PI						
PURIFICAT	ION	l:		AF	PI						
LITERATU	RE	REFI	ERE	NCES	:						

NAME	2, 3, 4-Tr	meth	nylpentane			STRUCTURAL	FORMUL	

						CH3CH CH		
Mole % Pur. 99.	Ref. Mo	lecul mul:		Molecular Veight 114.2	24	3000	3	
		Ref.	w	T	Ref.	I		Ref.
F. P. °C			dt/dP			f to		
F.P. 100%	-109.210	-2	°C/mm			g °K		
B. P. ℃			25°C BP	0.7192 0.04761	2	h		
760 mm 100	113.467 53.404	2 2	t	0.0370	5	f' to		
30	27.06	4	30 mm	0.6585	4	g' '° <u>K</u>		
10 1	7.1 -25,65	2 4	ΔHm cal/g			h'		
Pressure	-23.03	Ė	ΔHv cal/g			m 300 to	0.0265 0.0014	4
mm 25°C	27.01	4	25°C 30 mm	79.52 79.39	5		-0-0646	4
t _e	1051.	5	BP	68.37	2	m' 700 to	0, 1079	4
Density g/ml 20°C	0.71906	2	t _e (d, e)	66.45	5	n' 1000 °K	0.0012	4
t 25	0.71503	2	ΔHv/T _e	66, 82	5	o'	-0.0 ₆ 40	4
⁴ 30	0.71099	4	d 20 to	19.04	5	Surface tension		
a b	0.73517	4	e 120 °C	82.84 0.1276	5	dynes/cm. 20°C	21.14	2
Ref. Index	-0.0380	1	d' to			30 40	20.22 19.31	2 2
n _D 20°C		2		0.35/	-	Parachor [P]		
25 30	1.40198 1.39931	2 4	d _c g/ml v _c ml/g	0.256 3.913	2 2	20°C		
"C"	0,7491	4	tc°C	295.	2	30 40		
MR (Obs.)		2	P _c mm	20976.	2	Sugd.	346.2	5
MR (Calc.		5	PV/RT 25°C	1 0000	ا ۔	Exp. L.1.%/wt.		
(nD-d/2)	1.04469	2	30 mm	1.0000 1.0000	5	u. Dispersion	96.9	2
Dielectric		5	BP	0.9566	4 5	Flash Point C	2.	5
A 25 to B 150 °C		2 2	te t _C	0.9470 0.265	2	Fire Point		
c '-32 -	217.526	2	∆Hc kcal/m	1220.61	2	M. Spec.		
A* 25 to		5	ΔHf ΔFf	-60.98	2 2	Ultra V. X-Ray Dif.		
B* 135 °C	1230.23	5	Viscosity	2,54		Infrared	Yes	2
·	_		centistokes			Solubility in *Acetone		
t _k to			η °C			Carbon tet.	&	
t _x °C	+					Benzene Ether	∞ ∞	
B'°	<u>-</u>					n-Heptane	&	
C' to	+		B ^V to C			Ethanol Water	oc	ł
A'* to B'* °C			(B ^V) - to			Water in		
Acl 150 to		5	(A ^V) °C					
Bc tc °C	1695.6	5 5	c _p liq. °K					
Cryos. A°			1 -	0 2070-				
consts. B°		2 2	c _p vap.300°K 400	0.39703 0.50217	2			
t _e °C	125.62	5	c _v vap.					
$T_R = 0.7$				·		grams/100 gra	ms solven	t
	CES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:		A	PI					
PURIFICA'	TION:	A	PI					
LITERATU	RE REFERE	NCE	5:					

								N o. 40	
NAME	2, 2, 3	, 3-Tetra	methylbutane			STF	UCTURAL	FORMULA	<u>. </u>
		-			\neg		СН ₃ СН	[2	
					\dashv		сн ₃ с - с		
Mole	Ref.	Molecul	lar c u	Molecular			³ċн ₃ ċн		
% Pur.		Formul		Weight 114.2	24			3	
		Ref.			Ref.				Ref.
F.P. *C	+100.69	2	dt/dP			f	to		1
F.P. 100%			*C/mm	0.0140		g	<u>*</u> K_		1
B. P. °C	1,0,42	١,	25°C BP	0.8149 0.0476	3	h	· 1		
760 mm 100	106.47 54.70	# 2 3	te	0.0373	5	f'	to		
30	31.33	7, 2	30 mm	0.6604	4	g'			
10 1	13.09		ΔHm cal/g			h'	1		
	-10.03		ΔHv cal/g			m	300 to	-0.0414	
Pressure mm 25°C	20, 85	4 3	25°C	89.36#	2	n o	_600 °K	0.0017 -0.0 ₆ 75	4
t _e	1035.	5	30 mm BP	75.46 66.2	2	<u> </u>	1	-0.0613	Ľ
Density		7	t	65.02	5	m'	700 to	0.1205	
g/ml 23°C			t (d, e)	64.92	5	n' o'	1 1000 K	0.0013 -0.0 ₆ 46	4
d ₄ 25	0.82		AHv/Te	19.80	5	L.	<u> </u>	-0.0640	1
	0.83		d 20 to	78.03	5		ace tension	21 14	١,
ь	-0.03		120 °C		5	gyne	ss/cm, 20°C 30	21.14	2
Ref. Index	<u> </u>		a' to				40	19.31	2
n _D 20°C	1.46	95 31		0,239	2	Par	chor [P]		
25 30			d _c g/ml v _c ml/g	4.189	2	i	20°C 30		
"C"	+		tc °C	270.8	2		40		Ì
	 		P _c mm	18620.	3	1	Sugd.	346.2	5
MR (Obs.) MR (Calc.	39.14	4 5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	/	.	25°C	1.0000	5	D.	u.		l
Dielectric			30 mm BP	0.9599	4		ersion		<u> </u>
A 101 to	6, 87	665 2	t _e	0.9515	5		sh Point °C Point		ļ
B [160 °C	1327.8	2	tc	0.262	4	M S		ļ	
<u> </u>	226.0	2	ΔHc kcal/m	1218.59 -64.23	2 2		a V.		ł
A* 101 to	1.31	013 5	ΔFf	3, 13	2		ay Dif.	i	ļ
B* ∟135 °C	- 1230.0	"	Viscosity	 		 	ared	ļ	
°	_	Ì	centistokes				bility in + etone	į	1
tk to			∦ η °⊂				rbon tet.		
A' -20 to		864≠ 3	1				nzene		ļ
B' 101 °C						Eth n-F	er Ieptane	ĺ	
C'	233.63		B _v to			Eth	anol		l
A'* -20 to			A ^V O				ter ter in		ł
B'* 101 °C		5	(B ^V) to			_ <u>""</u>	ter in	ļ	├
Ac 160 to	8. 798 3213. 7	823 5	(A ^V) •C						1
Bc tc_°C	438.9	5	cp liq. 20°C			l			ŀ
Cryos. A°	0.00		- 1	0.497	3 2				İ
consts. B°	"	-	c _p vap.300°K 400	0.50217	2				l
te °C	117.83	5	c _w vap.	1					
# solid state	#104.9	°C #h	eat of sublimati	on $T_{D} = 0.7$	5 T.	+ ~ ~	me/100 area	ne eolwani	
REFERENC	ES: 1-De	ow 2-A1	PI 3-Lit. 4-	Calc from de	t da	ta 5.	Calc by for	mula	
SOURCE:	-		PI, Lit.				by 1011		
PURIFICAT	ION:		PI, Lit.						
			5: 3 JACS 74,	883 /1052\ D	w	Scott	at al. 21 TAC	S 71	
				003 (1752) D.	₩.	SCOLL (cai, J. JAC	, <u>, , , , , , , , , , , , , , , , , , </u>	
3447 (1949)	win. P.	seyer et	aı.						

								No. 41	
NAME _	n-Nonane					ST	RUCTURAL	FORMUL.	A
							CH ₃ (CH ₂).	CH.	
Mole % Pur. 99. 9	97 Ref. Mo. 1	ecul mula		Molecular Veight 128.25	50			, 5113	
	1	Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-53.519	2	dt/dP °C/mm 25°C	2 ((0	۰	f g	300 to _600_°K	0.0496 0.0013	5 5
B.P. °C 760 mm 100 30	150.798 87.899 60.16	2 2 4	BP t _e 30 mm	3.660 0.04967 0.0365 0.6943	5 2 5 4	f' g'	to *K	-0.0 ₆ 35	5
10 1	39.0 3.6	2	ΔHm cal/g	28.83	3 2	h'			
Pressure mm 25°C t _e	4.351 1127.4	5	AHv cal/g 25°C 30 mm BP	86.54 82.67 68.44	2 4 4	m n o	600 •K	0.0294 0.0015 -0.0 ₆ 52	4 4
Density g/ml 20°C dt 25 4 30	0.71763 0.71381 0.70999	2 2 4	te te (d, e) AHv/Te	66. 42 66. 00 19. 38	5 5	m' n' o'	700 to	0.1222 0.0013 -0.0 ₆ 45	4 4
a b Ref. Index	0. 73290 -0. 0 ₃ 762	4 4	d 60 to e 165 °C d' 10 to e' 60 °C	92.11 0.1570 89.29 0.1102	4 4 4		face tension es/cm, 20°C 30 40	22.92 21.95 21.00	2 2 2
ⁿ D 20°C 25 30	1.40542 1.40311 1.39984	2 2 4	d _c g/ml v _c ml/g t _c °C	0.244 4.094 322.	2 2 2	Par	20°C		
"C"	0.7509	4	P _c mm	17279.	2	ľ	40 Sugd.	385.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	43.842 43.762 1.04660	2 5 2	PV/RT 25°C 30 mm	1.0000 1.0000	5	1	L.1.%/wt. u. persion	3.2 11.8 97.2	3' 3' 2
A 60 to	6.93513	2	BP t e t c	0.9380 0.9482 0.250	5 2		sh Point C e Point	31.0	31
B 185 °C C	1428.811 201.619 1.43093	2 2	ΔHc kcal/m ΔHf	1369.70	2	Ult	Spec.	Yes	1
B* 180 °C K	1351.7 24.	4 5	ΔFf Viscosity			Infr	Ray Dif. ared ubility in	Yes	2
t _k 180 to t _x 270 °C	-0.15644 227. 381.	5 5 5	7 80 °C 100 120	0.5502 0.4743 0.4151	2 2 2	A c	cetone rbon tet.	80 80	1 1 1
A' 20 to B' 60 °C	7.30801 1630.71 219.5	5 5 5	140 BV 30 to	0, 3675 437, 7	2	n-	her Heptane hanol	8 8 8	1 1 1
A'* 20 to B'* 60 °C	1.76177 1536.61	5 5	A ^V 90 °C (B ^V) 90 to	2.50135 427.1	4		ater ater in		
Acl 185 to Bc t _c °C Cc —	7,43583 1842,3 254.0	5 5 5	(A ^V) 150 °C c _p liq. 300°K 400	2.53159 0.39979 0.50256	4 5 5		cosity tistokes 40°C	0.7921 0.6515	2 2
Cryos. A° consts. B°	0.03856	32	c vap.300K 400	0.39649 0.50136	2		60 110 150	0.4430 0.3473	2 2
t _e °C T _R = 0.77	166.351 T-	5	c _w vap.	L	لــــا	L	(100		<u> </u>
	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da		rams/100 gra -Calc. by for		<u> </u>
SOURCE:		AI		uc			by 101		
PURIFICAT	ION:		PI						
LITERATUE	RE REFEREI	CES	S: 3 NBS Circ.	514; 3' NFF	A C	irc.;	3 ² JACS <u>76</u> ,	333 (1954)	,
Finke et al.									

2-Met Ref. 143. 26 81. 0 53. 15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 1. 403 1. 400 1. 398 43. 762 1. 046 1. 969	5 34 2 255 2 66 4 90 4 4778 4 31 2 38 2 33 4 40 4 2 5 5 4 2.	ar C ₉ H ₂₀	2.527 0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	250 Ref. 52554 22425555444445555555	f g h f' g' h h' m n o o o Surfacdynes	CTURAL CH3 CH3CH (CH 300 to 600 °K to 600 °K 700 to 1000 °K 700 to 1000 °K ce tension /cm. 20°C 30 40 thor [P] 20°C 30 40	2) ₅ CH ₃	Ref 5 5 5 5 4 4 4 4 4 4 4
-80. 4 143. 26 81. 0 53. 15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 43. 762 1. 046	Ref. 2 2 2 4 2 5 5 5 6 4 4 6 6 4 6 6 6 6	a C9H20 dt/dP	2.527 0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 4 2 5 5 4 4 4 4 4 5 5 5 5	f g h f' g' h n n n n n n n n n	300 to 600 °K to 600 °K 700 to 1000 °K 700 to 1000 °K 20°C 30°C 30°C 30°C 30°C 30°C 30°C 30°C 3	0.0496 0.0013 -0.0 ₆ 35 0.0294 0.0015 -0.0 ₆ 52 0.1222 0.0013 -0.0 ₆ 45	5 5 5 4 4 4 4 4 4 2 2
-80. 4 143. 26 81. 0 53. 15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 43. 762 1. 046	Ref. 2 2 2 4 2 5 5 5 6 4 4 6 6 4 6 6 6 6	a C9H20 dt/dP	2.527 0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 4 2 5 5 4 4 4 4 4 5 5 5 5	f g h f' g' h n n n n n n n n n	300 to 600 °K to 600 °K 700 to 1000 °K 700 to 1000 °K 20°C 30°C 30°C 30°C 30°C 30°C 30°C 30°C 3	0.0496 0.0013 -0.0 ₆ 35 0.0294 0.0015 -0.0 ₆ 52 0.1222 0.0013 -0.0 ₆ 45	5 5 5 4 4 4 4 4 4 2 2
-80. 4 143. 26 81. 0 53. 15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 43. 762 1. 046	Ref. 2 2 2 4 2 5 5 5 6 4 4 6 6 4 6 6 6 6	a C9H20 dt/dP	2.527 0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 4 2 5 5 4 4 4 4 4 5 5 5 5	g h f' g' h' m n o o m' n' o' Surfacdynes	600 °K to 600 °K 700 to 1000 °K 700 to 1000 °K 700 to 1000 °K 700 to 700 t	0.0496 0.0013 -0.0635 0.0294 0.0015 -0.0652 0.1222 0.0013 -0.0645	5 5 5 4 4 4 4 4 4 2 2
143, 26 81, 0 53, 15 32, 2 -2, 5 6, 554 1142. 0, 713 0, 709 0, 705 -0, 0 ₃ 7 1, 400 1, 398 43, 762 1, 046	2 2 2 4 2 5 5 5 5 5 5 5 5 5 6 4 4 9 9 9 4 4 8 1 2 2 8 8 2 8 3 3 4 4 8 3 3 4 4 8 3 3 4 4 2 2 5 5 4 4 2 2 .	dt/dP	0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	5 2 5 4 2 4 2 5 5 5 4 4 4 4 4 4 4	g h f' g' h' m n o o m' n' o' Surfacdynes	600 °K to 600 °K 700 to 1000 °K 700 to 1000 °K 700 to 1000 °K 700 to 700 t	0.0496 0.0013 -0.0635 0.0294 0.0015 -0.0652 0.1222 0.0013 -0.0645	5 5 5 4 4 4 4 4 4 2 2
143, 26 81, 0 53, 15 32, 2 -2, 5 6, 554 1142. 0, 713 0, 709 0, 705 -0, 0 ₃ 7 1, 400 1, 398 43, 762 1, 046	2 2 2 4 2 5 5 5 5 5 5 5 5 5 5 5 5 5 5 6 4 4 6 6 6 6	*C/mm 25*C BP te 30 mm AHm cal/g AHv cal/g 25*C 30 mm BP te (d, e) AHv/Te d 55 to e 160 *C d' 10 to e' 55 *C d g/ml v ml/g t_c *G P_c mm PV/RT 25*C	0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 5 4 2 4 2 5 5 5 5 4 4 4 4 4 5 5	g h f' g' h' m n o o m' n' o' Surfacdynes	600 °K to 600 °K 700 to 1000 °K 700 to 1000 °K 700 to 1000 °K 700 to 1000 °K 700 to 1000 °K 1000 to	0.0013 -0.0 ₆ 35 0.0294 0.0015 -0.0 ₆ 52 0.1222 0.0013 -0.0 ₆ 45	5 5 4 4 4 4 4 4 2 2
81. 0 53.15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 0. 753 43. 88 43. 762 1. 046	2 4 4 2 5 5 5 5 5 5 6 4 2 2 6 6 4 4 4 6 6 6 4 4 6 6 6 4 4 6 6 6 6	25°C BP t t 30 mm AHm cal/g AHv cal/g 25°C 30 mm BP te(d,e) AHv/Te d 55 to e 160 °C d' 10 to e' 55 °C d_c g/ml v_c ml/g t_c °C P_c mm PV/RT 25°C	0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 5 4 2 4 2 5 5 5 5 4 4 4 4 4 5 5	h f' g' h' m n n n' n' o' o' Surfactynes	700 to 1000 °K 700 to 1000 °K ce tension /cm. 20°C 30 40 thor [P] 20°C	0.0294 0.0015 -0.0652 0.1222 0.0013 -0.0645	4 4 4 4 4 2 2
81. 0 53.15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 0. 753 43. 88 43. 762 1. 046	2 4 4 2 5 5 5 5 5 5 6 4 2 2 6 6 4 4 4 6 6 6 4 4 6 6 6 4 4 6 6 6 6	BP te 30 mm AHm cal/g AHv cal/g 25°C AHv cal/g 25°C AHv/Te d 55 to e 160 °C d' 10 to e' 55 °C	0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 5 4 2 4 2 5 5 5 5 4 4 4 4 4 5 5	f' g' g' h' m o o m' n' o' Surfact dynes	300 to 600 °K 700 to 1000 °K ce tension /cm. 20°C 30 40 thor [P] 20°C 30	0.0294 0.0015 -0.0652 0.1222 0.0013 -0.0645	4 4 4 4 4 2 2 2
81. 0 53.15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 0. 753 43. 88 43. 762 1. 046	2 4 4 2 5 5 5 5 5 5 6 4 2 2 6 6 4 4 4 6 6 6 4 4 6 6 6 4 4 6 6 6 6	30 mm AHm cal/g AHv cal/g 25°C 30 mm BP te (d,e) AHv/Te d 55 to e 160 °C d' 10 to e' 55 °C d g/ml v ml/g t_c °C P_c mm PV/RT 25°C	83. 2 79. 78 68. 3 66. 31 66. 26 19. 67 86. 55 0. 1274 86. 24 0. 1216	2 4 2 5 5 5 4 4 4 4 4 5 5	g' h' m n o m' n' o' Surface dynes	300 to 600 °K 700 to 1000 °K ce tension /cm. 20°C 30 40 thor [P] 20°C 30	0.0015 -0.0652 0.1222 0.0013 -0.0645 21.88 20.94	4 4 4 4 2 2
32.2 -2.5 6.554 1142. 0.713 0.705 0.705 0.725 -0.037 1.400 1.398 43.762 1.046	2 5 5 5 5 5 5 5 5 5 6 6 4 6 6 6 6 6 6 6 6	ΔHm cal/g ΔHv cal/g 25°C 30 mm BP te te (d,e) ΔHv/Te d 55 to e 160 °C d' 10 to e' 55 °C d_c g/ml v_c ml/g t_c °C P_mm PV/RT 25°C	83. 2 79.78 68. 3 66. 31 66. 26 19. 67 86. 55 0. 1274 86. 24 0. 1216 311. 0 16595.	2 4 2 5 5 5 4 4 4 4 5 5	h' m n o m' n' o' Surface dynes	300 to 600 °K 700 to 1000 °K ce tension /cm. 20°C 30 40 chor [P] 20°C 30°	0.0015 -0.0652 0.1222 0.0013 -0.0645 21.88 20.94	4 4 4 4 2 2
-2.5 6.554 1142. 0.713 0.705 0.705 0.725 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	5 5 5 5 5 5 6 4 2 6 6 4 4 778 4 4 8 1 1 2 2 3 3 4 4 2 3 3 4 4 4 4 4 4 4 4 4 4 4	ΔHv cal/g 25°C 30 mm BP te (d,e) ΔHv/Te d 55 to e 160 °C d' 10 to e' 55 °C d g/ml v ml/g t c °C P mm PV/RT 25°C	79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216 311.0 16595.	4 2 5 5 5 4 4 4 4 5 5	m n o o m' n' o' o' Surfactives	700 to 1000 °K 700 to 1000 °K 700 to 20°C 30	0.0015 -0.0652 0.1222 0.0013 -0.0645 21.88 20.94	4 4 4 4 2 2
0.713 0.705 0.705 0.705 -0.037 1.400 1.398 0.753 43.88 43.762	5 34 2 255 2 66 4 90 4 4778 4 31 2 38 2 33 4 40 4 2 5 5 4 2.	25°C 30 mm BP te te (d,e) ΔHv/Te d 55 to e 160 °C d' 10 to e' 55 °C d g/ml v ml/g tc °C P mm PV/RT 25°C	79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216 311.0 16595.	4 2 5 5 5 4 4 4 4 5 5	n o m' n' o' Surfac dynes	700 to 1000 °K 700 to 1000 °K 700 to 20°C 30	0.0015 -0.0652 0.1222 0.0013 -0.0645 21.88 20.94	4 4 4 4 2 2
0.713 0.705 0.705 0.705 -0.037 1.400 1.398 0.753 43.88 43.762	5 34 2 255 2 66 4 90 4 4778 4 31 2 38 2 33 4 40 4 2 5 5 4 2.	30 mm BP t t e (d, e) ΔHv/T d 55 to e 160 °C d' 10 to e' 55 °C d g/ml v _c ml/g t _c °G P _c mm	79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216 311.0 16595.	4 2 5 5 5 4 4 4 4 5 5	o m' n' o' Surfac dynes	700 to 1000 °K 1000 °K 20°C 30 40 20°C 3	0.1222 0.0013 -0.0 ₆ 45 21.88 20.94	4 4 4 2 2
0.713 0.705 0.705 0.725 -0.037 1.403 1.400 1.398 0.753 43.88 43.762	34 2 255 4 90 4 4778 4 31 2 28 2 33 4 30 4 2 5 54 2.	BP te (d, e) ΔHv/Te d 55 to e 160 °C d' 10 to e' 55 °C d g/ml vc ml/g tc °G Pc mm PV/RT 25°C	68. 3 66. 31 66. 26 19. 67 86. 55 0. 1274 86. 24 0. 1216 311. 0 16595.	2 5 5 5 4 4 4 4 5 5	Surface dynes	1000 °K ce tension /cm. 20°C 30 40 chor [P] 20°C 30	0.1222 0.0013 -0.0 ₆ 45 21.88 20.94	4 4 2 2
0.709 0.705 0.729 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	25	AHv/T _e d 55 to e 160 °C d' 10 to e' 55 °C d g/ml v _c ml/g t _c °C P _c mm PV/RT 25°C	66. 26 19. 67 86. 55 0. 1274 86. 24 0. 1216 311. 0 16595.	5 5 4 4 4 4 5 5	Surface dynes	1000 °K ce tension /cm. 20°C 30 40 chor [P] 20°C 30	0.0013 -0.0 ₆ 45 21.88 20.94	4 4 2 2
0.709 0.705 0.729 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	25	AHv/T _e d 55 to e 160 °C d' 10 to e' 55 °C d g/ml v _c ml/g t _c °C P _c mm PV/RT 25°C	19.67 86.55 0.1274 86.24 0.1216 311.0 16595.	5 4 4 4 4 5 5	Surfac dynes y	/cm. 20°C 30 40 thor [P] 20°C 30	21.88 20.94	2 2
0.729 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	90 4 1778 4 31 2 108 2 133 4 30 4 2 5 54 2.	d 55 to e 160 °C d' 10 to e' 55 °C d _c g/ml v _c ml/g t _c °G P _c mm PV/RT 25°C	86.55 0.1274 86.24 0.1216 311.0 16595.	4 4 4 4 5 5	dynes	/cm. 20°C 30 40 thor [P] 20°C 30	20.94	2
-0.0 ₃ 7 1.403 1.400 1.398 0.753 43.88 43.762 1.046	778 4 31 2 98 2 33 4 30 4 2 5 64 2.	e 160 °C d' 10 to e' 55 °C d g/ml v ml/g t e G P mm PV/RT 25°C	0.1274 86,24 0.1216 311.0 16595.	4 4 4 5 5	dynes	/cm. 20°C 30 40 thor [P] 20°C 30	20.94	2
1.403 1.400 1.398 0.753 43.88 43.762 1.046	31 2 08 2 33 4 30 4 2 5 54 2.	d' 10 to e' 55 °C d g/ml vc ml/g tc °G P mm PV/RT 25°C	86. 24 0. 1216 311. 0 16595.	4 4 5 5		40 thor [P] 20°C 30		
1.400 1.398 0.753 43.88 43.762 1.046	08 2 33 4 30 4 2 2 54 2.	d g/ml vc ml/g tc *G Pc mm PV/RT 25*C	311. 0 16595.	5 5		hor [P] 20°C 30	20.00	۴
1.400 1.398 0.753 43.88 43.762 1.046	08 2 33 4 30 4 2 2 54 2.	t _c *G P _c mm PV/RT 25°C	16595. 1.00 0 0	5		20°C		1
0.753 43.88 43.762 1.046	30 4 2 2 5 4 2.	P _C mm PV/RT 25°C	16595. 1.00 0 0	5			i i	i
43.88 43.762 1.046	2 5 5 4 2.	P _c mm PV/RT 25°C	16595. 1.00 0 0	-		40		ĺ
43.762 1.046	2 5 54 2.	PV/RT 25°C	1.0000	-	<u> </u>	Sugd	385.2	5
1.046	4 2.	25°C		=	Evn	L. 1. %/wt.	0,74	3
 		30			· ·	u.	2.9	3
1.70.) [5]	BP	1.0000 0.9610	5 4	Dispe		98.4	2
6, 917	-	t _e	0.9495	5		Point °C	32.	5
1410.0	2	tc		$oldsymbol{ol}}}}}}}}}}}}}}}}}}$	Fire l		 	-
206.0	2	ΔHc kcal/m			M Spe Ultra			
1.374		ΔHf ΔFf			X-Ray	y Dif.	ļ	
1321.0	5	Viscosity		\vdash	Infrar		ļI	<u> </u>
1		centistokes					_	
		η °C		1 1			80	
7 097	14 5						80	İ
1501.9	5			\perp			80	ı
214.3	5				Etha	nol	∞	l
							1	ĺ
		v.						
1799.9	5			<u> </u>				ĺ
255.7	5				1			İ
		c _D vap.300°K	0.39649	2	1			İ
		400			İ			l
<u> </u>	5	v (up .15.0 C	0.501	لئا	L		l	<u> </u>
								<u> </u>
ES: 1-Do			alc. from det	t. dat	a 5-C	alc. by for	mula	
		~						
E REFE	RENCES	3 ASTM Spe	c. Tech. Pub	. No.	109			
	1501. 9 214. 3 1.554 1412. 9 7.396 1799. 9 255. 7 159. 28 T _c ES: 1-Do	214.3 5 1.5590 5 1412.9 5 7.39626 5 1799.9 5 255.7 5 159.28 5 T _C ES: 1-Dow 2-AF ION: AF	7.0924 5 1501.9 5 214.3 5 1.5590 5 1412.9 5 7.39626 5 1799.9 255.7 5 159.28 5 T _C ES: 1-Dow 2-API 3-Lit. 4-C API ION: API	7 °C 7 °C 7 °C	7 °C 7, 0924 5 1501, 9 5 214, 3 5 1, 5590 5 1412, 9 5 7, 39626 5 1799, 9 5 255, 7 5 6	7. 0924 5 1501. 9 214. 3 5 1. 5590 5 1412. 9 5 1799. 9 255. 7 5 C _p liq. 300°K 0.39979 5 C _p vap.300°K 0.39649 2 400 0.50136 2 C _p vap.300°K 0.39649 2 400 0.50136 2 C _p vap.300°K 0.381 3 T _C ES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-C	Centistokes	Centistokes Acetone Carbon tet. Sensene Sensen

								No. 43	
NAME _	3-Meth	yloc tane				ST	RUCTURAL	FORMUL	A
					1		ÇН ₃		
<u> </u>						(CH ₃ CH ₂ CH (C	н.).сн.	
Mole		Molecul		Molecular	.	•	30112011 (0	72/4	
% Pur.		Formula	7 20	Weight 128.25					
	1	Ref.			Ref.	<u> </u>			Ref.
F, P. °C	-107.6	2	dt/dP			f	300 to	0.0496	5
F.P. 100%	<u> </u>		°C/mm 25°C	2,595	5	g	'_6 <u>0</u> 0_• <u>K</u>		5
B. P. °C 760 mm	144.18	2	BP	0.04966	4	h_	l	-0.0635	5
100	81.	2	t _e	0.0360	5	f'	to		
30	53.71	4	30 mm	0.6920	4	g'	<u>*K</u>		
10	33. -2.1	2 5	∆Hm cal/g			h'	<u> </u>		l
Pressure	 	+-	ΔHv cal/g			m n	300 to	0.0294 0.0015	4
mm 25°C	6.367	5	25°C	83.4	2	0	'_000_E	-0.0652	4
te	1149.4	5	30 mm BP	79.76 68.5	4 2	<u> </u>			 -
Density			t _a	66.51	5	m' n'	700 to	0.1222	4
g/ml 20°C	0.720 0.716		t _e (d, e)	66.47	5	٥'	<u> </u>	-0.0645	4
d ₄ 25 30	0.712		ΔHv/T _e	19.67	5		<u> </u>	-	\vdash
a	0.736	3 4	d 50 to	86.45	4		face tension es/cm. 20°C	22.34	2
ь	-0.037		160 °C	0.1245 86.57	4 4	8'	30	21.40	2
Ref. Index			e' 50 °C	0.1267	4		40	20.46	2
n _D 20°C	1.406		d _c g/ml			Par	rachor [P] 20°C		
30	1.401		v_ml/g		_		30		
"C"	0,750	8 4	<u>_</u> c	313.8	5		40		_
MR (Obs.)	43.73	2	P _c mm	16791.	5			385.2	5
MR (Calc.	43.762	5	PV/RT 25°C	1,0000	5	Exp	o. L.1.%/wt. u.		
(nD-d/2)	1.045		30 mm	1.0000	5	Dis	persion	97.5	2
Dielectric	1.977		BP	0.9638	4	Fla	sh Point C		
A 50 to B 195 °C	6.910 1411.0	2 2 2	te tc	0.9525	"	Fir	e Point		
c 1122	206.0	2	ΔHc kcal/m	 			Spec.		
A* 50 to	1.360	1 5	ΔHf				ra V. Ray Dif.		1
B* 170 °C		5	ΔFf	ļ			ared		
K	1		Viscosity centistokes			Sol	ubility in +		
t _k – tō	•		η °C			Ac	etone	oc	1
t⊈ °C			•				rbon tet. enzene	οο οο	
A' 20 to	7.041			1		Et	her	00	
B' _ 50 °C	1480.2	5 5	B _v to				Heptane hanol	οο οο	Ì
A'* 20 to	1,510		ĂV I °C				ater	~	l
B' ≠ 50 °C		5	(B ^V) to	1		W	ater in		
Acl 195 to	7. 388		(A ^V) °C						
Bc tc °C	1803.0	5 5	c liq. 300°K	0.39979	5				
Cc	256.2		P 400	0.50256	5				
Cryos. A° consts. B°	1		c _p vap.300°K 400	0.39649 0.50136	2 2				
t _e °C	160.52	5	c vap.	0.30130	-				
$T_R = 0.7$	L		L <u>. </u>	<u> </u>	Li	٠.		<u> </u>	
		2 41	DT 2 144 4	Cala from 1	- د ه		Cala ha far		ıt
REFERENC	<u></u>		PI 3-Lit. 4- PI	Calc, from de	t. da	ua 5	-Calc. by for	IIIUI A	
SOURCE:									
PURIFICAT			PI						
LITERATU	RE REFE	RENCES	: :						

								No. 44	<u> </u>
NAME	4 - Methyl	octane				STF	RUCTURAL	FORMULA	
<u> </u>				· · · · · · · · · · · · · · · · · · ·	\neg		сн3		
					\neg	СН	₃ (сн ₂) ₂ сн (CH ₂),CH,	
Mole		lolecul		Molecular Weight 128.2	50		3. 2.2 .	2.3 3	
% Pur.	<u></u>	ormul	*	Weight 120.2					Ref.
E D 46	-113,2	Ref.		1	Ref	<u> </u>	1 1		-
F.P. *C F.P. 100%		+-	dt/dP *C/mm	1		f g	300 to 600 °K	0.0496 0.0013	
B. P. °C		+	25°C	2.445	5	h		-0.0635	1
760 mm 100	142.48	2 2	BP t _e	0.049 0.0360	2 5	f'	to		
30	80. 52.53	4	30 mm	0,6882	4	g'			
10 1	32. -2.80	2 5	AHm cal/g			h'	l		
Pressure	-2.80	+3	ΔHv cal/g			m	300 to	0.0294	
mm 25°C	6.756	5	25°C 30 mm	83.4 79.63	2 4	n o	_600 •K	0.0015 -0.0 ₆ 52	
t _e	1141.1	5	BP	68.2	2	m'	700 to	0,1222	4
Density g/ml 20°C	0.7199	2	t _e (d, e)	66.23	5	n'	1000 K	0.0013	4
at 25	0.7160	2	ΔHv/T _e	19.68	5	0'	i	-0.0 ₆ 45	4
	0.7121	4	d 55 to	 	4		ace tension		
a b	0.7355 -0.0 ₃ 778	5 5	e 160 °C	0.1271	4	dyne	s/cm, 20°C 30	22.34 21.40	2
Ref. Index		\top	d' 10 to		4 4	L	40	20.46	2
n _D 20°C	1.4061	2 2	d _c g/ml			Par	achor [P]		
30	1.4015	4	ll V mi/g	1	_		20°C 30		
"C"	0.7515	4	tc °C	311.0	5		40	305 3	_
MR (Obs.)		2	P _c mm	16810.	,	- FC		385.2	5
MR (Calc. (nD-d/2)	1.0462	5 2	25°C	1.0000	5	Exp	. L.1.%/wt. u.		
Dielectric	1.977	5	30 mm BP	1.0000 0.9615	5 4	Disp	persion	97.5	2
A 50 to		2	t _e	0.9502	5		sh Point °C Point		
B <u>[165 °C</u>	1406.0	2	tc			M S			
C	206.0	2	ΔHc kcal/m ΔHf			Ultr	a V.		
A* 50 to B* 170 °C		5	ΔFf				ay Dif. ared		
к — — –	-		Viscosity			 	bility in +		
t _k			centistokes 7°C			Ac	etone	∞	
t _x • •			'				rbon tet. nzene	00 00	
A' 20 to B' <u>55</u> °C		5				Eth		oc	
C, L 33 7	208.5	5	B ^V to				Heptane nanol	00 00	
A!# 20 to		5	AV °C	_		Wa	ter ter in		
	7 3010	5	(B ^V) to			- wa	PET III		-
Ac 165 to	7.39191 1794.2	5	(A ^V) °C						
Cc - c-	255.6	5	c _p liq. 300°K 400	0.39979 0.50256	5				
Cryos. A° consts. B°			c _p vap.300°K	0.39649 0.50136	2				
t _e °C	158.42	5	c _w vap.	0.50130					
$T_{\mathbf{R}} = 0.7$				J	لــــــا	+ gra	ams/100 grai	ns solveni	t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	t. dat				
SOURCE:		AI	PI						
PURIFICA'		AI							
LITERATU	RE REFER	ENCES	5:						

								No. 45	
NAME	3-Ethylhe	ptane	•			ST	RUCTURAL	FORMUL	A
							C ₂ H ₅		
Mole % Pur.		lecul		Molecular Weight 128.25	50	C	н ₃ сн ₂ сн (с	H ₂) ₃ CH ₃	
		Ref.			Ref.				Ref.
F.P. °C F.P. 1009	<u>.</u>	├	dt/dP °C/mm			f	300 to	0.0496	5
B. P. °C		 	25°C	2.457	5	g h	'_6 <u>0</u> 0_ °K	0.0013 -0.0 ₆ 35	5
760 mm 100	143.0	2 2	BP t _e	0.0496 0.0359	5	f'	to	0.0633	Ť
30	80.3 52.7	4	30 mm	0.6904	4	g'	'° <u>K</u>		
10	32. -2.65	5	∆Hm cal/g			h'			L_
Pressure			ΔHv cal/g 25°C	02 5		m n	300 to 600 °K	0.0294 0.0015	4
mm 25°C	6.715 1152.5	5	30 mm	83.5 79.44	2 4	0		-0.0 ₆ 52	4
Density	1	+-	BP t_	68.5 66.55	5	m'	700 to	0.1222	4
g/ml 20°0	0.727 0.723	2 2	te te (d, e)	66.51	5	n' o'	1000 •K	0.0013 -0.0 ₆ 45	4
d ₄ 25 30	0.719	4	е	19.73	5	Sur	face tension	О	
a b	0.743	4	d 50 to e 160 °C	85.82 0.1211	4 4		es/cm. 20°C	22.81	2
Ref. Index	-0.03798	4	d' 10 to	87.17	5	•	30 40	21.87 20.93	2
ⁿ D 20°C	1.4093	2	d _c g/ml	0.1407	+-	Par	achor [P]		
30	1.4070 1.4045	2 4	v _c mi/g	212 1	5		20°C 30		
"C"	0.7496	4	t _c °C P _c mm	313.1 16895.	5		40 Sugd.	385 2	5
MR (Obs.) MR (Calc.		2 5	PV/RT	1	+	Exp	. L.1.%/wt.	303.2	<u> </u>
(nD-d/2)	1.046	2	25°C 30 mm	1.0000 1.0000	5 5		u. persion	96.6	,
Dielectric	+ -: 700	5	B P	0.9682	4		sh Point C	28.	5
A 50 to B 1195 °C		2	te t _c	0.9575	5	Fir	e Point		Ĺ
c	206.	2	∆Hc kcal/m		t		Spec. ra V.		
A* 50 to B* 170 °C		5	ΔHf ΔFf			X-I	Ray Dif.		
к	-		Viscosity				ability in +		_
t _k — tō			centistokes 7°C			Ac	etone	∞ 0	ĺ
'x							rbon tet.	0 0	
A' 25 to B' _55 °C		5 5			\sqcup		her Hep ta ne	oc oc	
C'	205.	5	B ^V to A ^V °C			Et	hanol ater	∞ ∞	
A'* 25 to B'* 55 °C		5	$\frac{\mathbf{B}^{\mathbf{v}}}{(\mathbf{B}^{\mathbf{v}})} - \frac{\mathbf{v}}{to}$	-			ter in		
Ac 195 to	7 37800	5	(A ^V) °C						
Bc t _c °C	1793.5 - 256.2	5	c _p liq. 300°K	0. 39979	5				
Cryos. A°		Ť	c vap.300°K	0.50256 0.39649	5 2	1			1
consts. B		ļ_		0.50136	2				
t_e °C $T_R = 0.7$	159.42	5	c vap.	1	$oxed{oxed}$	L	(100		<u> </u>
	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t de		ams/100 gra		t
SOURCE:			PI	11 om de	ua		by 101		
PURIFICA	TION:	A	PI						
LITERATU	RE REFERE	NCE	5:						

							No. 46
NAME	4-Ethylhe	ptane				STRUCTURAL	FORMULA
Ī						Ç ₂ 1	ł.
					$\neg \uparrow$	CH ₃ (CH ₂) ₂ CH	
Mole % Pur.		olecul ormul		Molecular Weight 128.2	50	3 2 2	22 3
		Ref.	 		Ref	<u> </u>	Ref
F. P. *C			dt/dP	1		f 300 t	0.0496 5
F.P. 1007	9		°C/mm		_	g 600°	
B. P. *C 760 mm	141.2	2	25°C BP	2.3002 0.050	5 2	h	-0.0635 5
100	79.	2	t _e	0.0358	5	f' t	
30 10	51.4 31.	4 2	30 mm	0,6864	4	g' ' <u>°</u> '	` -
ĭ	-3.41	5	AHm cal/g				0 0304 4
Pressure			ΔHv cal/g 25°C	83.5	2	m 300t	
mm 25°C	7.174 1147.7	5	30 mm	79.27	4	•	-0. 0 ₆ 52 4
Density	+	+-	BP t	68.3 66.43	2 5	m' 700t	
g/ml 20°C		2	te te (d, e)	66. 33	5	n' 1000°1	S 0.0013 4 -0.045 4
dt 25	0.726 0.722	2 4	AHv/Te	19.79	5		°
a	0,746	4	d 50 to		4	Surface tension dynes/cm. 20%	
ь	-0.038	4	-a,- -155 %		4 4	30	21.87 2
Ref. Index		2	e' 50 °C		4	40	20.93 2
D 25	1.4073	2	d _c g/ml			Parachor [P]	
30	1.4049	4	tc *C	311.0	5	30 40	
"C"	0.7472	4	P _c mm	17072.	5	H .	d. 385.2 5
MR (Obs.) MR (Calc.		5	PV/RT			Exp. L.1.%/wt	.
(nD-d/2)	1.0446	2	25°C 30 mm	1.0000 1.0000	5	u. Dispersion	96.6 2
Dielectric	1.987	5	BP	0.9676	4	Flash Point °C	26.0 5
A 50 to B 190 °C		2 2	t _e t _c	0.9582	5	Fire Point	
c 1.70 c	206.	2	ΔHc kcal/m	 	\vdash	M Spec.	
A* 50 to		5	ΔHf ΔFf			Ultra V. X-Ray Dif.	1
B* ∟ ¹⁷⁰ °	1305.	5	Viscosity	 	-	Infrared	
c	_		centistokes			Solubility in Acetone	l 1
t _k to			η •ο			Carbon tet.	&C
A' 20 to	_1	5				Benzene Ether	eo eo
B' _ 55 °C		5	B ^V to	+		n-Heptane	
A'* 20 to	200.	5	B to A °C			Ethanol Water	∞
	1.270	5	(B ^V) to	-		Water in	
Ac 190 to	7, 38178	5	(A ^V) °C	1			
Bc tc_°C	256.0	5	cp liq. 300°K		5		
Cryos. A°		+	400 c _p vap.300°K	0.50256	5 2	Į.	
consts. B°			400	0.50136	2		
t _e °C	157.31	5	c _v vap.				
$T_R = 0.8$						f grams/100 gr	ams solvent
	CES: 1-Dow	2-A1		Calc. from det	t. da	ta 5-Calc. by fo	rmula
SOURCE:		AI					
PURIFICA		AI					
LIIEKATU	RE REFERE	NCES	i:				

								No. 47	
NAME	2, 2-Dime	thylh	neptane			ST	RUCTURAL	FORMUL.	A
							CH ₃		
Mole % Pur.		ecul		Molecular Weight 128,25	50	C	H ₃ C (CH ₂) ₄ CH ₃	CH ₃	
		Ref.			Ref.				Ref.
F.P. °C F.P. 1009	-113.00	2	dt/dP °C/mm			f g	300 to 600 K	0.0496 0.0013	5 5
B. P. °C	122 (0	_	25°C BP	1.582 0.049	5 2	h		-0.0635	5
760 mm	132.69 71.0	4	t _e	0.0368	5	f'	to °K		
30 10	43.82 23.3	5	30 mm	0.6775	4	g' h'	<u>~</u>		
1	-11.	5	ΔHm cal/g ΔHv cal/g	 	-	m	300 to	0.0294	4
Pressure mm 25°C	11.06 1097.6	5	25°C 30 mm	78.8 76.62	2 4	n o	_6 <u>0</u> 0_•K	0.0015 -0.0 ₆ 52	4
Density	+	-	BP te	64.8 63.00	2 5	m'		0.1222	4
g/m1 20°0 dt 25 4 30	0.7066	2 2	t _e (d, e) ΔHv/T _e	62.91	5	n' o'	[10 <u>00 *K</u>]	0.0013 -0.0 ₆ 45	4
a 30	0.7027	4	d 45 to	82.44	4		face tension es/cm, 20°C	20.80	2
ь	-0.0378	4	145 °C	0.1330 81.70	4 4	ชังก	30	19.88 18.97	2 2
Ref. Index		2	e' 45 °C	0.1160	4	Par	achor [P]	10.97	۴
D 25	1.3993	2	d g/ml vc ml/g			1	20°C		
"C"	0.7527	4	vc ml/g tc °C	295.2	5		3 0 4 0		
MR (Obs.)		2	P _c mm	15656.	5	<u> </u>		385.2	5
MR (Calc. (nD-d/2)		5	PV/RT 25°C	1.0000	5	Exp	u. L.1.%/wt.		
Dielectric	1.0404	-	30 mm BP	1.0000 0.9506	5 4		persion	99.5	2
A 40 to B 165 °C		2 2	t e t c	0.9390	5	Fir	sh Point °C e Point		
С	208.00	2	ΔHc kcal/m ΔHf			Ult	Spec. ra.V.		
A* 40 to B* 155 °C		5	ΔFf				Ray Dif.		
к — — -	_		Viscosity centistokes				ubility in +	_	
t _k to			η •c			Ca	etone rbon tet.	88	
A' 10 to		5		1			nzene her	80	
B' 1_45 °C	1440.55 215.8	5	B _v to				Heptané hanol	∞	
A'* 10 to B'* 45 °C	1.4984	5	$\frac{\mathbf{A}^{\mathbf{V}}}{ \mathbf{B}^{\mathbf{V}} } - \frac{\mathbf{C}}{\mathbf{to}}$			w.	ater ater in	80	
Ac 165 to	7, 33533	5	(A ^V) °C						
Bc t _c °C	1733.9 256.9	5 5	c_ liq. 300°K	0.39979	5				
Cryos, A° consts, B°	0.042	2	c _p vap.300°K	0.50256 0.39649 0.50136	5 2 2				
t _e °C	146.94	5	c _v vap.						
$T_{\mathbf{R}} = 0.7$	7 T _c			-		+ g1	ams/100 gra	ms solven	t
	CES: 1-Dow			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:	TION		PI						
PURIFICA	TION: RE REFERE		PI 						
ERAIL	NE REFERE	40E	·						
									-

———								No. 48	
NAME	2,3-D	imethylhe	ptane		_	STR	UCTURAL 1	FORMULA	4
						CF	сн ₃ 1 ₃ сн сн - (с	:н.).сн.	
Mole	Ref.	Malasul		Molecular		0.	сн ₃	7112/30113	
% Pur.	Rei.	Molecula Formula		Weight 128.2	250		0113		
		Ref.			Ref				Ref
F.P. *C			dt/dP			f	300 to	0.0496	5
F.P. 1007	T		°C/mm		_	g	_ 600°K	0.0013	
B.P. °C			25°C BP	2,182 0,0496	5 4	h		-0.0635	5
760 mm 100	140.5 78.	2 2	t _e	0.0362	5	f'	to		
30	50.3	4	30 mm	0.6885	4	g'			
10 1	29.4	5 5	ΔHm cal/g			h'			
Pressure	+	-+-	ΔHv cal/g			m	300 to	0.0294 0.0015	
mm 25°C	7.73	8 5	25°C 30 mm	81.6 78.51	2 4	n o		-0.0 ₆ 52	
te	1137.3	5	BP BP	67.3	2				
Density g/ml 20°C		[]	t _e ,,,	65.44	5	m'	700to 1000°K	0, 1222 0, 0013	4
	0.72		[t (a, e)	65.33	i 1	o'	·	-0.0645	
d ₄ 25	0.71		ΔHv/T _e	19.54	5	Surf	ace tension		\vdash
	0.74		d 50 to		4 4		s/cm. 20°C	22.34	2
ь	-0.03	78 4	d' 10 to	84.66	4	y	30 40	21.40 20.47	2 2
Ref. Index		85 2	e' 50 °C	0, 1223	4	B	chor [P]	20.41	-
ⁿ D 25	1.40	062 2	d g/ml v ml/g			Para	20°C		1
30	1.40		t _c °C	309.5	5		30		
"C"	0.75		Pcmm	16649.	5		40 Sugd.	385.2	5
MR (Obs.)			PV/RT	 	\vdash	Exp.	L.1.%/wt.		+
MR (Calc. (nD-d/2)	1.04		25°C	1.0000	5	1	u.		1
Dielectric	1.98	4 5	30 mm BP	1.0000 0.9619	5 4		ersion	97.2	2
A 50 to	6.88	7 2	te	0.9517	5		h Point °C Point	26.0	5
B 175°C	1392.0	2	t _c			M Sp			
C	207.0	2	ΔHc kcal/m ΔHf			Ultra	a V.		
A* 50 to B* 165 °C	1.34 1302.4	21 5	ΔFf	I		X-Ra Infra	ay Dif.		1
K	-		Viscosity			ļ			-
t. 1	-	i i	centistokes 7°C	ľ			bility in +	∞	
t _x to			η ℃				bon tet.	oc	
A' 10 to	7.02	69 5		ĺ	1 1	Eth	zene er	8 0	
B' L 50 °C		5	B ^V l to	 	\vdash		leptane	∞	j
	213.7	5	B' to	ļ		Wat	anol	oc	1
A'* 10 to B'* 50 °C		68 5	(B ^V) to	1			er in		
Ac 175 to		579 5	(A ^V) °C	ĺ	l				İ
Bc tc_'	1782.3	5	c _p liq. 300°K	0.39979	5				
Cc	257.3	5	400	0.50256	5				
Cryos. A ^c consts. B ^c			c _p vap.300°K 400	0.39649 0.50136	2 2				
t _e °C	156, 34	5	c _w vap.	0.30130					
$T_{\mathbf{R}} = 0.7$		لــُـــــــــــــــــــــــــــــــــــ	<u> </u>			+ ~~~	ma/100	no acless :	<u> </u>
REFEREN		ow 2-AF	PI 3-Lit. 4-0	Calc. from det	def		ms/100 gran		<u> </u>
SOURCE:		AF		Jule, II om de	. udi	<u>)-</u>	Care. by for		
PURIFICA	TION:	AF							
LITERATU									
IBAA10	NE REF	ENENCES) ,						

No. 49 2,4-Dimethylheptane NAME STRUCTURAL FORMULA CH3CH - CH2CH (CH2)2CH3 Ċнз ĊHą Mole Ref. Molecular Molecular Weight 128, 250 $C_{9}H_{20}$ % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g _°K 25°C 1.606 4 B. P. ℃ h BP 0.049 2 133.5 760 mm 2 0.0359 5 ſ١ to 71. 100 2 g' <u>• K</u> 44.23 .30 4 30 mm 0.6772 4 10 23.8 5 h' ∆Hm cal/g 5 -9.6 300 to 0.0294 m ∆Hv cal/g Pressure 0.0015 | 600 °K n 25°C 79.9 2 mm 25°C 10.738 4 o -0.0₆52 4 30 mm 76.85 5 1131.2 t_e 5 BP 2 65.9 700 to 1000 °K m' 0.1222 4 Density 64.85 5 te te (d, e) n' 0.0013 g/ml 20°C 4 0.716 2 64.07 5 ۰, -0.0645 4 $\mathbf{d_{4}^{t}}$ 25 0.712 2 AHV/Te 5 19.73 30 0.708 Surface tension d 45 82.27 4 to 0.732 a 4 dynes/cm. 20°C 21.30 1<u>50</u> °C 0.1226 4 -0.0₃796 Ъ 4 20.38 2 30 ă¬` 10 to 83.87 19.46 2 40 Ref. Index e¹ 45 0.1587 5 20°C 1.4033 [P] ⁿD Parachor d_c g/ml 25 1.4010 2 20°C vc ml/g tc °C 30 1.398 4 30 297.7 5 40 "C" 0.7501 4 P_c mm 16179. 5 Sugd. 385.2 5 MR (Obs.) 43.7 2 PV/RT Exp. L.l.%/wt. MR (Calc.) 43.762 5 25°C 1.0000 5 (nD-d/2) 1.0453 2 30 mm 1.0000 5 97.8 2 Dispersion Dielectric BP 0.9610 5 Flash Point C 0.9642 A 40 to 6.869 2 Fire Point В 1360. 2 M. Spec. Ultra V. С 208. 2 AHc kcal/m ΔHf A* 40 to 5 1.3129 X-Ray Dif. ΔFf B*[160 °C 1267. Infrared ĸ Viscosity Solubility in centistokes Acetone to tk tx Carbon tet. °C Benzene 25 to 6.7010 Ether В' 45 °C 1277. 5 n-Heptane Bv | Av | 200. 5 to Ethanol ۰c Water 25 to 1.1957 5 B'* 45 °C Water in 1199. (B^V)| to Acl 165 to 7.34717 5 (A^V) °C Bc tc °C 1741.6 5 c_p liq. ۰ĸ 5 Cc' 257 3 cp vap 300°K Cryos. A 0.39649 2 consts. B° 400 0.50136 c vap. te °C 148.44 5 T_{R} $= 0.77 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

								No. 50	
NAME	2,5-Dime	hylhe	ptane			STR	UCTURAL I	FORMULA	
						CH	CH (CH) C		
						CH	CH (CH ₂) ₂ C		3.
Mole		lecul		Molecular			Ċн ₃ с	H ₃	
% Pur.	Fo	rmul	a 920	Weight 128.2	$\overline{}$				
		Ref.		,	Ref				Ref.
F.P. *C F.P. 100%	<u> </u>		dt/dP	ļ		f	300 to	0.0496	
	<u> </u>	1	*C/mm 25*C	1.847	5	g	600 <u>*</u> K_	0.0013	
B. P. *C 760 mm	136.0	2	BP	0.049	2	h_	 	-0. 0 635	5
100	74.1	2	t _e	0.0362	5	f' g'	to		
30 10	46.9 26.2	4 2	30 mm	0,6802	4	h'	' ' -		
1	-8.4	5	∆Hm cal/g				300 to	0,0294	4
Pressure			ΔHv cal/g 25°C	70.0	2	m n	600°K	0.0274	
mm 25°C	9.338 1119.5	5	30 mm	79.9 77.80	5	0		-0.0 ₆ 52	4
Te te	1119.5	13	BP	66.4	2	m'	700 to	0.1222	4
Density g/ml 20°C	0.715	2	te te (d, e)	64.56 64.48	5	n'	1 <u>1000°K</u>	0.0013	4
dt 25	0.711	2	ΔHv/T	19.52	5	٥'	i	-0.0645	4
4 30	0.707	4	d 45 to		4	Surf	ace tension		
a b	0.731	4	e 150 °C		4	dyne	s/cm. 20°C	21.30	2
Ref. Index	-0.038	+ +	a' 10 to	82.29	4	•	30 40	20.38 19.46	2 2
n _D 20°C		2	e' i 45 °C	0.0957	4	Para	chor [P]		
45	1.4015	2	d g/ml vc ml/g	Į.			20°C		
30	1.3989	4	t _c •C	300.7	5		30 4 0		
"C"	0.7526	4	P _c mm	16142.	5			385.2	5
MR (Obs.) MR (Calc.		5	PV/RT		Н	Exp.	L. 1. %/wt.		
(nD-d/2)	1.046	2	25°C	1.0000	5		u.		_
Dielectric	1.971	5	30 mm BP	1.0 00 0 0.9591	5 4		ersion	97.8	2
A 45 to	6.881	2	t _e	0.9484	5		h Point °C Point	23.0	5
B 1170°C		2	t _c		Ш	M Sp			
C 45.	207.	2	ΔHc kcal/m ΔHf			Ultra	a V.		
A* 45 to B* 160 ℃		5	ΔFf			X-Ra Infra	ay Dif.		
к — — –	-	-	Viscosity				bility in +		
c	-	1	centistokes 7°C				tone	80	ĺ
t _k to			η •⊂				bon tet.	60	
A' 10 to	7.2515	5			1 1	Eth	zene er	80 80	
B' _ 50 °C		5	B ^V to	 	\vdash		leptane	60	
	224.4		B' to			Eth Wat	anol	•	
A'* 10 to B'* 50 °C		5	725. — — —	1			ter in		L
Ac 185 to	7,35739	5	(A ^V) to						
Bc t °C	1753.3	5	c _p liq. 300°K	0,39979	5				İ
Ce	256.0	5	400	0.50256	5				
Cryos, A° consts, B°			c _p vap.300°K 400	0.39649 0.50136	2 2				
t _e °C	151.04	5	c _w vap.						
$T_R = 0.7$	7 T _C					+ gra	ms/100 gran	ns solveni	<u> </u>
REFERENC	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from det	dat		Calc. by for		
SOURCE:		A1	PI						
PURIFICAT	TION:	AJ	PI						
LITERATU	RE REFERE	NCES	3:						
1									

								No. 51	·
NAME	2,6-1	Dimethyl	heptane			ST	RUCTURAL	FORMUL	A
<u> </u>						C	н ₃ сн (сн ₂) ₃	сн - сн	
Mole % Pur.	Ref.	Molecu Formul	lar C ₉ H ₂₀	Molecular Weight 128.2	50		- 1	Ċн ₃	
		Rei	4		Ref.				Ref.
F. P. ℃	-102.9	2	dt/dP			ſ	300 to	0.0496	5
F.P. 100%			°C/mm			g	_6 <u>0</u> 0_ ° K	0.0013	5
B. P. ℃	l		25°C BP	1.780	5 2	h		-0.0 ₆ 35	5
760 mm 100	135.21		te	0.0363	5	ſ١	to		
30	46.18	4	30 mm	0.6793	4	g'	'° <u>K</u>		
10	25.5	4 5	∆Hm cal/g			h'	ļ		
Pressure	 		∆Hv cal/g			m n	300 to	0.0294 0.0015	
mm 25°C	9, 70		25°C 30 mm	79.8 77.55	2	0	-000 Tr	-0.0652	
t _e	1117.7	5	BP	66.2	2	m'	700 to		4
Density g/ml 20°C	0.70	89 2	te (3 a)	64.35	5	n'	1000 K	0.1222 0.0013	4
dt 25	0.70		te (d, e)	t	5	0'		-0.0 ₆ 45	4
⁴ 30	0.70	09 4	ΔHv/T _e	19.49		Sur	face tension		!
a	0.72		d 45 to e 150 °C	83.44 0.1275	4		es/cm. 20°C	20.83	2
b	-0.03	80 4	d 10 to	82.45	4	,	30 40	19.92 19.00	2
Ref. Index		07 2	e' 45 °C	0.1061	4	Par	achor [P]		Ħ
25	1.39	83 2	d g/ml vc ml/g				20°C		
30	1.39		± °C °	298.6	5		30 4 0		
"C"	0.75		P _c mm	15863.	5			385.2	5
MR (Obs.) MR (Calc.			PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	Die	u. persion	98.9	2
Dielectric	1.96	2 5	∥ RD	0.9599	4		sh Point C	22.0	5
A 45 to	6.87		t _e	0.9488	5		e Point	22.0	
B 1_170 °C	1366.0	2 2	tc AHc kcal/m		-	M.	Spec.		
A* 45 to	1.33		ΔHf				ra V. Ray Dif.		
B*[160 °C		5	ΔFf	ļ.,			ared		
K	1		Viscosity centistokes			Sol	ability in +		
t _k to	•		7 °C				etone rbon tet.	00	
t _x °C			1			Be	nzene	∞ ∞	
A' 10 to B' 50 °C	7.15 1510.6	10 5					her Vantana	00	
c'	220.1	5	B _v to				Heptane hanol	oc oc	
A** 10 to	1.61		A C	_			ter		
B'* 50 °C	1419.0	5	(B ^V) to	1			ter in		├
Ac 170 to Bc t _c °C	7.34 1741.9	814 5	(A ^V) °C						
Cc C	255.7	5	c _p liq. 300°K	0.39979	5				
Cryos. A°			c _p vap300°K	0.50256 0.39649	5 2				1
consts. B°	ļ		n 400	0.50136	2				
t _e °C	150.20	5	c _v vap.	<u> </u>	\perp	<u> </u>			<u> </u>
$T_{\mathbf{R}} = 0.7$							ams/100 gra		t
REFERENC	CES: 1-D			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			API						
PURIFICAT			API						
LITERATU	RE REF	ERENCE	S:						
1									

							No. 52	
NAME	3, 3-E	imethyll	neptane			STRUCTURAL CH3	FORMUL	A.
						сн ₃ сн ₂ с -		
Mole	Ref.	Molecu	lar C ₉ H ₂₀	Molecular Weight 128.	350	Ċн ₃		
% Pur.		- T		Weight 128.	Ref	r	·	Ref.
F, P. *C	T	Ref	1	T	Kei	 	1	101.
F.P. 100%	†		dt/dP *C/mm		} }	f t	K.	1
B. P. °C			25°C BP	1.833	5 2	h ,		ĺ
760 mm 100	137.3	2 2	t.	0.0367	5	f' t		
30	46.9	4	30 mm	0.6897	4	- :	K_	l
10 1	26. -8.4	5	ΔHm cal/g			h'		<u> </u>
Pressure	† — —		ΔHv cal/g	-0.5		m 300t		
mm 25°C	9.45	3 5	25°C 30 mm	79.5 76.72	5	0	-0.0652	
t _e Density	1124.		BP	65.8	2	m' 700 t		
g/ml 20°C	0.72		te te (d, e)	64.01 63.94	5	n' 1000°	K 0.0013	4
dt 25 4 30	0.72 0.71		AHv/T	19.28	5	o'	-0.0645	4
	0.71		d 45 to		4	Surface tension dynes/cm. 20°		2
ь	-0.03		150 °C		4	30	21.09	2
Ref. Index			e' 45 °C		4	40	20.17	2
ⁿ D 20°C	1.40		d _c g/ml			Parachor [P]	۱.	İ
30	1.40		v _c ml/g t _c °C	304.5	5	30	١	l
"C"	0.75	03 4	P _c mm	16081.	5	40 Sug	d. 385.2	5
MR (Obs.) MR (Calc.)	43.7 43.76	2 2	PV/RT	+	\vdash	Exp. L.1.%/wt		H
(nD-d/2)	1.04		25°C	1.0001	5	u.		١,
Dielectric	1.98	4 5	30 mm BP	1.0000 0.9591	5	Dispersion Flash Point °C	97.4	2
A 45 to			ţ.	0.9484	5	Fire Point		1
B <u> 170°C</u>	1385.	2 2	t _c	+	\vdash	M Spec.		
A* 45 to	1.33	05 5	ΔHf			Ultra V. X-Ray Dif.		1
B* ∟160 °C	1296.	5	ΔFf Viscosity	+	\vdash	Infrared		<u> </u>
c		İ	centistokes			Solubility in Acetone	+	1
tk C			η •α			Carbon tet.		
t _x *C		30 5	1			Benzene Ether		
B' _ 50 ℃		5 5	B ^v to	 	\vdash	n-Heptane	ĺ	
A'* 10 to	1.36		B to to ℃	1		Ethanol Water		1
B'* 50 °C		5	(B ^V),	-		Water in		1
Ac 170 to	7.35		(A ^V) •C	:				
Bc tc_C	1779.0 260.9	5 5	cp liq. °K					
Cryos. A°	1		c _p vap.300°K	0.39649	2			
consts. B°	<u> </u>		400	0.50136			}	
t _e °C	152.73	5	c _v vap.	1	<u> </u>			<u></u>
$T_{\mathbf{R}} = 0.77$						† grams/100 gr		t
REFERENC	ES: 1-D			Calc. from de	t. da	ta 5-Calc. by fo	rmula	
SOURCE: PURIFICAT	TON:		PI					
LITERATU			PI s.					
	NE KEFI	an ence	J.					

No. 53 3, 4-Dimethylheptane NAME STRUCTURAL FORMULA CH₃ CH3CH2CH CH (CH2)2CH3 Mole Ref. Molecular Molecular ċнз $C_{9}H_{20}$ % Pur Weight 128, 250 Formula Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °<u>K</u> g 25°C 2.178 5 B.P. °C h BP 0.050 760 mm 140.6 2 t_e 0.0360 5 f١ 100 78. to 2 g' <u>°K</u> 30 50.31 4 0.6899 4 30 mm 10 29.4 4 h' AHm cal/g 1 -5.09 5 m 300 to 0.0294 4 AHv cal/g Pressure n 1_600 °K 0.0015 0.0₆52 4 4 25°C 81.7 2 mm 25°C 7.742 o 30 mm 78.35 5 1147. 5 t_e ВP 67.8 2 700 to 0.1222 m 4 Density 65, 92 5 t_e (d, e) 0.0013 1000 °K g/ml 20°C 0.7314 2 65.90 o' -0.0645 4 25 $\mathbf{d_4^t}$ 0.7275 2 AHv/Te 19.66 5 30 0.7236 4 Surface tension ď 50 to 84, 22 4 0.74699 a 4 dynes/cm. 20°C 22.80 2 °C 0.1168 å-1<u>55</u> 4 ь -0.03777 4 30 21.87 2 10 to 85.01 5 20.94 2 40 Ref. Index e' 50 0.1325 5 20°C 1.4111 n_D 2 [P] Parachor dc g/ml 25 1.4089 2 20°C vc ml/g t_°C 30 30 t_c 311.1 5 40 "C" 4 0.7483 P_c mm 16953. 5 Sugd 385.2 5 MR (Obs.) 43.55 43.762 2 5 PV/RT Exp. L.1.%/wt. MR (Calc.) 25°C 1.0001 5 (nD-d/2)1.0454 2 u. 30 mm 1.0000 5 5 Dispersion 2 96.8 0.9692 Dielectric ВP Flash Point C A 50 to 0.9588 5 6.897 2 Fire Point B 175 °C 1400. 2 M. Spec. С 208.0 2 AHc kcal/m Ultra V. A* 50 to ΔHf 1.3395 5 X-Ray Dif. ΔFf B* 165 °C 1306.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. ^{t}x °C Benzene A' 25 to 6.9320 Ether B' L 50 °C 1418.1 5 n-Heptane B_V C' 209.7 5 to Ethanol °C A'* 25 to B'* 50 °C Water 5 5 1.4076 Water in 1332.9 (B^V)| to Acl 175 to 7.37833 5 (AV) °C Bc tc °C 1794.8 5 c_p liq. ۰ĸ Cc 258.9 5 Cryos. A c_p vap.300°K 0.39649 2 consts. B° 0.50136 2 400 c vap. te °C 156.83 5 $T_R = 0.77 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 54		
NAME	3, 5-Dim	ethylhe	eptane		STRUCTURAL FORMULA				
						CU CU CU CU C	u cu c	***	
					$\neg \neg$	CH3CH2CH CH2C	n - Cn ₂ C	′ ¹¹ 3	
Mole		lolecul		Molecular	50	Ċн ₃ С	H ₃		
% Pur.	<u></u>	ormul	7 20 1	Weight 128.2				Ref	
	T	Ref.		r	Ref.			Ker.	
F.P. *C F.P. 100%	 	+	dt/dP *C/mm	l		f to	1	1	
B. P. °C	-	+	25°C	1.812	5	g K		l	
760 mm	136.0	2	BP	0.049 0.0363	2 5	l	 	╁	
100 30	73.9 46.59	4 4	t _e 30 mm	ì	4	g' to		1	
10	26.	2		0.6824	+	h'	1		
1	-8.	5	ΔHm cal/g		-	m 300 to	0.0294	4	
Pressure		1 - 1	ΔHv cal/g 25°C	80.1	2	n 600°K	0.0015	4	
mm 25°C	9.495 1124.	5 5	30 mm	77.40	5	° ;	-0.0 ₆ 52	4	
Density	 	+	BP	66.4 64.57	2 5	m' 700 to	0.1222	4	
g/ml 20°C	0.723	2	t _e t _e (d, e)	64.52	5	n' 1000 K			
dt 25 4 30	0.719	2 4	AHv/Te	19.51	5	<u> </u>	-0.0 ₆ 45	4	
	0.715		d 45 to	83.13	4	Surface tension	33. 77		
a - b	0.739	4 4			4	dynes/cm. 20°C	21.77	2 2	
Ref. Index	1	1	d' 10 to		5	40	19.93	2	
n _D 20°C	1.4067	2	d g/ml	0.1231	<u> </u>	Parachor [P]			
25 30	1.4044	2 4	vc ml/g tc C			20°C 30			
"C"	0.7493	4	tc *C	302.3	5	40		[
MR (Obs.)	43.6	2	P _c mm	16335.	5	Sugd.	385.2	5	
MR (Calc.)		5	PV/RT	1 0000	_	Exp. L.1.%/wt.	l	ł	
(nD-d/2)	1.045	2	25°C 30 mm	1.0000 1.0 000	5	u. Dispersion	96.9	2	
Dielectric			BP	0.9626	5	Flash Point °C	70.7	 -	
A 45 to		2	t _c	0.9519	5	Fire Point			
B [_160 °C	208.0	2 2	ΔHc kcal/m		├	M Spec.			
A* 45 to	1,33769	-+	ΔHf		Ì	Ultra V. X-Ray Dif.		l	
B* 160 °C		5	ΔFf		<u> </u>	Infrared	1	ł	
K	1	1 1	Viscosity centistokes	}	1	Solubility in +			
t _k to	1	1 1	7 °c	1		Acetone		1	
'x 1		$oldsymbol{ol}}}}}}}}}}}}}}}}}}$,		l	Carbon tet. Benzene			
A' 25 to B' <u>50</u> °C	6.97157 1423.06	7 5 5			ĺ	Ether		ĺ	
c, L 30 S	212.4	5	B ^v to			n-Heptane Ethanol		l	
A'* 25 to	1.44661	5	Av i °C	l	1	Water		1	
	1336.7	5	(B ^V) to]		Water in		-	
Ac 160 to	7.2756	5 5	(A ^V) °C		L				
Bc tc_C	246.8	5	c _p liq. °K						
Cryos, A°	1	\dashv	c _p vap.300°K	0.39649	2	[
consts. B°	<u></u>		400	0.50136			1	1	
t _e °C	151.27	5	c _v vap.				1	İ	
$T_{\mathbf{R}} = 0.77$	T _c			•	•	+ grams/100 grai	ms solven	<u> </u>	
	ES: 1-Dow	2-AF	I 3-Lit. 4-0	Calc, from de	t. da				
SOURCE:		AF							
PURIFICAT	ION:	AF	PI						
	RE REFERI	ENCES	:						

No 55

								No. 55		
NAME	4, 4-Dime	thylh	eptane		STRUCTURAL FORMULA					
					CH ₃					
					СH ₃ (СH ₂) ₂ С (СH ₂) ₂ СH ₃					
Mole	Ref. Mo	ecul	ar 1	Molecular		01.	.3(0112/20 (2,201.3		
% Pur.		mula		Weight 128.25	0		ĊH ₃			
		Ref.			Ref.				Ref.	
F. P. ℃		1.	dt/dP			f	to			
F.P. 100%			°C/mm		_	gl	<u>°K</u>			
B. P. ℃		1	25°C BP	1.680 0.0496	5 4	h				
760 mm 100	135.2	2 2	te	0.0365	5	f'	to			
30	73. 45.16	4	30 mm	0,6865	4	g'	<u>•K</u>			
10	24.5	5	ΔHm cal/g			h'				
1	-9.	5	ΔHv cal/g			m	300 to	0.0294	4	
Pressure mm 25°C	10.32	5	25°C	79.5	2	n	_600 • K	0.0015	4	
t _e	1128.	5	30 mm BP	76.26	5 2	نـــــا		-0.0 ₆ 52	7	
Density			t _e	65.9 64.05	5	m'	700 to	0.1222	4	
g/ml 20°C		2	te (d, e)	64.11	5	n' o'	1000 °K	0.0013 -0.0 ₆ 45	4 4	
dt 25	0.721 0.717	2 4	AHv/T _e	19.37	5			5.06.3	\Box	
a 30	0.717	4	d 45 to	81.45	5		face tension	22 01	,	
b	-0.03797	4	1150 °C	0.1150	5	gyne	es/cm. 20°C 30	22.01 21.09	2 2	
Ref. Index		\Box	d' 10 to	83.52 0.1609	5		40	20.17	2	
n _D 20°C		2	d _c g/ml	1	<u> </u>	Par	achor [P]			
25 30	1.4053 1.4026	2 4				l	20°C			
"C"	0,7485	4	tc °C	301.7	5		40			
MR (Obs.)	43.6	2	P _c mm	15996.	5		Sugd.	385.2	5	
MR (Calc.)		5	PV/RT		_	Exp	. L.1.%/wt.			
(nD-d/2)	1.045	2	25°C 30 mm	1.0000	5	Dis	u. persion	97.4	2	
Dielectric			BP	0.9673	5		sh Point C	7	Ė-	
A 45 to	6.858	2	t _e	0.9562	5		Point			
B (160 °C	- 1373. 210.	2.	tc AHc kcal/m	-		M.	Spec.			
A* 45 to	1, 30972	5	ΔHf				a V.			
B* 160 °C	1281.5	5	ΔFf		ļ		ay Dif. ared			
к ———]		Viscosity			l——	bility in +			
t _k - to	•		centistokes り °C		ĺ	Ac	etone			
t _x °C		1	'	}	1		rbon tet. nzene			
A' 25 to	6, 62226	5					her			
B' 1_50°C	1255.3	5	B _v to		\vdash		Heptane			
A'* 25 to	198.8	5	B' to A' ℃				hanol iter			
B'* 50 °C	1178.6	5	(B ^V) - to	1	1		ter in			
Acl 160 to	7.2592	. 5	(A ^V) °C							
Bc tc ℃	1684.1	5			 	1				
Cc — —	249.6	5	р.			l				
Cryos. A° consts. B°			c _p vap.300K 400	0.39649 0.50136	2 2					
t _e °C	150,77	5	c _v vap.	0.50136	١	I				
T _R = 0.75			L_ <u>*</u>	L	<u></u>	+ -			<u> </u>	
		2 .	DI 2 T-4 4	Colo ()			ams/100 gra		t	
	ES: 1-DOW	_	PI 3-Lit. 4-	Caic, irom de	ι. αa	1 (A)	-Caic. by for	IIIUIA		
SOURCE:			PI							
PURIFICAT			PI							
LITERATU	RE REFERE	NCES	5:							
1										
1										
L										

No. 56 STRUCTURAL FORMULA NAME 3-Ethyl-2-methylhexane C₂H₅ сн₃сн сн (сн₂)₂(сн₃) ĊН₃ Ref. Molecular Molecular Mole C9H20 % Pur Weight 128, 250 Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 1.929 B. P. °C h 0.050 ž BP 760 mm 138.0 2 ſ١ 0.0361 5 to 100 75.5 g' °K 30 47.98 30 mm 4 0.6870 4 10 27. 2 h' ∆Hm cal/g 1 5 -6. 300 to 0.0294 4 AHv cal/g Pressure 600 °K 0.0015 4 n 25°C 81.7 mm 25°C 8.744 -0.0₆52 o 4 30 mm 77.57 67.1 5 2 t_e 1140. 5 BP 0.1222 700 to m' 4 Density g/ml 20°C 5 65.40 te (d, e) 0.0013 11000 °K n* 4 5 0.731 2 65.24 ٥' -0.0₆45 4 ď4 25 0.727 AHV/T 5 19.65 30 0.723 4 Surface tension 83.14 to 50 dynes/cm. 20°C a 0.74699 4 22.80 2 155 °C 0.1163 4 ь -0.0₃797 4 30 21.87 20.94 2 ď٠ 10 to 86,20 5 ı 40 2 Ref. Index 0.1799 e' 50 ٠c n_D 20°C 1.4120 [P] Parachor d_c g/ml 25 1.4097 2 20°C ml/g 30 1.4059 4 t_C 30 •C 306.7 5 40 "C" 0.7479 5 Pc mm 16524. 4 Sugd. 5 385.2 MR (Obs.) 43.6 2 PV/RT Exp. L. l. %/wt. 43,762 MR (Calc.) 25°C 1.0000 5 u. (nD-d/2)1.046 2 30 mm 1.0000 5 Dispersion 96.6 2 Dielectric BP 0.9703 Flash Point °C 0.9594 A 45 to 6. 872 2 Fire Point 1381.0 M Spec. C 208.0 2 AHc kcal/m Ultra V. ΔHf A* 45 to 1.31689 5 X-Ray Dif. B*| ΔFf 165 °C 1288.5 Infrared ĸ Viscosity Solubility in centistokes Acetone t_k Carbon tet. •c Benzene ۸i 25 to 6.55701 Ether B١ 1224.4 <u>50</u> ℃ n-Heptane B C' Āv 193.1 to Ethanol °C Water 25 to 1.05788 Water in B'* 50 °C (BV) 1150.8 to Ac | 160 to 7,2707 5 5 (A^V) °C Bc tc °C 1691.5 liq. Сp °K Cc 247.5 5 Cryos. Aº c_p vap.300°K 0.39649 2 consts. B° 400 0.50136 2 c_v vap. te °C 153,99 5 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME	4-Ethy	l-2-met	hylhexane			STRU	JCTURAL	FORMUL	
			· · · · · · · · · · · · · · · · · · ·			CH,	сн-сн ₂ сн	- CH ₂ CH	2
Mole % Pur.	Ref.	Molecu Formu		Molecular Weight 128.2	50	-	Ċн ₃ сс		
		Re	f.		Ref.				Ref.
F. P. ℃			dt/dP			f	to		
F.P. 100%			°C/mm			gl	<u>°K</u>		1
B.P. °C			25°C BP	1.60471	2	h			
760 mm 100	133.8	2 2	te	0.03630	5	f'	to		
30	44.31		30 mm	0.6820	4	g'	<u>*K</u>		[
10	24.0	2	ΔHm cal/g	1	+-	h'			
1	-9.24	4	ΔHv cal/g		 	m	300 to	0.0294	4
Pressure mm 25°C	10.71		25°C	80.1	2	n	[600 • K	0.0015	
t _e	10.71 1127.	9 4	30 mm	76.35	5	0		-0.0 ₆ 52	4
Density		-+-	BP	66.4 64.805	2 5	m'	700 to	0.1222	4
g/ml 20°C	0.72	3 2	te te (d, e)	64.685	5	n' l	1000 •K	0.0013	4
d 25 4 30	0.71		AHV/Te	19.675	5	8.		-0.0 ₆ 45	4
4 30	0.71		d 45 to	81,27	4	Surfac	e tension		
a b	0.73	898 4 796 4	e 150 °C	0.1111	4	dynes	/cm. 20°C	21.77	2
	-0.03	170 4	- 10 to	84.96	5	•	30 4 0	20.85 19.93	2
Ref. Index	1.40	68 2	e' 45 °C	0.1945	5	Parac		17.75	-
25	1.40	46 2	d _c g/ml			1 4140	20°C		
30	1.40	15 4	v _c ml/g t _c °C	299.9	5		30		
"C"	0.74	86 4	P _c mm	16102.	5		40 Sugd	385.2	5
MR (Obs.)	43.7	2	PV/RT	-	-	Fvn	L. 1. %/wt.		1
MR (Calc.) (nD-d/2)	43.76 1.04		25°C	1.0000	5	Exp.	u.		
	1.01		30 mm	1.0000	5	Dispe	rsion	96.9	2
Dielectric A 40 to	(05	4 -	BP t _e	0.9694	5		Point °C		
B 160 °C	6.85	4 2	t c	0.7571	ĺ	Fire I	Point		
c '	209.0	2	AHc kcal/m			M. Sp			
A* 40 to	1.30	397 5	ΔHf			Ultra X-Ray			
B* 160 °C	1270.45	5	ΔFf			Infrar			
к — — — c	-	1	Viscosity centistokes			Solubi	lity in +		
t _k T to		ļ	η °c			Acet			
t _x °C		1	'			Benz	on tet.		
A' 25 to	6.40		1			Ethe	r		
B' 1_45 °C	1146.31 188.1	5 5	B. to			n-He Etha	ptane		
A1* 25 to	0.92		AV. C			Wate			
B'* 45 ℃	1076.64		(B ^v) to		1	Wate	r in		
Ac 160 to	7, 25	32 5	(A ^V) °C						
Bc tc °C	1669.5	5	c _p liq. °K	T	T	1			
Cc = -	248.2	5		•					
Cryos. A° consts. B°		1	c _p vap.300°K	0.39649	2				
	140 22	-+-	- 400 c vap.	0.50136	2	ļļ			1
t _e °C	149.23	5		<u> </u>	<u></u>	L			<u> </u>
$T_{\mathbf{R}} = 0.75$							ns/100 gra		t
REFERENC	ES: 1-D			Calc. from de	t. da	ta 5-C	alc. by for	mula	
SOURCE:			PI						
PURIFICAT	ION:	A	PI						
LITERATU	RE REF	ERENCE	CS:						

								No. 58	
NAME	3-Eth	yl-3-met	hylhexane			STR	UCTURAL 1		
							CH ₃		
l					$\overline{}$	CH	₁₃ сн ₂ с - (с	CH ₂) ₂ CH ₃	
Mole % Pur.	Ref.			Molecular Weight 128.2	250		Ċ ₂ H ₅		
70 Fur.		Formul Ref		weight 120.2	Ref				Ref
F.P. °C	T	IXE1.	dt/dP	T		١,	1		-
F.P. 100%			*C/mm	}		f g	to eK		1
B. P. *C			25°C BP	1.991 0.050	5 2	h			
760 mm 100	140.6 77.0	2 2	t _e	0.0369	5	f	to		
30	48.68		30 mm	0.7007	4	g'	<u>*</u> K_		ł
10 1	27.0 -6.75	2 5	ΔHm cal/g	Ť		h'			
Pressure	-0.73		ΔHv cal/g			m (300 to	0.0294	
mm 25°C	8.62		25°C 30 mm	80. 2 76. 37	5	n l	6 0 0 °K	0.0015 -0.0 ₆ 52	4
t _e	1143.	5	BP	66.6	2	m'	700 to		-
Density g/ml 20°C	0.74	1 2	te te (d, e)	65.02 64.86	5 5	n'	1000 °K	0.1222 0. 00 13	4
dt 25	0.74		Le (d, e)		1 1	0'		-0.0 ₆ 45	
⁴ 4 30	0.73	3 4	ΔHv/T _e	19.39 81.55	5	Surfa	ce tension		
a b	0.75		e 155 ℃	0.1063	4	dyne	s/cm. 20°C	23. 22 22. 30	2 2
Ref. Index		171 1	a' 10 to		5 5	•	30 40	21.38	2
n _D 20°C	1.41			0.1011	H	Para	chor [P]		
25 30	1.41		v ml/g	į			20°C 30		
"C"	0,74) 'c '	312.5	5		40		ŀ
MR (Obs.)	+	2	P _c mm	16423.	5			385.2	5
MR (Calc.	43.76	2 5	PV/RT 25°C	1,0000	5	Exp.	L.1.%/wt.		}
(nD-d/2)	1.04	4 2	30 mm	1.0000	5	Disp	u. ersion	95.9	2
Dielectric	4		BP	0.9654 0.9550	4	Flas	h Point °C		
A 45 to B 165 C		3 2 2	te tc	0.7550		Fire	Point		
c	212.0	2	ΔHc kcal/m	<u> </u>		M Sp Ultra			İ
A* 45 to			ΔHf ΔFf				y Dif.		
B* _170 °C	1309.87	5	Viscosity	 	\vdash	Infra			<u> </u>
¢	_	- 1	centistokes	1			ility in + tone		l
tk to		1	η ∘c			Car	bon tet.		
A' 25 to	6,57	634 5				Ben Ethe	zene		
B' _ 50 °C	1258, 52	5	B ^V to	 	-	n-H	eptane		İ
C1	198.1	5	B to A °C			Eth: Wat	nol er		l
A'* 25 to B'* 50 °C	1.06		(B ^V) to	1			er in		
Ac 165 to	7.26	97 5	(A ^V) °C						"
Bc tc_°C	1728.4 253.4	5	c _p liq. °K	1	$\vdash \vdash$				
Cryos. A°			-11	0.30(40	,	1			
consts. B°			c _p vap.300°K 400	0.39649 0.50136	2 2	İ			l
t _e °C	157.00	5	c _v vap.						
$T_R = 0.7$	5 T _c					+ gra	ms/100 gran	ns solvent	
REFEREN	CES: 1-D			Calc, from de	t. dat				
SOURCE:		AI	PI						
PURIFICA'		AI							
LITERATU	RE REF	ERENCES	S:						

					~				No. 59	<u> </u>	
NAME	4-E	thyl-3	-met	hylhexane		STRUCTURAL FORMULA					
ı İ						- 1	C ₂ H ₅				
							C	н ₃ сн ₂ сн-сн	- CH2CH	3	
Mole % Pur.	Re		lecul mul		Molecular Weight 128.2	50		сн	_		
, Fui.		1 FO.	_		Weight 120.1					Ref.	
	1		Ref.		 	Ref.			r	Kel.	
F.P. °C F.P. 100%	+		-	dt/dP °C/mm			f ~	to K			
B.P. °C	†		-	25°C	2, 101	5	g h	<u> </u>			
760 mm	140.4	ŀ	2	BP	0.05 0.0362	2 5	f'	1		 	
100 30	77.0 50.)	2 2	t _e	0. 69251		g'	to K			
10	29.		2	30 mm	0. 69251		h'			1	
1	-5.7	:	5	ΔHm cal/g		-	m	300 to	0.0294	4	
Pressure			ا ۔ ا	ΔHv cal/g 25°C	81.9	2	n	600 °K	0.0015	4	
mm 25°C	1147.	0593	5	30 mm	77.76	5	٥		-0.0652	4	
Density	+		H	BP	67.8 66.15	2 5	m'	700 to	0.1222	4	
g/ml 20°C	0.	742	2	te te (d, e)	66.01	5	n' o'	1000 °K	0.0013	4	
d ₄ 25		738	2	AHv/Te	19.73	5			-0.0 ₆ 45		
	+	734	4	d 50 to	83, 22	4		face tension	22.25		
a b	-0.		4 4	e_ 155 °C	0.1098	4	dyn	es/cm. 20°C 30	23.27 22.34	2 2	
Ref. Index	†			d' 10 to	86. 09 0. 1676	5	ا	40	21.40	2	
n _D 20°C	1	116	2	d _c g/ml	+ 3.23.0	<u> </u>	Par	achor [P]			
25 30		114	2	v mi/g				20°C		İ	
"C"		112	-	t _c °C	312.3	5		30 40			
	43.	459	2	P _c mm	17008.	5		Sugd.	385.2	5	
MR (Obs.) MR (Calc.			5	PV/RT			Exp	. L.1.%/wt.			
(nD-d/2)	1.0	145	2	25°C 30 mm	1.0000 1. 0 000	5	Die	u. persion	95.5	2	
Dielectric				RP	0.9694	5		sh Point °C	75.5	1-	
A 45 to	6.1	385	2	t _e	0.9591	5		e Point			
B (165 ℃	1399. 209. ()	2 2	te c AHc kcal/m		-	M.	Spec.			
A* 45 to	+	32622	5	ΔHf	i			ra V.	}	ļ	
B*[165 °C			5	ΔFf				lay Dif. ared			
к — — —				Viscosity			Solv	ability in +			
t _k – to	-			centistokes り °C			Ac	etone			
tva i °C				'		1		rbon tet.			
A' 25 to		3329	5					her			
B' 1_50°C	- 1271.1 - 197.1		5	B _v to	 			Heptane hanol	}		
A'* 25 to		2712	5	ĂV I °C				ter			
B'* 50 ℃			5	(B ^V) to	1		Wa	ter in		<u> </u>	
Ac 165 to		865	5	(A ^V) °C							
Bc∟t _c °C Cc	1716.2 249.3		5	c _p liq. °K							
Cryos, A°			-	8	0.20112	_					
consts. B	1			c _p vap300°K 400	0.39649		l				
te °C F	156.	72	5	c vap.		1			1		
$T_{\mathbf{R}} = 0.7$	5 T _C				<u> </u>		+ 91	ams/100 gra	ms solver	t t	
REFERENC		Dow	2-A	PI 3-Lit. 4-	Calc, from de	t. da		-Calc. by for			
SOURCE:			-	PI							
PURIFICAT	ION:			PI							
LITERATU		FERE									
	KE.	. LAE	CES	••							
t .											

							No. 60				
NAME	2, 2, 3-Tri	meth	ylhexane			STRUCTURAL FORMULA					
						СН ₃ - С СН - (СН ₂) ₂ СН ₃					
	TT		· ·		\neg		(CH ₂) ₂ CH	3			
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 128.2	50	СH ₃ CH ₃					
		Ref.			Ref			Ref.			
F,P. °C			dt/dP			f to					
F.P. 1007	5		*C/mm	, 551	5	g <u> </u>					
B. P. °C 760 mm	133,60	2	25°C BP	1.551 0.0490	2	h		_			
100	71.0	2	t _e	0.0369	5	f' to		ŀ			
30 10	43. 23.	2 2	30 mm	0.6863	4	b'		ŀ			
1	-11.	5	ΔHm cal/g			m 300 to	0,0294	4			
Pressure mm 25°C	11 272	5	ΔHv cal/g 25°C	78.1	2	n 600°K	0.0015	4			
t _e	11.373 1114.	5	30 mm BP	75.47 64.8	5 2	<u> </u>	-0.0 ₆ 52	4			
Density		<u> </u>	t_	63. 05	5	m' 700 to	0.1222 0.0013				
g/ml 20°0	0.7292 0.7254	2 2		63.02	5	n' 1000°K	-0.0645				
d ₄ 25	0.7216	4	ΔHv/T _e	19.17	5	Surface tension		<u> </u>			
•	0.7444	4	d 45 to	80.62 0.1184	4 4	dynes/cm. 20°C	21.86	2			
Ref. Index	-0.03756	4	d' 10 to	81.66	5	30 40	20.96 20.06	2			
n _D 20°0	1.4105	2		0.1423	,	Parachor [P]					
25 30	1.4082	2	d g/ml vc ml/g			20°C 30					
"C"	0.7494	4	T _C T	300.9	5	40					
MR (Obs.		2	P _c mm	16053.	5		385.2	5			
MR (Calc. (nD-d/2)	1,0459	5 2	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		ŀ			
Dielectric		+-	30 mm BP	1.0000 0.9601	5	Dispersion	97.3	2			
A 40 t		2	t_	0.9490	5	Flash Point °C Fire Point		ŀ			
B L160 •	211.0	2 2	t _c	ļ		M Spec.					
A* 40 to	+	5	ΔHc kcal/m ΔHf			Ultra V.					
B* 150°		5	ΔFf			X-Ray Dif. Infrared					
K C		ŀ	Viscosity centistokes	·		Solubility in +					
t _k to		ľ	η °c			Acetone Carbon tet.					
t 'C 'C A' 25 te		Ļ				Benzene					
B' _ 45 °		5				Ether n-Heptane					
C'	205.1	5	B ^V to A ^V °C			Ethanol Water					
A'* 25 to B'* 45 °C		5	(BV)	-		Water in					
Acl 160 to	7 2477	4	(A ^V) ₁ °C								
Bc tc_°	C 1678.6	1	c _p liq. °K								
Cryos. A	250.9	+-	c _p vap.300°K	0.39649	2						
consts. B			400	0.50136	2						
t _e °C	148.67	5	c _v vap.		L						
$T_R = 0.7$						grams/100 gran	ns solvent	t			
	CES: 1-Dow			Calc, from det	t. da	ta 5-Calc. by for	mula				
SOURCE:	TION	Al									
PURIFICA	TION: TRE REFERE	Al									
LILERAIC	ne nefekt	NUES);								

										No. 61				
NAME	2, 2, 4-Trimethylhexane								STRUCTURAL FORMULA					
								CH ₃						
<u> </u>								сн ₃ с сн ₂ сн-сн ₂ сн ₃						
Mole		Ref.	Mo	ecul	ar C ₉ H ₂₀	Molecular			сн ₃ сн					
% Pur.			For	mula	C9 ¹¹ 20	Weight 128.25	0			-3				
				Ref.			Ref.				Ref.			
F. P. ℃		-120.0		2	dt/dP			f	to					
F.P. 1009	•				°C/mm			g	<u>^K</u> _					
B.P. °C	1				25°C BP	1.17286	2	h						
760 mm	1	126.54 65.	•	2 2	t _e	0.0370	5	f'	to					
30		38.		2	30 mm	0.6767	4	g'	' <u>*</u> K					
10	ı	17.		2	ΔHm cal/g		-	h'						
1	+	-16.		5	ΔHv cal/g	1	_	m	300 to	0.0294	4			
Pressure mm 25°C	ŀ	15, 53	. 0	5	25°C	75.6	2	n o	_6 <u>0</u> 0_ <u>•</u> K	0.0015				
t _e	1	083.	•	5	30 mm BP	73.75	5 2			-0.0652	4			
Density	Ť				t _e	63.4	5	m'		0.1222	4			
g/ml 20°C	:	0.71		2	t _e (d, e)	61.76	5	n' o'	1000 •K	0.0013 -0.0 ₆ 45	4 4			
d ₄ 25 30		0.71 0.70		2 4	ΔHv/T _e	19.14	5		L	6.3	\vdash			
a	+	0, 73		4	d 40 to	78.13	4		face tension	20 51	2			
b		-0.03		4	e 140 °C 10 to	0.1164 79.26	4 5	g yn	es/cm. 20°C 30	20.51 19.67	2			
Ref. Index	:				e' 40 °C	0.1463	5		40	18.83	2			
n _D 20°C	;	1.40		2	d _c g/ml			Par	achor [P]					
25 30		1.40 1.39		2 4	vc ml/g				20°C 30					
"C"	+	0.75		4	vc ml/g tc °C	288.7	5		40					
MR (Obs.)	+	43.76		2	P _c mm	15468.	5		Sugd.	385.2	5			
MR (Calc.		43.76		5	PV/RT	1 0000	_	Exp	. L.1.%/wt.					
(nD-d/2)		1.04	55	2	25°C 30 mm	1.0000	5	Dis	u. persion	98.8	2			
Dielectric					RP	0.9601	4	<u> </u>	sh Point C	70.0	÷			
A 35 to		6, 83	91	2	t _e	0.9496	5		e Point					
B 1_150 °C	- '	344.0 213.0		2	tc AHc kcal/m	 		M.	Spec.					
A* 35 to	+	1, 31	004	5	ΔHf				ra V.					
B* 150 °C		256.4	001	5	ΔFf				Ray Dif. ared					
к	1				Viscosity			í 	ability in +					
	-				centistokes 7 °C			Ac	etone					
t _x °C	:				'				rbon tet. enzen e					
A' 25 to		6, 63	114	5					her					
B' 40°C	- 1	241.8		5 5	B _v to				Heptane					
A'* 25 to	+	1, 12	724	5	B to				hanol ater					
B'* 40 °C		1, 12	1 34	5	(B ^V) to	-			ter in					
Ac 150 to	+	7. 24	30	5	(A ^V) °C									
Bc tc °C		651.4	-	5			-	1		ł				
Cc	1	252.1		5	P -						1			
Cryos, A° consts. B°		0.06	•	2	cp vap.300°K	0.39649				İ				
	-+	140 (5		\vdash	c _v vap.	0.50136	2							
t _e °C T _R = 0.7		140.65		5	V	L	L	L	4100	<u> </u>	<u> </u>			
					DT 2 T::	0.1			rams/100 gra		t			
	υE	5: 1-E	ow		PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula				
SOURCE:				AP										
PURIFICA				AP										
LITERATU	RE	REF	ERE	VC ES	5:									
1														
L														

							No. 62		
NAME	2,2,5-Tr	imeth	ylhexane		- 1	STRUCTURAL FORMULA			
Γ						CH ₃			
						СН ₃ С - (СН ₂)2	CH CH ₃		
Mole	Ref. Mo	lecul		Molecular			Ċн ₃		
% Pur.	Fo	rmul	a 09 ¹¹ 20	Weight 128.2	50	J	3		
		Ref.			Ref			Ref.	
F, P. *C	-105.780	2	dt/dP	1		f to		i	
F.P. 100%			*C/mm		ا ۔ ا	g <u>K</u>		ı	
B. P. *C		1 .	25°C BP	1.109 0.04838	5 2	ъ .			
760 mm 100	124. 084 63. 092	2 2	t	0.0372	5	f' to			
30	36.37	2	30 mm	0.6676	4	g' ' <u>*</u> K_		İ	
10 1	16.17	5	ΔHm cal/g			h'			
	-18.2	13	ΔHv cal/g			m 300 to	0.0294		
Pressure mm 25°C	16,593	5	25°C	74.86	2	n 600 K	0.0015	4	
t _e	1066.	5	30 mm BP	74.14 62.9	5 2		-0.0 ₆ 52	-	
Density	1		t.	61.44	5	m' 700 to	0.1222	4	
g/ml 20°C	0.70721	2	t (d, e)	61.24	5	n' 1000 K	0.0013		
dt 25 4 30	0.70322	2 4	AHv/Te	19.21	5		-0.0 ₆ 45		
a	0.72315	4	d 40 to	78.80	4	Surface tension	30.04		
b	-0.03793	4		0.1281	4	dynes/cm. 20°C	20.04 19.60	2 2	
Ref. Index	1		d' 10 to e' 40 °C	76.45 0.0634	5 5	40	18. 29	2	
n _D 20°C	1.39972	2	d _c g/ml	0.0051	۲	Parachor [P]			
25 30	1.39728	2 4	vc ml/g tc °C			20°C 30			
"C"	0.7538	4	tc °C	282.1	5	40		1	
	 	-	P _c mm	15064.	5	Sugd,	385.2	5	
MR (Obs.) MR (Calc.)	43.935 43.762	5	PV/RT			Exp. L.1.%/wt.			
(nD-d/2)	1.04612	2	25°C 30 mm	1.0000 1.0000	5 5	u.	99.0	2	
Dielectric			BP	0.9458	4	Dispersion Flash Point °C	99.0	<u> </u>	
A 35 to	6.83531	2	t _e	0.9347	5	Fire Point			
B 1_145 ℃		2	t _c		Ш	M Spec.	t		
<u>C</u>	210.737	2	ΔHc kcal/m ΔHf	l		Ultra V.			
A* 35 to B* 150 °C	1.33661 1244.71	5	ΔFf			X-Ray Dif.		l	
K 130 0	1244.71		Viscosity		$\vdash \lnot$	Infrared Solubility in +		-	
¢ .—=	ļ		centistokes	1		Solubility in + Acetone			
t _x to t _x *C	İ		η •c			Carbon tet.			
A' 25 to	7,43548	5				Benzene Ether			
B' 40 °C	1637.27	5			\sqcup	n-Heptane			
	238.4	5	B ^V to	l		Ethanol		l	
A!* 25 to	1.88305	5	AV C			Water Water in			
	1534.28	5	(B ^V) to	}					
Ac 145 to	7, 2339	5	(A ^V) °C						
Cc c	248.5	5	c _p liq. ∘K	1			1		
Cryos. A*	0.0265	2	cp vap.300°K	0.39649	2		1	i	
consts. B°			400	0.50136	2				
t _e °C	137.04	5	c _v vap.	1			1		
$T_{R} = 0.75$						+ grams/100 gra	ms solveni		
REFERENC	ES: 1-Dow	2-AI		alc. from det	dat				
SOURCE:		AI	PI						
PURIFICAT	ION:	AI	PI						
LITERATUE	RE REFERE	NCES	3:						

	-							No. 63		
NAME _	2, 3, 3-Tri	meth	ylhexane			ST	RUCTURAL	FORMUL	A	
							CH ₃			
Mole % Pur.	Ref. Mol	ecul	ar C ₉ H ₂₀	Molecular Veight 128.2	50	сн ₃ сн-с (сн ₂) ₂ сн ₃ сн ₃ сн ₃				
<i>N</i> 1 U.1.	1 1 1 1 1 1 1	Ref.	, <u></u>		Ref.	Γ			Ref.	
F. P. °C	-116,800	2	dt/dP			f	to			
F.P. 100%			°C/mm 25°C	1.749	5	g	<u>•</u> K_			
B.P. °C 760 mm	137.68	2	BP	0.0505	4	h				
100 30	74. 46.	2 2	t _e 30 mm	0.0371 0.6982	5	f' g'	to			
10	25.	2	ΔHm cal/g	0.0782		h'				
Pressure	-9.	5	ΔHv cal/g		_	m	300 to	0.0294	4	
mm 25°C	10.01	5	25°C 30 mm	78.7 75.38	2	n o	1_6 <u>0</u> 0_•K	0.0015 -0.0 ₆ 52	4	
t _e	1130.	5	BP	65.2	2	m'	700 to	0.1222	4	
Density g/ml 20°C	0.738	2	te te (d, e)	63.43 63.43	5	n' o'	1000 •K	0.0013	4	
dt 25 4 30	0.734 0.730	2	AHv/Te	19.06	5	o.		-0.0 ₆ 45	4	
a 30	0.754	4	d 45 to	80.49	5		face tension es/cm. 20°C	22.41	2	
ъ	-0.0 ₃ 797	4	d 155 ℃ to	0.1110 82.65	5	8,	30	21.49	2	
Ref. Index	1.4141	2	e' 45 °C	0.1581	5	Par	achor [P]	20.59	2	
25 30	1.4119 1.4092	2	d g/ml vc ml/g			1 ***	20°C			
"C"	0.7467	4	tc°C	307.8	5		30 4 0			
MR (Obs.)	43.4	2	P _c mm	16059.	5			385.2	5	
MR (Calc.) (nD-d/2)		. 5 . 2	PV/RT 25°C	1.0000	5	Exp	u. L.1.%/wt.			
Dielectric	1.045	-	30 mm BP	1.0000 0.9629	5		persion	96.3	2	
A 45 to	6.8474	2	t_	0.9517	5		sh Point °C e Point			
B (165 °C	1391.0 213.0	2	tc AHc kcal/m			<u> </u>	Spec.			
A* 45 to	1.29980	5	ΔHf				ra V. Ray Dif.			
B*[165 °C	1298.6	5	ΔFf Viscosity		<u> </u>		ared			
c			centistokes				ability in +			
t _k to to			η °C			Ca	rbon tet.			
A' 25 to	6,55736	5					nzene her			
B' 45 °C	1244.81 199.0	5 5	Bv to				Heptane hanol			
A!# 25 to	1.05181	5				Wa	ater			
B'* 45 °C Acl 165 to	7 2550	5	(B ^V) to				ater in		├	
Bc tc °C	7. 2550 1713. 7	5 5	(A ^V) °C c_liq, °K							
Cc — —	254.3	5	Р.							
Cryos. A° consts. B°	0.0447	2	c vap.300°K 400	0.39649 0.50136						
t _e °C	153.61	5	c _y vap.	L		L		L	<u>Ļ</u>	
T _R = 0.77		2-4	PI 3-Lit. 4-0	Calc from de	t d=		ams/100 gra		t	
SOURCE:	E.S. 1-DOW		PI	Carc. Hom de	t. ua		-Caic. by 101	IIIuIa .		
PURIFICAT	ION:		PI							
LITERATUR	E REFERE	CES	3:							
i										

							No. 64		
NAME	2, 3, 4-Tri	meth	ylhexane			STRUCTURAL FORMULA			
						CH ₃			
34-3	2 ()			V-11		сн ₃ сн сн сн			
Mole % Pur.		lecul rmul		Molecular Weight 128.2	50	Ċн ₃ Ċн	¹ 3		
		Ref.			Ref			Ref.	
F.P. °C			dt/dP			f to			
F.P. 100%			*C/mm 25*C	1.914	5	g <u>K</u>		ĺ	
B.P. °C 760 mm	139.0	2	BP	0.050	2	h	ļ	<u> </u>	
100	76.	2	t _e	0, 0365	5	f' to g'K		ĺ	
30 10	48. 27.	2	30 mm	0, 6952	4	h'	1		
1	-7.	5	ΔHm cal/g ΔHv cal/g		\vdash	m 300 to	0,0294	4	
Pressure mm 25°C	9, 000	5	25°C	80.0	2	n 600°K	0.0015		
t _e	1144.	5	30 mm BP	76.57 66.5	5 2	0	-0.0652		
Density			l t_	64.75	5	m' 700 to n' , 1000 °K	0.1222 0.0013	4	
g/ml 20°C	0.7392 0.7354	2 2	le (d, e)	64.70	5	0'	-0.0645		
d ₄ 25	0.7316	4	ΔHv/T _e	19.38	5	Surface tension		<u> </u>	
a b	0.7544	4	d 50 to	81.85 0.1104	4 4	dynes/cm. 20°C	22.80	2	
Ref. Index	-0.0376	-	d' 10 to	83.76 0.1505	5	30 40	21.87	2	
n _D 20°C	1.4144	2	d _c g/ml	0,1303	احا	Parachor [P]			
25 30	1.4120	2 4	I V mi/g		_	20°C 30		ĺ	
"C"	0.7441	4	t _c °C	310.8	5	40			
MR (Obs.)	43.39	2	P _c mm	16713.	5	<u> </u>	385.2	5	
MR (Calc.) (nD-d/2)	43.762 1.0448	5 2	25°C	1,0000	5	Exp. L.1.%/wt. u.		ĺ	
Dielectric	1.0440	-	30 mm BP	1.0000 0.9689	5 4	Dispersion	96.4	2	
A 45 to	6.867	2	te	0.9594	5	Flash Point °C Fire Point		İ	
B 1_16 <u>5</u> °C C	1395.	2	t _c		\vdash	M Spec.			
A* 45 to	1,3076	5	ΔHc kcal/m ΔHf			Ultra V.		ĺ	
B* 165 °C	1300.	5	ΔFf			X-Ray Dif. Infrared		İ	
K ———			Viscosity centistokes			Solubility in +			
t _k to	Ì		7 °C			Acetone Carbon tet.			
A' 25 to	6, 6857	5				Benzene			
B' _ 50 °C	1302.	5				Ether n-Heptane			
C'	202.3	5	B ^V to			Ethanol Water			
A'* 25 to B'* 50 °C	1.1738 1223.	5	(B ^V) - to			Water in			
Ac 165 to	7,2717	5	(A ^V) °C					l	
Bc tc °C	1715.5 251.9	5	c _p liq. °K		\vdash			ĺ	
Cryos, A°	231.7	,	c _p vap.300°K	0, 39649	2				
consts. B°			400	0.50136	2				
t _e °C	155.27	5	c _w vap.						
$T_{\mathbf{R}} = 0.76$						grams/100 gran	ms solvent	t	
	ES: 1-Dow			alc. from det	t. dat	ta 5-Calc, by for	mula		
SOURCE:	ION.	AF							
PURIFICAT		AF							
TILEKATUR	RE REFEREI	vCES	•						

NAME	2,3,	5-Tri	meth	ylhexane		STRUCTURAL FORMULA CH ₃			
Mole % Pur.	Ref.		ecula mula		Molecular Veight 128.25	50	сн ₃ сн сн сн сн ₃	₂ сн сн ₃	
			Ref.			Ref.			Ref.
F.P. °C F.P. 1009	-127.8		2	dt/dP °C/mm			f to		
B. P. °C 760 mm 100 30 10	131.3- 69. 42. 21.	4	2 2 2 2 5	25°C BP te 30 mm AHm cal/g	1.442 0.049 0.0369 0.6814	5 2 5 4	h f' to g' °K		
Pressure mm 25°C t _e	12.3°	76	5 5	ΔHv cal/g 25°C 30 mm BP	77.19 75.27 64.9	2 5 2	m 300 to n 600 °K o 700 to	0.0294 0.0015 -0.0 ₆ 52 0.1222	4 4 4
Density g/ml 20°0 dt 25 4 30	0.7 0.7	179 139	2 2 4	t _e (d, e) ΔHv/T _e d 40 to	63.35 63.21 19.38	5 5 5	n' 1000 °K		
a b Ref. Index	-0.0	3788 3796	4	e 145 °C 10 to e 40 °C	80.13 0.1159 80.03	4 5 5	dynes/cm. 20°C 30 40	21.27 20.37 19.49	2 2 2
ⁿ D 20°0 25 30		037	2 2 4	d g/ml vc ml/g tc °C	0.1137	5	Parachor [P] 20°C 30		
"C"	0.74	194	4	P _c mm	15745.	5	40 Sugd	385.2	5
MR (Obs. MR (Calc. (nD-d/2)	1.0	62	2 5 2	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9586	5 5 5	Exp. L.1.%/wt. u. Dispersion Flash Point °C	97.8	2
A 40 to B 155 °C C	1	505	2 2 2	te t _c ΔHc kcal/m	0.9475	5	Fire Point M. Spec. Ultra V.		
A* 40 to B* 155 °C K		2039 0	5 5	ΔHf ΔFf Viscosity			X-Ray Dif. Infrared		
c t _k t _c t _x °C	;	(030	-	centistokes 7 °C			Solubility in Acetone Carbon tet. Benzene		
A' 25 to B' 45 °C C' A'* 25 to	216.2	6029 0 3569	5 5 5	B ^V to A ^V C			Ether n-Heptane Ethanol Water		
B'* 45 °C Acl 155 to Bc tc °C	7.2		5 5 5	(B ^V) to (A ^V) °C			Water in		
Cryos. A'	250.3		5	c _p liq. °K c _p vap.300°K 400	0.39649 0.50136				
t _e °C	145.9	6	5	c _v vap.	0.30136		+		
T _R = 0.)o	2 A 1	PI 3-Lit. 4-	Cala from de	• 4-	grams/100 gra		IT.
SOURCE:	CE3: 1-1		AP		Calc. Irom de	ua	ia 3-Caic, by 10		
PURIFICA	TION		AP						
LITERATI		ERE							

							No. 66	
NAME	2, 4, 4-Tr	imetl	hylhexane			STRUCTURAL	FORMULA	L
						ÇI	I,	
<u> </u>			————		\dashv	сизси сизс		
Mole		lecul		Molecular		ċн ₃ ċн		
% Pur.	Fo	rmul		Weight 128.	_			
		Ref.		,	Ref			Ref.
F.P. *C	-113.380	2	dt/dP	1	Į.	f to		
F. P. 1007	`	-	*C/mm 25*C	1, 356	5	8 <u>*K</u>		
B. P. °C 760 mm	130,648	2	BP	0.0496	2	h +		Ь—
100	68.069	2	t _e	0.0373	5	f' to	1	1
30 10	40.62 19.86	2 2	30 mm	0.6860	4	g' 'K_	1	1
10	-14.2	5	ΔHm cal/g	İ		h'		<u> </u>
Pressure		+	ΔHv cal/g			m 300 to n 600 °K	0.0294 0.0015	
mm 25°C		5	25°C 30 mm	76.1 74.15	5		-0.0652	
t _e	1096.	5	BP	63.9	2		ļ. — —	├
Density		١.	t _e	62.41	5	m' 700 to n' 1000 K	0.1222	4
g/ml 20°0	0.72381 0.72007	2 2	'e (", ",	62.26	5	0'	-0.0645	4
dt 25 4 30	0.71632	4	ΔHv/T _e	19.14	5	Surface tension		┢
	0.73875	5	d 40 to		4	Surface tension dynes/cm. 20°C	21.17	2
Ъ	-0.03744	5	d 145 to		4 5	30	20.33	2
Ref. Index		1.	e' 40 °C		5	40	19.49	2
ⁿ D 20°0	1.40745 1.40515	2 2	d _c g/ml			Parachor [P] 20°C		l
30	1.40300	4	V mi/g	205 (١.	30		1
"C"	0,7499	4	1. ℃	295.6	5	40		
MR (Obs.		2	P _c mm	15689.	5		385.2	5
MR (Calc.) 43.762	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.		ĺ
(nD-d/2)	1.04554	2	30 mm	1.0000	5	u. Dispersion	98.2	2
Dielectric			BP	0.9533	4	Flash Point *C		
A 40 t		2	t _e t _c	0.9421	5	Fire Point		l
B 1_155 *	2 1368.723 214.047	2	ΔHc kcal/m	 	\vdash	M Spec.		
A* 40 to		5	∆Hſ		į	Ultra V.		ļ
B* 155 °		5	ΔFf	<u> </u>	_	X-Ray Dif. Infrared		Ì
к — —		1	Viscosity		l	Solubility in +	<u> </u>	<u> </u>
\$,	1	centistokes °C		l	Acetone		1
t _X 0	5	1	'		l	Carbon tet. Bensene		l
A' 25 to	6.79316	5	1	1]	Ether		l
	C 1339.10	5	B ^V to	 	├	n-Heptane		i
C'	211.3	5	A ^V C	1	l	Ethanol Water		1
A'* 25 to B'* 40 °C	1.27805 C 1255.58	5 5	175v. — — -	-	1	Water in		L
Ac 155 to		5	1	1	1			
Bc t *	C 1684.3	5		+	\vdash	1		l
Ce	254.2	5	c _p liq. ∘K	1	j	1		
Cryos. A'		2	c _p vap.300°K 400	0.39649 0.50136	2			
t _e *C	145.03	5	c _v vap.					L
$T_{\mathbf{R}} = 0.$						+ grams/100 gram	ms solvent	
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		AI	PI					
PURIFICA	TION:	AJ	PI					
LITERATU	RE REFERE	NCES	S:					

NAME	3, 3, 4-Tr	imeth	nylhexane		STRUCTURAL FORMULA CH ₃			
Mole		lecul rmula		Molecular Veight 128.2	:50	сн ₃ сн ₂ с - сн сн ₂ сн ₃ сн ₃ сн ₃		
		Ref.	I		Ref.	. Re		
F.P. °C F.P. 100	-101.20	2	dt/dP °C/mm			f to g °K		
B. P. °C 760 mm 100 30	140.46 77. 48. 27.	2 2 2 5	25°C BP t _e 30 mm	1.983 0.049 0.0375 0.7011	5 2 5 4	h to g' °K h'		
Pressure mm 25°C		5 5	ΔHv cal/g 25°C 30 mm BP	78.8 76.23 65.5	2 5 2	m 300 to 0.0294 4 0.0015 4 0.0015 4 0.005 4		
Density g/ml 20°6 dt 25 4 30	0.7454 0.7414 0.7374	2 2 4	te (d, e) ΔHv/Te	63.85 63.67 19.07	5 5 4	n' 1000 °K 0.0013 4 0' -0.0645 4 Surface tension		
a b Ref. Inde:		4 4	d 50 to e 155 °C d' 10 to e' 50 °C	81.89 0.1167 81.54 0.1094	4 5 5	dynes/cm. 20°C 23.27 2 30 22.31 2 40 21.35 2		
ⁿ D 20°0 25 30	1.4154 1.4130	2 2 4	d g/ml vc ml/g tc °C	312.0	5	Parachor [P] 20°C 30 40		
"C"	0.7458	4	P _c mm	16275.	5	Sugd. 385.2 5		
MR (Obs. MR (Calc (nD-d/2)	1, 0451	5 2	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9514	5 5 4	Exp. L.1.%/wt. u. Dispersion 95.3 2		
A 45 to B 165 °C	6.8557	2 2 2	te tc ΔHc kcal/m	0.9391	5	Flash Point °C Fire Point M. Spec. Ultra V.		
A* 45 to B* 165 °C K		.5 .5	ΔHf ΔFf Viscosity			X-Ray Dif. Infrared Solubility in +		
c t _k			centistokes γ°C			Acetone Carbon tet. Benzene		
B' 50 °C	1462.49 217.7	5 5 5	B ^v			Ether n-Heptane Ethanol Water		
B'* 50 °C	1372.08	5	(B ^V) to			Water in		
Acl 165 to Bc t _c °C	7, 2624 1725, 1 253, 4	5 5 5	(A ^V) °C c _p liq. °K					
Cryos. A consts. B		2	c vap.300°K 400 c vap.	0.39649 0.50136	2 2			
$T_R = 0$.		1	<u>*</u>	L	L	† grams/100 grams solvent		
		2-A	PI 3-Lit. 4-	Calc, from de	t. da	ata 5-Calc. by formula		
SOURCE:			PI			· · · · · · · · · · · · · · · · · · ·		
PURIFICA	TION:	A	.PI					
	JRE REFERE	NCES	S:					
L								

										No. 68		
NAME		3, 3-D	ethy	lpent	ane			STR	STRUCTURAL FORMULA			
								C ₂ H ₅				
								СН	CH2C CH	I ₂ CH ₃		
Mole % Pur.		Ref.		lecul rmul		Molecular Weight 128.2	50		Ċ ₂ н ₅			
				Ref.			Ref				Ref.	
F.P. °C	Τ.	-33.11	0	2	dt/dP	I		f	to			
F.P. 1009					°C/mm	3 370	ا ۔ ا	g	<u>*</u> K_			
B. P. °C 760 mm	١,	46.16	.8	2	25°C BP	2.379 0.05153	5 2	h			ļ	
100		81.04	4	2	t _e	0.0371	5	f' g'	to to		İ	
30 10		52.40 30.71		2 2	30 mm	0.7164	4	h'	'^_ 		ŀ	
1	\perp	-4.20		5	ΔHm cal/g			m	300 to	0.0294	4	
Pressure mm 25°C		7.16	0	5	ΔHv cal/g 25°C	80.8	2	n	_6 <u>00</u> °K _	0.0015	4	
t _e	11	57.	0	5	30 mm BP	76.44 67.1	5 2	°	İ	-0.0 ₆ 52	4	
Density						65.46	5	m'	700 to	0.1222	4	
g/ml 20°0	1	0.75 0.75		2 2	te (d, e)	65.38	5	o'	T000 .K	0.0013 -0.0 ₆ 45	4	
dt 25 4 30	\perp	0.74		4	ΔHv/T _e	19.23	5	Surf	ace tension	-	-	
a	Т	0.76		4	d 50 to		4 4		s/cm. 20°C	23.75	2	
Ref. Index	+	-0.03	/16	4	d' 10 to	84.78	5	•	30 4 0	22.83 21.92	2	
n _D 20°0		1.42	051	2	e' j 50 °C	0.1591	5	Para	chor [P]			
25 30	ł	1.41 1.41		2 4	d g/ml vc ml/g				20°C 30			
"C"	+	0.74		4	, c	324.3	5	1	40			
MR (Obs.	+	43, 11		2	P _c mm	17130.	5			385.2	5	
MR (Calc. (nD-d/2))	43.76		5 2	PV/RT 25°C	1.0000	5	Exp.	L.1.%/wt. u.			
Dielectric	+	1.04	312	-	30 mm BP	1.0000 0.9643	5		ersion	94.3	2	
A 50 t		6.89	262	2	te	0.9526	5		h Point °C Point			
B 175°	2 14	51.24	5	2	tc		\Box	M S			-	
A* 50 to	+	1.33		5	ΔHc kcal/m ΔHf	1		Ultr	a V.			
B* 175 °C		53.96		5	ΔFf			Infra	ay Dif. ared			
K c					Viscosity centistokes		1 1	Solu	bility in +			
t _k to					7 00				tone bon tet.			
t _x ; •(753					Ber	nzene			
A' 25 to B' 55 °C	12	6.54 72.18		5				Eth n-H	er Ieptane			
C'		98.5		5	B ^V to				anol			
A'* 25 to B'* 55 °C		1.03 94.11	683	5		-			ter ter in			
Ac 175 to	, —	7.30	74	5	(A ^V) to	ł						
Bc tc_°	C 17	93.2		5	c _p liq. °K							
Cryos, A	_	0, 02	2.3	2	•	ŀ	,					
consts. B		0.02		Ľ	c _p vap.300°K 400	0.39649 0.50136	2 2					
t _e °C		63.42		5	c _w vap.							
$T_R = 0.7$								+ gra	ms/100 gran	ns solvent	t	
REFEREN	CES	: 1-D	ow	2-AI		Calc. from det	t. da	ta 5-	Calc. by form	nula		
SOURCE:				AI					· · · · · · · · · · · · · · · · · · ·			
PURIFICA				AI								
LITERATU	KE	KEF	EREI	NCES) :							

								No. 69		
NAME	3-Ethyl-2	, 2 - di	imethylpentane			STRUCTURAL FORMULA				
			······							
			·				СН ₃			
Mole	Ref. Mo	1.	,	Valarula m		CI	н ₃ с - сн - с	H ₂ CH ₃		
% Pur.	Kei. Mo	mula	ar C ₉ H ₂₀	Molecular Weight 128.2	50		ĊH ₃ Ċ ₂ H ₅			
- W 1 u1.	1 1 1 0	_		Vergnt 120.2	-				D 6	
	1	Ref.			Ref.				Ref.	
F. P. °C	-99.2	2	dt/dP			f	to			
F.P. 100%	ļ	\sqcup	°C/mm	, 521	ا ۔ ا	gl	, <u>•</u> <u>K</u>			
B. P. ℃			25°C B P	1.521 0.050	5 2	h				
760 mm 100	133.83	2	te	0.0371	5	f'	to			
30	71. 43.	2 2	90 mm	0.6905	4	g'	•K			
10	22.	2		0.0903	+ -	h'				
1	-10.86	5	ΔHm cal/g		_		300 to	0.0294	4	
Pressure			ΔHv cal/g			m n	600 °K	0.0274	4	
mm 25°C	11.572	5	25°C	78.3	2	0		-0.0652	4	
t _e	1114.	5	30 mm BP	74.88 64.9	5 2					
Density			te	63.36	5	m'	700 to	0.1222	4	
g/ml 20°C	0.7348	2	t _e (d, e)	63.23	5	n'	1000 •K	0.0013	4	
dt 25	0.7310	2	AHV/Te	19.25	5			-0.0 ₆ 45	*	
	0.7272	4	d 45 to	79.63	4	Surf	ace tension			
a	0.74998	4	e 150 °C	0.1101	4	dyne	s/cm. 20°C	22.38	2	
ь	-0.03756	4	a 10 to	83.01	5	8	30	21.45	2 2	
Ref. Index		,	e' 45 °C	0.1883	5	L	40	20.52	-	
ⁿ D 20°C	1.4123	2 2	d _c g/ml			Para	chor [P]			
30	1.4077	4	V _C m1/g				20°C 30			
"C"	 		tc °C	302.2	5		40			
	0.7469	4	P _c mm	16099.	5			385.2	5	
MR (Obs.)	43.46	2	PV/RT		†	Exp	L.1.%/wt.			
MR (Calc.) (nD-d/2)	43.762 1.0449	5 2	25°C	1.0000	5	JAP.	u.			
	1.0447	-	30 mm	1.0000	5	Disp	ersion	96.3	2	
Dielectric			B P	0.9599	5	Flas	h Point C			
A 40 to	6. 8482	2	t _e	0.9489	5	Fire	Point			
B (160 °C	1376.0	2	t _c			М. 5	Spec.			
	213.0	2	ΔHc kcal/m ΔHf			Ultr		l		
A* 40 to B* 160 °C	1.31046 1286.16	5 5	ΔFf				ay Dif.			
K Loo C	1200.10		Viscosity		—	Infra	red			
c			centistokes				bility in +		'	
t _k to			η °c				etone			
t _x °C			i '		١.,		rbon tet. nzene			
A' 25 to	6, 34515	5				Eth				
B' 1_45 °C	1130.32	5	B _w to	 			le pta ne			
C'	189.0	5					anol		1	
A'* 25 to	0.85689	5		.	'	Wa:	ter ter in			
B'* 45 °C	1060.39	5	(B ^V) to	1				ļ	\vdash	
Acl 160 to	7.2549	5	(A ^V) °C			H				
Bc tc °C	1694.6 253.8	5	c _p liq. °K					1		
	255.0			1	1	H				
Cryos, A° consts, B°			c vap.300°K	0.39649	2 2					
	+	\vdash	- 400	0.50136	-					
t _e °C	149.00	5	c _w vap.		L	L				
$T_{R} = 0.75$	Tc					† gra	ams/100 gra	ms solver	t	
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-	Calc. by for	mula		
SOURCE:			.PI							
PURIFICAT	ION.		.PI							
LITERATU	RE REFERE	NCES	5:							
İ										
1										
L										

							No. 70		
NAME	3-Ethyl-2,	3-di	methylpentane			STRUCTURAL FORMULA			
] [C	н ₃		
<u> </u>						сн,-сн с	CH ₂ CH ₃		
Mole % Pur.	Ref. Mo	lecul	аг а С ₉ Н ₂₀	Molecular Weight 128.	250	Ċн ₃ Ċ	2 ^H 5		
		Ref.		weight	Ref			Ref.	
F, P. *C	1	1	dt/dP	T	-	f	to		
F.P. 100%			°C/mm	1		g	<u>°</u> K.		
B. P. *C			25°C BP	2.0224 0.050	5 2	ь	1		
760 mm 100	142. 77.	2 2	t	0.0374	5	f'	to		
30	49.	2	30 mm	0.7083	4	g' '	<u>•</u> K_		
10	28. -7.09	2 5	AHm cal/g			h'		<u> </u>	
Pressure		Ť	ΔHv cal/g			m 300 n 600			
mm 25°C	8.591	5	25°C 30 mm	79.3 75.71	5		-0.0652		
Te te	1138.	5	BP	65.8	2	m' 700		┼	
Density g/ml 20°C	0.754	2	te (d, e)	64, 22 64, 05	5	n' 1000	*K 0.0013	4	
at 25	0.750	2	ΔHv/T	19.08	5	o'	-0.0 ₆ 45	4	
	0.746	4	d 50 to	+	4	Surface tension			
a b	-0.02797	4	<u> </u>	0.1066	4	dynes/cm. 20		2 2	
Ref. Index	1	 	d' 10 to		5	40		2	
n _D 20°C		2	d _c g/ml	1		Parachor [P]			
30	1.417	2 4	ll V mi/g	21/4	ا ۔ ا	30)°C		
"C"	0.7390	4	ic ic	316.4	5	40)	١.	
MR (Obs.)		2	P _c mm	16454.	3		igd. 385.2	5	
MR (Calc. (nD-d/2)) 43.762 1.042	5 2	25°C	1.0000	5	Exp. L.1.%/v u.	vt.		
Dielectric	1.042	-	30 mm BP	1.0000	5	Dispersion	94.6	2	
A 45 to	6,853	2	t _e	0.9589 0.9477	5	Flash Point °	С		
B <u>170</u> °C	1414.	2	t _c			Fire Point		┼	
C	214.	2	ΔHc kcal/m ΔHf			M Spec. Ultra V.	l		
A* 45 to B* 170 °C		5	ΔFf			X-Ray Dif.	l	}	
K	2	-	Viscosity			Infrared	+	┼	
t ₁ - t ₀	-		centistokes 7 °C			Solubility in Acetone	·	1	
t _x to			7 ℃			Carbon tet. Benzene	ŀ		
A' 25 to		5				Ether		1	
B' _ 50 °C	201.9	5	B ^V to	<u> </u>	\vdash	n-Heptane Ethanol			
A'* 25 to		5	A ^v i °C			Water			
	1207.57	5	(B ^V) to	7		Water in		<u> </u>	
Ac 170 to	7.2640	5	(A ^V) •C						
Bc tc_°C	256.6	5	c _p liq. °K			Į.			
Cryos. A°		Ť	c _p vap.300°K	0.39649	2	Į			
consts. B°			400	0.50136	2	Į.			
t _e °C	158.42	5	c _v vap.		L l				
$T_R = 0.7$						grams/100	grams solven	t	
	CES: 1-Dow	2-AI		Calc, from de	t. dat	ta 5-Calc. by	formula		
SOURCE:		AI							
PURIFICA'		AI							
LITERATU	RE REFERE	NCES	5:						
}									
L									

No. 71 3-Ethyl-2, 4-dimethylpentane NAME STRUCTURAL FORMULA C2H5 CH3CH-CH CH CH3 Ref. Mole Molecular Molecular Ċнз C9H20 ĊH3 % Pur. Weight 128, 250 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -122.2 2 dt/dP f to °C/mm 25°C g _°<u>K</u> 1.684 5 B. P. °C h BP 0.0503 760 mm 136.73 2 2 2 t_e 0.0366 5 ſ١ to 100 73. g' °<u>K</u> 30 0.6959 45. 30 mm 4 10 24. 2 h' ∆Hm cal/g -8.42 m 300 to 0.0294 AHv cal/g Pressure n 600 °K 0.0015 25°C 80.0 2 mm 25°C 10.228 o -0.0652 4 30 mm 75.34 5 1145. t_e ΒP 66.0 2 m 700 to 0.1222 4 Density 64.43 5 5 te te (d, e) 0.0013 n' [10<u>00 °</u>K g/ml 20°C 0.7379 2 64.32 ٥' -0.0645 4 0.7341 d_4^t 25 2 AHv/Te 19.40 5 30 0.73031 4 Surface tension d 50 to 79.98 4 0.75308 4 dynes/cm. 20°C 22.80 2 150 ℃ 0.1022 4 ь -0.0₃757 4 21.87 2 30 ă٦ 10 to 85.71 20.94 2 40 Ref. Index 50 °C 0,2286 5 20°C 1.4137 n D [P] Parachor d_c g/ml 25 1.4115 2 20°C vc ml/g 30 1.4091 4 30 t_c °C 307.6 5 40 "C" 0.7461 4 P_c mm 16350. 5 Sugd. 385. 2 5 MR (Obs.) 43.40 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 43.762 5 25°C 1.0000 5 (nD-d/2) 1.0488 2 30 mm 1.0000 5 Dispersion 96.4 2 Dielectric ВP 0.9753 5 Flash Point C 6.8524 te tc 0.9650 A 45 to 2 Fire Point 1389.0 B ∟165 °C 2 M. Spec. 213.0 2 AHc kcal/m Ultra V. **AHf** A* 45 to 1.2849 5 X-Ray Dif. ΔFf B*[165 °C 1291.5 Infrared Viscosity Solubility in centistokes c Acetone to Carbon tet. œ °C t^ Benzene 00 A' 25 to 6.10396 Ether œ B١ 45 °C 1029.13 n-Heptane œ B^V A 5 177.0 to Ethanol œ A'* °C Water 25 to 0.62992 5 B'* Water in 45 °C 965.97 5 (B^V) to Acl 165 to 7,2601 5 (A^V)| °C Bc tc °C 1711.9 c_p liq. ۰ĸ Cc 5 254.4 Cryos. A° consts. B° c_p vap300°K 0.39649 2 0.50136 2 c vap. te °C 5 153.13 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 72		
NAME	2, 2, 3, 3-7	[etra	methylpentane			STRUCTURAL FORMULA			
						CH ₃ CH ₃			
						сн ₃ с - с - с	H_2CH_3		
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 128.2	50	Ċн ₃ Ċн ₃			
		Ref.	F		Ref.			Ref.	
F.P. ℃	-9.90	2	dt/dP			f to			
F.P. 1009			*C/mm 25*C	1.893	5	g <u>*K</u> _		1	
B. P. °C 760 mm	140, 274	2	BP	0.05124	2	h		_	
100	75.705	2	t _e	0.0372	5	f' to			
30 10	47.43 26.05	2 2	30 mm	0.7064	4	h'		İ	
1	-10.3	5	ΔHm cal/g		-	m 300 to	0.0294	4	
Pressure mm 25°C	0.535	_	ΔHv cal/g 25°C	76.4	2	n _600 °K_	0.0015	4	
t _e	9.525 1150.3	5	30 mm BP	75.17 65.7	5 2	° .i	-0.0 ₆ 52	4	
Density			t.	63.93	5	m' 700 to n' 1000 °K	0.1222 0.0013	4	
g/ml 20°C	0.75666 0.75299	2 2	l e (d, e)	63.98	5	o' 11000 X	-0.0645	4	
dt 25 4 30	0.74932	4	ΔHv/T _e	19.05	5	Surface tension		 	
a L	0,77133	4	d 50 to e 160 °C	80.00 0.1020	4	dynes/cm. 20°C	23.38	2	
b Ref. Index	-0.03731	4	d' 10 to	77.77	5	30 40	22.48 21.60	2 2	
n _D 20°0		2	e' j 50 °C	0.0550	5	Parachor [P]			
25 30	1.42140	2 4	d g/ml vc ml/g			20°C 30			
"C"	0,7440	4	t _c *C	316.6	5	40		ĺ	
MR (Obs.)		2	P _c mm	16709.	5		385,2	5	
MR (Calc. (nD-d/2)) 43.762	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.			
Dielectric	1.04527	2	30 mm BP	1.0000 0.9714	5	Dispersion	95.9	2	
A 45 t	6. 82876	2	t _e	0.9606	5	Flash Point °C Fire Point			
B [180 °C	1397.483	2	t _c			M Spec.		_	
C 45 45 4	213.703	5	ΔHc kcal/m ΔHf	1		Ultra V.			
A* 45 to B* 170 °(5	ΔFf			X-Ray Dif. Infrared			
к ——-	-[1	Viscosity		[]	Solubility in +			
k		į	centistokes 7 °C	1		Acetone			
'x		L	•		1	Carbon tet. Benzene			
A' 25 to B' 50 °C		5				Ether n-Heptane			
C'	242.0	5	B ^V l to			Ethanol			
A'* 25 to B'* 50 °C		5	AV I C			Water Water in			
Ac 180 to		5	(B ^V) to						
Bc t °C	1788.9	5			-				
Cc — —	265.7	5	c _p liq. °K						
Cryos. Acconsts. B		2	c _p vap.300°K 400	0.39649 0.50136	2				
t _e ℃	157.18	5	c _v vap.						
$T_R = 0.7$						grams/100 grai		<u> </u>	
	CES: 1-Dow			Calc. from det	t. da	ta 5-Calc. by for	mula		
SOURCE:	TION.	AF							
PURIFICA	RE REFERE	AF							
LILERAIU	ne refekt	NUES);						

r										No. 73	
NAME		2, 2,	3,4-7	Cetra	methylpentane			STRUCTURAL FORMULA			
1								сн, сн,			
			T					снзс - сн-сн снз			
Mole		Rei	. Mo	ecul	ar CH 1	Molecular		•	-, ,	n Cn ₃	
% Pur.				mula	C ₉ H ₂₀	Weight 128.2	50		CH ₃ CH ₃		
				Ref.			Ref.				Ref.
F. P. ℃		-121.0)9	2	dt/dP			f	to		
F.P. 100	%				°C/mm	ł		g	<u>*K</u>		
B. P. ℃					25°C BP	1.434	4	h	l	}	
760 mm		133.0		2	t _e	0.05030 0.0374	5	f'	to		
30	- 1	41.8		2	30 mm	0.6938	4	g¹	' <u>*K</u>		
10		20.8		2	ΔHm cal/g	3.0750	├	h'			
1		-13.0)	5	ΔHv cal/g	 	-	m	300 to	0.0294	4
Pressure mm 25°C	.		- 7.4	١. ا	25°C	76.4	2	n	_6 <u>0</u> 0 •K	0.0015	
t _e	1	12.5 1114.4		4 5	30 mm	73.89	5	0		-0.0652	4
Density	-			-	BP	63.9 62.23	5	m'	700 to	0.1222	
g/ml 20°	c	0.7	73895	2	t _e (d, e)	62.23	5	n' o'	[1000 •K		
at 25	I		73524	2	ΔHv/T _e	18.93	5			-0.0645	<u> </u>
	4		73153	4	d 45 to	78.48	4		face tension		
a b			75378	4	e 150 °C	0.1096	4	dyn	es/cm. 20°C	21.98 21.11	2
Ref. Inde:	\exists	-0. (3738	-	d' 10 to	80.12	5	"	30 40	20.25	2 2
n _D 20°		1 4	11472	2		0.1488	5	Par	achor [P]		
45	Ì		1246	2	d _c g/ml				20°C		
30	_	1.4	1026	4	vc ml/g tc °C	302.3	5		30		
"C"		0.7	467	4	P _c mm	15991.	5		40 Sugd.	385.2	5
MR (Obs.		43.4		2	PV/RT		<u> </u>	Evr	L. l. %/wt.		╁╌╴
MR (Calc (nD-d/2)	۱۲.	43.7	762)4524	5 2	25°C	1.0000	4	DA	u.		
Dielectric	\exists	1.0	74324	-	30 mm BP	1.0000	5	Dis	persion	96.8	2
A 40 to		- (2172	-	t _e	0.9613 0.9503	4 5		sh Point C		
B 160 °C		1374.0	33173 042	2 2	t e	1,755		Fir	e Point		
c '	-	214.7		2	AHc kcal/m				Spec.		
A* 40 to	,	1.2	9082	5	ΔHf ΔFf				ra V. Ray Dif.		
B*[160 °C	드	1282.8	3	5					ared		
C					Viscosity centistokes			Sol	ubility in +		
t _k -te					η °c				etone		
t _x °(i '	l			rbon tet. enzene		
A' 25 to			6205	5		1		Et	her		
B' 1_45 °	ے	1239.1 201.8		5	B ^v to				Heptane hanol		
A'* 25 to	\forall		5786	5	A ^v I ^c C				ater		
B'* 45 °C		1161.4		5	(B ^V) to	1		W	ater in		
Ac 160 to		7.2	8064	5	(A ^V) °C		1				
Bc tc °		1735.3	3	5	c _p liq. °K						
Cc -		261.7		5		ļ					
Cryos, A consts. B		0.0	027	2	cp vap.300°K	0.39649	2				
	-	140		 –	c, vap.	0.50136	2				- '
t _e °C	لِ	148.2	:9	5	, v r.	L	L	L		İ	<u> </u>
$T_{\mathbf{R}} = 0$.									rams/100 gra		ıt
REFEREN	CI	ES: 1-	Dow			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:					.PI						
PURIFICA	TI	ON:		A	PI						
LITERAT	UR	E REI	FERE	NCES	3:						
1											

						No. 74			
NAME	2, 2, 4, 4-7	etra	methylpentane			STRUCTURAL FORMULA			
						CH ₃ CH ₃			
					\dashv	сн ₃ с - сн ₂ с - сн ₃			
Mole		lecul		Molecular Weight 128.	250	ch ₃ ch ₃			
% Pur.	FC	rmul		Weight 128.					
	-66.54	Ref.		1	Ref.				
F.P. °C F.P. 100%		+	dt/dP *C/mm	ļ		f to to			
B. P. *C	†	 	25°C	0.968	4	h			
760 mm	122. 284	2	BP t _e	0.04932 0.0378	2 5	f' to			
100 30	60.154	2 2	30 mm	0.6778	4	g' ' <u>*</u> K_			
10	12,50	2	ΔHm cal/g	0.01.10	÷	h'			
1	-24.7	5	ΔHv cal/g	 	\vdash	m 300 to 0.0294 4			
Pressure mm 25°C	20.0524	5	25°C	71.0	2	n _600 °K 0.0015 4 -0.0252 4			
t _e	1073.	5	30 mm BP	71.44 61.2	5 2				
Density				59.69	5	m' 700 to 0.1222 4 n' , 1000 °K 0.0013 4			
g/ml 20°C	0.71947 0.71563	2 2	te (d, e)	59.66	5	n' 1000°K 0.0013 4 0' -0.0 ₆ 45 4			
d ₄ 25	0.71178	4	ΔHv/T _e	18.72	5	-			
a	0.73482	4	d 35 to		4	Surface tension dynes/cm. 20°C 20.37 2			
ь	-0.03762	4	a 135 °C		*	30 19.46 2 40 18.55 2			
Ref. Index		2	e' j °C			 			
n _D 20°C	1.40459	2	d _c g/ml			Parachor [P] 20°C			
30			d g/ml v ml/g t °C	283.0	5	30			
"C"	0.7535	4	Pcmm	14836.	5	40 Sugd. 385.2 5			
MR (Obs.) MR (Calc.		2 5	PV/RT	 		Exp. L. 1. %/wt.			
(nD-d/2)	1.04720	2	25°C 30 mm	1.0000	5	u.			
Dielectric			BP	1.0000 0.9538	4	Dispersion 101.0 2 Flash Point °C			
A 30 to		2	t _e	0.9431	5	Fire Point			
B (_155 °C	216, 093	2	t _c	 	\vdash	M Spec.			
A* 30 to	+	5	ΔHf			Ultra V. X-Ray Dif.			
B* 145 °C		5	ΔFf			Infrared			
K		1	Viscosity centistokes			Solubility in +			
t _{te (} to			7 °C		1	Acetone Carbon tet.			
'x '	1		, '	j		Benzene			
A' 25 to B' <u>35</u> °C		5				Ether n-Heptane			
c, C == =	313.4	5	B ^V to			Ethanol			
A!# 25 to		5	AV I _ °C	_		Water			
B'* 35 °C	+	5	(B ^V) to	1		Water in			
Ac 155 to	7.2058 1635.6	5	(A ^V) °C						
Cc c-	256.0	5	c _p liq. °K						
Cryos. A° consts. B°	0.0273	2	c _p vap.300°K 400	0.39649 0.50136	2 2				
t _e °C	135.74	5	c _w vap.						
$T_R = 0.7$	5 T _C					+ grams/100 grams solvent			
REFERENC	CES: 1-Dow	2-AI	PI 3-Lit. 4-(Calc, from det	da				
SOURCE:		AI	PI						
PURIFICAT	TION:	AI	PI						
LITERATU	RE REFERE	NCES	:						

								No. 75	
NAME	2, 3, 3, 4	-Tetra	methylpentane			STRU	JCTURAL	FORMUL.	A
					1		CH ₃		
						CF	13СН С - С	CH-CH.	
Mole	Ref. M	folecul	ar C ₉ H ₂₀	Molecular	1		ch3ch3c		
% Pur.	F	ormul	a 5920	Weight 128.2	50		011301130	3	
		Ref			Ref.				Ref.
F.P. °C	-102.123	2	dt/dP			f	to		
F.P. 1009	•		°C/mm		1	gl	°K		
B. P. °C			25°C	1.997	5	h			
760 mm	141.551	2	BP t	0.04223 0.0376	5	f'	to		
100 30	76. 980	2 2	t _e 30 mm		4	g'	•K		
10	48.64 27.20	2		0.7083	 4	h'			
1	-8.44	5	AHm cal/g		└	m	300 to	0.0294	4
Pressure			ΔHv cal/g	1	1	n l	600 °K	0.0015	4
mm 25°C	8.867	5	25°C 30 mm	77.8 75.53	5	0 !		-0.0 ₆ 52	4
t _e	1131.	5	BP	65.1	2	 	1 700 1		_
Density			te (d.s)	63.31	5	m' n'	700 to	0.1222 0.0013	4
g/ml 20°0			te (d, e)	63.29	5	0'		-0.0645	4
d ₄ 30	0.7511		ΔHv/T _e	18.84	5	 		•	<u> </u>
a	+	-	d 50 to	81.00	4		e tension	23.31	2
b	0.7691		_•1160 °C	0.1123	4	gynes,	/cm. 20°C 30	22.44	2
Ref. Index		\top	d' 10 to e' 50 °C	80.20 0.0959	5	1	40	21.58	2
n _D 20°0				0.0737	1	Parac	hor [P]		
– 2 5	1.4200		d g/ml		ł	l	20°C		
30	1.4179		t _c °C	317.1	5	ŀ	30		
"C"	0.7437	4	P _c mm	16742.	5	l	40 Sugd.	385.2	5
MR (Obs.)		2	PV/RT		<u> </u>	Fyn	L.1.%/wt.		<u> </u>
MR (Calc. (nD-d/2)	1.0448	6 2	25°C	1.0000	5	Exp.	u.		
		0 2	30 mm	1.0000	5	Disper	rsion	95.5	2
Dielectric			BP	0.9554 0.9434	4 5	Flash	Point °C		
A 45 to			t _e t _c	0.7454		Fire F	Point		
B (175 °C	1417.473 214.705	2 2	ΔHc kcal/m	 	 	M. Sp			
A* 45 to			ΔHf			Ultra			
B* 170 °C		5 5	ΔFf			X-Ray Infrar			
к — — –	-		Viscosity		Ì	ļ	lity in +		
t,	-		centistokes		İ	Acet			
		ļ	η °C				on tet.		
t _x °C		9 5				Benz			
B' 50 °C		´ š			ļ	Ethe:	ptane		
C'	222.9	5	B _v to			Ethai			
A1# 25 to	1.4862	6 5	A °C	.[1	Wate			
B'* 50 °C		5	(B ^V) to			Wate	1111		
Acl 175 to		5	(A ^V) °C	<u> </u>]			
Bc tc °C		5	c _p liq. °K			1			
	257.6	-	1		1.				
Cryos. A° consts. B°		2	c vap.300°K	0.39649 0.50136	2 2				
t _e °C	+	5	c vap.	0.30130	-				
$T_{\mathbf{R}} = 0.7$	157.71 6T-		<u> </u>	L	L	H-+	/100		<u> </u>
			DI 2 1 4 4	Colo (1:	- د. ج		ns/100 gra		<u> </u>
	CES: 1-Dow	2-A		Calc. from de	ı. az	14 3-C	arc. by for		
SOURCE:	TION.								
PURIFICA		AF							
LITERATO	RE REFER	ENCE	:						
1									
L									

								No. 76	
NAME	n-Decane					STR	UCTURAL :	FORMULA	
							CII (CII)	CII	
Mole	7, 1,			Molecular			CH ₃ (CH ₂)	8 ^{CH} 3	
Mole % Pur.	Ref. Mo	rmul	ar C ₁₀ H ₂₂	Weight 142.2	276				
		Ref.			Ref.				Ref.
F.P. *C	-29.661	2	dt/dP			f	l to		
F.P. 100%			*C/mm 25*C	10,537	5	g	<u>*K</u> _		
B. P. *C 760 mm	174, 123	2	BP	0.05172	2	h	<u> </u>		
100	108.582	2	t _e	0.0364	5	f'	to to		
30 10	79.65 57.56	4 5	30 mm	0.7246	4	g' h'		f l	
ī	20.59	5	AHm cal/g	48.24	3		1 300 to	0,0256	4
Pressure			ΔHv cal/g 25°C	86, 28	2	m n	600 °K	0.0256	
mm 25°C	1.366	5	30 mm	79.99	5	٥		-0.0 ₆ 47	4
Density	1170.0	-	BP	66.0 63.55	2 5	m'	700 to	0.1099	4
g/ml 20°C		2	te te (d, e) AHv/T	64.74	5	n' o'	17000 .K		4
dt 25 4 30	0.72625 0.72245	2 2	ΔHv/Te	19.41	5		1	-0.0 ₆ 40	4
	0.74525	4	d 80 to		4		ace tension s/cm. 20°C	23.92	2
ъ	-0.03759	4	-a, - -190 °C		4 5	y	30	22.96	2
Ref. Index	3		e' 80 °C		5		40	22.02	2
ⁿ D 20°C	1.41189	2 2	d g/ml v ml/g	0.236	2	Par	achor [P] 20°C		
30	1.40734	4	v ^c ml/g	4.231 346.	2 2		30		
"C"	0.7510	4	t _c °C	15808.	2		40 Suad	424.2	5
MR (Obs.)		2	PV/RT	13000.	-	Fyn	. L. l. %/wt.	121.2	
MR (Calc. (nD-d/2)	48.38 1.04686	5 2	25°C	1.0000	5	DAP	u.		
Dielectric	1	<u> </u>	30 mm BP	1.0000 0.9345	5		ersion	98.0	2
A 75 to	6.95367	2	te	0.9187	5		sh Point °C Point	(
B 1 210 °C	1501.268	2	t _c	0,246	2	M S			
C	194.480	2	ΔHc kcal/m	1516.63	2	Ultr	a V.	,	
A* 75 to B* 200 °C		4	ΔFf				ay Dif. ared		
к — — —	1		Viscosity				bility in +		
t _k	-		remtistokes	0.5164	2	Ace	etone	ŀ	
t _x ; •0			130	0.4502	2		rbon tet. nzene		
A' 25 to	7.33883	5	150 170	0.3976 0.3543	2 2	Eth	er		
S, C 50 C	213.8	5	B 70 to	462,90	4		deptane nanol		
A'* 25 to	1.83020	4	AV 125 °C	2.50523	4	Wa	ter		
B'* 80 °C	1626.19	4	(BV) 125 to	464.77	4	Wa	ter in		
Ac 210 to Bc t _c °C	7.3363 1809.1	5	(A ^V) 180 °C	2.50071	4		osity]	
Cc _ c_	232.0	5	c _p liq. °K			cent η	tistokes 80°C	0.6544	2
Cryos. A°	0.05824	3	cp vap.300°K	0.39599	2	'	120	0.4814	
consts. B°	 		400	0.50072	2				
t _e °C	192.66	5	c _v vap.	<u> </u>	<u></u>	<u>L,</u>		i	Ĺ
$T_R = 0.78$							ams/100 grai		
	ES: 1-Dow	2-AI		Calc. from det	da	ta 5-	Calc. by for	mula	
SOURCE:	TON.	AI							
PURIFICAT		AI		202 (105 (1 = :					
LITERATU	RE REFEREI	NCES	: 3 JACS <u>76,</u>	333 (1954) Fi	nke e	et al			

No. 77 2-Methylnonane NAME STRUCTURAL FORMULA CH3CH (CH2)6CH3 Mole Ref. Molecular Molecular Weight 142.276 Ċн₃ $C_{10}H_{22}$ % Pur Formula Ref. Ref. Ref. F. P. °C -74.50 2 dt/dP f to F.P. 100% °C/mm °K g 25°C 6.8858 5 B.P. °C BP 0.05176 5 166.8 760 mm 2 4 t_e 5 ſ١ 0.0357 101.27 100 °K g' 30 72.38 4 30 mm 0.7231 5 10 50.35 5 h١ ∆Hm cal/g 13.55 5 m to AHv cal/g Pressure °K n 25°C 81,75 5 2.206 mm 25°C 5 o 30 mm 76.89 5 122.39 5 te ВP 66.18 5 m' Density to t_e (d, e) 63.96 5 n' °K g/ml 20°C 0.7281 2 63.94 5 ۰' d₄ 25 0.7242 2 ΔHv/T_e 19.79 5 30 0.7203 4 Surface tension d 10 to 85.10 5 0.7437 4 22.21 5 dynes/cm. 20°C °C 190 0.1134 5 ь -0.0378 4 21.27 5 30 to 20 84.32 5 40 20.36 5 Ref. Index e' °C 70 0.1026 5 20°C 1.4099 [P] n_D Parachor d_c g/ml 0.245 5 1.4076 25 2 20°C vc ml/g 4.076 5 4 30 1.4053 30 t_c 336. 5 40 "C" 0.7496 4 P_c mm 15364. 5 424.2 5 Sugd MR (Obs.) 48.40 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 25°C 1.0000 5 (nD-d/2)1.0459 2 30 mm 1.0000 5 5 Dispersion 99. 2 Dielectric 1.988 5 BP 0.9693 Flash Point C 0.9568 5 70 to 6.93010 5 Fire Point 0.242 1485.28 В 1_225 °C 5 M. Spec. C 5 AHc kcal/m 200. Ultra V. ΔHf A# 70 to 1.3949 5 X-Ray Dif. ΔFf B*[190 °C 1388.6 Infrared ĸ Viscosity Solubility in c centistokes Acetone to Carbon tet. °C t_x Benzene 10 to 7.2905 Ether В' 7<u>0 °C</u> 1688.1 5 n-Heptane B_v | C 218. 5 to Ethanol °C 10 to 70 °C Water A'* 1.7817 5 Water in B'* 1592.7 (B^V) 5 to 7.7412 Acl 225 to 5 (A^V)| °C Bc tc °C 2254.1 °C cp liq. °K Сc 298.5 5 Cryos. A° c_p vap. °K consts. B° c vap. t, °C 186.55 5 $T_R = 0.82 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 78	
NAME	3-Methyln	onan	•			STRUCTURAL	FORMULA	¥.
ĺ								
	- 1 ·					CH ₃ CH ₂ CH (CH	2)5CH3	
Mole % Pur.		lecul rmul		Molecular Weight 142.2	76	Ċн ₃		
		Ref.			Ref			Ref.
F.P. °C	-84.80	2	dt/dP			f to		
F. P. 1009 B. P. *C	<u> </u>	—	*C/mm 25*C	7.8111	5	g <u>*K</u> _		1
760 mm	167.8	2	BP	0.05101	5 5	h		<u> </u>
100 30	103.08 74.40	4	t _e 30 mm	0.03565 0.7190	5	f' to		ĺ
10	52.48	5	ΔHm cal/g	0.1170	-	h'		
<u>1</u>	15,75	5	ΔHv cal/g	 		m to		
Pressure mm 25°C	1.907	5	25°C	83.37	5	n •K		
t _e	1200.1	5	30 mm BP	78.24 66.38	5			
Density g/ml 20°C	0.7334	2	te te (d, e)	64.04 64.01	5	m' to		
dt 25	0.7296	2	ΔHv/T _e	19.82	5	o'		
	0.7258 0.7486	4	d 75 to	<u> </u>	5	Surface tension	22.0/	_
a b	-0.0376	4	_e <u> 185</u> °C	-1/	5	dynes/cm. 20°C	22.86 21.93	5
Ref. Index			d' 15 to		5	40	21.02	5
n _D 20°0	1.4125	2 2	d _c g/ml	0.243	5	Parachor [P] 20°C		
30	1.4080	4	v _c ml/g t _c °C	4.117 336.8	5	30		
"C"	0.7486	4	P _c mm	16121.	5	40 Sugd.	424.2	5
MR (Obs.) MR (Calc.		2	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0458	2	25°C 30 mm	1.0000	5	u. Dispersion	98.	2
Dielectric		5	BP	0.9536 0.9384	5 5	Flash Point °C		
A 75 to B <u>226</u> °C		5	te t _c	0.234	5	Fire Point		_
С	200.	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		-
A* 75 to B*_200 °C		5	ΔFf			X-Ray Dif. Infrared		
к —	-	-	Viscosity			Solubility in +		+
c	, -		centistokes 7°C			Acetone		Ì
'x '			'			Carbon tet. Benzene		
A' 15 to B' 75 °C		5				Ether n-Heptane		
Cı C	218.	5	B ^V to			Ethanol		ŀ
A'* 15 to B'* 75 °C		5	A ^V °C	_		Water Water in		
B'* 75 °C	+	5	(B ^V) to	1				\vdash
Bc t _c °(2291.	5			-			
Cryos. A	270.	5	F	ł				
consts. B			c _p vap. *K					1
t _e °C	186.5	5	c _v vap.	1				
$T_{R} = 0.8$	~					f grams/100 gran	ns solven	t
	CES: 1-Dow	2-AI		Calc. from det	da	ta 5-Calc. by for	nula	
SOURCE:	TION.		PI					
PURIFICA	RE REFERE		PI ·.					
LILERATU	AL REFERE	NUES);					
L								

							No. 79)
NAME	4-Meth	ylnonane				STRUCTURAL	FORMUL	ıΑ
						Сн (сн) сн (сн / сн	
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 142.2	76	сн ₃ (сн ₂) ₂ сн (сн ₃	C11 ₂ / ₄ C11	3
		Ref.			Ref.			Ref
F. P. ℃	-98.7	2	dt/dP			f to		
F. P. 1009	<u> </u>		°C/mm 25°C	7 1296	5	g '° <u>K</u>		
B. P. °C 760 mm	165.7	2	BP	0.05085	5	h		├
100	101.18	4	t _e	0.03566	5	f' to g'° <u>K</u>		
3 0 10	72.61 50.79	4 5	30 mm	0.7158	5	h'		
1	14.25	5	ΔHm cal/g	 	├ ─┤	m to		+-
Pressure mm 25°C	2.10	59 5	ΔHv cal/g 25°C	82.72	5	n		Ì
t _e	1195.8) 5	30 mm BP	77.78	5	0		<u> </u>
Density			t_	65.98 63.70	5	m' to		
g/ml 20°0	0.73		t _e (d, e)	63.64	5	n' <u>°K</u>		
d ^t 25 4 30	0.72		ΔHv/T _e	19.82	5	Surface to raise		+-
a	0.74	_ : 1 . [d 70 to	86.99 0.1268	5	Surface tension dynes/cm. 20°C	22.72	5
ь	-0.03	78 4	d' 15 to	85.31	5	30 40	21.77 20.84	5
Ref. Index		23 2	e' 70 °C	0.1037	5	Parachor [P]	20.04	+-
D 25	1.410	00 2	d g/ml	0.242 4.127	5 5	20°C		
30 "C"	1.40		vc ml/g tc °C	333.	5	30 40		
MR (Obs.	0.749	-	P _c mm	15903.	5	Sugd.	424.2	5
MR (Calc.		5	PV/RT		_	Exp. L. l. %/wt.		
(nD-d/2)	1.04	61 2	25°C 30 mm	1.0000	5	u. Dispersion	98.	2
Dielectric			BP	0.9542 0.9399	5 5	Flash Point °C	, - ,	一
A 70 to		04 5	te tc	0. 247	5	Fire Point		
<u> </u>	200.	5	∆Hc kcal/m		\Box	M. Spec. Ultra V.		
A* 70 to			ΔHf ΔFf			X-Ray Dif.		
B*[_190°C	1414.3	5	Viscosity	 		Infrared		┼
° - to	_		centistokes			Solubility in TACetone		
t _k to			່າ °⊂			Carbon tet.		
A' 15 to						Benzene Ether		
B' 1_ 70°C	2 1708.0 218.	5 5	B ^V to		\vdash	n-Heptane Ethanol		
A'* 15 to			B ^V to C			Water		
B'* 70°C		5	(B ^V) to			Water in		ـ
Acl 222 to	7.785		(A ^V) °C					
Bc t _c °C	2249.3	5 5	c _p liq. °K					
Cryos, Acconsts, B	-		c _p vap. °K					
te °C	184.13	5	c _w vap.					
$T_R = 0.8$			L			grams/100 gra	ms solver	at
REFEREN	CES: 1-De	ow 2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:		AF						
PURIFICA		AF						
LITERATU	JRE REFE	RENCES	3:					
L								

							N o. 80	
NAME	5-Methyln	onane	•			STRUCTURAL	FORMUL	A
1								
						сн ₃ (сн ₂) ₃ сн (с	H ₂) ₃ CH ₃	
Mole % Pur.		lecul rmul		Molecular Weight 142.2	76	CH ₃		
		Ref.			Ref			Ref.
F.P. °C	-87,70	2	dt/dP			f to		
F.P. 1009			*C/mm 25*C	6, 8851	5	g <u>°K</u> _		
B. P. °C 760 mm	165.1	2	BP	0.0508	5	h		
100	100.6	4	t _e	0.03560	5	f' to		
30 10	72.05 50.26	4 5	30 mm	0.7148	5	א יב _ ב. או יב _ ב.		ĺ
i	13.77	5	ΔHm cal/g	.		m to		├
Pressure	2 .0/2		ΔHv cal/g 25°C	82,52	5	n K		}
mm 25°C	2.1860 1197.0	5	30 mm	77.64	5	° '		<u> </u>
Density			BP t.	65.87 63.73	5	m' to	•	
g/ml 20°C		2	te te (d, e)	63.55	5	n' <u>*K</u> _		1
dt 25 4 30	0.7288 0.7250	2 4	ΔHv/T _e	19.61	5			₩
	0.7478	4	d 70 to		5	Surface tension dynes/cm. 20°C	22.76	5
ь	-0.0376	4	d' 1 25 to		5	y 30	21.83	5
Ref. Index		2	e' 70 °C		5	Parachor [P]	20.93	1
D 25	1.4100	2	d g/ml	0.244 4.098	5	20°C		i
30	1.4077	4	v _c ml/g t _c °C	333.	5	30 40		j
"C"	0.7489	4	P _c mm	16015.	5	Sugd.	424.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT	1 0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0459	2	25°C 30 mm	1.0000	5	u. Dispersion	98.	2
Dielectric	1.994	5	BP	0.9542 0.9421	5	Flash Point °C		
A 55 to B 222 °C		5	te t _c	0.247	5	Fire Point		
<u>c</u>	200.	5	ΔHc kcal/m			M Spec. Ultra V.		1
A* 55 to B* 190 °C	1.4783	5	ΔHf ΔFf			X-Ray Dif.		1
K Line	1407.5		Viscosity			Infrared Solubility in +		├-
t _k	. 	i i	centistokes 7°C			Acetone		
1 × (;		1]	Carbon tet. Benzene		1
A' 10 to B' 55 °C		5 5				Ether		
B' L 55 °C	218.	5	B ^V to			n-Heptane Ethanol		
A'* 10 to		5	AV C			Water Water in		1
B'* 55 °C		5	(B ^V) to	1				+-
Ac 222 to		5	(A ^V) °C	 		ļ		
Ce	294.	5	c _p liq. ∘K					
Cryos. A° consts. B°			c _p vap. °K	1				
t _e °C	183.47	5	c _v vap.	<u> </u>				
$T_R = 0.8$						† grams/100 gran		<u>t</u>
SOURCE:	CES: 1-Dow	2-AI	PI 3-Lit. 4- PI	Calc. from det	da:	ta 5-Calc, by for	nula	
PURIFICA'	rion:		PI					
	RE REFERE							

No. 81 3-Ethyloctane NAME STRUCTURAL FORMULA CH3CH2CH (CH2)4CH3 Ċ₂н₅ Mole Ref. Molecular Molecular $C_{10}H_{22}$ Weight 142, 276 % Pur Formula Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm °<u>K</u> g 25°C 7.8654 5 B.P. °C h BP 0.05101 5 760 mm 168. 2 t_e 0.0356 5 f١ to 100 103.2 4 g¹ <u>°к</u> 30 74.52 4 30 mm 5 0.7192 10 52.60 5 h* AHm cal/g 15.86 5 m to ∆Hv cal/g Pressure ۰ĸ n 25°C 83.42 5 mm 25°C 1.8929 o 30 mm 78.27 5 te 1202.5 5 ΒP 5 66.42 m to Density t_e (d, e) 5 64.16 n' °K g/ml 20°C 0.740 2 64.05 5 ٥' 25 $\mathbf{d_4^t}$ 0.736 2 AHV/Te 19.85 5 30 0.732 4 Surface tension 75 to 87.72 5 0.756 dynes/cm. 20°C 23.70 5 ٠C 0.1268 5 Ъ -0.0380 4 22.69 5 30 ٦·٦ 10 86.02 5 to 21.71 5 40 Ref. Index 75 5 0.1039 ⁿD 20°C 1.416 [P] Parachor d_c g/ml 0.245 5 25 1.414 2 20°C vc ml/g 4.085 30 1.411 4 30 337. 5 tc 40 "C" 0.7479 4 16170. P_c mm 5 Sugd. 424,2 5 MR (Obs.) 48.3 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 25°C 1.0000 5 (nD-d/2)1.046 2 30 mm 1.0000 Dispersion 96. 2 Dielectric 2,005 5 BP 0.9535 5 Flash Point C 0.9398 A 75 to te tc 7.0031 5 Fire Point 0.247 5 1517. B | 227 °C 5 M. Spec. C 200. 5 AHc kcal/m Ultra V. AHf A* 75 to 1.4945 5 X-Ray Dif. ΔFf B*|_200 °C 1427.6 Infrared K Viscosity Solubility in centistokes c Acetone $\mathbf{t_{k}^{t}}$ to Carbon tet. °C Benzene A' 10 to 7.3654 Ether B' 7<u>5 °C</u> 1722.5 n-Heptane B_v | 218. 5 to Ethanol °C Water A'* 10 to 1.8550 5 Water in B'* 75 °C (BV) 1626.7 5 to Acl 227 to 7.8210 (A^V) 5 °C Bc tc °C 2294.9 5 cp liq. °K Cc 298.1 5 Cryos. A° ۰ĸ cp vap. consts. Bo c_v vap. te °C 186,69 $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc, by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

	4 Fabriles					STRUCTURAL 1	No. 82	
NAME	4-Ethyloc	tane	·			STRUCTURAL	ORMULA	
Mole % Pur.		lecul rmul		Molecular Weight 142.2	76	сн ₃ (сн ₂) ₂ сн с ₂ н ₅	(СН ₂) ₃ СН ₃	
		Ref.			Ref		F	Ref
F.P. °C F.P. 100%			dt/dP *C/mm			f to		
B. P. °C 760 mm 100	168. 103.2	2 4	25°C BP t _e	7.8654 0.05101 0.03560	5 5 5	h fo to		
30 10	74.52 52.60	4 5	30 mm	0.7192	5	g' ' <u>*</u> K_		
1	15.86	5	ΔHm cal/g		<u> </u>	m to	-	
Pressure mm 25°C t _e	1.8929 1202.5	5	ΔHv cal/g 25°C 30 mm BP	83.42 78.27 66.42	5 5 5	n •K_		
Density g/ml 20°C dt 25 d4 30	0.740 0.736	2 2	te te (d, e) AHv/Te	64.16 64.05 19.85	5 5	m' to n' *K_		
a b	0.732 0.756 -0.0 ₃ 80	4 4	d 75 to e 185 °C d' 10 to	87.72	5 5 5	Surface tension dynes/cm. 20°C	23.70 22.69	5 5
Ref. Index n _D 20°C 25	1.416 1.414	2 2	e' 75 °C	0.1 0 39 0.245	5 5	Parachor [P] 20°C	21.71	5
30 "C"	1.411 0.7479	4	v _c ml/g t _c °C P _c mm	4.085 337. 16170.	5	30 40 Sugd.	424.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	48.3 48.38 1.046	2 5 2	PV/RT 25°C 30 mm	1.0000	5	Exp. L.1.%/wt. u. Dispersion	96.	2
Dielectric A 75 to	2, 005 7, 0031	5	BP t _e	0.9535 0.9398	5	Flash Point °C Fire Point		
B [227 °C C	1517. 200.	5 5	t _c AHc kcal/m AHf	0.247	5	M Spec. Ultra V.		
A* 75 to B* 200 °C K	1.4945 1427.6	5	ΔFf Viscosity centistokes			X-Ray Dif. Infrared Solubility in +		
tk to tx °C	7.3654		7 °C			Acetone Carbon tet. Benzene		
B' ∟75 °C	1722.5 218.	5 5 5	B ^V to			Ether n-Heptane Ethanol Water		
A'* 10 to B'* 75 °C Ac 227 to	1.8550 1626.7 7.8210	5 5	(B ^V) to			Water in		_
Bc tc_°C	2294.9 298.1	5	c _p liq. °K		-			
Cryos. A° consts. B°			c _p vap. *K					
t _e °C	186.69	5	c _w vap.	<u> </u>	<u> </u>	1		
REFERENCE		2 - AT	DT 3_144 A 6	'alo from de		grams/100 grant ta 5-Calc. by form		
SOURCE:	20. 1-DOW		PI 3-LAC, 4-C	aic. irom del	. da	ua 5-Caic, by for	nula	
PURIFICAT	ION:		PI					
	RE REFERE							

							т			No. 8	,,,	
NAME	2,	2-D:	imet	hyloc	tane			STRUCTURAL FORMULA				
[CH ₃				
ļl									сн ₃ с - (сн) CH		
Mole	F	lef.	Mo	ecul:	ar -	Molecular	ĺ			2/5		
% Pur.	-	1	For	mula		Weight 142.2	76		ĊН ₃			
				Ref.			Ref.	Γ			Ref.	
F.P. °C					dt/dP	1	1	f				
F.P. 100%	,				°C/mm	1		g	to K			
B. P. °C	1				25°C	3.864	5	h				
760 mm	155			2	BP	0.05104 0.0366	5		 		+-	
100		. 3		4	t _e	1	Į.	f' g'	to °K			
30 10		. 71 . 9		4 5	30 mm	0.7159	5_	h'	! -]		
i		. 4		5	ΔHm cal/g				<u> </u>		+	
Pressure					∆Hv cal/g		1	m n	to °K			
mm 25°C	4	. 23	7	5	25°C 30 mm	75.85	5	0	! <u></u>			
t _e	1161			5	BP	72.94 62.29	4	<u> </u>			4	
Density					t.	60.38	5	m' n'	to or or or or or or or or or or or or or			
g/ml 20°C		. 72		2	te (d, e)	60.32	5	0'		1		
d ₄ 25 30		. 72 . 71		2 4	ΔHv/T _e	19.29	5	ļ	L	ļ	+-	
		. 73		4	d 60 to	79.98	5		face tension	l	_	
a b		.03		4	_e_ _1 <u>70 °C</u>	0.1142	5	g yr	ies/cm. 20°C 30	21.77	5	
Ref. Index				\vdash	d' 10 to	77.84 0.0794	5	ľ	40	20.04	5	
n _D 20°C		. 40	82	2				Pa	rachor [P]			
25		.40		2	d g/ml	0.240 4.170	5		20°C			
30		. 40		4	vc ml/g tc °C	318.5	5		30 40	1		
"C"		. 75	04	4	P _c mm	14927.	5			424.2	5	
MR (Obs.)		. 47		2	PV/RT		 	Ext	o. L.1.%/wt.		-	
MR (Calc. (nD-d/2)		. 04	60	5 2	25°C	1.0000	5		u.			
Dielectric		. 98:		5	30 mm BP	1.0000	5	Dis	persion	100.	2	
A 60 to					t _e	0.9500 0.9369	5		sh Point °C			
B 205 °C		. 96	21	5	te	0.24	5	Fir	e Point			
c '	210			5	AHc kcal/m				Spec.			
A* 60 to	1	. 46	537	5	ΔHf		Ì		ra V. Ray Dif.			
B*[180 °C	1398	. 3		5	ΔFf	-	ļ		rared			
K	1				Viscosity centistokes		1	Sol	ubility in +			
t _k -to	-				η °C		1		etone			
t _x °C					'		1		arbon tet. enzen e			
A' 0 to		. 32	603	5					her			
B' _ 60 °C				5	B ^v to	+	 		Heptane			
	228			5	B to C				:hanol ater			
A'* 0 to B'* 60 °C		. 817	246	5	⊢.=v. 	-	1		ater ater in			
Acl 205 to	-	.71	10	5								
Bc tc °C				5		 	┼					
Cc	- 296	•		5	c _p liq. °K	1						
Cryos, A°		_			c _p vap. °K	1])		1		
consts. B°												
t _e °C	172	. 21		5	c _v vap.							
$T_{R} = 0.81$	T _c							+ g:	rams/100 gra	ms solve	nt	
REFEREN	CES:	1-D	ow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula		
SOURCE:				AP	I							
PURIFICA	TION:			AP	PI							
LITERATU		EFI	EREI	NCES								
1					· ·							
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1												
L												

								N	o. 84
NAME	7	2,3-D	imet	thylo	ctane		- 1	STRUCTURAL FOR	MULA
				_				CH ₃	
1		1			<u>-</u>			сн ₃ сн сн (сн ₂)	₄CH,
Mole		Ref.	Мо	lecul	ar c u	Molecular		ĊH ₃	4)
% Pur.				rmul		Weight 142.2	76	3	
_				Ref.			Ref.		Ref
F. P. *C	Т				dt/dP			f to	
F.P. 1007	•				°C/mm	1	1. 1	g *K	
B. P. *C	1			t	25°C	5.5982	5	h	
760 mm		63.8		2	BP	0.0519 0.0367	5	f' to	
100 30		98.0 68.9		4 4	t _e 30 mm	0,7287	5	g' '°K_	
10		46.7		5		0.7207	-	h' I	1
1		9.6		5	ΔHm cal/g			<u> </u>	
Pressure	T				ΔHv cal/g	70.4/	-	m to	
mm 25°C		2,82	8	5	25°C 30 mm	78.46 74.79	5	0	
t _e	111	83.		5	BP	63.73	5	⊩ .!	
Density	.		_,		t _e	61.68	5	m' to K	
g/ml 20°0	7	0.73		2 2	t _e (d, e)	61.60	5	", ' "	
dt 25 4 30		0.72		4	ΔHv/T _e	19.27	5		
	+	0.75		4	d 70 to		5	Surface tension dynes/cm. 20°C 23	. 39 5
ь		-0.03		4	180 °C		5		.41 5
Ref. Index	. -				d' 25 to		5	40 21	.47 5
n _D 20°0		1.41		2		0, 241	5	Parachor [P]	
25 30		1.41 1.41		2 4	d g/ml v ml/g	4, 154	5	20°C	
	+-				t _c ml/g	331.8	5	30 40	
"C"		0.74	83	4	P _c mm	15320.	5	Sugd. 424	. 2 5
MR (Obs.		48.28		2	PV/RT	 	\vdash	Exp. L.1.%/wt.	
MR (Calc. (nD-d/2)	"	48.38 1.04	60	5 2	25°C	1.0000	5	u.	
Dielectric	+	2.00		5	30 mm BP	1.0000	5	Dispersion 97	. 2
A 70 t	-	6.97		5	t _e	0.9480 0.9340	5	Flash Point °C	
B _ 220°			••	5	tc	0.24	5	Fire Point	
<u> </u>		08.		5	ΔHc kcal/m			M Spec. Ultra V.	
A* 70to		1.47	511	5	ΔHf			X-Ray Dif.	
B* 190°	2 14	32.2		5	ΔFf			Infrared	
K — — -					Viscosity centistokes			Solubility in +	
t _k	5				7 0	:	1	Acetone	
t _x i •	٦				l '			Carbon tet. Benzene	
A' 25to		7.33	426	5				Ether	
B' _ 70°		26.8 26.		5	B ^V to		-	n-Heptane	
	\rightarrow			-	B' to			Ethanol Water	
A'* 25to	16	1.81	744	5		-		Water in	
	-		20	5		1			
Ac 220to	23	7.79 01	20	5	(A ^V) °C				
Cc '— c—	- 3	06.		5	c _p liq. °K		l i		
Cryos. A					c _p vap. °K		i i		
consts. B	-			-	c _v vap.				1
t _e °C		82.10		5	о _{V} Р .	<u> </u>	L		
								grams/100 grams	
REFEREN	CES:	1-D	o w			Calc. from de	t. da	ta 5-Calc, by formul	a
SOURCE:	TION				PI				
LITERATU			ים פי		PI				
					•				

NAME	2, 4-Di	imethylog	tane			STRUCTURAL	FORMU	
Mole % Pur.	Ref.	Molecul Formula		Molecular Weight 142, 27	76	сн ₃ сн-сн ₂ сн сн ₃ сн ₃		¹ 3
		Ref.		T	Ref.			Ref.
F.P. °C F.P. 1009			dt/dP °C/mm			f to		
B. P. °C 760 mm 100 30 10	153. 88.51 60.03 38.3 1.9	2 4 4 5	25°C BP t _e 30 mm AHm cal/g	3.5488 0.05088 0.0367 0.7132	5 5 5	h to g' - °K		
Pressure mm 25°C t _e Density	4.651 1155.	14 5 5	ΔHv cal/g 25°C 30 mm BP	75.24 72.49 61.90 60.02	5 5 5	m to		
g/ml 20°0 dt 25 d4 30	0.726 0.722 0.718	24 2	t _e (d, e) AHv/T _e	59.97 19.27	5 5	n'eK_		
a b	0.742 -0.038		d 60 to e 170 °C d' 25 to	79.33 0.1139 77.20	5 5 5	Surface tension dynes/cm. 20°C 30 40	22.00 21.04 20.12	5 5 5
Ref. Index n _D 20°C 25 30		69 2	e' 60 °C d g/ml vc ml/g tc °C	0.0785 0.237 4.214 314.7	5 5 5	Parachor [P] 20°C 30	20.12	
"C"	0.750	03 4	P _c mm	14677.	5	40 Sugd	424. 2	5
MR (Obs.) MR (Calc. (nD-d/2)) 48.38 1.046		PV/RT 25°C 30 mm	1.0000	5	Exp. L.1.%/wt. u. Dispersion	98.	2
Dielectric A 60 to B 200 °C	6.956		BP te t _c ΔHc kcal/m	0.9500 0.9370 0.24	5 5 5	Flash Point C Fire Point M. Spec.		
A* 60 to B* 180 °C K	1.462		ΔHf ΔFf Viscosity centistokes			Ultra V. X-Ray Dif. Infrared Solubility in		
t _k to		217 5	η °C			Acetone Carbon tet. Benzene Ether		
B' _60 °C C' A'* 25 to			B ^V to A ^V °C	-		n-Heptane Ethanol Water Water in		:
B'* 60 °C Ac 200 to Bc tc °C Cc	7.687	78 5 5 5	(B ^V) to (A ^V) °C c _p liq. °K			water in		
Cryos. Acconsts. Bo			c _p vap. °K					
t _e °C T _R = 0.81	169.95 T _c	5	c _v vap.		<u></u>	grams/100 gra	ms solve	nt
REFEREN		ow 2-A	PI 3-Lit. 4-	Calc. from de	t. da	ata 5-Calc. by for		
SOURCE:			PI					
PURIFICA	TION:	А	PI					
LITERATU	RE REFI	ERENCES	5:					

						p	No. 86	<u>, </u>
NAME	2,5-Dimet	hylo	tane			STRUCTURAL	FORMUL	A
						כם כם וכם / כם	ICH \ C	u
Mole % Pur.		lecul rmul	Lucha I	Molecular Weight 142.2	76	ch ₃ ch (ch ₂) ₂ ch	(CH ₂ / ₂ C	**3
		Ref.	F		Ref	1		Ref
F,P. °C	I		dt/dP			f to		T
F.P. 100%			°C/mm		١	g°K_		1
B. P. °C	150		25°C BP	4.3755 0.05134	5	h ;		1
760 mm 100	158. 92.9	2 4	te	0.0365	5	f ¹ to		T
30	64.15	4	30 mm	0.7204	5	g'		
1 0 1	42.2 5.5	5	ΔHm cal/g			h'		
Pressure	<u> </u>	\vdash	ΔHv cal/g			m to		
mm 25°C	3.7001	5	25°C 30 mm	76.71 73.55	5			
t _e	1170.	5	BP	62.81	5	m' to		┼─
Density g/ml 20°C	0.736	2	te (d.e)	60.85	5	m' to		
t 25 d ₄ 30	0.732	2	e '-, -,	60.79	1	o'	İ	
⁴ 30	0.728	4	ΔHv/T _e	19.29	5	Surface tension	· .	+
a L	0.752	4	d 65 to e 175 °C		5	dynes/cm. 20°C	23.19	5
b	-0.038	4	d' 25 to	78. 73	5	30 40	22.19 21.23	5
Ref. Index	1.414	2	e' 65 °C		5	Parachor [P]		+-
25	1.412	2	d g/ml v ml/g	0.240 4.160	5 5	20°C		
30	1.409	4	tc C	323.2	5	30 40		1
"C"	0.7486	4	P _c mm	15080.	5		424.2	5
MR (Obs.) MR (Calc.)	48.3 48.38	2 5	PV/RT		\vdash	Exp. L.1.%/wt.		T
(nD-d/2)	1.046	2	25°C	1.0000	5	u.		
Dielectric	2.000	5	30 mm BP	1.0000 0.9500	5	Dispersion	99.	2
A 65 to	6.9658	5	te	0.9366	5	Flash Point °C Fire Point		
B 210 °C	1499. 209.	5	t _c	0, 24	5	M Spec.		
A* 65 to	1,46620	5	ΔHc kcal/m ΔHf			Ultra V.		1
B* 185 °C	1409.0	5	ΔFf			X-Ray Dif. Infrared	ĺ	1
K			Viscosity		ļ	Solubility in +		+
t _k			centistokes 7°C		İ	Acetone	ĺ	1
tÇ i •C			'		ł	Carbon tet. Bensene		1
A' 25 to	7, 32751	5				Ether	1	
B' ∟ 65 °C	1703.2 227.	5	B ^V to		\vdash	n-Heptane, Ethanol	1	
A'* 25 to	1.81284	5	AV I °C		}	Water	1	
B'* 65 °C		5	(BV) to		l	Water in		
Ac 210 to	7.7373	5	(A ^V) °C		1		ĺ	
Bc tc_°C	2217. 299.7	5	c _p liq. °K			1	ĺ	
Cryos. A°		 	1 -				1	
consts. B°			c _p vap. °K			1		
t _e °C	175.63	5	c _v vap.	į		1		
$T_{\mathbf{R}} = 0.81$	T _C					grams/100 gran	ns solver	ıt.
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc, from de	t. da			
SOURCE:		A	PI					
PURIFICATI	ION:	A	PI					
	E REFERE							
		_						

NAME	2,6-Dimet	hylo	tane			STRUCTURAL FORMULA				
						сн ₃ сн (сн ₂) ₃ с	н сн,сн	ا ۱		
Mole % Pur.		lecul rmuli		Molecular Weight 142.2	76		н ₃			
		Ref.			Ref.			Ref.		
F.P. °C F.P. 1009	6		dt/dP °C/mm			f to				
B. P. °C 760 mm	158.54	2	25°C BP	4.4828 0.05137	5	h		\sqcup		
100 30	93.4 64.62	4	t _e 30 mm	0.0366 0.7210	5	f' to to° <u>K</u>				
10	42.6 5.9	5	∆Hm cal/g			h'		\vdash		
Pressure mm 25°C		5	ΔHv cal/g 25°C	76.90	5	$\begin{bmatrix} \mathbf{m} & \mathbf{to} \\ \mathbf{n} & - \mathbf{K} \end{bmatrix}$				
t _e	1171.	5	30 mm BP	73.69 62.92	5			1		
Density g/ml 20°(0.7285	2	te te (d, e)	60.95	5	m' to				
d ^t 25 4 30	0.7245 0.7205	2 4	ΔHv/T _e	19.30	5	Surface tension		\vdash		
a b	0.7449	4 4	d 70 to		5	dynes/cm. 20°C	22.26	5		
Ref. Index	-0.0380	+	d' 25 to	78.92	5	8 30 40	21.29 20.36	5 5		
n _D 20°0	1.4113	2	d _c g/ml	0, 237	5	Parachor [P]				
25 30	1.4089	2 4	I v mi/g	4.217	5	20°C 30				
"C"	0.7516	4	t _c °C	322.7	5	40	424.2	_		
MR (Obs.		2	P _c mm	14866.	3		424.2	5		
MR (Calc. (nD-d/2)	1.0471	5 2	25°C	1.0000	5	Exp. L.1.%/wt. u.				
Dielectric		5	30 mm BP	1.0000 0.9500	5	Dispersion	99.	2		
A 70 to		5	t	0.9366	5	Flash Point C Fire Point				
B (210 °C	1502. 209.	5	tc ΔHc kcal/m	0. 24	5	M. Spec.				
A* 70 to		5	ΔHf ΔFf			Ultra V. X-Ray Dif.				
B*[185°C	1411.6	5	Viscosity	-		Infrared Solubility in +		\vdash		
			centistokes γ °C			Acetone Carbon tet.				
A' 25 to	7. 32953	5				Benzene Ether				
B' 1_70°C	1705,9 227.	5	B ^V to	†		n-Heptane Ethanol				
A'* 25 to	1.81464	5	_A <u>V</u>	_		Water				
B'* 70 °C		5	(B ^V) to			Water in		+		
Ac 210 to		5	(A ^V) °C		\vdash					
Cc	– 299.	5	c _p liq. °K							
Cryos. Acconsts. B			c _p vap. °K							
t _e °C	176.23	5	c _v vap.							
$T_{\mathbf{R}} = 0.81$						f grams/100 gra		nt		
	CES: 1-Dow			-Calc. from de	t. da	ta 5-Calc. by for	mula			
SOURCE:	TION		PI							
PURIFICA			.PI							
LILERATO	IRE REFERE	NCE	:							
1										

							No. 88
NAME	2,7-Din	nethylo	octane			STRUCTURAL FO	
Mole % Pur.	Ref. M	lolecul Formul	ar C ₁₀ H ₂₂	Molecular Weight 142.2	276	CH ₃ CH-(CH ₂) ₄ C CH ₃	н сн ₃ н ₃
		Ref.			Ref.		Re
F. P. *C	-54.	2	dt/dP			f to	
F.P. 100%)		*C/mm	4 7300	5	g <u>*K</u> _	
B. P. *C 760 mm	150.07		25°C BP	4.7380 0.05150		h	
100 mm	159.87 94.6	2	t _e	0.0367	5	f' to	
30	65.7	4	30 mm	0.7230	5	g'K	
10 1	43.7 6.8	5	∆Hm cal/g			h'	
Pressure	+	+-	ΔHv cal/g			m to	
mm 25°C	3.3920		25°C 30 mm	77.28 73.96	5		
t _e	1173.	5	BP	63.08	5	m' to	
Density g/ml 20°C	0.7242	2	te (d. e)	61.09 61.02	5	n' 'K	
dt 25	0.7202	2	e (-, -,	1	5	0'	
⁴ 4 30	0.7162	4	ΔHv/T _e	19.27	-	Surface tension	
a	0.7402	4	d 70 to		5	dynes/cm. 20°C	21.73 5
b Def Zerlee	-0.0380	4	a' _ 25 to	79.32	5		20.79 5 19.87 5
Ref. Index		2	e' j 70 °C		5	Parachor [P]	
25	1.4062	2	d g/ml vc ml/g	0, 235 4, 261	5	20°C	
30	1.4038	4	tc °C	323.6	5	30 40	
"C"	0.7514	4	P _c mm	14735.	5	Sugd. 4	24.2 5
MR (Obs.) MR (Calc.		2 5	PV/RT	<u> </u>		Exp. L.1.%/wt.	
(nD-d/2)	1.0465	ž	25°C	1.0000	5	u.	98.6 2
Dielectric	1.984	5	30 mm BP	0.9490	5		70.0 2
A 70 to	6.9697	5	t _e	0.9354	5	Flash Point °C Fire Point	
B [_2]0 °C		5	t _c	0,24	13	M Spec.	
A* 70 to	209. 1.4701	5 5	ΔHc kcal/m ΔHf	İ		Ultra V.	
B* 185 °C		5 5	ΔFf			X-Ray Dif. Infrared	
к — — —	7	- 1	Viscosity			Solubility in +	
\$. -		centistokes 7 °C			Acetone	:
tî t	;		'			Carbon tet. Benzene	
A' 25 to					1	Ether	
B' _ 70 °C	227.	5	B ^V to	 	\vdash	n-Heptane Ethanol	
A'* 25 to			ĀV ∤ °C			Water	
B'* 70 °C		5	(BV) to			Water in	
Ac 210 to		5	(A ^V) °C				
Bc tc_°C	2225.	5	c _p liq. °K				
Cryos. A°	1	<u> </u>	i	j			
consts. B°			c _p vap. °K				
t _e °C	177.69	5	c _v vap.				
$T_R = 0.8$	l T _c		· · · · · · · · · · · · · · · · · · ·			+ grams/100 grams	solvent
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	t. dat	ta 5-Calc, by form	
SOURCE:		A	PI				
PURIFICAT	TION:	A	PI				
	RE REFER	ENCES):				

NAME	3, 3-Dimethyloctane						STRUCTURAL FORMULA			
	·	r					c	сн ₃ н ₃ сн ₂ сн (с	H-) (CH-	
Mole % Pur.	Ref.	Mo! For	ecul:		Molecular Weight 142.2	76	Ū	сн ₃	2/4	
			Ref.			Ref.				Ref.
F. P. ℃				dt/dP			f	to		
F.P. 100%				°C/mm			g	<u>•K</u>		
B. P. ℃	1,7,2		,	25°C BP	5.0083 0.05163	5	h	1		
760 mm 100	161.2 95.7		2	te	0.0367	5	f'	to		
30	66.78	В	4	30 mm	0.7250	5	g'	<u>K</u>		-
10	44.7		5 5	∆Hm cal/g		Γ	h'	<u>i</u>		<u> </u>
Pressure	†			∆Hv cal/g			m n	to °K		1
mm 25°C	3.19	931	5	25°C 30 mm	77.66 74.23	5	0	! ' 		
t _e	1176.		5	BP	63.25	5	m'	1 1 4-		+-
Density g/ml 20°C	0.7	200	,	t _e	61.24	5	n'	to °K		
dt 25 4 30	0.73		2	te (d, e)	61.16	5	0'			
^d 4 30	0.7		4	ΔHv/T _e	19.26	5	Sur	face tension		+
a	0.75		4	d	81.99 0.1162	5		es/cm. 20°C	23.57	5
b	-0.0	3 / 8	4	di 7 25 to	79.72	5	8	30 40	22.59	5
Ref. Index	1.41	165	2	e' 70 °C		5		rachor [P]	21.03	+-
45	1.41	142	2	d g/ml	0.242 4.136	5	F.	20°C	1	
30	1.41		4	v _c ml/g t _c °C	328.5	5		30 40	i	
"C"	0.74		4	P _c mm	15304.	5			424. 2	5
MR (Obs.) MR (Calc.)			2 5	PV/RT	 	1	Exp	L.1.%/wt.		
(nD-d/2)	1.04		2	25°C 30 mm	1.0000	5		u.	07	2
Dielectric	2.00	07	5	BP BP	1.0000 0.9480	5		persion	97.	 ' -
A 70 to	6.97	713	5	t _e	0.9342	5		sh Point C e Point	ł	
B (215 °C	1512. 208.		5	t _c	0, 24	5		Spec.		+-
A* 70 to	1.47	722	5	ΔHc kcal/m ΔHf			Ult	ra V.		
B* 190 °C		123	5	ΔFf				Ray Dif. rared		1
к — — —	-			Viscosity			ļ	ubility in +		+-
t _k to	-			centistokes り °C			A	etone		
t _x °C				'				arbon tet. enz e ne		
A' 25 to B' 70 °C	7.33	308	5				Et	her		
c' - '- =	1716.0 226.		5	B _v to				Heptane hanol		
A'* 25 to	1,81	491	5	A °C			w.	ater		
B'* 70 °C			5	(B ^V) to				ater in		+-
Ac 215 to	7.76	597	5	(A ^V) °C	1					
Bc tc °C	304.		5	c _p liq. °K						
Cryos. A° consts. B°				c _p vap. °K						
t _e °C	179.15	;	5	c vap.						
$T_R = 0.8$			لـــــا	L			+ ~	rams/100 gra	ms solve	nt.
REFERENC		Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da				
SOURCE:				PI						
PURIFICAT	ION:			PI						
LITERATU		EREI								

							No. 90	
NAME	3, 4-Dime	hylo	ctane			STRUCTURAL I	ORMULA	4
						CH ₃		
			1		\neg	CH3CH2CH CH (CH ₂) ₃ CH ₃	3
Mole % Pur.		le cul		Molecular Weight 142.	276	ĊH ₃		-
		Ref.			Ref			Ref.
F.P. °C			dt/dP			f to		
F.P. 1009	6		*C/mm 25*C	6, 1341	. 5	g <u>*K</u>		1
B. P. °C 760 mm	166.	2	BP	0.05208	5	h +		-
10 0	99.9	4	t _e	0.0367	5	f' to		
30 10	70.7 48.4	5	30 mm	0.7320	-	h'		
1	11.1	5	ΔHm cal/g ΔHv cal/g	 	\vdash	m to		\vdash
Pressure mm 25°C	2,5603	5	25°C	79.08	5	n ' <u>*K</u> _		
te	1190.	5	30 mm BP	75.22 64.10	5	I		_
Density	0.746	1	•	62.01	5	m' to		
g/ml 20°0 dt 25 4 30	0.742	2	te (d, e)	19.27	5	0'		1
	0.738	4	ΔHv/T _e		5	Surface tension		
a b	0.762 -0.0 ₃ 8	4 4	e 185 °	0.1167	5	dynes/cm. 20°C	24.47 23.44	5
Ref. Index		t	d' 25 to		5	40	22.44	5
n _D 20°0	1.4182 1.4159	2 2	d _c g/ml	0.242	5	Parachor [P] 20°C		
30	1.4135	4	v _c m1/g t _c °C	4,128	5 5	30		
"C"	0.7456	4	P _c mm	15521.	5	40 Sugd	424.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT	-	Н	Exp. L.1.%/wt.		
(nD-d/2)	1.045	2	25°C 30 mm	1.0000 1.0000	5 5	u. Dispersion	97.	2
Dielectric		5	BP	0.9480	5	Flash Point °C	71.	+-
A 70 t B 225 •		5	te t _c	0.9339	5 5	Fire Point		
c Lang	208.	5	ΔHc kcal/m		\vdash	M Spec.		
A* 70 to		5	ΔHf ΔFf			Ultra V. X-Ray Dif.		
B* 195 °C	2 1439.7	5	Viscosity	 	\vdash	Infrared		<u> </u>
c	_	1	centistokes			Solubility in + Acetone		
t _x t ₀			η °	1		Carbon tet. Benzene		
A' 25 to		5		i		Ether		
B' _ 70 °	2 1734.7 226.	5	B ^V to		\Box	n-Heptane Ethanol		
A'* 25 to		5	A _ °C	<u>: </u>		Water		
B'* 70 °C		5	(B ^V) to	1		Water in		\vdash
Ac 225 to		5	(A ^V) •C		Ш			
Ce	310.	5	c _p liq. °K					
Cryos. Accounts. B			c _p vap. °K					
t _e °C	184.62	5	c _w vap.					
$T_R = 0.0$	4	L	II	1	Ц	+ grams/100 gran	ns solven	t
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc. from det	t. dat	ta 5-Calc. by form		
SOURCE:			PI					
PURIFICA			PI					
LITERATU	RE REFERE	NCES	5 :					
}								

101 No. 91 3, 5-Dimethyloctane NAME STRUCTURAL FORMULA CH3CH2CH CH2CH (CH2)2CH3 Mole Ref. Molecular Molecular CH₃ CH₃ $C_{10}H_{22}$ % Pur Weight 142, 276 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g <u>°K</u> 25°C 9.4812 B. P. °C h ВP 0.05150 5 760 mm 160. 2 t_e 0.0367 5 ſ١ 100 94.7 4 to 0.7231 5 °<u>K</u> 4 g' 65.82 30 30 mm 5 10 43.8 h١ 5 ∆Hm cal/g 6.9 to AHv cal/g Pressure n ۰ĸ 25°C 77.33 mm 25°C 3.3674 o 5 30 mm 74.00 1172. 5 te BР 63.06 5 m to Density 61.07 te te (d, e) °K g/m1 20°C 0.736 2 60.99 5 o' 25 0.732 $\mathbf{d_4^t}$ 2 ΔHv/T_e 19.27 5 30 0.728 4 Surface tension d 70 to 81,65 5 0.752 44 dynes/cm. 20°C 23.19 <u>°C</u> 0.1162 5 -0.038 ь 30 22.19 āΠ 25 to 79.38 5 5 21.23 40 Ref. Index 70 °C 0.0817 5 $^{\mathbf{n}}\mathbf{D}$ 1.413 20°C [P] Parachor d_c g/ml v_c ml/g t_c °C 5 0.240 25 1.411 2 20°C 4.170 30 1.408 4 30 t_c 325.8 5 40 "C" 0.7468 4 P_c mm 15110. 5 Sugd. 424.2 5 MR (Obs.) 48.2 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48 38 5 25°C 1.0000 (nD-d/2)1.045 2 30 mm 1.0000 Dispersion 97. 2 Dielectric 2.039 5 ВP 0.9480 5 Flash Point C 0.9343 5 A 70 to 6.9710 5 Fire Point 0.24 B 1_215 °C 1508. 5 M. Spec. C 209. 5 AHc kcal/m Ultra V. ΔHf A# 70 to 1.47305 5 X-Ray Dif. ΔFf B* 185 °C 1418.2 Infrared K Viscosity Solubility in centistokes Acetone to t_k t_x Carbon tet. °C Benzene A' 25 to 7.33132 Ether В¹ 70 °C 1711.9 5 n-Heptane B_v | C' 227. 5 to Ethanol °C Water 1.81589 5 25 to 70 °C Water in (B^V) 5 1611.5 to Ac 215 to 7.7550 5 (A^V) °C Bc tc °C 2243. 5 c_p liq. °K Cc 301. 5 Cryos. Aº c_p vap. °K consts. B° c vap. te °C 177.78 $= 0.81 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

	_					No. 9	2		
NAME	3,6-Dime	thylo	ctane			STRUCTURAL FORMULA			
						CH CH CH (CH) CH CH	CU		
34-1-	2.4			Molecular		Сн ₃ сн ₂ сн (сн ₂) ₂ сн сн ₂ сн ₃ сн ₃	СП3		
Mole % Pur.		olecul ormul		Molecular Weight 142.	276	cn ₃ cn ₃			
	_	Ref.	l .		Ref		Ref.		
F.P. °C			dt/dP			f to			
F.P. 1009			*C/mm 25*C	4,7691	.5	g <u>*K</u>			
B. P. °C 760 mm	160.	2	BP	0.05150	5	h			
100	94.7	4	t _e	0.0367	5	f' to o'K	Ì		
30 10	65.8 43.8	4 5	30 mm	0.7231	5	g'			
i	6.9	5	ΔHm cal/g		<u> </u>				
Pressure			ΔHv cal/g 25°C	77, 33	5	m to			
mm 25°C	3.3674 1172.	5	30 mm	74.00	5	0	į		
Density	+	+	BP	63.06	5	m' to			
g/m1 20°0		2	te (d, e)	60.99	5	n' °K			
dt 25 4 30	0.7324	2	AHV/Te	19.27	5	8. 1			
1 30	0.7519	4	d 70 to		5	Surface tension dynes/cm, 20°C 23.22	5		
ь	-0.0378	4	e 175 °C		5	30 22.25	5		
Ref. Index			e¹ 70 °C		5	40 21.31	5		
n _D 20°0	1.4145	2 2	d _c g/ml	0.241	5	Parachor [P] 20°C	1		
30	1.4099	4	ıı v xnı/ø	4.147 326.4	5	30	-		
"C"	0.7491	4	-	15209.	5	40 Sugd. 424. 2	5		
MR (Obs.)		2	P _c mm PV/RT	13207.	۲	Exp. L.1.%/wt.	+-		
MR (Calc. (nD-d/2)	1.0463	5 2	25°C	1.0000	5	u.			
Dielectric		5	30 mm BP	1.0000 0.9480	5	Dispersion 97.	2		
A 70 t		5	te	0.9343	5	Flash Point °C Fire Point	ı		
B (215 °C		5	t _c	0.24	5	M Spec.	+		
C	209.	5	ΔHc kcal/m ΔHf		-	Ultra V.			
A* 70 to B* 185 °C		5	ΔFf		L	X-Ray Dif. Infrared	į		
к — — -	-		Viscosity			Solubility in +	_		
'k	-	1	centistokes 7 °C			Acetone			
t _x						Carbon tet. Benzene			
A' 25 to B' 70 °C	7.33132 C 1711.89	5			1	Ether	-		
c, - 10	227.	5	B ^V to		\Box	n-Heptane Ethanol			
A'* 25 to		5	AV C]		Water			
	1611.5	5	(B ^V) to			Water in	+		
Ac 215 to	7.7581	5	(A ^V) °C		L	1			
Cc Cc	302.	5	c _p liq. ∘K		1				
Cryos, A			c _p vap. °K						
t _e °C	177,78	5	c _v vap.						
$T_R = 0.8$	i	1	L	L	<u> </u>	+ ==== /100			
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from de	+ 4-	grams/100 grams solve sta 5-Calc. by formula	nt		
SOURCE:		AP		Jane. Hom de	ue	J-Carc. by formula			
PURIFICA	TION:	AP	· · · · · · · · · · · · · · · · · · ·						
	RE REFERE								
			: -						

No. 93 4, 4-Dimethyloctane NAME STRUCTURAL FORMULA сн3 $CH_{3}(CH_{2})_{2}C - (CH_{2})_{3}CH_{3}$ Mole Ref. Molecular Molecular Weight 142.276 ĊH3 $C_{10}H_{22}$ % Pur. Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g _°<u>K</u> 4.9725 5 B. P. °C h ВP 0.05160 5 760 mm 161. 2 0.0367 5 ſ١ to 100 95.6 4 g' <u>•K</u> 30 30 mm 5 66.64 4 0.7246 10 44.6 5 h! ∆Hm cal/g 7.6 1 5 m to AHv cal/g Pressure °K n 25°C 77.62 5 mm 25°C 3,2176 o 30 mm 74.20 5 1175. 5 te BP 63.23 5 m' to Density 5 t_e (d, e) 61.22 n' g/ml 20°C °K 2 0.737 61.15 ۰, $\mathbf{d_{4}^{t}}$ 25 0.733 2 ΔHv/Te 19.27 5 30 0.729 4 Surface tension Т 70 to ď 81.95 5 0.753 a 4 180 °C 25 to dynes/cm. 20°C 23.31 5 0.1163 ь -0.038 4 22.31 30 ď٦ 79.68 5 5 21.35 40 Ref. Index e' 70 °C 0.0821 5 ⁿD 20°C 1.414 Parachor [P] d_c g/ml 0.240 5 25 2 1.412 20°C vc ml/g t °C 4.168 327.4 5 30 1.409 4 30 **t**c 5 40 "C" 0.7475 4 P_c mm 15157. 5 Sugd. 424.2 5 48.2 MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 5 25°C 1,0000 5 (nD-d/2)1.046 2 u. 30 mm 1.0000 5 Dispersion 97. 2 2,000 Dielectric 5 BP 5 0.9480 Flash Point C 0.9342 5 A 70 to 6.9720 5 Fire Point 0.24 5 1511. 1 215 °C M. Spec. 208. 5 AHc kcal/m Ultra V. ΔHf A* 70 to 1.47331 X-Ray Dif. ΔFf B* 190 °C 1421.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to $\mathbf{t_k} \mid \mathbf{t_x} \mid$ Carbon tet. °C Benzene 25 to 7.33180 Ether B' L 70 °C 1715.69 n-Heptane B_v | 226. 5 Ethanol to °C Water A'* 25 to 1.81600 5 Water in B'* 70 °C (B^V)| 1615.2 to Ac 215 to 7,7641 (A^V) 5 °C Bc tc °C 2257. c_p liq. ۰ĸ Cc 303. 5 Cryos. A° c_p vap. °K consts. B° c vap. te °C 178,92 $T_{\mathbf{R}} = 0.81 \, T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

										No. 94	
NAME	4,5-1	Dimethy	loctar	ıe				STR	UCTURAL 1	FORMUL	A
									СН		
		<u> </u>			Т		$\neg \dashv$	CH ₂ (C	:н ₂) ₂ сн-сн		Ηą
Mole	Ref.			C ₁₀ H ₂₂		Molecular	7.	,	Čн ₃		,
% Pur.		Form		10 22		Weight 142.2		_			Ref
E D 86	Т	Re	_			ı	Ref.		1		Kei
F.P. °C F.P. 100%	 			t/dP *C/mm				f g	to K		1
B. P. *C	†		-(1	25°C		5.2038	5	h			1
76 0 mm 100	162.13		, ,,	BP t _e		0.05173 0.0367	5 5	f'	to		+
30	96.5	4		30 mm		0.7264	5	g'			
1 0 1	45.4 8.3	5		Hm cal/	g'g			h'	l		
Pressure	8.3	- -	⊣ Δ	Hv cal/	 g			m	to to		1
mm 25°C	3.06)	25°C 30 mm		77.92 74.41	5 5	n o	•K		
t _e	1178.	5	<u>'</u>	BP		63.41	5	m'	to		+
Density g/ml 20°C	0.74	70 2		te (d, e)		61.39 61.31	5 5	n'	K K		
at 25	0.74	32 2	- 11	te (d, e)		19.26	5	٥'	, – – – – I		
	0.73				to	82,26	5		ace tension		
a b	0.76		_e	180	_ ℃	0.1162	5	dyne	s/cm. 20°C 30	24.60 23.62	5
Ref. Index	13	<u> </u>	a		_ to C	79.99 0.0826	5 5		40	22.66	5
n _D 20°C			, II—			0, 246	5	Para	chor [P]		1
25 30	1.41		v	c g/ml c ml/g		4.062	5		20°C 30		
"C"	0.74		_∥ Կ	: 'C		331.6	5		40		
MR (Obs.)				c mm		15661.	5	<u> </u>		424.2	5
MR (Calc.)	48.38	. 5	, P	V/RT 25°C		1.0000	5	Exp.	L.1.%/wt. u.		1
(nD-d/2)	1.04		_ ։	30 mm		1.0000	5	Disp	ersion	95.8	2
Dielectric A 70 to	2.01 6.97			BP ^t e		0.9480 0.9341	5		h Point °C		
B 220 °C		17 5	. L	t c		0.24	5	I 	Point		╄
С	208.	5	⊣ I ∓	Hc kcal	/m			M S			
A* 70 to B* 190 °C		193 5	, 11	Hf Fí		1		X-R	ay Dif.		1
K L Z	-	١		iscosity				Infra			+
¢to	-	j		entistoke					bility in +		j
tk to		ŀ	7		•c				bon tet.		
A' 25 to						İ	1	Eth	er		1
B' ∟ 70 °C	1719.3 226.	5		v	to		-		leptane anol		
A'* 25 to	+	-+	—∥ .	v j	٠c			Wa			j
	1618.8	5		3 v) —	to		1	Wa	ter in		╄
Ac 220 to				Δ Υ)	°C			l			1
Bc tc_°C	2295. 307.	5	c,	liq.	°K						1
Cryos. A°			_	vap.	°K						1
consts, B°			- 11		••						-
∙t _e °C	180, 22	5	c,	vap.							
$T_R = 0.8$								+ gra	ms/100 gran	ns solver	nt
REFERENC	ES: 1-D		API	3-Lit.	4-0	alc. from det	. da	ta 5-	Calc. by form	nula	
SOURCE:			API								
PURIFICAT			API								
LITERATU	RE REF	ERENC	ES:								

No. 95 4-Propylheptane STRUCTURAL FORMULA NAME CH3(CH2)2CH (CH2)2CH3 Mole Ref. Molecular Molecular Weight 142, 276 Ċ₃H₇ $C_{10}H_{22}$ % Pur Formula Ref. Ref. Ref. <u>F.P. °C</u> dt/dP f to F.P. 100% °C/mm g <u>°K</u> 25°C 5.1849 5 B.P. °C h BP 0.05169 760 mm 162. 2 5 0.0367 ſ١ to 96.4 100 4 0.7261 g' <u>• K</u> 5 67.5 30 4 30 mm 10 45.3 5 h١ ∆Hm cal/g 8.3 5 to m AHv cal/g Pressure °K n 25°C 77.91 5 3.0744 mm 25°C o 30 mm 74.41 1178. 5 te BP 63.41 5 m' to Density te te (d, e) 5 61.38 n' g/ml 20°C °K 0.7364 2 5 61.30 ۰, ď4 25 0.7326 2 ΔHv/T_e 19.27 5 30 0.7288 4 Surface tension 65 **to** 82, 26 5 0.7516 4 dynes/cm. 20°C 23, 24 180 °C 0.1164 5 ь -0.0376 4 30 22.29 5 aח 25 to 79.98 5 40 21.37 5 Ref. Index e١ 65 °C 0.0826 5 20°C 1.4150 [P] ⁿD Parachor 5 d_c g/ml 0.242 25 1.4127 2 20°C vc ml/g t °C 4.135 30 1.4105 4 30 ^tc 329.6 5 40 "C" 0.7498 4 P_c mm 15334. 5 Sugd. 424.2 5 MR (Obs.) 48.38 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 5 (nD-d/2)2 1.0468 30 mm 1.0000 5 Dispersion 96. 2 Dielectric 2,002 5 BP 0.9480 5 Flash Point C 0.9342 5 65 to 6.9734 5 Fire Point 0.24 В _220 °C 1515. 5 M. Spec. С 208. 5 AHc kcal/m Ultra V. ΔHf A* 65 to 1.47356 5 X-Ray Dif. ΔFf B*| 190 °C 1425.4 Infrared ĸ Viscosity Solubility in c centistokes Acetone to Carbon tet. °C t_x Benzene 25 to 7.33227 Ether B' 1719.5 65 °C 5 n-Heptane Bv | Av | C 226. 5 Ethanol °C Water 25 to 65 °C 1.81612 5 Water in B'* 1619.0 (B^V)| to Acl 220 to 7,7776 5 (AV) °C 2278. Bc tc °C c_p liq. °K 5 Cc 305. c_p vap. Cryos. Aº °K consts. B° c_v vap. te °C 180.06 5 $T_{R} = 0.81 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 96	
NAME	4-Isoprop	ylher	otane			STRUCTURAL	FORMUL	A
								-
						сн ₃ (сн ₂) ₂ сн -	(CH ₂) ₂ CH	¹ 3
Mole % Pur.	Ref. Mo	lecul	ar C ₁₀ H ₂₂	Molecular Weight 142.	276	Ċ ₃ н ₇		
70 2 411.		Ref.		Weight 112.	Ref	<u> </u>		Ref
F. P. *C	Τ	Kei.		T	Ker		Г	1
F.P. 100%	 		dt/dP *C/mm		i	f to		1
B. P. °C	†		25°C	4. 7691	5	h .	l	}
760 mm 1 0 0	160. 94.7	2	BP t _e	0. 05150 0. 0367	5 5	f' to	 	+-
30	65.8	4	30 mm	0.7231	5	g' ' <u>°</u> K_]	1
10	43.8	5	ΔHm cal/g		\vdash	h'		
1	6.9	5	ΔHv cal/g	 	\vdash	m to		
Pressure mm 25°C	3.3674	5	25°C	77.33	5	n <u>*K</u> _	1	1
t _e	1172.	5	30 mm BP	74.00 63.06	5 5			_
Density			t_	61.07	5	m' to		
g/ml 20°C	0.741	2 2	-	61.00	5	n	1	
d 25 4 30	0.733	4	ΔHv/T _e	19.27	5	Surface tension		+
a	0.757	4	d 70 to		5 5	dynes/cm. 20°C	23.82	5
	-0.038	4	$\frac{1}{6}$ $\frac{1}{1}$ $\frac{175}{25}$ $\frac{6}{10}$		5	30	22.81	5
Ref. Index	1.417	2	e' 70 °C	0.0817	5	Parachor [P]	21,82	13
25	1.415	2	d _c g/ml v _c ml/g t _c °C	0, 242 4, 133	5	Parachor [P] 20°C	l	
30	1.412	4	tc °C	326.7	5	30		
_"C"	0.7486	4	P _c mm	15270.	5	40 Sugd	424.2	5
MR (Obs.) MR (Calc.)	48.3 48.38	5	PV/RT		\vdash	Exp. L.1.%/wt.		+-
(nD-d/2)	1.047	2	25°C	1.0000	5 5	u.		_
Dielectric	2.008	5	30 mm BP	1.0000 0.9480	5	Dispersion	97.	2
A 70 to	6.9710	5	te	0.9343	5	Flash Point °C Fire Point		
B	1508.	5	t _c	0.24	"	M Spec.	 	+
A* 70 to	1,47305	5	ΔHc kcal/m ΔHf	1		Ultra V.		
B* 187°C		5	ΔFf			X-Ray Dif. Infrared		
к — — —			Viscosity			Solubility in +		+
t _k	1		centistokes 7 °C			Acetone		1
t _x ; •c			•		1 1	Carbon tet. Benzene		
A' 25 to B' 70 °C		5				Ether		
B' ∟ 70°C	1711.9	5	B ^V to		\Box	n-Heptane Ethanol	l	
A'* 25 to	1,81589	5	AV °C	1		Water		
B'* 70°C	1611.5	5	(B ^V) to	1		Water in	ļ	┼
Ac 215 to	7.7600 2250.	5	(A ^V) °C					
Bc Ltc_°C	302.	5	c _p liq. °K					
Cryos. A°			c _p vap. °K					
consts. B°		igspace	•					
t _e °C	177.78	5	c _v vap.	L			<u> </u>	<u> </u>
$T_R = 0.8$						grams/100 grai		ıt
	ES: 1-Dow	2-AI		Calc. from det	t. dat	ta 5-Calc. by for	mula	
SOURCE:			PI					
PURIFICAT			PI					
LITERATU	RE REFERE	NCES	i:					

NAME	3-Ethyl-2-	meth	ylheptane	STRUCTURAL FORMULA				
					C ₂ H ₅			
Mole % Pur.	Ref. Mo	lecul	ar C ₁₀ H ₂₂	сн ₃ сн сн (с	CH ₂) ₃ CH ₃			
70 1 41.	1 1 10.	Ref.		Weight 142.2	Ref.			Ref.
F. P. °C			dt/dP	†				
F.P. 100	6	1	°C/mm			f to g =°K_		
B. P. °C			25°C	6.1341	5	h i		
760 mm	166.	2	BP t _e	0.05208	5 5	f' to		
100 30	99. 9 70. 7	4	30 mm	0.7320	5	g'° <u>K</u>		1
10	48.4	5	ΔHm cal/g	0.1320	+ -	h'		
1	11.1	5		+		m to		
Pressure	2 5/02	_	ΔHv cal/g 25°C	79.08	5	n		
mm 25°C	2,5603	5	30 mm	75,22	5	° '		
Density	1107.	-	BP	64.06	5	m' to		
g/ml 20°	0.746	2	t _e (d, e)	61.97	5	n' K	1	
dt 25	0.742	2	ΔHv/T _e	19.26	5	o'		
4 30	0.738	4	d 75 to	· · · · · · · · · · · · · · · · · · · ·		Surface tension		
a b	0.762	4	e 185 °C		5 5	dynes/cm. 20°C	24.47	5
	-0.038	4	d' 25 to	81.19	5	8 30 40	23.44	5
Ref. Inde:		2	e' 75 °C	·	5	Parachor [P]	1 22.13	+
ⁿ D 20°	1.416	2	d _c g/ml	0. 242	5	20°C		
30	1.413	4	vc ml/g tc °C	4,129 335.8	5	30		
"C"	0.7452	4	P _c mm	15514.	5	40 Sugd	424.2	5
MR (Obs. MR (Calc		2	PV/RT	 		Exp. L.1.%/wt.		
(nD-d/2)	48.38 1.045	5 2	25°C	1.0000	5	u.	_	
Dielectric		5	30 mm BP	1,0000 0,9475	5	Dispersion	96.	2
A 75 to		5	t_	0.9333	5	Flash Point °C Fire Point		
B 225°C	1530.	5	t _c	0.24	5		-	
c	208.	5	ΔHc kcal/m ΔHf			M. Spec. Ultra V.		
A* 75 to B* 195 °C		5 5	ΔFf			X-Ray Dif.		
K - 3	- 1137.7	١	Viscosity			Infrared		\vdash
·	_		centistokes			Solubility in Acetone		
t _k -to			η °C			Carbon tet.		
A' 25 to	ı	5				Benzene		
B' 75 °C		5			-	Ether n-Heptane		
C'	226.	5	Bv to			Ethanol		
A'* 25 to		5		_		Water Water in		
B'* 75 °C		5	(B ^V) to			.,, 8601 711	 	\vdash
Acl 225 to		5	(A ^V) °C		<u> </u>	1		
Bc tc °C	310.	5	c _p liq. °K					
Cryos. A			c _p vap. °K					
t _e °C	184.59	5	c, vap.					
$T_R = 0$.	i i		L	1	L	+ grams/100 gra	ms solven	ıt.
		2-A	PI 3-Lit. 4-	Calc, from de	t. da	A. 5 C-1. 1. 6-		
SOURCE:		AI						
PURIFICA	TION:	AI						
	JRE REFERE		· · · · · · · · · · · · · · · · · · ·					
								

							N o. 98	
NAME	4-Ethyl-2-	meth	nylheptane			STRUCTURAL I	ORMULA	4
						C.1. C.1. C.1. C.1.	1CT \ CT	
	7.//			N-1		CH ₃ CH-CH ₂ CH- CH ₃ C ₂ H ₄		⁻¹ 3
Mole % Pur.	Ref. Mo	rmul	ar C ₁₀ H ₂₂	Molecular Weight 142.2	76	0113 02115	,	
		Ref.			Ref			Ref.
F.P. °C			dt/dP			f to		
F.P. 100%			*C/mm 25*C	5 4013	5	g <u>*K</u> _		
B. P. °C 760 mm	160.	2	BP	5.4812 0.05146		h		1
100	94.7	4	t _e	0.0367	5	f' to		l
30 10	65.9 43.8	4 5	30 mm	0.7229	5	g'K_		
ĭ	7.0	5	∆Hm cal/g			h'		\vdash
Pressure			ΔHv cal/g 25°C	77, 38	5	m to		
mm 25°C	3.3554 1172.	5	30 mm	74.04	5	•		
t _e Density	1112.	-	BP	63.10	5 5	m' to		
g/ml 20°C	0.736	2	te te (d, e)	61.11	5	n' °K_		
d ₄ 25	0.732	2 4	ΔHv/Te	19.28	5	o'		
	0, 728	4	d 70 to		5	Surface tension		١.
a b	-0.038	4		0.1163	5	dynes/cm. 20°C	23.19 22.19	5
Ref. Index			d' 25 to		5 5	40	21.23	5
n _D 20°C	1.413	2	d g/ml	0.240	5	Parachor [P]		
30	1.411 1.408	2 4	d g/ml vc ml/g	4.163	5	20°C		
"C"	0.7468	4	1c C	325. 8	5	40	l	1
MR (Obs.)	48.2	2	P _c mm	15136.	5		424.2	5
MR (Calc.)		5 2	PV/RT 25°C	1.0000	5	Exp. L.l.%/wt. u.		1
(nD-d/2) Dielectric	1.045	5	30 mm	1.0000	5	Dispersion	96.	2
A 70 to	1.996 6.9737	5	BP t _e	0.9480 0.9 3 43	5 5	Flash Point °C		
B 1215°C	1509.	5	tc	0.24	5	Fire Point		├ ──
С	209.	5	ΔHc kcal/m			M Spec. Ultra V.		
A* 70 to B* 185 °C	1.47578 1419.2	5	ΔHf ΔFf	ł		X-Ray Dif.		1
K Lie C	1417.2]]	Viscosity			Infrared Solubility in +		
t to	ŀ		centistokes 7°C			Solubility in †		ļ
t _x to		1	7 ℃			Carbon tet.		1
A' 25 to	7, 33416	5				Benzene Ether		
B' ∟ 70 °C	1713. 227.	5	B ^V to	 	\vdash	n-Heptane		
A'* 25 to	1.81869	5	B' to	1		Ethanol Water		
B' ≠ 70 °C	1612.6	5	(B ^V) to	-		Water in		<u> </u>
Ac 215 to	7.7575	5	(A ^V) °C	ł				
Bc tc_°C	2243. 301.	5	c _p liq. °K	1	T			
Cryos. A°	301.	1	_					
consts. B°			c _p vap. °K					1
t _e °C	177.77	5	c _v vap.					
$T_{\mathbf{R}} = 0.81$	T _C					+ grams/100 gran	ns solven	t
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-(Calc. from de	t. dat	a 5-Calc, by form		
SOURCE:		A.	PI					
PURIFICAT			PI					
LITERATUR	RE REFERE	CES	:					

No. 99 5-Ethyl-2-methylheptane STRUCTURAL FORMULA NAME CH3CH (CH2)2CH CH2CH3 Mole Ref. Molecular Molecular Ċнз $C_{10}H_{22}$ Ċ₂H₅ Weight 142, 276 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g <u>°K</u> 4.6975 5 B. P. °C h BP 0.05150 5 760 mm 159.7 2 0.0367 5 ſ١ to 100 94.4 4 g' °<u>K</u> 65.5 30 4 30 mm 5 0.7229 10 43.5 5 h* ∆Hm cal/g 6.6 to m AHv cal/g Pressure ۰ĸ n 25°C 77.20 mm 25°C 3.42450 o 30 mm 73.90 1171. t_e 5 62.97 5 BP m' to Density 60.98 5 n' °K g/ml 20°C te (d, e) 0.736 60.91 5 o' $\mathbf{d_4^t}$ 25 0.732 AHv/Te 19.25 5 30 0.728 4 Surface tension 70 to 81.51 5 0.752 a 4 dynes/cm. 20°C 23.19 5 75 °C 25 to 175 0.1161 5 ь -0.038 4 30 22.19 ā٦ 79.24 40 21.23 5 e¹ Ref. Index 70 0.0815 n_D 20°C 1.4134 [P] Parachor d_c g/ml 0.240 5 25 1.4109 2 20°C vc ml/g t_°C 4.175 30 1.4086 4 30 ^tc 325.4 40 "C" 0.7475 4 P_c mm 15082. 5 Sugd 424.2 5 MR (Obs.) 48.2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 (nD-d/2)u. 1.045 2 30 mm 1.0000 5 Dispersion 96. 2 0.9480 Dielectric 1.998 5 BP 5 Flash Point C 0.9343 5 A 70 to 6.9681 5 Fire Point 0.24 5 B (215 °C 1506. M. Spec. AHc kcal/m С 209. 5 Ultra V. A* 70 to B* 185 °C ΔHf 1.47051 5 X-Ray Dif. ΔFf 1416.2 Infrared ĸ Viscosity Solubility in centistokes Acetone Carbon tet. °C ^tx_ Benzene 25 to 7, 32861 Ether B' |_ 70 °C 1709.8 n-Heptane B_v | 227. 5 to Ethanol °C Water A'* 25 to B'* 70 °C 1.81332 5 1609.4 Water in (B^V) to Acl 215 to 7.7503 5 (A^V)| °C Bc tc °C 2238. c_p liq. °K 301. 5 Cryos. A c_p vap. ٩K consts. B° c vap. te °C 177.45 $T_R = 0.81 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							_	No. 10	0
NAME	3-Et	hyl-3-	met	thylheptane			STRUCTURAL	FORMUL.	A.
						\dashv	C ₂ H ₅		
							сн ₃ сн ₂ с - (с	H_)_CH_	
Mole	Ref.	Mole	cul	ar .	Molecular		cH ₃	2/3	
% Pur.		For			Weight 142.2	76	3		
		F	Ref.			Ref.			Ref
F. P. °C	<u> </u>			dt/dP			f to		ì
F.P. 100%	L			°C/mm	5 5014	ا ۔ ا	g <u> •K</u> _	ļ	
B. P. ℃	1,42.0		٠, ١	25°C BP	5.5814 0.05189	5	h .		
760 mm 100	163.8		2 4	t _e	0.03670	5	f' to		
30	68.89)	4	30 mm	0.7289	5	g'	1	
10 1	46.67		5	ΔHm cal/g			h'		
Pressure	/		-	ΔHv cal/g	 		m to		ı
mm 25°C	2.83	8	5	25°C	78.41	5	n <u>*K</u> _	1	
t _e	1183.2		5	30 mm BP	74.75 63.70	5	<u> </u>		<u> </u>
Density				t_	61.65	5	m' to		
g/ml 20°C	0.75		2 2	e (4, 6)	61.57	5	n' ' <u>°K</u> -	†	
d ₄ 25	0.74		5	ΔHv/T _e	19.26	5		ļ ———	+
a	0.76		5	d 68 to		5	Surface tension dynes/cm, 20°C	25.01	5
b	-0.03		5	$\begin{vmatrix} \mathbf{e} \\ \mathbf{d} \end{vmatrix} = \begin{vmatrix} 182 \\ 20 \end{vmatrix}$		5	3 0	24.01	5
Ref. Index				e' 68 °C		5	40	23.04	5
n _D 20°C	1.42		2	d _c g/ml	0.247	5	Parachor [P] 20°C		
30	1.41		4	V mi/g	4.045	5	30		
"C"	0.74	159	4	, c	334.	5	40		_
MR (Obs.)	48.08	1	2	P _c mm	15790.	5		424.2	5
MR (Calc.)			5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		
(nD-d/2)	1.04		2	30 mm	1.0000	5	Dispersion	97.	2
Dielectric	2.01		5	BP	0.9480	5	Flash Point °C		t
A 68 to B <u>223</u> °C		739	5 5	te t _c	0.9340 0.24	5	Fire Point		
c =====	208.		5	∆Hc kcal/m		 	M Spec.	1	1
A* 68 to	1.47	24	5	ΔHf			Ultra V. X-Ray Dif.		
B* 200 °C	1431.2	1	5	ΔFf		-	Infrared		
K — — —			l	Viscosity centistokes			Solubility in +		T
t _k to		1		7 °c			Acetone Carbon tet.		
x '				•			Benzene		
A' 20 to B' 68 °C		114	5				Ether		
c	226.0		5	B ^V to			n-Heptane Ethanol		
A'* 20 to	1.81	47	5	AV C			Water		
B'* 68 °C	1625.2		5	(B ^V) to]		Water in	 	+-
Ac 223 to	7.80	61	5	(A ^V) °C					1
Bc Ltc_°C	2323.9		5	c _p liq. °K				[1
Cryos. A°	1		$\overline{}$	c _p vap. °K					
consts. B°	<u></u>			İ					
t _e °C	182.12	: T	5	c _v vap.					
$T_{\mathbf{R}} = 0.82$	2 T _c						grams/100 grai	ms solven	ıt_
REFERENC	ES: 1-D	ow 2	-AF	PI 3-Lit, 4-0	Calc. from det	t. da	ta 5-Calc. by for		
SOURCE:			AI						
PURIFICAT	ION:		AI	PI					
LITERATUI	RE REF	EREN	CES	:					

							No. 10	<u>,, </u>
NAME	4-Ethy	1-3-mei	hylheptan e			STRUCTURAL	FORMUL	A
						C ₂ H		
						сн ₃ сн ₂ сн сн	(CH ₂) ₂ CF	1,
Mole % Pur.	Ref.	Molect	la C ₁₀ H ₂₂	Molecular Weight 142.2	276	с́н,		•
-/0 T U I .		Re		Weight 142.7	Ref.			Ref.
F. P. ℃			dt/dP		Ker.	1	T	101.
F.P. 100%			°C/mm			f to		
B. P. °C			25°C	6. 3987	5	h		
760 mm 100	167.	2	BP t _e	0.05218 0.0367	5	f' to		\vdash
30	100.8	4	30 mm	0.7335	5	g'° <u>K</u>		
10	49.2	5	ΔHm cal/g	1	+ -	h'		1
1	11.7	5	ΔHv cal/g		\dagger	m to		
Pressure mm 25°C	2.44	54 5	25°C	79.37	5	n •K	-	
t _e	1192.	5	30 mm BP	75.43 64.27	5			↓
Density		. T	t_	62.17	5	m' to		
g/m1 20°C _t 25	0.75		t _e (d, e)	62.08	5	", <u> </u>	1	
d ₄ 25	0.74		ΔHv/T _e	19.27	5	Surface tension		+
a L	0.76		d 75 to		5	dynes/cm. 20°C	25.40	5
b D-C T-d	-0.03	8 4		81.49	5	8 30 40	24.34	5
Ref. Index	1,42	2 2	e' 75 °C	+	5	Parachor [P]	1	Ť
– <u>2</u> 5	1.41	9 2	d g/ml vc ml/g	0.245 4.081	5	20°C		
30	1.41		tc °C	338.4	5	30 40		
	0.74		P _c mm	15764.	5		424.2	5
MR (Obs.) MR (Calc.)	48.0 48.38	5	PV/RT 25°C	1 0000	_	Exp. L.1.%/wt.		
(nD-d/2)	1.04	6 2	30 mm	1.0000	5	u. Dispersion	96.	2
Dielectric	2.02	2 5	BP	0.9480	5	Flash Point °C	100	Ť
A 75 to	6.97 1534.	94 5	t _e	0.9338 0.24	5	Fire Point		ļ
B 1 230 ℃ C	207.	5	ΔHc kcal/m	+	+	M. Spec.		
A* 75 to	1.47	482 5	ΔHf		1	Ultra V. X-Ray Dif.		
B * 195 °C K	1443.3	5	ΔFf		₩	Infrared		
С			Viscosity centistokes		1	Solubility in +		
t _k to			7 °C			Acetone Carbon tet.		
t _x °C A' 25 to	7, 33	464 5	4			Benzene		
B'	1738.4	5	-		1	Ether n-Heptane		
C'	225.	5	B ^V to A ^V °C			Ethanol		
A'* 25 to B'* 75 °C	1.81 1637.9	669 5		-		Water Water in		
Acl 230 to	7.83	+	-					1
Bc _l t _c ℃	2367.	5		 	 	-		
Cc — —	313.	5	P -					
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	185.76	5	c, vap.					
$T_{\mathbf{R}} = 0.82$			1 -		1	grams/100 gra	m o cola:-	
REFERENC		ow 2-	API 3-Lit. 4	-Calc. from de	t. da	ata 5-Calc. by for		ııt
SOURCE:			PI	u		5 CLIC. D, 101		
PURIFICAT	ION:		PI					
LITERATUR								
			- 					
			T					

NAME								No. 10	·Z
	5 - E	thyl-3-	me	thylh e ptane			STRUCTURAL F	ORMUL	A
							CH CH CH CH C	u cu c	- 1.1
Mole % Pur.	Ref.	Mole Form		C ₁₀ H ₂₂	Molecular Weight 142.2	76	ch ₃ ch ₂ ch ch ₂ c ch ₃ c	2 ^H 5	′**3
			Ref.			Ref			Ref.
F.P. °C				dt/dP			f to		Ī
F.P. 1009	•			°C/mm	4 0000	ا ۔ ا	g <u>•K</u> _		1
B. P. °C 760 mm	161.		2	25°C BP	4.9880 0.05160	5	h		
100	95.6		4	t _e	0.0367	5	f' to		
30 10	66.6 44.6	ı	4 5	30 mm	0.7246	5	g' 'K_		
1	7.6		5	ΔHm cal/g			h¹		₩
Pressure	1	$\neg \uparrow$	\neg	ΔHv cal/g	77 (2	ا ۔ ا	m to to		
mm 25°C	3.20	076	5	25°C 30 mm	77.62 74.20	5 5	0		
te	1175.		5	BP	63, 23	5	m' to		+
Density g/ml 20°C	0.74	13	2	te t (d, e)	61.23	5 5	n'		
dt 25	0.73	39	2		19. 27	5	0'		1
4 3 0	0.73		4	ΔHv/T _e	81.95	5	Surface tension		
a b	-0.03		4	_e <u> </u> 179 °C		5	dynes/cm. 20°C	24.08 23.06	5
Ref. Index		-	-	d' 25 to	79.68 0.0821	5 5	30 40	22.07	5
n _D 20°C		6	2		0.0821	5	Parachor [P]		_
25 30	1.41		2	d _c g/ml v _c ml/g	4, 120	5	20°C		
"C"	1.41		4	tc °C	328.	5	30 40		
	0.74	149	4	P _c mm	15349.	5	1 . 1	424.2	5
MR (Obs.) MR (Calc.		,	2 5	PV/RT			Exp. L.l.%/wt.		
(nD-d/2)	1.04		2	25°C 30 mm	1.0000	5	u. Dispersion	96.	2
Dielectric	2.00)5	5	BP	0.9480	5	Flash Point °C	70.	+-
A 67 to			5	t _e	0.9342 0.24	5	Fire Point		
B 214 °C	208.		5	t _c AHc kcal/m	0.24	-	M Spec.		
A* 67 to	+	-	5	ΔHc Real/H	ĺ	i i	Ultra V.		
B* 189 °C		1	5	ΔFf			X-Ray Dif. Infrared		
K	1	-		Viscosity			Solubility in +		†
the Total	5		ı	centistokes 7°C			Acetone		
t _{x 1} *(_ i			•	1		Carbon tet. Benzene		
A' 25 to B' 67 °C			5				Ether		
č, – <u></u>	226.		5	B ^v l to			n-Heptane Ethanol		ŀ
A'* 25 to		417	5	A ^V I °C			Water		
B'* 67 °C	1617.6		5	(B ^V) to	1		Water in		+
Ac 214 to	7.77 C 2266.		5	(A ^V) °C					
Bc tc_°C	304.		5	c _p liq. °K					1
Cryos. A				c _p vap. °K	}				
te °C	178.92		5	c _v vap.					
$T_R = 0.8$	1				L		grams/100 gran		
REFEREN		ow 2	- AP	PI 3-Lit. 4-0	alc. from det	det			15
SOURCE:			AF		uei		a 3-care, by fort	·······a	
PURIFICA	TION:		AF						
LITERATU		ERENC							
				•					

NAME	3-Ethyl-4-	meth	ylheptane		STRUCTURAL FORMULA			
						сн ₃ сн ₃ сн ₂ сн сн (сн ₂) ₂ сн ₃		
Mole % Pur.		lecul rmula		Molecular Weight 142.2	76	C ₂ H ₅		
		Ref.		T	Ref.	Re		
F. P. ℃			dt/dP			f to		
F. P. 100	70	 	°C/mm 25°C	6. 3987	5	g° <u>K</u>		
B. P. °C 760 mm	167.	2	BP	0.05218	5 5	h		
100 30	100.8	4	t _e 30 mm	0.0367 0.7335	5	f' to to o'K		
10	49.2	5	ΔHm cal/g	0.7333		h'		
1	11.7	5	ΔHv cal/g	<u> </u>	-	m to		
Pressure mm 25°C	2.4454	5	25°C	79.37	5	"		
t _e	1191.	5	30 mm BP	75.43 64.20	5			
Density g/ml 20°	C 0.753	,	te (d.s)	62.10	5	m' to cK		
at 25	0.749	2 2	t _e (d, e) ΔHv/T _e	62.01	5	0'		
	0.745	4	d 75 to	83,83	5	Surface tension		
a b	0.769 -0.0 ₃ 8	4 4	_e_ _1 <u>85</u> °C	0.1175	5	dynes/cm. 20°C 25.40 5		
Ref. Inde:		 	d' 25 to e' 75 °C	81.49 0.0848	5	40 23.31 5		
ⁿ D 20°		2 2	d _c g/ml	0, 245	5	Parachor [P]		
30	1.417	4	V _C m1/g	4.084	5	20°C 30		
"C"	0.7450	4	t _c °C P _c mm	338.3 15750.	5	40 Sugd. 424. 2 5		
MR (Obs.		2	PV/RT	13130.		Exp. L.1.%/wt.		
MR (Calc (nD-d/2)	.) 48.38 1.046	5 2	25°C	1.0000	5	u.		
Dielectric	2.022	5	30 mm BP	1.0000 0.9470	5	Dispersion 96. 2 Flash Point ⁶ C		
A 75 to		5	t _e	0.9327 0.24	5	Fire Point		
B 1230 °C	5_ 1534. 207.	5	t _c ΔHc kcal/m		Ť	M. Spec.		
A* 75 to	1.47661	5	ΔHf ΔFf			Ultra V. X-Ray Dif.		
B*[195 °C	C 1443.8	5	Viscosity	 		Infrared		
c	_		centistokes			Solubility in TACetone		
t _k to			η °c			Carbon tet.		
A' 25 to		5				Benzene Ether		
B' _75 °	225.	5	B _v to	· · · · · · · · · · · · · · · · · · ·		n-Heptane Ethanol		
A1* 25 to	1.81669	5	AV I C	_[Water		
B'* 75 °	C 1637.9	5	(B ^V) to			Water in		
Ac 230 to	7.8325 C 2366.	5	(A ^V) °C	_	L			
Cc -	313.	5	c _p liq. °K					
Cryos. A consts. B			с _р va p. °К					
t _e °C	185.70	5	c _w vap.					
$T_{\mathbf{R}} = 0.8$						grams/100 grams solvent		
REFEREN	CES: 1-Dow			Calc. from de	t. da	ata 5-Calc. by formula		
SOURCE:		AI						
PURIFICA		AI						
LITERATI	URE REF E RE	NCE	5:					

							No. 104	4
NAME	4-Ethyl-	4-me	thylheptane			STRUCTURAL 1	FORMULA	
						C ₂ H ₅		
					\neg	СН ₃ (СН ₂)2 - (С	CH ₂) ₂ CH ₃	
Mole % Pur.		lecul rmul		Molecular Weight 142.2	76	ċн ₃		
		Ref.			Ref			Ref.
F.P. °C	T		dt/dP	1		f to		
F.P. 1007			°C/mm	/ 2007	_	gK_		
B. P. *C 760 mm	167.	2	25°C BP	6.3987 0.05218		h		
100	100.8	4	t _e	0.0367	5	f' to		
30 10	71.5 49.2	5	30 mm	0.7335	5	h' ' h		ĺ
1	11.7	5	ΔHm cal/g	_	L	m to		
Pressure mm 25°C	2 4454	5	ΔHv cal/g 25°C	79.37	5	n•K_		
t _e	2.4454 1191.	5	30 mm BP	75.43 64.20	5	<u>°</u> į.		
Density			I I	62.10	5	m' to		
g/ml 20°C	0.752 0.748	2 2	te (d, e)	62.01	5	o' '		ĺ
d ₄ 25 30	0.744	4	ΔHv/T _e	19.25	5	Surface tension		
a b	0.768	4	d 72 to e 186 °C		5 5	dynes/cm. 20°C	25.27	5
Ref. Index	-0.038	+	d' 25 to	81.49	5 5	30 40	24.21 23.18	5
n _D 20°0	1.421	2		0.244	5	Parachor [P]		
25 30	1.419 1.416	2 4	d g/ml v ml/g	4.091	5	20°C 30		
"C"	0.7443	4	t _c ·C	338.	5	40		_
MR (Obs.)		2	P _c mm PV/RT	15717.	2		424.2	5
MR (Calc. (nD-d/2)	1.045	5 2	25°C	1.0000	5	Exp. L.1.%/wt. u.		
Dielectric		5	30 mm BP	1.0000	5	Dispersion	97.	2
A 72 t	6.9794	5	te	0.9327	5	Flash Point °C Fire Point		
B [228 °C	2 1534. 207.	5	t _c	0.24	5	M Spec.		
A* 72 to		5	ΔHf			Ultra V. X-Ray Dif.		
B* 196 °C		5	ΔFf	ļ	L	Infrared		
K C		1	Viscosity centistokes	ļ		Solubility in +		
1 to 1 to			η •c			Acetone Carbon tet.		
A' 25 to		5	ŀ			Benzene Ether		
B' _ 72 °C	1738.4	5	B ^V to		-	n-Heptane		
A'* 25 to	225.	5	B ^V to		İ	Ethanol Water		
	1.81669	5	(B ^V) to	-		Water in		
Ac 228 to	7.8314	5	(A ^V) °C					
Bc tc_°C	2364. 312.	5	c _p liq. °K					
Cryos. A			c _p vap. °K					
consts. B		L.	Р					
t _e °C T _R = 0.8	185.70	5	c _v vap.	<u> </u>		l		<u>L_</u>
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Tale from de		grams/100 gran	ns solven	t
SOURCE:	1-DOM		PI 3-LR. 4-0	Jaic. Irom del	. da	ta 5-Calc. by for	nula	
PURIFICA	TION:		PI					
	RE REFERE	NCES	;;					
1								
1								
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L								

NAME	2, 2, 3-Trir	nethy	lheptane		STRUCTURAL FORMULA CH3CH3			
Mole % Pur.		lecula mula	ar C ₁₀ H ₂₂	Molecular Weight 142.2	76	сн ₃ с -сн-(сы сн ₃	H ₂) ₃ CH ₃	
		Ref.			Ref.		F	Ref.
F. P. ℃			dt/dP			f to	ŀ	
F.P. 100%			°C/mm		1 1	g °K	- 1	
B. P. °C			25°C	4. 37554	5	h l		
760 mm	158.	2	BP	0.05134	5			
100	92.9	4	t _e	0.0367	5	f' to	İ	
30	64.2	4	30 mm	0.7204	5_	g'° <u>K</u>		
10	42.2	5	∆Hm cal/g			h!		_
	3.5	-	ΔHv cal/g			m to	1	
Pressure mm 25°C	3,7001	5	25°C	76.71	5	n •K		- 1
t _e	1167.	5	30 mm	73.55	5			
Density	+	-	BP	62.68	5	m' to		
g/ml 20°C	0.742	2	t _e (d, e)	60.65	5	n' K_		
d ₄ 25 30	0.738	2	ΔHv/T _e		5	o'	İ	
4 30	0.734	4		19.25	-	Surface tension		
a	0.758	4	d 65 to	80.98	5	dynes/cm. 20°C	23.95	5
ь	-0.038	4	e 175 °C to	0.1158 78.73	5 5	8 30	22.93	5
Ref. Index			e' 65 ℃	0.0808	5	40	21.94	5
n _D 20°C	1.417	2	d _c g/ml	0,243	5	Parachor [P]		
45	1.414	2	v ml/g	4.122	5	20°C		
30	1.412	4	vc ml/g tc °C	324.1	5	30	1	
"C"	0.7476	4	P _c mm	15244.	5	Sugd.	424.2	5
MR (Obs.)	48.2	2	PV/RT	 	\vdash	Exp. L.1.%/wt.		
MR (Calc.)		5	25°C	1.0000	5	u.		
(nD-d/2)	1.046	2	30 mm	1.0000	5	Dispersion	97.	2
Dielectric	2.008	5	BP	0.9480	5	Flash Point C		
A 65 to	6.9658	5	t _e	0.9345 0.24	5	Fire Point		
B (210 °C	1499. 209.	5 5	t _c ΔHc kcal/m	+	1	M. Spec.		
			ΔHC RCa1/III			Ultra V.		
A* 65 to B* 185 °C	1.46977	5	ΔFf			X-Ray Dif. Infrared		
к — —			Viscosity			· · · · · · · · · · · · · · · · · · ·		
c		1 1	centistokes			Solubility in * Acetone		
t _k to			ŋ ° ℃			Carbon tet.		
L _x						Benzene		
A' 25 to B' 65 °C	7.32751 1703.2	5 5				Ether		
C 55 C	227.	5	B ^v to			n-Heptane Ethanol		
A** 25 to	1.81284	5	B ^V to A ^V °C			Water		
B'* 65 °C	1602.8	5	(B ^V) - to			Water in		
Acl 210 to	7,7423	5	(A ^V) °C					
Bc tc °C	2224.	5			\vdash			
Cc	301.	5	c _p liq. °K					
Cryos, A°			c _p vap. °K					
consts. B°		Ш	I -					
te °C	175.52	5	c _v vap.	L				
$T_{R} = 0.81$	T _c					† grams/100 gran	ns solvent	:
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by form	nula	
SOURCE:		AF	PI				 	
PURIFICAT	ION:	AF	PI					
LITERATUI	RE REFERE	NCES	S:					
L								-

No. 106 STRUCTURAL FORMULA NAME 2, 2, 4-Trimethylheptane сн3 сн₃с - сн₂сн (сн₂)₂сн₃ Molecular C₁₀H₂₂ Ċнз Molecular ĊH₃ Mole Ref. Weight 142.276 % Pur Ref. Ref Ref. F.P. °C F.P. 100% dt/dP f °C/mm °K_ g 25°C 2.6734 5 B. P. °C h ВP 0.05125 5 760 mm 147.7 2 t_e 5 0.03733 f to 82.96 100 4 g' °K 30 54.52 4 30 mm 0.7113 5 10 32.86 h AHm cal/g -3,27 1 m to AHv cal/g Pressure °K n 72.51 25°C 6.4065 mm 25°C 30 mm 70, 29 5 t_e 1142.1 5 5 BP 60.0 m¹ to Density g/ml 20°C te (d, e) 58.21 °K 0.7275 2 'n 58,16 5 0 0.7237 $\mathbf{d_4^t}$ 25 2 ΔHv/T_e 18.93 5 30 0.71994 Surface tension d 54 76.32 5 0.7427 to 4 dynes/cm. 20°C 22.13 0.1105 <u>l 16</u>4 ь -0.0376 4 21,22 5 30 ď to 74.39 5 5 20.33 40 Ref. Index 0.0752 5 $\mathbf{n}_{\mathbf{D}}$ 20°C 1.4092 2 [P] dc Parachor g/ml 0.231 5 25 1.4070 20°C v_c ml/g 4.329 5 30 1.4046 4 30 °C 309. tc 40 "C" 0.7490 4 Pç 14147. 5 5 mm Sugd. 424.2 MR (Obs.) 48.37 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 48.38 25°C 1.0000 5 (nD-d/2) 1.0454 2 30 mm 1.0000 5 Dispersion 100. 2 Dielectric 1.986 5 BP 0.9510 Flash Point °C 0.9383 5 6.8795 A | 54 to t_e 5 Fire Point 0.24 5 1434.1 M Spec. C 211. 5 AHc kcal/m Ultra V. ΔHf 54 **to** A* 1.3879 5 X-Ray Dif. ΔFf 174°C 1345.3 Infrared Viscosity Solubility in c centistokes Acetone to Carbon tet. ۰c Benzene 25 **to** A' 7,2458 Ether В' 1635.1 <u>54 °C</u> n-Heptane $\mathbf{B}^{\overline{\mathbf{v}}}$ 5 229. to Ethanol A I °C 25 to Water A!* 1.7359 5 Water in B'* 54 °C 1535.2 5 (BV) to Ac | 195 to 7,5887 5 (AV) °C Bc tc_°C 2068.0 liq. °K Сp 292.5 Cc Cryos. Aº cp vap. °K consts. B° t_e ℃ c_v vap. 164.30 5 $T_{R} = 0.80 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

								No. 1	.07
NAME	2, 2, 5	-Trir	neth	lheptane		i	STRUCTURAL	FORMUI	LA
							СН3		
 						-	CH ₃ C - (CH ₂) ₂ C	н сн сн	
Mole	Ref.	Mo	lecul	ar c u	Molecular				3
% Pur.		For	mula	C ₁₀ H ₂₂	Weight 142.2	76	Ċн ₃ Ċ	Н ₃	
	-		Ref.			Ref.			Ref.
F. P. ℃				dt/dP			f to		1
F.P. 100%				°C/mm		_	g		1
B. P. ℃	1			25°C B P	2.900 0.05036	5	h		l
760 mm	148.		2	te	0.0366	5	f [†] to		T
30	56.0		4	30 mm	0.7055	5	g' <u>°K</u>		
10	34.5		5	ΔHm cal/g	1	+	h'		1
1	-1.5		5	ΔHv cal/g	 	+-	m to		T
Pressure mm 25°C			ا ۔ ا	25°C	73.88	5	n		
mm 25°C	5.79	708	5	30 mm	71.52	5	° '		1_
Density	+		Ì	BP	61.21 59.42	5	m' to		
g/ml 20°C	0.72	26	2	t _e (d, e)	59.38	5	n' <u>*K</u>	4	
dt 25	0.72		2	ΔHv/T _e	19.32	5			
	0.71		4	d 56 to	77.79	5	Surface tension		
a b	-0.0		4	e 164 °C		5	dynes/cm. 20°C	21.95	5
Ref. Index	+	<u>, </u>	<u> </u>	d' 25 to		5	8 30 40	20.99	5
n _D 20°C	1.40	9	2	e' 56 °C	+	5	Parachor [P]	} 	_
25	1.40		2	d g/ml	0.239 4.176	5	20°C		
30	1.40		4	vc ml/g tc °C	308.	5	30		
"C"	0.75	502	4	P _c mm	14639.	5	40 Sugd	424.2	5
MR (Obs.)	48.5		2	PV/RT	+	┼──	Exp. L.1.%/wt.		+-
MR (Calc.) (nD-d/2)	48.38		5 2	25°C	1.0000	5	u.		
Dielectric	1.98		5	30 mm BP	1.0000	5	Dispersion	100.	2
A 56 to	6.95		5	t.	0.9520 0.9395	5	Flash Point °C		
B 192 °C		, 32	5	t _c	0.24	5	Fire Point		+
c	211.		5	∆Hc kcal/m			M. Spec. Ultra V.		
A* 56 to	1.45	980	5	ΔHf ΔFf			X-Ray Dif.	ļ	
B*[174 °C K	1372.3		5		 	├	Infrared		
c c	İ			Viscosity centistokes			Solubility in +		
t _k _ to				η °c			Acetone Carbon tet.		
'x	İ., .						Benzene		
A' 25 to	7.32	2254	5				Ether	l	
B' 1_56 °C	1665.3		5	B _v to			n-Heptane Ethanol		
A₩ 25 to	1,81	153	5	A l °C			Water		
B'* 56 °C	1565.1		5	(B ^V) to	-		Water in		ļ
Ac 192 to	7.65	49	5	(A ^V) °C	1	1			
Bc tc °C	2085. 290.		5	c liq. °K			1		
Cc	270.		-						
Cryos, A° consts, B°				c _p vap. °K					
t _e °C	164 34	<u> </u>	5	c _v vap.					
$T_{\mathbf{R}} = 0.80$	164.36		ر ا	I . •	1	L	+ /122	L	
					<u> </u>		grams/100 gra		nt
REFERENC	ES: 1-L	νοw			Caic, from de	t. da	ta 5-Calc. by for	muia	
SOURCE:			AI						
PURIFICAT			AI						
LITERATU	RE REF	ERE	NCE	5:					
l									

							No. 10	8
NAME	2, 2,	6-Trimet	hylheptane			STRUCTURAL		
			, ,					-
						CH ₃	CII CII	
Mole	Ref.	Molecul		Molecular		сн ₃ с - (сн ₂) ₃	LH CH3	
% Pur.		Formul		Weight 142.2	76	CH ₃	Ċн ₃	
		Ref.			Ref			Ref.
F, P. *C	T		dt/dP			f to		T
F.P. 100%			°C/mm			g LK	ĺ	
B. P. *C			25°C BP	2.9204 0.05039	5	h		İ
760 mm 100	148.2	2 4	te	0.0366	5	f' to		
30	56.2	4	30 mm	0.7059	5	g' ' <u>•</u> K_	1	
10	34.6	5	ΔHm cal/g	 		h'		
1	-1.3	1 5	ΔHv cal/g	+	\vdash	m to		
Pressure mm 25°C	5.75	29 5	25°C	73.92	5	n <u>*K</u>		
t _e	1144.	- / 5	30 mm	71.54	5	0		
Density	 		BP te	61.22 59.43	5	m' to		
g/m1 20°C	0.71		te (d, e)	59.39	5	n' °K	-	1
dt 25 4 30	0.71		AHv/Te	19.31	5	<u> </u>	ļ	1_
a 30	0.71		d 56 to		5	Surface tension	21 17	-
b	-0.03				5	dynes/cm. 20°C	21.17	5
Ref. Index	<u> </u>		d' 25 to		5	40	19.39	5
n _D 20°C	1.40		d _c g/ml	0, 237	5	Parachor [P]	Ī —	
25 30	1.40		V_ml/g	4, 214	5	20°C 30		
"C"	0.75	-	tc °C	308.	5	40		l
MR (Obs.)	+		P _c mm	14510.	5	Sugd.	424.2	5
MR (Calc.)	48.56 48.38		PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	u. Dispersion	98.	2
Dielectric	1.97	6 5	BP	0.9520	5	Flash Point °C	70.	+-
A 56 to			t _e	0.9395 0.24	5	Fire Point		
B <u>[182 °C</u> C	211.	5 5	t _c ΔHc kcal/m	+	ļ -	M Spec.		1
A* 56 to	1.45		∆Hf			Ultra V.		
B* 175 °C		5	ΔFf			X-Ray Dif. Infrared	1	
к — — —	` <u> </u>		Viscosity			Solubility in +	 	+
,	-		rentistokes °C	ļ		Acetone		}
tx C			7			Carbon tet. Benzene		1
A' 25 to	7.32					Ether		
B' ∟ 56 °C	1665.6 229.	5 5	B ^V to	+	\vdash	n-Heptane		
A'* 25 to	1.81		B to			Ethanol Water		
B'* 56 °C		5	(B ^V) to	-[Water in		
Ac 182 to	7.65		(A ^V) °C	1				1
Bc t _c °C	2083.	5	c _p liq. °K					
	289.	- 3		ŀ				
Cryos. A° consts. B°			c _p vap. °K				1	
t _e °C	164.59	5	c _v vap.			L <u>.</u>	L	<u>L</u>
$T_R = 0.8$						grams/100 gra		ıt
REFERENC	ES: 1-D			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		Al						
PURIFICAT		Al						
LITERATU	RE REFI	ERENCES	:					

r							No. 109	
NAME	2, 3, 3-Trin	nethy	lheptane			STRUCTURAL	FORMULA	
						CH ₃		
						снзсн с (с	H ₂) ₂ CH ₂	
Mole		lecul	C10H22	Molecular	_,	ိင်မ₃င်မ₃	23 3	
% Pur.	Fo	rmula		Weight 142.2	,			
		Ref.			Ref.	ļ	R	lef.
F. P. °C		ļ	dt/dP			f to		
F.P. 100%		\vdash	°C/mm 25°C	4.76914	5	g '° <u>K</u>		
B. P. °C 760 mm	160.	2	BP	0.05150	5	h		
100	94.7	4	t _e	0.0367	5	f' to		
30 10	65.8 43.8	4 5	30 mm	0.7231	5_	g' <u>K</u>		
10	6.9	5	∆Hm cal/g		L	h'		
Pressure			∆Hv cal/g			m to		
mm 25°C	3.3674	5	25°C 30 mm	77.33 74.00	5	-		
t _e	1172.	5	BP	63.07	5	-	ļ	
Density			t _e	61.08	5	m' to		
g/ml 20°C t 25	0.7488 0.7448	2 2	te (d, e)	61.00	5	0'		
d ^t 25 4 30	0.7408	4	ΔHv/T _e	19.27	5	Sunface Assuran	 	
a	0.7648	4	d 66 to	81.64	5	Surface tension dynes/cm. 20°C	24.84	5
b	-0.0 ₃ 80	4	$\begin{vmatrix} -\mathbf{e} \\ \mathbf{d} \end{vmatrix} - \begin{vmatrix} 178 & ^{\circ}\mathbf{C} \\ 25 & \mathbf{to} \end{vmatrix}$	0.1161 79.38	5	8 30	23.79	5
Ref. Index	. 4303	_	e' 66 °C	0.0817	5	40	22.78	5
ⁿ D 20°C	1.4202 1.4178	2 2	d _c g/ml	0.245	5	Parachor [P] 20°C		
30	1.4153	4	V mi/g	4.075 328.	5	30		
"C"	0.7461	4				40	424 2	_
MR (Obs.)	48.10	2	P _c mm	15519.	5		424.2	5
MR (Calc.)	48.38	5	P V/RT 25°C	1.0000	5	Exp. L.1.%/wt.	!	
(nD-d/2)	1.0458	2	30 mm	1.0000	5	Dispersion	96.	2
Dielectric A 66 to	2,017	5	BP t _e	0.9480 0.9343	5 5	Flash Point C		
	6.9710 1508.	5	t c	0.24	5	Fire Point		
c '	209.	5	∆Hc kcal/m			M. Spec. Ultra V.		
A* 66 to	1.47305	5	∆Hf ∆Ff	i		X-Ray Dif.		
B*[_188 °C	1418.2	5	Viscosity			Infrared		
c			centistokes			Solubility in +	į	
t _k to			η °C			Acetone Carbon tet.		
'x		1_1				Benzene		
A' 25 to B' 66 °C	7.33132 1711.9	5 5				Ether		
c, '- 💆 -	227.	5	BV to			n-Heptane Ethanol		
A'* 25 to	1.81589	5		!		Water		
B'* 66 °C	1611.5	5	(B^V) to			Water in		
Acl 214 to	7.7679	5	(A ^V) °C					
Bc t _c °C	2261. 304.	5	c _p liq. °K					
Cryos. A°		+	1	1				
consts. B°			c _p vap. °K	İ				
t _e °C	177.78	5	c _v vap.	ĺ				
$T_{R} = 0.81$	T _c			1		grams/100 gra	ms solvent	
REFERENCE		2-A	PI 3-Lit. 4-0	Calc. from de	t. da			_
SOURCE:		AF						
PURIFICATI	ON:	AF	· · · · · · · · · · · · · · · · · · ·					
LITERATUR								
	KUFURE	. TO ES	••					

_									No. 11	.0
NAME		2,3,	4 - T r	imet	hylheptane			STRUCTURAL		
							\neg	CH ₃		
		Т					\dashv	сн,сн сн сн	(CH-)-CH	I.
Mole		Ref.	Mo	lecul	ar C ₁₀ H ₂₂	Molecular	١,	cн ₃ сн		-3
% Pur.	_		Fo	rmul	a 10 22	Weight 142.2	_	3 3	3	la . c
	_			Ref.		T	Ref		т	Ref.
F.P. °C F.P. 1009	4			-	dt/dP *C/mm	ļ		f to		1
B.P. °C	+				25°C	5.4068	5	g ' <u>*</u> !	-	
760 mm		163.		2	BP	0.05179	5 5			+-
100 30		97.3 68.3		4	t _e 30 mm	0.7276	5	f' to		1
10		46.1		5	ΔHm cal/g	- 0.12.0	H	h'		1
1	4	9.0		5	ΔHv cal/g		\vdash	m to		†
Pressure mm 25°C		2.93	72	5	25°C	78.20	5	n º1	:	
t _e		181.	12	5	30 mm BP	74.61 63.58	5 5	0		
Density	\top				te (d.e)	61.55	5	m' to		
g/ml 20°0		0.75		2	e (C, C)	61.47	5	n' ' <u>°</u>	'	
dt 25 4 30		0.74 0.74		2 4	ΔHv/T _e	19.27	5			
a	+	0.76		4	d 68 to		5	Surface tension dynes/cm. 20°C	25.13	5
Ъ	\perp	-0.03		4	$\frac{ \mathbf{e} }{\mathbf{d}} = \frac{181}{1} \cdot \frac{\circ}{25} \cdot \frac{\circ}{100}$		5	3 0	24.08	5
Ref. Index		1 42			e' 68 °C		5	40	23.05	5
n _D 20°0		1.42		2 2	d g/ml	0.245	5	Parachor [P]		1
30		1.41		4	l vc mii/g	4.078	5	30	'	
"C"		0.74	53	4	t _c °C	15637.	5	40 Su aa	1. 424.2	5
MR (Obs.		48.0		2	PV/RT	13037.	-	Exp. L.1, %/wt.		╁
MR (Calc. (nD-d/2)	۱۲.	48.38 1.04		5 2	25°C	1.0000	5	u.		
Dielectric	+	2.01		5	30 mm BP	1.0000 0.9480	5 5	Dispersion	97.	2
A 68 t	-	6, 97		5	te	0.9341	5	Flash Point °C Fire Point		1
B 224 °		519.		5	t _c	0.24	5	M Spec.	+	╁
C (2	+	208.		5	ΔHc kcal/m	j		Ultra V.	İ	ļ
A* 68 to B*, 191 °		1.47 428.9	382	.5 5	ΔFf			X-Ray Dif. Infrared		
к – –	-				Viscosity				 	- -
t t	-				centistokes 7°C			Solubility in +		
t _x t					7 ℃	1		Carbon tet. Benzene		1
A' 25 to		7.33	275	5				Ether		1
B' _ 68 °	<u> </u>	723.3 226.		5	B ^V to	<u> </u>	\vdash	n-Heptane Ethanol	İ	
A'* 25 to	+	1.81	623	5	Av C			Water		
B'* 68 °		622.8	023	5	(BV) to	1		Water in		<u> </u>
Ac 224 to	0	7.79	55	5	(A ^V) °C					
Bc tc_°	딕	305. 308.		5	c _p liq. °K		\vdash			
Cryos. A	+				1 -			1		
consts. B					P					
t _e °C	T	181.20		5	c _v vap.					
$T_R = 0.8$	32 1	`c						grams/100 grams	ms solver	ıt
REFEREN	CE	S: 1-D	ow	2-AF	PI 3-Lit. 4-0	Calc. from det	t. dat	ta 5-Calc. by fo		
SOURCE:				A.	PI					
PURIFICA	TIC	N:		A.	PI					
LITERATU	JRE	REF	CRE	CES	:					
L										

No. 111 2, 3, 5-Trimethylheptane NAME STRUCTURAL FORMULA CH₃ ch,ch ch ch,ch ch,ch, Molecular C₁₀H₂₂ Mole Ref. Molecular Weight 142,276 Ċнз ĊH, % Pur. Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm ٩Ķ g 25°C 3.8869 5 B. P. °C h ВP 0.05219 5 760 mm 157. 91.05 2 ſ١ 0.03744 5 to 100 4 g' °K 30 62.05 4 30 mm 0.7253 5 10 39.97 5 h' ∆Hm cal/g 5 1 3.12 m to AHv cal/g Pressure n °К 25°C 75.08 mm 25°C 4.2557 5 5 o 30 mm 72.14 5 te 1164.4 BP 61.37 5 m' Density to te te (d, e) 59.43 5 n' g/ml 20°C °K 0.741 2 5 59.36 ٥' dt4 25 0.737 2 AHV/Te 18.88 5 30 0.733 4 Surface tension 62 to 79.19 d 5 0.757 4 23.82 5 dynes/cm. 20°C 175 0.1135 5 °C -0.038 4 30 22.81 5 ăח 25 to 77.06 40 21.82 5 Ref. Index e' 62 °C 0.0793 5 20°C 1.416 2 [P] ⁿD d g/ml vc ml/g t °C Parachor 0.231 5 25 1.414 20°C 5 4.327 30 1.411 4 30 $^{\mathbf{t}}\mathbf{c}$ 323 5 40 "C" 0.7469 4 P_c mm 14494. 5 5 Sugd. 424.2 MR (Obs.) 48.2 2 PV/RT Exp. L.1.%/wt. 48.38 MR (Calc.) 5 25°C 1.0000 5 (nD-d/2)u. 1.046 2 30 mm 1.0000 5 2 Dispersion 97. Dielectric 2,005 5 0.9480 RP Flash Point C 0.9343 5 A 62 to 6.89044 5 5 Fire Point 0.24 1468.2 B | 210°C M. Spec. Ultra V. С 209.2 5 AHc kcal/m ΔHf A* 62 to 1.39506 1379.2 5 X-Ray Dif. ΔFf B* 185 °C Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. $\mathbf{t_{x}}$ °C Benzene 25 to 7,24970 Ether B١ <u>62 °C</u> 1669.6 5 n-Heptane B_v C' 227.2 5 to Ethanol °C Water 25 to A'* 1.73638 5 5 62 °C Water in B'* (B^V) 1569.4 to Ac 210 to 7.66447 5 (AV) °C 2190.1 Bc tc °C c_p liq. °K 302.1 5 Cryos. A° c_p vap. °K consts. B° c, vap. t_e °C 174.74 5 $T_{R} = 0.81 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** LITERATURE REFERENCES:

							No. 11	2
NAME	2,3,	6-Trimet	hylheptane			STRUCTURAL	FORMUL	A
						СН3		
	1. (сн ₃ сн сн (сн	2)2CH-CH	H ₃
Mole % Pur.	Ref.	Molecul Formul		Molecular Weight 142.2	76	Ċн ₃	Ċн ₃	
		Ref.	1		Ref.			Ref
F.P. °C			dt/dP	1		f to	T	
F.P. 100%			°C/mm	2 (2(2	ا ۔ ا	g	.	1
B. P. *C 760 mm	155.7	2	25°C BP	3.6863 0.05206	5	h		
100	89.92	4	te	0.03743	5	f' to		
30 10	60.99		30 mm	0.7234	5	g' 'K h'	1	
ì	2.22		ΔHm cal/g	<u> </u>		 :		+
Pressure			ΔHv cal/g 25°C	74,71	5	m to		
mm 25°C	4.50 1161.4	93 5	30 mm	71.88	5	۰		
Density		-	BP t	61.18 59.26	5	m¹ to		
g/ml 20°C			te te (d, e)	59.19	5	n' ' °K	-	
dt 25 4 30	0.73		ΔHv/T _e	18.88	5	<u> </u>	 	+
8	0,75		d 60 to		5	Surface tension dynes/cm. 20°C	23.00	5
ь	-0.03		-e - - 73 ℃ d' - 25 to		5	y 30	22.01	5
Ref. Index	1 41	25 2	e' 60 °C		5	Pomobon [D]	21.05	5
D 25	1.41		d g/ml v ml/g t C C	0.229 4.368	5	Parachor [P] 20°C		1
30	1.40		tc °C	320.	5	30 40	1	j
"C"	0.74		P _c mm	14287.	5	Sugd	424.2	5
MR (Obs.) MR (Calc.)	48.25 48.38	2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric	1.99	5 5	BP	0.9485	5	Flash Point °C	 	+-
A 60 to		867 5	te t _c	0.9349	5	Fire Point		
B <u>L²⁰7</u> • <u>C</u>	1463.3	5	ΔHc kcal/m	 	ļ -	M Spec.		
A* 60 to	1.39		ΔHf ΔFf			Ultra V. X-Ray Dif.		ĺ
B* 183 °C	1374.3	5	Viscosity	 		Infrared		
c	_		centistokes		1	Solubility in +		1
tk to	l		η ∘c			Acetone Carbon tet.		
t _x C	7.24	889 5		ļ		Benzene Ether		
	1664.6	5	B ^V to	<u> </u>	-	n-Heptane	1	
C' A'* 25 to	1.73	5 605 5	A to			Ethanol Water		
	1564.5	5	(B ^V) to	-		Water in		
Ac 207 to	7.64		(A ^V) °C					
Bc tc C	2165.1 299.6	5	c _p liq. °K	+		1		
Cryos. A*	2/7.0		c _p vap. *K					
consts. B°								
t _e *C	173, 29	5	c _v vap.	<u> </u>		L <u>.</u>		
T _R = 0.81						† grams/100 gra		nt
REFERENCE:	೬5: 1-D		PI 3-Lit. 4-0	Calc, from det	da	ta 5-Calc. by for	mula	
PURIFICAT	ION:		PI					
LITERATU								-
			•					

NAME	2,4,4	4-Tri	meth	ylheptane		STRUCTURAL	3	LA	
Mole % Pur.	Ref		lecul rmula		Molecular Weight 142.2	276	сн ₃ сн-сн ₂ с сн ₃ сн		H ₃
			Ref.			Ref.		***************************************	Ref.
F. P. °C				dt/dP			f to		-
F.P. 100	6			°C/mm			g °E		
B.P. °C				25°C BP	3.3037 0.05179	5 5	h	ŀ	1
760 mm 100	153. 87.5	6	2 4	te	0.03743	5	f' to		T
30	58.8		4	30 mm	0.7194	5	g' ' <u>°F</u>		
10	36.9		5	ΔHm cal/g		+	h'		1
<u> </u>	0.3	555	5	ΔHv cal/g		+-1	m to		T
Pressure mm 25°C	5.0	832	5	25°C	73.95	5	n P	<u>- </u>	İ
t _e	1108.	032	5	30 mm	71.33	5	Li		
Density	†			BP t _e	60.68	5	m' to		
g/ml 20°0			2	te (d, e)	58.73	5	n'		
d ₄ 25	0.7		2 4	AHv/Te	18.85	5			
·	0.7		4	d 59 to	77.99	5	Surface tension	22.04	_
a b	-0.0		4	_e_ <u> _168</u>	0.1133	5	dynes/cm. 20°C	23.06	5
Ref. Inde	,		\vdash	d' 20 to e' 59 °C	75.89 0.0775	5	40	20.18	5
n _D 20°			2		0, 208	5	Parachor [P]		T
25 30	1.4		2 4	d _c g/ml v _c ml/g t _c °C	4.804	5	20°C		
"C"	0.7		4	tc °C	307.	5	30 40		
MR (Obs.			2	P _c mm	12706.	5	Sugd	. 424.2	5
MR (Calc.			5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0	46	2	25°C 30 mm	1.0000	5 5	u. Dispersion	97.	2
Dielectric	1.9	94	5	BP	0.9485	5	Flash Point C	 /	+-
A 59 to		8497	5	t _e	0.9352 0.24	5	Fire Point		
B 1197 °C			5	tc ΔHc kcal/m	0.24	۲	M. Spec.	1	+
A* 59 to	209.9	9259	5	ΔHc Real/III			Ultra V.		
B* 178 °C			5	ΔFf			X-Ray Dif. Infrared		
к ——	_			Viscosity			Solubility in +		
t _k — to	-			centistokes り °C			Acetone	1	
t _x °C	;			,			Carbon tet. Benzene		
A' 20 to	7.2	4720	5 5				Ether		
B' 59 °C	1654.4 228.		5	PA	†	+	n-Heptane	1	
A!* 20 to		2524	5	B ^V	1		Ethanol Water		
B'* 59 °C		3534	5	(B ^V) to	-		Water in		
Ac 197 to	7.5	8044	5	(A ^V) °C					1
Bc tc °C	2068.0		5	c _p liq. °K	T	\vdash			
Ce	281.1		5						-
Cryos, A' consts, B'				c _p vap. °K					
t _e °C	167.8	1	5	c _w vap.					
$T_R = 0.8$			ب	<u> </u>	1		grams/100 gr	ame colsec	
REFEREN		Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by fo		111
SOURCE:			AF		-3.0, 110m de	ua	5 Care, by 10		
PURIFICA	TION:		AF					·	
LITERATU		ERF							
				•					

						No.	o. 114
NAME	2,4,5-Tri	meth	ylheptane			STRUCTURAL FOR	MULA
						CH ₃	
			T			CH3CH CH2CH CH C	H ₂ CH ₃
Mole		lecul		Molecular Weight 142.2	76	сн ₃ сн ₃	- ,
% Pur.	1 1 10	rmul		weight 142.2	Ref	r	Ref
F, P. *C	Γ	Ref.		T	Kei		- Kei
F.P. 100%	 	-	dt/dP •C/mm			f to g °K	
B. P. *C	†		25°C	3.8869	5	h	
760 mm 100	157. 91.05	2 4	BP t _e	0.05219 0.0374	5	f' to	
30	62.05	4	30 mm	0.7253	5	g' ' <u>°</u> K_	
10 1	39.97 3.12	5	ΔHm cal/g			h!	
Pressure	31.2	-	ΔHv cal/g			m to to	
mm 25°C	4.2557	5	25°C 30 mm	75.08 72.14	5	n ' *K	
t _e	1165.1	5	BP	61.40	5	l — !	
Density g/ml 20°C	0.741	2	t _e (d, e)	59.48 59.39	5	m' to n' °K_	
at 25	0.737	2	ΔHv/T _e	18.89	5	o'	
	0.733	4	d 62 to	79.16	5	Surface tension	
a b	0.757 -0.0 ₃ 8	4	e 175 °C	0.1131	5		.82 5 .81 5
Ref. Index	1 3	Ť	a' 20 to		5		.82 5
n _D 20°C	1.4160	2		0,231	5	Parachor [P]	
25 30	1.4137	2 4	d g/ml vc ml/g tc °C	4.325	5	20°C	ĺ
"C"	0.7469	4	t _c ·C	323.	5	40	
MR (Obs.)	48.2	2	P _c mm	14500.	5	Sugd. 424	.2 5
MR (Calc.) (nD-d/2)		5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.	
Dielectric	1.046 2.005	5	30 mm BP	1.0000	5	Dispersion 97	. 2
A 62 to	6,89044	5	t _e	0.9485 0.9348	5	Flash Point °C	
B [210 °C	1468.2	5	t _c	0.24	5	Fire Point	
С	209.	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.	
A* 62 to B* 185 °C	1.39418	5	ΔFf		1	X-Ray Dif. Infrared	
к — — —			Viscosity			Solubility in +	
t _k	1		centistokes 7°C	1		Acetone	
tx C			"			Carbon tet. Benzene	
A' 20 to	7.24970	5		}		Ether	
B' ∟62 °C	1669.6	5	B ^V to		1	n-Heptane Ethanol	
A** 20 to	1,73638	5	ĀV °C	1		Water	
B'* 62 °C		5	(B ^V) to			Water in	
Ac 210 to	7.6647 2190.	5	(A ^V) °C				
Bc tc_°C	302.	5	c _p liq. °K				
Cryos. Aº			c _p vap. °K				
t _e °C	174 74	5	c _v vap.				
$T_{R} = 0.81$	174.74	٠		l	L	+ (100	
REFERENC		2-AI	PI 3-Lit. 4-0	Calc from det	t de	grams/100 grams sta 5-Calc. by formula	
SOURCE:		Al				5 Cure. by formula	•
PURIFICAT	ION:	Al					
	E REFERE						

							No. 1	15
NAME _	2,4,6-Tr	imethy	lheptane			STRUCTURAL	FORMUI	LΑ
	11					сн ₃ сн сн ₂ сн с	сн,сн с	Н,
Mole	Ref. M	lolecul	ar C ₁₀ H ₂₂	Molecular		ch, ch,	-,	,
% Pur.	F	ormul		Weight 142.2				To 6
		Ref.		+	Ref.			Ref.
F.P. °C F.P. 100%			dt/dP *C/mm		1	f to	1	
B. P. ℃		+	25°C	2.3773	5	g '° <u>K</u>	1	1
760 mm	144.8	2	BP	0.05099	5	h		+-
1 00 30	80.41	4	t _e	0.0373	5	g' to	į	1
10	52.13 30.6	4 5	30 mm	0.7071	├ ⁵_	h' .		1
1	-5.3	5	ΔHm cal/g	 	┼—	m to	†	+
Pressure mm 25°C		١.	ΔHv cal/g 25°C	71.69	5	n		
t _e	7.2873 1133.7	5	30 mm	69.68	5	0		
Density		+	BP te	59.47 57.73	5	m' to		
g/ml 20°C	0.7225	2	te (d, e)	57.68	5	n' <u>*K</u>	-	
d ₄ 25	0.7184 0.7143	2 4	AHv/T _e	18.92	5	ļ	ļ	—
a	0.7389	4	d 52 to	75.43	5	Surface tension dynes/cm. 20°C	21.53	5
ь	-0.0382	4	-161 °C	0.1102	5	8 30	20.57	5
Ref. Index			e' 52 °C	73.53	5	40	19.63	5
ⁿ D 20°C	1.4071 1.4046	2 2	d _c g/ml	0.226	5	Parachor [P] 20°C		}
30	1.4022	4	v_mi/g	4.418	5	30		
"C"	0.7505	4	t _c °C	303.	5	40	424.2	-
MR (Obs.)	48.48	2	P _c mm	13721.	5	<u> </u>	424.2	5
MR (Calc.) (nD-d/2)	48.38	5 2	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		
	1.0459		30 mm	1.0000	5	Dispersion	97.	2
Dielectric A 52 to	1.980 6.8737	5	BP t _e	0.9510	5	Flash Point C		
B 188 °C	1422.6	7 5	t ^e c	0.24	5	Fire Point	ļ	+
С	211.5	5	∆Hc kcal/m	1		M. Spec. Ultra V.		
A* 52 to	1.3850	4 5 5	ΔHf ΔFf	ì		X-Ray Dif.		
B*[171 °C K	1334.4	"	Viscosity		1	Infrared	-	+
c t, — -to			centistokes		ł	Solubility in Acetone		
t _k to t _x °C			η °C			Carbon tet.		
A' 5 to	7.2422	6 5				Benzene Ether	l	
B' 1_52 °C	1623.5	5	-v	 	-	n-Heptane		
	229.	5	B ^V to °C		}	Ethanol Water		
A'* 15 to B'* 52 °C	1.7334 1523.7	4 5 5	(B ^V) - to	-		Water in		
Ac 188 to	7.5565		(A ^V) °C					
Bc t _c °C	2021.3	5	c _p liq. °K			1		
Cc	288.3	1 3	4 1	-				
Cryos. A° consts. B°			c _p vap. °K					1
t _e °Ç	161.00	5	c _v vap.	1				
$T_R = 0.80$	T _c			.4		grams/100 gra	ms solve	nt
REFERENCE		2-A	PI 3-Lit. 4-	Calc. from de	t. da	ata 5-Calc. by for		
SOURCE:		AI						
PURIFICATI	ON:	AI	PI					
LITERATUR	E REFER	ENCES	5:			***		

														N o. 11	6
NAME		2,5,5	-Tr	imetl	nylher	otane					STR	UCTURA	L I	FORMUL	
		<i>,</i> -, -			-,					\dashv			CH		
											CH.	сн (сн.		сн ₂ сн	
Mole		Ref.	Мо	lecul	ar ,	~ u		Molecul	ar		3	сн ₃	ĊH	7.	3
% Pur.			Fo	rmul	<u>a '</u>	C ₁₀ H ₂	2	Weight	142.2	76		3		-3	
				Ref.						Ref					Ref.
F.P. ℃				L.	dt/			1		[[f		to		1
F.P. 1009	•			ļ		C/mm S°C		1 3	2598	5	g	i '	<u>•</u> K_		1
B. P. °C 760 mm	1.	52.80		2	B				05184	5	h	L			<u> </u>
100		87.3		4	t _e			0.	0375	5	f'	' ' 1	to		1
30 10		58.5		4	30	mm		0.	7195	5	g'	¦ '	_•K_		
1		36.6 0.1		5	ΔН	m cal	/g				h'	<u> </u>			┼
Pressure	+					v cal/	g			_	m n	! ;	to °K		
mm 25°C	1	5.16	21	5		°C mm		73. 71.		5 5	0				1
t _e	111	53.		5	В			60.	60	5	m'		to		\vdash
Density g/ml 20°0	.	0.73	68	2	ţ.	(d, e)		58. 58.		5 5	n'	¦ i	°ĸ		
_a t 25	1	0.73	28	2	,			1		1 1	٥'	i — —			
⁴ 30	\perp	0.72		4	d d	Hv/Te	9 to	18. 77.		5	Surf	ace tensi	on		
		0.75		4	a e	1 17	0 °C		1125	5	dyne	s/cm. 2		23.29	5
Ref. Index	_	-0.03	80	4	a,-	$_{\rm i}$ $_{\rm i}$	5 to	75.		5	•	3 4		22.29 21.32	5
n _D 20°C		1.41	36	2	e'		9 °C	+	0773	5	Para	chor [P			+
- 25		1.41	12	2	d _C	g/ml ml/g °C			230 351	5		2	0°C		1
30	┿	1.40		4	tc	°C′		317.	331	5		3 4			
"C"	4_	0.74		4	P	mm		14258.		5				424.2	5
MR (Obs.) MR (Calc.		48.21 48.38		2 5		/RT		<u> </u>		\Box	Exp.	L.1.%/	wt.		
(nD-d/2)	Ί	1.04		2		•c			0000 0000	5 5		u.			
Dielectric					BI	mm P			9485	5		ersion		97.	2
A 59 t		6.87	94	5	te				9351	5		h Point ° Point	C		l
B (205°C		51. 10.		5	tc		,	0.		3	M S				†
C	-		710	5	ΔH	c kcal f	/m				Ultr	a V.			
A* 59 to B* 180 °C	13	1.38 62.	110	5	ΔF						X-R Infra	ay Dif. Ted			
к — — -	-					cosity						bility in	+		+
t ₁ to	-					tistok	°C				Ace	tone			
t _k -to					7		·					bon tet.			
A' 10 to	,	7.24	154	5							Eth				ŀ
B' L 59 °C		51.6		5	B	1		+		\vdash		leptane			
		28. 1.72	002	5	Av	i	to ℃	į			Wat	anol ter			
A'* 20 to B'* 59 °C		1.12 51.5	903	5	(B ^V		- to					er in			
Ac 205 to	,	7,62	27	5	(AV		°C								
Bc t °C	21:			5	c _p		°K	-							
Ce	+ 4	98.		5	1										İ
Cryos, A ^c consts, B ^c				1	Сp	vap.	°K								1
t _e °C	+-	70.01		5	c _v	vap.									
$T_R = 0.8$	1 Tc			L	I			L		I1	+ ~ ~ ~	me/100	~ F 9 F	ns solven	
REFEREN		1 - De	ow w	2-AI	PI 3	-Lit.	4-0	Calc. fr	om det	t. dat		Calc. by			
SOURCE:					PI										
PURIFICA	TION	:			PI										
LITERATU			ERE												
					••										
1															
1															

							No. 117
NAME	3,3,4-Tr	imet	hylheptane			STRUCTURAL	FORMULA
						CH ₃	
						CH ₃ CH ₂ C CH	CH ₂) ₂ CH ₃
Mole % Pur.	Ref. Mo	lecul	ar C ₁₀ H ₂₂	Molecular Weight 142.2'	76	Ċн ₃ ċн ₃	
- /		Ref	r	T T T T T T T T T T T T T T T T T T T	Ref.	r	Ref.
F.P. °C		1	dt/dP			f to	
F.P. 1009	•		°C/mm			g <u>K</u>	
B. P. °C			25°C BP	5.1760 0.05288	5	h	
760 mm	164. 97.1	2 4	te	0.0374	5	f' to	
30	67.7	4	30 mm	0.7358	5	g'° <u>K</u>	
10	45.3 7.9	5	AHm cal/g		Γ	h'	
Pressure	+	 	ΔHv cal/g			m to	
mm 25°C	3.1136	5	25°C 30 mm	77.06 73.54	5	" ' <u>-</u> -	
t _e	1184.	5	BP	62.56	5	m' to	
Density g/ml 20°C	0.757	2	t _e (d, e)	60.51 60.42	5	n'	1
dt 25	0.753	2		18.88	5	o'	
4 30	0.749	4	ΔHv/T _e	81, 27	5	Surface tension	
a b	0.773 -0.0 ₃ 8	4	e 183 °C	0.1141	5	dynes/cm. 20°C	25.95 5
Ref. Index		+	d' 20 to	79.12 0.0824	5	30 40	24.87 5 23.82 5
n _D 20°C		2		0.0824	5	Parachor [P]	
25 30	1.422 1.420	2	d g/ml vc ml/g	4.246	5	20°C	
"C"	0,7444	4	vc ml/g tc °C	336.	5	30 40	
MR (Obs.)		2	P _c mm	15079.	5	Sugd.	424.2 5
MR (Calc.		5	PV/RT 25°C	1,0000	5	Exp. L.l.%/wt.	
(nD-d/2)	1.046	2	30 mm	1.0000	5	u. Dispersion	97. 2
Dielectric		5	B P	0.9480 0.9337	5	Flash Point °C	
A 68 to		5	t c	0.7337	5	Fire Point	
c '	208.	5	∆Hc kcal/m			M. Spec.	
A* 68 to		5	ΔHf ΔFf	1	ŀ	Ultra V. X-Ray Dif.	
B*[193 °C	1403.7	5	Viscosity		 	Infrared	
c	_		centistokes		İ	Solubility in *Acetone	
		1	∥ າ °⊂	1	}	Carbon tet.	
A' 20 to		5				Benzene Ether	
B' 68 °C	1695.6	5			 	n-Heptane	
C1	226.	5	B ^V to C			Ethanol Water	i i
A'* 20 to B'* 68 °C		5	$\frac{ \mathbf{B}^{\mathbf{v}} }{ \mathbf{B}^{\mathbf{v}} } - \frac{ \mathbf{v} }{ \mathbf{t} }$	1		Water in	
Ac 227 to	7.7400	5	(A ^V) °C				
Bc tc °C	2310.	5	c _p liq. °K		\vdash	· ·	
Cc	313.	5	41	I			
Cryos, A° consts, B°			c _p vap. °K				
t _e °C	182.73	5	c _v vap.				
$T_R = 0.8$		<u> </u>	ш. :	1	ь	grams/100 gra	ms solvent
		2-A	PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc. by for	
SOURCE:		AF					
PURIFICA	TION:	AF					
	RE REFERE						

							No. 11	8
NAME	3, 3, 5	-Trime	thylheptane			STRUCTURAL	FORMUL.	A.
						сн,		
			T				сн сн,с	Н
Mole	Ref.	Molecu		Molecular Weight 142.2	76		Ċн ₃	,
% Pur.		Formu		Weight 142.2				Ref.
E D 40	T	Re	T	1	Ref.		T	Ker.
F.P. C F.P. 1007			dt/dP •C/mm		1	f to	İ	
B. P. *C	1		25°C	3. 6798	5	h .	1	
760 mm 100	155.68		BP t _e	0.05207 0.0374	5	f' to	-	+
30	89.9 60.96	4 4	30 mm	0.7234	5	g' ' <u>*</u> K_	1	1
1 0 1	38.9	5	ΔHm cal/g	1		h'		
Pressure	+		ΔHv cal/g	 		m to		
mm 25°C	4.51		25°C	74.69	5	n •K	┨	1
t _e	1164.	5	30 mm BP	71.86 61.26	5	<u> </u>	 	+
Density g/ml 20°C	. 0 74	20 2	te te (d, e)	59.35	5	m' to	ĺ	1
	0.74			59.28	5	o' =	1	ļ
dt 25 4 30	0.73	48 4	ΔHv/T _e	18.91	5	Surface tension	 	T
a b	0.75		d 61 to		5	dynes/cm. 20°C	24.05	5
Ref. Index	-0.03	80 4	- a' - 20 to	76.66	5	30 40	23.03 22.04	5
n _D 20°C			e' j 61 °C		5	Parachor [P]	 	+
25 30	1.41		d _c g/ml v _c ml/g t _c °C	0.232 4.303	5	20°C	l	l
"C"	0.74		ե [©] °C ຶ	322.	5	30 40	-	
MR (Obs.)		2	P _c mm	14542.	5	Sugd.	424.2	5
MR (Calc.		5	PV/RT	1 0000	_	Exp. L.1.%/wt.		T^{T}
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric	<u> </u>		BP	0.9500	5	Flash Point °C		╁
A 61 to B 209 °C		77 5	t _e	0.9365 0.24	5	Fire Point		
C C	209.	5	ΔHc kcal/m			M Spec.	1	
A* 61 to			ΔHf ΔFf	1		Ultra V. X-Ray Dif.	ł	
B* ∟183 °C	1373.2	5	Viscosity			Infrared	<u></u>	
c	_]	- 1	centistokes	1		Solubility in +	ĺ	
t _k to		į	7 °	;		Carbon tet.		
t _x °C		789 5	4			Benzene Ether		
B' _ 61 °C	1664.2	5	-		-	n-Heptane		1
C'	227.	5	B ^V to			Ethanol Water		
A'* 20 to B'* 61 °C	1.73	506 5		-	}	Water in		1
Ac 209 to	7 65		(A ^V) to	1				
Bc t °C	2176.	5	c _p liq. •K	+	 	1		
CE	301.	5	-ll ⁻					1
Cryos. A° consts. B°			c _p vap. °K	.			}	
t _e °C	173.35	5	c _v vap.					
$T_R = 0.8$			ш	т	ь	+ grams/100 gra	ms solver	+
REFEREN		ow 2-A	PI 3-Lit. 4-	Calc. from de	t. da			·*
SOURCE:			API					
PURIFICA'	TION:	1	IPI			· · · · · · · · · · · · · · · · · · ·		
LITERATU	RE REFE	ERENCE	S:					

No. 119 3, 4, 4-Trimethylheptane NAME STRUCTURAL FORMULA CH₃ сн₃сн₂сн с (CH₂)₂CH₃ Mole Ref. Molecular Molecular Weight 142.276 ҁҥҙҁҥ C10H22 % Pur Formula Ref Ref. Ref. F. P. °C dt/dP f to F.P. 100% °C/mm °K g 25°C 5,1760 5 B.P. °C h ΒP 5 0.05288 760 mm 164. 2 0.0374 5 f١ to 100 97.1 4 <u>°к</u> g' 30 67.7 4 30 mm 5 0.7358 10 5 45.3 h! ∆Hm cal/g 1 7.9 5 m to AHv cal/g Pressure ٩K n 25°C 77.06 mm 25°C 3.1136 5 o 30 mm 73.54 5 te 1184. 5 BP 62.56 5 m' Density to 60.51 5 n' g/ml 20°C °K te (d, e) 0.757 2 60.42 5 01 d_4^t 25 0.753 ΔHv/T_e 18.88 5 30 0.749 4 Surface tension d 68 to 81.27 5 0,773 dynes/cm. 20°C 25.95 5 0.1141 5 183 ь -0.038 4 24.87 30 5 dיi 20 to 79.12 5 40 23.82 5 Ref. Index e¹ 68 °C 0.0824 5 20°C 1.424 [P] ⁿD Parachor d_c g/ml 5 0.236 25 2 1.422 20°C vc ml/g 5 4.246 1.420 4 30 30 t_c 5 336. 40 "C" 0.7444 4 P_c mm 15079. 5 424.2 5 Sugd. MR (Obs.) 47.9 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 48.38 5 25°C 1.0000 5 (nD-d/2)1.046 2 30 mm 1.0000 Dispersion 97. 2 Dielectric 2,028 5 0.9480 5 BP Flash Point C 0.9337 A 68 to 6.8988 5 Fire Point 0.24 5 B 1227 °C 1494. M. Spec. 208. C 5 AHc kcal/m Ultra V ΔHf A* 68 to 1.39669 5 X-Ray Dif. ΔFf B*[193 °C 1403.7 Infrared ĸ Viscosity Solubility in centistokes Acetone to $t_{\mathbf{k}}$ Carbon tet. °C t_x Benzene 20 to 7, 25291 Ether В' 68 °C 1695.6 n-Heptane B_v C' 226. 5 to Ethanol A'* 20 to B'* 68 °C °C Water 5 1.73706 Water in 1595.4 (B^V)| to Ac | 227 to 7.7400 5 (AV) °C °C 2310. Bc _tc cp liq. ۰ĸ Cc 313. 5 Cryos. Aº c_p vap. °K consts. B° c vap. te °C 182.73 5 $T_R = 0.82 T$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

Т					т		No. 12	0
NAME _	3, 4, 5-7	Trimeth	ylheptane			STRUCTURAL	FORMUL	A
						CH ₃		
					\neg	CH3CH2CHCHCF	сн,сн,	
Mole	Ref.	Molecul	ar C ₁₀ H ₂₂	Molecular	_,	с́н, с́н		
% Pur.			1022	Weight 142.2	_			_
	·	Ref.	ļ	T	Ref			Ref
F, P, °C			dt/dP	Į		f to	Į.	1
F.P. 100%		\perp	*C/mm 25*C	5,1294	5	g <u>K</u>	1	
B. P. °C 760 mm	164.	2	BP	0.05288	5	h	<u> </u>	-
100	96.97	4	t _e	0. 03729	5	f' to		1
30	67.56	4	30 mm	0.7358	5	g' 'K_	-	1
10 1	45.16 7.77	5 5	ΔHm cal/g			h'	Ļ	↓_
Pressure	 ```	+-	ΔHv cal/g			m to		1
mm 25°C	3.146	5	25°C	76.96	5	n •K	1	
te	1190.3	5	30 mm BP	73.47 62.56	5 5	<u> </u>		ــ
Density		$\neg \vdash \vdash$	t_	60.76	5	m' to		
g/ml 20°C	0.759	2	te (a, e)	60.44	5	n' <u>°K</u> _	1	
dt 25 4 30	0.755	2 4	AHv/Te	18.96	5	<u> </u>	 	+
a	0.775	4	d 68 to	81.12	5	Surface tension dynes/cm. 20°C	26,22	1
b	-0.038	4	$\frac{183}{4}$ $\frac{\circ}{1}$		5	30 30	25.13	i
Ref. Index		$\neg \neg \neg$	d' 20 to		5	40	24.07	1
n _D 20°C	1.424	2	d _c g/ml	0,245	5	Parachor [P]		
25 30	1.422	2 4	IV mile	4. 080	5	20°C	424.19 424.19	4
"C"		-	tc •C	335.9	5	40	424.19	4
	0.7424		P _c mm	15182.	5	Sugd.	424.2	5
MR (Obs.) MR (Calc.)	47.832 48.380	5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.044	2	25°C	1.0000	5	u.	ļ	j
Dielectric	2, 028	5	30 mm BP	1.0000 0.9480	5 5	Dispersion		
A 68 to	6. 8988	30 5	t _e	0.9383	5	Flash Point °C Fire Point		
B 225 °C		5	t _c	0.232	5		ļ	+
С	208.	5	ΔHc kcal/m	ļ		M Spec. Ultra V.	İ	
A* 68 to	1.3893		ΔHf ΔFf		1	X-Ray Dif.		
B* L ¹ 93 °C	1401.63	5	Viscosity	 	\vdash	Infrared	<u> </u>	1
c			centistokes			Solubility in +		1
tk to			η •c		l	Acetone Carbon tet.	l	1
_x '	- 2526					Benzene	Í	
A' 15 to B' 68 °C	7.2529	5 5			L_	Ether n-Heptane		
č, 🗀 🐃 🖰	226.	5	B ^V to			Ethanol		
A** 20 to	1.7370		A ^V I °C			Water		
B'* 68 °C	1595.25	5	(B ^V) to			Water in	ļ	+
Ac 225 to	7.7435		(A ^V) oC					
Bc tc_°C	2315.88 314.01	5 5	c _p liq. °K			1		
Cryos. A°	3	\dashv						1
consts. B°			c _p vap. °K			H		
te °C	182.78	5	c _w vap.					
$T_{\mathbf{R}} = 0.82$			<u> </u>	L	L	+ arams /100	ma asless	<u> </u>
REFERENC		v 2-AE	PI 3-Lit. 4-0	Calc. from det	- 4-	grams/100 gra- ta 5-Calc. by for		
SOURCE:		AP		Jaic. Hom del	. ua	a J-Carc, by for		
PURIFICAT	ION:							
		AP						
LITERATUI	CE KEFEF	CENCES	:					

							No. 17	-1
NAME	3-isopr	op yl-2- r	nethylhexane			STRUCTURAL	FORMUI	JA
						C ₃ H ₇		
						сизси си (CH ₂) ₂ CH ₂	
Mole	Ref.			Molecular		с́н ₃	2.2 3	
% Pur.		Formul	a C ₁₀ H ₂₂	Weight 142.2	76			
		Rei	·	ļ	Ref.		T	Ref.
F. P. °C			dt/dP			f to		
F.P. 100%			°C/mm 25°C	5,4068	5	g '° <u>K</u>	-	1
B. P. °C 760 mm	163.	2	BP	0.05179	5	h		↓
100	97.3	4	t _e	0.0367	5	f! to	1	1
30	68.3	4	30 mm	0.7276	5	g' '° <u>K</u>		
10 1	46.1	5	∆Hm cal/g			h' .	ļ	1
Pressure	/		ΔHv cal/g		Ţ-	m to		İ
mm 25°C	2.93	72 5	25°C	78.20	5	n •K	-	
t _e	1181.	5	30 mm BP	74.61 63.58	5	 		+
Density			t_	61.55	5	m' to		
g/ml 20°C	0.75		le (a, e)	61.47	5	",	1	1
d ^t 25 4 30	0.74		ΔHv/T _e	19.27	5			+
a	0.76		d 68 to	82.56	5	Surface tension dynes/cm. 20°C	25.13	5
ь	-0.03		_e	0.1164	5	8 30	24.08	5
Ref. Index			e' 68 °C		5	40	23.05	5
ⁿ D 20°C	1.42		d _c g/ml	0.245	5	Parachor [P] 20°C		
30	1.41		v _c mi/g	4.075	5	30		
"C"	0.74	53 4	1 _	333.	5	40		
MR (Obs.)	48.0	2	P _c mm	15637.	5		424. 2	5
MR (Calc.)	48.38	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.04		30 mm	1.0000	5	Dispersion	98.	2
Dielectric	2.01		B P	0.9480	5	Flash Point C		+-
A 68 to B 224 °C	6.97- 1519.	46 5	t _c	0.9341 0.24	5	Fire Point		
c '	208.	5	∆Hc kcal/m	 	 	M. Spec.	į	
A* 68 to	1.47	382 5	ΔHf	i		Ultra V. X-Ray Dif.	ľ	
B* 191 °C	1428.9	5	ΔFf		├ ─	Infrared		
K			Viscosity centistokes		l	Solubility in +		
tkto	i		η °C	+		Acetone Carbon tet.		
t _x °C			11			Benzene		
A' 20 to B' 68 °C	7.33 1723.3	275 5				Ether	l	i
C,	226.	5	B _v to			n-Heptane Ethanol	ļ	}
A** 20 to	1.81	623 5] A' °C	_		Water		
B'* 68 °C	1622.8	5	(B ^V) to	-		Water in	·	┼
Acl 224 to	7.79		(A ^V) °C					1
Bc tc °C	2305. 308.	5	c _p liq. °K			1	1	
Cryos. A°	333.	-+-			1			1
consts. B°			c _p vap. °K					
t _e °C	181.20	5	c _v vap.		1		1	
$T_{R} = 0.82$	T _c		<u> </u>	1		grams/100 gra	ms solve	nt
		ow 2-A	PI 3-Lit. 4-	Calc, from de	t. da	ata 5-Calc. by for		
SOURCE:			PI					
PURIFICAT	ION:		PI					
LITERATUE								
	ADFI	-ALINOE	.					

							No. 12	.2
NAME	3, 3-1	Diethylhe:	xane			STRUCTURAL		
		,			\dashv	C ₂ H ₅		
						CH ₃ CH ₂ C (н) сн	
Mole	Ref.	Molecul		Molecular		Ċ ₂ H ₅	2/20113	
% Pur.		Formul	101122	Weight 142.2	_	T		
		Ref.		, 	Ref.			Ref.
F.P. °C			dt/dP	1	1	f to		İ
F. P. 100%	<u>'</u>		*C/mm 25*C	6, 2085	5	g <u>°K</u>	-	
B. P. °C 760 mm	166.3	2	BP	0.05212	5	h	-	┿
100	100.2	4	t _e	0.0367	5	f' to		
30 10	70.9 48.6	4 5	30 mm	0,7325	5	h'	1	
1	11.3	5	ΔHm cal/g			<u> </u>	 	┿
Pressure			ΔHv cal/g 25°C	79.16	5	m to	j	
mm 25°C	2.52°	72 5	30 mm	75.27	5	•		
t _e Density	1170.		BP	64.12 62.03	5	m' to		
g/ml 20°C	0.76	7 2	te te (d, e)	61.94	5	n' ''K	1	
at 25	0.76		ΔHv/T	19.26	5	o' i		
	0.75		d 71 to		5	Surface tension		
a b	0.78 -0.0 ₃		e 185 °C	0.1170	5	dynes/cm. 20°C	27.35 26.22	5
Ref. Index	<u> </u>		d' 20 to		5	40	25.13	5
n _D 20°C	1.42			0,251	5	Parachor [P]		
25 30	1.42		d g/ml vc ml/g	3.981	5	20°C 30		
"C"	0.74	-	, c	340.	5	40	1	
MR (Obs.)	+	2	P _c mm	16196.	5		424.2	5
MR (Calc.		5	PV/RT		5	Exp. L.1.%/wt.		
(nD-d/2)	1.04	4 2	25°C 30 mm	1.0000	5	u. Dispersion	95.	2
Dielectric	2.03	9 5	BP	0.9475	5	Flash Point °C	 /3.	┿
A 71 to			t _e t _c	0.9333 0.24	5	Fire Point	1	
B [230 °C	207.	5 5	ΔHc kcal/m	+	 	M Spec.		
A* 71 to			ΔHf			Ultra V. X-Ray Dif.		
B* 195 °C		5	ΔFf	1	ļ	Infrared		
K C			Viscosity centistokes			Solubility in +		\top
t _t to			7 000			Acetone Carbon tet.		
`x '	1		,	İ		Benzene		
A' 20 to B' 71 °C		375 5				Ether n-Heptane		
c,	225.	5	B ^V to			Ethanol		ļ
A'* 20 to			AV C	_		Water Water in		
B'* 71 °C		5	(B ^V) to			Water III		+-
Ac 230 to	7.84	19 5	(A ^V) °C		ļ			
Cc C-c-	315.	5	c _p liq. °K					İ
Cryos. A°	T		c _p vap. °K	1			1	1
consts. B°	184.93	5	c _v vap.					
$T_R = 0.82$			L	İ	<u> </u>	+ grams/100 gra	ma salwa	
REFEREN		ow 2-AF	PI 3-Lit. 4-	Calc from de	t. da	ta 5-Calc. by for		16
SOURCE:			PI					
PURIFICA	TION:	Al	PI					
LITERATU								
ļ								

No. 123 3, 4-Diethylhexane NAME STRUCTURAL FORMULA C2H5 снзснзсн сн сн,сн, Mole Ref. Molecular Molecular с₂н₅ $C_{10}H_{22}$ Weight 142, 276 % Pur Formula Ref. Ref Ref. F. P. °C dt/dP f to F.P. 100% °C/mm ۰ĸ g 25°C 5, 1849 5 B. P. °C h BP 0.05169 5 760 mm 162. 2 0.0367 5 f١ to 100 96.4 4 g' °<u>K</u> 30 67.5 4 30 mm 0.7261 5 10 5 45.3 h! AHm cal/g 5 8.3 1 m to ∆Hv cal/g Pressure ۰ĸ n 25°C 77.91 3.0744 mm 25°C 5 o 30 mm 74.41 1178. 5 t_e BP 63.41 5 m' to Density te te (d, e) 61,40 5 n' g/ml 20°C °K 0.754 2 61.31 5 ٥' ď4 25 0.750 2 ΔHv/Te 5 19,27 30 0.746 4 Surface tension 67 to 82.25 a 5 0.770 44 dynes/cm. 20°C 25,54 ᇷᅱ 180 °C 0.1163 5 ь -0.0₃8 24.47 5 30 25 to 79.98 5 40 23.43 Ref. Index 67 °C 5 e' 0.0826 20°C 1.420 2 [P] ^{n}D d g/ml vc ml/g t °C Parachor 0,247 5 ž 25 1.418 20°C 4.050 5 30 1.415 4 30 t_c 332. 5 40 "C" 0.7407 4 P_c mm 15718. 5 5 Sugd. 424.2 MR (Obs.) 47.8 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 5 25°C 1,0000 5 11 . (nD-d/2) 1.043 2 30 mm 1.0000 94.6 2 5 Dispersion Dielectric 2.016 5 RP 0.9480 5 Flash Point C 0.9342 67 to 6.9734 5 Fire Point 0.24 B | 223 ℃ 1515. 5 M. Spec. Ultra V. 5 С 208. AHc kcal/m ΔHf A* 67 to 1.47356 5 X-Ray Dif. ΔFf B*| 190°C 1425.4 Infrared ĸ Viscosity Solubility in centistokes Acetone tk tx to Carbon tet. °C Benzene 25 to 7.33227 Ether B١ 67 °C 1719.5 5 n-Heptane C' B_v 226. to Ethanol °C A'* Water 25 to 1.81612 5 Water in 67 °C (B^V)| R!# 1619.0 5 to Acl 223 to 7.7900 (AV) 5 °C Bc tc C 2296. 5 ۰ĸ cp liq. 307. Cryos. A° c_p vap. °K consts. B° c vap. te °C 180.06 5 $T_R = 0.82 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

							No. 12	24
NAME	3-Ethyl-	2, 2-dir	methylhexane			STRUCTURAL		
-	· · · · · · · · · · · · · · · · · · ·					сн ₃		
						Сн ₃ с сн (CH ₂) ₂ CH ₂	
Mole % Pur.		lolecul ormul		Molecular Weight 142.	276	сн ₃ с ₂ н ₅	2.2 3	
N Ful.		Ref		weight 112.	Ref			Ref.
F.P. °C		1	dt/dP			f to	T	1
F.P. 100%	İ		°C/mm		_	g*K	ļ	1
B. P. *C 760 mm	159.		25°C BP	4.2134 0.05239	5	h		
100	92.8	2 4	t _e	0.0375	5	f' to		
30 10	63.7 41.5	4 5	30 mm	0.7284	5	g' ' <u>°</u> K	-	1
i	4.5	5	ΔHm cal/g		<u> </u>	<u> </u>		+
Pressure			ΔHv cal/g 25°C	75.63	5	m to		1
mm 25°C	3.8974 1170.	5	30 mm	72.53	5	<u> </u>		
Density	†		BP te	61.67 59.70	5	m' to		
g/ml 20°C	0.749 0.745	2 2	[[a,e)	59.62	5	n' °K	-	
d ₄ 25	0.741	4	AHv/T _e	18,87	5	Surface tension	<u> </u>	+
•	0.765	4	d 64 to		5	dynes/cm. 20°C	24.87	5
<u> </u>	-0.038	4	d' 25 to	77.63	5	30 40	23.82 22.80	5
Ref. Index n _D 20°C	1.420	2	e' 64 °C		5	Parachor [P]		+-
D 25 30	1.418 1.415	2 4	d g/ml v ml/g	0.234 4.281	5	20°C		
"C"	0.7456	4	tc °C	327.	5	30 40	-	1
MR (Obs.)	48.1	2	P _c mm	14748.	5		424.2	5
MR (Calc.)	48.38	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		1
(nD-d/2)	1.046	2	30 mm	1.0000	5	Dispersion	97.	2
Dielectric A 64 to	2.016 6.8921	5	BP t _e	0.9475 0.9336	5	Flash Point °C		
B ¿?13 ℃	1475.	5	t _c	0.24	5	Fire Point	 	+
C	209.	5	ΔHc kcal/m ΔHf		İ	M Spec. Ultra V.	ĺ	1
A* 64 to B* 187 °C	1.39567 1386.2	' 5 5	ΔFf		<u> </u>	X-Ray Dif. Infrared		
к — — —			Viscosity			Solubility in +	<u> </u>	+
\$	1		centistokes °C	:		Acetone		İ
*x 1			•			Carbon tet. Benzene	!	
A' 25 to B' 64 °C	7.24984 1676.7	5 5				Ether n-Heptane	İ	
C'	227.	5	B ^V to			Ethanol		
A'* 25 to B'* 64 °C	1.73581	5	AV °C	-		Water Water in	1	
Ac 213 to	7,6877	5	(B ^V) to	1				\top
Bc t °C	2227.	5				1		
Cc	306.	5	•					
Cryos, A° consts, B°			c _p vap. °K					
t _e °C	177.00	5	c _w vap.		L		<u> </u>	
$T_{\mathbf{R}} = 0.81$						f grams/100 gra		<u>ıt</u>
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE: PURIFICAT	ION:	AI						
LITERATUI	· · · · · · · · · · · · · · · · · · ·							
			••					

No. 125 4-Ethyl-2, 2-dimethylhexane NAME STRUCTURAL FORMULA CH₃ CH2CH CH2CH3 Mole Ref. Molecular Molecular ĊН3 Ċ₂H₅ $C_{10}H_{22}$ Weight 142, 276 % Pur Formula Ref. Ref. Ref. F. P. °C dt/dP f to F.P. 100% °C/mm 25°C °K g 2.6651 B. P. °C h BP 0.05084 5 760 mm 147. 2 ^te 0.0371 5 f١ to 100 82, 7 4 <u>°к</u> g' 30 54.4 4 30 mm 0.7080 5 10 32.8 5 h! ∆Hm cal/g 5 -3,2 1 m to AHv cal/g Pressure °K n 25°C 72.79 5 mm 25°C 6.4021 5 5 o 30 mm 70.57 te 1140. BP 60,28 5 m' to Density 58.50 5 n' g/ml 20°C te (d, e) °K_ 0.733 5 58.46 ٥' dt 4 25 0.729 2 ΔHv/T_e 19.07 5 30 0.725 4 Surface tension 54 76.62 5 to 0.749 4 dynes/cm. 20°C 22.81 163 °C 0.1112 5 ь -0.038 4 30 21.82 5 25 to 74.67 5 40 20.87 5 Ref. Index e' 54 °C 0.0752 5 20°C 1.4131 [P] nD Parachor d_c g/ml 5 0.235 2 25 1.4107 20°C vc ml/g t °C 4, 249 30 4 1.4083 30 t_c 308. 5 40 "C" 0.7501 4 P_c mm 14390. 5 Sugd. 424. 2 5 MR (Obs.) 48.4 2 PV/RT Exp. L. 1. %/wt. MR (Calc.) 48.38 25°C 1,0000 5 (nD-d/2) 1.0466 2 1.0000 30 mm 5 Dispersion 97. 2 Dielectric 1.997 5 BP 0.9510 Flash Point °C 0.9384 5 54 to 6.9052 5 Fire Point 0.24 B 1192 °C 5 1441. M. Spec. Ultra V. C 211. 5 AHc kcal/m ΔHf A* 54 to 1.41441 5 X-Ray Dif. ΔFf B* 173 °C 1352.4 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. °C $\mathbf{t_x}$ Benzene 25 to 7.27326 Ether B١ 5<u>4 °C</u> 1643.1 5 n-Heptane B_v C 229. 5 to Ethanol ۰c Water 25 to 1.76324 5 54 °C Water in B'* (BV) 1543.0 5 to Ac 192 to 7,6092 5 (A^V)| °C Bc tc °C 2068. °C c_p liq. °K Cc 291. 5 Cryos. Aº c_p vap. °K consts. B° te °C c vap. 163.38 5 $= 0.80 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

No. 126 STRUCTURAL FORMULA NAME 3-Ethyl-2, 3-dimethylhexane C2H5 сн₃сн с (CH₂)₂CH₃Mole Ref. Molecular сн3сн3 Molecular $C_{10}H_{22}$ Weight 142.276 % Pur Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP ſ to °C/mm °K g 25°C 6.3714 5 B. P. °C 0.05337 5 BP 760 mm 2 169. t_e 5 ſ١ 0.0375 to 100 101.5 g¹ °K 30 5 71.8 4 30 mm 0.7432 10 49.2 5 h' ∆Hm cal/g 5 1 11.4 to AHv cal/g Pressure °K n 25°C 78.52 mm 25°C 2.4826 5 o 30 mm 74.56 5 t_e 1197. 5 BP 63.33 5 to Density g/ml 20°C m 5 te (d, e) 61.19 °K 0.765 2 61.09 5 01 $\mathbf{d_{4}^{t}}$ 0.760 2 AHV/Te 18.86 5 30 0.755 4 Surface tension 82,85 72 to 5 0.785 4 .27.06 a dynes/cm. 20°C <u>188</u> ℃ 0.1155 Ъ 4 5 -0.021 30 25.67 ď to 80.63 5 25 5 40 24.34 Ref. Index e' 72 °C 5 0.0846 ď 20°C 1.427 2 [P] Parachor d_c g/ml 5 0.225 25 2 1.424 20°C ml/g v_c 4.447 5 30 1.421 4 30 °C t_c 339. 5 40 "C" 0.7415 4 P_c mm 5 5 14482. Sugd 424.2 MR (Obs.) 47.8 2 5 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 (nD-d/2) 1.045 2 30 mm 1.0000 5 Dispersion 2 97. Dielectric 2.036 5 0.9470 BP 5 Flash Point °C 0.9323 5 A 72 to 6.9053 5 Fire Point t_c 0.24 B 1229 °C 1513. 5 M Spec. C 207. AHc kcal/m Ultra V ΔHf A* 72 to 1.40029 5 X-Ray Dif. ΔFf B* 198 °C 1422.1 Infrared ĸ Viscosity Viscour, centistokes °C Solubility in Acetone to Carbon tet. •c Benzene to A' 25 7.25592 5 Ether В' <u>.c</u> 1714.5 5 n-Heptane вŸ 225. Ethanol $\hat{\boldsymbol{A^{\boldsymbol{v}}}}$ °C 25 to 1.73827 Water Water in B'* 72 °C 1614.2 5 (BV) to Ac | 229 to 7.7601 (A^V) °C Bc L 2347. _tc__ °С 5 сp liq. ۰ĸ Cc 314 5 Cryos. A° c_p vap. °K consts. B° t_e °C c, vap. 188.39 $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc, by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 127
NAME	4-Ethyl-2,	3-di	methylhexane			STRUCTURAL	FORMULA
						CH ₃	
ļ ————————————————————————————————————						сн ₃ сн сн сн	сн,сн,
Mole		lecul	ar C ₁₀ H ₂₂	Molecular	74	Ċн ₃ с ₂ н	
% Pur.	Fo	rmul		Weight 142.2		T	
<u> </u>		Ref		 	Ref.	l	Ref
F.P. °C F.P. 100°	<u>.</u>	┼	dt/dP °C/mm			f to	
B. P. °C	<u> </u>	┼	25°C	5.1760	5	g '° <u>K</u>	
760 mm	164.	2	BP	0.05288	5	h	
100 30	97.1	4	t _e	0.0374	5	f' to	1
10	67.7 45.3	4 5	30 mm	0.7358	5_	h'=	
1	7.9	5	ΔHm cal/g			m to	
Pressure			ΔHv cal/g 25°C	77.06	5	n K	
mm 25°C	3.1136 1184.	5	30 mm	73.54	5	•	
Density	1101.	+-	BP	62.56	5 5	m¹ to	
g/ml 20°0	0.759	2	te te (d, e)	60.51	5	n' K_	
d ₄ 25 30	0.755	2	AHV/T	18.88	5		
	0.751	4	d 68 to	81,27	5	Surface tension	
a b	0.775 -0.0 ₃ 8	4	_e_ _183 °C	0.1141	5	dynes/cm. 20°C	26.22 5 25.13 5
Ref. Index		Ť	d' 25 to e' 68 °C	79.12 0.0824	5	40	24.07 5
n _D 20°0		2	d _c g/ml	0.236	2	Parachor [P]	
25	1.422	2 4	v _c mi/g	4.233	5	20°C 30	
"C"	0.7424	4	tc°℃	336.	5	40	
MR (Obs.			P _C mm	15139.	5	Sugd.	424.2 5
MR (Calc.		5	PV/RT 25°C		_	Exp. L.l.%/wt.	
(nD-d/2)	1.044	2	30 mm	1.0000	5	u. Dispersion	97. 2
Dielectric		5	BP	0.9480	5	Flash Point C	
A 68 to		5	te tc	0.9337 0.24	5	Fire Point	
B 1227 °C	7 1494. 208.	5	ΔHc kcal/m		-	M. Spec.	
A* 68 to	1,39669	5	ΔHf	1	1	Ultra V. X-Ray Dif.	
B* 193 °C		5	ΔFf			Infrared	
K c			Viscosity centistokes			Solubility in +	
t _k -to	-		7 °C			Acetone	
'x			,			Carbon tet. Benzene	
A' 25 to B' 68 °C	7. 25291	5				Ether	
c' -00	1695.6 226.	5	B ^V to C			n-Heptane Ethanol	
A'* 25 to	1.73705	5				Water	
B'* 68 °C	1595.4	5	(B ^V) to			Water in	
Acl 227 to		5	(A ^V) °C				
Bc tc C	; 2313. - 314.	5	c _p liq. °K				
Cryos, A		1	c _p vap. °K				
consts. B		L	ll .			l	
te °C	182.73	5	c _v vap.	}			
$T_{R} = 0.8$	32 T _c					grams/100 gra	ms solvent
		2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for	
SOURCE:		AI	PI				
PURIFICA	TION:	Al	PI				
LITERATU	RE REFERE	NCE	S:				
1							
1							

							No. 128	3		
NAME	3-Ethyl-2	, 4-di	methylhexane			STRUCTURAL FORMULA				
					\neg	C ₂ H ₅				
					\dashv	сизси си си	CH,CH	ı		
Mole	Ref. Mo	lecul	ar C u	Molecular		с́н ₃ с́н		•		
% Pur.	Fo	rmul	a C ₁₀ H ₂₂	Weight 142.2	76	3				
 		Ref.		_	Ref.			Ref.		
F.P. °C			dt/dP			f to				
F.P. 100%	↓	L	*C/mm 25*C	5.1760	5	g <u>*K</u>				
B. P. °C 760 mm	164.	2	BP	0.05288	5	h				
100	97.1	4	t _e	0.0374	5	f' to				
30 10	67.7 45.3	4 5	30 mm	0,7358	5	g' K_				
1	7.9	5	ΔHm cal/g			h'		ļ		
Pressure		†	ΔHv cal/g			m to				
mm 25°C	3, 1136	5	25°C 30 mm	77.06 73.54	5 5			ļ		
t _e	1184.	5	BP	62.56	5	m' to				
Density g/ml 20°C	0.759	2	te (d a)	60,51	5	n' _ °K				
	0.755	2	le (a'e)	1	ł i	0'		l		
d ₄ 25	0.751	4	ΔHv/T _e	18.88	5	Surface tension				
a .	0.775	4	d 68 to		5	dynes/cm. 20°C	26.22	5		
ь	-0.038	4	d' 25 to	79.12	5	30 40	25.13 24.07	5		
Ref. Index	1.424	2	e' 68 °C		5	Parachor [P]		 -		
D 25	1.422	2	d _c g/ml	0.236	5	20°C		l		
30	1.419	4	tc °C	4. 233 336.	5	30		1		
"C"	0.7424	4	P _c mm	15139.	5	40 Sugd.	424.2	5		
MR (Obs.)	47.8	2	PV/RT	1	Ħ	Exp. L.1.%/wt.		 		
MR (Calc.) (nD-d/2)	1.044	5 2	25°C	1.0000	5	u.				
Dielectric	2,028	5	30 mm BP	1.0000 0.9480	5	Dispersion	97.	2		
A 68 to		5	t_	0.9337	5	Flash Point °C Fire Point		l		
B [227 °C		5	tc	0.24	5			├		
С	208.	5	ΔHc kcal/m ΔHf	1		M Spec. Ultra V.		l		
A* 68 to		5	ΔFf			X-Ray Dif.		İ		
B*	1403.7	,	Viscosity			Infrared		 		
°	_		centistokes			Solubility in + Acetone		1		
t _x to t _x °C		1	η •⊂			Carbon tet.				
A' 25 to		5		-		Benzene Ether		1		
B' _ 68 °C	1695.6	5	<u> </u>			n-Heptane		ĺ		
C'	226.	5	B ^V to			Ethanol Water				
A'* 25 to B'* 68 °C		5		4		Water in				
Ac 227 to	 	5	1.0.1							
Bc t °C	2313.	5						1		
Cc — — -	314.	5	c _p liq. °K					l		
Cryos. A° consts. B°			c _p vap. °K							
t _e °C	182.73	5	c _v vap.							
$T_R = 0.87$						grams/100 gram	ns solven	t		
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula			
SOURCE:			PĪ							
PURIFICAT			PI				-			
LITERATU	RE REFERE	NCES	3:							

					т		No. 1	
NAME _	4-Ethyl-2,	4-di:	methylhexane		STRUCTURAL FORMULA			
				C ₂ H ₅				
			1			сн,сн сн,с	сн ₂ сн	.
Mole	Ref. Mo	lecul:	ar C H 1	Molecular	1	сн, сн		3
% Pur.	Fo	rmula	ar C ₁₀ H ₂₂	Weight 142.2	76		3	
		Ref.			Ref.			Ref.
F. P. °C			dt/dP			f to		
F.P. 100%			°C/mm		_	g ° <u>K</u>		
B. P. ℃			25°C BP	4.0445 0.05230	5	h		
760 mm 100	158.	2	t _e	0.0374	5	f' to		
30	62.9	4	30 mm	0.7269	5	g' :° <u>K</u>		
10	40.7 3.8	5 5	∆Hm cal/g			h'		
	1 3.0	-	ΔHv cal/g		1	m to		
Pressure mm 25°C	4.0755	5	25°C	75.35	5	n <u>*K</u>		
t _e	1168.	5	30 mm BP	72.33 61.56	5			
Density			t _e	59.61	5	m' to		1
g/m1 20°C		2	t _e (d, e)	59.53	5	n' <u>°K</u> _		
dt 25 4 30	0.743 0.739	2 4	ΔHv/T _e	18.88	5	<u> </u>		+-
a	0.763	4	d 63 to	79.44	5	Surface tension dynes/cm. 20°C	24.60	5
រិ	-0.038	4	e 176 °C to	0.1132	5	8 30	23.56	5
Ref. Index			e' 63 °C	77.34 0.0797	5	40	22,55	5
ⁿ D 20°C		2	d _c g/ml	0.233	5	Parachor [P]		
25 30	1.417	2 4	V _C m1/g	4.292	5	20°C 30		
"C"	0.7459	4	t _c °C	326.	5	40		
MR (Obs.)	48,1	2	P _c mm	14685.	5	Sugd.	424.2	5
MR (Calc.)		5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.045	2	25°C 30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric	2.013	5	BP	0.9485	5	Flash Point C	/	┿
A 63 to	6.8907	5	t _e	0.9347	5	Fire Point		
B 212 °C C	_ 1472. 209.	5	t _c ΔHc kcal/m	0.24	-	M. Spec.		1
A* 63 to	1.39351	5	ΔHc kcai/in			Ultra V.		-
B* 186 °C		5	ΔFf		<u> </u>	X-Ray Dif. Infrared		İ
к — — —			Viscosity			Solubility in +		_
t _k – to	-		centistokes 7°C			Acetone		
t _x ' °C			1			Carbon tet. Benzene		
A' 25 to	7.24922	5		ł		Ether		
B' 1_63 °C	1672.9	5 5	B _v to		-	n-Heptane		
A!* 25 to			B to C		1	Ethanol Water	!	i
B!* 63 °C	1.73555 1572.8	5	(B ^V) to	İ	1	Water in		
Ac 212 to	7,6778	5	(A ^V) °C	1				
Bc _i t _c ℃	2211.	5			1	1		
Cc	304.	5	P -					
Cryos. A° consts. B°			c _p vap. °K	1				
t _e °C	175.91	5	c vap.				}	
$T_R = 0.8$		ر ا	L	L	L	+ ~~~ - /100		
	ES: 1-Dow	2 4	PI 3-Lit. 4-	Cala (1		grams/100 gra		nt
SOURCE:	, E3; 1-D0W	AP		Caic. Irom de	aa	ata 5-Calc. by for	mud	
PURIFICAT	TON	AF						
							·····	
LILEKATU	RE REFERE	NCES	> :					
L								

No. 130 STRUCTURAL FORMULA NAME 3-Ethyl-2, 5-dimethylhexane C_2H_5 CH3CH CH CH2CH CH3 Molecular C₁₀H₂₂ Ref. Ċнз Mole Molecular Ċна Weight 142.276 % Pur Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 3.8827 B. P. *C h BP 0.05220 5 760 mm 157. 2 5 f' 0.0374 te to 100 91.0 4 °K g' 30 4 30 mm 5 62.0 0.7254 10 39.9 5 h! ∆Hm cal/g 1 3.1 5 to ΔHv cal/g Pressure n °K 25°C 75.06 mm 25°C 4,2614 5 o 30 mm 72.13 61.39 5 1165. 5 t_e 5 BP Density g/ml 20°C to m 59.5 5 te te °K n' 0.741 2 (d, e) 59.38 5 01 25 0.737 $\mathbf{d_{4}^{t}}$ AHv/Te 18.88 5 30 0.733 4 Surface tension 79.14 62 to 5 0.757 4 dynes/cm. 20°C 23.82 175 °C 0.1131 5 ь -0.038 4 30 5 22.81 ď٠ 77.04 5 5 40 21.82 e' 62 °C Ref. Index 0.0792 5 20°C 1.416 2 [P] $\mathbf{n}_{\mathbf{D}}$ Parachor d g/ml vc ml/g 0.231 5 1.414 25 2 20°C 4.328 5 4 30 1.411 30 °C tc 323 5 40 "C" 0.7469 4 P_c mm 14490. 5 424.2 5 Sugd. MR (Obs.) 48.2 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 5 1.0000 25°C (nD-d/2) 1.046 2 30 mm 1,0000 5 2 Dispersion 97. Dielectric 2,005 5 0.9485 BP 5 Flash Point °C 0.9348 A | 62 to 5 6.8894 5 Fire Point tc 0.24 5 [210 °C 1468. M Spec. C 209. 5 AHc kcal/m Ultra V. ΔHf A* 62 to 1.39312 5 X-Ray Dif. ΔFf B* 185 °C 1378.6 Infrared ĸ Viscosity Viscour, centistokes °C Solubility in c Acetone t_x to Carbon tet. °C Benzene 25 to 7,24860 Ether B' ∟<u>62 °C</u> 1669.1 5 n-Heptane вŸ CI 227. Ethanol to $\hat{\mathbf{A}^{\boldsymbol{V}}}$ 1.73529 °C Water 25 to 62 °C 5 Water in B'* 1569.0 (BV) to Ac | 210 to 7.6637 (AV) ۰c 2190. 5 Bc _tc_ °C cp liq. ۰ĸ Cc 302 5 Cryos. A° consts. B° c_p vap. ۰ĸ te °C c, vap. 174.77 5 $T_R = 0.81 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME	4-Ethyl-3, 3-dimethylhexane					STRUCTURAL FORMULA				
Mole % Pur.	Ref. Molecular C10H22 Molecular Weight 142.276					сн ₃ сн ₂ с сн сн ₂ сн ₃				
		Ref.			Ref.				Ref.	
F. P. °C			dt/dP			f	to		\top	
F.P. 1009			°C/mm	1		gl	*K		1 1	
B. P. °C	<u> </u>		25°C	5.3948	5	h				
760 mm	165.	2	BP	0.05298	5				+	
100	98.0	4	t _e	0.0374	5	f'	to			
30	68.5	4	30 mm	0.7373	5	g'	' <u>*K</u>		1 1	
10	46.1 8.6	5	ΔHm cal/g			_h' i				
	- 0.0	- 3	AHv cal/g		$\dagger \lnot$	m	to			
Pressure mm 25°C	2 0743	_	25°C	77.35	5	n	<u>•K</u>		1 1	
t _e	2.9762	5	30 mm	73.75	5	ľi				
Density	+	+-	BP	62.74	5	m'	l to			
g/ml 20°C	0.764	2	te te (d, e)	60.67	5 5	n'	<u>•</u> K_		1 1	
	0.760	2			i I	0'				
dt 25 4 30	0.756	4	ΔHv/T _e	18.89	5	S	face tension		+	
a	0.780	4	d 69 to	81.57	5		es/cm. 20°C	26.92	5	
∙ъ	-0.038	4	_e1 <u>84</u> °C	0,1142 79,42	5	8	30	25.81	5	
Ref. Index	:		e' 69 °C	0.0828	5		40	24.73	5	
n _D 20°C		2		†	5	Par	achor [P]			
1 23	1.425	2	d _c g/ml	0.238 4.200	5		20°C			
. 30	1.422	4	v _c ml/g t _c °C	338.	5		30			
_"C"	0,7424	4	P _c mm	15307.	5	l	40 Sugd	424.2	5	
MR (Obs.)		2	PV/RT	13301.	+-1	<u> </u>		101.0	+	
MR (Calc.		5	25°C	1,0000	5	Exp	. L.1.%/wt. u.	ļ		
(nD-d/2)	1.045	2	30 mm	1.0000	5	Dist	persion	97.	2	
Dielectric	2.036	5	BP	0.9480	5		sh Point C		+	
A 69 to	6.9001	5	t _e	0.9337	5		e Point			
B 1228 °C		5	tc	0.24	5		Spec.		+	
С	208.	5	ΔHc kcal/m ΔHf	1			a V.			
A* 69 to			ΔFf				lay Dif.		1	
B*[194 °C	1407.3	5	Viscosity		\vdash	Infr	ared			
c			centistokes]	li		ıbility in +		1	
t _k T to			ŋ °c				etone rbon tet.			
t _x °C	;	1	•				nzene		1	
A' 25 to							her	<u> </u>		
B' 1_69 °C		5	B _v to		+		Heptane			
	226.	5	A to				hanol			
A'* 25 to			I⊢.⊸v.— — — -	-			ter ter in	1		
B'* 69 °C		5		1					+-	
Ac 228 to	7.7571	5	(A ^V) °C		\sqcup			1		
Bc tc °C	2336. 316.	5	c _p liq. °K	1					1	
	+		11							
Cryos. A ^c consts. B ^c			c _p vap. °K	1						
	+	5	c, vap.	1						
	183.87		I. •	J	Ш	L		L		
$T_{R} = 0.8$							ams/100 gra		nt	
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula		
SOURCE:		AI	PI							
PURIFICA	TION:	AI	PI							
LITERATURE REFERENCES:										
1										
1										
1										

							No. 132	2		
NAME	3-Ethyl-3, 4-dimethylhexane					STRUCTURAL FORMULA				
					С ₂ н ₅					
			T		\neg	сн ₃ сн ₂ с с		3		
Mole % Pur.	Ref. Mo	lecul rmul	ar C ₁₀ H ₂₂	Molecular Weight 142.2	76	Ċн ₃ Ċн	¹ 3			
70.2.3.1		Ref.			Ref	i		Ref.		
F. P. °C	T		dt/dP	Τ		f to				
F. P. 1007	5		*C/mm		i _ i	g*K_				
B. P. *C	170		25°C BP	6.6033 0.05354	5	h				
760 mm 100	170. 102.3	2 4	te	0.0375	5	f' to				
30 10	72.5	4 5	30 mm	0.7452	5	g' K_		1		
1	49.8 12.0	5	ΔHm cal/g	L		h'		<u> </u>		
Pressure			ΔHv cal/g 25°C	78.71	5	m to				
mm 25°C	2.3895 1200.2	5	30 mm	74.67	5	•				
Density	1200.2	-	BP	63.45 61.30	5	m' to				
g/ml 20°0		2	te te (d, e)	61.19	5	n' °K_				
dt 25	0.768 0.764	2	AHv/Te	18.85	5					
1 50	0.788	4	d 73 to		5	Surface tension dynes/cm. 20°C	28.07	5		
b	-0.038	4	e 190 °C		5	30	26.92	5 .		
Ref. Index			e' 73 °C		5	40	25.81	5		
n _D 20°0	1.431	2 2	d _c g/ml	0.239	5	Parachor [P] 20°C				
30	1.426	4	v _c ml/g t _c °C	4.183 346.	5	30		l		
"C"	0.7412	4	P _c mm	15572.	5	40 Sugd.	424.2	5		
MR (Obs. MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.				
(nD-d/2)	1.045	2	25°C 30 mm	1.0000	5	u. Dispersion	97.	2		
Dielectric	2.048	5	BP	0.9470	5	Flash Point °C	71.	-		
A 73 t		5	t _e t _c	0.9322 0.24	5	Fire Point		<u> </u>		
B (235 °C	207.	5	ΔHc kcal/m	+		M Spec.				
A* 73 to		5	ΔHf ΔFf			Ultra V. X-Ray Dif.				
B* L200 °	2 1423.7	5	Viscosity	 	-	Infrared		ļ		
·	_		centistokes			Solubility in + Acetone				
t _k to			7 ℃			Carbon tet.				
A' 25 to		5		1		Benzene Ether		l		
B' _ 73 °	21716.1 225.	5	B ^V to	-	-	n-Heptane				
A'* 25 to		5	B to	1		Ethanol Water				
B'* 73 °		5	(BV) to	-1		Water in				
Ac 235 to	7.8101	5	(A ^V) °C	ı						
Bc tc_°	2428. - 325.	5	c _p liq. °K					ļ		
Cryos. A	•		c _p vap. °K					1		
consts. B			•							
t _e °C	189.56	5	c _v vap.	<u> </u>		<u> </u>				
T _R = 0.8		•				grams/100 gram		t		
	CES: 1-Dow	2-AI		Calc. from det	da:	ta 5-Calc. by for	mula			
SOURCE:	TION		PI PI							
	PURIFICATION: API LITERATURE REFERENCES:									
	KEFERE		••							

							No. 1	33
NAME	2, 2, 3, 3-7	[etra:	methylhexane	STRUCTURAL FORMULA CH ₃ CH ₃				
-						сн₃ċ ċ (с	CH,),CH,	
Mole	Ref. Mo	lecul	ar C ₁₀ H ₂₂	Molecular		с́н ₃ с́н ₃	2.2 3	
% Pur.	Fo	rmula	1022	Weight 142.	276			
		Ref.			Ref.			Ref.
F.P. °C	-54.00	2	dt/dP		1	f to		1
F.P. 100%	1		°C/mm	1 21/2	_	g ° <u>K</u>	_	i
B. P. ℃			25°C BP	4. 3468 0. 05280	5	h		
760 mm 100	160.31 93.6	2	t _e	0.0376	5	f' to		T
30	64.4	4	30 mm	0.7322	5	g'° <u>K</u>	.]	1
10	42.1	5	ΔHm cal/g	1	†	h'	1	1
1	4.9	5	ΔHv cal/g	 	+-	m to		
Pressure mm 25°C	3.7784	5	25°C	75.62	5	n•K		1
t _e	1175.	5	30 mm	72.45	5	0	l	
Density		\vdash	BP	61.65	5	m' to		
g/ml 20°C	0.76446	2	te te (d, e)	59.67 59.58	5	n' <u>*</u> K	4	
d ₄ 25 30	0.76089	2	AHv/Te	18.79	5			1
	0.75732	4	d 64 to	79.70	5	Surface tension		
a b	0.77873 -0.03713	4	_e _l1 <u>79 °C</u>	0.1126	5	dynes/cm. 20°C	26.99 25.99	5
Ref. Index	-5.03.25	+	d' 25 to	77.63	5	40	25.02	5
n _D 20°C	1.42818	2	e' 64 °C	0.0804	5	Parachor [P]	†	+-
45	1.42600	2	dc g/ml	0.241 4.150	5	20°C		
30	1.42390	4	vc ml/g tc °C	334.	5	30		
"C"	0.7439	4	P _c mm	15392.	5	40 Sugd	424.2	5
MR (Obs.)	47.900	5	PV/RT	+	+	Exp. L.1.%/wt.		+-
MR (Calc.) (nD-d/2)	48.38 1.04595	5 2	25°C	1.0000	5	u.		
Dielectric		+	30 mm BP	1.0000	5	Dispersion	97.	2
A 64 to	2.040 6.8728	5	t	0.9485 0.9345	5	Flash Point C		
B 225 ℃	1472.	5	t _c	0.24	5	Fire Point		↓
c ———	209.	5	∆Hc kcal/m			M. Spec. Ultra V.		
A* 64 to	1.37326	5	ΔHf ΔFf	1		X-Ray Dif.		
B* 189 °C	1382.5	5			-	Infrared		
c			Viscosity centistokes			Solubility in +		
t _k Tto			η °C			Acetone Carbon tet.		
t _x °C						Benzene		
A' 25 to B' 64 °C	7.22868 1673.1	5	}			Ether		
B' 1_64_°C	227.	5	B ^V to			n-Heptane Ethanol		
A'* 25 to	1.71446	5	B ^V to C			Water		
B'* 64 °C	1573.0	5	(B ^V) to			Water in		1
Acl 225 to	7.7086	5	(A ^V) °C					
Bc tc °C	2282.	5	c _p liq. °K					
Cc	314.	5						
Cryos, A° consts, B°			c _p vap. °K				1	
t _e °C	178,66	5	c vap.					1
$T_{\mathbf{R}} = 0.82$		L 3	V -	L		+	<u> </u>	
						grams/100 gra		nt
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for	rmula	
SOURCE:		AI						
PURIFICAT		AI						
LITERATU	RE REFERE	NCES	S:					
			····					

							No. 13	4
NAME	2, 2, 3	, 4-Tetra	methylhexane			STRUCTURAL	FORMUL	——— А
<u> </u>			· · · · · · · · · · · · · · · · · · ·			CH ₃ CI	I ₂	
1						сн, с сн с		ı
Mole	Ref.	Molecu		Molecular		3 сн3 сн3	2 3	,
% Pur.		Formu	1	Weight 142.2	_			
		Ref		·	Ref.			Ref.
F.P. °C F.P. 100%			dt/dP			f to	1	Į
B. P. *C	 		*C/mm 25*C	3, 4772	5	g <u>*K</u>	·	1
760 mma	154.9	2	BP	0.05231	5	h		┼
100	88.9	4	t _e	0.0377	5	f' to		
30 10	59.9 37.8	4 5	30 mm	0.7244	5	h'	1	
ì	1.1	5	ΔHm cal/g			 	 	+
Pressure	1		ΔHv cal/g 25°C	74.03	_	m to		
mm 25°C	4.82		30 mm	74.03 71.31	5		1	
Te	1157.	5	BP	60.57	5	m' to	 	†
Density g/ml 20°C	0.75	48 2	te (d, e)	58.67 58.59	5	n']	
t 25	0.75	08 2	ΔHv/Te	18.73	5	o'		ł
⁴ 4 30	0.74		d 60 to		5	Surface tension		
a b	0.77		e 172 %		5	dynes/cm. 20°C	25.65	5
Ref. Index	-0.03	0 7	d 25 to	75.98	5	30 40	24.57	5
n _D 20°C	1.42	26 2	e' j 60 °C	+	5	Parachor [P]		
2.5	1.42		d g/ml vc ml/g	0.233 4.293	5	20°C		Ì
30	1.41		tc °C	323.	5	30 40	ł	1
"C"	0.74		P _c mm	14608.	5		424.2	5
MR (Obs.) MR (Calc.)	47.96		PV/RT	 		Exp. L.1.%/wt.	†	T
(nD-d/2)	1.04		25°C	1.0000	5	u.		
Dielectric	2.02	4 5	30 mm BP	1.0000 0.9470	5	Dispersion	98.	2
A 60 to	6.86	21 5	t _e	0.9333	5	Flash Point °C Fire Point	}	
B [210 °C		5	t _c	0,24	5	M Spec.		+-
C	210.	5	ΔHc kcal/m	1		Ultra V.	İ	
A* 60 to B* 182 °C		036 5	ΔFf	-	1	X-Ray Dif. Infrared		
к ———			Viscosity			Solubility in +	-	+-
cto	-		centistokes °C	.		Acetone		1
tk to			7	`		Carbon tet. Benzene		
A' 25 to			1			Ether		
B', ∟ 60 °C		5	B ^V to		 	n-Heptane	j	
	228.		∦ B' ∤ to			Ethanol Water	1	
A'* 25 to B'* 60 °C		956 5	(B ^V) to	- l		Water in		
Ac 210 to	7,63	65 5	(A ^V)	1		H		
Bc t °C	2174.	5	c _p liq. °K		† i	1	1	
Ce	304.	5	-1,				1	
Cryos. A° consts. B°			c _p vap. °K	•				Ì
t _e °C	172.42	5	c _v vap.					
$T_R = 0.8$	l T _c					+ grams/100 gra	ms solver	nt
REFERENC	ES: 1-D	ow 2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:		A	PI					
PURIFICAT	ION:	A	PI					
LITERATU	RE REF	ERENCE	S:					

							No. 13	<u> </u>		
NAME	2, 2, 3, 5-7	Cetra:	methylhexane		1	STRUCTURAL FORMULA				
i F						CH ₃				
}L						сн3с сн с	, cu cu			
Mole	Ref. Mo	lecul	ar 1	Molecular	l		-,			
% Pur.		rmule	C ₁₀ H ₂₂	Weight 142.2	76	ĊН ₃ ĊН ₃	CH ₃			
		Ref.		T Total	Ref.	T		Ref.		
F. P. °C		1	dt/dP		1		T			
F.P. 100%	1	 	*C/mm	i	l	f to				
B. P. °C			25°C	2.6618	5		1			
760 mm	148.4	2	BP	0.05176	5	h	 	-		
100	83.11	4	t _e	0.03783	5	f' to	[[
30 10	54.49 32.73	4 5	30 mm	0.7153	5_	g' <u>°K</u>	·			
1 10	-3.54	5	∆Hm cal/g		l	h'				
Pressure		1	ΔHv cal/g]	!	m to	1 1			
mm 25°C	6.4707	5	25°C	72.10	5	n <u>•K</u>	-			
t _e	1138.6	5	30 mm BP	69.89	5					
Density			t _e	59.34 57.55	5	m' to				
g/m1 20°C		2	te (d, e)	57.48	5	n' K	-			
dt 25 4 30	0.7336	2	AHv/Te	18.69	5					
	0.7294	4	d 55 to	76,01	5	Surface tension				
a b	0.755 -0.0 ₃ 84	4 4	_e 1_165 °C	0.1123	5	dynes/cm. 20°C	23.41	5		
	-0.0304	1	d' 25 to	73.98	5	8 30 40	22.36 21.34	5		
Ref. Index	1.4142	2	e' 55 °C	0.0750	5	Parachor [P]	 	Ť		
¹¹ D 25	1.4117	2	d _c g/ml	0.225	5	20°C				
30	1.4092	4	v _c ml/g t °C	4.444 310.	5	30				
"C"	0.7471	4	-	ľ	l	40		_		
MR (Obs.)	48,21	2	P _c mm	13805.	5		424.2	5		
MR (Calc.)	48.38	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.				
(nD-d/2)	1.0453	2	30 mm	1.0000	5	u. Dispersion	98.	2		
Dielectric	2.000	5	BP	0.9470	5	Flash Point °C	1	F		
A 55 to	6.8462	5	t _e	0.9338	5	Fire Point				
B (199 °C		5	t ^e c	0.24	3	M. Spec.		 		
	211.	5	ΔHc kcal/m ΔHf	1		Ultra V.		ŀ		
A* 55 to B* 175 °C	1.36082	5	ΔFf		-	X-Ray Dif.				
K L S	. 1337.3]]	Viscosity			Infrared	-	ļ		
c	.1		centistokes	[1	Solubility in T				
t _k Tto	İ		ກ		,	Acetone Carbon tet.				
'x						Benzene				
A' 25 to B' 55 °C	7.21049	5		1	l	Ether		İ		
B' _55 °C	1624.23 229.	5	B ^V to			n-Heptane Ethanol				
A'* 25 to	1,70077	5	B ^V to A °C			Water]	1		
B'* 55 ℃	1524.4	5	(B ^V) to		1	Water in		<u></u>		
Ac 199 to	7,56018	5	(A ^V) °C							
Bc tc °C	2064.8	5		 	 	1		1		
Cc	293.8	5	c _p liq. °K	ľ	1	N .	1	1		
Cryos. A°			c _p vap. °K					1		
consts. B°	_		l -							
t _e °C	165.04	5	c _v vap.	L	<u></u>			L		
$T_{\mathbf{R}} = 0.81$	I T _C					grams/100 gra	ms solven	t		
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for	rmula			
SOURCE:		AI	PI		-					
PURIFICAT	ION:	AI	PI							
	RE REFERE									
			••							

No. 136 2, 2, 4, 4-Tetramethylhexane STRUCTURAL FORMULA NAME CH₃ CH₃ сн₃ċ сн₂с CH2CH3 Molecular Weight 142, 276 Mole Ref Molecular ĊH3 ĊНз C10H22 Formula % Pur Ref. Ref Ref. dt/dP F.P. *C F.P. 100% f to *C/mm 25*C <u>°K</u> g 3.2530 5 B, P. °C h BP 0.05218 760 mm 153.3 2 0.0378 5 f t_e to 100 87.44 4 g' °K 30 58.55 30 mm 0.7222 5 10 36.58 5 h† ∆Hm cal/g 1 -0.06 m to AHv cal/g Pressure n °K 25°C 73.54 mm 25°C 5.1912 o 30 mm 70.95 1152.6 5 t_e BP 60.26 5 to m' Density te (d, e) 58.38 5 g/ml 20°C n' °K 0.7470 58.30 5 o' ď4 25 0.7428 2 ΔHv/Te 5 18.72 30 4 0.739 Surface tension d 77.56 5 59 to ℃ . 0.7638 4 dynes/cm. 20°C 24.60 _171 °C 0.1129 e ь -0.0384 4 30 23.51 5 āï 75.47 1 5 40 22.45 Ref. Index e' 0.0772 5 **n**D [P] 20°C 1.4208 2 Parachor d_c g/ml 0.228 5 25 1.4183 2 20°C ml/g 5 4.385 30 1.4159 4 c 30 $\mathbf{t_c}$ 318. 5 40 "C" 0.7489 4 mm 14184. 5 Sugd. 424.2 5 MR (Obs.) 48.28 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 48.38 25°C 1,0000 (nD-d/2) u. 1.0473 2 30 mm 1.0000 Dispersion 99. 2 Dielectric 2.019 5 BP 0.9470 Flash Point °C T 59 to 0.9334 t_e 6.85770 5 Fire Point tc 0.24 В L206 °C 1444.2 M Spec. 5 AHc kcal/m 210. Ultra V. ΔHf A* 59 to 1.36755 5 X-Ray Dif. ΔFf B* 181 °C 1356.4 Infrared Viscosity Solubility in c centistokes Acetone t_k to Carbon tet. •c Benzene A' 25 to 7.21851 Ether В' 59 °C 1644.5 5 n-Heptane 228. Ethanol ÁV °C Water A'* 1.70692 to Water in (BV) B'* 59 °C 1544.5 5 to (AV) Ac | 206 to 7.60949 5 °C Bc _tc_ 2135.9 liq. °K Cc 299.6 5 Cryos, Aº c_p vap. ۰ĸ consts. B° c_v vap. 170,60 5 $T_R = 0.81 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 137 2, 2, 4, 5-Tetramethylhexane STRUCTURAL FORMULA NAME CH₃ сн₂сн сн сн₃ Molecular Mole Ref. Molecular Ċн₃ ĊH3 C10H22 Weight 142.276 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g °K 2,6050 5 B. P. °C h BP 0.05173 5 760 mm 147.88 2 0.03784 5 f to 100 82.64 4 g' <u>°К</u> 30 mm 30 54.05 4 0.7146 5 10 32.31 5 h' ∆Hm cal/g 5 -3.93 m to AHv cal/g Pressure °K n 25°C 71.94 5 6.6267 1137.1 mm 25°C o 30 mm 69.77 5 5 t_e BP 59.25 5 m' to Density te (d, e) 5 57.47 n' <u>°K</u> g/ml 20°C 0.73546 57.39 5 ۰, **d**₄ 25 0.73161 2 ΔHv/T_e 5 18.68 30 0.7278 4 Surface tension 54 to 75.83 0.7508 5 dynes/cm. 20°C 23.12 1 <u>65</u> °C 0.1121 5 ь -0.0₃774 30 22.16 5 d٦ 25 to 73.81 5 5 40 21.23 Ref. Index e¹ 54 0.0748 5 20°C 1.41318 ^{n}D [P] Parachor d_c g/ml 0.228 5 25 1.41095 2 20°C vc ml/g t_°C 4.391 30 1.40861 4 30 $^{\mathbf{t}}_{\mathbf{c}}$ 5 310. 40 "C" 0.7477 4 P_c mm 13973. 5 424.2 5 Sugd. MR (Obs.) 48, 257 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 (nD-d/2) 1.04545 2 30 mm 5 Dispersion 99. 2 1,0000 Dielectric 1.997 5 BP 0.9470 Flash Point C 0.9339 7 54 to 6.8445 5 Fire Point 0.24 В 199 °C 1422.1 M. Spec. Ultra V. C 211. 5 AHc kcal/m ΔHf A* 54 to 1.35962 5 X-Ray Dif. ΔFf B*|175 °C 1335,1 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. $\mathbf{t_{x}}$ °C Benzene 7,20914 25 to Ether 54 °C 1621.9 n-Heptane B^V | 5 229. Ethanol °C Water A'* 25 to B'* 54 °C 1.69963 5 Water in (B^V)| 1522.1 to Acl 199 to (A^V)| 7.56096 5 °C °C Bc tc 2065.9 c_p liq. °K 294.6 Cryos. A° c_p vap. °K consts. B° t_e °C c vap. 164.45 5 $T_R = 0.81 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

								No. 13	8	
NAME	2, 2, 5	5,5-1	Tetra	methylhexane			STRUCTURAL FORMULA			
						$\neg \neg$	ÇН ₃	CH ₃		
i.							СН3С (СН2)			
Mole	Ref.	Mo	lecul		Molecular	1	Ċн ₃	Ċн ₃		
% Pur.			rmul		Weight 142.2	76	O.1. ₃	0113		
			Ref.	<u> </u>		Ref.			Ref.	
F.P. °C	-12.60		2	dt/dP			f to		T	
F.P. 100%	1			°C/mm	1	H	g L _ °K		1	
B. P. *C	1			25°C BP	1.7150 0.05083	5	h		1	
760 mm 100	137.46		2 4	t _e	0.03782		f' to			
30	45.41		4	30 mm	0.6999	5	g' '°K_	1	1	
10	24.12		5	ΔHm cal/g			h'		1	
11	-11.33		5	ΔHv cal/g	+	H	m to			
Pressure mm 25°C	10.50	0.3	_	25°C	68.95	5	n '°K_	ļ	1	
t _e	10.50	03	5	30 mm	67.53	5	<u> </u>	i		
Density	+		-	BP	57.58 55.96	5	m' to			
g/ml 20°C	0.71		2	te te (d, e)	55.92	5	n' •K_			
d ₄ 25	0.71		2	ΔHv/T _e	18.68	5	o' ¦			
	0.71		4	d 45 to	_	5	Surface tension			
a b	0.73		4 4	e 153 °C		5	dynes/cm. 20°C	21.08	5 5	
Ref. Index	-0.03	<u>·/</u>	-	d' 25 to		5	40	19.27	5	
n _D 20°C	1.40	550	2			5	Parachor [P]		+	
- 25	1.40	316	2	d g/ml vc ml/g	0.222 4.504	5	20°C			
30	1.40		4	tc °C	293.	5	30 40			
"C"	0.75		4	P _c mm	13225.	5	Sugd.	424.2	5	
MR (Obs.)	48.57		2	PV/RT	+	\vdash	Exp. L.1.%/wt.		+	
MR (Calc.) (nD-d/2)	48.38 1.04		5 2	25°C	1.0000	5	u.			
Dielectric	1.97		5	30 mm BP	1.0000 0.9510	5	Dispersion	102.	2	
A 45 to	+		5	te	0.9389	5	Flash Point °C	İ	i	
B 180 °C		,,,	5	tc	0.24	5	Fire Point		+	
С	213.		5	AHc kcal/m			M Spec. Ultra V.			
A* 45 to	1.33	826	5	ΔHf ΔFf	1		X-Ray Dif.	İ	1	
B* 163 ℃	1293.1		5	Viscosity	 	-	Infrared		1	
c	1			centistokes			Solubility in +		1	
tk to				γ •c			Acetone Carbon tet.			
X '			Ļ				Benzene	1		
A' 25 to B' 45 °C	7.19 1579.0	208	5				Ether n-Heptane			
c, - 12 -	231.		5	B ^V to			Ethanol			
A'* 25 to		745	5	_AV_ °C	_		Water		1	
B'* 45 °C	1479.7		5	(B ^V) to	ì	1	Water in		+	
Ac 180 to	7.46	624	5	(A ^V) °C					1	
Bc tc °C	284.4		5	c _p liq. °K						
Cryos. A°	1 -0		-	il					1	
consts. B°				c _p vap. °K						
te °C	152.84		5	c _w vap.						
$T_R = 0.8$	OT _C						+ grams/100 gran	me solver		
REFERENC		ow .	2-AF	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for			
SOURCE:				PI						
PURIFICAT	'ION:			PI						
LITERATU		CR EN								
DAM I U	ADF	-REI	-CES	•						

No. 139 2, 3, 3, 4-Tetramethylhexane STRUCTURAL FORMULA NAME CH₃CH₃ CH3CH C CH CH2CH3 Ref. Mole Molecular Molecular $C_{10}H_{22}$ ĊH₃ĊH₃ Weight 142.276 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C ٩K g 5.2091 5 B. P. ℃ h BP 0.05316 5 760 mm 164.59 2 0.03760 5 ſ١ 97.44 t_e to 100 4 g' <u>°K</u> 67.92 30 30 mm 0.7381 5 10 45.45 5 h' ∆Hm cal/g 7.97 5 to AHv cal/g Pressure n °K 25°C 76.93 5 mm 25°C 3.0990 o 30 mm 73.40 5 1186.1 5 te BP 62.41 5 m to Density te te (d, e) 60.34 5 n' °K g/ml 20°C 0.7694 2 60.25 5 ٥' 0.7648 d_4^t 25 2 AHv/T 18.80 5 30 0.7602 4 Surface tension T d 68 to 81,12 5 0.7878 4 dynes/cm. 20°C 27.69 5 183 <u>°C</u> 0.1137 5 å, ь -0.0392 4 5 30 26.39 25 to 78.99 5 40 25.13 5 Ref. Index e' 0.0824 5 [P] ^{n}D 20°C 1.4297 Parachor 5 d_c g/ml 0.231 25 2 1,4269 20°C 4.334 vc ml/g t °C 5 30 1.4242 4 30 ^tc 336. 5 40 "C" 4 0.7416 P_c mm 14785. 5 Sugd. 424.2 5 MR (Obs.) 47.74 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 5 1.0000 25°C 5 (nD-d/2)1.0450 2 30 mm 1.0000 5 Dispersion 97. 2 Dielectric 1.901 5 BP 0.9480 5 Flash Point C 0.9336 5 A 68 to 6.88337 5 Fire Point 0.24 5 1490.2 B | 226 °C 5 M. Spec. С 208. 5 AHc kcal/m Ultra V. ΔHf A* 68 to 1.38067 5 X-Ray Dif. ΔFf B*[193 °C 1399.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to t_k Carbon tet. °C Benzene 25 to 7.2364 Ether B١ 68 °C 1691.2 5 n-Heptane B_v | 5 226. to Ethanol °C Water A'* 25 to 1.72056 5 B'* 68 °C (BV) Water in 1591.0 5 to Ac| 226 to 7.7251 5 (A^V)| °C Bc _tc °C 2307.0 c_p liq. °ĸ 5 Cc 313.3 c_p vap. °K Cryos. A consts. B° c_v vap. te °C 183.49 5 T_{R} = 0.82 Tgrams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 14	0
NAME	2, 3, 3	,5-Tetra	methylhexan e			STRUCTURAL :	FORMUL	A.
						CH ₃		
			T			сн ₃ сн с сн		
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 142.2	76	Ċн ₃ Ċн ₃	Ċн ₃	
		Ref			Ref			Ref
F, P. °C	Γ		dt/dP		-		1	-
F.P. 100%	†	\dashv	*C/mm	}		f to	j	
B. P. *C	†	\neg	25°C	3.2148	5	h	ł	
760 mm 100	153.	2	BP t _e	0.05215 0.03777	5	f' to		+-
30	87.19 58.31	4 4	30 mm	0.7217	5	g' '°K_		
10	36.35	5	ΔHm cal/g		۲	h'	ŀ	1
1	-0.26	5	ΔHv cal/g	 	-	m to		
Pressure mm 25°C	5.258	35 5	25°C	73.46	5	n <u>*K</u> -	ł	ł
t _e	1151.7	5	30 mm BP	70.89	5	<u> </u>		<u> </u>
Density	† · · · · · · ·		t.	58.34	5	m' to	1	
g/ml 20°C	0.74		t _e (d, e)	58.26	5	n' ' <u>°K</u> -		
d ^t 25 4 30	0.742		ΔHv/T _e	18.72	5	<u> </u>	L	┼
	0.76		d 58 to	77.47	5	Surface tension dynes/cm, 20°C	24.47	5
ь	-0.038		$\frac{170}{4}$ $\frac{1}{1}$ $\frac{170}{25}$ $\frac{90}{100}$		5 5	3 0	23.43	5
Ref. Index			e' 58 °C		5	40	22.43	5
ⁿ D 20°C	1,419		d_ g/ml	0.230	5	Parachor [P]		1
30	1.414		d g/ml vc ml/g tc °C	4.347	5	30		1
"C"	0.747	79 4		318.	5	40		_
MR (Obs.)	48. 223	3 2	P _c mm	14305.	5	(424.2	5
MR (Calc.)		, 5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		1
(nD-d/2)	1.04		30 mm	1.0000	5	Dispersion	97.	2
Dielectric	2.019		BP t _e	0.9470 0.9334	5	Flash Point °C		
A 58 to B 206 °C	6.857	754 5 5	tc	0.24	5	Fire Point		<u> </u>
c	210.	5	∆Hc kcal/m			M Spec. Ultra V.		i
A* 58 to	1.36		ΔHf ΔFf			X-Ray Dif.		
B* 180 °C	1355.4	5	Viscosity	 	\vdash	Infrared		
·			centistokes			Solubility in +		
tk to	ł		η •c	1	1	Acetone Carbon tet.		1
t _x C	7,218	358 5			1	Benzene		
B' _ 58 ℃		5 5				Ether n-Heptane		ł
C'	228.	5	B ^V to			Ethanol		
A'* 25 to B'* 58 °C	1.707		├ ─ ~₩`	-1		Water Water in		1
Ac 206 to	7,611	5	(B ^V) to					T
Bc t C	2137.7	5	(A ^V) °C	ļ	<u> </u>			1
Cc '- c-	300.1	5	c _p liq. °K					j
Cryos. A° consts. B°			с _р vap. °К					
t _e °C	170.26	5	c _v vap.					
$T_{\mathbf{R}} = 0.8$						grams/100 grai	ns solven	t
REFERENC	ES: 1-Do	w 2-AP	I 3-Lit. 4-0	Calc. from det	t. da	ta 5-Calc. by for	mula	
SOURCE:		AI	PI					
PURIFICAT	ION:	AI	PI					
LITERATUI	RE REFE	RENCES	:					

No. 141 2, 3, 4, 4-Tetramethylhexane NAME STRUCTURAL FORMULA CH₃CH₃ CH3CH CH C CH2CH3 Mole Ref. Molecular Molecular Ċнз C10H22 ĊH % Pur Weight 142,276 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm ۰<u>K</u> g 25°C 4.7322 5 B. P. ℃ h BP 0.05290 5 162.2 760 mm 2 t_e 0.03758 5 ſ١ 100 95.38 4 to g' °K 30 66.01 4 30 mm 5 0.7344 10 43.66 5 h' ∆Hm cal/g 1 6.38 5 m to AHv cal/g Pressure n ۰ĸ 25°C 79.29 5 mm 25°C 3.4403 5 o 30 mm 72.95 te 1179.4 5 ΒP 62.03 5 m to Density 60.01 5 5 t_e (d, e) <u>°K</u> g/ml 20°C 0.7639 2 59.93 o' 25 $\mathbf{d_{4}^{t}}$ 0.7596 2 ΔHv/Te 5 18.81 30 0.7553 4 Surface tension 66 to 80.44 5 0.7811 44 a dynes/cm. 20°C 26.91 0.1135 -0.0386 h 30 25.71 5 an) 25 5 66 °C 78.31 5 40 24.56 Ref. Index 5 e' 0.0814 20°C 1.4270 2 [P] nD Parachor 0,233 5 d_c g/ml 25 1.4244 2 20°C vc ml/g t °C 5 4.290 30 1.4218 4 30 t_c 333. 5 40 "C" 0.7425 4 5 14863 P_c mm Sugd. 424.2 5 MR (Obs.) 47.82 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 5 1.0000 25°C (nD-d/2)1.0451 2 30 mm 1.0000 Dispersion 2 97. Dielectric 2.036 5 ВP 0.9480 5 Flash Point C 0.9338 5 A 66 to te t 6.88230 5 Fire Point 5 1482.1 0.24 B 1224 °C M. Spec. С 208. 5 AHc kcal/m Ultra V. ΔHf A* 66 to 1.38187 5 X-Ray Dif. ΔFf B*[191 °C 1392.1 Infrared Viscosity Solubility in centistokes Acetone to °C Carbon tet. $\mathbf{t_{\underline{x}}}$ Benzene A' | 25 to 7.23713 Ether B' _66_°C 1683.1 5 n-Heptane B^V A C' 5 226. to Ethanol °C Water A1* 25 to 1.72212 5 Water in B'* 66 °C 1582.9 5 (B^V)| to Ac 224 to 7.70922 5 (A^V)| °C Bc t_c °C 2278.0 cp liq. °K Cc 311.2 5 Cryos. A° consts. B° °K cp vap. c_v vap. te °C 180.75 5 $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

							No. 14	2	
NAME	2, 3, 4	.5-Tetra	methylhexane			STRUCTURAL			
		,			\dashv	CH ₃	CH ₃	_	
					\dashv	снзсн сн сн сн сн			
Mole	Ref.	Molecul	ar c u	Molecular	1	Ċн ₃ сн			
% Pur.		Formul	C ₁₀ H ₂₂	Weight 142.2	76		3		
		Ref.			Ref			Ref.	
F.P. °C			dt/dP			f to			
F.P. 100%			*C/mm 25*C	4, 4596	5	g <u>K</u>		1	
B. P. °C 760 mm	161.	2	BP	0.05290	5	h +		↓_	
100	94.20	4	l ^t e	0. 03768	5	f' to		1	
30 10	64.87 42.55	4 5	30 mm	0.7334	5	h'			
1	5.33	5	ΔHm cal/g				 	+-	
Pressure			ΔHv cal/g 25°C	75, 77	5	m to			
mm 25°C	3.679	54 5	30 mm	72.55	5	•]		
Density			BP	61.69	5	m' to			
g/ml 20°C	0.75		te te (d, e)	59.60	5	n' °K			
d ₄ 25	0.75		AHv/Te	18.76	5	o'			
a 30	0.74		d 65 to		5	Surface tension	35 05	-	
b	-0.038		<u></u>	0.1131	5	dynes/cm. 20°C	25.95 24.87	5	
Ref. Index			d' 25 to		5	40	23.81	5	
n _D 20°C 25	1.424			0,233	5	Parachor [P]			
30	1.422		d g/ml v ml/g t °C	4. 294	5	20°C 30		1	
"C"	0.74	14 4	1 _c 10	331.	5	40		1_	
MR (Obs.)	47.95) 2	P _c mm	14801.	5	Sugd.	424.2	5	
MR (Calc.)		, 5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		}	
(nD-d/2) Dielectric	1.046		30 mm	1.0000	5	Dispersion	98.	2	
A 65 to	6, 87	-	BP t _e	0.9480 0.9339	5	Flash Point °C			
B 222 °C		5	tc	0.24	5	Fire Point		↓	
С	208.	5	AHc kcal/m			M Spec. Ultra V.			
A* 65 to	1.372		ΔHf ΔFf			X-Ray Dif.		İ	
B* ∟190 °C	1384.4	5	Viscosity	 		Infrared		-	
·			centistokes			Solubility in + Acetone			
tk to C	ļ		7 ℃			Carbon tet.]	
A' 25 to	7, 226	562 5				Benzene Ether		1	
B' ∟ 65 °C		5	B ^V to	+	-	n-Heptane		1	
	226. 1.712	5	B' to			Ethanol Water		ļ	
A'* 25 to B'* 65 °C		217 5	(B ^V) - to	4		Water in		İ	
Ac 222 to	7, 691	+	(A ^V) °C						
Bc t °C	2261.0	5	c _p liq. °K						
Cc —	310.6	5						ĺ	
Cryos. A° consts. B°			c _p vap. °K				·		
t _e °C	179.43	5	c _v vap.						
$T_{R} = 0.87$						+ grams/100 gra		t	
REFERENC	ES: 1-Do			Calc, from det	da:	ta 5-Calc, by for	mula		
SOURCE:		A1	PI						
PURIFICAT		Al							
LITERATUI	RE REFE	RENCES	:						

NAME	3, 3, 4, 4-	Tetra	methylhexane		STRUCTURAL FORMULA CH ₃ CH ₃			
Mole % Pur.		lecul rmula		Molecular Veight 142.2	76	сн ₃ сн ₂ с с сн ₂ сн сн ₃ сн ₃	3	
		Ref.			Ref.		Ref.	
F.P. °C F.P. 1009	6		dt/dP °C/mm			f to g *K		
B.P. °C 760 mm 100 30 10	170.0 102.23 72.43 49.73 11.85	2 4 4 5	25°C BP t _e 30 mm	6.5635 0.05361 0.03757 0.7456	5 5 5	f' to g'°K		
Pressure mm 25°C t _e	2.4070 1200.2	5 5	ΔHv cal/g 25°C 30 mm BP	78.61 74.59 63.37	5 5 5	m to n - °K		
Density g/ml 20°0 dt 25 4 30	0.7783 0.7742	2 2 4	t _e (d, e) ΔHv/T _e d 72 to	61.23 61.12 18.82 82.91	5 5	n' °K_ o' Surface tension		
a b	0.7988 -0.0 ₃ 82	4	e 190 °C to	0.1150 80.74	5	dynes/cm. 20°C 29.61 30 28.39 40 27.20	5 5 5	
Ref. Index n _D 20°0 25 30		2 2 4	e' 72 °C d g/ml vc ml/g tc °C	0.0849 0.241 4.150 348.	5 5 5	Parachor [P] 20°C 30		
"C"	0.7406	4	P _c mm	15745.	5	40 Sugd. 424, 2	5	
MR (Obs. MR (Calc. (nD-d/2) Dielectric) 48.38 1.0456	2 5 2	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9470	5 5 5	Exp. L.1.%/wt. u. Dispersion 97.	2	
A 72 to B 236 °C	6. 89630	5 5 5	t e t c	0.9322 0.24	5	Flash Point C Fire Point M. Spec.		
A* 72 to B* 200 °C K c t _k to	1421.7	5	ΔHf ΔFf Viscosity centistokes η °C			Ultra V. X-Ray Dif. Infrared Solubility in + Acetone Carbon tet.		
A' 25 to B' 72 °C C' A'* 25 to	1714.0 225.	5 5 5	B ^V to A ^V °C			Benzene Ether n-Heptane Ethanol Water		
B'* 72 °C Ac 236 to Bc t _c °C	7,81479	5 5 5	(B ^V) to (A ^V) °C c _p liq. °K			Water in		
Cryos, Acconsts, B		Ť	c _p vap. °K					
t _e °C T _R = 0.8	189.58	5	c _v vap.			<u> </u>		
		2 ^	DI 2 124 4	Cala (many)		grams/100 grams solve	nt	
	CES: 1-Dow			Laic, irom de	t. da	ita 5-Calc, by formula		
SOURCE: PURIFICA	TION	AJ A1	PI					
	IRE REFERE							

No. 144 NAME 3-Isopropyl-2, 4-dimethylpentane STRUCTURAL FORMULA C₃H₇ сизси си си сиз Ref. Mole Molecular Molecular ĊНą ĊН₃ $C_{10}H_{22}$ % Pur. Weight 142, 276 Formula Ref. Ref Ref. F.P. C F.P. 100% -81.70 2 dt/dP to °C/mm <u>°K</u> g 25°C 3.8935 B. P. *C 0.05219 5 BP 157.04 2 760 mm 5 0.03742 ſ١ t_e to 100 91.08 4 g' °K 62.08 4 0.7254 5 30 30 mm 10 40.00 5 h' ∆Hm cal/g 5 1 3.151 to ł AHv cal/g Pressure °K n 25°C 75.09 4.2478 5 mm 25°C o 72.15 30 mm t_e 1165.2 5 5 61.42 BP ١ to Density 5 m 59.50 te (d, e) °K g/ml 20°C 5 0.75830 2 59.42 I ۰, 0.75457 $\mathbf{d_{4}^{t}}$ 5 AHv/T 18.90 30 0.7508 4 Surface tension 62 to 79.17 5 0.7732 dynes/cm. 20°C 26.13 5 _175_ 0.1130 ь -0.0374 4 30 5 25.11 ď٠ to 5 77.08 25 5 40 24.12 Ref. Index e' °C 62 0.0793 5 n_D 20°C 1.42463 [P] Parachor d_c g/ml 0.241 5 25 1.42248 2 20°C ml/g 4.145 1.42017 4 30 c 30 •C 5 327. tc 40 "C" 0.7441 4 P_c mm 15233. 5 Sugd. 424.2 5 MR (Obs.) 47.939 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 1.04548 u. (nD-d/2) 2 1,0000 5 97. 30 mm Dispersion 2 0.9485 Dielectric 2.029 5 BP Flash Point °C 5 0.9348 A 62 to 6.89055 5 Fire Point 0.24 L213 °C 1468.4 M Spec. Ultra V C 209. 5 ΔHc kcal/m ΔHf A* 62 to 1.3942 5 X-Ray Dif. ΔFf B* 185 °C 1379.1 5 Infrared ĸ Viscosity Viscos., centistokes °C Solubility in Acetone t_x Carbon tet. °C Benzene 25 to 7.2498 Ether Вı 1669.7 62 °C 5 n-Heptane $\mathbf{B}^{\mathbf{v}}$ 5 227. to Ethanol $\overset{\checkmark}{A}^{V}$ °C Water 5 25 to 1.7364 Water in B'* 62 °C 1569.6 5 (BV) to Ac | 213 to 7.68916 (A^V) °C Bc tc_°C 2224.8 5 cp liq. °K Cc 307.2 5 Cryos, A° °K cp vap. consts. B° c_v vap. te °C 174.81 5 $T_R = 0.81 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME	3, 3-Dieth	yl - 2	-methylpentane		STRUCTURAL FORMULA			
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 142.2	76	сн ₃ сн с сы сн ₃ с ₂ н ₅	H ₂ CH ₃	
		Ref.	I		Ref.		Rei	
F.P. °C			dt/dP			f to		
F.P. 100%			°C/mm			g °K		
B. P. °C			25°C	7.8588	5	h		
760 mm	174.	2	BP	0.05386				
100	105.87	4	t _e	0.03746	5	f' to		
30	75.87	4	30 mm	0.7507	5	g'° <u>K</u>		
10	53.01	5	∆Hm cal/g		ГΠ	h'		
	14.00	-	ΔHv cal/g	 	$\dagger \neg$	m to	1	
Pressure mm 25°C	1.9757	5	25°C	79.99	5	n•K		
t _e	1211.6	5	30 mm	75.57	5	0	ł	
Density			BP	64.20	5	m' to		
g/ml 20°C	0.780	2	t _e (d, e)	61.97	5 5	n'K	}	
d ₄ 25 30	0.775	2		1	1 1	o'		
4 30	0.770	4	ΔHv/T _e	18.87	5	Surface tension		
a	0.800	4	d 76 to	84.36	5		29.25 5	
ь	-0.001	4	_e194 °C d'	0.1159 82.16	5		27.78 5	
Ref. Index			e' 76 °C	0.0869	5	40	26.36 5	
n _D 20°C	1.435	2	d _c g/ml	0.229	5	Parachor [P]		
25	1.432	2	vc ml/g	4.361	5	20°C		
30	1.429	4	tc°C	348.	5	30 40		
"C"	0.7400	4	P _c mm	14985.	5	Sugd. 4	24.2 5	
MR (Obs.)	47.597	2	PV/RT		+	Exp. L 1.%/wt.		
MR (Calc.) (nD-d/2)		5	25°C	1.0000	5	u.		
	1.045	2	30 mm	1.0000	5		97. 2	
Dielectric	2.059	5	BP	0.9470	5	Flash Point C		
A 76 to	6.9118	5	te t	0.9319 0.24	5	Fire Point		
B 1236 ℃	1531.5 206.	5	ΔHc kcal/m	0.24	\vdash	M. Spec.		
A* 76 to	1.40206	5	ΔHc kcai/in	1		Ultra V.		
B* 204 °C	1439.9	5	ΔFf	1		X-Ray Dif. Infrared		
K			Viscosity					
·			centistokes	i		Solubility in T		
t _k to		İ	∥ າ °⊂			Carbon tet.	1	
X						Benzene		
A' 25 to B' 76 °C	7. 2589 1733. 4	5				Ether		
c, -,	224.	5	B ^V to			n-Heptane Ethanol	1	
A!* 25 to	1.7394	5	B ^V to A ^V °C		1 1	Water		
B'* 76 °C	1633.1	5	$(\mathbf{B}^{\mathbf{v}})$ = $\frac{1}{to}$	1		Water in		
Acl 236 to	7,8257	5	(A ^V) °C					
Bc tc °C	2451.7	5		+	+-+		1	
Ce C	323.8	5	c _p liq. °K					
Cryos. A°			c _p vap. °K	1				
consts. B°			li e				İ	
t _e °C	194.10	5	c _v vap.					
$T_{R} = 0.82$	2 T _C					grams/100 gram	s solvent	
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by form		
SOURCE:		Al						
PURIFICAT	ION.	Al						
	RE REFERE							
DI BARTO	KD KDI BKD	IVC EX	.					
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L								

							No. 146	٤
NAME	3-Ethyl-2	2, 2, 3	-trimethylpent	ane		STRUCTURAL	FORMULA	A.
l T					\neg	CH ₃ C ₂ H ₅		
					\neg	сн ₃ с с	CH ₂ CH ₃	
Mole	Ref. Mo	lecul	ar C ₁₀ H ₂₂	Molecular Weight 142.2	76	ĊH ₃ ĊH ₃		
% Pur.	<u> </u>	_		weight 142.2		T		Ref.
	T	Ref.		T	Ref.			Kei.
F.P. °C F.P. 100%	+	 	dt/dP *C/mm			f to		
B. P. °C	†		25°C	6.0208	5	h		
760 mm	168.	2	BP t	0.05345 0.03756	5	f' to		+-
100 30	100.45 70.75	4	t _e 30 mm	0.7429	5	g' 'K_		1
10	48.14	5	ΔHm cal/g	1	1	h'		
1	10.41	5	ΔHv cal/g		-	m to		
Pressure mm 25°C	2.6452	5	25°C	77.98	5	n <u>*K</u> _		
t _e	1196.0	5	30 mm BP	74.14 63.06	5	0		
Density	<u> </u>			60.95	5	m' to		1
g/ml 20°C		2	te te (d, e) AHv/T	60.84	5	n' °K_	1	
dt 25	0.777 0.773	2	AHv/T _e	18.83	5			├—
a .	0.7970	4	d 71 to		5	Surface tension dynes/cm. 20°C	29.40	5
ь	-0.0380	4	e 187 °C		5	30	28, 21	5
Ref. Index			e' 71 °C		5	40	27.06	5
n _D 20°C	1.436	2 2	d _c g/ml	0.242	5	Parachor [P] 20°C		
30	1.431	4	t _c °C	4.131 345.	5	30		
"C"	0.7407	4	P _c mm	15744.	5	40 Sugd	424,2	5
MR (Obs.)		2	PV/RT	13.11.	-	Exp. L.1.%/wt.	727.2	
MR (Calc. (nD-d/2)) 48.38 1.046	5 2	25°C	1.0000	5	u.		1
Dielectric	2,062	5	30 mm BP	1.0000 0.9480	5	Dispersion	97.	2
A 71 to		5	te	0.9334	5	Flash Point °C Fire Point		İ
B 1234°C	1504.2	5	t _c	0.24	5	M Spec.		┼
_ c	207.	5	ΔHc kcal/m ΔHf	1		Ultra V.		
A* 71 to B* 197 °C		5	ΔFf	ļ		X-Ray Dif. Infrared		
K	-		Viscosity			Solubility in +		┼
c	-1		centistokes °C	.		Acetone		1
t _k t ₀		ļ	η ∘ο			Carbon tet. Benzene		
A' 25 to		5				Ether		
B' ∟ 71 °C	225.	5	B ^V to	<u> </u>	\vdash	n-Heptane		İ
A1# 25 to		5	AV C			Ethanol Water		l
	1605.2	5	(B ^V) to	<u>-</u> ļ		Water in		ļ
Ac 234 to	7. 79459	5	(A ^V) °C	i				
Bc tc °C	2410.3 324.7	5	c _p liq. °K					
Cryos. A°	 	ا `	11 -	1				
consts. B°			P	•				
t _e °C	187.37	5	c _v vap.					
$T_{\mathbf{R}} = 0.8$	2T _c					grams/100 gran	ns solven	t
REFEREN	CES: 1-Dow	2-A1	PI 3-Lit. 4-	Calc. from det	t. da	ta 5-Calc. by for		
SOURCE:		A	PI					
PURIFICA?	rion:	A	PI					
LITERATU	RE REFERE	NCES	5:					
1								
1								
}								
L								

г							No. 1	41		
NAME	3-Eth	yl-2, 2, 4	-trimethylpenta	ne		STRUCTURAL FORMULA				
1						CH ₃ CH ₃				
<u> </u>						сн, с сн				
Mole	Ref.	Molecul:	ar C ₁₀ H ₂₂	Molecular	İ					
% Pur.		Formula	10 ¹¹ 22 V	Weight 142.2	276	ċн ₃ ċ ₂ н ₅				
		Ref.			Ref.			Ref.		
F. P. °C			dt/dP			f to				
F.P. 100%	,		°C/mm		l	g ° <u>K</u>	ł			
B. P. °C			25°C	3.5358	5	h	1			
760 mm	155.3	2	BP	0.05235 0.03680	5		 	+		
100 30	89.23		t _e		l	g' to		ľ		
10	60.23		30 mm	0.7250	5_	h'	· [1		
1	1.39		AHm cal/g		<u> </u>	<u>}</u>	 	+		
Pressure			ΔHv cal/g		l _	m to		ł		
mm 25°C	4.73	70 5	25°C 30 mm	74.15 71.40	5		1			
t _e	1160.4	5	BP	60.74	5	├ ─ ,	<u> </u>	+		
Density			t	58.84	5	m' to				
g/ml 20°C			te (a, e)	58.76	5	"	1			
dt 25 4 30	0.75		AHv/T _e	18.76	5	I		+		
a	0.77		d 60 to	78.15	5	Surface tension	25.96	5		
b	-0.03		e 173 °C to	0.1121	5	dynes/cm. 20°C	24.88	5		
Ref. Index			d' 25 to	76.10 0.0782	5	40	23, 82	5		
n _D 20°C				0,234	5	Parachor [P]		T		
25 30	1.41		d g/ml	4, 272	5	20°C				
	1.41		tc °C	324.	5	30 40	Į.			
"C"	0,74		P _c mm	14706.	5		424.2	5		
MR (Obs.)			PV/RT		 	Exp. L.1.%/wt.		1		
MR (Calc. (nD-d/2)	1.04		25°C	1.0000	5	u.				
Dielectric	2,02		30 mm BP	1.0000 0.9485	5	Dispersion	98.	2		
A 60 to			t	0.9349	5	Flash Point °C				
B (211 °C		5 5	t _c	0.24	5	Fire Point	ļ	4		
c '	209.	5	AHc kcal/m			M. Spec.				
A* 60 to	1.36	84 5	ΔHf	İ		Ultra V. X-Ray Dif.				
B* <u>1</u> 8 <u>3 °C</u> K	1363.7	5	ΔFf			Infrared	1	İ		
C			Viscosity centistokes			Solubility in +				
t _k – tō	-		η °c			Acetone				
t _x °C			•			Carbon tet. Benzene				
A' 25 to					ļ	Ether				
B' _60_°C	1653.1 227.	5	B ^V to		t	n-Heptane	ļ			
A'* 25 to			B ^V to C		ŀ	Ethanol Water				
B'+ 60 °C		5	(B ^V) to	1	1	Water in				
Ac 211 to			(A ^V) °C	1						
Bc _i t _c °C	2183.2	5		 	+	1				
Cc	304.6	5	c _p liq. °K			1		1		
Cryos. A°			c _p vap. °K	1	1			1		
consts. B°	+	-	i)	1						
t _e °C	172.95	5	c _w wap.	L	L	L				
$T_{R} = 0.8$	l T _c					grams/100 gra	ms solve	nt		
REFEREN	CES: 1-D	ow 2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ata 5-Calc. by for	mula			
SOURCE:		AF	PI							
PURIFICA'	TION:	AF	PI							
LITERATU										
L										

No. 148 NAME 3-Ethyl-2, 3, 4-trimethylpentane STRUCTURAL FORMULA С₂H₅ Сн₃сн с с CH CH3 $\dot{c}_{H_3}\dot{c}_{H_3}\dot{c}_{H_3}$ Mole Ref. Molecular Molecular $C_{10}H_{22}$ Weight 142.276 Formula % Pur. Ref. Ref. Ref F.P. C F.P. 100% dt/dP to °C/mm <u>•ĸ</u> g 25°C 6.3989 5 B. P. *C h BP 0.05358 760 mm 169.44 2 4 f' 0.03759 5 to 100 101.72 °K g¹ 71.94 4 30 30 mm 0.7450 5 49, 26 5 10 h' ∆Hm cal/g 5 11.43 1 to m AHv cal/g Pressure °K n 25°C 78.42 mm 25°C 2.4751 5 5 o 74.45 63.25 5 1198.6 30 mm t_e BР 5 m to Density 5 te (d, e) 61.11 °K g/ml 20°C 0.7773 2 61,00 5 ۰, 25 0.7733 2 d4 ΔHv/Te 18.81 5 4 30 0.7693 Surface tension ī 82,71 72 5 0.7933 -0.0₃80 44 28.85 dynes/cm. 20°C •c 0.1149 189 ь 5 30 27.67 ď to 5 80.53 25 5 40 26.54 . Ref. Index e' 72 0.0846 5 20°C [P] n_D 1,4333 2 Parachor d_c g/ml 0.240 5 25 2 1.4310 20°C , c 4.168 ml/g °C 30 1.4285 4 30 t_c 5 347. 40 "C" 0.7399 4 P_c mm 15653. 5 5 Sugd 424.2 MR (Obs.) 47.60 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 (nD-d/2)1.0447 2 1.0000 30 mm 2 Dispersion 97. Dielectric 2.054 5 5 RP 0 9470 Flash Point °C 72 to 0.9322 5 6.8938 5 Fire Point tc 0.24 _235 °C 1509.9 В M Spec. C 207 5 AHc kcal/m Ultra V ΔHf A* | 72 to 1.38832 5 X-Ray Dif. ΔFf B* 199 °C 1419.1 Infrared ĸ Viscosity Viscos.., centistokes °C Solubility in c Acetone to Carbon tet. •c Benzene 7.24360 A 25 to 5 Ether B١ 1711.2 _ 72 °C 5 n-Heptane 225. 5 to Ethanol °C 25 to Water A'* 1.72596 Water in B'* 72 °C 1610.9 (BV) to Ac | 235 to 7.80480 (AV) °C Bc _tc_ 2426.8 ٠c 5 c_p liq. °K Cc 325.7 5 Cryos. A° c_p vap. °K consts. B° c, vap. 188.95 5 $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME

2, 2, 3, 3, 4-Pentamethylpentane

159 No. 149 STRUCTURAL FORMULA

NAME	_,_,_,		, ,	-	CH ₃ CH ₃					
			T			снас с сн сна				
Mole % Pur.		lecul: rmula	ar C ₁₀ H ₂₂	Molecular Veight 142.2°	76	ch3ch3ch3				
		Ref.			Ref.		Ref.			
F.P. °C F.P. 100%	-36.45	2	dt/dP °C/mm			f to g°K				
B. P. °C 760 mm 100 30 10	166.05 98.72 69.13 46.60 9.01	2 4 4 5 5	25°C BP t _e 30 mm AHm cal/g	5.5392 0.05328 0.03759 0.7402	5 5 5	h to g' °K h'				
Pressure mm 25°C t _e Density g/ml 20°C	1190.4	5 5	ΔHv cal/g 25°C 30 mm BP t _e t _e (d, e)	77.37 73.71 62.70 60.63 60.52	5 5 5 5	m to co m' to to co co co co co co				
d ₄ 25 30 a	0.77675 0.7734 0.7934	2 4	ΔHv/T _e d 69 to	18.82 81.56	5	Surface tension dynes/cm. 20°C 29.26	5			
b Ref. Index	-0.0 ₃ 67	4	e 1 185 °C d' 25 to e' 69 °C	0.1136 79.45 0.0831	5 5 5	30 28.27 40 27.30	5			
ⁿ D 20°C 25 30	1.43412 1.43210	2 2 4	dc g/ml vc ml/g tc °C	0.249 4.012 346.	5 5 5	Parachor [P] 20°C 30 40				
MR (Obs.) MR (Calc.		2	P _c mm PV/RT	16236.	5	Sugd. 424. 2 Exp. L.1.%/wt.	5			
(nD-d/2) Dielectric	1.04602	5 2 5	25°C 30 mm BP	1.0000 1.0000 0.9480	5 5 5	u. Dispersion 97.	2			
A 69 to B 235 °C		5 5 5	te tc AHc kcal/m	0.9335 0.24	5	Flash Point C Fire Point M. Spec.				
A* 69 to B* 195 °C K c t _k to	1405.4	5	ΔHf ΔFf Viscosity centistokes γ °C			Ultra V. X-Ray Dif. Infrared Solubility in + Acetone Carbon tet.				
A' 25 to B' 69 °C A'* 25 to B'* 69 °C	7.2383 1697.1 225.	5 5 5 5	B ^V to A ^V C To			Benzene Ether n-Heptane Ethanol Water Water in				
Ac 235 to Bc t _c °C Cc		5 5 5	(A ^V) °C c _p liq. °K							
Cryos. A° consts. B°			c _p vap. °K							
t_e °C $T_R = 0.8$	185.15 2 T _C	5	c _w wap.		<u></u>	grams/100 grams solve:	nt			
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da					
SOURCE:	TION.	AI AI				Name of Association and Associ				
PURIFICA LITERATU	RE REFERE									

No. 150 NAME 2, 2, 3, 4, 4-Pentamethylpentane STRUCTURAL FORMULA сн3 CH₃ сн3с сн с CH3 Mole Ref. Molecular Molecular с҅н₃с҅н₃с҅н₃ $C_{10}H_{22}$ Weight 142.276 % Pur. Formula Ref. Ref Ref. •c -38, 75 2 dt/dP to F.P. 100% °C/mm <u>°K</u> g 25°C 4.1689 5 B.P. °C h ВP 0.05270 5 760 mm 159.29 2 te 0.03768 5 f to 100 92.74 4 g' °K 63.52 4 30 30 mm 5 0.7307 10 41.28 5 h' AHm cal/g 1 4.20 5 to m AHv cal/g Pressure °K n 25°C 75.32 mm 25°C 3.9551 5 0 5 5 30 mm 72.24 te 1170.4 5 BP 61.41 Density m to 59.45 5 te (d, e) g/ml 20°C °K n' 0.76703 2 59.36 5 0 25 0.76361 dt 2 AHV/Te 18.77 5 30 0.7602 4 Surface tension 79.43 64 to 5 0.7807 -0.0₃68 27.35 4 dynes/cm. 20°C 178 °C 25 to 0.1131 Ъ 5 30 26.38 ď 77.32 5 40 25.44 5 Ref. Index e¹ 64 °C 0.0800 5 20°C 1.43069 [P] $\mathbf{n}_{\mathbf{D}}$ Parachor 5 d_c g/ml 0.244 25 2 1.42868 20°C v_c t_c ml/g °C 5 4.102 30 1.42660 4 30 333 5 40 "C" 0.7455 4 P_c mm 5 15544. Sugd. 424.2 5 MR (Obs.) 47.984 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 25°C 1.0000 5 1.04717 (nD-d/2)2 30 mm 1.0000 5 Dispersion 98. 2 Dielectric 2.047 5 RP 0.9475 Flash Point °C A 64 to 0.9335 6.8712 5 Fire Point t_c 5 1468.6 0.24 В 1_22<u>4 °C</u> c M Spec. 209. 5 AHc kcal/m Ultra V ΔHf A* 64 to 1.37443 5 X-Ray Dif. ΔFf B+ 188 °C 1379.3 Infrared ĸ Viscosity Centistokes Solubility in c Acetone to •c Carbon tet. Benzene 25 to A١ 7, 22785 5 Ether 1669.2 5 B <u>64</u> °C n-Heptane B^V A^V 227. 5 to Ethanol 1.71400 °C Water A'* 25 to 5 B'* 64 °C (BV) Water in 1569.1 5 to Ac | 224 to 7.70538 (A^{*}) | °C Bc tc_ ۰c 2275.4 5 c_p liq. ۰ĸ Сc 314.2 Cryos. A° c_p vap. ۰ĸ consts. B° t_e °C c_v vap. 177.44 5 $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

										No. 15	
NAME		n-Uı	ndeca	ne			_	STRUCTURAL FORMULA			
Mole		Ref.	Mo	lecul	ar C u	Molecular	\dashv		CH ₃ (CH ₂)	₉ CH ₃	
% Pur.		<u> </u>	For	mula	C ₁₁ H ₂₄	Weight 156.2	92				,
ļ	-			Ref.			Ref.			,	Ref.
F.P. °C		25.59	4	2	dt/dP			f	to		
F.P. 1009	•				°C/mm 25°C	30.42	5	g	' <u>°K</u>		'
B. P. °C 760 mm	1,0	95.89	0	2	BP	0.05356	5	h			
100	12	27.94	3	2	t _e	0.03641	5	f'	to		
30 10		97.90 74.97		4 4	30 mm	0.7526	5_	g'	<u>•</u> K		
l		36.60		5	ΔHm cal/g		\perp	_h'	200.4-	0.0355	-
Pressure					ΔHv cal/g			m n	300 to	0.0255 0.0014	4
mm 25°C		0.42	89	5	25°C 30 mm	86.4 77.55	2 5	0	_=	-0.0647	4
Densite	124	44.8		5	BP	63.5	2	m'	700 to	0.1104	4
Density g/ml 20°0	:	0.74	017	2	te te (d, e)	60.81	5 5	n'	1000 °K	0.0012	4
dt 25		0.73	655	2	AHV/Te	19.38	5	٥'		-0.0640	4
<u> </u>		0.73		4	d 98 to	91.59	5		face tension		
a b	١.	0.75 0.03-		4 4	_e_ _2 <u>17 °C</u>	0.1434	5	dyn	es/cm. 20°C 30	24.74 23.80	2 2
Ref. Index		••••		H	d' 25 to	89.44 0.1214	5 5		40	22.88	2
n _D 20°0	-	1.41		2	d _c g/ml	0,237	2	Par	achor [P]		
30		1.41		2 4	V_ mi/g	4.223	2			471.0 471.0	4
"C"	+	0.74		4	tc °C	367.	2		40	471.0	4
MR (Obs.	-	53, 12		2	P _c mm	14592.	2			463.2	5
MR (Calc.		52.99	8	5	PV/RT 25°C	1.0000	5	Exp	. L.1.%/wt.		1
(nD-d/2)		1.04		2	30 mm	1.0000	5	Dis	u. persion		ļ
Dielectric		2. 01		5	BP	0.9305	5	Fla	sh Point C		t^-
A 98 to B 258 °C		6.97 72.47		2 2	te tc	0.9126 0.242	2	Fir	e Point		<u> </u>
c -230		38. 02		2	ΔHc kcal/m	1663.55	2		Spec.		1
A# 98 to		1.53	000	5	ΔHf ΔFf	i			ra V. Ray Dif,		
B*[_227.°C	149	93,5		5	Viscosity	 	\vdash	Infr	ared		
c	_				centistokes				ability in +		
t _k to					7 110 °C	0.5977	2 2		etone rbon tet.	(
t _x °C		7.32	25	5	150	0.5160 0.4522	2		nzene		
B' 98 °C		76.4	25	5	170	0,4006	2		her Heptane		[
	20	06.		5	B ^V 100 to	491.2	4	Et	hanol	}	
A'* 25 to B'* 98 °C		1.85	11	5		₹. 49440	4		iter iter in		
Acl 258 to		7.93	52	5	(B ^V) to						
Bc tc °C	254	43.8	J.	5			╁─┤				
Cc	30	09.3		5	c _p liq. °K	1					
Cryos, A ^c consts, B ^c					c _p vap. °K						
t _e °C		7.09		5	c _v vap.						L
$T_R = 0.8$	3 T _c				-			+ gr	ams/100 gra	ms solven	t
REFEREN	CES:	1-D	ow			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:				AF							
PURIFICA				AF							
LITERATU	IRE .	REF	EREI	NCES	:						

			· · · · · · · · · · · · · · · · · · ·					No. 152	2
NAME	n-Dodecar	e				STR	UCTURAL	FORMULA	
					_		CH3(CH2)1	оCН ₂	
Mole % Pur.	Ref. Mo	lecul rmul	ar C ₁₂ H ₂₆	Molecular Weight 170.	328		3, 2,1	0 3	
		Ref.			Ref				Ref
F. P. ℃	-9.587	2	dt/dP			f	to		
F.P. 100%			*C/mm 25*C	89.64	5	g	<u>*K</u>		
B.P. °C 760 mm	216, 278	2	BP	0.05528	2	h			
100	146.142	2	t _e	0.03639	5	f' g'	to to		
30 10	115.1 91.45	4 5	30 mm	0.7771	5	h'	'A_ 		
i	51.84	5	ΔHm cal/g	51.69	31	m	300 to	0.0234	4
Pressure			ΔHv cal/g 25°C	86.0	2	n	_600 °K	0.0013	4
mm 25°C	0.1320 1297.3	5	30 mm	75.46	5	0	i — — — — — — — — — — — — — — — — — — —	-0.0643	4
Density	1-2/1.5	-	BP	61.3	2 5	·m¹	700 to	0.1115	4
g/m1 20°C	0.74869	2	te (d, e)	58.38 57.97	5	n'	1 <u>1000 °K</u>	0.0012	4
dt 25 4 30	0.74516 0.74163	2 4	AHV/T	19.37	5	<u>'</u>	İ	-0.0640	*
a 30	0.74183	4	d 150 to	91.58	5		ace tension s/cm. 20°C	25 44	2
ъ	-0.03706	4	_e _230 °C a' _25 to	0.1400 88.92	5	yne	30	25.44 24.51	2
Ref. Index			e' 150 °C	0.1170	5		40	23,60	2
ⁿ D 20°C	1.42160 1.41949	2	d _c g/ml	0.237	2	Para	chor [P]	510.9	4
30	1.41735	4	v mi/g	4.215	2 2		30	511.1	4
"C"	0.7486	4	1 -	386. 13604.	2		40 Su ad	511.1	4
MR (Obs.)	57.76	5	P _c mm	13604.	-	F		502.2	5
MR (Calc.) (nD-d/2)	57.616 1.04725	2 5	25°C	1.0000	5	Exp.	L.1.%/wt. u.		
Dielectric	 		30 mm	1.0000	5	Disp	ersion		
A 150 to	6,98059	2	BP t _e	0.9283 0.9084	5		h Point °C	165.	3
B 280 °C		2	tc	0.237	2		Point		
С	180.311	2	ΔHc kcal/m ΔHf	1810.48	2	M Sp Ultra			
A* 150 to B* 250 °C	1.5604 1546.9	5 5	ΔFf				ay Dif.		
к	1340.7	١	Viscosity			Infra			-
tto	-		centistokes 7 150 °C	0 5130	١, ١		bility in + etone		
tk to			7 150 °C	0.5128 0.4516	2 2		bon tet.		
A' 25 to	7.3157	5	190	0.4017	2 2	Eth			
B' ∟150 °C	1830.0	5 5	B ^V 140 to	0.3597 529.0	4		leptane anol		
A'* 25 to	1.8787	5	AV 220 °C	2.46116	4	Wat			
B'* 150 °C		5	(B ^V) to	1		Wat	ter in		
Ac 280 to	8.06653	5	(A ^V) °C	1					
Bc tc C	2780.0 322.0	5	c _p liq. °K						
Cryos, A°		Ė	c _p vap. °K	1					
consts. B°			_						
t _e °C	240.09	5	c _v vap.						
$T_{R} = 0.84$							ms/100 gran		:
	ES: 1-Dow		PI 3-Lit. 4-0	alc. from det	. da	ta 5-	Calc. by for	mula	
SOURCE: API									
PURIFICAT			PI						
LITERATUI	RE REFEREN	ICES	: 3 NFPA 325	; 3' JACS <u>76</u>	, 333	(1954) Linke et al	•	
l									

									No. 15	3
NAME	n-Tr	ideca	ne				ST	RUCTURAL	FORMUL	A
Mole	Ref	. Mo	ecul		Molecular			CH ₃ (CH ₂)	11 ^{CH} 3	
% Pur.		FOI			Weight 184.3					D (
	1		Ref.		+	Ref.			r	Ref.
F.P. °C	-5.39	92	2	dt/dP			f	to		
F.P. 100%	<u>'</u>			°C/mm 25°C	249.27	5	g	' <u>°K</u>		
B. P. °C 760 mm	225 4		١, ١	BP	0.0568	2	h			
100 mm	235.44 163.3	ŧ	2 2	t _e	0.03648	5	f'	to		
30	131.49	5	4	30 mm	0.7994	5	g'	<u>•K</u>		
10	107.09		5	ΔHm cal/g		\dagger	h'			
1	66.35		5	ΔHv cal/g		$\dagger \neg$	m	300 to	0.0241	
Pressure mm 25°C	0.00	3972	5	25°C	85.9	2	n o	_600_•K	0.0014	
t _e	1339.0	3712	5	30 mm	73.60	5			-0.0 ₆ 48	4
Density	+		\vdash	BP	59.1 55.97	5	m'	700 to	0.1133	
g/ml 20°C	0.75	564	2	te te (d, e)	55.47	5	n' o'	1000 •K		
dt 25 4 30	0.75		2	AHv/Te	19.30	5			-0.0 ₆ 39	-
	0.74		2	d 131 to	91.92	5		face tension		-
a b	-0.0		4 4	_e_ _2 <u>61 °C</u>		5	dyn	es/cm. 20°C 30	26.1 25.2	2 2
Ref. Index		3.2	┼╌┤	d' 25 to	88.78	5 5		40	24.3	2
n _D 20°C		256	2	e' 131 °C		+	Par	achor [P]		<u> </u>
45	1.42	234	2	d _c g/ml	0.24 4.231	2 2		20°C	550.9	4
30	1.42		4	v _c ml/g t _c °C	404.	2		30	551.3	4
"C"	0.74	476	4	Pcmm	12920.	2		40 Sugd.	551.6 541.2	5
MR (Obs.)			5	PV/RT	+	\vdash	Evr	. L. l. %/wt.		
MR (Calc. (nD-d/2)	62.23		2 5	25°C	1.0000	5		u.	}	
Dielectric	2.0		5	30 mm	1.0000	5	Dis	persion		
A 131 to	6.98		2	BP t _e	0.92225 0.9001	5		sh Point C		
B 302 °C			2	t _c	0.23	2		e Point		<u> </u>
c	172.90	0	2	∆Hc kcal/m	1957.40	2		Spec. ra V.		
A* 131 to	1.60	005	5	ΔHf ΔFf	1			Ray Dif.		
B*[271 °C	_ 1600.8		5	Viscosity	 	+-1	Infr	ared		
c	_			centistokes				ability in +		
t _k to				n 170 °C	0.5043	2 2		etone rbon tet.		
t _x °C	1			190 210	0.4468 0.3987	2	Be	nzene		
A' 25 to B' 131 °C		14/	5	230	0.3577	2		he <i>r</i> Heptane		
c, 131 9	190.9		5	B ^v 160 to	558.8	4		neptane hanol		
A'* 25 to	1.90	095	5	A 240 °C	₹. 44372	4	W	ater		
B'*131 °C			5	(B^V) to			W.	ter in		-
Acl 302 to	8.19	985	5	(A ^V) °C	1				ĺ	
Bc tc °C	3013.2		5	c _p liq. °K						
Cryos. A°	+ 333.4		 		1				1	
consts. B°	1		L	Р			1			
t _e °C	261.5		5	c _v vap.						
$T_R = 0.8$	5 T _c						+ g1	ams/100 gra	ms solven	t
REFEREN		Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			AI	PI						
PURIFICA'	TION:		AI	PI						
LITERATU		ERE	NCES	3:						
1										

										No. 154	4
NAME	n-Te	trad	ecan	8				STR	UCTURAL 1		
Mole	Ref.	Mo	lecul	ar C ₁₄ H ₃₀		Molecular Veight 198.3			сн ₃ (сн ₂) ₁	₂ CH ₃	
% Pur.		FO			<u>'</u>	veignt 170.3	Ref				Ref.
F. D. 46	T = 0/	2	Ref.		\neg		Kei				
F.P. C F.P. 100%	5.86		2	dt/dP *C/mm				f g	to K		
B, P, *C	<u>† </u>		Н	25°C	ı	711.4	5	h			İ
760 mm	253.57	,	2	BP t _e		0.0582 0.03632	2 5	f'	to		_
100 30	179.6		2 4	30 mm		0.8202	5	g'			
10	121.92	:	5	ΔHm cal/g	,	0.0202	Ť	h'	l		
1	80.13		5	ΔHv cal/g	`		-	m	300 to	0.0242	4
Pressure mm 25°C	0.01	172	5	25°C		85.7	2	n o	_6 <u>00</u> •K	0.0014	
te	1390.0		5	30 mm BP		71.85 57.5	5 2	l	l	-0.0 ₆ 48	1
Density				t.		54.19	5	m' n'		0.1121 0.0012	4
g/ml 20°C	0.76		2 2	t _e (d, e)		53.66	5	0'	11000 K	-0.0612	
dt 25 4 30	0.75		4	ΔHv/T _e		19.36	5	S., 6	*	•	
a	0.77	68	4	d 147 e 282		91.62	5		s/cm. 20°C	26.6	2
ь	-0.03	70	4	$\begin{bmatrix} -e_1 & -\frac{1}{25} & -\frac{282}{25} \end{bmatrix}$		0.1346 88.94	5	y'	30	25.7	2
Ref. Index	1.42	80	2	e' 147		0.1298	5		40	24.8	-
n _D 20°C	1.42		2	d _c g/ml		0.24	2	Para	chor [P] 20°C	590.6	4
30	1.42		4	t _c *C		4.184 422.	2 2		30	590.9	4
"C"	0.74	67	4	P _c mm		12160.	2		40 Sugd.	591.2 580.2	4.
MR (Obs.)	67.03		5	PV/RT	\dashv		H	Exp	L.1.%/wt.		<u> </u>
MR (Calc.) (nD-d/2)	66.85		5	25°C		1.0000	5		u.		
Dielectric	2,04		5	30 mm BP		1.0000 0.9232	5		ersion		
A 147 to	6.99	57	2	te		0.8996	5		h Point °C Point		
B ∟325°C	1725.46	,	2	¹ c	-	0, 23	2	M S			
C	165.75	_	5	ΔHc kcal/r ΔHf	n	2104.32	2	Ultr	a V.		l
A* 147 to B* 292 °C	1.62	54	5	ΔFf				X-R	ay Dif. red		
к ———				Viscosity					bility in +		
t _k	-			centistokes		0.4946	2	Ace	tone		
t _x °C	t			210		0.4402	2		rbon tet. nzene		l
A' 25 to	7.31	43	5	230 250		0.3941 0.3552	2 2	Eth	er		
B' ∟1 <u>47 °C</u>	1930.4		5	BV 180	to	588.9	4		leptane anol		
A'* 25 to	1.93	86	5	AV 260		Z. 42490	4	Wa	ter		
B'* 147 °C			5		to			Wat	ter in		_
Ac 325 to	8.36	982	5	(A ^V)	•c						
Bc tc °C	3312.8		5	c _p liq.	°K						
Cryos. A°	1 22.3				•ĸ						
consts. B°				_	*						
t _e °C	282.11		5	c _v vap.							
$T_R = 0.8$	στ _c							+ gra	ms/100 gran	ns solven	t
REFERENC	ES: 1-D	ow	2-AF	PI 3-Lit.	4-C	alc. from det	, dat				
SOURCE:			AI	PI							
PURIFICAT	ION:		AI	PI							
LITERATU	RE REFI	ERE	CES	:							
L											

No. 155 n-Pentadecane STRUCTURAL FORMULA NAME $CH_{3}(CH_{2})_{13}CH_{3}$ Molecular C₁₅H₃₂ Mole Ref. Molecular % Pur Weight 212.406 Ref. Ref. F.P. °C F.P. 100% 9.926 2 dt/dP to °C/mm ۰ĸ g 35521. 25°C B.P. °C h 0.0595 ΒP 2 270.63 760 mm 2 5 ſ١ t_e 0.03642 to 100 195.0 2 g' <u>°К</u> 30 4 30 mm 0.8390 5 161.57 10 5 136, 01 h١ ∆Hm cal/g 93.26 5 300 to 0.0239 m AHv cal/g Pressure n 600 °K 0.0014 25°C 85.7 2 mm 25°C 0.00233 5 ٥ -0.0648 30 mm 5 70. 26 te 1424.1 5 BP 55.6 2 m 700 to 0.1156 Density te (d, e) 52. 12 5 n' 1000 °K 0.0012 g/ml 20°C 0.7685 0.7650 5 2 51.51 o! -0.0639 4 25 $\mathbf{d_4^t}$ ΔHv/T_e 19.28 5 30 0.7615 4 Surface tension 160 91.99 d to 5 0.7825 4 dynes/cm. 20°C 27.1 2 1 290 °C 0.1344 ь -0.0370 4 30 26.2 2 آاتa 25 86.18 to 2 40 25.3 Ref. Index e' | 160 5 0.1130 20°C 1.4319 2 Parachor [P] n_D d_c g/ml 0.24 2 25 1,4298 2 20°C 630.6 vc ml/g 4.190 2 30 631.1 1.4277 4 30 4 $^{\mathbf{t}}_{\mathbf{c}}$ 437. 2 40 "C" 631.4 4 0.7461 4 11400. 2 P_c mm Sugd. 619.2 5 MR (Obs.) 71.67 5 PV/RT Exp. L. 1. %/wt. MR (Calc.) 71.47 25°C 1.0000 (nD-d/2)1.0476 5 30 mm 1.0000 5 Dispersion 2.05 Dielectric 5 BP 0.9170 5 Flash Point C te tc 0.8914 A 160 to 7,0017 2 Fire Point 0.23 2 B 1338 °C 1768.82 2 M. Spec. C 158.60 2 ∆Hc kcal/m 2251, 24 Ultra V ΔHf A* 160 to 1.6613 5 X-Ray Dif. ΔFf B* 310 °C 1693.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to 210 °C 0.4819 2 Carbon tet. °C t_x 230 0.4308 2 Benzene 250 0.3878 2 25 to 7.3123 Ether 270 0.3510 В' 160 °C 1973.3 5 n-Heptane B^V | 200 to A^V | 280 °C T200 to 176.6 5 608.0 Ethanol Z. 42601 A'* 25 to 4 Water 1.9646 5 B'*160 °C Water in (BV) 1891.6 5 to Acl 338 to 8.5317 5 (A V) 3600.5 Bc tc °C 5 c_p liq. ۰ĸ 5 Cc 369.1 c_p vap. Cryos. A° consts. B° c vap. te °C 301.06 5 $T_R = 0.86 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

									No. 156	<u>.</u>
NAME	n-He	xadeca	ane				STR	UCTURAL	FORMULA	.
ľ										
								CH ₃ (CH ₂) ₁	4 ^{CH} 3	
Mole % Pur.	Ref.	Mole For			Molecular Weight 226.4	32				
			Ref.			Ref				Ref.
F.P. °C	18, 16		2	dt/dP			f	l to		
F.P. 100%				*C/mm		_	g	<u>*K</u>		l
B. P. °C 760 mm	20/ 70	. T		71.87°C BP	127.43	5 2	h	L		
100 mm	286.79 209.51		2 2	t _e	0.03617		f'	to		
30 10	175.2	- 1	4	30 mm	0.8596	5	g'			
10	149.04 105.20		4 5	ΔHm cal/g			h'			<u> </u>
Pressure°(:		\dashv	ΔHv cal/g			m n	300 to	0.0240 0.0014	
mm 71.87		105	5	71.8 7° C 30 mm	79.39 68.43	5	0		-0.0648	4
t _e Density	1474.2		5	BP	54.3	2	m'	700 to	0,1230	4
g/ml 20°C	0.77	344	2	te (d, e)	50.76 50.13	5 5	n'	1 <u>1000 °K</u>	0.0012	
_d t 25	0.76	996	2	ΔHv/T	19.38	5	0'	i	-0.0 ₆ 40	4
	0.76		4	d 175 to	90.64	5		ace tension		
a b	0.78		4 4		0.1267	5	dyne	s/cm, 20°C 30	27.6 26.7	2
Ref. Index			Ť	d' 70 to e' 175 °C		5 5		40	25.8	2
n _D 20°C			2	1113	0.24	2	Para	chor [P]		
25 30	1.43		2 4	d g/ml vc ml/g tc °C	4. 195	2		20°C 30	671.0 671.5	4
"C"	0.74		4	1 -	452.	2	,	40	672.0	4
MR (Obs.)	+		5	P _c mm	10640.	2			658.2	5
MR (Calc.	76.08	8	2	PV/RT 71.87°C	1.0000	5	Exp.	. L.1.%/wt. u.		
(nD-d/2)	1.04		5	30 mm	1.0000	5	Disp	ersion		
Dielectric	2.06	-	5	BP t _e	0.9197 0.8937	5		h Point °C		
A 175 to B 358 °C			2 2	tc	0. 22	2		Point		
c	154.52		2	ΔHc kcal/m	2398.17	2	M S			
A* 175 to		090	5	ΔHf ΔFf			X-R	ay Dif.		
B* ∟330°C	1752.9		5	Viscosity	 	$\vdash \vdash$		red		
c	_			centistokes				bility in +		
t _k to		ļ		7 225 °C	0.4816 0.4322	2 2	Car	rbon tet.		
A' 70 to	7.33	309	5	265	0.3910	2	Ber Eth	nzene		
B' 175 °C	2036.4		5	285	0.3540	2	n-F	le ptane		
C'	172.5		5	B ^V 220 to A ^V 290 °C	626.8 7.42611	4 4	Eth Wa	anol		
A'* 70 to B'* 175 °C		108	5	(B ^V) to		-		ter in		
Ac 358 to	8,77	25	5	(A ^V) ₁ °C						
Bc t °C	4015.9	ł	5	c _p liq. °K		\vdash				
Cc	397.6		5	l -						
Cryos. A° consts. B°				c _p vap. °K						
t _e °C	319.7		5	c _w vap.	1					
$T_R = 0.8$	7 Т _с	,		·			+ gra	ms/100 gran	ns solven	t
REFEREN		ow 2	-AF	PI 3-Lit. 4-0	Calc. from det	t. dat				
SOURCE:			ΑI							
PURIFICAT	ION:		AI	PI						
LITERATU	RE REFI	EREN	CES	:						

		n-Hep	tade							No. 15	
NAME		n-nep	raue	Cane				ST	RUCTURAL	FORMUL	A
Mole % Pur. 99	o#	Ref.	Mo:	ecul		Molecular Weight 240.45			CH ₃ (CH ₂)	15 ^{CH} 3	
70 Ful. 77	. •	131	FUI	Ref.		Weight 240.45	Ref.	ī			Ref.
F.P. °C	T	21.98	0	2	dt/dP		1	f	1 40		-
F.P. 100	6	,0		-	°C/mm		1	g	to K		
B. P. °C					83.5°C BP	129.08	5 2	h			
760 mm		301.82 223.2		2 2	t _e	0.03618	5	f'	to		
30		188.36		4	30 mm	0.8733	5	g'	<u>•</u> K		
10		161.75 117.26		5	ΔHm cal/g			h'			
Pressure		111.20		-	ΔHv cal/g			m n	300 to	0.0238 0.0014	
mm 83.	5	0.03	572	5	83.5°C 30 mm	81.28 67.20	2 5	0	_600 <u>•</u> K	-0.0648	
t _e	1!	510.3		5	BP	52.8	2	m'	700 to	0.1161	├
Density g/ml 20°0	-	0.77	80 [≠]	2	t _e (d, e)	49.09 48.38	5	n'	1000 °K	0.0012	
dt 25		0.77	45	2	ΔHv/T _e	19.36	5	0'		-0.0 ₆ 39	4
		0.77		4	d 188 to	91.10	5	Sur	face tension		
a b		0.79		4	<u>e</u> 337 ℃	0.1269	5	dyn	es/cm. 20°C	28.0 [#]	2
Ref. Index	,	-0.03		-	d' 85 to e' 188 °C	88.3 0.1343	5		30 40	27.1 26.2	2
n _D 20°0		1.43	69 [‡]	2	d _c g/ml	0.1343	2	Par	achor [P]		
25 70	ł	1.43 1.41		2	v_ml/g	4.159	2		20°C 30	711.0	4
"C"	+	0.74		4	t	462.	2		40	711.6 712.1	4
MR (Obs.	\forall	80.96		2	P _c mm	9880.	2			697.2	5
MR (Calc.		80.70	6	5	PV/RT 83.5°C	1,0000	5	Ехр	. L.1.%/wt. u.		
(nD-d/2)	_	1.04		2	30 mm	1.0000	5	Dis	persion		
Dielectric		2.06		5	BP t	0.9176 0.8902	5		sh Point C		
A 188 to B 374 °C		7. 01 347. 82		2	te t _C	0.8702	2	<u> </u>	e Point		ļ
С		145.52		2	∆Hc kcal/m	2545.09	2		Spec. raV.		
A* 188 to		1.70		5	ΔHf ΔFf		1	X-F	lay Dif.		
B*[347°C	- 1	771.33		5	Viscosity				ared		<u> </u>
t, to	_				centistokes				ibility in Tetone		
t _k to					7 240 °C 260	0.4804 0.4330	2	Ca	rbon tet.	Ì	
A' 85 to		7, 30	95	5	280	0.3920	2		nzene her		
B' 1188°C		052.2		5	300 B _v 230 to	0.3550	2		Heptane		
A'* 85 to	-	1.96	829	5	A 310 °C	659.1 7.40037	4		hanol iter		
B'* 188 °C	_ 1	959.6	020	5	(B ^V) to				ter in		
Acl 374 to		8. 92	20	5	(A ^V) °C		1				
Bc tc C	43	310.1 414.4		5	c _p liq. °K					1	
Cryos. A	-	-1 -, 7		-	il .		İ				
consts. B					P						
t _e °C		336.6		5	c _v vap.						
$T_R = 0.8$	38 T	2			for undercool	ed liquid		+ gr	ams/100 gra	ms solver	nt
REFEREN	CES	: 1-D	ow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:					PI						
PURIFICA	TIO	N:		A	PI						
LITERATU	JRE	REF	ERE	NCES	5: 3 JACS 77 (2	2019) 1955 Sch	oere	r A. Ā	. et al.		
1											
1											
			709	·C							
# purity a	ppli	es to n	'ם	-							
L											

									No. 158	В
NAME	n-	Octade	cane			_	STR	UCTURAL	FORMULA	
Mole % Pur. 99.	9# R	ef. M	olecul	ar C ₁₈ H ₃₈	Molecular Weight 254.4	84		сн ₃ (сн ₂) ₁	6 ^{CH} 3	
			Ref.			Ref			************	Ref
F.P. °C		. 180	2	dt/dP	Ţ		f	to		
F.P. 100%		_		*C/mm 25*C	94805,4	5	g	<u>*</u> K		
B. P. °C 760 mm	316	12	2	BP	0.0630	2	h			
100	236	•	2	t _e	0.03608	5	f'	to		
30 10	173		4 5	30 mm	0.8890	5	g'		l	
1	128		5	ΔHm cal/g	57.65	3	h'	1 2001	0.022/	-
Pressure	†		\top	ΔHv cal/g	05.04		m n	300to 600°K	0.0236 0.0014	
mm 25°C	0.	.0473	5	25°C 30 mm	85.3 * 65.74	5	0		-0.0649	
t _e	1549	. 3	5	BP	51.5	2	m'	1 700 to	0, 1163	4
Density g/ml 20°C	1 0	. 7819 [‡]	2	te (d, e)	47.69 46.96	5	n'	1000°K	0.0012	4
at 25	0.	. 7785 ‡	2		I	5	01		-0.0 ₆ 39	4
4 30		. 7751	4	ΔHv/T _e	19.38	_	Surf	ace tension	,	
a L		. 7955	4	d 201 to		5 5		s/cm. 20°C	28.4 [‡]	2
b D. C. T. A.	+ -0.	. 0368	+-	a - 75 to	88.09	5	•	30 40	27.5 26.6	2 2
Ref. Index n _D 20°C	1	. 4390 [‡]	2	e' 201 °(+	5	Par	chor [P]		
D 25] 1.	.4369 [‡]	2	d g/ml v ml/g	0.24 4.322	2 2	1 41.	20°C	751.3	4
70		. 4191	3	tc °C	477.	2		30 4 0	751.9 752.2	4
"C"	0.	.7446	4	P _c mm	9880.	2		Sugd.		5
MR (Obs.) MR (Calc.)		. 61	5	PV/RT		\vdash	Exp	L.1.%/wt.		
(nD-d/2)	1	. 324 . 0480	5	25°C	1.0000	5		u.		
Dielectric	 	. 07	5	30 mm BP	1.0000 0.9178	5		ersion		ļ
A 201 to		. 0156	2	t.	0.8893	5		h Point °C Point		ĺ
B 1387 °C			2	tc	0.22	2	M S			
C	139.		2	ΔHc kcal/m	2692.01	2	Ultr			
A* 201 to B* 363 °C		. 7241 . 7	5	ΔFf	ļ		X-R Infra	ay Dif.		ł
K L	1			Viscosity						-
t. to	-		1	centistokes	0.701	2		bil ity i n † etone		1
t _k to				7 190 °C	0.616	2	Car	rbon tet.		
A' 25 to	7	. 3094	5	230	0.547	2	Ber Eth	nzene		l
B' [201 °C			5	250	0.490	2	n-F	l eptane		
C'	157.		5	B ^V 180 to		4 4	Eth Wa	anol		
A'* 25 to B'* 201 °C		. 03584 . 1	5		-)			ter in		
Acl 387 to	+	. 1876	5	(A ^V) to	1					
Bc t °C	4810.	. 2	5		· 	-				
Cc	449	. 57	5	c _p liq. ∘K						
Cryos, A° consts. B°				c _p vap. *K						
t _e °C	353.		5	c _v vap.						
$T_{\mathbf{R}} = 0.88$				for underco	oled liquid		+ gra	ms/100 gran	ns solven	t
REFERENC	ES: 1	-Dow	2-AI	PI 3-Lit. 4-	Calc. from det	da	ta 5-	Calc. by form	mula	
SOURCE:				PI						
PURIFICAT				PI						
LITERATU	RE RI	EFERE	NCES	3 JACS 77	(2019) 1955 S c	hoer	er A.	A. et al.		
# purity ap	plies	to n _D 7	0°C							

No. 159 n-Nonadecane NAME STRUCTURAL FORMULA $CH_3(CH_2)_{17}CH_3$ Mole Ref Molecular Molecular $C_{19}H_{40}$ % Pur. 99.5# Weight 268.510 3 Formula Ref. Ref. Ref. F.P. °C F.P. 100% 32.1 dt/dP f to °C/mm g °K 54.24°C 8682.84 5 B.P. °C h BP 0.0640 2 329.7 760 mm 2 0.03599 f١ 5 to 100 248. 2 <u>«к</u> g' 30 212.40 30 mm 0.9035 5 10 184.9 5 h! AHm cal/g 40.78 3 138.8 5 300 to 0.0237 m AHv cal/g Pressure°C 0.0014 n _600**°K** 54. 24℃ 83,23 5 5 mm 54.24 0.001 5 -0.0649 0 4 30 mm 1586.4 64.38 5 te BP 50.3 700 to m' 0.1164 4 Density te te (d, e) 46.37 5 'n' 0.7855 1000 °K 0.1165 g/ml 20°C 2 5 45.64 ۰, -0.0639 4 dt 4 25 0.7821 2 ΔHv/T_e 0.7787 19.40 5 30 4 Surface tension 212 89.88 5 to 28.7[‡] 27.8[‡] a 0.7991 4 dynes/cm. 20°C 2 <u> 369</u> °C 0.1201 5 -0.0₃68 4 2 30 ă٦ to 89.7 5 40 26.9 2 Ref. Index e¹ 212 0.1192 5 20°C 1.4409 [P] ⁿD dcg/ml vcml/g tc°C Parachor 2 0.24 1.4388 2 25 20°C 791.2 4.097 70 1.4211 3 30 791.8 4 ^tc 487. 2 792.2 40 4 "C" 0.7442 4 P_c mm 9120. 2 Sugd. 775.2 5 MR (Obs.) 90.25 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 89.942 54.24°C 1.0000 (nD-d/2) 1.0481 4 1.0000 30 mm 5 5 Dispersion Dielectric 2,08 5 BP 0.9184 Flash Point °C 0.8885 5 A | 212 to 7.0192 2 Fire Point 0.22 2 B | 403 °C 2 1917.0 M. Spec. Ultra V. C AHc kcal/m 2838, 94 133.5 2 2 ΔHf A# 212 to 1.7427 5 X-Ray Dif. ΔFf B*| 379 °C 1839.43 Infrared Viscosity Solubility in centistokes Acetone 190 °C 0.757 2 Carbon tet. °C 210 0.662 2 Benzene 230 0,586 2 25 to 7.31561 Ether B' 250 0,524 2127.54 5 212 °C n-Heptane B. 200 to 5 152 641.7 4 Ethanol A 1 260 ℃ 2.49284 Water A'* 25 to B'* 212 °C 2.06337 5 Water in (B^V) 2054.7 5 to Ac | 403 to 9.4412 5 (A^V) °C Bc t_c°C 5302. ۰ĸ cp liq. Cryos. A cp vap. °К consts. B° c, vap. te °C 368.49 $T_{R} = 0.89 T_{c}$ for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES: 3 JACS 77 (2019) 1955 Schoerer A.A. et al. # purity applies to nn 70°C

								No. 160)
NAME	n-Eicosa	ne		· · · · · · · · · · · · · · · · · · ·	_	STRUCTURA	L F	FORMULA	٤.
Mole	# Ref. Mo	lecul	ar C ₂₀ H ₄₂	Molecular		сн ₃ (сн	[[] 2 ⁾ 18 ⁽	CH ₃	
% Pur. 99.	9" 3 Fo		a 20 42	Weight 282.5		f			Ref
	1 2/ 2	Ref.	<u> </u>	r	Ref.				Ker.
F.P. °C F.P. 100%	36.8	2	dt/dP °C/mm	2000 41	٠, ا	g	to ºK_		
B. P. °C			61.35°C BP	8808.41 0.06492	5	h			L
760 mm 100	342.7 260.	2 2	te	0.03614		f'	to		
30	223.61	4	30 mm	0.9174	5	g'	_K_		1
1 0 1	195.7 148.9	5 5	ΔHm cal/g	59.11	3	h' m 300		0, 0236	_
Pressure°C			ΔHv cal/g	70.44	5	m 300 n 600		0.0236	1
mm 61.35	0,001	5	61.35°C 30 mm	79.66 63.08	5	0	-7	-0.0 ₆ 49	
Te eite	1608.	5	BP	48.8	2	m' 700	to	0.1162	4
Density g/ml 20°C	0.7887	2	te te (d, e)	44.80 43.99	5	n' 1000		0.0012	4
_d t 25	0.7853 ⁺	2	ΔHv/Te	19.29	5	o'		-0.0 ₆ 39	4
	0.7550	3	d 224 to	 	5	Surface tensi		±	
a b	0.8023 -0.0 ₃ 68	4 4	<u>e 383</u> °C	0.1199	5	dynes/cm. 2	0	29.0 [‡] 28.1 [‡]	2
Ref. Index	#		e' 61 to		5	4		27.2	2
n _D 20°C	1.4426 [‡] 1.4405 [‡]	2 2	d g/ml v ml/g	0.24	2	Parachor [P		831.3	4
70	1.4230	3		4.247	2	3		832.0	4
"C"	0.7439	4	t _c °C P _c mm	502. 8360.	2 2 ⁻	40		832.4 814.2	4 5
MR (Obs.)	94.90	4	PV/RT	0300.	-	Exp. L.1.%/			
MR (Calc.) (nD-d/2)	94.56 1.0482	2 4	61.35°C	1.0000	5	u.			
Dielectric	2.08	5	30 mm BP	1.0000 0.9122	5	Dispersion	_		<u> </u>
A 224 to	7.0225	2	te	0.8810	5	Flash Point ° Fire Point	١		
B 1417_°C C	1948.7 127.8	2 2	t _c	0. 21 2985. 86	2	M Spec.			
A* 224 to B* 393 °C	1.77166 1874.62	5	ΔHf ΔFf	2,00,00		Ultra V. X-Ray Dif. Infrared			
K — — —			Viscosity centistokes		ŀ	Solubility in	+		
t _k to			7 190 ℃	0.814	2	Acetone Carbon tet.			
t C A' 25 to	7 30070	اــِــا	210 230	0.710 0.627	2 2	Benzene	1		
B' 224 °C	7.30970 2155.78	5	250	0.560	2	Ether n-Heptane			
C'	146.	5	B ^V 180 to	651.4	4	Ethanol			
A'* 25 to B'* 224 °C	2.07885 2085.2	5	A ^V 260 °C (B ^V) to	Z. 50316	4	Water Water in			
Ac 417 to	9.6945	5	(A ^V) °C						
Bc t _c °C	5807. 513.	5	cp liq. °K						
Cryos. A° consts. B°	513.	3	c _p vap. °K						
t _e °C	382.84	5	c _w vap.						
$T_{R} = 0.89$	T _c		≠ for undercod	oled liquid		grams/100	gram	ns solveni	t
	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc. from det	da	ta 5-Calc. by	forn	nula	
SOURCE:		AP							
PURIFICAT	ION:	AP	I						
LITERATUE	RE REFERE	NCES	: 3 JACS 77 (2019) 1955 Sch	noere	er A.A. et al.			
# purity app	lies to n _D 70°	С							

										No. 16	
NAME		n	-Hene	icosa	ne			ST	RUCTURAL	FORMUL	.A
									CH ₃ (CH ₂) ₁	9СН3	
Mole % Pur. 9	9.9 ¹	Re:	f. Mo	lecul rmula	ar C ₂₁ H ₄₄	Molecular Veight 296.	562				
				Ref.			Ref.				Ref.
F.P. °C F.P. 100°	,	40.	5	2	dt/dP	1		f	to		
B. P. °C	-			+	°C/mm 114.64°C.	150.40	5	g	°K		
760 mm		356.		2	BP	0.0657 0.0360	4 5	-h -f'			
100		271. 232.		4	t _e 30 mm	0.9765	5	g'	to °K		
10	- .	203.	07	5	ΔHm cal/g	38.44	3	h'			
1	+	152.	94	5	ΔHv cal/g		Ť	m	to		
Pressure mmll4.	54	0.		5	114.64°C.	68.7	5	n o	°K		
t _e	1	607.	4	5	30 mm BP	58.50 47.05	5	m'			₩
Density g/ml 20%	_	0	7917‡	2	t _e	43.69	5	n'	to o		ŀ
t 25	٦	0.	7883 <u>*</u>	2	te (d, e)	43.30 19.33	5	٥'			
4 70	\perp	0.	75831	3	ΔHv/T _e d 232 to	80.06	5	Sur	face tension		1
a b			8053 0368	4	e 397 °C	0.0926	5	dyn	es/cm. 20°C 30	26. 91 26. 00	5
Ref. Inde:	+		,	+	d' 115 to e' 232 °C	78.58 0.0862	5	"	40	25.11	5
n _D 20%		1.	4441 [‡]	2	d _c g/ml	3.0002	+	Par	achor [P]		T
70			4420 [*] 42471	2	v_ml/g				20°C ⋅ 30		1
"C"	\top	0.	7434	4	II _*				40	853.2	5
MR (Obs.		99.		2	P _c mm		+	Form	Sugd. L.1.%/wt.	055.2	+
MR (Calc (nD-d/2)	.)	99.	178 0483 [≠]	5 2	114.64°C.	1.0000	5	Exp	u.	4	l
Dielectric	+	2.		5	30 mm BP	1.0000	5		persion	98. ≠	2
A 115 to B 492 •	,		47174	5 5	t e t	0.8948 0.8624	5	Fir	sh Point °C e Point		
c '		171.	5	5	∆Hc kcal/m		1		Spec. ra V.		
A* 115 to B* 410 °		2.3 343.	24451	5	ΔHf ΔFf			X-I	Ray Dif.		
к — —	- -				Viscosity				ability in +		+
t _k to					centistokes η °C			Ac	etone rbon tet.		
t _x °C				╁					nzene her		
B'°	길			1	DV I		+	n-	Heptane		
A'* 110 to	+	2	15834	5	B ^V				hanol ater		
B'* 233 °C		309.		5	(B ^V)				ter in		
Acl to				1	(A ^V)						
Bc tc °C	-				c _p liq. °K						
Cryos. A consts. B					c _p vap. °K						
t _e °C		396.	79	5	c _w vap.	1					
# for und	erco	oled	liquio	i.				+ gı	rams/100 gra	ms solve	nt
REFEREN	CES	: 1.	-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	et. da				
SOURCE:			API								
PURIFICA			API								
LITERAT	JRE	RE	FERE	NCE	S: 3 JACS 77 (2	:019) 1955 S ci	hoere	r A.A	. et al.		
l purity app	lies	to n	7 0°C	C. and	ı d ^t 4						

							No. 162	2
NAME	n-Doc	osane	•			STRUCTURAL 1		
Mole % Pur. 99.	71 Ref. Mo	lecul rmul		Molecular Weight 310.5	88	СН ₃ (СН ₂) ₂₀ СН	¹ 3	
		Ref.			Ref			Ref.
F. P. °C	44.4	2	dt/dP			f to		
F.P. 100%	<u> </u>		*C/mm 122.90°C.	149.36	5	g <u>*K</u>		
B. P. °C 760 mm	368.6	2	BP	0.0666	4	h		
100	282.39	4	t _e	0.0359	5	f' to		
30 10	243.15 212.86	4 5	30 mm	0.9920	5	g' K		
1	161.88	5	∆Hm cal/g	37.67	3	h'		├
Pressure			ΔHv cal/g 122.90°C.	67.10	5	m to		
mm122.9		5	30 mm	67.19 57.28	5	0		
t _e	1637.4	3	BP	46.00	5	m' to		\vdash
Density g/ml 20°C	0.7944	2	te te (d, e)	42.60 43.38	5	n' K		
_d t 25	0.7910	2	ΔHv/T _e	19.82	5	o'		
	0.76311	3	d 242 to	79.14	5	Surface tension		
a b	0.8080 -0.0368	4	e 410 °C	0.0899	5	dynes/cm. 20°C	27.12 26.20	5
Ref. Index		<u> </u>	d' 110 to		5 5	40	25.31	5
n _D 20°C	1.4455	2		0.0024	+	Parachor [P]		
25 70	1.4435 [‡] 1.4260 ¹	2	d g/ml v ml/g	1		20°C		
"C"	0.7431	4	t _c °C		1 1	40		
MR (Obs.)	+	2	P _c mm			Sugd.	892.2	5
MR (Calc.	1 103 706	5	PV/RT 122.90°C.			Exp. L.1.%/wt.		
(nD-d/2)	1.0483	2	30 mm	1.0000	5	u. Dispersion	98. [≠]	2
Dielectric	2.09	5	BP	0.8943	5	Flash Point °C		
A 242 to B 499°C		5	te t _c	0.8609	"	Fire Point		<u> </u>
c Live	169.3	5	ΔHc kcal/m		1	M Spec.		
A* 242 to	2.28329	5	ΔHf			Ultra V. X-Ray Dif.		
B* 420 °C	2401.2	5	ΔFf		-	Infrared		
с			Viscosity centistokes			Solubility in +		
tk Tto		ļ	η •c			Acetone Carbon tet.		1
x I					1	Benzene		
A' B'						Ether n-Heptane		1
C'			B ^V to	(Ethanol		
A'* 110 to		5	AV _ °C			Water Water in		
B'* 242 °C		5	(B ^V) to	1				†
Ac to			(A ^V) °C	ļ	1			
Cc C		ļ	c _p liq. °K					
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	410.47	5	c _v vap.					
	ercooled liquid					+ grams/100 gran	ns solven	t
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from de	t. dat	ta 5-Calc. by for	nula	
SOURCE:	AF	PII						
PURIFICA?					_			
LITERATU	RE REF ER EI	NCES	5: 3 JACS 77 (2019) 1955 Sc	hoere	r A.A. et al.		
l purity app	olies to n _D ^{70°6}	C. an	d d ^t					

No. 163 NAME n-Tricosane STRUCTURAL FORMULA $CH_{3}(CH_{2})_{21}CH_{3}$ Molecular C23H48 Mole Ref. Molecular % Pur. 99.61 Weight 324.614 Ref. Ref. Ref. F.P. °C F.P. 100% 47.6 2 dt/dP f to 130.85°C. 25°C g ١ °K 151.91 B. P. °C h 0.0675 BP 2 760 mm 380, 2 2 0.0358 5 ſ١ to 100 292.84 4 g' °К 30 1.0068 5 253.03 4 30 mm 10 222,27 5 h' 39.74 ∆Hm cal/g 3 170.48 5 m to ∆Hv cal/g Pressure°C mm130.85 ۰ĸ 65.77 n 130.85°C. 5 0.1 5 o 30 mm 56.07 5 1666.0 te 5 BP 44.99 5 m' | to Density 41.56 te te (d, e) 5 0.7969# n' °K g/ml 20°C 5 42.38 ٥, $\mathbf{d_{4}^{t}}$ 25 0.7935⁷ 0.7641 AHv/T 19.83 5 70 d 252 Surface tension to 78.11 5 a 0.8105 dynes/cm. 20°C 27.31 e | 442 d' | 125 1 423 0.0871 <u>°C</u> 5 ь -0.0368 4 30 26.39 to 76.16 40 25.49 5 Ref. Index e' | 252 0.0794 1.4468 20°C ⁿD Parachor [P] d_c g/ml 1.4448 25 20°C 2 vc ml/g t_°C 70 1,4276 30 t_c 40 "C" 0.7428 4 P_c mm Sugd. 931.2 5 MR (Obs.) 108.80# 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 108.414 5 130.85°C. 1.0000 1.0484[‡] (nD-d/2) u. 2 98. [‡] 30 mm 1.0000 2 Dispersion Dielectric 2.09 5 BP 0.8936 Flash Point °C 0.8594 A 252 to 7.51789 Fire Point 2537.9 B _506°C M. Spec. С ΔHc kcal/m 167.1 5 Ultra V. ΔHf A* 252 to 2.31840 5 X-Ray Dif. ΔFf B* 433 °C 2456.0 Infrared ĸ Viscosity Solubility in centistokes Acetone to $\mathbf{t}_{\mathbf{k}}$ Carbon tet. °C Benzene ۸' Ether B١ n-Heptane B_v | C' Ethanol °C Water A** 125 to 2.22338 B'* 252 °C Water in (B^V)| 2420.4 Acl (A^V)| to Bc °C cp liq. ۰ĸ Cc Cryos. A° c_p vap. °K consts. B° c_v vap. t, °C 423,63 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: LITERATURE REFERENCES: 3 JACS 77 (2019) 1955 Schoerer A. A. et al. 1 purity applies to $^{70^{\circ}\text{C}}$. and 4

							No. 164	
NAME	n-Te	tracc	sane			STRUCTURAL	FORMULA	¥
						CH (CH) C	·u	
Mole	1 Ref. Mo	lecul	ar	Molecular		CH ₃ (CH ₂) ₂₂ C	^{,11} 3	
% Pur. 99.	9 ¹ 3 Fo	rmul	C ₂₄ H ₅₀	Weight 338.6	40			
	·	Ref.		<u> </u>	Ref.			Ref.
F.P. °C F.P. 100%	50.9	2	dt/dP *C/mm			f to		İ
B. P. *C	 	-	138.4°C.	154.37	5	g <u>*K</u>		
760 mm	391.3 302.82	2 4	BP t _e	0.0683 0.0358	5	f' to		\vdash
100 30	262.46	4	30 mm	1.021	5	g' '°K_		
10 1	231.3	5	ΔHm cal/g	38,74	3	h'		L
Pressure (H	ΔHv cal/g			m to		
mm 138, 4	0.1	5	138, 4°C. 30 mm	64.4 54.93	5 5	0		
Densites	1694.4	5	BP	44.05	5	m' to		
Density g/ml 20°C	0.7991	2	te te (d, e)	40.59 40.35	5 5	n'		
dt 25 4 70	0.7958 0.76571	2 3	ΔHv/T	19.40	5	o'		<u> </u>
a 10	0.8123	4	d 1260 to		5	Surface tension dynes/cm. 20°C	27.47	5
ь	-0.0366	4	$\frac{e}{d}$, $\frac{1}{135}$ $\frac{6}{to}$		5	30	26.58	5
Ref. Index	1.44804	2	e' 260 °C		5	40	25.70	5
25	1.4460	2	d g/ml vc ml/g			Parachor [P] 20°C		
70	1.42861	3	t _c °C			30 40		
"C"	0.7427	5	P _c mm			Sugd.	970.2	5
MR (Obs.) MR (Calc.)	113.44 [‡] 113.032 ,	2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0484	2	138.4°C. 30 mm	1.0000	5	u. Dispersion	98.≠	2
Dielectric	2.10	5	BP	0.8933 0.8583	5	Flash Point °C	,	 -
A 260 to		5	t _e t _c	0.8383		Fire Point		ļ
<u> </u>	165.1		∆Hc kcal/m			M Spec. Ultra V.		
A* 260 to B* 450 °C		5	ΔHf ΔFf			X-Ray Dif.		
к — — —	- 2307.3		Viscosity			Infrared Solubility in +		
t _k	-		centistokes 7°C			Acetone		
t _{x 1}			'			Carbon tet. Benzene		
A' to B' C						Ether		
c,	-1	1	B ^v to			n-Heptane Ethanol		
A'* 135 to			AV °C	_[Water Water in		
B'* 260 °C	 	5	(B ^V) to	1				\vdash
Bcit °C			(A ^V) °C	+	\vdash			
Cc		1-	c _p liq. ∘K					
Cryos. A° consts. B°			c _p vap. °K					1
te °C	436.24	5	c _w vap.					
	rcooled liquid					+ grams/100 gran	ns solven	t
	ES: 1-Dow	2-AI		Calc. from de	t. dat	ta 5-Calc, by for	mula	
SOURCE:	YON.	API						
PURIFICAT	RE REFERE	API		(3010) 1055 5				
LIIERATU.	RE REFERE	NUES	5: 3 JACS 77	(2019) 1955 S c	hoer	er A.A. et al.		
l purity ann	lies to n _D	.	t					
Party app	D	and	4					

								No. 16	5
NAME	n-Penta	cosan	e			ST	RUCTURAL	FORMUI	A
Mole % Pur. 9	9.8 ¹ Ref. M	lolecul ormul	ar C ₂₅ H ₅₂	Molecular Weight 352.6	66		CH ₃ (CH ₂) ₂₃	CH ₃	
		Ref.		T	Ref.				Ref.
F. P. ℃	53.7	2	dt/dP			f	to		T
F.P. 1009	6		°C/mm 145.7°C.	156.7	5	g	•K		
B. P. °C 760 mm	401.9	2	BP	0.0690	2	_h_	l		
100	312.36	4	t _e	0.0357	2	f'	to		
30 10	271.48	4 5	30 mm	1.0345	5	g'	*K		
10	186.55	5	ΔHm cal/g	39.13	3	h'			+
Pressure	С	1	ΔHv cal/g	(2.00	_	m n	to °K		
mm 145.		5	145.7°C.	63.08 53.82	5 5	0			
t _e	1720.0	5	BP	43.12	5	m'	to		+
Density g/ml 20°0	0.8012	<u> </u>	t _e (d, e)	39.64 40.36	5	n'	°K		
at 25	0 7070		ΔHv/T _e	19.84	5	٥'			
	0.7693		d 270 to	76.09	5		face tension		
a b	0.8144 -0.0366	4	e 447 °C	0.0820	5	dyn X	es/cm. 20°C 30	27.63 26.73	5
Ref. Index		.	d' 140 to e' 270 °C	73.82 0.0737	5		40	25.86	5
n _D 20°0		2	d _c g/ml	1 0.0.3.	+-	Paı	achor [P]		
25 70	1.4471	2	V_ml/g				20 °C 30		
"C"	0.7424	5	₁₁ -		1 1		40		
MR (Obs.	·	2	P _c mm	ļ	$\perp \perp \downarrow$			1009, 2	5
MR (Calc.) 117.65	. 5	PV/RT 145.7°C.	1.0000	5	Exp	L.1.%/wt.		
(nD-d/2)	1.0485		30 mm	1.0000	5	Dis	u. persion	98. [‡]	2
Dielectric		5	BP t te	0.8927 0.8570	5	Fla	sh Point °C		1
A 270 to B 500 °C		5 5	t e				e Point		_
с '	163.1	5	ΔHc kcal/m				Spec. ra V.		
A* 270 to			ΔHf ΔFf			X-1	Ray Dif.		
B*[460 °C	2560.0	5	Viscosity	 	\dagger		ared		4
·	_		centistokes				ubility in E		
t _k to			η °C			Ca	rbon tet.		
A' I		+-					nzene her		
В'	_		- <u>v</u>	+	+-	n-	Heptane		
C'			B ^V				hanol ater		
A'*140 to B'*270 °C		5	(B ^V)	-			ter in		
Acl to		Ť	(A ^V)						
Bc t _c °C			c _p liq. °K		+				
Cc			13						
Cryos, Acconsts, B			c _p vap. *K						
t _e °C	448.26	5	c _w vap.			L,_			
	rcooled liqui						rams/100 gra		nt
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc, from de	et. da	ta 5	-Calc. by for	mula	
SOURCE:		API							
PURIFICA		API							
LITERATU	IRE REFER	ENCE	S: 3 JACS 77 ((2019) 1955 S c	hoere	r A.	A. et al.		
lpurity app	lies to nD	C. and	$\mathbf{d}_{4}^{\mathbf{t}}$						

							No. 166)
NAME	n-Hexaco	osane				STRUCTURAL		
Mole	Ref. M	olecul	ar	Molecular		CH ₃ (CH ₂) ₂	4 ^{CH} 3	
% Pur. 99.	8 ¹ 3 F	ormul	ar C ₂₆ H ₅₄	Weight 366.	692			
	1 5/ 4	Ref.			Ref		r	Ref.
F.P. °C F.P. 1009	56.4	2	dt/dP °C/mm			f to		
B.P. °C	412.2	1	152.75°C. BP	158.97 0.0698	5 2	h		İ
760 mm 100	412.2 321.64	2 4	t _e	0.0356	5	f' to		ĺ
30 10	280.25 248.22	4 5	30 mm	1.0477	5	g' °C		
1	194.18	5	ΔHm cal/g ΔHv cal/g	 	₩	m to		—
Pressure mm152.7		5	152.75°C.	61.84	5	n 'K		
t _e	1746.1	5	30 mm BP	52.78 42.24	5	m' to		-
Density g/ml 20°(0.8032	2	te te (d, e)	38.72 39.42	5	m' to K		İ
dt 25	0.7998 ⁷ 0.7704 ¹	2	ΔHv/T _e	19.82	5	0'		<u> </u>
a 10	0.8168	4	d 278 to		5	Surface tension dynes/cm. 20°C	27.79	5
ь	-0.0368	4	d' 140 to	72.68	5	30 40	26.86 25.95	5
Ref. Index	~ 1 45017	2	e' 278 °C	0.0710	5	Parachor [P]	23.73	-
D 25	1.4501 1.4481 1.43101	2	d g/ml vc ml/g			20°C 30		ĺ
"C"	0.7421	5	te of		l	40	1040 3	
MR (Obs.		2	P _c mm			Sugd. Exp. L.1.%/wt.	1048.2	5
MR (Calc. (nD-d/2)	1.0485	5 2	152.75°C. 30 mm	1.0000	5	u.	98.≠	
Dielectric	2,103	5	BP	1.0000 0.8924	5	Dispersion Flash Point °C	98.	2
A 278 t	7.57689 2692.73	5	t _e t _c	0.8560	5	Fire Point		
С	161.2	5	AHc kcal/m		<u> </u>	M Spec. Ultra V.		
A* 278 to B* 470 °C	2.41467 2.609.5	5	ΔHf ΔFf			X-Ray Dif.		
к — — -	-		Viscosity			Infrared Solubility in +	 	-
t _k			centistokes 7°C	:		Acetone Carbon tet.		
t'x °(-	_	·			Benzene		
B'		İ	V 1	ļ 	ļ	Ether n-Heptane		
A'* 140 to	2,32296	+_	B ^V to		İ	Ethanol Water		
B' # 278 °		5 5	(BV)	-		Water in		_
Ac to			(A ^V)]		
Bc tc_°			c _p liq. °C					
Cryos. A' consts. B'			с _р vap. °К					
	459.98	5	c _w vap.	}				
# for unde	rcooled liquid	1				grams/100 gra		<u> </u>
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE: PURIFICA	TION	API	•					
		API	S: 3 JACS 77 (2	(019) 1955 Sch	nere	r A A et al		
			(2	/, 1/33 GCR	1 E1	. A.A. C. M.		
purity app	lies to nD	and	d 4					

No. 167 n-Heptacosane NAME STRUCTURAL FORMULA CH3(CH2)25CH3 Molecular 380.718 Molecular Mole % Pur. 99.01 C27H56 Formula Ref Ref. Ref. 59.0 F. P. °C 2 dt/dP f to F.P. 100% °C/mm g °K 159, 55°C. 161,13 B. P. °C h BP 0.070 2 760 mm 422.1 2 ^te 0.03463 5 ſ١ 330,55 to 100 g' °К 288.67 4 30 30 mm 1.0544 5 256.26 5 10 h' 37.93 3 ∆Hm cal/g 201.54 5 to m ΔHv cal/g 159.55°C. Pressure°C ۰ĸ n 60.65 0.1 5 mm159.55 o 30 mm 51.8 5 1770.4 5 t_e ВP 41.4 5 m' to Density 5 38.6 te te (d, e) n' ۰ĸ g/ml 20°C 0.8050 2 38.76 ۰, 0.8016 0.7732 d_4^t 25 2 19.90 5 AHv/T 70 3 Surface tension d 285 74.30 5 to 0.8186 4 dynes/cm. 20°C 27.93 0.0779 470 d 150 °C 26.99 26.09 ь -0.0368 4 30 71.58 to 5 40 Ref. Index e' | 285 0.0685 5 1.4511 ⁿD 20°C [P] 1.4491 Parachor d_c g/ml 2 25 20°C 1.43211 vc ml/g 70 3 30 ^tc 40 "C" 0.7420 4 P_c mm Sugd 1087.2 5 127.37# MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 126.886 5 1.0000 159,55°C. 1.0486 (nD-d/2) u. 2 98.≠ 1.0000 5 30 mm 2 Dispersion Dielectric 2.11 5 BP 0.8919 5 Flash Point °C 0.8548 5 A 285 to 7.59371 27**40**.08 5 Fire Point B 1520 °C 5 M. Spec. AHc kcal/m С 159.3 5 Ultra V. ΔHf A* 285 to 2.4430 265**6.**4 5 X-Ray Dif. ΔFf B*|5 00 °C Infrared ĸ Viscosity Solubility in centistokes Acetone to t_k ∫ t_x ∣ Carbon tet. °C Benzene A' Ether В' n-Heptane B_v | C' to Ethanol °C Water A'* 150 to 2.35017 5 B'* 285 °C Water in 2614.8 (B^V)| Aclto (A^V) Bc t_c °C cp liq. °K Cc Cryos. A. c vap. °K consts. B° c_w vap. te °C 471.24 5 for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES: 3 JACS 77 (2019) 1955 Schoerer A.A. et al. 1 purity applies to nD 70°C. and dt

										No. 168	
NAME		n-O	ctaco	sane				STR	UCTURAL I	FORMULA	
Mole % Pur. 99	. 0 ¹	Ref.	Mo Fo	lecul	ar C ₂₈ H ₅₈	Molecular Weight 394.	744		СН ₃ (СН ₂) ₂₆ (сн ₃	
				Ref.			Ref.	1			Ref.
F. P. *C		61.4		2	dt/dP	T		f	to		
F. P. 1009	1				*C/mm	162.21	5	g	°K		l
B.P. °C 760 mm		431.6		2	0.1 mm BP	163.21	2	<u> </u>			
100 mm	-	339.1	0	4	t _e	0.03475		f'	to		
30		296.7 263.9		4 5	30 mm	1.0682	5	g'	°K		
10 1		208.5		5	ΔHm cal/g	39.14	3	h'			<u> </u>
0.1		166.0	6	5	ΔHv cal/g	50.50	_	m (to °K		l
Press. mn		727 2		[]	0.1 mm 30 mm	59.50 50.8	5				Į
t _e	1.	737.2		5	BP	40.6	5	m' l	to		
Density g/ml 20°0	:	0.8	067‡	2	te (d, e)	37.8 37.77	5	n' i	•K		1
at 25		0.8	033₹	2	ΔHv/T	19.79	5	0'			ĺ
	_		7501	3	d 295 to		5	Surfa	ce tension		T
a b		0.8 -0.0		4 4	e 480 °		5	dyne	s/cm. 20°C	28.06 27.12	
Ref. Index	+		,	H	d' 162 to		5		40	26.21	5
n _D 20°0		1.4	520‡	2		0.0000	1	Para	chor [P]		
25 70	1		50 0 * 3 30 1	2 3	d g/ml vc ml/g	1	1	l	20°C		
"C"	╅╌	0.7		4	, c	1	l		40		l
MR (Obs.)	+	132.0		2	P _c mm			L	Sugd,	1126.2	5
MR (Calc.		131 5	Ω4.	5	PV/RT		_	Exp.	L.1.%/wt.		
(nD-d/2)	4_		487 [‡]	2	0.1 mm 30 mm	1.0000	5	Disp	u. ersion	98. ≠	2
Dielectric		2.1	1	5	BP	0.8912	5		h Point °C		-
A 295 to B 1520 °C		7.6 7 8 5.8	0972 0	5 5	t _e t _c ΔHc kcal/m	0.8535	5	M Sp			-
A* 295 to B* 490 °C		2.4 702.8	7084	5	ΔHf ΔFf			Ultra X-Ra Infra	y Dif.		
c k — to	-				Viscosity centistokes 7 °C	:		Ace	oility in +		
A' to	+				•			Ben Eth			
B' L _ <u>'</u>	-			1 1	B ^V to				eptane anol		
A'* 162 to B'* 295 *C		2.3 658.5	7498	5 5	$\frac{ \mathbf{A}^{\mathbf{V}} }{ (\mathbf{B}^{\mathbf{V}}) } - \frac{\circ}{\circ}$	<u>:</u>		Wat Wat	er er in		ļ
Ac to Bc tc *C					(A ^V) c _p liq. •k	:					
Cryos, Acconsts, B				\prod	c _p vap. *F	:					
te °C		480.		5	c _w vap.	1	ļ	1			
# for unde	rco	oled l	quid					+ gra	ms/100 gran	ns solven	t
REFEREN	CES	: 1-D	ow	2-AF	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:				AP	I						
PURIFICA				AP							
LITERATU	RE	REF	ER EI	NCES	: 3 JACS 77 (2019) 1955 S ch	oere	er A. A.	et al.		-
l purity app	lies	to n _D	70°C	and o	d ^t						

								No. 169	
NAME _	n-Nonacosane					STRUCTURAL FORMULA			
						CH (CH) CH			
Mole Ref. Molecular C Molecular						$CH_3(CH_2)_{27}CH_3$			
Mole % Pur. 99.		mul	C ₂₉ H ₆₀ V	Molecular Veight 408.7	770				
Ref. Ref.									Ref.
F.P. °C	63.7	2	dt/dP			f	to		
F.P. 100%			°C/mm 0.1 mm	165,23	5	g	°K		
B.P. °C 760 mm	440.8	2	BP	0.071.	5.	_h_			
100	347.38	4	t _e 30 mm	0.0347 1.0796	5	f' g'	to °K		
30 10	304.59 271.43	4 5	ΔHm cal/g	1.0796	-	h'			
1 0, 1	215.40	5 5	ΔHv cal/g		\vdash	m	to		
Press. mm	172, 36	2	0.1 mm	58.40	5	n o	*K		
t _e	1816.9	5	30 mm BP	49.9 39.8	5 5	m'			
Density g/ml 20°C	0.8083 [‡]	2	te te (d, e)	37.1 36.94	5 5	n'	to K		
dt 25	0.8049, 0.8015	2	ΔHv/T _e	19.78	5	۰'			
		4	d 302 to	72.49	5		face tension	20±	_
a b	0.8219≠ -0.0368≠		e 491 °C	0.0742	5	dyn	es/cm. 20°C 30	28.18 [‡] 27.25 [‡]	5
Ref. Index	4		d' 169 to e' 302 °C	69.47 0.0643	5		40	26.33 [‡]	5
ⁿ D 20°C	1.4529 1.4508	2	d _c g/ml			Par	achor [P]		
30	1.4488	4	vc ml/g t _c °C				30		
"C"	0.7418	4	P _c mm				40 Sugd.	1165.2	5
MR (Obs.) MR (Calc.)	136.65# 136.122 /	2 5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	1.0487	2	0,1 mm 30 mm	1.0000 1.0000	5 5	Dis	u. persion	98.¥	2
Dielectric	2.11	5	BP	0.8910 0.8527	5		sh Point °C		_
A 302 to B 500 °C	7.62529 2830.55	5	te t	0.0527			e Point		
c			ΔHc kcal/m ΔHf				Spec. ra V.		
B* 501 °C	2.49691 2746.2	5	ΔFf				Ray Dif. ared		
к ———			Viscosity				ability in +	ļ ———	-
t _k – to			centistokes 7°C			Ac	etone		
'x			•				rbon tet.		
A' to B' °C					1		he <i>r</i> Heptane		
C'			B ^V to A ^V °C			Et	hanol		
A'* 169 to B'* 302 °C	2.39938	5	$\frac{A^{\vee}}{ B^{\vee} } - {^{\circ}C}$				ater ater in		
Acl to		Ť	(A ^V)						
Bc tc C			c _p liq. °K						
Cryos. A°	 		[]						
consts. B°			Р *						
t _e °C	492.51	5	c _v vap.			L			
# for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula									
	ES: 1-Dow			Calc. from de	et. da	ta 5	-Calc. by for	mula	
SOURCE: API PURIFICATION: API									
PURIFICATION: API LITERATURE REFERENCES: 3 JACS 77(2019) 1955 Schoerer A. A. et al.									
January 1755 Schoerer A. A. et al.									

							No. 17	0
NAME	n-	Triac	ontane			STRUCTURAL	FORMULA	L
						CH (CH)	CU	
					\neg	сн ₃ (сн ₂) ₂₈	0113	
Mole 99.	0 ¹ Ref. M	olecul ormul	ar C ₃₀ H ₆₂	Molecular Weight 422.	796			
	1311	Ref.		weight ibb.	Ref	f		Ref.
F. P. *C	65.8	2	dt/dP	1	IX.		T	1
F.P. 1009	-	+-	*C/mm			f to g °K	İ	ł
B.P. °C		1	0.1 mm BP	167.20 0.072	5	h .		1
760 mm 100	449.7 355.40	2	t	0.03469	5 5	<u>f</u> ' + to		
30	312.16	4	30 mm	1.0907	5	g' °K		ļ
10 1	278.65 222.00	5 5	ΔHm cal/g			h'		
0. 1	178.46	5	ΔHw cal/g	ĺ		m to		1
Press. mn			0.1 mm 30 mm	57.33 49.0	5			l
t _e	1838.9	5	BP	39.1	5	m' l to	<u> </u>	-
Density g/ml 20°0	0.8097		te t (d, e)	36.3	5	n' 'K		l
dt 25	0.8064		t _e (d, e) ΔHv/T _e	36.13 19.77	5	°'		İ
	0.8031		d 309 to	<u> </u>	5	Surface tension		
a b	0.8229 -0.0366		<u>e 510</u> °C	0.0720	5	dynes/cm. 20°C	28.29 27.37	5 5
Ref. Index		,	d' 170 to		5 5	40	26.49	
n _D 20°0	1.4536	₹ 2		3.6023	-	Parachor [P]		
70	1.4516 1.4348	i 2	d _c g/ml v _c ml/g t _c °C			20°C		l
"C"	0.7415	4	11 -			40		
MR (Obs.)	+	2	P _c mm				1204.2	5
MR (Calc. (nD-d/2)) 140.74	. 5	PV/RT 0,1 mm	1.0000	5	Exp. L.1.%/wt. u.	١.	
Dielectric	1.0487		30 mm	1.0000	5	Dispersion	98. ≠	2
A 309 to		5 4 5	BP t _e	0.8906 0 .8517	5	Flash Point °C		
B 481 °C		5	‡ _c	0.051		Fire Point		├
С			ΔHc kcal/m			M Spec. Ultra V.		1
A* 309 to B* 511 °(2.5 2 25 2789.6	7 5	ΔHf ΔFf			X-Ray Dif.		l
к – -	2 2.0%	1	Viscosity			Infrared Solubility in +	ļ	
t to	-		centistokes 7 °C			Solubility in + Acetone		1
t _k t _o		ļ	η · c			Carbon tet.		
A' to		\top				Benzene Ether		1
B' 'C	2)		B ^V to	 	 	n-Heptane Ethanol	}	
A'* 175 to	2,4240	7 5	AV C			Water		
B'* 309 °C		5	(BV)	1		Water in	ļ	₩
Ac to			(A ^V)					
Bc tc_*C	-		cp liq. °K					
Cryos. A	1	\top	c _p vap. °K					
consts. B			P .					
t _e °C	502.65	2	c _w vap.	<u> </u>				L
	ercooled liqui					fgrams/100 gra	ms solven	t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from det	da:			
SOURCE:		AP	I					
PURIFICA		AP						
LITERATU	RE REFERE	ENCES	3 JACS 77	(2019) 1955 Sc	hoer	er A. A. et al.		
l purity ap	plies to n _D ⁷⁰	°c.						

Name									No. 171	<u> </u>
Molecular Ref Molecular C31 H 64 Weight 436,822	NAME	n-He	entria	contane		_	ST	RUCTURAL	FORMUL	A
Molecular Ref Molecular C31 H 64 Weight 436,822										
Ref. Ref.				ar C ₃₁ H ₆₄	Molecular Weight 436.82	22		CH ₃ (CH ₂) ₂₉	сн ₃	
F.P. 100%			Ref.			_				Ref.
F. P. 100% Section S			2	dt/dP			f	to		
S. P. C 100 362.87 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 319.23 4 30 30 319.23 4 30 30 319.23 4 30 30 319.23 4 30 30 30 30 30 30 30	F.P. 100%		-		169 01	ا ۔ ا	g	*K		
100		450	١.				_h _	L	:	
1										
1					1.0818					
Density Section Sect	1	228.17	5		<u> </u>			to		_
1859.3 5 30 mm 48.1 38.4 5 31.6 35.			5		56.29	5	n			
Density g/ml 20°C 0.8011		1859.3	5	30 mm	48.1	5				
A	Density	0.0111	<i>‡</i>	11						
A 30		0.8111	# 2			1 1		-A-		
a	d4 30	0,8045	4		L	\perp	Sur	face tension		
Ref. Index								es/cm. 20°C		
The content of the		-0.0366	4	d' 180 to	67.46	5	8			
1.4503		1.4543	∮ 2		0.0607	5	Par			-
NR (Obs.) 145,93	_ 25	1.4523	' 2	II V mi/g				,20°C		
MR (Obs.) 145.93\frac{2}{145.358} 2 PV/RT 0.1 mm 1.0000 5		+		t _c °C ັ						
MR (Calc.) 145.358 5 1.0487 2 0.1 mm 1.0000 5 5 0.8905 5 5 0.8511 5				P _c mm				Sugd.	1243.2	5
Dielectric 2.11 5 BP 0.8905 5 Flash Point °C Fire Point C C C C C C C C C C C C C C C C C C C	MR (Calc.)	145.358	5		1 0000	_	Exp			
A 320 to	<u> </u>	 	\rightarrow	3 0 mm	1.00 0 0	5	Dis		98.≠	2
R 485 °C 2914, 17 5 C AHc kcal/m AH		+	_	t_						
A 320 to A 320 to B 518 °C 2829. 5	B 485 °C			t c						
A				∆Hc kcal/m						
Viscosity centistokes Solubility in + Acetone Carbon tet. Benzene Ether n-Heptane Ethanol Water Water in		2829.								
Carbon tet Benzene Ether Carbon tet Benzene Ether Carbon tet Benzene Ether Carbon tet]				\vdash
A' to B' °C	t _k to						Ac	etone		
B' °C	X I	 								
A **180 to 2.44635 5	B' °C				-					
B**320 °C 2782.1 5 (B')		2 44/2		B' to						
Acl to Bc tc °C Cc					•					
Cryos. A°	Ac to		1							
Cryos. A* consts. B* c p vap. *K te *C 512.12 5 c vap. # for undercooled liquid	Bc_tc_°C	-								
te °C 512.12 5 Cv vap.	Cryos. A*		1	13						
# for undercooled liquid		512 12	5	c _v vap.						
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API	# for unde			<u> </u>	L	L	+ ,	ams/100 gra	ms solven	Ļ_
SOURCE: API PURIFICATION: API				PI 3-Lit. 4-	Calc. from de	t. da				
								<u>-</u>		
LITERATURE REFERENCES:	PURIFICAT	ION:	AP	I						
	LITERATU	RE REFER	ENCE	S:						

							No. 172	;
NAME	n-De	otria	contane			STRUCTURAL	FORMU LA	L
						CH (CH) C	u	
Mole	Ref. Mo	lecul	ar -	Molecular		СН ₃ (СН ₂) ₃₀ С	⁻¹ 3	
% Pur.	Fo	rmul	ar C ₃₂ H ₆₆	Weight 450.8	48			
		Ref.			Ref.			Ref.
F.P. C F.P. 100%	69.7	2	dt/dP *C/mm			f to		
B.P. °C		-	0.1 mm	171.00	5	g °K		
760 mm	467.	2	BP	0.073 0.0347	5 5	$\frac{1}{f} + \frac{1}{f}$		
100 30	370.97 326.88	4	t _e 30 mm	1.1119	5	g' °K		
10	292.69 234.83	4 5	ΔHm cal/g			h'		
1 0. 1	190.32	5	ΔHv cal/g			m to		
Press. mm			0.1 mm 30 mm	55.37 47.1	5	n ¦ °K		
t e Doneity	1884.1	5	BP	38.24	5	m' i to	 	_
Density g/ml 20°C	0.8124	2	t _e (d, e)	35.0 34.67	5 5	n' 'K		
dt 25 4 30	0.8124 [‡] 0.8091 [‡] 0.8058	2	ΔHv/T	19.71	5	01		
1 30	0.8256	4	d 324 to		5	Surface tension dynes/cm, 20°C	28.50	5
b	-0.0366	4	a 186 to		5	30 ×	27.58	5
Ref. Index	1 45507	2	e' 324 °C		5	40	26.69	5
D 25	1.4530	2	d _c g/ml			Parachor [P] 20°C		
30	1.4516	4	vc ml/g tc °C			30 40		
"C"	0.7412	4_	P _c mm			II .	1282.2	5
MR (Obs.) MR (Calc.	1 140 076	2 5	PV/RT		_	Exp. L.1.%/wt.		
(nD-d/2)	1.0488 ₹	2	0.1 mm 30 mm	1.000 0 1.0000	5	u. Dispersion	98.≠	2
Dielectric	2,12	5	BP t _e	0.8910 0.8500	5	Flash Point °C		 - -
B 1530 °C		5	tc	0.0500		Fire Point		<u> </u>
С		<u> </u>	ΔHc kcal/m ΔHf			M Spec. Ultra V.		ł
A* 324 to B* 530 °C		5	ΔFf		İ	X-Ray Dif. Infrared		
к — —	-	-	Viscosity	1		Solubility in +	 	┢
t _k			rentistokes °C		l	Acetone		
1 × 1			! '		1	Carbon tet. Benzene		
A' to					L	Ether n-Heptane		}
C'			B ^V to			Ethanol		
A'* 186 to B'* 324 °C		5	$\begin{bmatrix} A^{\mathbf{v}} \\ (B^{\mathbf{v}})_1 \end{bmatrix} - {}^{\mathbf{c}}$	-		Water Water in		
Act to	+	 _	(A ^V)					
Bc tc C	<u>-</u>		c _p liq. °K	 				
Cryos. A		\vdash	c _p vap. °K					
consts. B°	<u> </u>	<u> </u>	11			Ĭ		
te °C	522.46	5	c _v vap.	<u> </u>	L_	L	L	<u> </u>
	rcooled liquid CES: 1-Dow		DV 2 V.			grams/100 gram		<u>t</u>
SOURCE:	759: 1-DOM	2-Al		Calc, from de	t, da	ta 5-Calc. by for	mula	
PURIFICA:	TION:	AP						
	RE REFERE							
L								

No. 173 n-Tritriacontane NAME STRUCTURAL FORMULA Molecular C33H68 $CH_{3}(CH_{2})_{31}CH_{3}$ Mole Molecular Weight 464.874 % Pur. Ref. Ref Ref. 71.4 2 F. P. °C dt/dP f to F.P. 100% ۰ĸ °C/mm ١ g 172.76 0.1 mm 5 B.P. °C h 0.074 5 BP 475. 760 mm 2 ^te 0.03462 f† to 378.17 4 100 g' ٠ĸ 5 1.1221 30 mm 30 333.70 4 10 299.19 h' ∆Hm cal/g 5 240.77 1 m to ∆Hv cal/g 195.81 5 0.1 n 54.42 5 5 0.1 mm 30 mm Press. mm 0 46.5 t_e 1900.8 5 BP 37.1 5 m Density to 5 te te (d, e) 34.3 0.8136 'n g/ml 20°C 2 5 33.99 25 0.8103 2 dt4 ΔHv/T_e 19.69 5 30 0.8070 4 330 to Surface tension 5 68.70 0.8268 28.59 dynes/cm. 20°C 5 0.0665 5 529 °C -0.0366 ь 4 30 27.67 5 to a¬ 196 65,66 5 5 40 26.78 5 Ref. Index e' 330 °C 0.0574 20°C 1.4557 ⁿD Parachor [P] 1.4536 d_c g/ml 25 2 20°C vc ml/g 30 1.4517 4 30 t_c "C" 0.7412 40 4 P_c mm Sugd 1321.2 5 MR (Obs.) 155.21# PV/RT Exp. L.1.%/wt. MR (Calc.) (nD-d/2) 154.594 0.1 mm 30 mm 1.0000 5 5 1.0488 2 98.≠ 1.0000 2 Dispersion Dielectric 2.12 5 BP 0.8895 Flash Point °C 0.8491 5 A 330 to 7.68082 5 Fire Point B 1 539 °C 2997.60 C M. Spec. ΔHc kcal/m Ultra V. ΔHf A*| 330 to 2.59301 5 X-Ray Dif. ΔFf B*| 539 °C 2912.0 Infrared K Viscosity Solubility in c centistokes Acetone to ۰, Carbon tet. ٠c t i Benzene A١ to Ether В' °C n-Heptane B^V | C' to Ethanol °C 2.48943 A'* 185 to Water 5 B'# 330 °C Water in 2862.0 (B^V) Acl (AV) to Bc °C c_p liq. °K Сc Cryos. A* cp vap. ۰ĸ consts. B° c, vap. te °C 531.48 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 5-Calc. by formula 3-Lit. 4-Calc. from det. data SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

									No. 174	
NAME		n-	Tetr	atria	contane		\neg	STRUCTURAL	FORMULA	١.
Mole % Pur,		Ref.	Mo Fo	lecul rmul	ar C 34 ^H 70	Molecular Weight 478.9	00	сн ₃ (сн ₂) ₃₂	СН ₃	
				Ref.			Ref			Ref.
F. P. *C	T	73.1		2	dt/dP	1		f to		
F.P. 1007	5				*C/mm		1 1	f to g °K		
B.P. °C	+				0.1 mm	174.29	5	h		
760 mm	-1	482.		2	BP	0.074 0.03457	5		ļ	
100	-	384.4		4	t _e	1,1310	5	g' to		
30 10	ı	339.6 304.8		4 5	30 mm	1.1310	-	h'	Ì	
ì		245.9		5	AHm cal/g	<u> </u>		<u> </u>	 	
0.1		200.6	2	5	∆Hv cal/g	52.44	ا ـ ا	m to		
Press. mn	n	1918.5		5	0.1 mm 30 mm BP	53.44 45.7 36.4	5 5 5	0		<u> </u>
Density	T		4		te (d.e)	33.7	5	m' to		i i
g/ml 20°0	۱:	0.8	148‡	2	t (d, e)	33.32	5	n' 'K		
dt 25 4 30		0.8	115 [#]	2 4	AHV/Te	19.69	5	<u> </u>		
	+				d 336 to	67.89	5	Surface tension		
a b		0.8 -0.0		4	e 537 °C		5	dynes/cm. 20°C	28.68 27.77	5
Ref. Index	.+		3-0	÷	d' 190 to		5	40	26.87	5
n _D 20°0		1.4	563‡	2		0.0557	5	Parachor [P]		
25	-	1.4	542	2	d g/ml v ml/g		1	20°C		
30	\perp	1.4	524	4	v ^c ml/g t _c °C			30		
"C"	\perp	0.74	410	4	Pcmm		ĺ	40 Sugd	1360.2	5
MR (Obs.		159.8		2	PV/RT	 	-	Exp. L.1.%/wt.	1300.2	+
MR (Calc. $(nD-d/2)$	7	159.2	12 488 [≠]	5	0.1 mm	1,0000	5	u.	,	
	+			2	30 mm	1.0000	5	Dispersion	98. [‡]	2
Dielectric	_	2.17		5	BP	0.8895	5	Flash Point °C		
A 336 t B 547 C		7.69 3031.78	916 4 8	5	t _e t _c AHc kcal/m	0.8486	3	Fire Point M Spec.	-	
A* 336 to B* 547 °C		2.61 2945.7	1320	5	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared		
к ——	-				Viscosity			Solubility in +	 	+
t _k					centistokes 7°C			Acetone		
t i to		·						Carbon tet. Benzene		
B' °(Ether n-Heptane		
C'	4				B ^V to	.		Ethanol	1	1
A'*190 to B'*336 *		2.50 2895.2	0973	5 5	(BV)	-		Water Water in		_
Ac to					(A ^V)					
	7				c _p liq. °K	Í			1	
Cryos, Acconsts, B					c _p vap. °K		'			
t _e °C	\perp	539.49		5	c _v vap.	L	<u>L_</u>	L	<u> </u>	<u></u>
# for und								grams/100 gran		<u>t</u>
REFEREN	UE	ວ: 1-D	ow	2-AI		Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:				AP						
PURIFICA				AP						
	·KI	SKEF	ER EI	NCES	o:					
				_						

								No. 175	
NAME	n-Penta	triac	ontane			ST	RUCTURAL		
Mole % Pur.	Ref. Mo	lecul:	C ₃₅ H ₇₂	Molecular Weight 492.9	26		CH ₃ (CH ₂) ₃₃	CH ₃	
· · · · · · · · · · · · · · · · · · ·		Ref		T	Ref				Ref.
F.P. *C	74.7	2	dt/dP			f	to		
F.P. 100%			°C/mm	Ì		g	l °K		
B.P. °C	1		0.1 mm BP	176.04 0.0754	5	h	l		
760 mm 100	490. 391.68	2 4	t _e	0.03453		f¹	to		
30	346.47	4	30 mm	1.141	5	g'	°K		1
10 1	311.36 251.90	5	∆Hm cal/g			h'			
0.1	206.10	5	ΔHv cal/g			m n	to o		
Press. mm	1938.5	5	0.1 mm 30 mm	52.60 45.0	5	0			
t _e	1730.3	Ľ	BP	35.8	5	m'	to		-
Density g/ml 20°C	0.8157	2	te te (d, e)	33.1 32.73	5 5	n'	°K		
dt 25	0.8124	2		19.67	5	o '	i		
	0.8093	4	d 343 to		5	Sur	face tension		
a b	0.8289	4 4	e 1546 °C	0.0641	5	dyn	es/cm. 20°C	28.74 27.85	5
Ref. Index	<u> </u>	╁	d' 200 to		5	•	30 40	26.95	5
n _D 20°C	1.4568	2	d _c g/ml	0.0541	5	Par	achor [P]		
25 30	1.4568 1.4548 1.4529	2 4	v_mi/g	1			20°C		
"C"	0.7408	4	t _c °C				30 40		
MR (Obs.)	164.49#	2	P _c mm				Sugd.	1399.2	5
MR (Calc.)	163 83	5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	1.0489	2	0.1 mm 30 mm	1.0000	5	Dis	u. persion	98. [≠]	2
Dielectric	2,12	5	BP	0.8892	5		sh Point °C		
A 343 to B 505 °C	7.70371 3070.74	5	te t	0.8478	5	Fir	e Point		
c	147.	5	AHc kcal/m				Spec.		
A* 343 to	2,63438	5	ΔHf ΔFf				ra V. Ray Dif.		
B* 546 °C K	2984.4	5				Infr	ared		
c			Viscosity centistokes				ubility in +		
t _k to	ł		η °C				etone rbon tet.		
A' to		 				Be	nzene		
B' C					\perp		he <i>r</i> Hep tane		
<u>c' </u>			B ^V to C			- Et	hanol		
A'* 200 to B'* 343 °C	2.52835	5	⊢	-			ater ater in		
Acl to	2932.1	5	(B ^V) (A ^V)						
Bc tc °C				+	\vdash				
Cc			c _p liq. °K						
Cryos. A° consts. B°			c _p vap. °K						
te °C	548, 62	5	c _v vap.			<u> </u>			<u></u>
	cooled liquid						ams/100 gra		t
	ES: 1-Dow			-Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		API							
PURIFICAT		API							
LILERATUR	RE REFERE	NCES	:						

									No. 176	,
NAME	n-Hex	atriac	ontan	e			STRUC	TURAL :		
Mole 99	.0 Ref		lecul		Molecular Weight 506.	952	CI	н ₃ (СН ₂) ₃₄	сн3	
/ Pul.		1 - F C	Ref		weight 500.	Ref				Ref.
F.P. ℃	76.	2.	2	dt/dP			f			-
F.P. 100%				*C/mm			g	to °K		
B.P. °C				0.1 mm BP	177.75 0.0759	5 5	h			l
760 mm 100	497. 397.	98	2 4	te	0.0355	5	Ţ, T	to		
30	352.	43	4	30 mm	1.155	5	g'	°K	ļ	
10 1	317.0 257.		5	∆Hm cal/g			h'		ļ	<u> </u>
0.1	210.		5	ΔHv cal/g			m	to °K		
Press.mn	1938.		5	0.1 mm 30 mm BP	43. 65 35, 2	5 5 5	o			ļ
Density g/ml 20°C		8169 [‡] ,	2	te (d, e)	31.23	5	n'	to °K	İ	
_a t 25	0.	8136₹	2	ΔHv/T _e	31.7 19.09	5	0'			
		8103	4	d 349 t		5		tension		
a b Ref. Index	-0.	3301 0 ₃ 66	4 4	e 554 °c	0.0584	5	dynes/	m. 20°C 30 40	28.85 27.93 27.03	5 5 5
n _D 20°C	:1.	4573‡ 4	2		<u> </u>	+	Parach			
25 30	1.1	4554 [≠] 4534	2 4	d g/ml vc ml/g tc °C				20°C 30	ļ	
"C" MR (Obs.)	0.	7406	4	t _c °C P _c mm				40	1438.2	5
MR (Calc.	1 169	449	=	PV/RT			Exp. L	.1.%/wt.		
(nD-d/2)		1489 [‡]		0.1 mm 30 mm	1.0000	5	Disper	u. ion	98. [‡]	2
Dielectric	2.		5	BP t _e	0.8852 0.8397	5	Flash F			1
A 349 to B 1560 °C		713 44 00	5 5 5	t _c AHc kcal/m			M Spec	•		
A* 349 to B* 560 °C		65726 4	5 5	ΔHf			Ultra V X-Ray Infrare	Dif.		
K				Viscosity centistokes 7°C			Solubili Acetor	ne		
A' to	+			•			Carbo: Benze: Ether			
B' °				B ^V to			n-Hep Ethano Water			
A'+ to B'+ °C	<u>: </u>			(BV)			Water	in		-
Ac to Bc t _c °C Cc				c _p liq. °k	:	\vdash				
Cryos. A° consts. B°				c _p vap. °F	:					
t _e °C	556.		5	c _w vap.						L
for unde								/100 grai		t
REFERENC	CES: 1-	Dow	2-AF		Calc. from de	t. da	ta 5-Cal	c. by for	mula	
SOURCE:	rio»:		AP							
PURIFICA:		PPP	AP							
- LARIU	NE REI	· EAĒ,	HCE3							

No. 177 n-Heptatriacontane NAME STRUCTURAL FORMULA CH3(CH2)35CH3 Molecular C37H76 Ref. Mole Molecular Weight 520.978 % Pur. Ref Ref. F.P. °C F.P. 100% 77.7 2 dt/dP f to °C/mm ۰ĸ g 0.1 mm 179.09 B. P. °C h BP 0.0764 5 504. 760 mm 2 0.3449 5 ſ١ to °K 404.28 100 4 g' 1.1590 5 30 mm 30 358.39 4 10 322.74 5 h' AHm cal/g 262.32 1 215.74 5 AHv cal/g 0.1 °К 50.91 0.1 mm Press. mm o 30 mm 5 ^te 43.6 1973.0 5 ΒP 34.6 5 mi Density t_e (d, e) 32.0 5 n' ۰ĸ g/ml 20°C 0.8179 2 31.54 0.8146 o' d_4^t 25 19.61 5 AHv/T 30 0.8113 4 a 355 Surface tension 65.75 to 5 0.8311 dynes/cm. 20°C 28.93 1 562 °C 0.0618 ь -0.0366 4 30 d' 210 61.96 28,00 to 40 27.10 Ref. Index e' 355 ٠c 5 5 20°C 1.4578 n_D [P] Parachor 1.4559 d_c g/ml 25 2 20°C vc ml/g t_ °C 30 1.4539 4 30 t_c "C" 40 0.7405 4 1477.2 5 P_c mm Sugd. MR (Obs.) 173.77≠ 2 PV/RT Exp. L.1.%/wt. 173.066 MR (Calc.) 1.0489 0.1 mm 1.0000 2 5 (nD-d/2) 98. [‡] 30 mm 1.0000 Dispersion 2 2.13 5 Dielectric BP 0.8888 5 Flash Point °C A 355 to t t 0.8465 7.72381 Fire Point B 1570 °C 3138.26 5 M. Spec. c 144. 5 AHc kcal/m Ultra V. ΔHf 2.67222 A*| 355 to X-Ray Dif. B*| 570 °C ΔFf 3051.3 Infrared Viscosity Solubility in centistokes to °C Acetone Carbon tet. °C ŧς Benzene ٨' to Ether B' °C n-Heptane Bv Av C to Ethanol 1 °C A**210 to 2.56511 2997.5 Water B'*355 °C Water in (B^V) Acl (AV) to Bc ۰c c_p liq. ۰ĸ Cc Cryos. A° c_p vap. consts. B° c, vap. te °C 564.63 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME											
	AME n-Octatriacontane								UCTURAL 1	FORMULA	
									CH ₃ (CH ₂) ₃₆	СН	
Mole % Pur.		Re	f. Mo	lecul rmul	ar C ₃₈ H ₇₈	Molecular Weight 535.0	004		2,36	503	
				Ref.			Ref				Ref.
F.P. *C		79.	0	2	dt/dP	T		ſ	to		
F.P. 100	•				*C/mm 0.1 mm	180,62	5	g	°K		
B.P. °C 760 mm		511.		2	BP	0.076	5	<u>h</u> _	+		
100	ı	410. 364.		4	t _e	0.03444	5	f' g'	to to		
30 10	1	328.	43	4 5	30 mm	1.1017		h'			ŀ
1 0. 1		267. 220.		5	ΔHr cal/g	 	\vdash	m	l to		m
Press. mr	n	1990.	,	5	0.1 mm 30 mm	50.12 42.9	5	n o	•к		ļ
t e Density		1770.		3	BP	34.1	5	m'	l to		\vdash
g/ml 20°	5	0.	8188‡	2	te (d, e)	31.4 31.0	5	n'	•K		
dt 25 4 30		0.	81887 8155 8122	2	ΔHv/Te	19.59	5	٥'	<u>.</u>		
a	-		8320	4	d 361 to		5		ace tension s/cm. 20°C	28.99	5
ь		-0.	0366	4	d' 570 to		5	3,	30	28.07	5
Ref. Inde:		1.	4583‡	2	e' 361 °C	0.0502	5	Par	40 achor [P]	27.17	5
- 25		1.	4564 ^r	2	d g/ml				20°C		
30	\dashv		4543 7404	4	tc °C	1			30 40		
MR (Obs.	Н	178.		2	P _c mm				Sugd.	1516.2	5
MR (Calc		177	691	5	PV/RT 0.1 mm	1 0000		Exp	L.1.%/wt.		
(nD-d/2)	4		0489	2	30 mm	1.00 0 0 1.0000	5	Disp	u. ersion	98.≠	2
Dielectric		7.	73405	5	BP t _e	0.8889 0.8458	5		h Point °C		
B 521°	디	3172.		5	t _c	<u> </u>		M S			\vdash
A* 361 t	,	2.	69112	5	ΔHf			Ultr			
B* ∟580°	듸	3085.		5	ΔFf Viscosity	 	\vdash		red		L
c	_				centistokes				bility in + etone		
t _x t	c				η ∘c			Ca	rbon tet.		
A' t	•							Eth	nzene ner		
B' ∟ °	의				B ^V to	 	\vdash		leptane anol		
A'* 216 t			8304	5	A ^V °C			Wa	ter		
B* 361 •		3030.2	:	5	(B ^V)	_		Wa	ter in		+-
	င္ပါ				(A ^V)	_					
	듸			L	c _p liq. •K						
Cryos, A consts. B					c _p vap. °K						
t _e °C		572.6	3	5	c _v vap.						
# for und									ams/100 grai		t
REFEREN	CI	ES: 1-	Dow	2-AI		Calc. from de	t. da	ta 5-	Calc. by for	mula	
SOURCE: PURIFICA	T	ON		API	• • • • • • • • • • • • • • • • • • • •						
LITERATI	_		FERE								

<u></u>					- т			No. 17	9
NAME	n-Nona	riaco	ntane			ST	RUCTURAL	FORMUL	A
			т		\dashv		CH ₃ (CH ₂) ₃₇	сң	
Mole % Pur.		olecul ormul		Molecular Veight 549.	030		J 2 31	3	
		Ref			Ref.				Ref.
F.P. *C	80.3	2	dt/dP			f	to		
F.P. 100	6		°C/mm		ا ۔ ا	g	*K		
B.P. *C]		0.1 mm BP	182.18 0.076	5	_h _			
760 mm 100	518. 416.88	2	t _e	0.0345	5	f'	to		İ
30	370.30	4	30 mm	1.1755	5	g'	*K		
10 1	334.10 272.69	5	∆Hm cal/g			h' j			-
0.1	225.33	5	ΔHv cal/g 0.1 mm	40.37	ا ۔ ا	m	to •K		
Press. mr	n 2007.1	5	30 mm	49.37 42.3	5	l ° ¦			
t _e Density	 	. -	BP	33.6	5	m'	to		
g/ml 20°	0.8197	2	t _e (d, e)	30.9 30.43	5	n'	°K		
dt 25 4 30	0.8164 0.8131	2	AHv/Te	19.54	5	لـنـــا			
a 30	0.8329	4	d 366 to	64.11	5		face tension es/cm. 20°C	29,07	5
Ъ	-0.0366	4	d' 220 to	0.0589 60.35	5	8,44	30	28.14	5
Ref. Index	1 4500	4	e' 366 °C	0.0488	5	<u> </u>	40	27.24	5
ⁿ D 20°0	1.4588 1.4568	2 2	d _c g/ml			Par	achor [P] 20°C		
30	1.4549	4	v _c ml/g t _c °C				30		
"C"	0.7404	4	P _c mm			ļ	40 Sugd.	1555.2	5
MR (Obs. MR (Calc.		2	PV/RT		1	Exp	. L.1.%/wt.		
(nD-d/2)	182.302 1.0490	5 2	0.1 mm	1.0000	5	_	u.	98. [≠]	
Dielectric		5	30 mm BP	1.0000 0.8881	5		persion	70.	2
A 366 to			t _e	0.8451	5		sh Point °C e Point		
B 1523 °C	3207.87	5	tc AHc kcal/m		-	M.	Spec.		T
A* 366 to	2,7103	, 5	ΔHf				a V. Lay Dif.		
B* 587 °C	3120.5	5	ΔFf				ared		
K — —			Viscosity centistokes		ł		bility in +		
t _k			η °c				etone rbon tet.		
, x	_					Ве	nzene		
A' to B' *C					ļ		her Hep tane		
C'			B ^V to C			Et	hanol		
A!* 220 to B!* 366 °C			$\frac{A^{\vee}}{(B^{\vee}) } - {^{\circ}C}$				ter ter in		
Acl to		5	(A ^V)						
Bc t *C					 				
			р -						
Cryos, A' consts, B'			c _p vap. *K						
t _e °C	580,62	5	c _v vap.						
	ercooled liqu						ams/100 gra		t
	CES: 1-Dow		PI 3-Lit. 4-0	Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		AP							
PURIFICA		AP							
LITERAT	JRE REFER	ENCE	5:						

							No. 180	
NAME	n-'	Tetracor	ntane			STRUCTURAL	FORMULA	
Mole % Pur.	Ref.	Molecul Formul	ar C ₄₀ H ₈₂	Molecular Weight 563.0	56	CH ₃ (CH ₂) ₃₈	СН ₃	
		Ref	T		Ref			Ref.
F.P. *C	81.5	2	dt/dP		-	 		
F.P. 100%			*C/mm		1	f to		ĺ
B. P. °C	†		0.1 mm	183.66	5	h		İ
760 mm	525.	2	BP t _e	0.077 0.03460	5	f' + to		
100 30	423.19 376.27	4 4	30 mm	1,1871	5	g' °K		
10	339.79	5	ΔHm cal/g	<u> </u>	1	h'		1
1	277.91	5		 	├-	m l to		
0.1 Press. mm	230.17	- 3	ΔHv cal/g 0.1 mm	48,68	5	n °K		
t e	2025.3	5	30 mm	41.7	5	0		
e Density	<u> </u>		BP	33.1 30.5	5	m' to		
g/ml 20°C	0.82		te (d, e)	29.78	5	n' *K		
d ^t 25 4 30	0.81		AHv/Te	19.41	5	0' '		
	0.81		d 372 to		5	Surface tension	26	
a b	0.83 -0.03		e 585 °C		5	dynes/cm. 20°C	29.13 28.20	5
Ref. Index			d' 225 to		5	40	27.30	5
n _D 20°C	1.45	93 🗜 2		0.0478		Parachor [P]		
45	1.45	73" 2	d g/ml vc ml/g		1	20°C		
30	1.45		tc °C			30 40	l	
"C"	0.74		P _c mm		1	Sugd.	1594.2	5
MR (Obs.) MR (Calc.)	187.69 ³ 186.92	2 5	PV/RT		 	Exp. L.1.%/wt.		
(nD-d/2)	1.04	90 [#] 2	0.1 mm	1.0000	5	u.	98.≠	
Dielectric	2.13	5	30 mm BP	1.00 0 0 0.8881	5	Dispersion	98.	2
A 372 to	7.75	334 5	t _e	0.8446	5	Flash Point °C Fire Point		
B 1526 °C	3240.23	5	t _c		$oxed{oxed}$	M Spec.	<u> </u>	├
C	1 22	·	ΔHc kcal/m ΔHf			Ultra V.		
A* 372 to B* 595 °C	3152.5	652 5	ΔFf		l	X-Ray Dif.		
K P'S C	3132.3	1	Viscosity	<u> </u>	†	Infrared	ļ	
°			centistokes			Solubility in + Acetone		ļ
tk to c		1 1	7 ℃			Carbon tet.		1
A' to	 				1	Benzene Ether		
B'°C			<u> </u>		!	n-Heptane		
C'			B ^V to			Ethanol		
A'* 225 to B'* 372 °C	2.616		L'	4		Water Water in		1
	3095.5	5	(B ^V)		1			
Ac to			(A ^V)	 	 	4		
Cc — -			c _p liq. °K		1	Į.		1
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	588.65	5	c _w vap.					<u> </u>
# for unde	rcooled lie	quid				grams/100 grai	ms solven	t
REFERENC	ES: 1-Do			Calc. from det	t. da	ta 5-Calc. by for	mula	
SOURCE:		AP	[
PURIFICAT	ION:	AP						
LITERATUI	RE REFE	RENCES	:					

No. 1 NAME Methyl fluoride STRUCTURAL FORMULA Fluoromethane CH₃F Mole Ref. Molecular Molecular % Pur Formula CH₃F Weight 34.034 Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm ۰ĸ 1 g 25°C B. P. °C h 0.0238 BP 2 760 mm -78.35 0.0326 5 ſ١ t_e 100 -108.68 2 to g' <u>• к</u> 30 -122.19 2 30 mm 0.3392 5 10 -132.50 2 h' ∆Hm cal/g 1 -149.60 5 m to AHv cal/g Pressure n °K 25°C mm 25°C o 30 mm 130.76 511.5 5 t_e BP 118.46 5 m' to Density te te (d, e) 120.56 5 5 n' ۰ĸ g/ml 20°C 120.37 ٥' 25 d t 4 ΔHv/T_e 21.83 5 30 -125 Surface tension d 96.49 dynes/cm. 20°C 5 <u>-65</u> °C 0.2804 ь 30 á⊓Ì to 40 Ref. Index e¹ °C 20°C [P] ⁿD Parachor d_c g/ml 0.275 5 25 20°C vc ml/g tc °C 3.63 5 30 30 3 44.6 40 "C" P_c mm 91.8 5 44080. 3 Sugd MR (Obs.) PV/RT Exp. L.1.%/wt. MR (Calc.) 6.668 5 25°C (nD-d/2)1.0000 30 mm 5 Dispersion Dielectric BP 0.9700 Flash Point C 0.9782 5 A -122 to 7.09761 2 Fire Point 0.275B 1_-20°C 740.22 M. Spec. С 253.89 2 AHc kcal/m Ultra V. A* - 122 to ΔHf 1.26539 5 X-Ray Dif. AFf B*|_-65°C 688.81 Infra red Viscosity Solubility in centistokes c Acetone to Carbon tet. 00 tx °C Benzene œ A' to Ether an B' <u>°C</u> n-Heptane 00 B_v | C to Ethanol °C Water A1* to Water in (BV) B'* °C to Acl -20to 7.81764 (A^V)| °C Bc tc °C 1113.00 cp liq. ۰ĸ Сс 306.57 4 Cryos, A° consts, B° °К cp vap. c_v vap. t_e °C -85.23 $T_R = 0.75T$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: Literature PURIFICATION: Literature LITERATURE REFERENCES: 3 Chem. Rev. 52, No. 1, 117 (1953), Kolbe and Lynn; 3' Physia 14, 104 (1948), Nichels and Wassenaar.

							No. 2	
NAME	Perfluor	0-n-l	nexane			STRUCTURAL	FORMULA	
						an (an)	C.P.	
Mole	Ref. Mo	lecul		Molecular		CF ₃ (CF ₂) ₄	CF ₃	
% Pur.		rmul		Weight 338.0	60			
		Ref.			Ref.			Ref.
F.P. °C	-87.1	3	dt/dP			f to		
F. P. 1009 B. P. °C	-	├	*C/mm 25*C	0.1045	5	g <u>*K</u>		
760 mm	57.11	3	BP	0.03827 0.03573	5	h f' to		-
100 30	8.27 -13.52	5	t _e 30 mm	0.5473	5	g' to		l
10	-30.1	5	ΔHm cal/g	0.5415	-	h'		
Pressure	-57.8	5	ΔHv cal/g	1		m to		
mm 25°C	218.9	5	25°C	22.50	5	n <u>*K</u> _		
t _e	890.9	5	30 mm BP	24.14 21.00	5	<u> </u>		├
Density g/ml 20°	1,6995	3	te (d, e)	19.62 20.8	5	m' to		
at 25	1.6851	4	ΔHv/Te	19.80	5	0'		
	1.6695	3	d -14 to	 	5	Surface tension		
a b	1.7612 -0.0029	4		0.0444	5	dynes/cm. 20°C	11.92 10.93	3
Ref. Inde			e' '°C			40	10.10	3
n _D 22°0	1.2515	3	d g/ml vc ml/g			Parachor [P] 20°C	369.4	4
30			v _c ml/g t _c °C	Ì		30	367.9	4
"C"		<u> </u>	P _c mm	ł		40 Sugd.	365.6 364.8#	4 5
MR (Obs. MR (Calc.		4 5	PV/RT			Exp. L.1.%/wt.		Ť
(nD-d/2)	0.4146	4	25°C 30 mm	0.9839 1.0000	5	u. Dispersion		
Dielectric			BP	0.9600	5	Flash Point °C		
A -14 t		4	t _e t _c	0.9559	5	Fire Point		
B L 66 °	C 1205.37 227.0	4	ΔHc kcal/m			M Spec.		
A* -14 to		5	ΔHf ΔFf			Ultra V. X-Ray Dif.		
B* L 65 °	<u>C</u> 1130.9	5	Viscosity	· · · · · · · · · · · · · · · · · · ·		Infrared		L
c	_	1	centistokes	0.41/0		Solubility in + Acetone		
t _x t			7 20 ℃ 40	0.4160 0.3270	3	Carbon tet.		i
A' t		T	50	0.2916	3	Benzene Ether		
B' 3	드	1	B 10 to	487.4	4	n-Heptane Ethanol		
A'* t	<u>- </u>		AV 60 °C	3.95674	4	Water		1
B!* °	- 	<u> </u>	(B ^V) to			Water in		-
Ac t			(A ^V) °C					
Cc - c-		<u> </u>	c _p liq. °K					
Cryos, A consts, B			c _p vap. °K					
t _e °C	61.81	5	c _v vap.					
						grams/100 grai	ns solven	t
REFEREN	CES: 1-Dow	2-A1	PI 3-Lit. 4-0	Calc, from det	. da			
SOURCE:		Li	t.					
PURIFICA		Li						
LITERAT	JRE REFERE	NCES	3: 3 JACS <u>74,</u>	3771 (1952) St	iles	and Cady		
# atomic:	ref. index for	Fluor	ine in polyfluor	ro compounds	= 1.2	22		
# atomic	P] for Fluorin	e in]	polyfluoro comp	oounds = 24.				

No. 3 Perfluoro-2-methylpentane STRUCTURAL FORMULA NAME CF2CF (CF2)2CF3 Molecular C6F14 Mole Ref. Molecular Weight 338.060 % Pur Formula Ref. Ref. Ref. F. P. °C dt/dP f to F.P. 100% °C/mm °K g 25°C 0.1065 5 B.P. °C h ΒP 0.03878 5 760 mm 57.73 3 0.03613 5 f١ to 100 8.30 5 °<u>K</u> g' 30 -13.69 5 30 mm 0.5521 10 -30.5 5 h' AHm cal/g -58.3 5 m to AHv cal/g Pressure °K n 25°C 22.43 5 216.8 mm 25°C 5 5 o 30 mm 23.90 5 831. t_e BP 5 21.00 m' to Density te te (d, e) 19.36 5 n' g/ml 20°C °K 1.7326 3 5 20.89 ٥' ď4 25 1.7169 ΔHv/T_e 19.62 5 30 1.7011 4 Surface tension -14 to 23.34 5 1.7946 3 dynes/cm. 20°C 12.12 60 ℃ 0.0406 ь -0.0031 4 11.41 30 3 ăח 40 10.49 3 Ref. Index e' °C 22°C 1.2564 3 [P] nD Parachor d_c g/ml 25 20°C 364.1 vc ml/g 30 30 365.2 4 t_c 364.2 40 4 "C" 0.1688 4 P_c mm Sugd. 364.8 MR (Obs.) 31.544 31.588 5 PV/RT Exp. L. l. %/wt. MR (Calc.) 25°C 0.9933 5 (nD-d/2) 0.3901 5 1.0000 30 mm 5 Dispersion Dielectric BP 0.9600 Flash Point °C 0.8954 A -14 to 7.08320 Fire Point B 1 73 °C 1198.63 M. Spec. Ultra V. C 227.5 4 ∆Hc kcal/m ΔHf A* -14 to 2.02668 5 X-Ray Dif. ΔFf B*| 65 °C 1118.1 Infrared ĸ Viscosity Solubility in centistokes Acetone 0.4538 20 °C 3 Carbon tet. t<u>x</u> °C 30 0.3979 3 Benzene 50 0.3130 3 to Ether B١ °C n-Heptane C' 10 to Ethanol **v** 1 60 °C 3,91880 A'* Water to Water in B'* °C (B^V) to Acl (A^V)| to °C Βc ۰c ۰ĸ c_p liq. Cc' Cryos. A° cp vap. °K consts. B° c, vap. te °C 5 60.39 $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula Lit. SOURCE: **PURIFICATION:** LITERATURE REFERENCES: 3 JACS 74, 3771 (1952) Stiles and Cady

Atomic ref. index for Fluorine in polyfluoro compounds = 1.22

-									No. 4	
NAME	Methyl ch	loride	•			_	STRUCTURA	L F	ORMUL	A.
	Chlorome	thane								
Mole % Pur.		olecul ormul			cular	,	Сн	3Cl		
<u> </u>		Ref			int 30.1	Ref				Ref.
F.P. °C	-97.720	31	dt/dP				f			
F.P. 100%		十	*C/mm				g	°K		İ
B. P. °C		1.	25°C BP	1	0.0081 0.0305	5 5	h ;	1		
760 mm 100	-24.22 -62.91	2 2	te	-	0.0340	5	<u></u>	to		
30	-80.03	2	30 mm		0.4292	5	g'	°K		ł
10 1	-93.05 -114.58	2 5	AHm cal/g	1			h'			
Pressure	1	+-	ΔHv cal/g				, ,	to °K		
mm 25°C	4309.7	5	25°C 30 mm	- -	89.54 114.01	5 5	n	~		
t _e	662.9	5	BP BP		101.30	5	 !	-		├
Density g/ml-24°C	. 0 007	32	t _e (d. a)		102.04	5		to K		İ
t -20	0.997	32	'e (d, e)		102.02	1 1	o'			1
4 -10	0.973	32	ΔHv/T _e		20.55	5	Surface tension	on l		t -
a b	0.9576	4		C	95.79 0.2276	5 5	dynes/cm20	°C		1
Ref. Index	-0.02184	4		: <u>-</u>			3 -10 40			
n _D 20°C		İ	<u> </u>	<u>'C</u>	0.252	H	Parachor [P]			\vdash
D 25 30	1		d _c g/ml v _c ml/g		0.353 2.833	3' 3'	20	°C		l
"C"	+	+-	tc °C °		143.1	3	30 40			
		+	P _c mm	50	084.	3	lli e	gd.	110.4	5
MR (Obs.) MR (Calc.		5	PV/RT			_	Exp. L.1.%/v	vt.		
(nD-d/2)			25°C 30 mm		0.9073 1.0000	5 5	u. Dispersion]
Dielectric			BP		0.9660	5	Flash Point °C	-		
A -80 to			te t _c		0.9692 0.2752	5 4	Fire Point			1
B [_40_°C	902.45 243.60	2 2	ΔHc kcal/m			\vdash	M Spec.			
A* -90 to		+	ΔHf				Ultra V. X-Ray Dif.			
B* _ 40 °C		5	ΔFf	4		Ь	Infrared			1
K — — –			Viscosity centistokes				Solubility in	+		
t _k to			7	c		1 1	Acetone Carbon tet.		00	1
'x 1			₫ '				Benzene	- 1	oc oc	
A' to			l				Ether	- 1	∞	1
c, – – –	-		B ^V to				n-Heptane Ethanol	- 1	90 90	ı
A'* to		1	ĂV °	<u> </u>			Water	- 1	~	1
B'* °C			(B ^V)				Water in	\dashv		┼
Ac 40 to		4	(A ^V)					- 1		1
Bc Ltc_°C	317.5	4	c _p liq. •	ĸ						l
Cryos, A° consts, B°			c _p vap. °I	к						
t _e °C	-27.34	4	c _w vap.							
							f grams/100 g	ram	s solven	t
	CES: 1-Dow	2-A1	PI 3-Lit. 4	-Calc.	from det	. da	ta 5-Calc. by	form	nula	
SOURCE:										
PURIFICA?										
mans; 3 ² La	RE REFERE	NCES	5: 3 Chem. Re	v. <u>52</u> , 1	No. 1, 117	(195	3), Kolbe and L	ynn;	3' Timn	ner-

	-						No. 5
NAME	Methylene	chlo	ride			STRUCTURAL	FORMULA
	Dichloron	netha	ne			CH CI	
Mole % Pur. 99.8	Ref. Mol	ecul:	ar CH ₂ Cl ₂	Molecular Weight 84.94	0	CH ₂ C1	2
	1	Ref.			Ref.		Ref
F.P. °C F.P. 100%	-95.14	1	dt/dP °C/mm			f to	
B. P. °C 760 mm 100 30 10	39.75 -7.27 -28.18 -44.12 -70.55	1 1 4 4 5	25°C BP t e 30 mm AHm cal/g	0.0575 0.03692 0.03406 0.5248	5 4 5 5	h to g' °K	
Pressure mm 25°C t _e Density	435.86 841.21	5	ΔHv cal/g 25°C 30 mm BP	80.55 89.20 78.18 77.74	5 5 5	m to	
g/ml 20°C dt 25 d4 30	1.32554 1.31630 1.30700	1 1 4	t _e (d, e) ΔHv/T _e d -28 to	77.71 20.91 84.63	5 5	n'	
a b Ref. Index	1.36277 -0.00178	4	e 43 °C to co	0.1622	5	dynes/cm. 20°C 30 40	28.00 1 26.41 1 24.84 1
ⁿ D 20°C 25 55	1.42416 1.42115 1.41519	1 1 1	d g/ml vc ml/g tc °C	252.	5	Parachor [P] 20°C 30 40	149.4 4 149.6 4 149.9 4
"C"	0,4252	4	P _c mm	56400.	5		147.6 5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	16.357 16.552 0.76139	4 5 4	PV/RT 25°C 30 mm BP	0.9727 1.0000 0.9600	5 5 5	Exp. L.1.%/wt. u. Dispersion	
A -28 to B 121 °C C	7.07138 1134.6 231.	1 1 5	te tc ΔHc kcal/m	0.9573	5	Flash Point °C Fire Point M. Spec.	
A* -28 to B* 53 °C K	1.46352 1063.0	5 5	ΔHf ΔFf Viscosity			Ultra V. X-Ray Dif. Infrared	
c t _k to t _x °C	!		centistokes 7 20 °C 40 60 80	0.3868 0.3530 0.3274	1 1 1	Solubility in + Acetone Carbon tet. Benzene Ether	
B' °C C' to			B ^V 10 to A ^V 50 °C	0.3000 182.34 2.9656	4 4	n-Heptane Ethanol Water	
B'* °C		\sqcup	(B ^V) 50 to	223, 35	4	Water in	
Acl 121 to Bc t _c °C Cc	7.50819 1462.59 278.60	5 5 5	(A ^V) 90 °C c _p liq. °K	Z. 8447	4		
Cryos. A° consts. B°	0.02286	1	c vap. °K				
t _e °C T _R = 0.75	42.63 T	5	Α	L	L	+ //05	
REFERENC		24	PI 3-Lit. 4-0	Calc from do	+ 4-	† grams/100 gra ta 5-Calc. by for	
SOURCE:	20. 1-DOW		ow, dist.	Cuic, irom de	t. ua	J-Caic, by for	*******
PURIFICAT	ION:		Dow, dist.				
	RE REFERE						
L							

NAME	Chlorofor	m			- 1	STRUCTURAL FORMULA			
	Trichloro	meth	ane			Cl			
Mole % Pur, 99	Ref. Mo	lecul rmul	** CHCL 1	Molecular Weight 119.3	89	сı с≀ н с≀			
70 1 41. 77.		Ref.		weight 11713	Ref		` 	Ref	
F.P. °C	-63.59	1	dt/dP	T		f to		\vdash	
F.P. 1007			°C/mm			gK			
B. P. °C			25°C BP	0.1205 0.04107	5	h ,			
760 mm 100	61.73	4 5	t _e	0.0358	5	f' to		T	
30	-13.06	5	30 mm	0.5718	5	g'			
10 1	-30.39 -58.93	5	ΔHm cal/g	17.62	4	h¹			
Pressure	-50.75	Ť	ΔHv cal/g			m to			
mm 25°C	197.4	5	25°C 30 mm	62.11 65.66	5 5	n ! ' *K-			
t _e	906.	5	BP	57.30	5			╁	
Density g/ml 20°C	1,48316	1	te (d.e)	56.68	5	m' to			
dt 25	1.47985		(4,0)	56.68	5	0'			
⁴ 30	1.47650	4	ΔHv/T _e	19.87	5	Surface tension		一	
a	1.49652	4	d -13 to e 70 °C	64.20 0.1118	5 5	dynes/cm. 20°C	27.10	1	
b Def Zede	-0.0362	4	d' to	ļ		30 40	25.78 24.53	1 1	
Ref. Index		1	e' i •c		\vdash	Parachor [P]		+-	
45	1.44293	1	d g/ml			20°C	183.7	4	
50	1.42822	1	t _c m1/g	271.	5	30 40	182.2 180.5	4	
"C"	0.3984	4	P _c mm	39741.	5	Sugd.	184.8	5	
MR (Obs.) MR (Calc.		4 5	PV/RT			Exp. L.1.%/wt.			
(nD-d/2)	0.70435	4	25°C 30 mm	0.9990 1.0000	5 5	u. Dispersion			
Dielectric	4.639	1	BP	0.9610	5	Flash Point °C		+-	
A -15 t		4	te t	0.9561	5	Fire Point			
B L135.6	227.	4 5	t _c ΔHc kcal/m		\vdash	M Spec.		П	
A* -15 to		5	ΔHf			Ultra V. X-Ray Dif.			
B*i 90°0		5	ΔFf			Infrared			
K			Viscosity centistokes	Ì		Solubility in +		Т	
the Tee			77 20.°C	0.3789	1	Acetone Carbon tet.			
*x `			30 40	0.3481 0.3206	1 1	Benzene			
A' to		i	50	0.2969	ī	Ether n-Heptane		1	
c,	-1		B 10 to	338.5	4	Ethanol			
A'* to			Av 60 °C	7.42528	4	Water Water in			
B'* *((B ^V) to]		Water III		+-	
Ac 135 to	7.3362 C 1498.	5	(A ^V) °C	ļ	L				
Cc -c-	276.	5	c _p liq. ∘K		1 1				
Cryos, Acconsts, B	0.02418	1	c _p vap. °K						
te °C	67.32	5	c _v vap.	ł					
$T_R = 0.7$	'5 T _c					f grams/100 gran	ns solven	t	
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-(alc. from det	t. dat				
SOURCE:		D	o w						
PURIFICA			istillation						
LITERATU	RE REFERE	NCES):						

							No. 7
NAME _	Carbon T	etrac	hloride		_	STRUCTURAL	FORMULA
						C C1	
Mole % Pur. 99.	Ref. Moi	ecul mula		Molecular Veight 153.83	8	C C1 ₄	
	1	Ref.			Ref.		Re
F.P. °C F.P. 100%	-22.99	1	dt/dP °C/mm 25°C	0.1972	5	f to	
B.P. °C 760 mm 100	76.54 21.81	1 4	BP te	0. 04322 0. 0360	5	h f' to	
30	-2.31	4	30 mm	0.6040	5	g' <u>°K</u>	
10 1	-20.62 -50.82	5 5	ΔHm cal/g	5.04	4	h' i	ļ — — —
Pressure mm 25°C t _e	115.2 960.	5	ΔHv ca1/g 25°C 30 mm	50.39 52.31	5	m to	
Density g/ml 20°C	1.59397 1.58429	1 1	t _e t _e (d, e)	46.41 45.89 45.90	5 5 5	m' to	
d ₄ 30	1.57456	4	ΔHv/T _e	19.74	5	S	
a b	1.63277 -0.00190	4	d -5 to e 90 °C d' to	52.14 0.0736	5 5	Surface tension dynes/cm. 20°C 30 40	26.84 1 26.17 1 25.51 1
Ref. Index n _D 20°C 25	1.46005 1.45704	1 1	d _c g/ml	0.5576 1.7934	3	Parachor [P]	219.7 4
"C"	1.45409 0.3817	4	v _c ml/g t _c °C P _c mm	283.15 34200.	3	30 40 Sugd.	221.0 4 222.1 4 222.0 5
MR (Obs.) MR (Calc.) (nD-d/2)	26.44 26.286 0.66306	5 2 5	PV/RT 25°C	0.9973	5	Exp. L.1.%/wt. u.	
Dielectric	1		30 mm BP	1.0000 0.9703	5	Dispersion	L
A -15 to B 138 °C	6.93390 1242.43	4	te tc	0.9645 0.2718	5 4	Flash Point C Fire Point M. Spec.	None None
C A* -5 to B* 100 °C	1.51283 1155.8	4 5 5	ΔHc kcal/m ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared	
c t _k to C			Viscosity centistokes 7 20 °C 40 60	0.6091 0.47364 0.3825	1 1 1	Solubility in + Acetone Carbon tet. Benzene	
B' _ °C			B ^V 10 to A ^V 70 °C	493.5 Z. 10154	4 4	Ether n-Heptane Ethanol Water	
A'* to B'* °C Ac 138 to	7, 3703	5	(B ^V) to			Water in	
Bc t _c °C	1584. 277.	5	c _p liq. °K				
Cryos, A° consts, B°	0.00629	1	c _p vap.300°K	0,206	31		
t _e °C	84.39	5	c vap.	L		+ /: 22	
TR = 0.75		2. 4	DI 3_14 4 4	Tale from de	+ 4-	ta 5-Calc. by for	
SOURCE:	±3; 1-D0₩		w, Lit. 4-0	Jaic, irom de	aa	ia 3-Caic, by Ioi	
PURIFICAT	ION:		stillation, Lit.				
			5: 3 Young; 3'	Timmerman	s		

	N - 41 - 1		11			STRUCTURAL 1	No. 8	
NAME			mochloride			SIRUCTURAL	ORMULA	
L	Bromoch	loror	nethane			CH ₂ C1 E	r	
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 129.3	99	-		
		Ref.			Ref.			Ref
F.P. ℃			dt/dP			f to		
F.P. 100%			*C/mm 25*C	0, 1514	5	g <u>*K</u> _		
B. P. °C 760 mm	68.11	4	BP	0.04074		h		
100	16.69	5	t _e	0.035 0	5	f' to	1	
30 10	-5.89 -22.98	5	30 mm	0.5644	5	g' ' <u>*</u> K_		
1	-51.09	5	ΔHm cal/g			h'		
Pressure			ΔHv cal/g 25°C	40.40	_	m to		
mm 25°C	147.2	5	30 mm	60.68 64.81	5 5			
t _e	925.	5	BP	55.43	5	m' to		_
Density g/ml 20°C	1.93439	1	te te (d, e)	54.69 54.64	5	n' ' ' ' ' ' ' ' ' K		
t 25	1.92292	1	ΔHv/T	20.36	5	o'		
	1.91143	4	d -6 to	64.06	5	Surface tension	1 22 22	
a b	-0.00225	4		0.1267	5	dynes/cm. 20°C	33.32 31.87	1
Ref. Index			d' i to			40	30.39	1
n _D 20°C	1.48376	1	d _c g/ml	0,625	5	Parachor [P]		
25 30	1.48076 1.47761	1 4	V mi/g	1.60	5	20°C	160.7	4
"C"	0.3297	4	tc °C	297.	5	40	161.3	4
MR (Obs.)	19.13	4	P _c mm	45600.	5	Sugd.	161.3	5
MR (Calc.) (nD-d/2)	19.450	5	PV/RT 25°C	0.9919	5	Exp. L.1.%/wt.		
	0.51657	4	30 mm	1.0000	5	Dispersion		
Dielectric A -6 to	(0(/24		BP t _e	0.9618 0.956 3	5 5	Flash Point °C		
B 1297 °C	6.86624 1132.3	4	tc	0.265	5	Fire Point		_
С	216.	4	∆Hc kcal/m			M Spec. Ultra V.		
A* -6 to B* 84 °C	1.41396	5	ΔHf ΔFf			X-Ray Dif.		
B*	1060.4	٦	Viscosity		 	Infrared Solubility in +		
c t. to			centistokes	0.3486	1	Solubility in + Acetone		
tk to			7 20 ℃	0.2949	i	Carbon tet.		
A' to			60	0.2569	1	Benzene Ether		
B', ∟ _ °C		1	B ^V 10 to	222.0	4	n-Heptane		
A¹* to			B' 10 to A' 70 °C	323.8 Z.43796		Ethanol Water		
B'* *C			(B ^V) to	1		Water in		
Ac to			(A ^V) °C					
Bc tc_°C			cp liq. °K					
Cryos. A°		\vdash	-	1				
consts. B°			р -]				
t _e °C	.74, 32	5	c _v vap.					
$T_{\mathbf{R}} = 0.75$	Тc					f grams/100 gran	ns solvent	
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc, from det	da	ta 5-Calc. by for		
SOURCE:		Do	·					
PURIFICAT			stillation					
LITERATUR	E REFERE	CES	:					

								No. 9	
NAME	Bromotric	hloro	methane	·		STI	RUCTURAL	FORMUL	.A
							CCLB	-	
Mole % Pur. 99.	Ref. Mo	ecul		Molecular Weight 198,29	7		C C1 ₃ B		
		Ref.			Ref.				Ref
F. P. °C F. P. 100%	-5.65	1	dt/dP °C/mm		_	f g l	to		
B. P. °C 760 mm	104.70	١,	25°C BP	0.5246 0.04655	5 4	h			
100	45.93	1 1	t _e	0.03682	5	f'	to		
30 10	20.13	4	30 mm	0.6450	5_	g'	<u>K</u>		
1	0.60 -31.53	5	AHm cal/g	3.05	4	h' i			├
Pressure			ΔHv cal/g	44.21		m	to oK		l
mm 25°C	38.40	5	25°C 30 mm	44.56	5	0			
t _e	1027.13	5	BP	38.61	5	m'	to		†
Density g/ml 20°C	2.01221	1	te te (d, e)	37.86 37.84	5	n'	<u>•</u> K_		
dt 25 d4 30	2,00130	1	AHv/Te	19.30	5	o' <u>'</u>			
a 30	1.99037	4	d 20 to	45.97	5		ace tension	21 2/	١.
b b	2.05583 -0.00216	4	116 °C	0.0703	5	dyne	s/cm. 20°C 30	31.26 29.98	1
Ref. Index			d' to			L	40	28.72	1
ⁿ D 20°C	1.50633	1	d _c g/ml			Para	chor [P]	222.0	
50	1.50342	1 1	I v mi/g	210	_		20°C 30	233.0 233.2	4
"C"	0.3308	4	, .	318.	5		40	233.3	4
MR (Obs.)	29.294	4	P _c mm	30812.	5			235.7	5
MR (Calc.)		5	PV/RT 25°C	0.9996	5	Exp.	L.1.%/wt.		
(nD-d/2) Dielectric	0.50023	4	30 mm	1.0000	5	Disp	ersion		
A 20 to	6,86625	.	BP te	0.9580 0.9492	5		h Point C		
B 170°C	1294.08	1 1	t c				Point		-
c	220.	5	ΔHc kcal/m ΔHf			Ultr	Spec. a.V.		
A* 20 to B* 126 °C	1.55053	5 5	ΔFf				ay Dif.		
к — =	1207.0	,	Viscosity			Infra	····		\vdash
t, to		i	centistokes 7 20 °C	0,8138	1		bility in [†] etone		
t _k to °C			40	0.6091	i		rbon tet.		
A' to			60 80	0.4788 0.3893	1 1	Eth	nzene Mer		
B' ° <u>C</u>			BV I to	577.74	4		leptane		
A'* to	 	\vdash	A 50 ℃	3.9400	4	Wa	anol ter		
B'* °C			(B ^V) 50 to	528.82	4	Wa	ter in		<u> </u>
Ac 170 to	7.29174	5	(A ^V) 90 °C	2. 0931	4				
Bc tc °C	1647.18	5	c _p liq. °K						
Cryos. A°	0.00426	1	c _p vap. °K						
t _e °C	115.72	5	c _v vap.						
$T_{\mathbf{R}} = 0.75$	1	Щ	u	L	L	+ gr	ams/100 gra	ms solver	ıt.
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-	Calc. by for	mula	
SOURCE:			w, dist.						
PURIFICAT	ION:	Do	w, dist.						
	RE REFERE	NCE	5:						
		-							

NAME	Methyl b	i			T	STRUCTURAL	No. 10	
NAME	Bromom					SIRUCIURAL	CIMIO LI	•
Mole	Ref. M	olecul	ar	Molecular		CH ₃ Br		
% Pur. 99.	71 1 F	ormul		Weight 94.9	-			D. 6
	1 24 25	Ref.			Ref			Ref
F.P. °C F.P. 100%	-94.07	1	dt/dP *C/mm			f to		i
B. P. °C	† ———	+	25°C	0.0187		g (°K		
760 mm	+3.56	2	BP	0.0339 0.0346		I		
100 30	-39.40 -58.37	2 2	t _e 30 mm	0.4752	((g' to		
10	-72.78	2	ΔHm cal/g	1	H	h'		l
1	-96.56	5	ΔHv cal/g	 	\vdash	m to		
Pressure mm 25°C	1633.0	5	25°C	56.99	5	n °K		İ
t _e	739.7	5	30 mm BP	67.72 59.74	5 5	1		<u> </u>
Density			t t	59.83	5	m' to		1
g/ml-10°C	1.73676		te (d, e)	59.57	5	n' l °K		l
d ^t -5 4 30	1.14500	1 1	ΔHv/T _e	20.49	5			├ ─
a b	1.73676 -0.00241		d -58 to	0.1289	5 5	Surface tension dynes/cm,-10°C		
Ref. Index		T	d' to			40		
ⁿ D 20°C	[d g/ml v ml/g			Parachor [P] 20°C		
30	ļ			194.0	3,	30		
"C"		T	-	174.0	,	40	124.1	5
MR (Obs.)			P _c mm		\vdash	Sugd.	124.1	-
MR (Calc.) (nD-d/2)	14.583	5	25°C 30 mm	0.9399 1,000 0	5 5	Exp. L.1.%/wt. u. Dispersion		
Dielectric	<u> </u>		BP	0.9625 0.9632	5	Flash Point °C		
A -58 to B 53°C		2 2	t _e	0.9632]]	Fire Point		<u>L</u> _
_c '_42 9	238.32	2	AHc kcal/m	 		M Spec.		1
A* -58 to B* 13 °C	1.44534 920.93	5	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared		
K — — —			Viscosity	1		Solubility in +		\vdash
te to	-		centistokes 7 -30 °C			Acetone Carbon tet.	∞ ∞	
t _x i °C			-20 -10	0.2575 0.2356		Benzene	&	1
A' to B' _ °C			0	0.2174		Ether	× ×	[
c,		1 1	B" -40 to	256.30	4	n-Heptane Ethanol	×0	1
A'* to			AV -10 °C	-	l I	Water Water in		
B'* °C		\vdash	(B ^V) -10 to	251.07 Z.4183	4 4	water in		\vdash
Bc t _c °C			c _p liq. °K					
Cryos. A° consts. B°	0.02120	1	c _p vap. °K					
t _e °C	2.86	5	c _v vap.	<u> </u>		<u> </u>	L	<u> </u>
REFERENC	ES: 1-Dow	2- A F	PI 3-Lit. 4-	Cala from det		grams/100 gran		<u> </u>
SOURCE:			w, Lit.	care. Irom det	. ual	ta 5-Calc. by for	u.a	
PURIFICAT	ION:		w, dist.					
LITERATU		NCES		ans; 3' Kirk a	nd O	thmer, Encyc. of	Chemical	
- demology	<u>-</u> , 031 (17 1 0)							

							No. 11				
NAME _	Ethyl chlor	ide		·		STRUCTURAL FORMULA					
	Chloroetha	ne									
Mole	Ref. Mo	ecul	. ,	Molecular	-	CH₂ClCH₃					
% Pur.		mula		Veight 64.51	7						
		Ref.			Ref.	I		Ref.			
F, P. °C			dt/dP			f to		T			
F.P. 100%			°C/mm		- 1	g °K					
B. P. °C			25°C BP	0.0245	5	h					
760 mm	+12.27	2	t _e	0.0348	5	f' to		T			
30	-51 .5 9	2	30 mm	0.4896	5	g' <u>*K</u>					
10	-66.43 -90.93	2 5	ΔHm cal/g			h'					
Pressure	-70.73	-	ΔHv cal/g			m to					
mm 25°C	1199.0	4	25°C	88.07	5	n •K	-{	1			
t _e	764. 2	5	30 mm BP	102.92 90. 5 5	5 5	- -	ļ	+			
Density	0.03366		t_	90.52	5	m' to		1			
g/m1 0°C	0.92390 0.90280	3	te (d, e)	90.40	5	0'	1				
d ₄ 10 25	0.87062	4	ΔHv/T _e	20.49	5	Surface tension		+-			
a	0.92572	4	d -52 to e 15 °C	92.92 0.1938	5	dynes/cm. 5°C	21.20	3			
b Def Tel	-0.00203	4	d' to	1		8 10 40	20.64	3			
Ref. Index	1.3790	3	e' °C			Parachor [P]	 	+			
, 10	1.3738	4	d g/ml vc ml/g			.5°℃	151.2	4			
30 "C"	0.549(tc °C °			10 40	152.3	4			
	0.5486	4	P _c mm			Sugd	149.4	5			
MR (Obs.) MR (Calc.)	16. 137 16. 303	4 5	PV/RT			Exp. L. 1. %/wt.					
(nD-d/2)	0.8171	4	25°C 30 mm	0.9492 1.0000	5	u. Dispersion	1				
Dielectric			BP	0.9620	5	Flash Point °C		+			
A -50 to	6.94914	2	te	0.9618	5	Fire Point					
B 1_70 °C_	1012.77 236.67	2 2	tc AHc kcal/m		-	M. Spec.					
A* -50 to	1.25525	5	ΔHf	1		Ultra V. X-Ray Dif.	ľ				
B*_40 °C	945.56	5	ΔFf			Infrared	l				
K ———			Viscosity centistokes			Solubility in +					
t _k to			η 5°C	0.3196	3	Acetone Carbon tet.	1				
t _x °C			10	0.3090	3	Benzene		Ì			
A' to B' C						Ether n-Heptane					
c, '			B _v -5 to	230.8	4	Ethanol	i	l			
A¹* to			AV 1 20 °C	2. 67499	4	Water Water in	ł				
B'* °C		\vdash	(B ^V) to			water in	 	+-			
Ac 70 to Bc t _c °C			(A ^V) °C		 	4	1				
Cc	250.	5	c _p liq. °K								
Cryos, A°			c _p vap. °K					1			
te °C	12, 42	5	c vap.								
$T_R = 0.75$		ر ر	I . •	L	L	grams/100 grams/	me colec-				
		2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ata 5-Calc. by for					
SOURCE:	Litez										
PURIFICAT											
			3: 3 Timmerm	ans							

							No. 12	
NAME	1,2-Dicl	loro	ethane			STRUCTURAL	FORMUL	A
Mole % Pur. 99	Ref. Mo	lecul	ar C ₂ H ₄ Cl ₂	Molecular Weight 98.96	6	сн ₂ с1 с н	1 ₂ C1	
		Ref.			Ref			Ref
F. P. ℃	-35.66	1	dt/dP			f to		
F. P. 1009	•		°C/mm		_	gK_		ļ
B. P. °C	00.45	١.	25°C BP	0.2647 0.04192	5	h '		
760 mm 100	83.47 29.83	1 4	t _e	0.0342	5	f' to		T
30	5, 83	5	30 mm	0.6038	5	g'K_		ł
10 1	-12.54 -43.11	5	∆Hm cal/g	20.03	4	h'		↓
Pressure		 	ΔHv cal/g	22.22		to °K		1
mm 25°C	80.3	5	25°C 30 mm	83.92 86.31	5	ö <u>-</u> -		ļ
te	963.	5	BP	76.38	5	m' to		+-
Density g/ml 20°0	1.25309	1	te te (d, e)	76.88 75.40	5	n' K		1
_d t 25	1.24579	1	ΔHv/T _e	20,88	5	o'		
	1.23847	1	d 6 to	87.06	5	Surface tension		Γ.
a b	1.28209	4	_e _ _92_ °C		5	dynes/cm. 20°C	32.48 31.06	1
Ref. Index			d' to			40	28.27	5
n _D 20°0		1		0.517	5	Parachor [P]		Τ.
25 30	1.44210	1 4	d g/ml v ml/g t _c °C	1.935	5	20°C 30	188.5 188.6	4
"C"	0.4704	4	N -	306.	5	40		
MR (Obs.	21.01	4	P _c mm	49035.	5		186.6	5
MR (Calc.	21.170	5	PV/RT 25°C	0.9994	5	Exp. L.1.%/wt. u.		1
(nD-d/2)	0.81821	4	30 mm	1.0000	5	Dispersion		1
Dielectric		1	BP t _e	0.9560 0.9499	5	Flash Point °C		
A 6 t		4 4	tc	0,260	5	Fire Point		₩-
C	232.	4	ΔHc kcal/m			M Spec. Ultra V.		ł
A* 6to B* 102°		5	ΔHf ΔFf	İ		X-Ray Dif.		
K LIE	12/4.2	'	Viscosity	1		Infrared Solubility in +		┼
t.	_		centistokes	0 (47)	١, ١	Solubility in + Acetone		
t _k to			7 20 °C	0.6671 0.5287	1 1	Carbon tet.		
A' to			60	0.4337	1	Benzene Ether		
B' '	일		B ^V 10 to	0.3982 441.1	4	n-Heptane		
A'* to	,	\vdash	A 80 °C	Z. 31484	4	Ethanol Water		
B'* °((B ^V) to	1		Water in		
Ac 161 to	7.6284	5	(A ^V) °C	1				
Bc tc_°	283.	5	c _p liq. °K					
Cryos. Acconsts. B	0.01770	1	c _p vap. °K					
t _e °C	91.19	5	c _v vap.					
$T_R = 0.7$	5 Т _с			•		grams/100 gran	ns solven	t
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from det	. dat	ta 5-Calc. by for		
SOURCE:		D	ow					
PURIFICA	TION:	D	istill a tion					
LITERATU	RE REFERE	NCES	5:					

No. 13 NAME STRUCTURAL FORMULA 1, 1, 2-Trichloroethane CHC12CH2C1 Molecular C2H3Cl3 Mole Ref. Molecular % Pur. 99.90 Weight 133.415 1 Formula Ref. Ref. Ref. F. P. °C -36.59 1 dt/dP £ to F.P. 100% °C/mm °K R 0.8090 25°C 5 B. P. °C h ВP 0.0458 5 760 mm 113.77 1 0.0356 5 ſ١ 55.78 100 4 °K g' 30.36 5 30 5 30 mm 0.6349 10 11.04 5 h' ∆Hm cal/g 20,68 4 -20.5 5 m to AHv cal/g Pressure ۰ĸ n 25°C 72.75 5 mm 25°C 22.49 5 o 30 mm 72.06 5 1055.5 te 5 BP 5 61.33 m' to Density 59.70 5 n' ۰ĸ g/ml 20°C 1.43968 te (d, e) 59.8 5 01 $\mathbf{d_4^t}$ 25 1.43194 ΔHv/T_e 19.97 5 30 1.42420 Surface tension 25 d 75.96 0.1286 5 1.47061 33.61 dynes/cm. 20°C 1 126 °C -0.00154 32.24 30 1 ď to 40 30.84 1 Ref. Index e' °C 20°C 1.47124 [P] ⁿD Parachor d_c g/ml 0.497 4 25 1.46868 1 20°C 223.1 vc ml/g 2.012 4 30 1.46600 4 30 223.2 4 339. t_c 5 40 223.1 4 "C" 0.4323 4 P_c mm 36263. 5 223.8 5 Sugd. MR (Obs.) 25.91 PV/RT Exp. L.1.%/wt. MR (Calc.) (nD-d/2) 26.037 2 25°C 1.0000 5 u. 0.75140 5 1.0000 30 mm Dispersion Dielectric 7.116 1 0.9603 5 BP Flash Point °C 0.9511 5 A 30 to 6.84165 Fire Point 0.255 5 B _186 ℃ 1262.6 M. Spec. Ultra V. 205. 4 С AHc kcal/m ΔHf A* 30 to 1.35368 5 X-Ray Dif. ΔFf B*[136 °C 1183.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to 20 °C 0.82145 t_k | 1 Carbon tet. °C t_x_ 40 0.63307 1 Benzene 60 0.50894 1 ۸ı to Ether 80 0.42113 В' °C n-Heptane C' B. 10 to 489.6 4 Ethanol A | 90 °C Ž. 23823 Water A1* to (B^V)| Water in B!# °C Ac | 186 to (AV) 7, 2567 Bc tc °C 1599. cp liq. °K 254 Cryos. Aº cp vap. 0.02484 1 °K consts. B° c, vap. t, °C 125,68 $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: Dow PURIFICATION: Distillation LITERATURE REFERENCES:

							No. 14	
NAME	Methyl	chlorof	orm			STRUCTURAL 1	FORMULA	L
						CH ₃ C C	L	
Mole % Pur. 99.	92 Ref. M	iolecul Formul	ar C ₂ H ₃ Cl ₃	Molecular Weight 133.4	125	01.30 0.	.3	
<u> </u>	<u> </u>	Ref		worgan 133.	Ref			Ref
F.P. °C	-30,41	1	dt/dP	T	-	f to		
F.P. 100%			°C/mm	ł		g LK		1
B. P. °C			25°C BP	0.1829 0.04250	5 4	h		l
760 mm 100	74.096 20.35	1 1	t,	0.03643		f' to		Г
30	-3.301	4	30 mm	0.5916	5	g' ' <u>*</u> K_		ļ
10 1	-21.22 -50.75	4 5	ΔHm cal/g	4.90	4	h'		
Pressure	-30.73	+ 3	ΔHv cal/g	<u> </u>		m to		
mm 25°C	123.37	5	25°C	57.86	5	n •K_		
t _e	925.86	5	30 mm BP	61.12 52.44	5	<u> </u>		<u> </u>
Density			t_	51.74	5	m' to		1
g/ml 20°C	1.3389		t (d, e)	51.71	5	" -		l
dt 25 4 30	1.3222		ΔHv/T _e	19.51	5	Sunfa as Assadas		├
	1.3724	7 4	d -3 to		5	Surface tension dynes/cm, 20°C	25.39	1
Ъ	-0.00164	4 4	-a,- -81 €	-1	,	y 30	24.11 22.87	1 1
Ref. Index		9 1	e' •c			40	22.81	<u> </u>
n _D 20°C	1.4351		d _c g/ml			Parachor [P] 20°C	223.7	4
50	1.4216	5 1	d g/ml vc ml/g tc *C	266.	5	30	223.8	4
"C"	0.4338	4	P _c mm	30274.	5	40 Sugd.	223.9 223.8	5
MR (Obs.)		4 5	PV/RT			Exp. L.1.%/wt.		 -
MR (Calc.) (nD-d/2)	0.7684	4	25°C	0.9867	5	u.		1
Dielectric	1		30 mm BP	1.0000	5	Dispersion		<u> </u>
A -3 to	6,90160	1	te	0.9405	5	Flash Point °C Fire Point		
B 1737.6	1202.60	1	t _c			M Spec.		
C 24	225.	5	ΔHc kcal/m ΔHf			Ultra V.		
A* -3 to B*, 91 °C	1.46948	5 5	ΔFf			X-Ray Dif. Infrared		
к — — –	1		Viscosity			Solubility in +		┢
t _k	-		rentistokes 20 °C	0.6282	ı	Acetone]
t _x °C			40	0.4899	1	Carbon tet. Benzene		
A' to			60 80	0.3962 0.3612	1	Ether		
B', ∟ _ °	-1		B ^V 10 to	495.84	4	n-Heptane Ethanol		ļ
A¹* to	 		AV 50 °C	2.1070	4	Water		1
B'* *C			(BV) 50 to	236.35	4	Water in		<u> </u>
Ac 131 to			(A ^V) 90 °C	7,8886	4			
Bc tc_*C	1524.70 269.86	5	cp liq. °K					
Cryos, A° consts, B°	0.00558		c _p vap. °K					
t _e °C	80.61	5	c _w vap.					
$T_R = 0.7$	5 T _C			<u> </u>		grams/100 gran	ns solveni	<u></u>
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	da	ta 5-Calc. by for		
SOURCE:		Do	ow, dist.					
PURIFICAT	ION:	Do	ow, dist.					
LITERATU	RE REFER	ENCES	:					

No. 15 sec. Butylbromide STRUCTURAL FORMULA NAME сн₃сн₂снсн₃ Molecular C4H9Br Mole Ref. Molecular Weight 137.028 Вr 99.56 % Pur. 1 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -112.65 1 dt/dP f to °C/mm °K g 25°C 0.3281 5 B. P. °C h ВP 0.04510 5 5 760 mm 91.44 4 0.0368 ſ١ to 100 34.61 5 g' °K 9.73 5 30 mm 30 0.6216 5 10 -9.08 5 h! 12,01 4 ∆Hm cal/g 5 -39.97 1 m to AHv cal/g Pressure n ۰ĸ 60.64 62.25 25°C 5 5 mm 25°C 64.65 5 o 30 mm 989. 5 te 53.67 5 BP m¹ to Density 52.73 5 te te (d, e) n' °K 1.26085 g/ml 20°C 5 1 52,70 o' 25 1.25354 d_4^t 5 AHv/Te 19.32 30 1.24624 4 Surface tension Т 10 to d 63.27 5 1,29004 25.01 a dynes/cm. 20°C 1 1 100 °C 0.1050 ь -0.00144 4 ăĦ 30 23.90 1 to 40 22,82 Ref. Index e' °C 20°C 1.43705 ⁿD Parachor [P] 1 0.412 5 d_c g/ml 25 1.43453 1 20°C 243.0 4 5 vc ml/g 2,425 50 1.42079 1 30 243.1 4 t_c 286. 5 243.0 4 40 "C" 0.4598 4 P_c mm 26756. 5 241.1 5 Sugd. MR (Obs.) 28.48 PV/RT Exp. L.1.%/wt. MR (Calc.) 28.437 25°C 0.9978 5 u. (nD-d/2)0.80662 4 30 mm 1.0000 5 Dispersion Dielectric BP 0.9579 5 Flash Point C 0.9507 5 A 10 to 6.82724 Fire Point 0,255 B 1146 °C 1229.08 M. Spec. AHc kcal/m С 220. 4 Ultra V. ΔHf A* 10 to 1.36870 5 X-Ray Dif. ΔFf B*[110 °C 1149.0 Infrared K Viscosity Solubility in centistokes Acetone to 20 °C 0.4692 1 t_k | Carbon tet. ٠c 40 0.3884 1 Benzene 60 0.3293 1 A' | to Ether 80 0.28431 R! ٠c n-Heptane B^V | 10 A^V | 90 C' 374.6 2.39320 4 to Ethanol °C 4 AI* Water to Water in (B^V)I B'* °C to Acl 146 to 7.2470 5 (AV) °C Bc_tc_°C 1556. 5 c_p liq. °К 266. Cryos. Aº 0.03217 1 cp vap. °K consts. B° c_v vap. °C 100.72 $T_{\mathbf{R}} = 0.75 T_{\mathbf{c}}$ grams/100 grams solvent 2-API 5-Calc. by formula REFERENCES: 1-Dow 3-Lit. 4-Calc. from det. data Dow SOURCE: Distillation PURIFICATION: LITERATURE REFERENCES:

							No. 16
NAME	1,2-1	Dibromo	-1, 1-dichloroet	hane	\dashv	STRUCTURAL	FORMULA
Mole % Pur. 99	Ref.	Molecu	lar laC ₂ H ₂ Cl ₂ Br ₂	Molecular Weight 256.7	.,	СН ₂ ВгС (Cl ₂ Br
70 Pur. 9	.97 1	Ref		weight 250.7	Ref	 	Re
	-66.85		1	T	I Kei	<u> </u>	1
F.P. C F.P. 1007		1	dt/dP •C/mm	1		f to	1 1
B. P. *C	+		25°C	11.0081	5	h .	
76 0 mm	178.30	1	BP	0.05269 0.03573		f' to	+
100 30	111.28 81.54	4 4	t _e 30 mm	0.7460	5	g' '°K	1
10	58.79	4	AHm cal/g	4.73	4	h'	
11	20.64	5	ΔHv cal/g		ا∸ ا	m to	
Pressure mm 25°C	1.34	82 5	25°C	46.37	5	n •K	1 1
t _e	1246.14	5	30 mm	43.51 37.75	5	<u> </u>	
Density	†	_	BP te	36.55	5	m' to	
g/ml 20°0			te (d, e)	36.50	5	n' ' <u>°K</u>	1 1
dt 25 4 30	2.25		ΔHv/T _e	19.87	5	1	
	2.30		d 81 to		5	Surface tension dynes/cm, 20°C	39.12 1
ъ	-0.00		$\begin{array}{c c} & -1 & 199 & \circ \\ \hline d' & 25 & to \\ \end{array}$		5 5	30 30	37.92 1
Ref. Index			d' 25 to	47.63 0.0504	5	40	36.83 1
ⁿ D 20°C			d _c g/ml			Parachor [P]	202 0
35	1.55		V mi/g	1	ا ۔ ا	20°C 30	283.9 4 284.2 4
"C"	0.32		11 -	420.	5	40	284.6 4
MR (Obs.	+		P _c mm	32870.	Ľ		288.4 5
MR (Calc.) 36.70) 5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.	
(nD-d/2)	0.42	558 4	30 mm	1.0000	5	Dispersion	
Dielectric			BP	0.9617 0.9482	5 5	Flash Point °C	
A 81 to B 246 °C	7.03° 2 1593.35	778 4	t _e t _c	0.7402		Fire Point	
C (530)	205.	5	AHc kcal/m		Н	M Spec.	
A* 81 to	1.75	159 5	MA			Ultra V. X-Ray Dif.	1
B* 209 °C	1493.7	5	ΔFf		-	Infrared	
C			Viscosity centistokes			Solubility in +	
the in the			7 20 °C		1 1	Acetone Carbon tet.	1 1
t _x ; •(_	40 60	1.0399 0.7852	1 1	Benzene	
A' 25 to B' 81 °C			80	0.6209	ī	Ether n-Heptane	
C'	-	1	B 10 to	687.51	4	Ethanol	1
A ¹ * to		T	AV 50 °C	3.8219	4	Water	1
B'* °(+		(B ^V) 50 to		4	Water in	
Ac to	7.452 3 1989.74	39 5	(A ^V) 90 °C	2.0944	4		1
Ccc-	258.0	5	c _p liq. ∘K				
Cryos. A' consts. B'		139 1	c _p vap. *K				
t _e °C	199.18	5	c _v vap.				
$T_{R} = 0.7$	5 Т _с					grams/100 gra	ms solvent
REFEREN	CES: 1-De	ow 2-A	PI 3-Lit. 4-	Calc, from det	t. dat	ta 5-Calc. by for	mula
SOURCE:		D	ow, dist.				
PURIFICA			ow, dist.				
LITERATU	RE REFI	RENCE	s:				

							No. 17
NAME	Ethyl br	omid	e			STRUCTURAL	FORMULA
	Bromoet	hane					
Mole	Ref. Mo	ecul	,	Molecular		C₂H₅Br	
% Pur. 99.				Weight 108.9	76		
	-	Ref.			Ref.		Res
F.P. °C F.P. 100%	-119.33	1	dt/dP			f to	
B. P. °C			°C/mm 25°C	0.0560	5	g '°K	-
760 mm	38, 35	2	BP	0.0382 0.0354	5	f' to	+
100 30	-10.00 -31.30	2 2	t _e 30 mm	0.5330	5	g'°K	
10	-47.45 -74.08	2	ΔHm cal/g		+ +	h'	
Pressure	-74.08	3	ΔHv cal/g		!	m to	
mm 25°C	468.60	5	25°C 30 mm	59.91 66.69	5	n •K	-
t _e	837. 3	5	BP	58. 32	5	m¹ to	
Density g/ml 20°C	1.45939	1	t _e (d, e)	57.98 57.73	5 5	n' °K	1
d ₄ 25	1.44917 1.43889	1 4	ΔHv/T	20.01	5	0'	
a	1.50075	4	d -32 to	62.93	5	Surface tension dynes/cm. 20°C	23.70 1
b	-0.00195	4	_e_ _41°C	0.1203	5	8 30	22.46 1
Ref. Index	1.42416	1	e' °C			40	21.45 5
25	1.42104	1	d _c g/ml v _c ml/g	0.507 1.9724	3	Parachor [P] 20°C	165.1 4
30 "C"	1.41776	4	tc °C	230.7	3	30 40	165.3 4
MR (Obs.)	0.3862	4	P _c mm	42000.	3	Sugd	163.1 5
MR (Calc.)	19.060 19.201	4 5	PV/RT 25°C	0.9716	5	Exp. L 1. %/wt.	
(nD-d/2)	0.6945	4	30 mm	1.0000	5	u. Dispersion	
Dielectric A -32 to	9.112 6.91995	2	BP t _e	0.9600	5	Flash Point C	
B 110 °C	1090.81	2	t _c	0.2873	5	Fire Point	ļ
C	231.71	2	ΔHc kcal/m ΔHf	!		M. Spec. Ultra V.	
A* -32 to B* 51 °C	1.42140 1019.18	5	ΔFf			X-Ray Dif. Infrared	
K			Viscosity centistokes			Solubility in +	
t _k to			7 0°C	0.3191	1	Acetone Carbon tet.	σο σο
t _x °C			10 20	0.2914 0.2690	1 1	Benzene	×
B'i °C			30	0.2503	1	Ether n-Heptane	&C &C
C'			B ^V -10 to A ^V 40 °C	291. 22 2. 43797	4	Ethanol Water	oc
A'* to B'* °C			(B ^v) to	2. 43171	*	Water in	
Ac 110 to	7.59098	5	(A ^V) °C				
Bc tc °C	1545.3 290.0	5	c _p liq. °K				
Cryos, A°	-/***	<u> </u>	c _p vap. °K				
consts. B°	<u> </u>						
t _e °C	41.19	5	c _v vap.	L		L	<u> </u>
T _R = 0.75		2 4	DI 2 I 4 4	Cala (grams/100 gra	
SOURCE:				Caic. from de	t. da	ta 5-Calc. by fo	rmula
PURIFICAT			rature on, literature				
				. 52. No. 1. 1	17 (1	.953) Kolbe and Ly	nn
					11	and Dy	
L							

										No. 18	
NAME		1,2-	Dibr	omoe	thane			STRUCTU	RAL	FORMULA	.
		Ethy	l e ne	dibro	mide			CU	B-CI	J R.	
Mole % Pur. 99	. 91	Ref.	Mo Fo	lecul	ar C ₂ H ₄ Br ₂	Molecular Weight 187.8	84	OI.	Br CI	.201	
				Ref.	Ī		Ref	Ī —			Ref.
F,P. °C	Ι	9.79		1	dt/dP			f	to		
F.P. 100%		9.85		4	*C/mm 25*C	1 5220	ا ۔ ا	g	_ <u>•</u> K		
B. P. *C 760 mm	Ι,	31.36		1	BP	1.5320 0.04803	5	h			
100	1 '	70.21		4	t _e	0.0356	5	f	to		1
30 10		43.04		5	30 mm	0.6820	5	g' '_ h'	_ <u>•</u> K_		
ì	-	12.74		5	ΔHm cal/g	12.76	4	<u> </u>	to		├
Pressure	T				ΔHv cal/g 25°C	52.48	5	m n	°K		1
mm 25°C	١,,	11.70 12.		5	30 mm	51.70	5	• -			ł
Density	+::	12.		-	BP	45.6	5	m' l	to		
g/ml 20°C	:	2.17	920	1	t _e (d, e)	44.58 44.60	5	n' '_	_ <u>•</u> K_		l
dt 25 4 30		2.16		1 4	AHv/T	19.99	5	0'			L_
	+	2.22		4	d 43 to	54.67	5	Surface te		38.51	1
ь	1	-0.00		4	$\begin{bmatrix} -\frac{1}{4} & -\frac{1}{46} & \frac{9}{46} \\ -\frac{1}{4} & -\frac{1}{25} & \frac{1}{46} \end{bmatrix}$		5 5	dynes/cm.	30	37.22	i
Ref. Index							5		40	35,13	5
ⁿ D 20°C	;	1.53		1 1	d g/ml vc ml/g tc °C	0.776	5	Parachor	[P] 20°C	214 0	4
30	1	1.53		4	v ml/g	1.288	5		30	214.8 215.0	4
"C"	Ι	0.32	37	4	1	42731.	5		40 Suad	214.0	5
MR (Obs.)		27.00		4	P _c mm	12131.	۲	Exp. L.1.	_ <u> </u>	214.0	-
MR (Calc. (nD-d/2)	"	26.96 0.44		5 4	25°C	1.0000	5	ū,			1
Dielectric	T	4.77		1	30 mm BP	1.0000 0.9645	5	Dispersion			<u> </u>
A 43 to	,†-	7.06		4	t _e	0.9541	5	Flash Poir Fire Point			1
B [_215°C		69.7		4	tc	0.255	5	M Spec.			\vdash
C 424	+-	20.	050	4	ΔHc kcal/m			Ultra V.		ĺ	1
A* 43 to B* 156 °C		1.67 74.2	003	5 5	ΔFf			X-Ray Dif. Infrared	•	1	
к — — -	1				Viscosity		1	Solubility	in +		
t _k	-			ļ	r 20 °C	0.79558	1	Acetone			ł
'x					40 60	0.60455	1 1	Carbon te Benzene	et.	i	
A' to					80	0.48166 0.39768		Ether			
č, –	-			1 1	B 10 to	503.1	4	n-Heptan Ethanol	е		
A'* to					AV 90 °C	Z. 17512	4	Water		0.071	١,
B'* *(\vdash	(B ^V) to			Water in		0.071	1
Ac 215 to	18	7.49	59	5	(A ^V) °C						ļ
Cc	2	76.		5	c _p liq. °K	}					
Cryos, A° consts, B°		0.01	508	1	c _p vap. °K						
t _e °C	_	45.84		5	c _w vap.						
$T_{R} = 0.7$								f grams/1	00 gran	ns solven	t
REFEREN	CES:	1-D	ow			Calc, from det	t. da	ta 5-Calc.	by for	mula	
SOURCE:				Do	·						
PURIFICA					stillation						
LITERATU	RE	REFI	ERE	NCES	:						

No. 19 NAME 1, 1, 2-Tribromoethane STRUCTURAL FORMULA Br Br в снсн, Molecular C2H3Br3 Mole Ref. Molecular % Pur. 99.96 1 Weight 266.792 Ref Ref. F.P. °C -29.21 1 dt/dP f to F.P. 100% °C/mm 25°C -29.19 4 g <u>°K</u> 18.078 B.P. °C h BP 0.05408 5 760 mm 188.93 1 0.0364 5 ſ١ to 100 120.43 4 g' <u>•к</u> 5 30 90.21 30 mm 0.7567 5 10 67.16 5 h١ AHm cal/g 8.1637 4 5 28.62 m to AHv cal/g Pressure ۰ĸ n 25°C 47.27 5 mm 25°C 0.7751 o 30 mm 43.33 1263. te 5 ΒP 36.76 5 m' to Density 5 5 35.43 n' °K g/ml 20°C te (d, e) 2,62111 1 35.29 o' $\mathbf{d_{4}^{t}}$ 25 2.61007 1 ΔHv/T_e 19.52 5 30 2.59903 4 Surface tension 90 to 49.33 5 2.66526 dynes/cm. 20°C a 44.36 1 211 °C 0.0665 ь -0.00221 30 43.12 1 aי٦ to 41.68 40 1 Ref. Index e' ⁿD 20°C 1.59336 Parachor [P] dc g/ml 25 1.59076 1 20°C 262.7 vc ml/g t °C 30 1.58797 4 30 263.0 4 ^tc 451. 5 40 263.1 "C" 4 4 0.2945 P_c mm 35925. 5 Sugd. 264. 9 5 MR (Obs.) 34.51 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.731 25°C 1.0000 (nD-d/2)0.28278 4 30 mm 1.0000 Dispersion 0.9529 Dielectric 5.609 1 ВP Flash Point C 90 to 0.9373 5 6.94373 1562.12 Fire Point В _270°C M. Spec. С 196. 4 AHc kcal/m Ultra V. ΔHf A* 90 to 1.68974 5 X-Ray Dif. ΔFf B* 221 °C 1469.3 Infrared ĸ Viscosity Solubility in c centistoke Acetone to 20 1.4784 Carbon tet. t_x °C 1.0190 40 1 Benzene 0.7616 60 1 A'ı to Ether 80 0,5984 В' °C n-Heptane B^V | 10 A^V | 90 c' 639.3 4 to Ethanol A'* °C 3.96698 Water to (BV) Water in B'* °C to Acl 270 to (AV) 7.3511 5 °C Bc tc °C 1964. c_p liq. ۰ĸ Сc 5 252. Cryos. A° 0.01843 1 ۰ĸ cp vap. consts. Bo c_v vap. 211.00 te °C $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula Dow SOURCE: Distillation PURIFICATION: LITERATURE REFERENCES:

	*	-1-1-					No. 20	
NAME	Isopropyl 2-Chlorop		***************************************		_	STRUCTURAL I	FORMUL	A
	1 1	Topa				(CH ₃) ₂ CH	Cl	
Mole	Ref. Mo	lecul		Molecular	- 1	5 -		
% Pur. 99.	90 1 Fc	rmul	a C3117C1	Weight 78.54	_			
		Ref.			Ref	,		Ref
F.P. °C	-117.18	1	dt/dP			f to		1
F.P. 100%	-117.15	1	*C/mm 25*C	0.04031	5	g <u> K</u>		
B. P. °C	1 24 0	١. ١	BP	0.04931 0.03704	4	h		
760 mm 100	34.8 -12.17	4	t _e	0.03466	5	f' to		
30	-32.93	4	30 mm	0.5198	4	g'		1
10 1	-48.69 -74.72	5 5	ΔHm cal/g	22.48	4	h'		
Pressure	1	1	ΔHv cal/g			m to		
mm 25°C	529.5	5	25°C	83.27	5	n •K		1
te	827.2	5	30 mm BP	93.66 81.53	5 5	1		—
Density			te .	81.11	5	m' to		
g/ml 20°C		1	[[a,e)	81.10	5	n' <u>*</u> K		1
d ₄ 25	0.85599 0.84936	1 4	ΔHv/T _e	20.52	5			┼
a	0,88926	4	d -32 to	87.76	5	Surface tension	10 14	_
b	-0.00124	4	<u>•</u> _ _40_ ° ⊆		5	dynes/cm. 20°C	18.16 17.01	5
Ref. Index			d' l to			40	15.87	5
n _D 20°C		1	<u> </u>	0.341	5	Parachor [P]		
25 30	1.37464	1	d g/ml v ml/g	2.934	5	20°C		
"C"	1.37149	4	t _c *C	212.	5	30 40		
	0.5855	4	P _c mm	35451.	5		188.4	5
MR (Obs.) MR (Calc.)	20.973	4 5	PV/RT			Exp. L.1.%/wt.		\top
(nD-d/2)	0.94630	4	25°C	0.9690	5	u.		
Dielectric	9.52	1	30 mm BP	1.0000 0.9601	5	Dispersion		
A -32 to		5	t _e	0.9578	5	Flash Point °C Fire Point		
B	1081.6	5	t _c	0.27	5	M Spec.		+
С	230.	5	ΔHc kcal/m ΔHf			Ultra V.		1
A* -32 to	1.33241	5	ΔFÍ	·		X-Ray Dif.		
B* <u> 47 °C</u>	1012.14	5	Viscosity	 		Infrared		Д
c	_		centistokes			Solubility in + Acetone		1
tk to	·		7 0 ℃	0.4499	1	Carbon tet.		
_ X '		\sqcup	10	0.4079 0.3724	1 1	Benzene		1
A' to			30	0.3410	i	Ether n-Heptane		1
c, – – –	1		B ^V 0 to	334.0	4	Ethanol		
A¹* to			AV 40 °C	₹. 43017	4	Water		
B'* °C	<u> </u>		(B ^V) to			Water in		 -
Ac 90 to	7.39133	5	(A ^V) °C					
Bc tc °C	270.6	5	c _p liq. °K					1
Cryos. A°	0.03661	1	-	ļ	i i			l
consts. B°	0.03001	Ľ	р •					
t _e °C T _R = 0.7	37.21	5	c _w vap.					\perp
		2 45				grams/100 gran		ıt
	ES: 1-Dow	2-AF		calc. from det	. da	ta 5-Calc. by form	nula	
SOURCE:		Do						
PURIFICAT			stillation					
LITERATU:	RE REFERE	NCES	:					

							No. 21		
NAME	Propylene	dicl	nloride		STRUCTURAL FORMULA				
1	1,2-Dichloropropane					CH3CHC1CH2C1			
Mole Ref. Molecular C3H6Cl2 Molecular Weight 112.992									
	1	Ref.			Ref.		·····	Ref.	
F.P. °C F.P. 1009	-100,53 -100,44.	1 4	dt/dP °C/mm			f to			
B. P. °C			25°C	0.4082	4	h l			
760 mm	96.37	ļ	BP t _e	0.04442 0.03573	5	f¹ to			
1 0 0 30	40.05 15.18	1 4	30 mm	0.6231	4	g' <u>K</u>		ll	
10	-3.7 -35.0?	5 5	ΔHm cal/g	13.53	4	h!			
Dragona -	-35.07		ΔHv cal/g			m to			
Pressure mm 25°C	49.67	4	25°C	77.00	5	n <u>•K</u>			
t _e	1003.0	5	30 mm BP	78.23 67.70	5	-		\vdash	
Density			t_	66.76	5	m' to		1 1	
g/ml 20°C	1.15597	1 1	te (a, e)	66.45	5	0'	Í		
dt 25 4 30	1.14273	4	ΔHv/T _e	19.89	5	Surface tension			
a	1.18241	4	d 15 to	80.17 0.1294	5	dynes/cm. 20°C	28.65	1	
Ь	-0.00131	4	d¹ to	0.12/4		8 30 40	27.37 26.15	1 1	
Ref. Index		1	e'			Parachor [P]	.20.13	H	
25	1.43679	1	d g/ml	0.41 2.44	5	20°C		4	
30	1.42389	1	vc ml/g tc °C	304.3	5	30 40	226.2 226.3	4 4	
"C"	0.5041	4	P _c mm	33300.	5		225.6	5	
MR (Obs.) MR (Calc.		4 5	PV/RT			Exp. L.1.%/wt.			
(nD-d/2)	0.86138	4	25°C 30 mm	0.9978 1.0000	5	u. Dienersion			
Dielectric	8.96	1	BP	0.9550	5	Dispersion Flash Point C			
A 15 to		4	t _e	0.9506 0.255	5	Fire Point			
B 1160 °C	1295.9	4 5	te c AHc kcal/m	0.255	-	M. Spec.			
A* 15 to		5	∆Hf			Ultra V. X-Ray Dif.			
B* 115 °C		5	ΔFf			Infrared			
K			Viscosity centistokes			Solubility in +			
t _k T to			7 20 °C	0.7415	1	Acetone Carbon tet.			
t _x °C			40 60	0.5830 0.4747	1 1	Benzene			
A' to			80	0.3978	i	Ether n-Heptane			
_ c' ' =			B _v 10 to	459.1	4	Ethanol	1		
A¹* to			A 90 °C	₹. 29983	4	Water Water in	0.275 0.132	1 1	
B'* °C	+	_	(B ^V) to			Water in	0.132	+	
Ac 160 to		5 5	(A ^V) °C				1		
Cc	265.2	5	c _p liq. °K						
Cryos, A° consts, B°		1	c _p vap. °K						
t _e °C	106.03	5	c _v vap.						
$T_R = 0.7$	'5 Т _с					grams/100 gra	ms solver	it	
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for	mula		
SOURCE:)ow						
PURIFICA	TION:	E	ow, Distillation	1					
LITERATURE REFERENCES:									
İ									

											No. 22	
NAME	NAME Trimethylene chloro bromide						T	STRUCTURAL FORMULA				
									сн ₂ с	CH2C	H ₂ Br	
Mole % Pur. 99	. 79	Ref.	Mo Fo	lecul	arC3H6ClBr	Molecular Weight 157.4	51					
		لستسل		Ref			Ref					Ref.
F.P. °C		58.87		1	dt/dP	T		f	Γ	to		
F.P. 1007	\perp				*C/mm 25*C	2.582	5	g	¦	_ <u>•</u> K_		1
B. P. *C 760 mm	1	43.36		1	BP	0.04891	4	h_				-
100 30		81.16		1	t _e	0.0359	5	f' g'		to °K		
10		53.57 32.4		5	30 mm	0.6921	4	h'	. –			
1	1	-3.0		5	ΔHm cal/g ΔHv cal/g	14.00	4	m	1	to		\vdash
Pressure mm 25°C	1	6.49	5	5	25°C	66. 92	5	n o	i '-	_ <u>.</u> K		1
t _e	11	27.0	-	5	30 mm BP	64.89 55.78	5 5	ļ	1			ــــ
Density		1 50	404	Ι,	t.	54.29	5	m' n'	!	to °K		
g/ml 20°0 dt 25 4 30	'	1.59 1.58		1 1	e (=, =,	54. 23	5	o'	· –	_ =		
⁴ 4 30	1				ΔHv/T _e	19.79	5	Surf	ace ter	sion		
a b		1.62 -0.0 ₃		4 4	_e _l _165 °C		5	dyne	s/cm.	20°C 30	36.11 34.81	1 1
Ref. Index	_	-0.03		+	d' i to	1		•		40	33.72	i
ⁿ D 20°C	;	1.48		1	d _c g/ml	0, 522	5	Para	chor			
30		1.48 1.47		1 1	V mi/g	1.914	5			20°C 30	241.7 241.8	4
"C"	T	0.40	14	4	t _c °C	379. 2 36000.	5			40 Su ad	242.3	4 5
MR (Obs.)		28.32		4	P _c mm	30000.	-	Exp	L.1.9	Sugd.	239.3	1-
MR (Calc. (nD-d/2)	'	28.68 0.68		5 4	25°C	1.0000	5	-	u.			ł
Dielectric	25°	9.43		1	30 mm BP	1.0000 0.9500	5 5		ersion h Poin			
A 53 t		7.03		1	t _e	0.9380 0.255	5 5		Point	t ·C		
B (215 °C		75.23 12.		1 1	t _c AHc kcal/m	0.233	-	M S				
A* 53 to	\top	1.59	125	5	ΔHf			Ultr:	a V. ay Dif.			
B* 175 °C	13	87.9		5	ΔFf Viscosity	ļ	-	Infra				
·	_				centistokes		1 1		bility i	n +		
t _k t ₀				1 1	7 20 °C	0.9148 0.7042	$\begin{vmatrix} 1 \\ 1 \end{vmatrix}$	Car	rbon te	t.		
A' to				1	60	0.5659	1	Ber Eth	er			
B' °	2				80 B ^V 10 to	0.4685 489.5	1 4		leptane	;		
A'* to	+			\vdash	A* 90 °C	Z. 28480	4	Wat	anol ter		1.84	1
B'* *(:				(B ^V) to	1		Wat	ter in			-
Ac 215 to	1,7	7.32 60.2	546	5 5	(A ^V) °C							
Cc C-	- 2	52.		5	cp liq. °K			1				
Cryos. Acconsts. B		0. 02	418	1	c _p vap. °K							
t _e °C		58.64		5	c _v vap.							
$T_R = 0.7$	_							+ gra	ms/10	0 gran	ns solven	t
REFEREN	CES:	1-D	ow	2-AF		Calc. from det	dat	a 5-	Calc.	by for	mula	
SOURCE:	FICE			Do								
PURIFICA'			ים מי		w, Distillation				<u></u>			
LIIBRAIO	KE.	REFI	SR EJ	NCES	:							
<u> </u>												

							No. 23		
NAME	l,3-Dibro	mop	ropane	İ	STRUCTURAL FORMULA				
	Trimethylene dibromide					Br(CH ₂) ₃ Br			
Mole % Pur. 99	Ref. Mo	10	B1 (C11 ₂	73.51					
		Ref.			Ref.		Ref.		
F.P. °C F.P. 100%	-34.20	1	dt/dP °C/mm			f to			
B. P. °C			25°C BP	6.1946 0.05144	5 5	h (
760 mm	166.67 100.82	1 4	t _e	0.0356	5	f' to			
30	71.32	5	30 mm	0.7421	5	g' <u>°K</u>			
10	48.64 10.41	5	ΔHm cal/g	16, 10	4	h'			
Pressure			ΔHv cal/g			m to			
mm 25°C	2.604	5	25°C 30 mm	54.25 52.47	5 5	·			
t _e	1206.5	5	BP	46.52	5	m' to	 		
Density g/ml 20°C	1.98009	ı	te te (d, e)	45.30 45.34	5	n' °K]]		
at 25	1.97119	1	AHv/T	19.93	5	o'			
	1.96229	4	d 71 to	56.92	5	Surface tension			
a b	2.01568	4	_e _ 186 °C	0.0624	5	dynes/cm. 20°C	39.93 1 38.60 1		
Ref. Index		<u> </u>	d' to			40	38.60 1 37.34 1		
n _D 20°C	1.52318	1	d _c g/ml	0.732	5	Parachor [P]			
25 30	1.52075	1 4	v mi/g	1.367	5	20°C	256.3 4 256.5 4		
"C"	0.3467	4	tc °C	414.	5	40	257.5 4		
MR (Obs.)		4	P _c mm	38811.	5		25 3 .0 5		
MR (Calc.	31.158	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.			
(nD-d/2)	0.53314	4	30 mm	1.0000	5	Dispersion			
Dielectric	7 10074	ļ.,	BP t	0.9583 0.9451	5 5	Flash Point C			
A 71 to B 242 °C	7.19874	4	t _e	0.25	5	Fire Point			
c	222.	4	ΔHc kcal/m			M. Spec. Ultra V.].		
A* 71 to	1.81483 1473.6	5	ΔHf ΔFf	ì		X-Ray Dif.			
B*[196°C	- 1473.0	,	Viscosity			Infrared			
\$ c	-		centistokes	1 0421	١, ١	Solubility in TACetone			
t _k to			40	1.0431 0.7838	1 1	Carbon tet.			
A' to			60 80	0.6192 0.5068	1 1	Benzene Ether			
B'°	-		B ^V 10 to		-	n-Heptane			
A'* to		<u> </u>	A 1 90 °C	523.7 2.22212	4	Ethanol Water			
B'* °C			(B ^V) to			Water in			
Acl 242 to		5	(A ^V) °C						
Bc t _c °C	2110. - 278.	5	c _p liq. °K						
Cryos. A° consts. B°	0.02865	1	c _p vap. °K						
t _e °C	185.62	5	c _v vap.				i		
$T_R = 0.7$	5 T _C					grams/100 gra	ms solvent		
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for			
SOURCE:		Do	w						
PURIFICA'	TION:	Di	stillation						
LITERATURE REFERENCES:									
1									
L									

							No. 24			
NAME	Propylene dibromide					STRUCTURAL FORMULA				
	1,2-Dibromopropane									
Mole Ref. Molecular C3H6Br2 Weight 201.910						CH ₂ Br CH B	rCH ₃			
% Pur. 99.	86 1 Fo	rmul	C ₃ H ₆ Br ₂	Weight 201.9	10					
		Ref.			Ref			Ref.		
F.P. °C F.P. 100%	-55.50	1	dt/dP *C/mm			f to				
B, P. *C	 	\vdash	25°C	2.167	5	g		ı		
760 mm	141.99	4	BP t _e	0.05044 0.0371	5	f' to		-		
100 30	78.24 50.21	5	30 mm	0.7011	5	g' ' <u>*</u> K_	İ			
10 1	28.86 -6.78	5	ΔHm cal/g	10.58	4	h'				
Pressure		۱Ť	AHv cal/g	50.04	_	m to				
mm 25°C	8.0368 1128.	5	25°C 30 mm	50.24 48.94	5					
t _e Density	1120.	-	BP te	42.04 40.90	5	m¹ to				
g/ml 20°C		1	e (4, 6)	40.85	5	n' ' <u>°K</u> _				
dt 25 4 30	1.92344	1 4	ΔHv/T _e	19.16	5	Surface tension		-		
8	1.96962	4	d 50 to _e 158_ °C		5	dynes/cm. 20°C	34.14	1		
b Ref. Index	-0.00184	4	d' to	1	5	30 40	32.95 31.80	1		
n _D 20°C	-,	1		0,605	5	Parachor [P]		_		
50 25	1.51737 1.50420	1	d g/ml vc ml/g	1.652	5	20°C 30	252.5 252.7	4		
"C"	0.3532	4	, c	371. 30719.	5	40	252.9	4		
MR (Obs.)	31.76	4	P _c mm PV/RT	30719.	3	Sugd. Exp. L.1.%/wt.	253.0	5		
MR (Calc. (nD-d/2)	31.584 0.55370	5 4	25°C	1.0000	5	u.				
Dielectric	4.369	1	30 mm BP	1.0000 0.95 34	5	Dispersion Flash Point °C				
A 50 to		4 4	t _e t _c	0.9409 0.255	5	Fire Point				
B (210 °C	1419.6 212.	5	ΔHc kcal/m	1	-	M Spec.				
A* 50 to	1.55314	5	ΔHf ΔFf			Ultra V. X-Ray Dif.	1			
B* 168 °C	1330.9	"	Viscosity		 	Infrared Solubility in +		<u> </u>		
ct	_		centistokes	0.9454	١, ا	Solubility in + Acetone				
t _k to			40	0.8456 0.6462	1	Carbon tet. Benzene				
A' 25 to			60 80	0.5168 0.4267	1 1	Ether				
B' L 50 °C	1		B ^V 10 to	498.5	4	n-Heptane Ethanol				
A'* to			AV 90 °C	₹. 21874	4	Water Water in	0.143 0.052	1		
B'* °C	<u> </u>	5	(B ^V) to				3.052	Ť		
Bcit C	1796.	5	c _p liq. °K		-					
Cryos. Aº	265.	5								
consts. B°	0.02272	1	P .							
t _e °C	157.85	5	c _w vap.					<u></u>		
TR = 0.7			NY 0 ***			grams/100 gran		<u> </u>		
SOURCE:	ES: 1-Dow		PI 3-Lit. 4-0	Calc, from det	t, da	ta 5-Calc. by for	mula			
PURIFICAT	ION:		stillation							
	RE REFERE									

							No. 25		
NAME	1, 2, 3-Tri	bron	nopropane		STRUCTURAL FORMULA				
							~·· ~		
Water 1974 Water 1974 August 1974						Br CH ₂ CH Br CH ₂ Br			
Mole % Pur. 99	Ref. Mo	ecul mul		Molecular Veight 280.81	18				
	.,-	Ref.	,	1	Ref.	T	Ref.		
F. P. °C	16.19	1	dt/dP		-	f to			
F.P. 1009			°C/mm			g °K			
B. P. ℃			25°C BP	74.77 0.0572	5 4	h			
760 mm 100	222.16 149.18	5	te	0.03669	5	f' to			
30	116.68	5	30 mm	0.8160	5_	g'° <u>K</u>			
10	91.79 50.01	5 5	ΔHm cal/g	20.24	4	h'			
Pressure			ΔHv cal/g			m to			
mm 25°C	0.1726	5 5	25°C 30 mm	48.76 43.94	5				
Density	1336.5	•	BP	37.27	5	m' to			
g/ml 20°C	2.42086	1	t _e (d, e)	35.83 35.70	5	n' L*K_			
dt 25	2.41104	1	AHv/T	19.34	5	0'			
a 30	2.40122	4	d 117 to	51.31	5	Surface tension	45 94		
b	-0.00196	4	e 247 °C to	0.0632	5	dynes/cm. 20°C	45.86 1 44.59 1		
Ref. Index			e' C			40	43.28 1		
ⁿ D 20°C	1.58621	1	d _c g/ml	1.2887	5	Parachor [P]	301.9 4		
50	1.57120	i	v _c ml/g	0.776 550.	5	30	302.2 4		
"C"	0.3153	4	t _c °C P _c mm	56554.	5	40 Sugd	302.4 4 303.9 5		
MR (Obs.)		4	PV/RT		-	Exp. L. 1, %/wt.	303.7		
MR (Calc. (nD-d/2)	39.349 0.3758	5 4	25°C	1.0000	5	u.			
Dielectric	6.256	1	30 mm BP	1.0000 0.9400	5	Dispersion			
A 117 to	7.09534	4	t _e	0.9234	5	Flash Point °C Fire Point	None None		
B 1_322 °C	1779.2 200.	4	tc ΔHc kcal/m	0.24	5	M. Spec.			
A* 117 to		5	∆Hf			Ultra V. X-Ray Dif.			
B*[_257 °C		5	ΔFf			Infrared			
K ———			Viscosity centistokes			Solubility in +			
t _k -to			າ 20 °C	3, 1764	1	Acetone Carbon tet.			
t _x			40 60	1.8776 1.2651	1 1	Benzene			
A' to			80	0.9249	i	Ether n-Heptane			
C'	_		B _V 35 to	85 <u>0</u> .07	4	Ethanol			
A'* to B'* °C			B 35 to A 85 °C	3.55946	4	Water Water in			
Acl 322 to		5	(B ^V) to (A ^V) °C						
Bc tc °C	2246.	5			-				
	204.	5	P			1			
Cryos. A° consts. B°	0.02811	1	c _p vap. °K						
t _e °C	247.0	5	c vap.						
$T_R = 0.7$						grams/100 gra			
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula		
SOURCE:		Do							
PURIFICA			stillation						
LITERATU	RE REFERE	NCES	S:						

NAME								No. 26)
Mole	NAME	n-Buty	yl chloz	ride			STRUCTURAL	FORMUL	A
Ref. Formula		l-Chlo	robuta	ne					
Ref.		81 Ref. N	Molecul Formul	ar C ₄ H ₉ C1		569	CH ₃ (CH ₂) ₃ Cl		
F.P. °C -123.2 1 dt/4P C/mm 0.2143 5 ft to to S ft to S ft to S S C/mm 0.358 5 ft to to S S C/mm 0.358 5 ft to to S S C/mm 0.358 5 ft to to S S C/mm 0.358 5 ft to to S C/mm 0.358 5 ft to to S C/mm					,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		<u> </u>		Ref
F.P. 100% B.P. °C To mm To make the control of t	F.P. °C	-123, 2					f l to		T
Total materials Total mate				*C/mm	0 2142	_ ا	11 1		1
100		70 44					h	i	
1				t _e	0.0358	5	11 - 1	İ	
1				30 mm	0.5959	5	н :		
Pressure mm 25°C				ΔHm cal/g			<u> </u>		—
mm 25°C 102,43 5	Pressure	1			0(2)	_	11		1
Density g/ml 20°C 0.88621 1 t t t t t t t t							II i		ł
g/ml 20°C 0.88621 1 t t (d, e) 76.95 5 5 1 o' o' o'		950.6	3	li .	78.18		m' i to	 	+
A 25	g/ml 20°C	0.8862	1 1	te (d. e)			n' 'K		1
Color		0.8808	30 I I	AHV/T			0'		į
Section Sect		+							
Ref. Index nD 20°C 1.40211 1 25 1.39953 1 30 1.39686 4 4 40 20.25 20°C 30 1.39686 4 40 20.25 20°C 30 1.39686 4 40 20.25 20°C 30 269.0 5 40 20.25 20°C 30 40 20.25 20°C 30 40 20.25 20°C 30 40 20.25 20°C 30 40 20.25 20°C 30 40 20.25 20°C 30 40 20.25 20°C 30 269.0 5 5 40 40 20.25 20°C 30 269.0 5 5 40 40 20.25 20°C 30 269.0 5 5 40 40 20.25 20°C 30 269.0 5 5 40 40 20.25 20°C 30 269.0 5 5 40 40 20.25 20°C 30°C 30°C 30°C 30°C 30°C 30°C 30°C 3				_e _l _80_ °C					5 5
No. 20°C 1.40211 1 3953 1 39686 4 4 4 4 4 4 5 269.0 5 30 30 30 30 30 40 40 40		-0.0210	70 -						5
30 1.39686 4 vc ml/g 2.3.370 5 30 30 40	n _D 20°C			'	 	-	Parachor [P]		
TC"	45			v ml/g					
MR (Obs.) 25.539 5 MR (Calc.) 25.440 4 (nD-d/2) 0.95901 4 Dielectric 7.39 31 A 0 to 6.93790 2 te 0.9517 5 Tere Point Carbon technique Tere Point Carbon technique Tere Point Carbon technique Tere Point Carbon technique Tere Point Carbon technique Tere Point Tere Point Carbon technique Tere Point Carbon technique Tere Point			\rightarrow		269.0	5			1
MR (Calc.)	MR (Obs.)	 			27641.	5	J	227.4	5
Dispersion Dis	MR (Calc.)	25.440	4		0 9936	5]	ł
A 0 to 6.93790 2 te 0.9517 5 Fire Point C B 123 °C 1227.43 2 2 24.10 2 AHc kcal/m AFf		 	\rightarrow	30 mm				l	1
B 123 °C 127.43 2 224.10 2 AHc kcal/m AHf AFf		+							
A* 0 to	B 123 °C	1227.43	2	t _c			ļ		+
Viscosity centistokes Visc	A* 0 to	1.3221	4 5	ΔHf			X-Ray Dif.		
tk to tk *** °C 0.5260 0.4626 3 3 Acceptone *** °C Acceptone *** °C **	к — — —	1140.00					Solubility in +		+
A' to B' C' C BV 5 to 324.9 4 Ether n-Heptane Ethanol water in A' to B' C C C C C C C C C							Carbon tet.	∞	
At 1310 7,13392 5 (A ^V) (A ^V) (A ^V) C _p liq. °K C _p vap. °K C _p vap. T _R = 0.75T _C REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula by formula 5-Calc. by formula by	A' to B' °C						n-Heptane	∞	
B'* °C		-		B 5 to	. ""			∞	
Bc tc cc cc cc cc cc cc	B'* °C		_	(B ^V)	2.59305	7		<u> </u>	\perp
Cryos. A° consts. B° c _p vap. °K c _v vap. t _e °C 85.87 5 c _v vap. T _R = 0.75T _C fgrams/100 grams solvent grams/100 grams solvent solve	Bc t °C	1410.8	5						
te °C 85.87 5 Cv vap. TR = 0.75Tc + grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc, by formula SOURCE: Dow PURIFICATION: Distillation	Cryos. A°	1 233.0		11	:				
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: Dow PURIFICATION: Distillation			5	c _w vap.					
SOURCE: Dow PURIFICATION: Distillation									<u>1</u> t
PURIFICATION: Distillation			2-AF	PI 3-Lit. 4-	Calc. from det	t. da	ta 5-Calc. by for	mula	
					ane: 31 MB C				
			DNCES	,. <i></i>	ano, y NDO				

									No. 1	
NAME	Ethene	(Eth	ylen	e)			ST	RUCTURAL	FORMUL	A
								CH,=CH,		
Mole	Ref.	Mo!	ecul	ar c v	Molecular			2 2		
% Pur.		For	mula		Weight 28.0	52				
	1	4	Ref.			Ref.			r	Ref.
F.P. °C F.P. 100%	-169.15	5'	2	dt/dP *C/mm			f	to •K		
B. P. °C				25°C	0.0334		g h	1		
76 0 mm 100	-103.71		2	BP t _e	0.0224 0.03495	5		to		
30	-144.00)	2	30 mm	0.3049	5	g'	°K		
10 1	-153.22 -168.30		2 5	ΔHm cal/g			h'			
Pressure	100.30			ΔHv cal/g			m n	300 to	0.0698 0.0011	4
mm 25°C		j		25°C 30 mm	129.08	5	ő	, 000 K	-0.0634	4
t _e	429.3		5	BP	115.39	2	m'	700 to	0.1300	4
Density g/ml-10°(0.38	34	31	t _e (d, e)	119.91 118.53	5	n'	1000 °K	0.0010	4
dt 0	0.34		3'	ΔHv/T	20.97	5	o'		-0.0 ₆ 32	4
a 30	0, 43	-	4	d -144 to	80.15	5		face tension		
b	-0.00		4	e -115 °C to	0.3398	5	dyn	es/cm. 20°C 30		
Ref. Index				e' °C				40		
ⁿ D 20°C		i		d _c g/ml	0.21	31	Par	achor [P]		
30	ļ			vc ml/g tc °C	4.76 9.90	3' 2		30		
"C"				P _c mm	38380.	2		40 Sugd.	101.2	5
MR (Obs.) MR (Calc.)		اها	5	PV/RT			Exp	L.1.%/wt.		ا ا
(nD-d/2)	10.70	´	<i>[</i>	25°C 30 mm	1.0000	5	ļ	u.		
Dielectric				RP	0.9681	5		persion sh Point °C		
A -144to	6.74	756	2	t e t	0.9622 0.290	5 3'		e Point		İ
B 1- 61°C	585. 255.	ŀ	2 2	ΔHc kcal/m	316, 20	2		Spec.		
A* -144to	0.81		4	ΔHf ΔFf				ra V. Ray Dif.		
B* - 80°C	528.72 23.1		4 4	Viscosity		\vdash	Infr	ared		
c	-0.21	634	4	centistokes				ubility in +		
t _k [-80 to t _x -40 °C	-70.0 38.2		4 4	η -125 °C -115	0.23	2 2	Ca	rbon tet.		
A' to		$\overline{}$		-105	0.16	2		nzen e her		
B'°C	.			B. to		\vdash		Heptane		
A¹* to	 		\dashv	A I C		1		hanol ater		
B'* °C				(B ^V)			W	ter in		ļ
Acl -61 to	7.30	63	5	(A ^V)						
Bc t _c °C	397.		5	c _p liq. °K]				
Cryos. A°	0.03	725	2	c vap.300°K	0.37252	2				
consts, B°	0.01	30	2	400	0.45986	2				1
t _e °C	-112.83		5	c _v vap.	L	L_J	L	4		<u> </u>
				ion pressure (t				Cala by for		t
SOURCE:	,E3: 1-D	-OW	API	PI 3-Lit. 4-	Caic, from de	ı. da	ua 5	-Caic. by for	muia	
PURIFICAT	ION.		API							
		EREN		3: 3 Comm. L	eiden 7° 180 -	h -	/T:	mermenal 14	thing of -	1.
3' Lange						, u, c	(1 III	ermans) Ma	ımas et a	1,
_										

							No. 2	
NAME	Propene	(Pro	pylene)			STRUCTURAL	FORMULA	
Mole	Ref. Mo	lecul		Molecular	7.0	с н ₃ сн=с	CH ₂	
% Pur.	F	$\overline{}$	a 30 1	Weight 42.0	-			Ref
	-185.25#	Ref.		T	Ref		,	Kei.
F.P. °C F.P. 1007	-185, 25	2	dt/dP *C/mm			f to g K		
B. P. *C	+	1	25°C			h .	İ	
760 mm	-47.70	2	BP t _e	0.0289 0.0351	2 5	f' to	1	
100 30	-84.12 -100.06	2 2	30 mm	0.3982	5	g' 'K_		
10	-112.11	2	ΔHm cal/g			h'		
1	-131.89	5	ΔHv cal/g		\vdash	m 300to	0.0622	
Pressure mm 25°C	8582.	5	25°C	140.78	5	n600°K_	0.0011	4 4
t _e	593.97	5	30 mm BP	118,24 104,62	5 2		-0.0629	
Density	#		t.	106.03	5	m' 700to n' 1000°K	0.0970	4
g/ml 20°C	0.5139 [‡] 0.5053 [‡]	2 2	te (a, e)	105.99	5	"- 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	-0.0634	
d ₄ 25	3.3333		ΔHv/T _e	20.31	5	Surface tension	"	
	0.5585	4	d -100 to e -53 °C	92.22 0.2600	5	dynes/cm. 20°C		
ь	-0.00123	4		1 0.2000		30 40		
Ref. Index			•' i •c			Parachor [P]		
D 25			d g/ml vc ml/g	0.233 4.302	2 2	20°C	[[
30			tc °C	91.9	2	30 40		
"C"		1	P _c mm	34504.	2		140.2	5
MR (Obs.) MR (Calc.		5	PV/RT		\vdash	Exp. L.1.%/wt.		
(nD-d/2)	13.30.		25°C 30 mm	1,0000	5	u. Dispersion		
Dielectric			BP	0.9632	5	Flash Point °C		
A -100 t	/ - /	2	te	0.9694 0.274	5 2	Fire Point		
C P 1-0.€.	785. 247.	2 2	t _c ΔHc kcal/m	460.43	2	M Spec.		
A* -100 to		5	ΔHf	100.15	-	Ultra V. X-Ray Dif.		
B*43 °C		5	ΔFf		\vdash	Infrared	1	
c			Viscosity centistokes			Solubility in +		
tk tc			7 -115°C	0.48	2	Acetone Carbon tet.		
t *C *C *A' tc		├	-105 - 95	0.40 0.35	2 2	Benzene		
B' *(Ш	Ether n-Heptane		
C'	1	\perp	B ^V to			Ethanol Water		
A'* to B'* °0			<u> </u>			Water in		
Ac -0.6 to	ļ	5	(A ^V) to					
Bc t *C	1020.	5	c _p liq. °K		$\vdash \dashv$			
Cc -	282.	5	-					
Cryos. Acconsts. Bo	0.0054	2	c _p vap.300°K 400 c _v vap.	0.36456 0.45392				
t _e °C	-52.97	5	VF.	l	$oldsymbol{ol}}}}}}}}}}}}}}}}}}$	l ₊	ll	
T _R = 0.7	CES: 1-Dow	3	DT 2 T/: / -			grams/100 gran		
SOURCE:	CES: I-DOW	2-AI		alc. from det	t. dat	ta 5-Calc. by for	mula	
PURIFICA	TION							
	RE REFERE	AI						
	tion pressure	NCES	•					
# at satura	tion pressure	(trip	le point)					

No. 3 l - Butene (Butylene) NAME STRUCTURAL FORMULA CH3CH2CH=CH2 Mole Ref. Molecular Molecular C4H8 Weight 56.104 % Pur Formula Ref. Reί. Ref. -1**85.3**5# F.P. °C F.P. 100% 2 dt/dP f to °C/mm °K g 25°C 0.01483 5 B. P. °C h 0.0337 2 BP 760 mm -6.26 2 0.0358 5 f† 100 -48.77 2 to g¹ -67.41 <u>°К</u> 30 30 mm 0.4556 5 10 -81.5 2 h! ∆Hm cal/g -104.7 5 1 0.0121 300 to m ΔHv cal/g Pressure n |_600 °K 0.0013 4 25°C 86.8 2 mm 25°C 2217. o -0.0₆52 4 30 mm 107.38 5 te 5 703. BP 93.36 2 m 700 to 0.1051 4 Density 93.80 5 te (d, e) n' 1000 °K 0.0011 4 4 g/ml 20°C 0.5951 5 2 93.81 o' 0.5888[‡] -0.0₆36 25 $\mathbf{d_{4}^{t}}$ ΔHv/Te 19.86 5 30 Surface tension d -67 to 91.92 5 0.6227 dynes/cm. 20°C 12.50 5 <u>-9 ℃</u> 5 0.2293 ᇷᅴ ь -0.0011 4 30 5 11.29 to 10.11 5 40 Ref. Index e' 20°C [P] ^{n}D Parachor d_c g/ml 0.233 2 25 20°C vc ml/g t_°C 2 4.296 30 30 t_c 146. 2 40 "C" 2 P_c mm 30172. Sugd. 179. 2 5 MR (Obs.) PV/RT Exp. L. l. %/wt. MR (Calc.) 20,205 5 25°C 0.9177 5 (nD-d/2)30 mm 1.0000 5 Dispersion Dielectric ВP 0.9518 5 Flash Point C t_e 0.9540 5 -67 to 6.84290 2 Fire Point 2 926. 1 0.277 B |_40 °C 2 M. Spec. С 240. 2 AHc kcal/m 607.37 2 Ultra V. ΔHf A# -67 to 1.13369 5 X-Ray Dif. ΔFf B*| _ <u>2°C</u> 865.72 Infrared ĸ Viscosity Solubility in c centistokes Acetone to -55°C 2 η 0.32 ٠c Carbon tet. -45 0.28 2 Benzene A' -35 0.25 2 to Ether °C B n-Heptane B_v | C' to Ethanol °C Water A1# to Water in B'* (B^V) °C to Acl 40 to 7.2793 5 (A^V) °C Bc tc °C 1186. cp liq. ۰ĸ Cc 5 277 cp vap.300 K 0.06002 Cryos. A° 2 0.36664 2 consts. B° 0.0045 2 2 400 0.46414 c_w vap. te °C -8, 229 5 $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES: # at saturation pressure # at saturation pressure (triple point)

						\top			
NAME	cis-2	-But	ene				STRUCTURAL	FORMULA	A
Mole % Pur.	Ref.	Mo	lecul:	ar C ₄ H ₈	Molecular Weight 56.10	04	CH ₃ CH=0	сн сн ₃	
			Ref.	<u> </u>		Ref			Re
F.P. °C	-138.91	0	2	34/37	Ι			·	1
F.P. 100%			┝┶╢	dt/dP *C/mm	ł		f to		1
B. P. *C	 		\vdash	25°C	0.0194	5	h	· -	1
760 mm	3.72		2	BP	0.0345 0.0352	2			╁
100 30	-39.82		2	t _e			f'		1
10	-58.94 -73.42		2 2	30 mm	0,4780	5	h'	7	į
1	-97.2		5	AHm cal/g		Ш	1 222	0,0176	4
Pressure				ΔHv cal/g			m 300 to		
mm 25°C	1604.0		5	25°C 30 mm	94.5 113. 3 8	2 5	0	-0.0628	
t _e	741.2		5	BP	99.46	2	} 		+
Density		#		te,,	99.61	5	m' 700 to		
g/ml 20°C	0.62	13′ _≠	2 2	t (a, e)	99.61	5	0, 11000 2	-0.0639	
d ₄ 25 30	0.60	94	4	AHv/T _e	20.23	5	0	<u> </u>	╁
a	0.64		4	d -59 to	100.29	5	Surface tension dynes/cm, 20°0		5
Ъ	-0.00		4	├ ॄ ,	0.2221	5	30	13.80	5
Ref. Index				d' l to	}		40	12.55	5
n _D 20°C	1				0,238	2	Parachor [P]]
25 30				d g/ml vc ml/g	4.206	2	2000	7	1
	 		-	tc •C	155.	2	30 40	l l	
"C"	ļ		Ш	P _c mm	31160.	2		d. 179.2	5
MR (Obs.)			ا ۔ ا	PV/RT		\vdash	Exp. L.1.%/wt		<u> </u>
MR (Calc.) (nD-d/2)	20.20	כי	5	25°C	0.8766	4	u.		1
Dielectric	 		\vdash	30 mm BP	1.0000	5	Dispersion		L.,
A -59 to	6.86	024	H	t _e	0.9637 0.9643	4 5	Flash Point °C		
B 541 °C	960.1	920	2 2	tc	0.276	2	Fire Point	<u> </u>	-
č — = =	237.		2	ΔHc kcal/m	605,73	2	M Spec.	ŀ	1
A* -59 to	1, 12	:61	5	ΔHf			Ultra V. X-Ray Dif.	ļ	
B* _ 13 °C			5	ΔFf		\sqcup	Infrared		
к — — —				Viscosity			Solubility in	.	
t _k	-{		1 1	centistokes 7°C	ì		Acetone	1	
tx c				7			Carbon tet.		
A' l to	†		$\vdash \vdash$		l		Benzene Ether	ŀ	
B'				- ·		$\vdash \dashv$	n-Heptane		
C'	<u> </u>		$oxed{oxed}$	B ^V to	}		Ethanol		
A'* to				AV C	ļ		Water Water in		
B'* °C				(B ^V) to	1			+	+
Ac 541 to Bc t °C	7.30 1229.	23	5	(A ^V) °C		igsquare			
Cc	275.		5	c _p liq. °K			1		
Cryos. A°	0.04	877	2	1	0.33794	2			
consts. B°	0.00		2	c _p vap300°K	0.43366				
te °C	3.07	,	5	c _w vap.					
$T_R = 0.7$	1		لــنــا	at saturation	Dressure		+ ~~~ /100		<u></u>
REFERENC		OW	2-AF				grams/100 gr		t
	,		AI		aic. Irom det	. dat	ta 5-Calc. by fo	rmula	
SOURCE:				·					
PURIFICAT			AI						
	RE REF	er en	CES	:					
LITERATU									
LITERATU									
LITERATU									

No. 5 trans-2-Butene NAME STRUCTURAL FORMULA CH₂CH=CHCH₂ Mole Ref. Molecular Molecular C₄H₈ % Pur Weight 56.104 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -105.550 2 dt/dP f to °C/mm ۰ĸ g 25°C 0.0181 B. P. ℃ h ВP 0.0345 2 760 mm 2 0.88 t_e 0.0356 5 f١ 100 -42.69 to 2 g' °K 30 0.4783 5 -61.82 30 mm 10 -76.3 2 h! AHm cal/g 1 -100.1 5 300 to 0.0816 m AHv cal/g Pressure 600 °K 0.0011 n 25°C 91.8 2 mm 25°C 1753. ٥ -0.0625 4 30 mm 110.28 5 $\mathbf{t_e}$ 729.7 5 96.94 ВP 2 700 to 0.0851 Density m 4 97.17 5 te te (d, e) n' 1000 °K 0.0011 g/ml 20°C 0.6042 2 97.17 0.5984 ٥' -0.0637 4 25 2 d_4^t ΔHv/Te 19.97 5 30 0.5925 4 Surface tension -62.0 to 97.13 5 0,6291 4 13.43 5 dynes/cm. 20°C -0, 2°C 0,2128 5 ᇷᅱ -0.0010 h 4 30 12.27 5 to 5 40 11.13 Ref. Index e' °C nD 20°C [P] Parachor d_c g/ml 0.238 2 25 20°C vc ml/g t °C 4.206 30 30 t_c 155. 2 40 "C" P_c mm 31160. 2 Sugd. 179.2 5 MR (Obs.) PV/RT Exp. L.1.%/wt. MR (Calc.) 20,205 5 25°C 0.9333 (nD-d/2) 30 mm 1.0000 5 Dispersion Dielectric BP 0.9595 4 Flash Point C 0.9606 5 A -62 to 6.86952 2 Fire Point 0.276 2 960.8 B 1 49°C 2 M. Spec. C 240. 2 AHc kcal/m 604,73 Ultra V. ΔHf A* -62 to 1.13404 5 X-Ray Dif. ΔFf B*| _ <u>10 °C</u> 895.9 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. t_x °C Benzene ۸' to Ether В' °C B^V A n-Heptane to Ethanol °C Water A'* to Water in B'* (B^V)| °C to (A^V)| Acl 49 to 7.3082 5 °C Bc t_c °C 1231. cp liq. °K Cc 5 278 c_p vap.300 K Cryos. A° 0.04177 0.37573 2 consts. B° 2 0.0058 2 400 0.46378 c vap. te °C -0.183 5 $T_{\mathbf{R}} = 0.75 \, \mathbf{T}$ at saturation pressure grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

	_								No. 6	
NAME	2-M	ethyl	prop	ene			STI	RUCTURAL	FORMULA	1
	Isob	utene	,					CH ₃ C = C	н,	
Mole	Ref.	Mo	lecul		Molecular	ļ		с́н ₃	2	
% Pur.			rmul		Weight 56, 1	04				
			Ref.			Ref.				Ref
F.P. °C	-140.35	0	2	dt/dP			f	to		
F.P. 100%				*C/mm 25*C	0,0145	4	g	<u>*K</u>		
B. P. *C 760 mm	-6.90	00	2	BP	0.0336	2	h	 		<u> </u>
100	-49.30	9	2	t _e	0.0355	4	f' g'	to *K		
30 10	-67.90 -81.95		2 2	30 mm	0.4645	5	h'	'		
1	-105.06		5	ΔHm cal/g		-	m	1 300 to	0.0498	4
Pressure	T			ΔHv cal/g 25°C	87.7	2	n'	_600 °K_	0.0012	4
mm 25°C	2278. 710.1		4 5	30 mm	107.14	5	٥	i	-0.0 ₆ 45	4
Density	+		-	BP	94.22 94.59	2	m'	700 to	0.1217	4
g/ml 20°C	0.59)42 [‡]	2	te te (d, e)	94.59	4	n' o'	1 1000 °K	0.0011 -0.0 ₆ 34	4
d ₄ 25	0.58	379 °	2 4	AHv/T	20.06	5		<u> </u>	-0.0634	1
	0.62		4	d -68 to		5		face tension es/cm. 20°C	12.42	_
ъ ~	-0.00		4	<u></u>		5	3,11	30	11.22	5
Ref. Index				e' %				40	10.04	5
ⁿ D 20°C				d _c g/ml	0.234	2	Par	achor [P] 20°C		
30				V_ mi/g	4.278 144.73	2 2	1	30		
"C"				t _c •C	29982.	2		40 Suad	179.2	5
MR (Obs.)	I	_		PV/RT	27702.	-	Fyn	. L.1.%/wt.	117.2	-
MR (Calc.) (nD-d/2)	20.20	15	5	25°C	0.9314	4	-	u.		Ì
Dielectric	 		Н	30 mm BP	1.0000	5		persion		
A -68 to	6, 84	134	2	t.	0.9646	5		sh Point °C e Point		i
B 39°C	923.2		2	t _c	0. 276	2		pec.		├
<u>C</u>	240.		2	ΔHc kcal/m	603.36	2	Ultr	a V.		
A* -68 to B*1 °C	861.1	.232	5	ΔFf	ł	1		ay Dif. ared		
к — — —	1			Viscosity	ļ.			bility in +	<u> </u>	
t _k to	-{			centistokes 7°C	:	1	Ac	etone		1
t C				'		1		rbon tet. nzene		
A' to						}	Etl	her		
B' •C	·			B ^V to	†			Heptane nanol		
A¹* to	1			AV I °C			Wa	ter		
B'* °C				(B ^V) to	7		W a	ter in		-
Ac 39 to	7.27	77	5	(A ^V) °C	:					
Bc _tc_*C	1183.		5	c _p liq. °K						
Cryos. A*	0.04	044	2	cp vap.300°K	0.38126	2				
consts. B°	0.00)5	2		0.47358	2				
t _e °C	-8.62	3	4	c _v vap.			L			<u></u>
$T_R = 0.7$				≠ at saturation				ams/100 gran	ns solven	t
REFERENC	ES: 1-D	ow	2-AF		Calc. from de	t. da	a 5	Calc. by for	mula	
SOURCE:				PI						
PURIFICAT				PI						
LITERATU	RE REF	EREI	NCES	:					-	

								No. 7	
NAME	l-Penten	e				ST	RUCTURAL	FORMUL.	A
Mole % Pur.		lecul rmula		Molecular Weight 70.13	10		CH ₃ (CH ₂) ₂ CI	H=CH ₂	
		Ref.	L		Ref.				Ref.
F.P. ℃	-165.220	2	dt/dP			f	to		
F.P. 100%		L	°C/mm 25°C	0.04297	ا ۔ ا	g	'• <u>K</u>		
B.P. °C 760 mm	29.968	2	BP	0.04297	5 5	h			
100	-17.92	2	t _e	0.0360	5	f'	to		
30 10	-38.91	2 2	30 mm	0,5247	5	g'	' ° K		
10	-54.8 -80.9	5	ΔHm cal/g	<u> </u>		h'			_
Pressure			ΔHv cal/g	04.00	_	m	300 to	0.0283 0.0013	
mm 25°C	648.8	4	25°C 30 mm	86.93 98.81	5 5	0		-0.0647	
t _e	813.4	5	BP	86.24	5	m'	700 to	0, 1035	4
Density g/ml 20°C	0.64050	2	t _e t _e (d, e)	85.88 85.88	5	n'	1000 °K	0.0011	4
dt 25	0.63533	2	ΔHv/T	19.74	5	ە'		-0.0 ₆ 37	4
	0.63012	4	d -39 to	91.71	5		face tension		
a b	0.66168 -0.0 ₃ 954	4	_e_ <u>32 °C</u>		5	dyn	es/cm. 20°C 30	15.57 14.50	5
Ref. Index		<u> </u>	d' to	į			40	13.45	5
n _D 20°C	1.37148	2	d _c g/ml	0, 227	5	Par	achor [P]		
25 30	1.36835	2 4	v _c ml/g t _c °C	4.409	5		20°C ∃		
"C"	0, 7764	4		188.	5		40		
MR (Obs.)		2	P _c mm	25113.	5	<u> </u>	<u> </u>	218.2	5
MR (Calc.	24.36	5	PV/RT 25°C	0, 9657	5	Exp	o. L. l. %/wt.		
(nD-d/2)	1.05123	2	30 mm	1,0000	5	Dis	persion	126.1	2
Dielectric	/ 2//	_	BP t	0.9600 0.9581	5		sh Point C		
A -39 to B 73 °C	6.84650 1044.9	2 2	t e t c	0.270	5		e Point		
С	234.	2	∆Hc kcal/m	754.25	2		Spec. ra V.		
A* -39 to B* 42 °C	1.16738 974.8	5	ΔHf ΔFf	1		X-1	Ray Dif.		
K - 12 0	- 7/4.8	3	Viscosity				ared		<u> </u>
°			centistokes				ubility in [†] etone		
t _k to t _x °C			7 -20 °C -10	0.30 0.27	2 2	Ca	rbon tet.		
A' to	1	t	0	0.24	2		enzene her		
B'°C	.		B _v to		-	n-	Heptane		
A!* to	+	-	B' to				hanol ater		
B'* °C			(B ^V) to	-		W	ter in		
Acl 73 to	7,2751	5	(A ^V) °C]				
Bc tc °C	1324. - 272.	5	c _p liq. °K		П				
Cryos. A°	0, 05995	2		0.3751	,				l
consts. B°	0.03993	2	c _p vap.300°K 400	0.37516 0.47198	2 2				
t _e °C	31.94	5	c _v vap.						
$T_R = 0.7$	5 T _c					+ g:	rams/100 gra	ms solven	t
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc, from de	t. da		-Calc. by for		
SOURCE:		AF	PI						
PURIFICAT	TION:	AF	PI						
LITERATU	RE REFERE	NCES	5:						

							No. 8	
NAME	cis-2-P	enten	•			STRUCTURAL F	ORMULA	
						CU CU CU-C	иси	
Mole	Ref. M	olecul	la r	Molecular		CH ₃ CH ₂ CH=C	псп ₃	
% Pur,		ormul		Weight 70.13	0			
		Ref			Ref.			Ref.
F.P. °C F.P. 100%	-151.390	2	dt/dP	1		f to		
B. P. *C	<u> </u>	+	*C/mm 25*C	0.0537	5	g <u>*K</u>		ĺ
760 mm	36.942	2	BP	0.03830 0.0358	2 5	f' to		<u> </u>
100 30	-11.42 -32.66	2 2	t _e 30 mm	0.5310	5	g' 'K_		
10 1	-48.7 -75.2	2 5	ΔHm cal/g	1	╅	h'		
Pressure	-15.2	13	ΔHv cal/g		1	m 300 to	-0.0106	4
mm 25°C	494.6	5	25°C 30 mm	91.48 102.91	5	n _600°K_	0.0013 -0.0 ₆ 46	
t _e	829.6	5	BP	89.12	5	m' 700 to	0.0568	4
Density g/ml 20°C	0.6556	2	t _e (d, e)	88.62 88.61	5	n' 1000°K	0.0012	4
dt 25 4 30	0.6504 0.6452	2 4	ΔHv/T _e	19.87	5	0'	-0.0 ₆ 41	4
a 30	0.6767	4	d -33 to	96.44	5	Surface tension	17.14	_
Ъ	-0.0398	4	-å,-¦ ⁴⁰ €		5	dynes/cm. 20°C	17.14 16.01	5
Ref. Index		Τ,	e' °C			40	14.89	5
n _D 20°C	1.3830	2 2	d _c g/ml	0.232 4.303	5	Parachor [P] 20°C		
30	1.3766	4	v _c ml/g t _c °C	210.	5	30 40		İ
"C"	0.7808	4	P _c mm	26762.	5	Sugd.	218.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT	0.045		Exp. L.1.%/wt.		
(nD-d/2)	1.0552	2	25°C 30 mm	0.9671	5	u. Dispersion	130.	2
Dielectric		↓	BP	0.9560 0.9535	5	Flash Point °C		Ē
A -33 to B <u>82</u> °C		2 2	t _c	0.268	5	Fire Point		
C	231.	2	ΔHc kcal/m	752.74	2	M Spec. Ultra V.		ĺ
A* -33 to B*50 °C		5	ΔHf ΔFf	ļ		X-Ray Dif.		
K	1002.4	1	Viscosity	1		Infrared Solubility in +		
cto	-	-	centistokes 7°C			Acetone		İ
tx c	;	1	'			Carbon tet. Benzene		
A' to		T				Ether		ĺ
J	-	1	B ^V l to			n-Heptane Ethanol		ĺ
A¹* to				_		Water Water in		ĺ
B'* °C		5	(B ^V) to					
Bc t °C	1350.	5	(A ^V) °C					ĺ
Ce	210.	5	c _p liq. °K					ĺ
Cryos. A° consts. B°	0.05768 0.0048	2 2	c _p vap 300 °K 400	0.34864 0.45016	2 2			
t _e °C	39.52	5	c _w vap.					
$T_R = 0.7$						grams/100 gram	ıs solvent	
	CES: 1-Dow	2-A1		Calc. from det	t, dat			
SOURCE:			PI					
PURIFICAT			PI					
LILEKATU	RE REFERE	NCES	5:					

NAME trans-2-Pentene STRUCTURAL FO		A.
Mole	:н ₃	
Mole	H ₃	
F.P. °C -140.244 2 dt/dP f to		
E D 100%		Ref
25°C 0 0527 5		
760 mm 36, 353 2 BP 0.03824 2 h		_
100 -12.00 2 te 0.0358 5 ft to to 30 mm 0.5325 5 gt to 4 co		
10 -49 4 2 50 mm		
1 -76.0 5 Arm cal/g	0406	4
Pressure 25°C 91 02 5 n 600 °K 0	0012	4
mm 25 C 505.5 5 30 mm 102.10 5 0 -0	9639	4
BP 88.92 5 720 1 0	0980	4
g/ml 20°C 0.6482 2 t (d,e) 88.44 5 n 1000 °K 0	0011	4
dt 25 0.6431 2 AHv/T _e 19.88 5	0636	4
0 6600 4 d -33 to 95.80 5 Surface tension		_
1 1 1 e 1 39 °C 1 0.1893 5 37 300 cm. 20 0 1 0	. 38 . 29	5
Ref. Index e ¹ °C 40 14	. 23	5
n _D 20°C 1.3793 2 d _c g/ml 0.237 4 Parachor [P] 20°C	l	İ
30 1 3729 4 V _C m1/g 4.416 4 30		
"C" 0 7825 4 C		_
MR (Obs.) 25.02 2 PV/RT Exp. L.1.%/wt.	. 2	5
MR (Calc.) 24.36 5 PV/RT Exp. L.1.%/wt. (nD-d/2) 1.0552 2 25°C 0.9653 5 u.		
Dielectric 30 mm		2
A 23 to 0.9536 5 Flash Point °C		
B 81 °C 1084.0 2 tc 0.268 5		
C 233. 2 AHc kcal/m 751.66 2 Ultra V.		
7-1 -35 to 1.22421 5 AFf X-Ray Dif.		
K Viscosity Salubility in the		
Acetone		
tk to Carbon tet.		
A' to Ether		
B' o n-Heptane C' to Ethanol		
A' to Ethanol Water		
B ¹ * °C (B ^V) to Water in		_
Acl 81 to 7.3347 5 (A ^V) °C		
Bc tc °C 1370. Cc 272. 5 c _p liq. °K		
Cryos. A. 0.05685 2 c_vap.300K 0.37131 2		
consts. B° 0.0052 2 P 400 0.46585 2		
t _e °C 38.86 5 c _v vap.		<u> </u>
$T_{R} = 0.75 T_{C}$ + grams/100 grams		t
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc, by formul	<u>a</u>	
SOURCE: API		
PURIFICATION: API		
LITERATURE REFERENCES:		

							No. 10	
NAME	2-Methyl-	l -bu	tene			STRUCTURAL :	FORMULA	Ł
Γ						CII CII C	611	
					一	CH ₃ CH ₂ C =	CH ₂	
Mole		lecul		Molecular	.	СН	3	
% Pur.	F0	rmul		Weight 70.13	_	<u> </u>		D . 6
	T	Ref.	·	T	Ref.			Ref
F.P. °C F.P. 100%	-137.560	2	dt/dP *C/mm	ĺ		f to		
B. P. *C	 		25°C	0.0449	5	-		Ì
760 mm	31.163	2	BP	0.03778	2	h f' to		
100 30	-16.54	2	t _e	0.0358	5	g' to 'K_		
10	-37.50 -53.4	2 2	30 mm	0.5239	5	h'	1	
1	-79.5	5	ΔHm cal/g	ļ		m 300 to	0.0201	4
Pressure			ΔHv cal/g 25°C	88.15	5	n 600 °K	0.0014	4
mm 25°C	610.0 813.	5	30 mm	100.15	5	•	-0.0 ₆ 56	4
t _e Density	1013.	-	BP	86.99	5	m' 700 to	0.1147	4
g/ml 20°C	0.6504	2	te (d, e)	86.62 86.62	5	n' 1000°K	0.0011	4
_a t 25	0.6451	2	ΔHv/T _e	19.83	5	o' ¦	-0.0 ₆ 37	4
	0.6397	4	d -38 to	 	5	Surface tension		
a b	0.6721 -0.0 ₃ 98	4	_e_ 33 °C		5	dynes/cm. 20°C	16.56 15.42	5
Ref. Index	-0.0370	1	d' to			40	14.30	5
n _D 20°C	1.3778	2				Parachor [P]		
- 25	1.3746	2	d g/ml	0.237 4.224	5	20°C		ļ
30	1.3713	4	tc °C	191.	5	30 40		
"C"	0.7769	4	P _c mm	26188.	5		218.2	5
MR (Obs.) MR (Calc.)	24.85 24.36	2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0526	2	25°C 30 mm	0.9618	5	u.	,,,,	2
Dielectric			BP BP	1.0000 0.9560	5	Dispersion	133.	-
A -38 to	6.87314	2	t _e	0.9541	5	Flash Point °C Fire Point		
B <u> </u>		2	t _c	0, 268	5	M Spec,		┢
C	233.	2	ΔHc kcal/m ΔHf	750.57	2	Ultra V.	ļ	
A* -38 to B* 43 °C	1.20404 986.94	5	ΔFf			X-Ray Dif. Infrared	ŀ	
K	1 /		Viscosity			Solubility in +		┢
cto	-		centistokes 7°C			Acetone		
t _x to t _x *C			7 ℃			Carbon tet.		
A' to		_				Benzene Ether		
B' L _ *C			B ^V to	 	-	n-Heptane		1
C'	<u> </u>		B' to			Ethanol Water		1
A'* to B'* °C	1		<u> </u>	-		Water in		L
Ac 75 to	7,3005	5						
Bc t °C	1334.	5		1				
Cc — —	271.	5	c _p liq. ∘K	1				
Cryos. A° consts. B°	0.05167 0.0043	2 2	c _p vap.300°K 400	0.38243 0.48068	2 2			
t _e °C	33.12	5	c _w vap.					
$T_{R} = 0.75$	Tc					+ grams/100 grai	ms solveni	<u> </u>
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	da			
SOURCE:		AI	PI					
PURIFICAT	ION:	AI	PI					
LITERATUI	RE REFEREI	CES	j:					

								No. 11	
NAME	3 - Meth	yl-l-bı	atene			ST	RUCTURAL	FORMUL	A
							сн _з сн сн	=CH ₂	
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 70.13	.0		сн ₃	-	
		Ref.	,	T	Ref.	Γ'			Ref
F.P. °C F.P. 100%	-168.495	2	dt/dP °C/mm			f g	to •K		
B. P. °C 760 mm	20.061	2	25°C B P	0.0326 0.03721 0.0363	5 2 5	h f'	to		_
100 30	-26.82 -47.3	2 2	t _e 30 mm	0.5128	5	g' ¦	<u>*K</u>		
10 1	-62.9 -88.3	2 5	ΔHm cal/g			_h'			_
Pressure mm 25°C	902.	5	ΔHv cal/g 25°C 30 mm	81.51 93.96	5	m n o	300 to	0.0231 0.0015 -0.0 ₆ 77	4
t _{e.} Density	784.	5	BP	82.20	5	m'	700 to	0.1380	4
g/ml 20°C dt 25 d4 30	0.6272 0.6219 0.6168	≠ 2 ≠ 2 4	t _e t _e (d, e) ΔHv/T _e	81.96 82.05 19.54	5 5	n' o'	1000 <u>°</u> K	0.0011 -0.0 ₆ 37	4
a b	0.6490 -0.0009	4	d -47 to e 21 °C	85,70	5		face tension es/cm. 20°C 30	15.57 14.50	5
Ref. Index			d' to e' °C				40	13.45	5
ⁿ D 20°C 25 30	1.3643 1.3611 1.3587	≠ 2 ≠ 2 4	dc g/ml vc ml/g tc °C	0.219 4.568 170.	5 5 5	Par	achor [P] 20°C 30		
"C"	0.7784	4	t _c °C P _c mm	23294.	5		40 Sugd.	218.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	24.94 24.36 1.0507	2 5 2	PV/RT 25°C 30 mm	0.9518 1.0000	5		L.1.%/wt.		
Dielectric			BP	0.9550	5		sh Point °C		H
A -47 to B -60 °C C	6, 8261 1013, 474 237,	8 2 2 2 2	te tc ΔHc kcal/m	0.9576 0.270 752.33	5 2	M.	e Point Spec.		-
A* -47 to B* 31 °C	1.1675 946.98	-+	ΔHf ΔFf			X-F	ra V. Ray Dif. Pared		
t _k to			Viscosity centistokes 7°C			Ac Ca	ability in + etone arbon tet.		
A' to B' °C						Et	nzene her Heptane		
A¹* to			B ^V to C	-		Et Wa	hanol ater ater in		
B'* °C	7,2588	5	(B ^V) to (A ^V) °C						\vdash
Bc t _c °C	1286. 274.	5 5	c _p liq. °K						
Cryos, A° consts, B°	0.0588 0.0047	2	c vap.300°K 400 c vap.	0.40596 0.50278	2				
t _e °C T _R = 0.75	20.93 T ≠.	5	c _v vap.	<u> </u>	L	L		<u> </u>	Ļ
			n pressure PI 3-Lit. 4-	Calc. from de	t. da		ams/100 gra		t
SOURCE:		AF			-,				
PURIFICATI	ION:	AF							
LITERATUR	E REFER	RENCES	5 :						

						No. 12	
NAME	2-Methyl	-2-b	utene			STRUCTURAL FORMULA	
						CH ₃ CH=C CH ₃	
Mole	Ref. Mo	lecul	ar	Molecular		Ċн ₃	
% Pur.		rmul		Weight 70.13	0		
		Ref.		•	Ref	R	ef.
F.P. °C F.P. 1007	-133.768	2	dt/dP *C/mm	1		f to	
B. P. °C	1	\vdash	25°C	0.0566	5	g <u>*K</u>	
760 mm	38.568	2	BP t _e	0.03844 0.0357	2 5	f' to	
100 30	-10.06 -31.48	2 2	30 mm	0.5360	5	g' ''K_	
10 1	-47.7 -74.5	2 5	AHm cal/g			h'	
Pressure	-14.5	-	ΔHv cal/g				4 4
mm 25°C	466.1	5	25°C 30 mm	91.86	5		4
t _e	834.3	5	BP	89.73	5	┣ ── ─ └──	4
Density g/ml 20°0	0.6623	2	te (d, e)	89.21 89.21	5	n' 1000°K 0.0011 4	4
dt 25 4 30	0.6570 0.6517	2	ΔHv/T	19.89	5	o' -0.0 ₆ 39	4
a 30	0,6838	4	d -31 to	97.01	5	Surface tension dynes/cm, 20°C 17.87	5
ь	-0. 0 210	4	-å,-¦⁴1- ° ⊆	0.1889	5	30 16.68 9	5
Ref. Index			e, C			l	5
ⁿ D 20°C	1.3874	2 2	d _c g/ml	0.242 4.128	5	Parachor [P] 20°C	
30	1.3809	4	t _c *C	204.	5	30	
"C"	0.7813	4	P _c mm	27547.	5	40 Sugd. 218. 2	5
MR (Obs.) MR (Calc.		2 5	PV/RT	2 2/15	_	Exp. L.1.%/wt.	_
(nD-d/2)	1.0563	2	25°C 30 mm	0.9645 1.0 0 00	5	u. Dispersion 135.	2
Dielectric			BP	0.9560	5	Flash Point °C	_
A -31 t		2 2	t _e t _c	0.9534 0.268	5	Fire Point	
_c	233.	2	∆Hc kcal/m	749.08	2	M Spec. Ultra V.	
A* -31 to B* 51 °C		5	ΔHf ΔFf	,		X-Ray Dif.	
B*	1026.4]	Viscosity			Infrared Solubility in +	
ct	_		centistokes 7°C	ł		Solubility in + Acetone	
t _x (7	ł		Carbon tet. Benzene	
A' to				ļ		Ether	
B' *	4		B ^V to			n-Heptane Ethanol	
A¹+ to			AV °C			Water Water in	
B1* °(+	_	(B ^V) to			water in	
Ac 85 to		5 5	(A ^V) °C	ļ			
Cc	2 72.	5	c _p liq. ∘K				
Cryos, Acconsts, B		2 2	c _p vap.300°K 400	0. 3 5962 0. 45 530	2 2		
t _e °C	41.32	5	c _v vap.	<u> </u>			
$T_R = 0.7$						grams/100 grams solvent	
	CES: 1-Dow	2-AI		alc. from det	da	ta 5-Calc. by formula	
SOURCE:	TION:		PI PI	 		· · · · · · · · · · · · · · · · · · ·	
	RE REFERE						
			••				

No. 13 1-Hexene NAME STRUCTURAL FORMULA CH3(CH2)3CH=CH2 Mole Ref. Molecular Molecular C 6H12 Weight 84.156 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% -139.819 2 dt/dP f to °C/mm 25°C g °Κ 0.1267 5 B. P. ℃ h ВP 0.04149 760 mm 63.485 2 2 ^te 0.0362 5 ſ١ 11.109 to 100 g' <u>°К</u> -11.88 30 30 mm 5 0.5748 2 10 -29.3 h١ ∆Hm cal/g -57.9 5 300 to 0.0258 m AHv cal/g Pressure 1 600 °K 0.0013 n 25°C 88,33 mm 25°C 187.2 5 o -0.0₆48 4 30 mm 93.50 5 5 905. t_e ВP 80.66 5 m' 700 to 0.1031 4 Density 79.72 5 te te (d, e) 1000 °K 0.0011 n' g/ml 20°C 0.67317 2 5 79.70 ٥' -0.0638 4 d_4^t 25 0.66848 2 AHV/T 5 19.60 30 0.66377 Surface tension d -12 91.48 5 to a 0.69199 dynes/cm. 20°C 17.87 ᇷᅴ <u>69</u> °C 0.1705 5 -0.03906 ь 4 30 16.86 5 to 15.86 5 40 Ref. Index e' ⁿD 20°C 1.38788 [P] Parachor d_c g/ml 0.230 25 1.38502 2 20°C vc ml/g t_°C 4.342 5 30 1.38219 4 30 t_c 5 228. 40 "C" 0.7696 4 P_c mm 22668. 5 Sugd. 257. 3 5 MR (Obs.) 29.492 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 28.978 25°C 0.9857 (nD-d/2)u. 1.05130 2 30 mm 1.0000 5 Dispersion 121.9 2 Dielectric BP 0.9550 5 Flash Point C 0.9499 A -12 to 6.86572 2 Fire Point 0.265 5 103 °C В 1152.971 2 M. Spec. AHc kcal/m С 226. 2 901.14 2 Ultra V. ΔHf A* -12 to 1.23301 5 X-Ray Dif. ΔFf B*[_ 79 °C 1079. Infra red ĸ Viscosity Solubility in centistokes Acetone to tk tx -10°C 0.37 2 Carbon tet. °C 20 0.26 2 Benzene 40 0.22 2 A' to Ether 60 0.19 В' °C n-Heptane B^V | C to Ethanol °C Water A'* to Water in (B^V)| B'* °C to Acl 103 to 7,2845 (A V) 5 °C Bc tc " ٠c 1445. 5 c_p liq. ۰ĸ Cc 265. 5 cp vap.300K Cryos, A 0.37763 2 consts. Bº 400 0.47566 2 c_v vap. t_e °C 69.08 5 $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{c}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 14	
NAME	cis-2-He	xene				STRUCTURAL	FORMULA	
	D-6 No	1 1		Mala -ula m		CH ₃ (CH ₂) ₂ CH	=CHCH ₃	
Mole % Pur.		lecul rmul		Molecular Weight 84.1	56			
		Ref.			Ref.			Ref.
F.P. C	-141.135	2	dt/dP			f to		
F. P. 1007	<u> </u>	<u> </u>	*C/mm 25*C	0,1530	5	g <u>°K</u> _		
B. P. *C 760 mm	68.84	2	BP	0.041	2	h		
100 30	15.85 -7.46	4	t _e	0.0361	5	f' to g'K_		
10	-25.13	5	30 mm ΔHm cal/g	0.5832	5	h'		
1	-54.23	5	ΔHv cal/g			m 300 to	0,0155	4
Pressure mm 25°C	151.	5	25°C	89.40	5	n 600 °K	0.0013 -0.0 ₆ 36	4
t _e	919.	5	30 mm BP	95.30 82.33	5	L		
Density	0 (0(0		t_	81.31	-5	m' 700 to n' 1000 °K	0.0978 0.0011	4
g/ml 20°0 dt 25 4 30	0.6869	2	t _e (d, e) ΔHv/T _e	81.28	5	0, 1	-0.0 ₆ 36	4
	0.6777	4	d -7 to	19.65	5	Surface tension		
a b	0.7053 -0.0 ₃ 89	4	<u>e l _75 °C</u>	94.03 0.1700	5	dynes/cm. 20°C	19.39 18.34	5 5
Ref. Index		Ė	d' to			40	17.31	5
n _D 20°0	1,3977	2	d _c g/ml	0, 238	5	Parachor [P]		
25 30	1.3948 1.3920	2 4	11 V m1/g	4.202	5	20°C 30		l
, "C"	0.7722	4	tc °C	238.	5	40 Sund	257.2	_
MR (Obs.		2	P _c mm	23801.	3	Exp. L.1.%/wt.	257.3	5
MR (Calc. (nD-d/2)) 28.978 1.0542	5 2	25°C	0.9847	5	u.		
Dielectric		亡	30 mm BP	1.0000 0.9540	5	Dispersion	125.	2
A -7 t	6.89962	5	t _e	0.9485	5	Flash Point °C Fire Point		
B 1_110.6	21184.6 226.	5	t _c AHc kcal/m	0.264	-	M Spec.		
A* -7 to		5	ΔHf	899.54	2	Ultra V. X-Ray Dif.		
B* L 85 °C		5	ΔFf		_	Infrared		
c c	1		Viscosity centistokes			Solubility in +		
tk to		į	η ℃	ļ		Acetone Carbon tet.		
t of		-				Benzene Ether		
B' •		1	B ^V to		 i	n-Heptane		
C' to			B' to			Ethanol Water		
A'* to			(B ^V) to			Water in		
Ac 110 to	7, 32029	5	(A ^V) °C					
Bc tc_"	C 1484. - 266.	5	c _p liq. °K					
Cryos. A		Ť	c _p vap.300°K	0.35874	2			
consts. B		L_	400	0.45867	2			
t _e °C	75.01	5	c _v vap.			l <u>, </u>		L
T _R = 0.7						grams/100 gram	ns solvent	<u> </u>
	CES: 1-Dow			alc. from det	t. da	ta 5-Calc. by for	mula	
SOURCE:	TION:	A1						
PURIFICA	RE REFERE	Al						
	NE REFERE		•					

No. 15 trans-2-Hexene NAME STRUCTURAL FORMULA CH3(CH2)2CH=CH CH3 Molecular C6H12 Molecular Mole Ref. % Pur Weight 84.156 Ref. Ref. Ref. F.P. °C F.P. 100% -132.970 2 dt/dP f to °C/mm 25°C °K g B.P. °C 0.1480 5 h BP 0.041 2 760 mm 67.87 2 t_e 0.0361 5 ſ١ to 100 15.01 4 g' <u>×</u> 30 4 -8.25 30 mm 0.5818 5 -25.87 5 10 h' ∆Hm cal/g 1 -54.90 5 300 to 0.0582 AHv cal/g Pressure n | 600 °K 0.0012 25°C 89.70 5 mm 25°C 157. 5 o -0.0631 4 30 mm 94.98 5 te 5 916. ВP 82.05 5 700 to 0.0760 m 4 Density 81.04 5 te te (d, e) n' 1000 °K 0.0012 g/ml 20°C 0.6784 2 5 81.02 o' -0.0642 4 25 d_4^t 0.6738 2 AHv/T 19.65 5 30 0.6692 4 Surface tension Т -8 to 93.58 5 a 0.6968 4 4 dynes/cm. 20°C 18.44 5 <u>°C</u> 74_ 0.1698 5 ь -0.0389 30 17.43 ď٠ to 40 16.45 5 Ref. Index e¹ [P] 20°C 1.3935 2 ⁿD Parachor d_c g/ml 0.235 5 25 1.3907 2 20°C vc ml/g t °C 5 4.257 30 1.3879 4 30 ^tc 236. 5 40 "C" 0.7741 4 5 P_c mm 23399. Sugd. 257.3 5 MR (Obs.) 29.64 PV/RT Exp. L. l. %/wt. MR (Calc.) 28.978 5 25°C 0.9937 5 (nD-d/2)1.0543 2 30 mm 1.0000 Dispersion 127. 2 Dielectric BP 0.9540 5 Flash Point C te t 0.9485 5 -8 to 6.89830 Fire Point 0.264 5 1181.0 5 M. Spec. С AHc kcal/m 226. 5 898.54 2 Ultra V. A*I ΔHf -8 to 1.26070 5 X-Ray Dif. ΔFf 8<u>4</u> °C 1106.0 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. $t_{\underline{x}}$ °C Benzene A'I to Ether B' °C n-Heptane Bv Av C to Ethanol °C Water A'* to Water in (B^V) B'* °C to Ac | 108 to (A^V)| ۰ç 7.3188 5 Bc tc °C 1479. 5 c_p liq. ۰ĸ Cc' 266. 5 c_p vap.300K Cryos. Aº 0.37763 2 consts. B° 0.47174 2 c_v vap. te °C 5 73.93 $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

							No.	16
NAME	cis-3	-Hexen e				STRUCTURA	AL FORMU	LA
						CII CII CII		
Mole % Pur.	Ref.	Molecul Formul	ar C ₆ H ₁₂	Molecular Weight 84.1	56	CH ₃ CH ₂ CH	:=Снсн ₂ Сн	¹ 3
		Ref.		weight oill	Ref	T T		Ref
F.P. °C	-137.82		dt/dP	T	1	f	to	
F.P. 100%			°C/mm	1		g	°K.	
B. P. *C			25°C BP	0.1410 0.041	5 2	h .		1
760 mm 100	66.44	2	t,	0.0361	5	f	to	
30	-9.43		30 mm	0.5797	5	g' '	_K_	
10 1	-26.98 -55.90	5	ΔHm cal/g			h'		
Pressure	-33.70		ΔHv cal/g		П		0to -0.02	
mm 25°C	166.	5	25°C 30 mm	88.52	5	n 1_60	0.06	
t _e	912.	5	BP	94.47 81.62	5 5			
Density g/ml 20°C	0.67	06 3	t _e ,	80.64	5	m' 70 n' 100	0to 0.06 0°K 0. 0 0	
	0.67		t _e (d, e)	80.62	5	o'	-0.06	
d ₄ 25	0.67		ΔHv/T _e	19.64	5	Surface tens	ion	_
	0.69		d -94 to		5	dynes/cm. 2	0°C 18.57	
b	-0.03	91 4		-			0 17.53 0 16.52	
Ref. Index n _D 20°C	1.39	47 2	e' i •c		\vdash	Parachor [F		+-
D 25	1.39	20 2	d g/ml	0.236 4.245	5		o.c	
30	1.38		v _c ml/g t _c °C	233.	5		0	-
"C"	0.77		P _c mm	23329.	5		ugd. 257. 3	5
MR (Obs.) MR (Calc.)	29.67 28.97		PV/RT		\vdash	Exp. L.1.%/		
(nD-d/2)	1.05		25°C 30 mm	0.9879	5	u.		
Dielectric			BP	1.0000	5	Dispersion	126.	2
A -94 to	6.89	493 5	t _e	0.9487	5	Flash Point ' Fire Point		
B 1 107.€	1175.4 226.	5	t _c	0.264	2	M Spec.		
A* -94 to			ΔHc kcal/m ΔHf	899.54		Ultra V.		- 1
B* 82 °C	1.25 1097.1	194 5	ΔFf			X-Ray Dif. Infrared		
K ———			Viscosity			Solubility in	+	\dashv
t to	-		centistokes 7 °C			Acetone		
tx C			'	Í		Carbon tet. Benzene		1
A' to						Ether		
B', ∟ _ <u>•</u> C	·		B ^V to	T	\vdash	n-Heptane Ethanol		
A¹* to			A ^V I °C		1	Water		
B'* °C			(B ^V) to	1	1 1	Water in		
Ac 107 to			(A ^V)					
Bc Ltc_°C	1472. 266.	5	c _p liq. °K	1				- [
Cryos. A°	1	- -	c _p vap.300°K	0.35303	2			-
consts. B°	 		400	0.45748	2			
t _e °C T _R = 0.79	72.33	5	c _v vap.	<u> </u>	Ш	L		
REFERENC		3 4 7	OT 2 1 2	2-1- (grams solv	ent
SOURCE:	, 1-DC		PI 3-Lit. 4-0 PI	Calc. from de	t. dat	ta 5-Calc. by	iormula	
PURIFICAT	ION:		PI PI					
LITERATU						·		

No. 17 trans-3-Hexene NAME STRUCTURAL FORMULA CH3CH2CH=CHCH2CH3 Molecular Mole Ref. Molecular C6H12 % Pur Formula Weight 84.156 Ref. Ref. Ref. F, P, °C -113.430 2 dt/dP f to F.P. 100% °C/mm °K g 25°C 0.1441 5 2 B. P. °C h ВP 0.041 760 mm 67.08 2 ſ١ 0.0361 5 100 14.30 4 g' °ĸ 30 -8.91 4 30 mm 0.5807 5 10 -26.50 5 h' ∆Hm cal/g -55.47 5 1 0.0191 m 300 to AHv cal/g Pressure 0.0014 4 n 600 °K 25°C 88.86 5 mm 25°C 162. 5 0 -0.0₆53 4 30 mm 94.68 5 5 t_e 914. BP 81.80 700 to m' | 0.0469 4 Density 80.81 5 te (d, e) n' 1000 °K 0.0013 4 g/ml 20°C 0.6772 2 5 80.78 ٥' -0.0647 4 0.6725 25 2 d₄ ΔHv/T_e 19.64 5 0.6678 30 4 Surface tension -9 to 93.17 5 0.6961 4 dynes/cm. 20°C 18.31 5 0.1696 5 å <u>73</u> °C ь -0.0391 17.28 16.27 4 30 5 to 40 5 Ref. Index e' ٠c 20°C 1.3943 ⁿD [P] Parachor d_g/ml 0.234 5 25 20°C 1.3916 2 vc ml/g t_ °C 5 4.270 30 1.3886 4 30 t_c 234. 5 40 "C" 0.7770 4 P_c mm 23237. 5 Sugd. 257.3 5 MR (Obs.) 29.75 2 PV/RT Exp. L. 1. %/wt. MR (Calc.) 28.978 5 25°C 0.9888 5 u. (nD-d/2) 1.0557 2 30 mm 2 1.0000 5 Dispersion 128. Dielectric BP 0.9540 5 Flash Point C 0.9486 5 ţ. A -9 to 6.89575 5 Fire Point 0.264 5 В 107 °C 1177.7 M. Spec. 226. С 5 AHc kcal/m 898,54 Ultra V ΔHf A*[-9 to 1.25454 5 X-Ray Dif. ΔFf 83 °C B* 1100.5 Infrared K Viscosity Solubility in c centistokes Acetone °c to Carbon tet. ٠c Benzene A' | to Ether °C_ B' n-Heptane B^V | C' to Ethanol °C Water A'* to Water in (B^V) B'* ۰c to Acl 107 to 7.3162 (A^{V}) 5 °C Bc tc ۰c 5 °K c_p liq. 5 266. c_p vap,300°K Cryos. Aº 0.37906 2 consts. B° 400 0.47768 2 c, vap. t_ °C 73,05 5 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

NAME	.8
Mole	LA
Ref.	
F. P. °C	Re
F. P. 100%	Re
B. P. °C 760 mm 60.7 2 100 8.77 4 30 1.4.06 4 30 mm 0.5710 5 1 10 1.5 1.5	
760 mm 60.7 2 100 8.77 4 30 mm 0.0361 5 1 10 1.4.06 4 10 -31.35 5	
100	-
10	-
Density September Septem	
This This	52 4
The color of the	
Density g/ml 20°C 0.6799 2 t e (d, e) 79.13 5 79.13 5 1000 °K 0.00 0.02 0.6720 4 d -1.4 to 90.21 5 30 17.70 16.64 6 -0.0393 4 d -1.4 to 90.21 5 30 17.70 16.64 6 -0.0393 4 d -1.4 to 90.21 5 30 17.70 16.64	56 4
g/ml 20°C 0.6799 2 t	
A	
C	<u> </u>
Delectric Price	5
Ref. Index np 20°C 1.3891 25 1.3891 2 2 30 1.3856 4 4	5
TC' 0.7666 4	5
TC' 0.7666 4	
MR (Obs.) 29. 47 MR (Calc.) 28. 978 (nD-d/2) 1.0506 2 Dielectric	-
MR (Cos.) 27. 4	5
Note Column Col	+3
30 mm	1
A -14 to 6,88772 5 te 0,9492 5 Flash Point °C B 100 °C 1154.7 5 C 227. 5 AHC kcal/m 897.54 2 AHG AFf AFf	2
B 100 °C 1154.7 5 C 0.265 5 M Spec.	
C	+
A 1 10 10 10 10 10 10	
Viscosity Centistoken Viscosity Carbon tet. Benzene Ether n-Heptane Ethanol Water Water in Viscosity Centistoken Viscosity Centistoken Viscosity Centistoken Viscosity Carbon tet. Carbon tet. Benzene Ether n-Heptane Ethanol Water Water in Viscosity Centistoken Viscosity Centistoken Viscosity Centistoken Viscosity Carbon tet. Carbon tet. Centistoken Viscosity Centistoken Viscosity Carbon tet. Centistoken Viscosity Centistoken Viscosity Carbon tet. Centistoken Viscosity Centistoken Viscosity Carbon tet. Carbon tet. Carbon tet. Carbon tet. Carbon tet. Carbon tet. Carbon tet. Carbon tet. Carbon tet. Carbon tet. Centistoken Viscosity Centistoken	
Centistokes Centistokes	
Carbon tet. Benzene Ether Benzene Ether	
A' to B' C' BV to AV °C Ether n-Heptane Ethanol Water in Ac 100 to 7.30907 5 Cp liq. °K Cp vap300 °K 0.38690 2	
B' \ _ °C \	
A'* to B'* °C AV °C Water water in	1
B'* °C	i
Ac 100 to 7,30907 5 (A ^V) °C	
Bc t _c °C 1447. 5 c _p liq. °K Cryos. A° c _p vap300°K 0.38690 2	
Cryos. A° c _p vap.300 °K 0.38690 2	1
Cryos. A	
Consts. B 400 0.48481 2	
t _e °C 65.92 5 c _v vap.	
T _R = 0.75 T _C [†] grams/100 grams solv REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula	<u>ent</u>
SOURCE: API PURIFICATION: API	
LITERATURE REFERENCES:	

No. 19 3-Methyl-1-pentene STRUCTURAL FORMULA NAME CH3CH2CH CH=CH2 ĊН3 Mole Ref. Molecular Molecular C6H12 Weight 84, 156 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% -153.0 2 dt/dP f to °C/mm g °K 25°C 0.0936 5 B.P. °C h ВP 2 0.040 760 mm 54.14 2 t_e 0.0362 5 ſ١ to 100 3.06 4 g' <u>°к</u> 30 -19.38 4 30 mm 0.5612 5 10 -36.38 5 h' ∆Hm cal/g 5 1 -64.34 300 to -0.0261 m AHv cal/g Pressure _600<u>°</u>K 0.0017 n 25°C 83,10 5 mm 25°C 264. 5 5 o -0.0682 4 30 mm 90.36 5 877. te BP 5 78.99 m' 700 to 0.1155 4 Density te (d, e) 77.39 5 0.0012 n' 1000 °K g/ml 20°C 4 0.6675 2 5 77.38 ٥' -0.0642 4 $\mathbf{d_{4}^{t}}$ 25 0.6628 2 AHV/T 19.63 5 30 0.6581 4 Surface tension ď -19 5 to 87.36 a 0.6865 4 dynes/cm. 20°C 17.24 5 ᇷᅴ <u>59 ℃</u> 0.1546 5 ь 4 -0.0390 30 16.24 5 40 15.25 5 Ref. Index e' ⁿD 20°C 1.3842 [P] Parachor d_c g/ml 0.234 5 25 1.3814 2 20°C vc ml/g t °C 4,276 30 1.3786 4 30 ^tc 213. 5 40 "C" 0.7691 4 P_c mm 22329. 5 Sugd. 257.3 5 MR (Obs.) 29.49 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 28.978 5 0.9798 25°C 5 (nD-d/2)u. 1.0504 2 30 mm 1. 6000 Dispersion 2 5 5 124 Dielectric BP 0.9540 Flash Point C 0.9498 5 ţe. -19 to 6,87729 Fire Point 0.265 5 1130.4 5 1_ 92 °C M. Spec. Ultra V. C 229. 5 AHc kcal/m 900.08 2 ΔHf A* -19 to 5 1.26723 X-Ray Dif. ΔFf B* 69 °C 1062.0 Infrared ĸ Viscosity Solubility in centistokes Acetone to $t_{\mathbf{x}}^{\mathbf{t}}$ Carbon tet. °C Benzene A'ı Ether B' °C n-Heptane Bv C' to Ethanol °C Water A'* to Water in ۰c (B^V) B'* to Acl 92 to (A^V)| 7.2989 5 °C Bc ۰c 1417. t_{c_}° c_p liq. ۰ĸ Сс 267. 5 c_p vap.300°K Cryos. A 0.41376 2 consts. B° 0.52759 2 400 c vap. te °C 5 58,61 $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

								No. 20	
NAME	4-M	ethyl	-1-p	en te ne			STRUCTURAL	FORMULA	.
Mole % Pur.	Ref.	Mo Fo	lecul	ar C ₆ H ₁₂	Molecular Weight 84.1	56	сн ₃ сн сн ₂ сн ₃	сн=сн ₂	
			Ref			Ref			Ref.
F, P. *C	-153, 63	3	2	dt/dP	1		f to	1	\Box
F.P. 100%				*C/mm			gK		
B. P. *C				25°C BP	0.0928 0.0404	5 2	h		1
760 mm 100	53.88		2 4	t _e	0.0361	5	f' to		
30	-19.58	3	4	30 mm	0.5607	5	g'	4	
10 1	-36.57 -64.51		5	AHm cal/g			h'	<u> </u>	L
Pressure				ΔHv cal/g			m 300 to n 600 °K	-0.0583 0.0015	
mm 25°C	267.		5	25°C 30 mm	82.95 90.29	5 5	0 -000 1	-0.0 ₆ 57	4
t _e	877.		5	BP	78.14	5	m' 700 to	0,0700	├
Density g/ml 20°C	0.66	642	2	te te (d,e)	77.41 77.40	5 5	n' 1000 °K		
at 25	0.65		2	e (4, 6)	1	5	0'	-0.0 ₆ 42	4
4 30	0.65	546	4	AHv/Te	19.65	5	Surface tension		<u> </u>
a b	0.68		4 4	d -20 to		5	dynes/cm. 20°C	16.90	5
Ref. Index	-0.0	372	-	a' to	5		30 40	15.90 14.91	5
n _D 20°C		328	2			├ ┊┤	Parachor [P]	1	Ť
25 30	1.37		2 4	d g/ml vc ml/g	0.233 4.295	5 5	20°C	1	Ì
"C"	1.37		4	tc •C	212.	5	30 40		ļ
MR (Obs.)	29.54		2	P _c mm	22185.	5	Sugd	257.3	5
MR (Calc.			5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.05	507	2	25°C 30 mm	0.9807 1.0000	5	u. Dispersion	124.	2
Dielectric				BP	0.9550	5	Flash Point °C	1	اٽ
A -20 to		7757	5	te t _c	0.9509 0.265	5 5	Fire Point		<u> </u>
C F 1 € 57.6	1130.0		5	ΔHc kcal/m	899,44	2	M Spec.		
A* -20 to	1,25	5996	5	ΔHf] ",,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		Ultra V. X-Ray Dif.		
B* _ 68 °C	1059.0		5	ΔFf			Infrared		
K — — —	i			Viscosity centistokes			Solubility in +		
t _k to			1	7 °C	;]]	Acetone Carbon tet.	į	l
x '	1		L				Benzene	ļ	ŀ
A' to							Ether n-Heptane		ļ
c,				B ^V to			Ethanol	ŀ	l
A¹* to				A ^V - •C	<u>: </u>		Water Water in		
B'* °C				(B ^V) to	Į.		W4.61 III		
Ac 91 to Bc t _c °C		991	5	(A ^V) •C	;			İ	ł
Cc ' c-	267.		5	c _p liq. °K	:				
Cryos, A° consts, B°				c _p vap.300°K	0.36171 0.46224	2 2			
t _e °C	58.35	5	5	c _v vap.					
$T_R = 0.7$						'	+ grams/100 gra	ms solven	t
REFERENC	CES: 1-D	ow	2-AF	PI 3-Lit. 4-	Calc, from de	t. dat	ta 5-Calc, by for		
SOURCE:			A.	PI					
PURIFICAT				PI					
LITERATU	RE REF	ERE	NCES	:					

No. 21 2-Methyl-2-pentene NAME STRUCTURAL FORMULA CH3CH2CH=C CH3 ĊН Mole Ref. Molecular Molecular C6H12 % Pur Weight 84.156 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -135.070 2 dt/dP f to °C/mm <u>_°K</u> g 25°C 0.1451 5 5 B. P. °C h ВP 0.0418 760 mm 67, 29 2 t_e 0.0362 5 f¹ 100 14.48 to 4 4 g' <u>°К</u> -8.75 30 5 30 mm 0.5811 5 10 -26.36 h' ∆Hm cal/g -55.34 5 300 to -0.0144 AHv cal/g 0.0014 -0.0₆58 Pressure n <u>60</u>0_°K 4 4 25°C 88.87 5 mm 25°C 160. 5 o 30 mm 94.73 5 914. 5 t_e ΒP 81.84 5 m 700 to 0.0644 4 Density te (d, e) 5 80.84 0.0012 n' 1000 °K g/ml 20°C 0.6863 2 -0.0642 80.82 o' 4 d_4^t 25 0.6815 2 AHv/Te 5 19.64 30 0.6767 4 Surface tension d -9 to 93.24 5 0.7055 4 dynes/cm. 20°C 19.32 7<u>3</u> °C 0.1695 5 -0.0393 ь 4 5 30 18.22 ā٦ to 40 5 17.16 Ref. Index e' °C ⁿD [P] 20°C 1.4004 Parachor d_c g/ml 5 0.237 25 1.3976 2 20°C vc ml/g t °C 4.216 5 30 1.3945 4 30 t_c 235. 5 40 "C" 0.7778 4 P_c mm 23580. 5 Sugd. 257.3 5 MR (Obs.) 29.75 PV/RT Exp. L. l. %/wt. MR (Calc.) 28.978 25°C 0.9837 5 (nD-d/2)1.0573 4 30 mm 2 1.0000 Dispersion 131. 5 Dielectric ВP 0.9540 Flash Point C 0.9486 5 -9 to 6.89488 5 Fire Point 0.264 5 B [_108 °C 1178.1 5 M. Spec. AHc kcal/m С 5 226. 896.14 Ultra V. ΔHf A*l -9 to 1.25698 5 X-Ray Dif. ΔFf B*| 83 °C 1101.9 Infrared ĸ Viscosity Solubility in centistokes Acetone to tk tx Carbon tet. °C Benzene A' to Ether В' ٠c n-Heptane B^V | C' to Ethanol •c Water A'* to Water in B'* °C (B^V)| to Acl 108 to (A^V)| 7.3157 5 5 °C Bc tc C 1476. c_p liq. ۰ĸ Cc 266. 5 c_p vap.300 K Cryos. A 0.36147 2 consts. B° 0.46343 2 c, vap. te °C 73.28 5 $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 22	
NAME	3-Methyl	-cis-	-2-pentene			STRUCTURAL	FORMULA	
						GY GY 6 - 6		
						CH ₃ CH ₂ C = C	нсн ₃	
Mole % Pur.		lecul rmul		Molecular Weight 84.1:	56	Ċн ₃		
		Ref.			Ref	<u> </u>		Ref.
F.P. *C	-138.445	2	dt/dP	T .		f to		
F.P. 100	6		°C/mm	0.1/27	ا ۔ ا	g <u>°K</u> _		
B. P. °C 760 mm	70.45		25°C BP	0.1627 0.0421	5 4	h ;		
100 mm	70.45 17.21	2 4	t _e	0.0362	5	f¹ to		
30 10	-6.21 -23.95	4	30 mm	0.5859	5	g' K_		
1	-53.19	4	ΔHm cal/g			h¹	0.0144	4
Pressure			ΔHv cal/g 25°C	89.77	5	m 300 to	-0.0144 0.0014	4
mm 25°C	140.97 923.5	4	30 mm	95.77	5	0	-0.0 ₆ 58	4
Density	723.5	1	BP	82.73 81.67	5	m' 700 to	0.0644	4
g/m1 20°0	0.6986	2	te te (d, e)	81.64	5	n' 1000 °K_	0.0012	
dt 25	0.6942 0.6898	2	ΔHv/T	19.64	5		-0.0 ₆ 42	*
1 30	0.7162	4	d -6 to	94.71	5	Surface tension dynes/cm, 20°C	20.75	5
ь	-0.03855	4	-å,-¦ ⁷⁷ %	0.1701	5	30	19.69	5
Ref. Inde			e, .c			40	18.65	5
n _D 20°0	1.4045 1.4018	2 2	d g/ml	0.242	5	Parachor [P] 20°C		
30	1.3990	4	d g/ml vc ml/g tc °C	4.140 243.	5	30		
"C"	0.7715	5	P _c mm	24391.	5	40 Sugd	2 57.3	5
MR (Obs.		2	PV/RT	213/1.	 	Exp. L.1.%/wt.	31.3	
MR (Calc. (nD-d/2)	28.978 1.0552	5 2	25°C	0.9815	5	u.		
Dielectric			30 mm BP	1.0000 0.9540	5	Dispersion	131.	2
A -6 t	6, 8985	5	t _e	0.9483	5	Flash Point °C Fire Point		
B 1_114°	C 1189.5 225.6	5	t _c	0.264	<u> </u>	M Spec.		-
A* -6 to		5	ΔHc Real/Hi ΔHf	896.78	2	Ultra V. X-Ray Dif.		
B≠ 87 °		5	ΔFf			Infrared		
K			Viscosity centistokes			Solubility in +		
%			η °c			Acetone Carbon tet.		Ì
X '	c		·	1		Benzene		
	č					Ether n-Heptane		
C'			B ^V to			Ethanol		
A'* t B'*			A ^V I C	ļ		Water Water in		
Ac 114 t		-	(B ^V) to	1				
Bc tc	7.3195 C 1492.3	5	(A ^V) °C		$\vdash \vdash$			
Cc -	266.4	5	c _p liq. °K					l
Cryos, A consts, B			c _p vap.300°K	0.36147				
t _e °C	76,82	5	400 c _w vap.	0,46343	2			
$T_{\mathbf{R}} = 0$		_ <u> </u>	II	L	L	+ ===== (100		L
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc from det	de	grams/100 grants ta 5-Calc. by for		<u> </u>
SOURCE:			PI	, de	. ud	J-Care, by for	u.a	
PURIFICA	TION:		PI					
	JRE REFERE							
[

No. 23

										No. 23	
NAME		3-M	ethyl	-tran	s-2-pentene			ST	RUCTURAL	FORMUL	A
									сн ₃ сн ₂ с = с	сн сн.	
		7							ĊH ₃	3	
Mole % Pur.		Ref.		lecul: rmula		Molecular Weight 84.15	6		3		
	_			Ref.		T	Ref.	Ε			Ref.
F.P. °C		-134.84	10	2	dt/dP			f	to		
F.P. 1009	%				°C/mm			g	• <u>K</u>		
B. P. °C			_		25°C BP	0.1468 0.0418	5	h			
760 mm		67.63 14.73		2 4	te	0.0362	5	ſ١	to		
30		-8.48	3	4	30 mm	0.5816	5	g'	<u>*</u> K		
10		-26.09 -55.11		5	ΔHm cal/g			_h'			l
Pressure					ΔHv cal/g			m n	300 to	-0.0144 0.0014	4
mm 25°C	;	158.		5	25°C 30 mm	89.07 94.85	5	0	<u></u>	-0.0658	
t _e	\dashv	915.		5	BP	82.38	5	m'	700 to	0.0644	4
Density g/ml 20°0	c l	0.69	942	2	t _e (d, e)	80.94 80.92	5 5	n'	1000 °K	0.0012	4
dt 25		0.68	398	2	ΔHv/T _e	19.64	5	0'	İ	-0.0 ₆ 42	4
	\dashv	0.68		4	d -8 to	93.46	5		face tension		
a b		0.71 -0.0 ₃		4 4	_e_ <u>74 °C</u>	0.1638	5	dyn	es/cm. 20°C 30	20.22 19.18	5 5
Ref. Index	<u>,</u>			\vdash	d' to				40	18.15	5
n _D 20°0		1.40		2	d _c g/ml	0.241	5	Par	achor [P]		
25		1.39		2 4	v ml/g	4.151	5		20°C 30		
"C"	7	0.77		4	tc °C	238.	5		40		
MR (Obs.	,	29.49		2	P _c mm	24093.	5			257.3	5
MR (Calc.	.)	28.97	78	5	PV/RT 25°C	0.9823	5	Exp	L. l. %/wt.		
(nD-d/2)	-	1.05		4	30 mm	1.0000	5	Dis		131.	2
Dielectric		4 00	9552	5	BP t te	0.9540 0.9486	5		sh Point C		
B 110°C		1179.4	1004	5	t ^e c	0.264	5		e Point		ļ
С		226.		5	ΔHc kcal/m ΔHf	896.78	2		Spec. ra V.		
A* -8 to	2	1.25	5531	5 5	ΔFf	i		X-I	Ray Dif.		
K - 31	ا-'	1103.0			Viscosity				ared		-
t, to	-1				centistokes n °C				ubility in ^T etone		
t _k to					7 °C				rbon tet.		
A' to									nzene her		
B'°	듸				B ^V to		+-1		Heptane		
A'* to				\vdash	B ^V to C				hanol ater		
B'* °C				<u>L</u>	(B ^V) to			Wa	ater in		ļ
Ac 110 to	2	7.31	167	5	(A ^V) °C		Ll				
Bc tc °C	-	1480. 266.		5 5	c _p liq. °K						
Cryos. A					c _p vap.300°K	0.36147	2				
consts. B					400	0.46343					
t _e °C		73.66	5	5	c _v vap.						
$T_R = 0.$	75	Tc						+ g1	ams/100 gra	ms solver	ıt
REFEREN	CI	ES: 1-I	Dow	2-A		Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:				AF							
PURIFICA	TI	ON:		AF	PI						
LITERATU	UR	E REF	ERE	NCES	S:						_
L											

T I										
NAME	4-M	ethyl	cis-	2-pentene			STR	UCTURAL	FORMULA	١.
							c	н,сн сн=с	нсн,	
Mole	Ref.	Mol	ecul		Molecular			с́нз	3	
% Pur.			mul		Weight 84.1	56				
			Ref.			Ref				Ref
F.P. °C	-134.4	30	2	dt/dP	1	1 1	f	to	l	
F.P. 100% B.P. *C	<u>'</u>			*C/mm 25*C	0.1005	5	g	<u>*</u> K	i	
760 mm	56.3	,	2	BP	0.0406	5	h			⊢
100 30	4.9		4 4	t _e 30 mm	0.0361	5	f' g'	to K	j	}
10	-34.6)	5		0.5645	-	h'	. -]	
11	-62.8	2	5	ΔHm cal/g ΔHv cal/g	 	\vdash	m	300 to	-0.0000	4
Pressure mm 25°C	243.		5	25°C	83.81	5	n	_600 •K		
t _e	883.	- 1	5	30 mm BP	91.13 78.78	5	۰	1	-0.0 ₆ 67	
Density				t _e	78.01	5	m' n'	700 to	0.1236 0.0011	
g/ml 20°C	0.6		2 2	t (d, e)	77.99	5	01	11000 1	-0.0 ₆ 36	4
d 25 4 30	0.6		4	ΔHv/T _e	19.64	5	Sunf	ace tension		-
	0.6		4	d -18 to		5		s/cm. 20°C	17.41	5
b	-0.0	392	4	d' to	<u>.</u>		y	30 40	16.38 15.37	5
Ref. Index		380	2	e' i •c		ļ	Para	chor [P]	13.31	Ť
- 25	1.3	349	2	d g/ml	0.234 4.265	5		20°C	ĺ	
30 "C"	1.3		4	d g/ml v ml/g t °C	217.	5		30 40	ł	l
	29.6		4 2	P _c mm	22487.	5			257.3	5
MR (Obs.) MR (Calc.			5	PV/RT	0.07/1	_	Exp.	L. 1. %/wt.		
(nD-d/2)	1.0		4	25°C 30 mm	0.9764 1.0 0 00	5	Disp	u. ersion	126.	2
Dielectric	<u> </u>			BP	0.9540 0.9496	5 5		h Point °C		F
A -18 to		3274	5	t _e t _c	0.264	5	Fire	Point		<u>_</u>
c — A	228.		5	ΔHc kcal/m	897.84	2	M S		ļ	ŀ
A* -18 to		5914	5	ΔHf ΔFf			Ultra X-R	ay Dif.	Ì	ì
B*	1064.5		5	Viscosity	 	\vdash	Infra			_
·	_			centistokes				bility in +	l	l
t _x to			ı	የ ℃	1		Cai	rbon tet.	1	
A' to							Ber Eth	nzene		ļ
B' '	2		1	B ^V to	 	\vdash	n-H	leptane		İ
A'* to	+			AV to	1		Eth Wat	anol ter		
B'* *((B ^V) ₁ to	-{			ter in		L
Ac 94 to		041	5	(A ^V) °C	J.					
Bc tc_C	1427. 267.		5	c _p liq. •K	 					ļ
Cryos. A	1 201.		∸┤	c _p vap300 °K	1	2				
consts. B°				400	0.48125					
t _e °C	61.0		5	c _v vap.	L					L
$T_R = 0.7$								ms/100 gra		t
REFEREN	CES: 1-I	low ?			Calc. from det	t. dat	a 5-	Calc. by for	mula	
SOURCE:			AF							
PURIFICA:		ED 5-	AF							
LITERATU	RE KEF	EKEN	CES	:						

No. 25 4-Methyl-trans-2-pentene NAME STRUCTURAL FORMULA CH3CH CH=CH CH3 Molecular C6H12 Ċнз Mole Ref. Molecular Weight 84.156 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% -140.810 2 dt/dP f to °C/mm ٩K g 25°C 0.1083 5 B. P. °C h ВP 0.0409 4 760 mm 58.55 2 f 0.0361 5 to 100 6.91 4 g' °К 30 -15.78 4 30 mm 0.5677 5 10 -32.98 5 h! AHm cal/g 5 1 -61,28 0.0441 300 to m AHv cal/g Pressure n 600 °K 0.0014 25°C 84.88 5 mm 25°C 224. 5 o 4 -0.0663 30 mm 91.88 5 889. 5 t_e BP 79.41 m' 700 to 0.1393 Density 4 78,60 5 n' g/ml 20°C 1000 °K 0.0011 te (d, e) 0.6686 2 5 78.58 ٥, -0.0636 4 d_4^t 25 0.6638 2 AHv/Te 19.64 5 30 0.6590 4 Surface tension T-10 d 89.23 to 5 0.6879 4 dynes/cm. 20°C 17.37 _7<u>0</u>_ 0.1677 ъ -0.0392 4 ăΠ 30 16.35 5 to 40 15.35 5 Ref. Index e¹ °C 20°C 1.3889 [P] ⁿD Parachor d_c g/ml 0.234 5 25 2 1.3859 20°C vc ml/g t °C 5 4.28 30 1.2831 4 30 t_c 220. 5 40 "C" 0.7768 4 P_c mm 22557. 5 5 Sugd. 257.3 MR (Obs.) 29.76 2 PV/RT Exp. L. l. %/wt. 29.441 MR (Calc.) 5 25°C 0.9814 (nD-d/2)1.0546 5 4 30 mm 2 1.0000 5 Dispersion 128. Dielectric BP 0.9540 Flash Point °C 0.9494 5 A -20 to 6.88584 Fire Point 0,264 5 В _ <u>97 °C</u> 1147.1 5 M. Spec. Ultra V. C 227.9 5 ∆Hc kcal/m 896,84 ΔHf A* -20 to 1.25965 5 X-Ray Dif. ΔFf B*| 70 °C 1073.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to °C tk tx Carbon tet. °C Benzene A۱ to Ether В' °C n-Heptane B_v | C' Ethanol °C Water A'* to Water in B'* °C (B^V) to 97 to Acl (A^V)| 7.3069 5 5 5 °C Βc 1437. t_c°C cp liq. °K Cc 267. c_p vap.300°K Cryos. A° 0.40330 2 consts. B° 0.49788 c_v vap. te °C 63.52 5 $\overline{T_R}$ $= 0.75 T_{c}$ grams/100 grams solvent 2-API REFERENCES: 1-Dow 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 26	
NAME	2-Ethyl-	l -but	ene			STRUCTURAL I	ORMULA	
						CH (CH) C	- CH	
						сн ₃ (сн ₂) с	- CH ₂	
Mole % Pur.		lecul		Molecular Weight 84.15	6	Ŭ,	2**5	
		Ref.			Ref	<u> </u>		Ref.
F.P. °C	-131.530	2	dt/dP	Ī		f to		
F.P. 100%			°C/mm		_	g K		
B, P. °C 760 mm	64,66	2	25°C BP	0.1328 0.0415	5 4	h		ļ
100	12.22	4	t _e	0.0361	5	f' to		ĺ
30 10	-10.84 -28.32	4 5	30 mm	0.5769	5	g' 'K_		ĺ
l i	-57.10	5	ΔHm cal/g			m 300 to	-0.0060	4
Pressure			ΔHv cal/g 25°C	87.83	5	n 600°K	0.0015	4
mm 25°C	177. 907.	5 5	30 mm	93.91	5	° ;	-0.0 ₆ 67	4
Density	+	 	BP te	81.15 80.21	5 5	m' 700 to	0.1283	4
g/ml 20°C		2 2		80.19	5	n' 11000°K	0.0011 -0.0 ₆ 36	4
dt 25 4 30	0.6847 0.6800	4	AHv/T _e	19.65	5		- 650	Ĺ
a	0.7083	4	d -5 to	92.08	5	Surface tension dynes/cm. 20°C	19.66	5
ь	-0.0391	4	-å, - 75 °C to	1	5	30 40	18.57 17.50	5
Ref. Index		2	e' i °C			Parachor [P]	17.50	<u>-</u> -
45	1.3941	2	d g/ml v ml/g	0.240 4.15	5	20°C		İ
30	1.3912	4	tc °C	232.	5	30 40		
MR (Obs.)	0.7680	4	P _c mm	23699.	5		257.3	5
MR (Calc.		5	PV/RT 25°C	0.9844	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0522	2	30 mm	1.0000	5	u. Dispersion	128.	2
Dielectric			BP t _e	0.9540 0.9488	5	Flash Point °C		
A -5 to B 106°C		5	tc	0.263	5	Fire Point		
c	226.7	5	AHc kcal/m	898.18	2	M Spec. Ultra V.		l
A* -5 to B*		5	ΔHf ΔFf			X-Ray Dif.		
к — — -		١	Viscosity			Infrared Solubility in +		-
's	, -		centistokes 7 °C			Acetone		
t _x °C	7		'			Carbon tet. Benzene		
A' to						Ether		
c	-		B ^V to			n-Heptane Ethanol		
A!* to			A ^V I C			Water		
B'* *(-	(B ^V) to]		Water in		
Ac 106 to	7.3153 1466.	5	(A ^V) °C		<u> </u>			
Cc —	267.	5	c _p liq. °K					
Cryos, A° consts, B°			cp vap 300 °K	0.38120 0.48363				
t _e °C	70,34	5	f 400 c, vap.	0.48363	۱'			
$T_{\mathbf{R}} = 0.7$		1 2	<u> </u>	L	<u> </u>	+ grams /100 s===	na aalaa	<u> </u>
	CES: 1-Dow	2-AI	PI 3-Lit. 4-(Calc. from det	da	grams/100 grar ta 5-Calc, by for		<u> </u>
SOURCE:			PI					
PURIFICA	TION:	A	PI				·····	
LITERATU	RE REFERE	NCES	3:					
}								
L		_						

								No. 27	
NAME	2, 3-Dime	thyl-	l-butene			ST	RUCTURAL	FORMUL.	A
							CH ₃		
					\dashv		сн ₃ сн с = 0	CH,	
Mole		ecul	ar C ₆ H ₁₂	Molecular	.		Ċнз	2	
% Pur.	For	mula	612	Weight 84.15	6				
· · · · · · · · · · · · · · · · · · ·		Ref.			Ref.	L			Ref.
F. P. °C	-157.27	2	dt/dP			f	to		
F.P. 100%			°C/mm 25°C	0 0005	_	g	' <u>°K</u>		
B. P. °C	/-	.	BP	0.0985 0.0406	5 4	h			
760 mm 100	55.67 4.41	2 4	te	0.0361	5	f¹	to		
30	-18.11	4	30 mm	0.5633	5	g'	'° <u>K</u>		l
10 1	-35, 17 -63, 25	5	ΔHm cal/g			h'			
	-03.25	3	ΔHv cal/g	 	$\vdash \lnot$	m	300 to	0.0285	4
Pressure mm 25°C	249.	5	25°C	84.74	5	n o	_6 <u>0</u> 0_•K	0.0015	4
t _e	881.	5	30 mm BP	90.92	5 5			-0.0675	
Density			t _e	78.61 77.85	5	m		0.1113	4
g/ml 20°C	0.6779	2	te (d, e)	77.84	5	n' o'	1000 •K	0.0012	4
d ₄ 25	0.6731 0.6682	2 4	ΔHv/T _e	19.64	5			-0.0642	-
a 30		4	d -15 to	87.90	5		face tension	,,,,,,	_ ا
a b	0.6973 -0.0 ₃ 92	4	_e_ _7 <u>0 °C</u>	0.1669	5	dyn	es/cm. 20°C 30	18.35 17.28	5
Ref. Index		Н	d' to e' °C				40	16.23	5
n _D 20°C	1.3904	2		0,238	5	Par	achor [P]		
25 30	1.3874	2	d _c g/ml v _c ml/g	4.20	5		20°C		
	1.3845	4	vc ml/g tc °C	217.	5	l	30 4 0		l
"C"	0.7689	4	P _c mm	22833.	5			257.3	5
MR (Obs.) MR (Calc.)	29.45 29.441	2 5	PV/RT		t	Exp	L.1.%/wt.		
(nD-d/2)	1.0515	2	25°C	0.9919	5	-	u.		
Dielectric			30 mm BP	1.0000 0.9540	5		persion	129.	2
A -15 to	6.88200	5	l t	0.9497	5		sh Point C e Point		
B 94 °C	1136.7	5	t ^e c	0.264	2				
С	228.4	5	ΔHc kcal/m	896.32	2		Spec. ra V.		
A* -15 to	1.25939	5	∆Hf ∆Ff			X-F	Ray Dif.		
B* _80 °C	1063.9	2	Viscosity			Infr	ared		
c			centistokes				ability in T		l
t _k to			η ° ⊂	1			etone rbon tet.		
t _x °C		1					nzene		
B' ℃							he <i>r</i> Heptane		
c' '			B _v to				hanol		
A¹* to			A C				ater		
B'* °C		\sqcup	(B ^V) to				ster in		-
Acl 94 to	7.3038	5	(A ^V)] °C						i
Bc tc °C	1425. 267.	5	c _p liq. °K						
Cryos. A°		┢╧┥		0.40924	2				
consts. B°			c _p vap,300°K 400	0.50620	2				
t _e °C	60.31	5	c _v vap.	1					
$T_{R} = 0.75$	L					+ ",	ams/100 gra	me solven	
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc, from de	t d=				
SOURCE:		AI			-u		22.0. 2, 101		
PURIFICAT	ION:	AI							
	E REFERE								
PLIERVION	L REFERE	ACES);						

_									No. 28	
NAME	3, 3	-Dim	ethyl	-1-butene			STE	UCTURAL	FORMULA	A
1								CH ₃		
		Γ		T				CH ₃ C CH	=CH ₂	
Mole % Pur.	Ref.		lecul rmul		Molecular Weight 84.1	56		Ċн ₃		
			Ref.			Ref	Γ			Ref.
F.P. °C	-115.20	,	2	dt/dP			ſ	to		
F.P. 1009				°C/mm		ا ۔ ا	g	<u>*</u> K		
B. P. *C 760 mm	41.2		,	25°C BP	0.0619 0.0392	5 4	h	· !	<u> </u>	
100	41.24 -8.20		2	t.	0.0363	5	f'	to		
30 10	-29.90		4 5	30 mm	0.5424	5	g'		1	
ı i	-46.38 -73.30		5	∆Hm cal/g	<u> </u>		h'	1 2004	0.0071	
Pressure				ΔHv cal/g	77,42	5	m n	300 to	0.0071	4
mm 25°C	425.9 841.0		5 5	25°C 30 mm	85.87	5	0		-0.0642	4
Density	041.0		,	BP	74.36 73.87	5 5	m'	700 to	-0.0237	4
g/ml 20°C	0.6	529	2	t _e (d, e)	73.86	5	n' o'	1_10 <u>00°K</u>		
dt 25 4 30	0.64		2	AHv/Te	19.58	5	L°.	1	-0.0659	4
1 30	0.64		4	d -30 to		5		ace tension	15.72	_
ь	- 0 . 0		4			5	yne	ss/cm. 20°C 30	14.71	5
Ref. Index				d' to		ll		40	13.71	5
n _D 20°0	1.37		2	d g/ml	0.230	5	Par	chor [P]	j	
30	1.30		4	d g/ml vc ml/g tc °C	4.348 192.	5		30		
"C"	0.77	05	4		21009.	5	j	40 Sugal	257 2	5
MR (Obs.)			2	P _c mm	21007.	اثا	Fyp	. L.1.%/wt.	257.3	13
MR (Calc. (nD-d/2)	29.44		5	25°C	0.9752	5	Exp	u.		
Dielectric	1	.,,	Ť	30 mm BP	1.0000 0.9550	5 5		ersion	124.	2
A 1 -30 to	6.84	1763	5	te	0.9521	5		h Point °C Point		
B ∟ 76 °C	1080.6		5	t _c	0, 265	5	M S		-	
C				ΔHc kcal/m	896.85	2	Ultr	a V.		
A* -30 to B*54 °C		1167	5	ΔFf				ay Dif. ared		
к – – -	1			Viscosity	l			bility in +		-
t _k	, 			centistokes 7 °C	1	ll	Ace	etone		
1 € 1				'				rbon tet. nzene		
A' to							Eth			
č, – - <u>-</u>	<u>- </u>			B ^V to				leptane anol		
A'* to				_A'_ •c	_i	1 1		ter ter in		
B'* *((B ^V) to			Wa	ter in	 	-
Ac 76 to		04	5	(A ^V) °C	<u> </u>		1			
Cc C-c-	268.		5	c _p liq. °K	1]			
Cryos, A ^c consts, B ^c]		c _p vap.300°K 400	0.36112 0.46224					
t _e °C	44.29	,	5	c _w vap.						
$T_R = 0.7$							+ gra	ams/100 gra	ms solven	t
REFEREN	CES: 1-D	ow	2-AI	PI 3-Lit. 4-	Calc. from det	t, dat	a 5-	Calc. by for	mula	
SOURCE:			A	PI						
PURIFICA				PI						
LITERATU	RE REF	EREN	ICES	3:						

							No. 29	
NAME	2,3-Dime	thyl-	2-butene			STRUCTURAL	FORMUL.	A
						CH ₃)	
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 84.15	6	сн ₃ с = с -	сн _{.3}	
	1	Ref.			Ref.			Ref
F.P. °C F.P. 100%	-74.280	2	dt/dP °C/mm 25°C	0.1776	5	f to		
B.P. °C 760 mm 100 30 10	73.21 19.63 -3.95 -21.82	2 4 4 5	t _e 30 mm	0.0424 0.0361 0.5899	5	f' to g'		
1	-51.26	5	ΔHm cal/g	 	-	m 300 to	0.0464	4
Pressure mm 25°C t _e	127. 931.	5 5	AHV cal/g 25°C 30 mm BP	90.14 96.73 83.55	5 5 5	n 600 •K	0.0011 -0.0 ₆ 28	4
Density g/ml 20°C dt 25 d4 30	0.7080 0.7034 0.6988	2 2 4	t _e t _e (d, e) ΔHv/T _e	82.44 82.41 19.65	5 5	m' 700 to n' 1000 K	0.0157 0.0013 -0.0 ₆ 48	4 4
a b	0.7264 -0.0 ₃ 90	4 4	d -10 to e 90 °C d' to	96.06 0.1708	5 5	Surface tension dynes/cm. 20°C 30 40	21.90 20.75 19.63	5 5 5
Ref. Index n _D 20°C 25 30	1.4122 1.4094 1.4065	2 2 4	d g/ml vc ml/g tc°C	0.246 4.06 248,	5 5 5	Parachor [P] 20°C 30 40	1,7.03	
"C"	0.7750	4	P _c mm	24865.	5	11	257.3	5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	29.59 29.441 1.0582	2 5 2	PV/RT 25°C 30 mm BP	0.9725 1.0000	5 5 5	Exp. L.1.%/wt. u. Dispersion	132.	2
A -10 to B 118 °C	6. 90 3 93	5	t t t	0.9540 0.9481 0.264	5	Flash Point C Fire Point		
A* -10 to B* 90 °C	1.25993 1124.0	5 5 5	ΔHc kcal/m ΔHf ΔFf	895.19	2	M. Spec. Ultra V. X-Ray Dif. Infrared		
K c to to C A' to to			Viscosity centistokes 7 °C			Solubility in + Acetone Carbon tet. Benzene		
B' _ °C C' to			B ^V to	_		Ether n-Heptane Ethanol Water Water in		
B'* °C Acl 118 to Bc tc °C Cc	7.3249 1506. 266.	5 5 5	(B ^V) to (A ^V) °C c _p liq. °K			water in		
Cryos. A° consts. B°			c _p vap.300°K 400	0.36397 0.45867	2 2			
t _e °C	79.91	5	c _v vap.			L		<u> </u>
TR = 0.75		2	DI 3 144 4	Cala face		grams/100 gra		t
SOURCE:	ES: 1-Dow	2-A		Calc. from de	t. da	ita 5-Calc. by for	muia	
PURIFICAT	ION:		PI					
	RE REFERE							

NAME	1-H	ept e n								
Ì		cpten					STRUCTURAL FORMULA			
Mole % Pur.	Ref.	Mo	lecul	ar C ₇ H ₁₄	Molecular Weight 98.18	2	сн ₃ (сн ₂) ₄ (:H=CH ₂		
			Ref.			Ref			Ref	
F. P. *C	-119.0	29	2	dt/dP			4 1		T	
F.P. 1007			H	*C/mm	ŀ		f to	1	ļ	
B, P. *C	†		\vdash	25°C	0.3655	5	h .]	Į	
760 mm	93.6		2	BP	0.04447 0.03624	4 5	f' to	+	+-	
100 30	37.4 12.6		2 2	t _e 30 mm	1	5	g' 'K		1	
10	-6.0		2		0.6189	۲	h' i	1	l	
1	- 3 6.9	5	5	ΔHm cal/g	ļ	<u> </u>	m 300 to	0,0251	4	
Pressure	-/-	_		ΔHv cal/g 25°C	87.79	5	n 600 °K	0.0013	4	
mm 25°C	56.3 987.5	5	5	30 mm	89.09	5	0	-0.0 ₆ 48	4	
Density	+ /313		┵┤	BP	76.31	5	m' 700 to	0,1118	4	
Density g/ml 20°C	0.6	9698	2	te t_ (d, e)	74.93 74.87	5	n' 1000 °K	0.0011	4	
at 25	0.6	9267	2	t _e (d, e) ΔHv/T _e		5	o'	-0.0 ₆ 37	4	
⁴ 4 30	0.6	8829	4		19.57		Surface tension		T	
•		1432	4	d 10 to		5	dynes/cm. 20°C	19.52	5	
b	-0.0	3855	4	_d'to	1		30 40	18.55 17.60	5	
Ref. Index		9980	2	e' i *C				17.00	۲-	
n _D 20°C		9713	2	d g/ml vc ml/g	0.235	5	Parachor [P] 20°C		1	
30	1.3	9451	4	t _c mi/g	4. 247 262.	5 5	30		l	
"C"	0.7	648	4	P mm	1	5	40 Suad	206.2	5	
MR (Obs.)	34.1	35	2	P _c mm PV/RT	20813.	1		. 296. 2	-	
MR (Calc.			5	25°C	0.9966	5	Exp. L.1.%/wt.			
(nD-d/2)	1.0	5131	2	30 mm	1,0000	5	Dispersion	118.2	2	
Dielectric	_		Ш	BP	0.9520 0.9441	5	Flash Point °C			
A 10 to		0069	2	te t _c	0.260	5	Fire Point			
B 1_128_*C	219.1		2 2	ΔHc kcal/m	1048,05	2	M Spec.			
A* 10 to		0572	5	ΔHf			Ultra V. X-Ray Dif.	1		
B* 113 °C			5	ΔFf	ļ	Ш	Infrared	1		
c c				Viscosity	ļ		Solubility in +	†		
i _k	-1			centistokes 7 0°C	0.44	2	Acetone			
t⊈i °C	7			20	0.35	2	Carbon tet. Benzene		1	
A' to			\Box	40 80	0.29 0.22	2 2	Ether	1	1	
B', L _ •	4				+ ··	┝╧┤	n-Heptane			
A'* to	+		$\vdash\vdash$	B to A C	j		Ethanol Water			
B'* °((B ^V) to	1		Water in		L	
Ac 128 to	7.3	119	5	(A ^V) °C	ļ					
Bc t *C	1560.		5		 	$\vdash \vdash \vdash$	1			
Cc — —	259.		5	c _p liq. °K						
Cryos. A° consts. B°				c _p vap.300°K 400	0.37960 0.47840	2				
t _e °C	102.7	5	5	c _v vap.					L	
$T_R = 0.7$							† grams/100 gra	ms solven	t	
REFEREN	CES: 1-D)ow	2-AF	PI 3-Lit. 4-(Calc. from det	dat	ta 5-Calc, by for			
SOURCE:			AI	PI						
PURIFICA'	rion:		AF	PI						
LITERATU	RE REF	ERE	CES	:						

		_					No. 31	l
NAME	cis-2-He	ptene)			STRUCTURAL	FORMUL	.A
						CH (CH) CH	-CH CH	
Mole % Pur.	Ref. Mo	ecul mula		Molecular Weight 98.18	2	сн ₃ (сн ₂) ₃ сн		
	T	Ref.			Ref.		т	Ref.
F.P. °C F.P. 100%			dt/dP °C/mm 25°C	0.4366	5	f to g° <u>K</u>		
B. P. °C 760 mm	98.5	2	BP	0.0449	4	h	 	<u> </u>
100 30	41.62 16.53	4	t _e 30 mm	0.0362 0.6280	5	f' to		
10	-2.50	5	ΔHm cal/g	0.0280	+ -	h'	İ	
Pressure	-33.90	5	ΔHv cal/g		$\dagger \neg$	m to		
mm 25°C	45.64 1001.	5 5	25°C 30 mm	90.00 90.19	5	n •K	-	
Density			BP te	77.53 76.09	5	m' to		
g/ml 20°C	0.708 0.704	2 2	'e (a, e)	76.04	5	", <u> </u>	†	
d ^t 25 4 30	0.700	4	d 12 to	19.59	5	Surface tension		\vdash
a b	0.724 -0.038	4	_e <u>115</u> °C	92.74 0.1544	5 5	dynes/cm. 20°C	20.79 19.84	5
Ref. Index			d' to			40	18.91	5
ⁿ D 20°C	1.406 1.403	2	d _c g/ml	0.247	5	Parachor [P] 20°C		
30	1.401	4	vc ml/g tc °C	4.054 272.	5	30		
"C"	0.7640	4	P _c mm	21783.	5	40 Sugd	296.2	5
MR (Obs.) MR (Calc.)	34.0 34.059	2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.052	2	25°C 30 mm	0.9976 1.0000	5	u. Dispersion	122.	2
Dielectric A 12 to	6.02(47	_	BP t _e	0.9515 0.9432	5	Flash Point °C		†
B 135 °C	6.93647	5 5	t _c	0.255	5	Fire Point M. Spec.		₩
C A* 12 to	1.33477	5	∆Hc kcal/m ∆Hf	1046.45	2	Ultra V.		
B* 118 °C	1213, 1	5	ΔFf		<u> </u>	X-Ray Dif. Infrared		
K			Viscosity centistokes			Solubility in +		
t _k to to			η °C			Acetone Carbon tet, Benzene		
A' to B' °C						Ether		
c'			B ^V to C			n-Heptane Ethanol	ļ	
A'* to B'* °C			F.=v.=			Water Water in		
Ac 135 to	7.3513	5	(B') to					
Bc tc °C	1605. 261.	5	c liq. °K					
Cryos, A° consts, B°			c _p vap. °K					
t _e °C	108.18	5	c _w vap.		1			
$T_{R} = 0.75$				<u> </u>	لــــــــــــــــــــــــــــــــــــــ	grams/100 gra	ms solver	nt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da			
SOURCE:		Al						
PURIFICAT		Al						
LILEKATU	RE REFERE	NCES	s :					
L								

						No. 32						
NAME	trans-2-Heptene					STRUCTURAL FORMULA						
			T.		\neg	$CH_3(CH_2)_3CH=CHCH_3$						
Mole % Pur.		lecul		Molecular Weight 98.18	2							
		Ref.			Ref	Re						
F.P. °C	-109.480	2	dt/dP			f to						
F.P. 1009			°C/mm	0 4354	ا ۔ ا	g <u>*K</u>						
B. P. *C 760 mm	07.05		25°C BP	0.4274 0.0449	5 4	h ,						
100	97.95 41.12	2 4	t.	0.0362	5	f' to						
30 10	16.06 -2.96	4 5	30 mm	0.6273	5	g' '°K_						
1	-34.31	5	∆Hm cal/g		Ш	h ¹						
Pressure		 	ΔHv cal/g	00.10	ا ۽ ا	m to r						
mm 25°C	46.55 999.3	5	25°C 30 mm	90.18	5 5	• =-						
t _e Density	799.3	3	BP	77.35	5	m' to						
g/ml 20°C	0.7012	2	te (d, e)	75.91 75.87	5 5	n' •K_						
dt 25 4 30	0.6969	2	ΔHv/T	19.57	5	0'						
	0.6926	4	d 16 to		5	Surface tension						
a b	0.7183 -0.0 ₃ 85	4	<u>• 108 °C</u>	0.1543	5	dynes/cm. 20°C 20.00 5						
Ref. Index			d' to			40 18.08						
ⁿ D 20°C	1.4045 1.4020	2 2	d _c g/ml	0,243	5	Parachor [P]						
30	1.3992	4	A	4,120	5	20°C						
"C"	0.7687	4	n ~	269.	5	40						
MR (Obs.)	34.28	2	P _c mm	21316.	5	Sugd. 296. 2						
MR (Calc. (nD-d/2)) 34.059 1.0539	5 2	25°C	0.9972	5	Exp. L.1.%/wt.						
Dielectric		-	30 mm	1.0000	5	Dispersion 124.						
A 16 to		5	BP t _e	0.9515 0.9433	5 5	Flash Point °C						
B 134 °C		5	tc	0, 255	5	Fire Point						
С	220.	5	ΔHc kcal/m	1045.45	2	M Spec. Ultra V.						
A* 16 to B* 118 °C		5	ΔHf ΔFf	ļ		X-Ray Dif.						
K LIO	- 1210.3	1	Viscosity	<u> </u>	\vdash	Infrared Solubility in +						
t - to	_		centistokes 7 °C			Solubility in +						
5k to		ĺ	7 ℃			Carbon tet.						
A' to		\vdash				Benzene Ether						
B' '	긱	ł	B ^V to	 	\vdash	n-Heptane						
A'* to	, 	 	A C			Ethanol Water						
B'* *(1	(BV) to	1		Water in						
Ac 134 to	7.3482	5	(A ^V) °C									
Bc tc °C	261.	5	c _p liq. °K	1	П							
Cryos, A		Ť	c _p vap. °K									
consts. B			11 -									
t _e °C	107.57	5	c _v vap.									
$T_R = 0.7$						grams/100 grams solvent						
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from de	t. dat	ta 5-Calc. by formula						
SOURCE:			PI									
PURIFICATION: API												
LITERATURE REFERENCES:												

No. 33 cis-3-Heptene NAME STRUCTURAL FORMULA CH3CH2CH=CH(CH2)2CH3 Mole Ref. Molecular Molecular Weight 98,182 C7H14 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g °K 25°C 0.3945 5 B. P. °C h ВP 0.0447 4 760 mm 95.75 2 t_e 0.0362 5 ft to 100 39.21 4 <u>°К</u> g' 30 14.29 4 30 mm 0.6239 5 10 -4,62 5 h! ∆Hm cal/g 5 1 -35.81 to m AHv cal/g Pressure ۰ĸ n 25°C 87.77 5 mm 25°C 52.36 5 0 30 mm 89.36 5 993.0 te 5 BP 5 76.83 m' to Density te (d, e) 75.44 5 n' g/ml 20°C °K 0.7030 2 75.39 5 ۰, d_4^t 25 0.6987 2 AHv/Te 19.58 5 30 0.6944 4 Surface tension 14 91.56 d 5 a 0.7202 4 20,21 dynes/cm, 20°C 105 °C 0.1539 5 ь -0.0385 4 19.22 5 30 ă٦ 5 40 18.26 Ref. Index e١ °C 20°C 1.4059 ^{n}D [P] Parachor d_c g/ml 0.244 5 25 1.4033 2 20°C vc ml/g 4.096 5 30 4 1.4006 30 ^tc 5 266. 40 "C" 0.7692 4 P_c mm 21322. 5 Sugd. 296.2 5 MR (Obs.) 34.30 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 5 25°C 0.9983 5 (nD-d/2) 1.0544 2 30 mm 1.0000 5 Dispersion 122. 2 Dielectric BP 0.9515 5 Flash Point C 0.9435 5 A 14 to 6.93161 1282.3 Fire Point 0.255 B 1 131 °C 5 M. Spec. Ultra V. C 221. 5 AHc kcal/m 1046.45 ΔHf A* 14 to 1.33298 5 X-Ray Dif. ΔFf B*[115°C 1203.1 Infrared ĸ Viscosity Solubility in centistokes Acetone $\mathbf{t}_{\mathbf{k}}$ to Carbon tet. ٠c t^ Benzene A١ to Ether B' °C n-Heptane B_v | C Ethanol °C Water A'* to Water in °C B'* (B^V) to Ac 131 to 7.3464 5 (A^V) °C 1591. 5 Bc_tc_° °C cp liq. °K Cc 261. 5 c_p vap. Cryos. A °K consts. B° c vap. t_e °C 105.09 5 TR $= 0.75 \, T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 34					
NAME	NAME trans-3-Heptene						STRUCTURAL FORMULA					
			1		$\neg \neg$	CH ₃ (CH ₂) ₂ CH=C	нсн ₂ сн ₃	3				
Mole % Pur.	Ref. Mo	olecul	ar C ₇ H ₁₄	Molecular Weight 98.1	82							
		Ref		weight /o.i.	Ref		******	Ref.				
F.P. °C	-136.63	2	dt/dP	T		f to	T	-				
F.P. 1009		Ť	*C/mm	1		g to						
B, P. *C			25°C BP	0.3933 0.0447	5 4	h						
760 mm 100	95.67 39.13	2 4	t	0.0362	5	f' to						
30	14.04	4	30 mm	0.6239	5	g' 'K_						
10 1	-4.70 -35.88	5	ΔHm cal/g			h'						
Pressure	-33.00	+-	ΔHv cal/g			m to						
mm 25°C		5	25°C 30 mm	83.77	5	n •K_						
t _e	993.	5	BP	89. 22 76. 79	5	ļ <u> </u>		 				
Density g/ml 20°0	0.6981	2	te (d.e)	75.40	5	m' to		İ				
t 25	0.6938	2	'e (", ",	75, 37	5	0'		ļ				
4 30	0.6895	4	ΔHv/T _e	19.57	5	Surface tension		\vdash				
a b	0.7152 -0.0 ₃ 85	4	d 10 to		5	dynes/cm. 20°C	19.65	5				
Ref. Index		+-	d' j to	5	}	30 40	18.68 17.79	5				
n _D 20°0	1.4043	2	e' i °(\vdash	Parachor [P]						
25 30	1.4017	2	d g/ml vc ml/g tc °C	İ		20°C						
"C"	1.3989	4	tc °C	266.	5	30 40						
MR (Obs.)	0.7717	+	P _c mm	21116.	5	Sugd.	296.2	5				
MR (Calc.		5	PV/RT	0.0004	5	Exp. L.1.%/wt.						
(nD-d/2)	1.0553	2	25°C 30 mm	0.9984 1.0000	5	u. Dispersion	124.	2				
Dielectric		<u> </u>	BP	0.9515	5	Flash Point °C		├				
A 10 t	6.93075 1281.8	5	te t _c	0.9435 0.255	5	Fire Point						
c	221.	5	ΔHc kcal/m	1045.45	2	M Spec.						
A* 10 to	1.32366	5	ΔHf			Ultra V. X-Ray Dif.						
B* 115°C	2 1200.5	5	ΔFf		\vdash	Infrared						
c	_		Viscosity centistokes		1 1	Solubility in +						
tk to			η •	;		Acetone Carbon tet.						
t _x i °C		₩				Benzene		1				
B'					\sqcup	Ether n-Heptane						
C'		1	B ^V to			Ethanol						
A'* to B'* *(-1		Water Water in						
Ac 131 to		5		1								
Bc t *C	C 1590.	5			\vdash	ļ						
Cc — —	261.	5	c _p liq. ∘K			1						
Cryos. As consts. B			c _p vap. °K									
t _e °C	105.03	5	c _v vap.									
$T_R = 0.7$						grams/100 grai	ns solven	t				
REFEREN	CES: 1-Dow	2-A1	PI 3-Lit. 4-	Calc. from de	t. dat	ta 5-Calc, by for						
SOURCE:		A	PI									
PURIFICATION: API												
LITERATURE REFERENCES:												

No. 35 2-Methyl-1-hexene STRUCTURAL FORMULA NAME $CH_3(CH_2)_3C = CH_2$ Molecular C7H14 Mole Ref. Molecular Weight 98.182 % Pur. Ref. Ref. Ref. F.P. °C F.P. 100% -102.840 2 dt/dP f to °C/mm 25°C g °K 0.3442 B. P. ℃ h ВP 0.0443 4 760 mm 92.00 2 t_e 0.03624 5 ſ١ to 100 35.93 11.22 4 g' <u>°K</u> 30 5 4 30 mm 0.6184 10 -7.52 5 h' ∆Hm cal/g -38.42 5 m to AHv cal/g Pressure n °K 25°C 89.66 5 mm 25°C 60.48 o 30 mm 88,25 5 982.3 5 te BP 75.88 5 m to Density 74.56 5 te te (d, e) °K g/ml 20°C 0.7030 2 74.52 o' 25 0,6986 d_4^t 2 AHV/T 5 19.57 30 0.6942 4 Surface tension to 89.97 5 0.7206 a dynes/cm. 20°C 20, 20 5 1110 0.1531 <u>°C</u> -0.0387 4 ь 19.19 5 30 ăח to 18,20 5 40 Ref. Index e' °C ⁿD 1.4034 [P] 20°C 2 Parachor d_c g/ml 25 1.4007 2 20°C vc ml/g t °C 30 4 1.3979 30 t_c 261. 5 40 "C" 0.7647 4 P_c mm 21115. 5 Sugd. 296.2 5 MR (Obs.) 34.11 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 5 25°C 0.9962 5 (nD-d/2)1.0520 2 30 mm 1.0000 2 5 Dispersion 125. Dielectric ВP 0.9515 0.9438 5 Flash Point C A 5 to B 127 °C t_e 6.92518 5 Fire Point 5 0.255 1268.0 M. Spec. 221.52 5 AHc kcal/m 1044.44 2 Ultra V. ΔHf A* 5 to 1.33076 5 X-Ray Dif. ΔFf B*|110 °C 1189.4 Infrared ĸ Viscosity Solubility in centistokes Acetone to η Carbon tet. $\mathbf{t_{\underline{x}}}$ °C Benzene A' to Ether B' °C n-Heptane Bv | Av | to Ethanol °C A'* Water to Water in B'* °C (B^V) to Ac | 127 to 7.3403 5 (A V) °C $\underline{\mathbf{B}}\mathbf{c} \, \underline{\hspace{1em}} \, \mathbf{t}_{\underline{\mathbf{c}}}$ ۰c 1574. cp liq. ۰ĸ Cc 5 261. Cryos. Aº °K c_p vap. consts. B° c_v vap. t_e °C 100.88 $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc, by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

	IAME 3-Methyl-1-hexene									STRUCTURAL FORMULA						
							Т				C	:н ₃ (сн	2)2CH	CH=C	^{:H} 2	
Mole % Pur.		Ref.		lecul: rmul:		C7H14		Molecul Weight		82			0	3		
				Ref.	Ī				,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	Ref						Ref
F.P. °C	\top			-	dt/	'AD		Γ			f	IT	to			
F.P. 100	-				•	/mm		1			g	! ∟	_ <u>*K</u>			ļ
B. P. *C	1				25 B	S°C			2587 0435	5 4	h					
760 mm 100		84. 28.99		2 4	t _e				0362	5	f'	1	to			
30		4.76		4		mm		0.	6063	5	g'	'_	_ <u>_</u> K_			
10 1		13.61 43.90		5 5	ΔH	m cal	/g			\Box	h'	<u> </u>				L
Pressure	┿	73.70		۲	ΔН	v cal/	g	†			m	(to	ļ		
mm 25°C	1 :	83.37		5		°C	•	83.		5	n o	! ' -	_ <u>"</u> K_	ł		1
t _e	9	60.		5	B) mm P		85. 73.		5		<u> </u>				<u> </u>
Density	.T	0 / 0	-		te			72.	82	5	m' n'	! !	to °K	ľ		
g/ml 20°0	1	0.69 0.69		2 2	1 t _a	(d, e)		72.		5	0,	i .–	,-	1		
dt 25 4 30		0.68		4		Hv/Te		19.		5	S	ace ter	eics.	 		
	1	0.71		4	d		5 to		69 1512	5		s/cm.		19.2	9	5
ь		-0.03	77	4	-å,-	$\frac{1}{1} - \frac{92}{1}$	_ to	1	1314	3	*		30 40	18.4 17.5		5
Ref. Index		1.39	7	2	e'	<u>i </u>	•c	<u> </u>			D			17.5	-	
ⁿ D 25	1	1.39		2	d _c	g/ml			241	5	Par	achor	[P] 20°C			
30		1.39	2	4	t _c	g/ml ml/g *C		249.	142	5			30			1
"C"		0.76	19	4	-	mm		20818.		5			40 Sugd.	296. 2	:	5
MR (Obs. MR (Calc.		34.0 34.05	^	2 5		/RT				\vdash	Exp	L.1.9				-
(nD-d/2)	"	1.04		2	25	°C			9956	5	_	u.				
Dielectric	1			\Box	30 B1	mm P			0000 9520	5		ersion		120.		2
A 5 t	•	6.91	598	4	te			0.	9450	5		h Poin Point	t °C			
B [118°		39.0		4	t _c				260	5	M S				-	\vdash
C	_	23.04		4	ΔH	c kcal, f	/m	1046.	98	2	Ultr	a V.				
A* 5 to	2 111	1.32 61.66		5	ΔF					1 1		ay Dif. ared				
K	-					cosity					<u> </u>	bility i	n +	-		├─
c	-					tistok	°C				Ace	etone	-			
t _x					7		·	ļ		1 1		rbon te nzene	t.			ĺ
A' t				\Box							Eth					
B' '	의			l î	В ^V	T	to	 		+		leptane	:			
A'* t	+-			\vdash	AV	i	.c				Wa	anol ter				
B'*					(B ^V		to	-				ter in				<u> </u>
Ac 118 t	>	7.33	18	5	(A ^V) i	°C									
Bc tc	15	39. 62.		5 5	Сp	liq.	°K			\Box						
Cryos. A	_	·		+	1		•ĸ									
consts. B					, F	vap.	r									
t _e °C		91.92		5	c^	vap.										
$T_R = 0$.												ms/10	0 gran	ns sol	vent	t
REFEREN	CES:	1 - De	w	2-AF		-Lit.	4-0	Calc. fr	om de	t. dat	a 5-	Calc.	by for	mula		
SOURCE:				AI												
PURIFICA				AF												
LITERATU	RE	REFE	ERE	NCES	:											

							No. 37	
NAME	4-Methyl	-1-h	exene			STRUCTURAL	FORMULA	4
			_			сн ₃ сн ₂ сн сн	-CH=CH-	
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 98.18	32	³ ² сн ₃	2 2	
		Ref.			Ref.]	Ref
F.P. °C F.P. 100%	-141.45	2	dt/dP °C/mm			f to		
B. P. °C 760 mm	86.73	2 4	25°C BP t _e	0. 2851 0. 0437 0. 0362	5 4 5	h ft to		
1 0 0 3 0	31.38 6.99	4	30 mm	0.6103	5	g'° <u>K</u>		
10 1	-11.51 -42.00	5 5	ΔHm cal/g		† –	h'		
Pressure mm 25°C	74.69	5	ΔHv cal/g 25°C 30 mm	84.06 86.78	5	m to		
t _e	967.8	5	BP	74.68	5	m' to	 	
Density g/ml 20°C dt 25 d4 30	0.6985 0.6942	2 2	t _e (d, e) ΔHv/T _e	73.46 73.43	5	n' *K_		
	0.6899	4	d 7 to	19.59 87.84	5	Surface tension		
a b	0.7156 -0.0 ₃ 84	4	_e 95 °C d' to		5	dynes/cm. 20°C 8 30 40	19.68 18.71 17.76	5 5 5
Ref. Index n _D 20°C	1.4000	2	d _c g/ml	0, 241	5	Parachor [P]		
25 30	1.397 3 1.3947	2	v_mi/g	4, 152	5	20°C 30		
"C"	0,7635	4	t _c °C	252. 2 0 890.	5	40	296.2	5
MR (Obs.) MR (Calc.)	34.08 34.059	2 5	PV/RT 25°C	0.9945	5	Exp. L.1.%/wt.		
(nD-d/2) Dielectric	1.0508	2	30 mm	1.0000	5	Dispersion	120.	2
A 7 to	6.92084	4	BP t _e	0.9520 0.9448	5	Flash Point C Fire Point		
B _121 °C_ C	1249.38 222.52	4	t _c ΔHc kcal/m	0.260	5	M. Spec. Ultra V.		
A* 7 to B* 105 °C	1.33154 1171.6	5 5	ΔHf ΔFf			X-Ray Dif. Infrared		
c t,to			Viscosity centistokes n °C			Solubility in + Acetone		
t _k to t _x °C			<i>"</i> 7 °C			Carbon tet. Benzene		
B'°C			B _v to	<u> </u>		Ether n-Heptane		
A'* to B'* °C			$\begin{bmatrix} \mathbf{B}^{v} & to \\ \mathbf{A}^{v} & l \\ \mathbf{B}^{v} l \end{bmatrix} = \frac{C}{to}$	-		Ethanol Water Water in		
Ac 121 to Bc tc °C	7.3364 1551,	5	(A ^V) °C					_
Cc — — —	262.	5	c _p liq. °K					
Cryos, A° consts, B°			c _p vap. °K					
t _e °C T _R = 0.75	94.98 T-	5	L		<u> </u>	+ ~~~~ (100		
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	† grams/100 gra		<u>. </u>
SOURCE:		AF		u	,,			
PURIFICAT	ION:	AF						
LITERATUE	RE REFERE	NCES	S:					

							No. 3	8
NAME	5- M e	ethyl-l	-he	xen e			STRUCTURAL FORMUI	
l f							a a (a) a a	
							CH ₃ CH (CH ₂) ₂ CH=CH ₂	
Mole % Pur.	Ref.	Molec	cula		Molecular Weight 98,18	,	ĊН ₃	
			ef.		weight 70.10	Ref		Ref.
F.P. °C	T		~	dt/dP		-	f to	-
F.P. 100%		-	ᅦ	°C/mm			g L K	
B. P. °C				25°C BP	0.2710 0.0436	5 4	h '	1
760 mm 100	85.31 30.13		2 4	t	0.0362	5	f' to	
30	5.82	: .	4	30 mm	0.6083	5	g' 'K_	-
10 1	-12.61		5	ΔHm cal/g			h'	
Pressure	 		\dashv	AHv cal/g			m to	
mm 25°C	79.11		5	25°C 30 mm	83.45 86.35	5 5	0	
Deneitu	963.7		4	BP	74.31	5	m' l to	+
Density g/ml 20°C	0.69	20	2	te te (d, e)	73.11	5	n' °K	
dt 25 4 30	0.68		2 4	ΔHv/T _e	19.58	5	0'	\perp
a 30	0.68		4	d 0 to	87.23	5	Surface tension	_
Ъ	-0.03		4			5	dynes/cm. 20°C 18.96 30 17.99	5
Ref. Index			\neg	d' to			40 17.05	5
n _D 20°C	1.39		2 2	d _c g/ml	0.249	5	Parachor [P] 20°C	1
30	1. 39		4	v mi/g	4,202	5	30	
"C"	0.76	45	4	tc °C	20522.	5	40 Sugd. 296.2	5
MR (Obs.)			2	PV/RT		⊢	Exp. L. l. %/wt.	+-
MR (Calc. (nD-d/2)	34.05 1.05		5 2	25°C	0.9941	5	u.	
Dielectric			ᅱ	30 mm BP	1.0000 0.95 2 0	5 5	Dispersion 120.	2
A 0 to	6.91	781	4	t _e	0.9449	5	Flash Point °C Fire Point	1
B (_119 •0	222.79		4	t _c	0.260	5	M Spec.	+
A* 0 to	+		5	ΔHc kcal/m ΔHf	1046.34	2	Ultra V.	ĺ
B* 100 ℃			5	ΔFf		\sqcup	X-Ray Dif. Infrared	
K	1		- 1	Viscosity centistokes	Ì		Solubility in +	1 -
t _k to				7 °C	1	1 1	Acetone Carbon tet.	İ
<u>'x '</u>			_	•	1	1	Benzene	
A' to		- 1					Ether n-Heptane	Ì
C'			_	B ^V to A ^V °C			Ethanol	
A'* to B'* °C			-				Water Water in	
Ac 119 to		32	5	V.	1			
Bc t °C	1544.	. !	5		ļ	\vdash		
Cc	262.		5	c _p liq. °K				
Cryos, A° consts. B°			1	c _p vap. °K				
t _e °C	93.39	- -	5	c _v vap.				
$T_R = 0.7$	75 T _C		il		L		grams/100 grams solve	nt
REFEREN		ow 2-	ΑP	I 3-Lit. 4-0	Calc, from de	t. dat		
SOURCE:			AP					
PURIFICA?	TION:		AP	I				
LITERATU	RE REFI	ERENC	ES:					
L								

						No. 39	
NAME	2-Methy	-2-h	exene			STRUCTURAL FORMULA	
						CH ₃ (CH ₂) ₂ CH=C CH ₃	
Mole % Pur.		lecul		Molecular Weight 98.18	82	CH ₃	
	<u> </u>	Ref.			Ref.	R	eſ.
F. P. °C	-130,350	2	dt/dP	{		f to	
F.P. 100% B.P. °C	'	├	°C/mm 25°C	0.3897	5	g <u>*K</u> _	
760 mm	95.41	2	ВР	0.0446	4	h	
1 0 0 3 0	38.91	4	t _e	0.0362	5	f' to to	
10	14.01	5	30 mm	0.6234	5	h'	
<u> </u>	-36.05	5	ΔHm cal/g ΔHv cal/g	 	$+$ \dashv	m to	
Pressure mm 25°C	52.60	5	25°C	87.56	5	$\begin{bmatrix} \mathbf{n} & \mathbf{-} & -\frac{\mathbf{e}\mathbf{K}}{2} \end{bmatrix}$	
t _e	991.4	5	30 mm BP	89.26 76.71	5 5		
Density			t_	75.32	5	m' to K	
g/ml 20°C	0.7082 0.7038	2 2	'e (d, e)	75.27	5	"	
d ₄ 25	0.6994	4	ΔHv/T _e	19.57	5	Surface tension	_
a	0.7257	4	d 14 to e 110 °C	91.42 0.1543	5	dynes/cm. 20°C 20.81	5
b	-0.0386	4	-d to	0.1313			5 5
Ref. Index		2	e'	-		Parachor [P]	_
- 25	1.4079	2	d g/ml v ml/g	0.2 4 2 4.13	5	20°C	
"C"	1.4052	4	v _c ml/g t _c °C	266.	5	30 40	
MR (Obs.)	0.7719 34.39	2	P _c mm	21467.	5		5
MR (Calc.		5	PV/RT 25°C	0.0073	_	Exp. L.1.%/wt.	
(nD-d/2)	1.0565	2	30 mm	0.9972	5	Dispersion 127.	2
Dielectric		<u> </u>	BP	0.9510 0.9430	5	Flash Point °C	
A 14 to B 132 °C	6. 93104 1281, 02	4	t _e t _c	0.259	5	Fire Point	
C	220.87	4	∆Hc kcal/m	1043.04	2	M. Spec. Ultra V.	
A* 14 to B* 110 °C	1.33367	5	ΔHf ΔFf	i		X-Ray Dif.	
K 110 0	1202.03	2	Viscosity			Infrared	
t ₁ to	-		centistokes	į		Solubility in * Acetone	
t _k to t _x °C			η °c			Carbon tet. Benzene	
A' to						Ether	
B'°C	-		B _v to	1	T	n-Heptane Ethanol	
A'* to		\vdash	A I °C			Water	
B¹* °C			(B ^V) to		1 1	Water in	
Acl 132 to	7.3460 1590.	5	(A ^V) °C		$oxed{oxed}$		
Bc t _c °C	261.	5	c _p liq. °K				
Cryos. A° consts. B°			c _p vap. °K				
t _e ℃	104.69	5	c _v vap.				
$T_R = 0.7$	1		U	 -	т	grams/100 grams solvent	
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da		
SOURCE:		AF	PI				
PURIFICAT	TION:	AF	PI				
LITERATU	RE REFERE	NCE	S:				-

							No. 40
NAME	3-Methyl	-cis-	2-hexene			STRUCTURAL F	ORMULA
						CH ₃ (CH ₂) ₂ C =	сиси.
Mole	Ref. Mo	11		Malasslan		сн.	,
% Pur.	Kei. Mo	rmul	ar C ₇ H ₁₄	Molecular Weight 98.18	82		,
		Ref.			Ref		Re
F.P. *C			dt/dP			f to	
F. P. 1007	<u> </u>	<u> </u>	*C/mm 25*C	0.3700	5	g <u>°K</u> _	
B. P. *C 760 mm	94.	2	BP	0.0445	4	h	
100 30	37.68 12.85	4	t _e 30 mm	0.0363 0.6214	5 5	f' to to	
10	-5.98	5	ΔHm cal/g	0.0214	۲	h'	
1	-37.04	5	ΔHv cal/g		\vdash	m to	
Pressure mm 25°C	55.75	5	25°C	86.96	5	n •K	
t _e	987.4	5	30 mm BP	88.84 76.35	5 5	 	
Density	0.7136		te te (d, e)	74.99	5	m' to	
g/ml 20°0 dt 25 d4 30	0.7120 0.7080	2 2	t _e (d, e)	74.95	5	0	
⁴ 30	0.7039	4	ΔHv/T _e	19.57	5	Surface tension	
a b	0.7282 -0.0 ₃ 80	4	e 115 °C		5	dynes/cm. 20°C	21.26 5
Ref. Index		+	d' to			30 40	20.29 5 19.34 5
n _D 20°0	1.410	2	d _c g/ml	0,245	5	Parachor [P]	
25 30	1.407 1.405	2	V mi/g	4.08	5	20°C	
"C"	0.7667	4	1 -	266.	5	40	
MR (Obs.)	34.2	2	P _c mm PV/RT	21745.	5	Sugd.	296.2 5
MR (Calc. (nD-d/2)) 34.059 1.054	5 2	25°C	0.9967	5	Exp. L.1.%/wt. u.	
Dielectric		+-	30 mm BP	1.0000 0.9510	5	Dispersion	127. 2
A 5 t		4	te	0.9431	5	Flash Point °C Fire Point	
B [131 °C	221.14	4	t _{c.}	0.259	5	M Spec.	
A* 5 to		5	ΔHc kcal/m ΔHf	1043.68	2	Ultra V.	
B* 115 °C		5	ΔFf			X-Ray Dif. Infrared	
K — — —	1		Viscosity centistokes	l		Solubility in +	
ւել			η °c			Acetone Carbon tet.	
t _x i °C		├		1		Benzene	
B'					\sqcup	Ether n-Heptane	
C'	_		B ^V to A ^V °C			Ethanol Water	
A'* to B'* °C			(B ^V) to			Water in	
Ac 131 to	7.3437	5	(A ^V) °C				
Bc tc_°C	1585.	5	c _p liq. °K	 	\vdash		
Cryos. A	202.	1,	-				
consts. B			р				
t _e °C	103.11	5	c _v vap.				
$T_{\mathbf{R}} = 0.7$						+ grams/100 gran	ns solvent
	CES: 1-Dew	2-AI		alc. from det	t. dat	ta 5-Calc. by form	nula
SOURCE:		AI	PI				
PURIFICA		AI				· · · · · · · · · · · · · · · · · · ·	
LITERATU	RE REFERE	NCES	:				

No. 41 3-Methyl-trans-2-hexene NAME STRUCTURAL FORMULA $CH_3(CH_2)_2C = CHCH_3$ Molecular C7H14 ĊН3 Mole Ref. Molecular Weight 98.182 % Pur Formula Ref. Ref. Ref. <u>F.P.</u> ℃ dt/dP f to F.P. 100% °C/mm 25°C g °K 0.3700 5 B. P. °C h BP 0.0445 4 760 mm 94. 37.68 2 0.0363 5 ſ١ 100 to 4 °K g' 30 12.85 4 30 mm 5 0.6214 10 -5.98 5 h ∆Hm cal/g -37.04 5 m to AHv cal/g Pressure °K n 25°C 86.96 5 mm 25°C 55.75 5 ٥ 30 mm 88.84 5 987.4 5 t_e BP 5 76.35 m' to Density 74.99 5 n' g/ml 20°C te (d, e) °K 0.7120 2 74.95 5 ۰, 25 0.7080 2 d_4^t AHv/T 19.57 5 30 0.7039 4 Surface tension 5 90,81 to 5 0.7282 4 dynes/cm. 20°C 5 21.26 1115_ °C 0.1539 5 ь -0.0₃80 4 .30 ď 20, 29 5 to 40 19.34 5 e' Ref. Index °C 1.410 20°C nD [P] Parachor d_c g/ml 0.245 5 25 1.407 ž 20°C vc ml/g 5 30 4.08 1.405 4 30 t_c 266. 5 40 "C" 0.7667 4 P_c mm 21745. 5 Sugd. 296.2 5 MR (Obs.) 34.2 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 5 2 25°C 0.9967 5 (nD-d/2)u. 1.054 i. óóóó 30 mm 5 Dispersion 127. 2 Dielectric BP 0.9510 Flash Point C 0.9431 5 5 **to** 6.9**28**21 1**27**5.5**0** Fire Point 0.259 5 B 131 °C M. Spec. Ultra V. 221.14 С 4 AHc kcal/m 1043,68 ΔHf A*[5 **to** 1.33242 5 X-Ray Dif. ΔFf B* 115 °C 1196.77 Infrared ĸ Viscosity Solubility in centistokes Acetone to °C Carbon tet. °C ŧÇ. Benzene to Ether B١ <u>•с</u> n-Heptane Bv | Av | C' to Ethanol °C Water A'* to ٠c Water in B'* (B^V)| to Ac 131 to 7.3437 5 (A^V) °C Bc_t_ 1585. 5 c_p liq. ۰ĸ Cc 5 262. Cryos, A cp vap. °K consts. B° te °C c vap. 103.11 5 $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc, by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

							No. 42	
NAME	4-Methyl	-cis-	-2-hexene			STRUCTURAL	FORMULA	
						сн ₃ сн ₂ сн сн	=CHCH	
						Ċн ₃	3	
Mole % Pur.	Ref. Mo	rmul	ar C ₇ H ₁₄	Molecular Weight 98.18	32	3		
		Ref.			Ref			Ref
F.P. °C			dt/dP			f to		
F.P. 1009	6		*C/mm 25*C	0,2900	5	g <u> </u>	İ	l
B, P, *C 760 mm	87.37	2	BP	0.0440	4	h		<u> </u>
100	31.74	4	t _e	0.0364	5	f' to	İ	ļ
30 10	7.27 -11.3	4 5	30 mm	0.6122	5	h'		
1	-41.8	5	ΔHm cal/g ΔHv cal/g	 	 	m to		
Pressure mm 25°C	73.49	5	25°C	84.01	5	n ' <u>*K</u> _		
t _e	969.6	5	30 mm BP	86.69 74.50	5	<u>°</u>		
Density			,	73.26	5	m' to		
g/ml 20°0	0.6996	2 2	t _e (d, e)	73.23	5	", ' <u>-</u> "-		
dt 25 4 30	0.6910	4	ΔHv/T _e	19.50	5	Surface tension		
a b	0.7168	4	_e _l _l10 °C	0.1521	5	dynes/cm. 20°C	19.81 18.83	5 5
Ref. Index	-0.0385	-	d' to	l .		40	17.87	5
n _D 20°0	1.4024	2		0,238	5	Parachor [P]		
25 30	1.3997	2	d g/ml vc ml/g tc °C	4.20	5	20°C		l
"C"	0.7667	4	,, ~	254.	5	40	206.2	_
MR (Obs.	34.20	2	P _c mm	20630.	-	Exp. L.1.%/wt.	296. 2	5
MR (Calc. (nD-d/2)) 34.059 1.0526	5 2	25°C	0.9948	5	u.		
Dielectric		<u> </u>	30 mm BP	1.0000	5	Dispersion	122.	2
A 0 t	6.89863	4	te	0.9446	5	Flash Point °C Fire Point		
B 1_122 •9	222.	4	t _c	0.260	5	M Spec.		
A* 0 to		5	ΔHf	1045.36	-	Ultra V. X-Ray Dif.	ļ	
B* 110°		5	ΔFf		ļ	Infrared		
K — —		1	Viscosity centistokes		Ì	Solubility in +		
tk to			η •c		ļ	Acetone Carbon tet.		
t i		-				Benzene Ether		İ
B' •			B ^V l to		├	n-Heptane		
A!* to			B' to A' °C			Ethanol Water		
B'* °			(B ^V) to	-	l	Water in		<u> </u>
Ac 122 to		5	(A ^V) °C					
Bc tc_°	261.	5	c _p liq. °K					
Cryos. A		Ė	c _p vap. °K					
consts. B		<u> </u>	łi [–]		1			
te °C	95.74	5	c _v vap.		L	<u> </u>	<u> </u>	<u> </u>
T _R = 0.7			DI 2 II			grams/100 gram	ns solven	t
SOURCE:	CE3: 1-DOW			aic, from de	t. da	ta 5-Calc. by for	mula	
PURIFICA	TION:		PI PI					
	RE REFERE							
			•					

No. 43 4-Methyl-trans-2-hexene NAME STRUCTURAL FORMULA CH3CH=CH CH CH2CH3 Molecular C7H14 ĊНз Mole Ref. Molecular Weight 98.182 % Pur Formula Ref. Ref Ref. F.P. °C -126.5 dt/dP f to F.P. 100% °C/mm °K g 25°C 0.2920 5 B. P. ℃ h ВP 4 0.0441 87.6 760 mm 2 0.0364 5 f١ to 100 31.9 4 °<u>K</u> g' 30 7.40 4 30 mm 0.6129 5 10 -11.2 5 h! ∆Hm cal/g -41.7 5 m to ∆Hv cal/g Pressure °K n 25°C 83.92 5 mm 25°C 72.99 5 0 30 mm 86.67 5 970.3 te 5 BP 5 74.49 m' to Density 73.25 te te (d, e) n' g/ml 20°C <u>°K</u> 0.6975 2 5 73.21 ٥' d_4^t 25 0.6932 2 ΔHv/T_e 5 19.48 30 0.6889 4 Surface tension 87,80 5 to a 0.7146 4 dynes/cm. 20°C 19.57 5 9<u>6</u> <u>°C</u> 0.1519 5 ь -0.0384 4 30 18.61 5 d٠٦ to 5 40 17.66 Ref. Index e١ °C 20°C 1.4023 [P] 2 nD dcg/ml vcml/g tc°C Parachor 5 0.236 25 1.3997 2 20°C 5 4.244 30 1.3970 4 30 $\mathbf{t}_{\mathbf{c}}$ 254. 5 40 "C" 0.7688 4 P_c mm 20512. 5 296.2 5 Sugd. MR (Obs.) 34.30 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 25°C 0.9943 5 (nD-d/2)u. 1.0536 2 30 mm 1,0000 Dispersion 124. 2 5 5 Dielectric ВP 0.9520 Flash Point °C 0.9446 5 7 to 6.89566 Fire Point 0.260 5 122 °C 1243.0 M. Spec. Ultra V. С 222. 4 AHc kcal/m 1044.38 2 ΔHf A*| 7 to 1.30573 X-Ray Dif. ΔFf B*|_106 °C 1165, 24 Infrared ĸ Viscosity Solubility in centistokes Acetone Carbon tet. °C t_x Benzene A' to Ether В' °C n-Heptane B_v | C' to Ethanol °C Water A1* to Water in B'* °C (B^V) to Acl 122 to 7.3106 5 (AV) °C Bc tc °C 1545. cp liq. ۰ĸ Cc 261. 5 Cryos. A° c_p vap. °K consts. B° c_w vap. t_e °C 96.01 $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

									No. 44	
NAME	5 - M	ethyl-ci	s-2-hexene				STR	UCTURAL		
Mole	Ref.			Molecu			CI	н ₃ сн сн ₂ сн сн ₃	=CH CH ₃	
% Pur.		Form		Weight	98.18					5 (
		Re	ef.			Ref			1	Ref.
F.P. *C F.P. 100%	<u> </u>	-+	dt/dP				f	to		
B. P. *C	 	-+	*C/mm 25*C	0	. 3119	5	g	i		
760 mm	91.		BP		. 0462	4 5	h f'	 		+
100 30	33.0 7.77		t _e	1	. 0379	5	g'	to *K	1	
10	-11.3	5	5 ATT 1/2		. 6301	-	h'	 i	1	
1	-42.5					\vdash	m	l to	ļ	+
Pressure	70.05	. .	ΔHv cal/g		. 01	5	n	<u>*</u> K_		
mm 25°C	70.05	, 5	30 mm	84	. 52	5	0	i		İ
Density	1		BP		. 46 . 17	5	m'	l to		
g/ml 20°C					. 13	5	n' o'	<u>*</u> K	ĺ	
dt 25 4 30	0.69		4 LT/T	18	. 71	5	٥.	<u> </u>		<u> </u>
a 30	0. 71		—	to 85	. 65	5		ace tension	10.05	١,
ь	-0.03			. <u>'</u> ⊆∣ ∘	. 1449	5	gyne	s/cm. 20°C 30	19.85 18.94	5
Ref. Index			- 	to °C		i i		40	18.04	5
n _D 20°C				0	. 215	5	Para	chor [P]		1
25 30	1.39		II v mi/g	4	. 64	5		20°C 30		
"C"	0.76		⊣ եւ Վ	260		5		40		
MR (Obs.)	34.0	2	P _c mm	18614	•	5			296.2	5
MR (Calc.)	34.05	9 5	PV/RT		. 9956	5	Exp.	L.1.%/wt.		
(nD-d/2)	1.05	0 Z	30 mm	1	.0000	5	Disp	u. ersion	122.	2
Dielectric	<u> </u>		BP		.9520 .9441	5	Flas	h Point °C		<u> </u>
A 0 to B 1127 °C					. 255	5	Fire	Point		
c Living	1212.86	' 4		m 1044	. 74	2	M S			
A* 0 to	1,16	082 5	ΔHf		•		Ultr.	a V. ay Dif.		
B* 110 °C	1133.96	, 5				\vdash	Infra			<u> </u>
С	}		Viscosity centistoke	.				bility in +		
t _k to]		7	°C				tone rbon tet.		1
x I			」 `				Ber	nzene		
A' to		-					Eth	er Ieptane		
c, – – –	-	į	B _v	to				anol		
A'* to			Av i	•c			Wa	ter ter in		
B'* °C			(B ^V)	to			W &	ter III		+
Ac 127 to Bc t _c *C	7.17 1522.	33 5		•c		Ш			1	-
Cc	264.	5		•к		1 1				1
Cryos. A°			c _p vap.	•ĸ						1
consts. B°	ļ		11 -							
t _e °C	100.21	5	c _v vap.	L						<u></u>
$T_{R} = 0.7$							+ gra	ms/100 gran	ms solver	nt
REFERENC	ES: 1-D	ow 2-	API 3-Lit.	4-Calc. f	rom de	t. dat	a 5-	Calc. by for	mula	
SOURCE:			API							
PURIFICAT			API							
LITERATU	RE REF	ERENC	ES:							

							No. 45			
NAME	5-Methy	l-trai	ns-2-hexene			STRUCTURAL FORMULA				
 						сн ₃ сн сн ₂ сн	I=CH CH ₃			
Mole % Pur.		lecul		Molecular Weight 98,18	2	Ċн ₃				
		Ref.			Ref.		Ref			
F.P. °C F.P. 100%			dt/dP °C/mm			f to g° <u>K</u>				
B. P. ℃	04		25°C BP	0.2618 0.0458	5 4	h				
760 mm 100	86. 28.55	2 4	t _e	0.0381	5	f' to				
30 10	3.58 -15.2	4 5	30 mm	0.6226	5_	g' <u>K</u>				
1	-46.0	5	AHm cal/g		<u> </u>	h' i	 			
Pressure			ΔHv cal/g 25°C	79.92	5	n				
mm 25°C	85.52 966.5	5	30 mm BP	83.02 71.01	5	0				
Density			t _e (d, e)	69.81	5	m' to				
g/ml 20°C	0.700	2		69.76	5	ō'	1			
dt 25 4 30	0.692	4	ΔHv/T _e	18.63	5	Surface tension				
a b	0.7161 -0.0 ₃ 79	4 4	d 0 to e 110 °C d' to	83.54 0.1456	5	dynes/cm. 20°C 8 30	19.85 5 18.93 5			
Ref. Index		2	e'		<u> </u>	Parachor [P]	18.02 5			
25	1.397	2	d _c g/ml	0.211 4.74	5	Parachor [P] 20°C				
30	1. 395	4	vc ml/g tc °C	253.	5	30 40				
"C"	0.7619	4	P _c mm	18186.	5		. 296.2 5			
MR (Obs.) MR (Calc.		5	PV/RT	0.0043	_	Exp. L.1.%/wt.				
(nD-d/2)	1.050	2	25°C 30 mm	0.9942 1.0000	5	u. Dispersion	124. 2			
Dielectric		ļ	BP	0.9520 0.9445	5	Flash Point °C				
A 0 to B (121°C	1183.25	4	t e t c	0.258	5	Fire Point M. Spec.				
A* 0 to	1,13430	5	ΔHc kcal/m ΔHf	1043.74	2	Ultra V.				
B*[110°C		5	ΔFf			X-Ray Dif. Infrared				
c	_	İ	Viscosity centistokes		1	Solubility in + Acetone				
t _k to			η °C			Carbon tet.				
A' to		 		}		Benzene Ether				
B'°			B ^V to		 	n-Heptane				
A¹* to	+	+	B ^V to C	1	l	Ethanol Water				
B'* °C		L	(B ^V) to	1		Water in	 			
Acl 121 to	7.1390	5	(A ^V) °C							
Bc t _c °C	1486. 263.	5 5	c _p liq. °K							
Cryos, A° consts, B°			c _p vap. °K							
te °C	94.60	5	c _v vap.		L					
$T_{\mathbf{R}} = 0.7$						† grams/100 gra				
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	rmula			
SOURCE:		AF								
PURIFICAT		AI								
LILEKATU	RE REFERE	NCE	:							

							No. 46	
NAME	2-Me	thyl-cis-	3-hexene			STRUCTURAL	FORMUL	A.
						сн ₃ сн сн=сн	сн.сн.	
Mole	1,			M-1		Ċн ₃	22	
Mole % Pur.	Ref.	Formul	ar C ₇ H ₁₄	Molecular Weight 98.1	82	3		
		Ref.			Ref			Ref
F.P. C			dt/dP			f to		T
F.P. 100%			*C/mm 25*C	0.2410	ا ۔ ا	g <u>"K</u>		
B. P. °C 760 mm	86.	2	BP	0.2618 0.0458	5 4	h		-
100	28.6	4	t _e	0.0381	5	f' to g'°K		ì
30 10	3.58 -15.2	4 5	30 mm	0.6226	5	p ₁	1	1
1	-46.0	5	ΔHm cal/g	 	\vdash	m to		+-
Pressure	05.53		ΔHv cal/g 25°C	79.89	5	n •K]	
mm 25°C	85.52 966.5	5 5	30 mm	83.02	5	<u> </u>		
Density	 		BP te	71.01 69.80	5 5	m' to		
g/ml 20°C	0.694		t _e (d, e)	69.76	5	n' <u>*K</u> _	1	
dt 25 4 30	0.690		ΔHv/T _e	18.63	5	<u> </u>		+
	0.710		d -5 to		5	Surface tension dynes/cm. 20°C	19.17	5
ь	-0.037		-a, -110- *		"	30	18.28	5
Ref. Index	1 200		e' i °C	1		40	17.40	5
ⁿ D 20°C	1.399		d _c g/ml			Parachor [P] 20°C	Į	
30	1.394		d g/ml vc ml/g tc °C	252.	5	30		1
"C"	0.766	-	Pcmm	17992.	5	40 Sugd	296.2	5
MR (Obs.) MR (Calc.)	34.2 34.059	2 5	PV/RT	 	1	Exp. L.1.%/wt.		†
(nD-d/2)	1.052		25°C 30 mm	0.9938	5	u.	122.	2
Dielectric			BP	0.9520	5	Dispersion Flash Point °C	122.	+-
A -5 to	6.722		t _e	0.9445	5	Fire Point		
B 1_121 °C	222.	4 4	t _c	1044.74	2	M Spec.		
A* -5 to	1,134	30 5	ΔHf		-	Ultra V. X-Ray Dif.		İ
B*	1105.83	5	ΔFf	 	├	Infrared		
c			Viscosity centistokes			Solubility in +		
tk to			η •α			Acetone Carbon tet.		1
t _x i °C		-+		1		Benzene	1	
B' ∟ °C	ļ			_		Ether n-Heptane		
C'		-	B ^V to			Ethanol Water		1
A'* to B'* °C	1			-		Water in		
Ac! 121 to	7,138	8 5	(A ^V) to	1				
Bc t C	1486.	5	c _p liq. °K		+-			
Ce	263.	5						
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	94.60	5	c _w vap.				<u> </u>	
$T_{R} = 0.75$						grams/100 gra		ı t
REFERENC	ES: 1-Do			Calc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE: PURIFICAT	ION.	AI AI						
LITERATUE								
			•					

No. 47 2-Methyl-trans-3-hexene NAME STRUCTURAL FORMULA CH3CH CH=CHCH2CH3 Ċн₃ Mole Ref. Molecular Molecular Weight 98,182 C7H14 % Pur. Formula Ref. Ref. Ref. F.P. °C dt/dP f. to F.P. 100% °C/mm °K g 25°C 0.2618 5 B. P. °C h ВP 0.0458 4 760 mm 86. 2 0.0381 5 f١ to 28.6 100 4 <u>°к</u> g' 30 3.58 4 0.6226 5 30 mm 10 -15.2 5 h! ∆Hm cal/g 5 1 -46.0m to AHv cal/g Pressure n °K 25°C 79.89 5 mm 25°C 85.52 5 o 30 mm 83.02 5 966.5 5 te 5 BP 71.01 m' to Density t_e (d, e) 69.80 5 g/ml 20°C n' °K 0.694 2 69.76 5 ٥' d_4^t 25 0.690 2 ΔHv/T_e 18.63 5 0.686 4 Surface tension 83.54 5 a 0.710 4 19.17 5 dynes/cm. 20°C 110 °C 0.1457 5 -0.0379 h 4 18.28 5 30 to 40 17.40 5 Ref. Index e' °C 20°C 1.399 2 [P] n_D Parachor dc g/ml 25 1.396 2 20°C vc ml/g t_ °C 30 1.394 4 30 252. 5 tc 40 "C" 0.7667 4 P_c mm 17992. 5 Sugd. 296.2 5 MR (Obs.) 34.2 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.059 5 25°C 0.9938 5 u. (nD-d/2) 1.052 2 30 mm 1.0000 5 Dispersion 2 124. Dielectric BP 0.9520 Flash Point C 0.9445 5 A -5 to 6.72252 Fire Point B 121 °C 1183.25 M. Spec. С 222. 4 AHc kcal/m 1044.74 Ultra V. ΔHf A* -5 to 1.13430 5 X-Ray Dif. ΔFf B* 110 °C 1105.83 Infra red K Viscosity Solubility in c centistokes Acetone $\mathbf{t_k}$ to Carbon tet. °C Benzene A to Ether B °C n-Heptane B^V A C' to Ethanol °C Water A'* to Water in (B^V)I R'* °C to Ac| 121 to 7.1388 (A^V)| 5 °C Bc tc °C 1486. 5 c_p liq. ۰ĸ 5 263. Cryos, A° c_p vap. °K consts. B° c_v vap. te °C 94.60 5 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 48	
NAME	3-Methyl-	cis-	3-hexene			STRUCTURAL	FORMULA	1
						CH ₃ CH ₂ CH=C	сн. сн.	
						сн		
Mole % Pur.	Ref. Mo	rmul	ar C ₇ H ₁₄	Molecular Weight 98.18	32		3	
		Ref			Ref		**************************************	Ref
F.P. °C			dt/dP	T		f to		
F.P. 100%			°C/mm			g K		1
B. P. *C			25°C BP	0.3640 0.0465	5 4	h ,		
760 mm 100	95.35 36.9	2 4	t.	0.0378	5	f' to		
30	11.46	4	30 mm	0.6358	5	g' ' <u>°</u> K_		
10 1	-7.8 -39.3	5	ΔHm cal/g			h'		L
Pressure	37.3	H	ΔHv cal/g			m to		
mm 25°C	58.69	5	25°C 30 mm	83.98	5	n ! ' *K-		
t _e	993.	5	BP	85.99 73.66	5		<u></u>	├-
Density g/ml 20°C	0.7132	2	te (d. e)	72.28	5	m' to		
dt 25	0.7089	2	t _e (d, e) ΔHv/T _e	72.23 18.76	5	0'		1
4 30	0.7046	4	d 0 to	87.67	5	Surface tension		一
a b	0.7303 -0.0 ₃ 84	4 4	e 115 °C		5	dynes/cm. 20°C	21.40	5
Ref. Index	-0.0304	1	d' to			30 40	20.38 19.37	5
n _D 20°C	1.4123	2			-	Parachor [P]		
25 30	1.4096 1.4070	2 4	d g/ml v ml/g			20°C		
"C"	0.7695	4	t _c *C	268.	5	30 40		
MR (Obs.)	34.28	2	P _c mm	19203.	5	Sugd.	296.2	5
MR (Calc.)	34.059	5	PV/RT 25°C	0.00(8	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0557	2	30 mm	0.9968	5	u. Dispersion	127.	2
Dielectric			BP	0.9515	5	Flash Point °C		
A 0 to B 133 °C	6.77018	4 4	t _e	0.9432	ا ا	Fire Point		
<u>c</u>	221.	4	ΔHc kcal/m	1043.68	2	M Spec. Ultra V.		
A* 0 to	1.17140	5	ΔHf ΔFf		1	X-Ray Dif.		
B* ∟115 °C	1151.0	5	Viscosity	 	\vdash	Infrared		<u> </u>
·			centistokes			Solubility in + Acetone		
tk to			η ∘c			Carbon tet.		
A' to						Benzene Ether		
B' <u>*C</u>			B ^v to	<u> </u>	-	n-Heptane		Ì
			A ^V to			Ethanol Water		
A'* to B'* *C			(B ^V) to	•		Water in		
Ac 133 to	7.1872	5	(A ^V) °C					
Bc tc_C	1543. 263.	5	c _p liq. °K	<u> </u>				
Cryos. A°	203.	-	-					
consts. B°			P					
t _e °C	105.07	5	c _v vap.	}				
$T_{\mathbf{R}} = 0.75$						grams/100 gran	ns solven	t
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-(Calc, from de	t. da			
SOURCE:		Al	PI					
PURIFICAT			PI					
LITERATUE	RE REFERE	NCES	i:					

No. 49 3-Methyl-trans-3-hexene STRUCTURAL FORMULA NAME CH3CH2CH=C CH2CH3 ĊН₃ Ref. Mole Molecular Molecular C_7H_{14} Weight 98.182 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g _°K 25°C 0.3413 5 B. P. °C h BP 0.0464 760 mm 93.55 2 0.0379 5 ſ١ to 100 35.3 4 g' <u>°К</u> 30 9.95 4 30 mm 0.6330 5 10 -9.2 5 h' ∆Hm cal/g 5 40.5 to m AHv cal/g Pressure °K n 25°C 83,22 5 mm 25°C 63.13 5 o 30 mm 85.44 987.5 5 t_e BP 73.14 5 m+ to Density 71.79 5 te te (d, e) n' °K g/ml 20°C 0.7099 2 71.74 ٥' d₄ 25 0.7056 2 AHv/Te 5 18,73 30 0.7013 4 Surface tension ď 0 86.91 5 to а 0.7271 4 dynes/cm. 20°C 21.01 ᇷᅱ 110 °C 0.1472 5 -0.0385 19.99 19.00 Ъ 4 5 30 5 40 Ref. Index e' 20°C 1.4107 2 ⁿD [P] Parachor d_c g/ml 25 1.4080 2 20°C vc ml/g t_°C 30 1.4053 4 30 t_c 265. 5 40 "C" 0.7702 4 P_c mm 18933. 5 Sugd. 5 296.2 MR (Obs.) 34.32 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.059 5 25°C 0.9963 5 (nD-d/2)1.0558 2 30 mm 1.0000 2 5 Dispersion 127. Dielectric BP 0.9515 Flash Point C 0.9433 5 A 0 to 6.75851 4 Fire Point B 130 °C 1219.73 M. Spec. Ultra V. ΔHc kcal/m C 4 221. 1043,68 2 ΔHf A* 0 to 5 1.16211 X-Ray Dif. ΔFf B*[110 °C 1140.86 Infrared ĸ Viscosity Solubility in centistokes Acetone to $\mathbf{t_k}$ Carbon tet. °C Benzene to Ether В' °C n-Heptane B_v | C' Ethanol °C Water A'* to Water in ٠c (B^V)| to Ac 130 to 7,1751 5 (A^{V}) °C Bc tc ° °C 1530. 5 c_p liq. ۰ĸ Cc 263. 5 Cryos. A c_p vap. °K consts. B° c vap. te °C 103.05 5 $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{c}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

						No. 50
NAME	2-Ethyl-	l-pen	tene			STRUCTURAL FORMULA
						CH (CH) C = CH
						$CH_3(CH_2)_2C = CH_2$ C_2H_5
Mole % Pur.	Ref. Me	olecul ormul		Molecular Weight 98.18	2	2115
		Ref.			Ref	R
F.P. °C			dt/dP	1		f to
F.P. 100%			*C/mm			g <u>*K</u>
B. P. ℃	0.4		25°C BP	0.3468 0.0464	5 4	h
760 mm 100	94. 35.7	2 4	te	0.0379	5	f! to
30	10.33	4	30 mm	0.6337	5	g' <u>°</u> K_
10 1	-8.8 -40.2	5 5	ΔHm cal/g			h!
Pressure	· · · · · · · · · · · · · · · · · · ·	T	ΔHv cal/g	00.41	ا ۔ ا	m to to
mm 25°C	61.99	5	25°C 30 mm	83.41 85.58	5 5	0
t _e Density	988.8	5	BP	73.27	5	m' to
g/ml 20°C	0.708	2	te t_ (d, e)	71.91	5 5	n' ' K
dt 25 4 30	0.704 0.700	2 4	ΔHv/Te	18.74	5	0'
	0.700	-	d 0 to		5	Surface tension
a b	-0.038	4	d, 110 °C		5	dynes/cm. 20°C 20.78 5
Ref. Index			d' to		1 1	40 18.93 5
n _D 20°C	1.405 1.402	2 2	d _c g/ml			Parachor [P]
30	1.399	4	v _c ml/g t _c °C	266.	5	30
"C"	0.7622	4		19065.	5	40 Sugd. 296. 2
MR (Obs.)	34.0	2	P _c mm PV/RT	17005.	┝┵┤	Exp. L.1.%/wt.
MR (Calc.) (nD-d/2)	34.059 1.051	5 2	25°C	0.9963	5	u.
Dielectric	1	╁╾┤	30 mm BP	1.0000 0.9515	5	Dispersion 125.
A 0 to	6, 76143	4	te	0.9433	5	Flash Point °C Fire Point
B <u>[131 °C</u>	1222.4	4	t _c			M Spec.
C	221.	5	ΔHc kcal/m ΔHf	1045.08	2	Ultra V.
A* 0 to B* 110 °C	1.16444 1143.39	5	ΔFf			X-Ray Dif. Infrared
к — — —			Viscosity			Solubility in +
\$	-	1 1	r centistokes	}		Acetone
x I			'			Carbon tet. Benzene
A' to B' ∟ °C						Ether
c,	1		B ^V to			n-Heptane Ethanol
A¹* to			AV I C	_		Water Water in
B'* °C		$\downarrow \downarrow$	(B ^V) to			Water in
Ac 131 to	7.1782 1534.	5	(A ^V) °C		\sqcup	
Cc Cc	263.	5	c _p liq. °K			
Cryos. A° consts. B°			c _p vap. °K			
t _e °C	103.56	5	c _v vap.			
$T_{R} = 0.75$	Tc		L	L		grams/100 grams solvent
	ES: 1-Dow	2-AF	I 3-Lit. 4-0	Calc. from de	t. dat	a 5-Calc, by formula
SOURCE:		AI				
PURIFICAT	ION:	AI	PI			
LITERATUI	RE REFERE	NCES	:			

rr							No. 51
NAME	3-Ethyl-	l-per	ntene			STRUCTURAL	FORMULA
						сн ₃ сн ₂ сн с	CH=CH-
34-1-	D-6 No			M-11		Ċ ₂ H ₅	
Mole % Pur.	Ref. Mo	rmula		Molecular Weight 98.18	32	25	,
		Ref.	7/		Ref.		Ref.
F.P. °C	-127.4	2	dt/dP			f to	
F.P. 100%)	├ ─	°C/mm 25°C	0,2520	5	g '° <u>K</u>	
B. P. °C 760 mm	85.13	2	BP	0.0461	4	h	
100	27.4	4	t _e	0.0384	5	f' to	
30 10	2.34 -16.5	4 5	30 mm	0.6244	5_	h'	
1	-47.3	5	ΔHm cal/g	 	-	m to	
Pressure mm 25°C	90.06	5	ΔHv cal/g 25°C	78.81	5	n K	
t _e	964.1	5	30 mm BP	82.03 70.24	5		
Density			11	69.06	5	m' to	
g/m1 20°C	0.6962 0.6917	2 2	t _e (d, e)	69.02	5	",	
d ₄ 25 30	0.6872	4	ΔHv/T _e	18.48	5	Surface tension	
a	0.7142	4	d -5 to e 105 °C	82.36 0.1423	5	dynes/cm, 20°C	19.42 5
b	-0.0388	4	d' to	0.1423		30 40	18.41 5 17.42 5
Ref. Index		2	e' °C			Parachor [P]	3
25 30	1.3954	2	dcg/ml vcml/g			20°C	
"C"	1.3928 0.7624	4	t _c °C	250.	5	30 40	
MR (Obs.)		2	P _c mm	17476.	5		296.2 5
MR (Calc.	34.059	5	PV/RT 25°C	0.9938	5	Exp. L.1.%/wt. u.	
(nD-d/2)	1.0499	2	30 mm	1.0000	5	Dispersion	121. 2
Dielectric A -5 to	(70122	١.	BP t _e	0.9520 0.9445	5	Flash Point °C	
B 1119 °C		4	t _c			Fire Point	
С	223.	4	ΔHc kcal/m ΔHf	1047.60	2	M. Spec. Ultra V.	
A* -5 to B* 105 °C		5	ΔFf			X-Ray Dif. Infrared	
к — — –	- //		Viscosity			Solubility in +	
t _k tō	-		centistokes り °C			Acetone	
t _x °C			•			Carbon tet. Benzene	
A' to B' °C						Ether	
č,' º	-		B _v to			n-Heptane Ethanol	
A¹* to			AV 1 C	_	ŀ	Water Water in	
B'* °C		-	(B ^V) to			Merci III	
Acl 119 to Bc tc °C	7, 1192 1481.	5	(A ^V) °C		ļ		
	265.	5	c _p liq. °K				
Cryos. A° consts. B°			c _p vap. °K				
t _e °C	93.70	5	c _v vap.				
$T_{\mathbf{R}} = 0.7$						† grams/100 gra	
	CES: 1-Dow		• • • • • • • • • • • • • • • • • • • •	Calc. from de	t. da	ta 5-Calc, by for	mula
SOURCE:			API				
PURIFICA			API				
LITERATU	RE REFERE	NCE	5:				

No. 52 NAME 2, 3-Dimethyl-1-pentene STRUCTURAL FORMULA $CH_3CH_2CH C = CH_2$ ĊH₃ĊH₃ Mole Ref. Molecular Molecular C7H14 % Pur. Formula Weight 98.182 Ref. Ref Ref. F.P. °C F.P. 100% -134.8 2 dt/dP to °C/mm °К g 25°C 0,2466 5 B. P. *C h BP 0.0457 4 2 760 mm 84.26 f' 0.0381 5 to 100 27.0 4 g¹ °K 2.07 4 0.6209 30 30 mm 5 -16.7 5 10 h' ∆Hm cal/g -47.4 5 1 to AHv cal/g Pressure °K n 25°C 79.05 mm 25°C 91.70 5 o 30 mm 82.33 5 te 961.5 5 BP 70,53 5 Density g/ml 20°C m to 69.37 te (d, e) °K 0.7051 2 69.33 5 o† 25 0.7008 2 dt AHV/T 18,62 5 30 0.6965 4 Surface tension 82.63 -5 5 0.7223 4 20.43 dynes/cm. 20°C 105 °C 0.1435 ь -0.0384 4 30 19.43 5 ٦ď to 40 18.45 5 Ref. Index e' •c n_D 20°C 1.4033 2 [P] Parachor d_c g/ml 25 1.4007 2 4 20°C v_c ml/g 30 1.3981 30 •C 250. 5 ŧ_c 40 "C" 0.7623 4 mm 18171. 5 5 Sugd. 296.2 MR (Obs.) 34.00 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 34.059 25°C 0.9933 (nD-d/2)1.0508 2 30 mm Dispersion 1.0000 5 125. 2 Dielectric BP 0.9520 5 Flash Point °C te 0.9446 A -5 to 6,72442 4 Fire Point 1_12<u>0 °C</u> 1181.0 M Spec C ΔHc kcal/m 223. 4 1043,42 Ultra V. ΔHf A*| -5 to 1.13745 X-Ray Dif. ΔFf B* 105 °C 1103.49 Infrared ĸ Viscosity Viscos..., centistokes °C Solubility in Acetone t_x | to Carbon tet. •c Benzene to Ether В' •c n-Heptane Вv C' Ethanol $\vec{A}^{\boldsymbol{v}}$ °C Water A'* to Water in B'* ۰c (BV) to Ac | 120 to 7.1424 (AV) °C 1485. Bc tc °C cp liq. °K 265. 5 Cc Cryos. Aº c_p vap. ۰ĸ consts. B° c_v vap. te °C 92.65 5 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 53

							No. 53			
NAME	2, 4-Dim	ethyl	-1-pentene			STRUCTURAL FORMULA				
1 1						C11 C11 C11 C				
					一	CH ₃ CH CH ₂ C = CH ₂				
Mole % Pur.		lecul rmuk		Molecular Weight 98,182	,	Ċн ₃ с	:н ₃			
76 Fur.	1 1 1 1 1	Ref	*************************************	Weight 98, 182	Ref.	T	Ref			
F, P, °C	-123,8	2	dt/dP	 	Kei.		1			
F. P. 100		+-	°C/mm	1		f to g &K				
B. P. ℃			25°C	0. 2261	5	h				
760 mm	81.64 24.7	2	BP t _e	0.0454 0.0381	5	f' to				
30	-0.04	4	30 mm	0,6173	5	g' <u>°K</u>				
10 1	-18.7 -49.2	5	ΔHm cal/g		T -	h'				
Pressure	-47.2	+-	ΔHv cal/g			m to				
mm 25°C	101.2	5	25°C	78.00	5	n <u>•K</u>				
t _e	954.0	5	30 mm BP	81.55 69.95	5	- <u></u>	ļ <u>-</u>			
Density	0 (043	2	t _e (d, e)	68.84	5	m' to				
g/ml 20°0	0.6943 0.6898	2		68.81	5	0'	l .			
d ₄ 25 30	0.6853	4	ΔHv/T _e	18.62	5	Surface tension				
a	0.7124	4	d 0 to e 95 °C	81.54 0.1420	5	dynes/cm. 20°C	19.20 5			
b	-0.0388	4	d' l to	0.1420		30 40	18.20 5 17.21 5			
Ref. Index		2	e' °C	ļ	-	Parachor [P]	1			
D 25	1.3959	2	d g/ml			20°C				
30	1.3952	4	vc m1/g tc °C	245.	5	30 40				
"C"	0.7656	4	P _c mm	17719.	5		296.2 5			
MR (Obs. MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.				
(nD-d/2)	1.0514	2	25°C 30 mm	0.9917 1.0000	5	u. Dispersion	,,,,			
Dielectric	:	T	BP	0.9520	5	Flash Point °C	125. 2			
A 0 to		4	t e t c	0.9449	5	. Fire Point				
B 1_115 °C	224.	4	ΔHc kcal/m	1042,74	2	M. Spec.				
A* 0 to		5	ΔHf	1042.74	-	Ultra V.				
B*[100 °C	1098.95	5	ΔFf			X-Ray Dif. Infrared				
K c		İ	Viscosity centistokes			Solubility in +				
t _k to		1	η °C			Acetone Carbon tet.				
* "C			,			Benzene				
A' to		1				Ether				
č,' 3	<u>^</u>		B ^v to			n-Heptane Ethanol				
A'+ to			A I °C	.		Water				
B'* °C		1	(B ^V) to			Water in	-			
Ac 115 to	7.1479	5	(A ^V) °C	_		1				
Bc tc °C	265.	5	c _p liq. °K	1						
Cryos. A		1	c _p vap. °K		1					
consts. B	<u>`</u>	 	!!							
t _e °C	89.69	5	c _v vap.	1	l	L				
$T_R = 0.7$						† grams/100 gra				
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula			
SOURCE:			PI							
PURIFICA			PI							
LITERATI	JRE REFERE	NCE	S:							
1										
L										

								No. 54	ł
NAME	3, 3-	Dim	ethyl-	-1-pentene			STRUCTURAL	FORMUL	A
							CH3		
		Γ					сн ₃ сн ₂ с сі	H=CH ₂	
Mole % Pur.	Ref.	Mo	lecul rmul	ar C ₇ H ₁₄	Molecular Weight 98.1	82	с́н ₃	-	
-/0_1 u1.		1 10	Ref		weight /o.1	Ref			Ref
F.P. °C	-134.3		2	dt/dP		-	f to	T	
F.P. 100%				°C/mm			g *K_		
B. P. ℃				25°C BP	0.1965 0.0452	5 4	h ¦		
760 mm 100	77.54		2 4	t _e	0.0383	5	f¹ to		Т
30 10	-3.5		4	30 mm	0.6116	5	g' '°K_	ł	
1	-22.1 -52.2		5 5	ΔHm cal/g			h'		-
Pressure				ΔHv cal/g 25°C	7/ 14	ا ۔ ا	m to		
mm 25°C	119.2		5 5	30 mm	76.14 80.18	5		1	
t _e Density	942.9		1 3	BP	68.68	5	m' to		1
g/ml 20°C			2	te (d, e)	67.63 67.60	5	n' °K_		
dt 25 4 30	0.6		2 4	AHv/Te	18.53	5			<u> </u>
	0.7		4	d -10 to		5	Surface tension dynes/cm, 20°C	19.54	5
ь	-0.0		4	- d - 100 %	0.1418	5	3 0	18.58	5
Ref. Index		204		e' °			40	17.64	5
n _D 20°C	1.39		2 2	d g/ml vc ml/g			Parachor [P] 20°C		
30	1.3	932	4	t _c °C	240.	5	30		1
"C"	0.7		4	P _c mm	17499.	5	40 Sugd.	296.2	5
MR (Obs.) MR (Calc.	34.0		5	PV/RT		\vdash	Exp. L.1.%/wt.		†
(nD-d/2)	1.04		2	25°C 30 mm	0.9909 1.0000	5	u. Dispersion	120.	2
Dielectric				BP	0.9525	5	Flash Point °C	120.	+
A -10 to		9410	4 4	te t _c	0.9458	5	Fire Point		<u> </u>
B [_1,1, _6]	224.	0	4	∆Hc kcal/m	1044.77	2	M Spec.		
A* -10 to		1439	5	ΔHf ΔFf	1		Ultra V. X-Ray Dif.		
B* 100 °C	1073.5	В	5	Viscosity	-	-	Infrared		<u> </u>
c	_			centistokes			Solubility in + Acetone		
t _x t _o				7 °	7		Carbon tet.		
A' to			\vdash		1		Benzene Ether		
B' °	<u>.</u>			B ^V to	-	 	n-Heptane		
A¹* to	 			B to			Ethanol Water		
B'* °C				(BV) to	<u>-</u>		Water in		-
Ac 111 to	7.1	122	5	(A ^V) •c					
Bc tc_°C	1447.		5	cp liq. °K					
Cryos. A°				c _p vap. °K	:				
te °C	85.1	3	5	c _v vap.					
$T_{R} = 0.7$	1			1			grams/100 grai	ns solver	ıt
)ow	2-AF	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:			Al						
PURIFICAT			AI						
LITERATU	RE REF	ERE	NCES	3:					

NAME

Mole

% Pur.

F.P. °C F.P. 100%

B. P. °C

100

30

10

1

Density

 d_4^t 25

n_D

"C"

a b

Pressure

mm 25°C

g/m1 20°C

Ref. Index

30

20°C

25

30

te

760 mm

3, 4-Dimethyl-1-pentene

Molecular

2

4

4

5

5

5

5

2

2

4

4

4

2

2

4

Formula Ref.

 $C_7 H_{14}$

°C/mm 25°C

30 mm

∆Hm cal/g

AHv cal/g

30 mm

te te (d, e)

ΔHv/T_e

-1 to

25°C

ВP

d

d٦

d_c g/ml

vc ml/g t °C

P_c mm

e'

t_c

dt/dP

BP

t_e

Ref.

81.

24.1

-19.3

-49.8

104.2

952.2

0.701

0.697

0.693

0.717

-0.038

1.3995

1.3969

1.3946

0.7599

-0.69

271 No. 55 STRUCTURAL FORMULA CH,CH CH CH=CH, Molecular 98.182 ĊH₃ĊH₃ Ref. Ref. f to °K g 0.2208 5 h 0.0454 4 0.0382 5 ſ١ to _<u>°K</u> g' 0.6168 5 h! m to n °K 78.34 5 0 81.22 5 69.65 5 m' to 68.55 n' °K 68.51 5 ۰' 18.58 5 Surface tension 81,13 5 - 89 °C dynes/cm. 20°C 19.95 5 5 0.1417 30 19.03 5 18.12 5 40 Parachor [P] 0,210 5 20°C 4.769 5 5 30 246. 40 17980. 5 Sugd. 296.2 5

MR (Obs.)	33.9	2	6	1.700.	الـــا	Duga.	270.2			
MR (Calc.) (nD-d/2)		5 2	PV/RT 25°C 30 mm	0.9907 1.0000	5	Exp. L.1.%/wt. u. Dispersion	120.	2		
Dielectric			BP	0.9520	5		120.	<u>-</u>		
A -1 to B 116 °C	6.71798 1170.33	4	te tc	0.9449	5	Flash Point C Fire Point				
c	224.	4	AHc kcal/m	1045, 32	2	M. Spec.	'			
A* -1 to B* 99 °C K	1.13434 1093.21	5	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared				
c t _k to c c			Viscosity centistokes n °C			Solubility in + Acetone Carbon tet. Benzene Ether				
B' i °C			B ^V to A ^V °C			n-Heptane Ethanol				
A ¹ * to B ¹ * °C			(B ^V) to			Water Water in				
Ac 116 to Bc t _c °C Cc —	7.1370 1472. 265.	5 5 5	(A ^V) °C c _p liq. °K							
Cryos. A° consts. B°			c _p vap. °K							
t _e °C	88.99	5	c _v vap.							
$T_{\mathbf{R}} = 0.75$	T _c			·	L.,I	grams/100 gra	ms solven	t		
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for	mula			
SOURCE:		A	PI							
PURIFICAT	ION:	A	PI							
LITERATUR	LITERATURE REFERENCES:									
1										
1										
1										

							No. 56	
NAME	4,4-Dim	ethyl	-1-pentene			STRUCTURAL	FORMULA	١.
						CH ₃		
					\neg	CH ₃ C CH ₂ C	H=CH ₂	
Mole % Pur.		lecul rmul		Molecular Weight 98.18	12	Ċн ₃		
		Ref.			Ref			Ref.
F. P. °C	-136.600	2	dt/dP			f to		
F.P. 100%			°C/mm	0.1//2	_	g <u>*K</u> _		l
B. P. °C 760 mm	72.40		25°C BP	0.1663 0.0448	5 4	h		
100	72.49 16.43	2 4	t _e	0.0384	5	f' to		
30 10	-7.87 -26.13	4 5	30 mm	0.6049	5	g' °K_		1
1	-55.93	5	ΔHm cal/g			h' to		├
Pressure			ΔHv cal/g 25°C	73.95	5	m to		
mm 25°C	144.6 928.4	5	30 mm	78.52	5	•		
Density	720.1	-	BP	67.25 66.30	5	m¹ to		
g/ml 20°C		2	te te (d, e)	66.27	5	n' ' <u>*K</u> -		ļ
dt 25 4 30	0.6785 0.6743	2 4	AHV/Te	18.46	5	o' ¦		_
2 30	0.6743	4	d -15 to	77.42	5	Surface tension dynes/cm. 20°C	17 02	_
ь	-0.0381	4	<u></u> q, 90 °€		5	y 30	17.93 17.03	5
Ref. Index			e' °C			40	16.15	5
n _D 20°C	1.3918	2 2	d g/ml	0.201	5	Parachor [P] 20°C		
30	1.3867	4	v _c ml/g t _c °C	4.985 230.	5	30		
"C"	0.7661	4	P _c mm	16735.	5	40 Sugd	296.2	5
MR (Obs.)		2	PV/RT	10.33.	-	Exp. L.1.%/wt.	270.2	
MR (Calc. (nD-d/2)) 34.059 1.0504	5 2	25°C	0.9880	5	u.		l _
Dielectric	•		30 mm BP	1.0000 0.9525	5	Dispersion	120.	2
A -15 to		4	t _e	0.9462	5	Flash Point °C Fire Point		
B 1_105 °C	1128.36	4	t _c AHc kcal/m	0.261	5	M Spec.		
A* -15 to		5	ΔHf	1043.65		Ultra V. X-Ray Dif.		
B* _ 90 ℃		5	ΔFf			Infrared		
K — — -		l	Viscosity centistokes	1		Solubility in +		
t _{te} to			η °c			Acetone Carbon tet.		
'x l						Benzene		
A' to						Ether n-Heptane		
c,	_		B ^V to			Ethanol		1
A'* to B'* °C			A ^V I C			Water Water in		ŀ
B'* °C		5	(B ^V) to					
Bc t °C	1421.	5			\vdash			
Ce	205.	5	c _p liq. °K	1				
Cryos, A° consts. B°			c _p vap. °K					
te °C	79.47	5	c _w vap.					
$T_{\mathbf{R}} = 0.7$						+ grams/100 gran	ns solven	t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5-Cafic, by form		
SOURCE:			PI					
PURIFICA'			PI					
LITERATU	RE REFERE	NCES	5:					

	·						No. 57		
NAME _	3-Ethyl-2	-pen	tene			STRUCTURAL FORMULA			
						CH ₃ CH ₂ C = C	H=CH,		
Mole % Pur.	Ref. Mo	ecul		Molecular Veight 98.18	2	c ₂ H ₅	3		
		Ref.			Ref.			Ref.	
F.P. °C F.P. 100%			dt/dP °C/mm			f to			
B. P. °C 760 mm 100 30	96. 01 37. 51 11. 98 -7. 27	2 4 4 5	25°C BP t _e 30 mm	0.3725 0.0465 0.0378 0.6369	5 4 5 5	h to g' °K			
Pressure mm 25°C t _e	57.19 993.9	5 5 5	ΔHv cal/g 25°C 30 mm BP	84.20 86.15 73.79	5 5 5	m to			
Density g/ml 20°C dt 25 4 30	0.7204 0.7159 0.7114	2 2 4	t _e (d, e) ΔHv/T _e	72.40 72.35 18.76	5 5 5	m' to n' Surface tension			
a b	0.7384 -0.0 ₃ 89	4	d 12 to e 106 °C d to	87.92 0.1472	5	dynes/cm. 20°C 30 40	22.28 21.17 20.09	5 5 5	
Ref. Index n _D 20°C 25 30	1.4148 1.4122 1.4094	2 2 4	d _c g/ml v _c ml/g t _c °C	0.217 4.598 270,	5 5 5	Parachor [P] 20°C 30			
"C"	0.7662	4	P _c mm	19361.	5	40 Sugd.	296. 2	5	
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	34.11 34.059 1.0546	2 5 2	PV/RT 25°C 30 mm BP	0.9967 1.0000 0.9510	5 5 5	Exp. L.1.%/wt. u. Dispersion	127.	2	
A 5 to B 134 °C C	6.77291 1233.83 221.	4 4 4	te tc	0.9426 0.258	5 5	Flash Point C Fire Point M. Spec.			
A* 5 to B* 116°C	1.17411	5	AHc kcal/m AHf AFf Viscosity	1044,30	2	Ultra V. X-Ray Dif. Infrared			
c t _k to C			centistokes 7°C			Solubility in Acetone Carbon tet. Benzene Ether			
B' °C C' to			B ^V to			n-Heptane Ethanol Water			
B'* °C	7. 1903	5	(B ^V) to (A ^V) °C			Water in		<u> </u>	
Bc tc °C Cryos, A°	1548. - 264.	5 5	c _p liq. °K						
consts. B°	ļ		P						
t _e °C	105.79	5	c _w vap.			L		<u> </u>	
T _R = 0.7	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t de	grams/100 grants 5-Calc. by for	ms solven	t	
SOURCE:		AI		oute. Itom de	ua	J-Calc. by for	******		
PURIFICAT	ION:	AI							
LITERATU	RE REFERE	NCES	S:						
			·						

							No. 58
NAME	·2,3-Dim	ethyl	-2-pentene		П	STRUCTURAL FO	
				·····		CH CH C - C	CH
			T		\neg	сн ₃ сн ₂ с = с сн ₃ сн	. Сп ₃
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 98.18	2	011301	*3
		Ref.	l		Ref		Ref
F.P. ℃	-118.3	2	dt/dP			f to	
F.P. 100%	1		*C/mm 25*C	0.3878	5	g <u> •K</u>	
B. P. *C 760 mm	97.46	2	BP	0.0470	4	h	
100	38.46	4	t _e	0.0381	5	f' to g'°K	
30 10	12.75 -6.63	4 5	30 mm	0.6412	5	h'	
1	-38, 37	5	ΔHm cal/g		\sqcup	m to	
Pressure	54, 94	5	∆Hv cal/g 25°C	84,24	5	n '•K_	
mm 25°C	998.2	5	30 mm	86.03	5	0	
Density			BP	73.68 72.27	5	m' to	
g/ml 20°C	0.7277 0.7234	2 2	t _e (d, e)	72.22	5	n' °K	l
dt 25 4 30	0.7190	4	ΔHv/T _e	18.64	5	Sumface Acceptant	
a	0.7450	4	d 5 to	87.89 0.1458	5	Surface tension dynes/cm. 20°C	23.20 5
ь	-0.0385	4	a' to			30 40	22.10 5 21.03 5
Ref. Index		2	•' i °C			Parachor [P]	21.03
45	1.4182	2	d g/ml v ml/g	j		20°C	
30	1.4156	4	tc °C °	273.	5	30 40	1
"C"	0.7688	4	P _c mm	19410.	5	Sugd. 2	96.2 5
MR (Obs.) MR (Calc.		2 5	PV/RT		_	Exp. L. l. %/wt.	
(nD-d/2)	1.0570	2	25°C 30 mm	0.9970 1.0000	5	u. Dispersion 1	30. 2
Dielectric			BP	0.9510	5	Flash Point °C	
A 5 to B 1_137.°C		4	te t _c	0.9424		Fire Point	
c Line	221.	4	AHc kcal/m	1041,82	2	M Spec.	
A* 5 to		5	ΔHf ΔFf	1	1 1	Ultra V. X-Ray Dif.	
B*	1153.70	5	Viscosity		\vdash	Infrared	
c	_		centistokes	1		Solubility in + Acetone	
t _x to			7 ℃			Carbon tet.	
A' to	 	\vdash				Benzene Ether	
B' *C	되		B ^v to		\vdash	n-Heptane	
A'* to	+		B' to			Ethanol Water	
B'* °C			(B ^V) to	1		Water in	
Ac 137 to		5	(A ^V) °C				
Bc tc_°C	1551. - 264.	5	cp liq. °K				
Cryos, A°	1	<u> </u>	c _p vap. °K				
consts. B°		_	P	l	1 1		
t _e °C	107.50	5	c _v vap.	L			
$T_R = 0.7$						f grams/100 gram	
	CES: 1-Dow			Calc. from de	t. dat	ta 5-Calc, by form	ula
SOURCE:	TON.		PI				
PURIFICAT			PI				
LILERATU	RE REFERE	NCES) :				

No. 59 2, 4-Dimethyl-2-pentene STRUCTURAL FORMULA NAME CH3C = CHCH CH3 ĊH₃ ĊH3 Mole Ref. Molecular Weight 98.182 Molecular C7H14 % Pur Formula Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm °K g 25°C 0.2398 5 B. P. °C h ВP 0.0456 4 760 mm 2 83.44 f١ 0.0381 5 to 100 26.25 4 g' <u>«к</u> 30 1.39 4 5 30 mm 0.6196 10 -17.33 5 h' AHm cal/g 5 1 -47.94 m to AHv cal/g Pressure °K n 25°C 78.72 5 mm 25°C 94.68 5 o 30 mm 82.09 5 $^{t}\underline{\mathbf{e}}$ 959.2 5 BP 70.30 5 m' Density to 5 69.15 t_e (d, e) n' °K g/ml 20°C 0.6955 2 69.11 5 01 25 0.6912 $\mathbf{d_4^t}$ 2 18.60 5 AHv/Te 0.6869 4 Surface tension -10 to 82.29 d 5 0.7126 -0.0₃84 44 19.34 dynes/cm. 20°C 105 0.1437 5 °C 18.38 ъ 30 to 17.44 5 40 Ref. Index e¹ ۰c 1.4040 20°C 2 [P] nD Parachor d_c g/ml 0.208 5 25 1.4013 20°C vc ml/g 5 4.81 30 1.3986 4 30 $^{\mathbf{t}}_{\mathbf{c}}$ 248. 5 40 "C" 0.7741 4 P_c mm 17809. 5 Sugd. 296. 2 5 MR (Obs.) 34.52 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.059 25°C 0.9930 5 (nD-d/2)1.0562 2 30 mm 1.0000 5 Dispersion 128. 2 Dielectric BP 0.9520 5 Flash Point C 0.9447 5 A -10 to 6.71947 4 Fire Point 0.259 1176.32 B 1118 °C_ 4 M. Spec. Ultra V. 223. C 4 AHc kcal/m 1041.34 2 ΔHf 1.13362 A* -10 to 5 X-Ray Dif. ΔFf B*[105 °C 1099.06 Infrared ĸ Viscosity Solubility in centistokes Acetone to °C Carbon tet. °C Benzene A١ to Ether B١ ٠c n-Heptane $\mathbf{B}^{\mathbf{v}}$ C' to Ethanol v | °C Water A'* to Water in B'* (B^V) °C to Ac 118 to 7.1369 5 (A^V)| °C Bc tc ۰c 1478. 5 °K c_p liq. 5 264. Cryos. A° c_p vap. ۰ĸ consts. B° c, vap. t_e °C 91.73 5 $= 0.75 T_{\rm C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

								No. 60	
NAME	3,4	-Dime	thyl-	-cis-2-pentene			STRUCTURAL 1	FORMULA	4
							сн ₃ сн с = 0	сн сн.	
Mole	Ref	Mo	lecul	27	Molecular		ċн ₃ ċн ₃	3	
% Pur.			rmul		Weight 98.18	2			
			Ref.			Ref			Ref
F.P. °C			\sqcup	dt/dP			f to		
F. P. 1007 B. P. °C	<u>'</u>		\vdash	*C/mm 25*C	0,2711	5	g <u>*K</u> _		
760 mm	87.		2	BP	0.0459	4	h		⊢
100 30	29.4 4.3		4 4	t _e 30 mm	0.0380	5	f' to		
10	-14.5		5	ΔHm cal/g	0.6251	-	h'		
1	-45.4		5	ΔHv cal/g	 	\vdash	m to		
Pressure mm 25°C	82.2	8	5	25°C	80.23	5	n °K		
t _e	969.4		5	30 mm BP	83.15	5 5			ـــ
Density		1 2		l t_	70.12	5	m' to		
g/ml 20°0	0.7		2 2	t _e (d, e)	70.09	5	o' ' = -		
d ₄ 25	0.7	05	4	ΔHv/T _e	18.66	5	Surface tension		
a b	0.7		4	d -5 to		5	dynes/cm. 20°C	21.37	5
Ref. Index		310	-	d' to			30 40	20.40 19.45	5 5
n _D 20°0	1.4		2		0,216	5	Parachor [P]		
25 30	1.4		2	d _c g/ml v _c ml/g t _c °C	4.63	5	20°C		İ
"C"	0.7		4		256.	5	40		
MR (Obs.	33.9		2	P _c mm	18790.	5		296.2	5
MR (Calc. (nD-d/2)	34.0		5 2	25°C	0.9944	5	Exp. L.1.%/wt. u.		
Dielectric			┝╧╢	30 mm BP	1.000 0 0.9520	5 5	Dispersion	127.	2
A -5 t		4284	4	te	0.9444	5	Flash Point °C Fire Point		
B (_124 •9			4	tc	0.259	5	M Spec.		┢
A* -5 to		5 215	5	ΔHc kcal/m ΔHf	1042.02	2	Ultra V.		
B* 105 °	1118.9		5	ΔFf		Ш	X-Ray Dif. Infrared		
K				Viscosity centistokes			Solubility in +		
the Te				η °c			Acetone Carbon tet.		
tÇ i °C			\sqcup				Benzene		
B'i °							Ether n-Heptane		
C'				B ^V to A ^V °C	İ	1 1	Etha nol		
A'* to					-		Water Water in		
Ac 124 to	7.1	618	5	(A ^V) to	1				
Bc t *	C 1506.		5	c _p liq. °K	·	\vdash			
Cryos. A	265.		5	11 -					
consts. B				c _p vap. °K					
t _e °C	95.7	2	5	c _v vap.					
$T_R = 0.7$							f grams/100 gran	ns solven	t
REFEREN	CES: 1-I	Dow	2-AF		Calc. from de	t. dat	ta 5-Calc. by form	nula	
SOURCE:				PI					
PURIFICA				PI	· · · · · · · · · · · · · · · · · · ·				
LITERATU	KE REF	ERE	CES	5:					

No. 61 43, 4-Dimethyl-trans-2-pentene NAME STRUCTURAL FORMULA CH3CH C = CHCH3 Ċн₃Ċн₃ Ref. Molecular Mole Molecular C7H14 Weight 98.182 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g <u>°K</u> 0.2711 5 B.P. °C h BP 0.0459 87. 760 mm 2 0.03801 5 f١ t_e to 100 29.43 4 g' <u>°К</u> 30 4.36 5 30 mm 0.6251 5 10 -14:53 4 h' ∆Hm cal/g 4 -45.44 m to AHv cal/g Pressure °K n 25°C 80.23 5 mm 25°C 82.28 4 o 30 mm 83.15 5 969.36 te 4 BP 5 71.33 m' to Density 70.12 5 n' °K g/ml 20°C te (d, e) 0.713 2 70.09 5 o' ď4 25 0.709 2 AHv/T 18.66 5 30 0.7050 4 Surface tension d -5 83,78 5 to a 0.7290 4 dynes/cm. 20°C 21.37 5 ᇷᅴ <u>96</u> °C 0.1430 5 ь -0.0378 4 30 20.40 to 19.45 5 40 Ref. Index e' 20°C [P] ⁿD 1.407 2 Parachor d_c g/ml 0.216 5 25 1.404 20°C vc ml/g t °C 5 4.633 30 1,402 4 30 t_c 256. 5 40 "C" 0.7604 5 P_c mm 18790. 5 Sugd. 296.2 5 MR (Obs.) 33.9 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 5 25°C 0.9939 5 (nD-d/2)1,050 2 u. 30 mm 1.0000 Dispersion 127. 5 2 Dielectric ВP 0.9520 5 Flash Point C 0.9444 5 A -5 to 6.74284 Fire Point 0.259 5 B | 124 °C 1197.23 M. Spec. Ultra V. С 4 223. AHc kcal/m 1042.02 2 ΔHf A* -5 to 5 1,15215 X-Ray Dif. ΔFf B* 105 °C 1118,91 Infrared ĸ Viscosity Solubility in centistokes Acetone to t_k Carbon tet. °C Benzene ۸' to Ether В' °C n-Heptane $\mathbf{B}^{\mathbf{v}^{\top}}$ C' Ethanol 'v | °C Water A'* to Water in B'* °C (B^V) to Ac | 124 to 7.1618 5 (A^V)| °C Bc _tc_ 1505.5 °C 5 c_p liq. °K Cc 265.11 5 Cryos. Aº c_p vap. °K consts. B° c, vap. te °C 95.72 5 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** LITERATURE REFERENCES:

							No. 62		
NAME	4,4-Dim	ethyl	-cis-2-pentene			STRUCTURAL	FORMUL	4	
						CH ₃			
					$\neg \uparrow$	сн, с сн=	снсн,		
Mole		lecul		Molecular	,	ĊH ₃			
% Pur.	. FO	rmul	, 	Weight 98.18	_	, , , , , , , , , , , , , , , , , , ,		Ref.	
	1 125 44	Ref.		T	Ref.			Ker.	
F.P. °C F.P. 100%	-135.46	2	dt/dP *C/mm	1	1	f to			
B. P. *C			25°C	0.2165	5	h .			
760 mm	80.42	2	BP t _e	0.0454 0.0382	5	f' to		┼	
100 30	23.53	4	30 mm	0,6160	5	g' 'K_			
10	-19.80	5	ΔHm cal/g	0.0100	 _	h'		Ì	
1	-50, 21	5	ΔHv cal/g	 	 	m to			
Pressure mm 25°C	106.6	5	25°C	77.16	5	n <u>*K</u> -			
t _e	950.6	5	30 mm BP	81.03 69.46	5	<u> </u>		ļ	
Density			t_	68.37	5	m' to			
g/ml 20°C	0.6996 0.6952	2 2	i e (4, 6)	68.34	5	" ' " -			
dt 25 4 30	0.6908	4	ΔHv/T _e	18.57	5		ļ		
a	0.7172	4	d -10 to		5	Surface tension dynes/cm, 20°C	19.79	5	
ь	-0.0 ₃ 86	4	-6, - -88 °C		٦	3 0	18.79	5	
Ref. Index			e'			40	17.80	5	
n _D 20°C	1.3998	2 2	d g/ml	0.208	5	Parachor [P] 20°C			
30	1.3970	4	t _c °C	4.819 244.	5	30			
"C"	0.7667	4	P _c mm	17725.	5	40 Sugd.	296.2	5	
MR (Obs.) MR (Calc.		2 5	PV/RT	 	<u> </u>	Exp. L.1.%/wt.	-/	1	
(nD-d/2)	1.0526	2	25°C	0.9915	5	u.		١.	
Dielectric			30 mm BP	1.0000 0.9520	5	Dispersion	122.	2	
A -10 to	6,71324	4	t _e	0.9450	5	Flash Point °C Fire Point			
B 1_115 °C	21166.67 224.	4	tc	0. 260	5	M Spec.		 	
A* -10 to	 	5	ΔHc kcal/m ΔHf	1042.05	2	Ultra V.			
B* _ 95 °C	1.13040	5	ΔFf			X-Ray Dif. Infrared			
K	1		Viscosity		1	Solubility in +		\vdash	
t _k	-1		centistokes 7 °C	İ		Acetone			
tî °C	;	l i	•			Carbon tet. Benzene			
A' to				1		Ether			
c,	<u>- </u>		B ^V to		 	n-Heptane Ethanol			
A'* to			A ^v i °C			Water			
B'* °((B ^V) to	1		Water in		 	
Ac 115 to	7. 1319	5	(A ^V) °C	L					
Bc tc_°C	1467. 265.	5	c _p liq. °K						
Cryos. A°			c _p vap. °K						
consts. B°				1					
t _e °C	88.34	5	c _v vap.					L	
$T_{\mathbf{R}} = 0.7$						grams/100 gran	ns solven	t	
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for	nula		
SOURCE:		A.	PI						
PURIFICA?			PI						
LITERATU	RE REFEREI	1CES):						

							No. 63	3	
NAME	4, 4-Dime	thyl	trans-2-penten	e		STRUCTURAL	FORMUL	A	
						CH ₃			
Mole % Pur.		ecul:		Molecular Weight 98.18	2	сн ₃ с сн=снсн ₃ сн ₃			
		Ref.			Ref.			Ref	
F.P. °C F.P. 100%	-115.235	2	dt/dP °C/mm			f to g*K			
B. P. °C 760 mm 100	76.75 20.24	2 4	25°C BP t	0.1913 0.0451 0.0383	5 4 5	h f' to g' c'K		_	
30 10	-4.27 -22.7	4 5	30 mm	0.6105	5	h'			
1	-52.8	5	ΔHm cal/g			m to		-	
Pressure mm 25°C t _e	122.9 940.6	5 5	ΔHv cal/g 25°C 30 mm BP	75.80 79.93 68.43	5 5 5	n•K			
Density g/ml 20°C dt 25 d4 30	0.6889 0.6845	2 2	te te (d, e) AHv/Te	67.40 67.37 18.51	5 5	m' to			
a b	0.6801	4	d -10 to e 95 °C	79.33 0.1420	5 5	Surface tension dynes/cm. 20°C	18.60	5	
Ref. Index	-0.0386	4	d' to e' °C			8 30 40	17.64 16.69	5	
ⁿ D 20°C 25 30	1.3982 1.3953 1.3928	2 2 4	d _c g/ml v _c ml/g t _c °C	0.203 4.928 237.	5 ·5 5	Parachor [P] 20°C 30			
"C"	0.7709	4	P _c mm	17097.	5	40 Sugd.	296.2	5	
MR (Obs.) MR (Calc.) (nD-d/2)	34.41 34.059 1.0538	2 5 2	PV/RT 25°C	0.9908	5	Exp. L.1.%/wt. u.			
Dielectric			30 mm BP	1.0000 0.9525	5	Dispersion	124.	2	
A -10 to B 110 °C C	6.68799 1145.0 224.	4 4 4	te t	0.9458 0.260	5	Flash Point ⁶ C Fire Point M. Spec.		-	
A* -10 to B* 95 °C	1.10939 1069.0	5 5	ΔHc kcal/m ΔHf ΔFf	1041.05	2	Ultra V. X-Ray Dif. Infrared			
c t _k to c			Viscosity centistokes			Solubility in + Acetone Carbon tet. Benzene			
A' to B' °C			B ^V to A ^V °C			Ether n-Heptane Ethanol Water			
A'* to B'* °C		_	(B ^V) to			Water in		-	
Ac 110 to Bc tc °C Cc	7.1055 1440. 265.	5 5 5	(A ^V) °C c _p liq. °K						
Cryos. A° consts. B°			c _p vap. °K				2		
t _e ℃	84.25	5	c _v vap.						
$T_{\mathbf{R}} = 0.75$	T _c					f grams/100 gra		ıt	
REFERENC	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula		
SOURCE:		AF	PI II						
PURIFICAT		AF							
LITERATUR	E REFEREI	NCES	:						

							No. 64	
NAME	2-Et	hyl-3-me	thyl-l-butene			STRUCTURAL		
-			,.		\dashv	C ₂ H		-
						сн _з сн с =		
Mole	Ref.	Molecul	27	Molecular	- 1		CH ₂	
% Pur.	Mer.	Formul	ar C ₇ H ₁₄	Weight 98.18	32	ĊН ₃		
		Ref.			Ref			Ref
F.P. °C	1		dt/dP	T		f to		t
F.P. 100%	 		*C/mm			f to		l
B, P, *C	†		25°C	0.2906	5	h	1	
760 mm	89.	2	BP	0.0460 0.0380	4 5	f' to	 	\vdash
100 30	31.23		t _e 30 mm	0.6272	5	g' ' °K		
10	-12.9	5		0.0272	 	h'	1	1
1	-43.9	5	ΔHm cal/g		 	m to		
Pressure			ΔHv cal/g 25°C	81.09	5	n '*K	_	1
mm 25°C	75.95 975.1	5 5	30 mm	83.90	5	•		
Pomeite	713.1		BP	71.87	5	m' to		
Density g/ml 20°C	0.71	5 2	te te (d, e)	70.61	5	n' 'K		ľ
_a t 25	0.71		e (a, c)	1	5	o'		
⁴ 4 30	0.70		ΔHv/T _e	18.68		Surface tension	 	
a	0.73		d -5 to		5	dynes/cm. 20°C	21.61	5
Ъ	-0.03	8 4	1-a,; 65			30 40	20.64 19.68	5
Ref. Index	1 , 4,	.0 2	e' i °C			l	17.00	1-
n _D 20°C	1.41		d _c g/ml	0.216	5	Parachor [P] 20°C	ŀ	
30	1.40	5 4	V_ mi/g	4.624 259.	5	30	1	1
"C"	0.76	35 4	11 -	ł	1 1	40		_
MR (Obs.)	34.0	2	P _c mm	18932.	5	Sugd	296.2	5
MR (Calc.	34.05		PV/RT 25°C	0.9949	5	Exp. L.1.%/wt. u.		
(nD-d/2)	1.05	2 2	30 mm	1.0000	5	Dispersion	125.	2
Dielectric			BP	0.9520	5	Flash Point °C	ļ	
A -5 to			te te	0.9442 0.259	5	Fire Point		[
B <u>1.125 °C</u> C	1200.64	4	t _c ΔHc kcal/m	1043, 42	2	M Spec.		
A* -5 to	1.14		AHf	1045.42	-	Ultra V.	j	1
B* 110 °C			ΔFf			X-Ray Dif. Infrared		
к — — —	1	i	Viscosity			Solubility in +	 	一
t _k	-	}	centistokes 7°C		1 1	Acetone	į.	
tx c	:]	ł	l 1			Carbon tet.	ļ	
A' to	†		1		1	Benzene Ether	Ì	1
B' L _ *	<u>.</u>	ŀ	B ^V to	 	1-	n-Heptane	ł	
C'	}		B ^V to A ^V C	1		Ethanol.		1
A'* to B'* °C	.1	[-		Water Water in		
Ac 125 to		91 5	1					1
Bc t °C	1509.	91 5	(A ^V) °C		\perp]
Cc	264.	5	c _p liq. °K					1
Cryos. A° consts. B°			c _p vap. °K					
t, °C	97.97	5	c _w vap.					
$T_{R} = 0.7$			<u> </u>	L		+ arame/100 c==	ma acles	
REFERENC		ow 2-A1	PI 3-Lit. 4-(Calc from de	+ 4-	grams/100 gra		<u>. </u>
SOURCE:	D		PI	Jane, Moni de	ua	ta 5-Calc. by for	u.a	
PURIFICAT	'ION:							
LITERATU			PI					
	ALFI	MENCES	··					

							No. 65	
NAME _	2, 3, 3-Tr	imet	hyl-1-butene			STRUCTURAL	FORMULA	
					1	CH ₃		
Mole % Pur.	Ref. Mo	lecul rmuk		Molecular Weight 98.1	82	сн ₃ с с сн ₃ с		
		Ref.			Ref.		R	ef.
F. P. °C	-109.85	2	dt/dP			f t		
F.P. 100%	 	├—	°C/mm 25°C	0.1987	5	gº1	의	
B. P. °C 760 mm	77.87	2	BP	0.0452	4	h		_
100 30	21.26	4	t _e	0.0383	5	f' t		
10	-3.31 -21.79	5	30 mm	0.6121	5	h'		
1	-51.98	5	ΔHm cal/g	+	+-i	m t	,	
Pressure mm 25°C	117 4	5	ΔHv cal/g 25°C	76, 23	5	n		
t _e	117.6 943.3	5	30 mm	80.29	5	0		
Density			BP te	68.75 67.70	5 5	m' t		
g/ml 20°C		2	e (4, 6)	67.67	5	n'	۱ ۲	
dt 25 4 30	0.7005 0.6960	2	ΔHv/T _e	18.53	5		+	
a	0.7230	4	d -10 to	79.82	5 5	Surface tension dynes/cm. 20°0		5
ь	-0.0388	4		0.1422	"	8 30	19.35	5
Ref. Index	1.4029	2	e' °C		1	40 Danaskan (D)	18.32	5
D 25	1.4000	2	d _c g/ml	0.209 4.789	5	Parachor [P]	;	
30	1.3973	4	v _c ml/g t _c °C	241.	5	30 40		
"C"	0.7617	4	P _c mm	17662.	5		1. 296.2	5
MR (Obs.) MR (Calc.)	33.98 34.059	2 5	PV/RT			Exp. L. l. %/wt		
(nD-d/2)	1.0504	Ž	25°C 30 mm	0.9903	5 5	u. Dispersion	124.	2
Dielectric			BP	0.9520	5	Flash Point C	122.	_
A -10 to	6.69701	4	te tc	0.9452 0.259	5	Fire Point		
B (115 °C)	1152.0 224.	4	ΔHc kcal/m	1042.08	2	M. Spec.		
A* -10 to	1.11770	5	ΔHf ΔFf		1 1	Ultra V. X-Ray Dif.		
B*[_95 °C K	1075.8	5	Viscosity	+	+-1	Infrared		
c			centistokes			Solubility in +		
t _k to t_ ⊢ °C		1	η ° C			Carbon tet.		
t _x °C A' to		 				Benzene Ether		
B' i°C	.	i i	B _v to	 	+	n-Heptane	1 1	
	+	-	B to			Ethanol Water		
A'* to B'* °C			(B ^V) - to	-		Water in		
Ac 115 to	7.1153	5	(A ^V) °C					
Bc tc °C	1450. 265.	5 5	c liq. °K	1	\top			
Cryos, A°	203.	1	P	1		İ		
consts. B°			P					
t _e °C	85.48	5	c _v vap.	1				
$T_{\mathbf{R}} = 0.75$	T _c					+ grams/100 gr	ams solvent	
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da	ta 5-Calc, by fo	rmula	
SOURCE:		AF	PI .					
PURIFICAT		AF				,		
LITERATU	RE REFERE	NCES	5:					

									No. 66	
NAME	1-00	tene					STR	UCTURAL	FORMULA	1
Mole % Pur.	Ref.	Mo Fo	lecul	ar C ₈ H ₁₆	Molecular Weight 112.2	08	c	:н ₃ (сн ₂) ₅ сн	=CH ₂	
			Ref.			Ref.				Ref.
F.P. ℃	-101.73	6	2	dt/dP			f	l to		
F.P. 100%				*C/mm	1 0424	_	g	°K		
B. P. °C 760 mm	121 20			25°C BP	1.0436	5 4	h	L		
100 mm	121.28		2 2	te	0.03622	5	f'	to		
30	35.33		4	30 mm	0.6583	5	g'	°K		
10 1	15.38 -17.5	•	5	∆Hm cal/g			h'	1		
Pressure	 		\vdash	ΔHv cal/g	9(93	,	m n	300 to	0.0250 0.0013	
mm 25°C	17.38	;	5	25°C 30 mm	86.82 85.36	5	o		-0.0649	
t _e	1064.		5	BP	72.70	5	m'	700 to	0.1070	4
Density g/ml 20°C	0.71	492	2	t _e (d, e)	70.95 70.86	5	n'	1000 °K	0.0011	
at 25	0.71	085	2		19.56	5	٥'	i	-0.0638	4
⁴ 30	0.70		4	d 25 to		5	Surf	ace tension		
a b	0.73		4 4	e 134 °C		5	dyne	s/cm. 20°C 30	20.79 19.86	5
Ref. Index			\vdash	d' to				40	18.94	5
n _D 20°C	1.40		2		0,240	5	Para	chor [P]		
25 30	1.40		2 4	d g/ml v ml/g	4.173	5		20°C 30		
"C"	0.76		4	t _c ·C	292.	5	ľ	40		
MR (Obs.)			2	P _c mm	19197.	5	<u></u>	Sugd.	335.2	5
MR (Calc.			5	PV/RT 25°C	1.0000	5	Exp.	L.1.%/wt.		
(nD-d/2)	1.05	124	2	30 mm	1.0000	5	Disp	u. ersion	116.8	2
Dielectric			1	BP	0.9500 0.9396	5	Flas	h Point °C		
A 0 to B 151 °C		263	2 2	t _e	0.255	5	Fire	Point		<u></u>
č	212.76	4	2	ΔHc kcal/m	1194.97	2	M S			
A* 25 to	1.37	023	5	ΔHf			Ultr X-R	a v. ay Dif.		
B* ∟140 °C	1272.1		5	ΔFf	· 	-	Infra			
c				Viscosity centistokes				bility in +		ĺ
tk to				7 40 °C	0.382 0.316	2 2		etone rbon tet.	:	1
x			\vdash	80	0.316	2		nzene	ŀ	ļ
B'				100	0.234	2	Eth n-F	ier Ieptane		
c'			oxdot	B ^V 20 to		5	Eth	anol		
A'* to B'* °C				AV 60 °C	_	5	Wa Wa	ter ter in		
Ac 151 to	+	70	5	(B ^v) 60 to	409.49 2.27184	5				Π
Bcit C	1664.		5		 	9				
Cc	252.		5	c _p liq. °K	1					1
Cryos, A° consts. B°				c _p vap.300°K	0.38117 0.48045	2 2				
t _e °C	133.78		5	c _v vap.	<u> </u>					
$T_R = 0.75$								ms/100 grai		t
REFERENC	CES: 1-D			PI 3-Lit. 4-	Calc. from de	t. da	ta 5-	Calc. by for	mula	
SOURCE:			API							
PURIFICAT			PI							
	NEF.	-dev Est		•						

No. 67 cis-2-Octene NAME STRUCTURAL FORMULA CH3(CH2)4CH=CHCH3 Molecular C8H16 Mole cular Mole Ref. Weight 112.208 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% -100.2 2 dt/dP f to °C/mm ۰ĸ g 25°C 1.1514 B. P. ℃ h ΒP 0.04870 5 760 mm 125.64 2 5 te 0.03714 ſ١ to 100 64.12 4 g¹ 30 37.93 4 30 mm <u>°К</u> 0.6758 5 10 16.52 5 h! ∆Hm cal/g 1 -17.84 5 m to AHv cal/g Pressure n °K 25°C 85.12 mm 25°C 16.067 5 o 30 mm 5 84.10 $\mathbf{t_e}$ 1076.47 5 BP 71.86 5 m' Density to 5 70.08 te (d, e) n' <u>°K</u> g/ml 20°C 0.7243 2 70.01 5 ۰' 25 d₄ 0.7201 ΔHv/T_e 19.08 5 30 0.7159 4 Surface tension 38 89.23 to 5 5 0.7411 -0.0₃84 44 dynes/cm. 20°C 21.91 1 140 °C 0.1382 ъ ā 30 20.90 5 20 to 87.21 40 19.93 1 Ref. Index e' 5 38 0.0838 20°C 1.4150 2 [P] ^{n}D Parachor d_c g/ml 0.25 5 25 1.4125 2 20°C 335, 21 4 5 vc ml/g t °C 4.006 30 1.4099 4 30 335.23 4 ^tc 300. 5 "C" 40 335.21 4 0.7624 4 P_c mm 18399. 5 Sugd 335.2 5 MR (Obs.) 38.794 38.677 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 25°C 1.0000 1.0529 2 (nD-d/2) 30 mm 1.0000 5 Dispersion 2 118. Dielectric BP 0.9495 5 Flash Point C t_e 0.9386 5 A 38 to 6.87711 Fire Point 0.254 5 B 1157 °C 1361.3 M. Spec. C 215. 4 ∆Hc kcal/m Ultra V ΔHf 1.30761 1277.39 A* 38 to X-Ray Dif. ΔFf B*[150 °C Infrared ĸ Viscosity Solubility in centistokes Acetone to °C °C Carbon tet. t_{x_} Benzene 20 to 7, 26268 Ether 38_**°C** 1562.64 n-Heptane B_v | C' 5 233. to Ethanol °C A!* 20 to 1.65772 Water B'* 38 °C Water in $(\mathbf{B}^{\mathbf{v}})$ 1463.16 5 to Acl 157 to 7.28738 (AV) 5 °C Bc_tc_°C 1683.60 c_p liq. °K Cc 256.69 5 Cryos. A° consts. B° c_p vap. °K te °C c, vap. 139.03 5 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 68 NAME trans-2-Octene STRUCTURAL FORMULA CH3(CH2)4CH=CHCH3 Mole Ref. Molecular Molecular $C_{8}H_{16}$ Formula Weight 112, 208 % Pur Ref Ref. Ref F.P. °C F.P. 100% -87.7 2 dt/dP f to °C/mm <u>°K</u> g ٠5 25°C 1,1227 B. P. °C BP 0.0487 760 mm 125.0 2 t_e ſ١ 0.0372 5 to 4 100 63.55 g¹ °K 30 36.56 4 30 mm 0.6748 5 10 16,02 h' ∆Hm cal/g -18,28 5 1 to AHv cal/g Pressure °K n 25°C 84.90 mm 25°C 16.52 5 o 30 mm 5 83.94 1075. 5 t_e ВP 71.69 5 to Density m 69.92 5 te (d, e) ٩K g/ml 20°C 0.7199 2 69.84 5 ۰, 25 0.7157 $\mathbf{d_{4}^{t}}$ AHv/T 19.06 5 30 0.7115 4 Surface tension 89.00 37 to 5 0.7367 4 dynes/cm. 20°C 21.38 0.1385 87.00 _138 °C 5 ъ -0.0384 4 30 5 20.39 ā٠ to 5 25 ı 5 19.44 40 Ref. Index e' 37 °C 0.0836 5 20°C 1.4132 ſ₽Ì $\mathbf{n}_{\mathbf{D}}$ Parachor d_c g/ml 5 0, 225 25 1.4107 2 20°C ml/g 4.44 5 30 1.4081 4 c 30 •c 299. 5 ŧč 40 "C" 4 0.7639 Pc mm 18206. 5 5 Sugd 335.2 MR (Obs.) 38.88 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 25°C 1.0000 (nD-d/2) 1.0533 2 30 mm 1.0000 Dispersion 120. 2 0.9495 Dielectric BP 5 Flash Point °C te 0.9386 5 A 37 to 6.87364 Fire Point 0.254 5 В 1357.6 M Spec. c 215. 4 ΔHc kcal/m Ultra V. ΔHf 1.30491 A* 37 to X-Ray Dif. ΔFf _148 °C B* 1273.8 Infrared ĸ Viscosity Viscozi, centistokes °C Solubility in Acetone to Carbon tet. ·c Benzene 25 to 7, 25977 Ether В' 1558.8 37 °C n-Heptane Ċ١ 233. 5 $\mathbf{B}^{\mathbf{v}}_{\mathbf{A}^{\mathbf{v}}}$ Ethanol °C Water A'* 25 to 1.66269 Water in B'* 37 °C 1461.4 5 (BV) to Ac | 156 to 7.2835 (AV) °C Bc _tc_° 1678. °C cp liq. °K Cc 256. 5 Cryos. A° consts. B° c_p vap. °K c_v vap. 138.31 te °C 5 $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 69 cis-3-Octene NAME STRUCTURAL FORMULA CH3(CH2)3CH=CHCH2CH3 Mole Ref. Molecular Molecular C8H16 % Pur. Weight 112, 208 Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP Í to °C/mm g °K 25°C 1.0317 5 B.P. °C h ΒP 0.0485 4 122.9 760 mm 2 0.0372 5 ſ١ t_e to 100 61.66 4 <u>°К</u> g' 30 34.76 4 30 mm 0.6726 5 14.28 10 5 h! ∆Hm cal/g -19.91 5 to m AHv cal/g Pressure ۰ĸ n 25°C 84.02 5 18.16 mm 25°C 5 o 30 mm 83.24 5 te 1069. 5 BP 71.17 5 m to Density 5 69.46 n' <u>°K</u> g/ml 20°C te (d, e) 0.721 2 69.39 5 ٥' d_4^t 25 0.717 2 ΔHv/T_e 19.05 5 30 0.713 4 Surface tension 35 87, 99 5 d to 0.7370 dynes/cm. 20°C 5 21.51 136 °C 0.1369 5 Ъ -0.0₃80 4 30 20.56 5 an' 86.04 5 40 19.64 Ref. Index e' 35 0.0807 5 20°C [P] 1.4135 ^{n}D Parachor 5 d_c g/ml 0.228 25 1.4111 2 20°C vc ml/g 4.393 5 30 1.4087 4 30 $\mathbf{t_c}$ 297. 5 40 "C" 0.7633 4 P_c mm 18324. 5 Sugd. 5 335.2 MR (Obs.) 38.8 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 25°C 1.0000 5 (nD-d/2)u. 1.053 2 30 mm 1.0000 Dispersion 2 5 5 119. Dielectric BP 0.9495 Flash Point °C 0.9388 5 A 35 to 6.87423 4 Fire Point 0.254 B 1155 °C 1353.4 M. Spec. Ultra V. C 216. 4 AHc kcal/m ΔHf A* 35 to B* 146 °C 1.30707 1269.7 5 X-Ray Dif. ΔFf Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. °C t⊊ i Benzene 25 to 7,26164 Ether В' 35 °C 1554.6 n-Heptane BV | C' 234. 5 Ethanol °C Water A'* 25 to B'* 35 °C 1.65709 5 Water in (B^V)| 1454.9 to Ac 155 to 7.2856 5 (A^V) °C 1675. Bc tc °С 5 c_p liq. °K Cc¹ 258. 5 c_p vap. Cryos. A °K consts. B° c vap. te °C 5 135.95 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 70	
NAME	trans-3-Octene					STRUCTURAL FORMULA		
14-1-	Def	Malan	lecular Molecular			CH ₃ (CH ₂) ₃ CH=C	HCH ₂ CH	3
Mole Ref. Mo % Pur. Fo		Formu	lecular C8H16 Molecular Weight 112,208					
		Ref			Ref.			Ref.
F,P. °C	-110.	2	dt/dP			f to		
F. P. 100%	<u>' </u>		*C/mm 25*C	1.0480	· ₅	g <u> •K</u>		
B. P. °C 760 mm	123.3	2	BP	0.0485	4	h		ــــ
100	62.0	1 4	t _e	0.0372	5	f' to	1	1
30 10	35.0°		30 mm	0.6732	5	h'	1	1
1	-19.6		ΔHm cal/g	 	-	m to		+
Pressure	17.0		ΔHv cal/g 25°C	84.17	5	n K		į
mm 25°C	17.89	5 5	30 mm	83.34	5	<u> </u>		
Density	 		BP te t (d.e)	71.27 69.54	5 5	m' to		
g/ml 20°C			'e \-', -'	69.47	5	n' <u>°K</u> _	}	
d ₄ 30	0.7		ΔHv/T _e	19.05	5	<u> </u>		├
a	0.7	320 4	d 35 to		5	Surface tension dynes/cm. 20°C	20.83	5
ь	-0.0	384 4	$\frac{140}{a}$, $\frac{1}{25}$ to		5	30	19.86	5
Ref. Index		126 2	e' 35 °C	0.0807	5	Parachor [P]	18, 92	5
D 25	1.4		d _c g/ml	0.225	5	20°C		Ì
30	1.40		t _c *C	4.44 295.	5	30 40		
"C"	0.70		P _c mm	17998.	5	Sugd.	335.2	5
MR (Obs.) MR (Calc.			PV/RT		\vdash	Exp. L.1.%/wt.		\vdash
(nD-d/2)	1.0		25°C 30 mm	1.0000	5 5	u. Dispersion	121.	2
Dielectric			BP	0.9495	5	Flash Point °C	121.	+-
A 35 to		7632 4	te t _c	0.9388 0.254	5 5	Fire Point		
B 1_153 °C	1355.7 216.	4 4	ΔHc kcal/m	+	+	M Spec.		
A* 35 to	1.30	0868 5	ΔHf			Ultra V. X-Ray Dif.		1
B* 146 °C	1271.9	5	ΔFf	 	\vdash	Infrared		
c	_		Viscosity centistokes			Solubility in +	}	l
t _k To			∥ 7 ° ° ° ° °			Acetone Carbon tet.		
t'x 00		5338 5			1 1	Benzene Ether		1
B' _ 35 °C	1557.0	5			\vdash	n-Heptane		Ì
C'	234.	5	B ^V to	.]		Ethanol Water		
A'* 25 to B'* 35 °C		5858 5	(B ^V) - to	-1		Water in		
Ac 153 to	7 21		(A ^V) °C	1				
Bc tc °C	1676.	5	c _p liq. °K		\vdash			
Cryos. A°	251.	- 1 3	-11 -	İ				1
consts. B°			J P -					
t _e °C	136.40	5	c _v vap.					
TR = 0.75 Tc + grams/100 grams solvent								
REFEREN	CES: 1-D			Calc, from de	t. dat	ta 5-Calc. by for	mula	
SOURCE: API								
PURIFICATION: API								
LITERATURE REFERENCES:								

No. 71 cis-4-Octene NAME STRUCTURAL FORMULA CH3(CH2)2CH=CH(CH2)2CH3 Mole Ref. Molecular Molecular C8H16 Weight 112, 208 % Pur Formula Ref. Ref. Ref. F.P. °C -118.7 2 dt/dP f to F.P. 100% °C/mm °K g 25°C 1.0168 5 B. P. °C h ΒP 0.0485 760 mm 122.54 2 t_e 0.0372 5 f١ to 61.33 100 4 <u>°К</u> g' 5 30 34.45 4 30 mm 0.6721 10 13.98 5 h' ∆Hm cal/g -20.18 5 m to AHv cal/g Pressure °K n 25°C 83.89 5 5 18.46 mm 25°C 5 o 30 mm 83.13 te 1068. 5 BP 5 71.07 m' to Density te te (d, e) 69.36 5 n' g/ml 20°C °K 0.7212 2 69.29 ٥' d₄ 25 0.7170 2 ΔHv/T_e 19.04 5 30 0.7128 4 Surface tension 35 d 87.85 to 5 a 0.7380 4 21.53 dynes/cm, 20°C 0.1369 5 136 ь -0.0383 4 30 20.54 5 ďΠ to 25 85.91 5 40 19.58 5 35 °C Ref. Index e' 0.0806 5 20°C 1.4148 [P] nD Parachor d_c g/ml 0.226 5 25 2 1.4124 20°C vc ml/g t_°C 4.42 30 1.4097 4 30 ^tc 295. 5 40 "C" 0.7653 4 P_c mm 18155. 5 Sugd. 335.2 5 MR (Obs.) 38.95 2 PV/RT Exp. L 1.%/wt. MR (Calc.) 38.677 5 25°C 1.0000 5 u. (nD-d/2)1.0542 30 mm 1.0000 5 Dispersion 120. 2 Dielectric BP 0.9495 Flash Point C 0.9388 A 35 to 6.87149 Fire Point 0.254 B (153 °C 1351.0 4 M. Spec. С 216. 4 AHc kcal/m Ultra V 1.30477 1267.4 ΔHf A* 35 to 5 X-Ray Dif. ΔFf B* 146 °C Infrared K Viscosity Solubility in c centistokes Acetone to t_k Carbon tet t_x ٠c Benzene 25 to 7.25919 Ether 5 5 5 B 3<u>5</u> ℃ 1552.2 n-Heptane B_v C 234. to Ethanol °C A Water 25 to 35 °C A'* 5 1.65489 Water in B'* 5 (BV) 1452.5 to Acl 153 to 7.2826 5 (A^V) °C Bc_tc_ °C 1672. 5 ۰ĸ c_p liq. Cc 257. 5 Cryos. A° c_p vap. °K consts. B° c vap. t_e °C 135.54 5 $= 0.75 T_{c}$ $\overline{T_R}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 72		
NAME	trans-4-	Octen	е			STRUCTURAL FORMULA			
Mole % Pur.	Ref. M	olecul ormul		Molecular Weight 112.2	208	CH ₃ (CH ₂) ₂ CH=CH	i(CH ₂) ₂ CI	H ₃	
		Ref.			Ref			Ref	
F.P. °C	-93.810	2	dt/dP	1		f to		\top	
F.P. 100%			*C/mm		H	gK_			
B. P. *C			25°C BP	1.0047 0.0485	5 4	h ¦			
760 mm 100	122.25 61.06	2	t _e	0.0372	5	f' to			
30	34.19	4	30 mm	0.6717	5	g'			
10 1	13.74	5	ΔHm cal/g			h'		<u> </u>	
Pressure	1-20.40	+-	ΔHv cal/g			m to			
mm 25°C	18.71	5	25°C	83.78	5	n <u>*K</u>			
te	1067.	5	30 mm BP	83.04 70.98	5	ļ <u>-</u>		₩	
Density			t_	69.27	5	m' to			
g/ml 20°C	0.7141 0.7099	2	le (a, e)	69.20	5	0' '		1	
d ₄ 25 30	0.7057	4	ΔHv/T _e	19.03	5	Surface tension		+-	
a	0.7309	4	d 34 to	87.72	5	dynes/cm. 20°C	20.70	5	
ь	-0.0383	4	-a, - 1 - 140 ° €		5	30	19.73	5	
Ref. Index	.	١,	e' 34 °C		5	40	18.80	5	
n _D 20°C	1.4118	2 2	d g/ml v ml/g	0.223	5	Parachor [P] 20°C			
3 0	1.4068	4	v _c ml/g t _c °C	4.48	5	30			
"C"	0.7676	4		17879.	5	40 Sugd	335.2	5	
MR (Obs.)		2	P _c mm PV/RT	17077.	۲	Exp. L.1.%/wt.	333.2	+-	
MR (Calc. (nD-d/2)	38.677 1.0548	5 2	25°C	1.0000	5	u.			
Dielectric	1,0348	+-	30 mm	1.0000	5	Dispersion	122.	2	
A 34 to	6,8690	4	BP t _e	0.9495 0.9389	5	Flash Point °C			
	1349.0	4	tc	0.254	5	Fire Point	 	┼	
С	216.	4	ΔHc kcal/m			M Spec. Ultra V.			
A* 34 to			ΔHf ΔFf			X-Ray Dif.	1	1	
B* ∟145 °C	1265.5	5	Viscosity	 	\vdash	Infrared	ļ	1	
°	_ji		centistokes			Solubility in + Acetone			
tk to			უ °⊂			Carbon tet.			
A' 20 to		5 5				Benzene Ether			
B' _ 34 °C	1550.1	5	<u> </u>	<u></u>	L.,	n-Heptane	}		
С'	234.	5	B ^V to			Ethanol Water			
A'* 20 to B'* 34 °C		5 5		-		Water in			
Ac 152 to		5		1				T	
Bc t °C	1668.	5	<u> </u>	 	1				
Cc — —	257.	5	c _p liq. °K					1	
Cryos. A° consts. B°			c _p vap. °K						
t _e °C	135,22	5	c _v vap.	<u> </u>		l <u>, </u>	<u></u>		
$T_R = 0.7$						grams/100 gran		ıt	
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula		
SOURCE:			PI						
PURIFICA?			PI						
LITERATU	RE REFER	ENCES	S:						

No. 73 2-Methyl-1-heptene NAME STRUCTURAL FORMULA $CH_3(CH_2)_4C = CH_2$ ĊH3 Mole Ref. Molecular Molecular C8H16 Weight 112, 208 % Pur Formula Ref Ref. F. P. °C -90.0 2 dt/dP f to F.P. 100% °C/mm 25°C g °K 0.8953 5 4 B. P. °C h BP 0.0482 760 mm 119.3 2 t_e 0.0373 5 f 58.44 to 100 4 g' °<u>K</u> 30 31.74 4 30 mm 0.6672 5 10 11.43 5 h' AHm cal/g -22.45 5 m to ∆Hv cal/g Pressure °K 'n 25°C 82.81 5 mm 25°C 21.24 5 o 30 mm 82,27 5 5 1058. t_e 5 BP 70,25 m' to Density te (d, e) 68.59 5 n' ۰ĸ g/ml 20°C 0.7205 2 5 68.52 o' d_4^t 25 0.7164 2 AHv/Te 19.00 5 30 0.7123 4 Surface tension d 31 to 86,63 5 0.7369 dynes/cm. 20°C 5 21.45 å °C 0.1373 5 Ъ -0.0381 4 30 20.48 5 20 to 84.80 40 19.54 5 Ref. Index e' 31 °C 0.0796 5 n_D 20°C 1.4123 [P] Parachor d_c g/ml 5 0.226 25 2 1.4098 20°C vc ml/g t_ °C 30 4.43 1.4073 4 30 t_c 5 291. 40 "C" 0.7617 4 P_c mm 17989. 5 Sugd. 335.2 5 MR (Obs.) 38.78 2 PV/RT Exp. L. 1. %/wt. MR (Calc.) 38,677 5 25°C 1.0000 5 (nD-d/2)2 u. 1.0521 30 mm 1.0000 5 Dispersion 2 122. Dielectric BP 0.9495 Flash Point C 0.9391 5 A 31 to 6.85239 Fire Point 0.254 B 150 °C 1331.7 M. Spec. Ultra V. AHc kcal/m C 216. 4 ΔHf A* 31 to 1.28966 X-Ray Dif. ΔFf B*[135 °C 1249.0 Infrared ĸ Viscosity Solubility in centistokes Acetone to t_k Carbon tet. °C Benzene 20 to 7.24294 Ether B' 3<u>1</u> °C 1532.2 5 n-Heptane B_v | 234 5 Ethanol °C Water A'* 20 to B'* 31 °C 1.64083 5 Water in (B^V)| 1433.1 5 to Ac | 150 to 7,2629 5 (A^V)| °C 1649. Bc_tc_°C 5 cp liq. °K Cc 257. 5 c_p vap. Cryos. A °K consts. Bº c, vap. te °C 131.90 5 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 74	
NAME	3-Methyl	-l-he	ptene			STRUCTURAL	FORMUL.	A
			Т		-	сн ₃ (сн ₂) ₃ сн с	H=CH ₂	
Mole	Ref. Mo	lecul		Molecular		ĊH ₃		
% Pur.	Fo	_		Weight 112.2				
		Ref.		T	Ref			Ref.
F.P. °C		-	dt/dP		1 1	f to		
F. P. 100%	-	1	*C/mm 25*C	0,6480	5	g <u> </u>		ł
B. P. °C 760 mm	111.	2	BP	0.0476	4	h		+
100	50.98	4	t _e	0.0375	5	f' to g'K_		l
30 10	24.69 4.82	4 5	30 mm	0.6566	5	h' K_	1	1
i	-27.81	5	ΔHm cal/g	L		 	 	+
Pressure			ΔHv cal/g	70.74	5	m to		
mm 25°C	30.47	5	25°C 30 mm	79.74 79.78	5		1	
t _e	1035.	5	BP	68.20	5	m' to	 	+
Density g/ml 20°C	0.711	2	t _e (d, e)	66.70 66.64	5 5	n'		1
_d t 25	0.707	2	te (d, e) ΔHv/Te	1	5	o']	
4 30	0.703	4		18.91		Surface tension	<u> </u>	1
a b	0.727	4	d 20 to e 1 130 °C		5	dynes/cm. 20°C	20.34	5
	-0.038	4	d' to			30 40	19.41 18.50	5
Ref. Index n _D 20°C		2	e' i °C	 	-	Parachor [P]		†
D 25	1,404	2	d g/ml vc ml/g	0.220 4.54	5 5	20°C		
30	1.402	4	tc°°C′°	278.	5	30 40		
"C"	0.7607	4	P _c mm	17269.	5		335.2	5
MR (Obs.) MR (Calc.		2	PV/RT		+	Exp. L.1.%/wt.	 	T
(nD-d/2)	38.677	5 2	25°C	1.0000	5	u.		
Dielectric	1	┿	30 mm BP	1.0000 0.9500	5	Dispersion	117.	2
A 20 to	6.82779	4	te	0.9403	5	Flash Point °C Fire Point		
B 140 °C	1298.6	4	tc	0.256	5	M Spec.	 	+-
C	218.	4	ΔHc kcal/m ΔHf			Ultra V.		
A* 20 to B* 125 °C		5	ΔFf			X-Ray Dif.		1
K 123 C	- 1210.7		Viscosity		T	Infrared Solubility in +		+
·	-		centistokes			Solubility in +		
t _k to			η ∘c			Carbon tet.		
A' to	 	+				Benzene Ether		
B' •C			B ^V to		\vdash	n-Heptane		
C'		_	B' to			Ethanol Water		
A'* to B'* °C				4		Water in		
Ac 140 to		5						1
Bc t °C	1610.	5	<u> </u>	1	\vdash			1
Cc — —	259.	5	c _p liq. °K					
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	122.59	5	c _v vap.					
$T_R = 0.7$						grams/100 grai	ms solver	ıt
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:		Al	PI					
PURIFICAT	TION:	A	PI			-		
LITERATU	RE REFERE	NCES	S:					

No. 75 4-Methyl-1-heptene STRUCTURAL FORMULA NAME CH3(CH2)2CH CH2CH=CH2 Molecular C8H16 ĊHą Mole Ref. Molecular Weight 112.208 % Pur Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm 25°C g _°K 0.6943 5 B. P. °C h BP 0.0478 4 760 mm 112.8 2 0.0374 5 ſ١ 52.59 te to 100 4 g' °<u>K</u> 30 26, 20 4 30 mm 0.6593 5 10 6.13 5 h١ ∆Hm cal/g -27.34 5 m to AHv cal/g Pressure ۰ĸ n 25°C 80.33 5 28.23 mm 25°C 5 o 30 mm 80,26 5 te 1041. 5 BP 5 68. **6**6 m' to Density te te (d, e) 5 67.14 n' g/ml 20°C °K 0.717 0.7132 67.08 5 o' $\mathbf{d_4^t}$ 25 AHV/Te 5 18.94 30 0.709 4 Surface tension d 20 to 83,77 5 0.7330 dynes/cm. 20°C 4 21.03 ᇷᅴ 1<u>30 °C</u> 0.1339 5 ь -0.0₃794 20.10 30 to 19.19 5 40 Ref. Index e' ⁿD 20°C 1.410 [P] Parachor d_c g/ml 0.224 5 25 1.408 2 20°C vc ml/g t_°C 5 4.47 30 1.406 4 30 ^tc 281. 5 40 "C" 0.7614 4 P_c mm 17639. 5 Sugd. 335.2 5 MR (Obs.) 38.8 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 25°C 1,0000 5 (nD-d/2)1.052 2 u. 30 mm Dispersion 1,0000 5 117. 2 Dielectric BP 0.9500 5 Flash Point C 0.9401 A 20 to 6.88388 Fire Point 0.256 B 1143 °C 1309.3 4 M. Spec. Ultra V. AHc kcal/m C 218. 4 ΔHf A* 20 to 5 1.28143 X-Ray Dif. ΔFf B* 130 °C 1227.2 Infrared ĸ Viscosity Solubility in centistokes Acetone to $\mathbf{t_k^t_x}$ Carbon tet. °C Benzene A' to Ether В' °C n-Heptane B_v | C to Ethanol °C Water A1* to Water in B'* °C (B^V)| to Ac 143 to (AV) 7.2514 5 °C Bc tc °C 1624. 5 5 c_p liq. °K Cc 259. Cryos. A cp vap. °K consts. B° c_w vap. te °C 5 124.61 $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

								No. 76		
NAME	5- M	ethyl-	l -he	ptene			STRUCTURAL FORMULA			
							כט כט כט וכט	\ CU-CU		
				· · · · · · · · · · · · · · · · · · ·		\neg	CH ₃ CH ₂ CH (CH ₂	1/2CH-CH	2	
Mole % Pur.	Ref.	Mole Form	cula mula	с ₈ н ₁₆	Molecular Weight 112.2	08	CH ₃			
			Ref.			Ref	T		Ref	
F.P. °C				dt/dP			f to			
F.P. 1009	•			°C/mm	0.700/	. 5	g <u>"K</u>		1	
B. P. *C 760 mm	113.3		2	25°C BP	0.7096 0.0478	4	h		<u> </u>	
100	53.09		4	t _e	0.0374	5	f' to		1	
30 10	26.70		4 5	30 mm	0.6593	5	h'			
1	-26.83		5	ΔHm cal/g		\vdash	m l to		\vdash	
Pressure mm 25°C	37.5		_	ΔHv cal/g 25°C	80.76	5	n K			
mm 25-C	27.51 1042.0	·	5	30 mm	80.53	5	°			
Density	 			BP te	68.84 67.29	5	m' to			
g/ml 20°0	0.71		2 2	t _e (a, e)	67.24	5	n' ' <u>°K</u> -	-	l	
d ₄ 25	0.70		4	ΔHv/T _e	18.96	5	Surface tension		╁	
	0.73		4	d 15 to e 1 135 °C		5	dynes/cm. 20°C	21.03	5	
b D. C. T. I.	-0.0	83	4	d' i to	5		30 40	20.10 19.19	5	
Ref. Index		94	2	e' ' •(+	Parachor [P]		Ť	
D 25 30	1.40		2	d g/ml vc ml/g tc °C	0.222 4.50	5	20°C		1	
"C"	0.76		4	-	281.	5	30 40		ł	
MR (Obs.		_	2	P _c mm	17537.	5	Sugd.	335.2	5	
MR (Calc.	38.67	77	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.		1	
(nD-d/2)	1.05	12	2	30 mm	1.0000	5	Dispersion	117.	2	
Dielectric		1061	4	BP t _e	0.9500 0.9401	5	Flash Point °C			
B 1143°		,,,,,	4	tc	0.256	5	Fire Point		┼	
<u>C</u>	217.47		4	ΔHc kcal/m ΔHf			M Spec. Ultra V.			
A* 20 to B* 130 °C		3101	5	ΔFf			X-Ray Dif. Infrared			
K	-			Viscosity			Solubility in +		╁	
t _k			i	r centistokes			Acetone			
'x 1					1		Carbon tet. Benzene			
A' to							Ether			
č, – – -	<u>-</u>		1	B ^V l to			n-Heptane Ethanol			
A'* to				A ^V I °C	_		Water Water in			
B'* *(-	_	(B ^V) to	1		Water III		\vdash	
Ac 143 to	7.25 C 1622.	01	5	(A ^V) °C		-				
Ce	436.		5	c _p liq. °K					1	
Cryos. Acconsts. B				c _p vap. °K						
t _e °C	125.16	· [5	c _v vap.						
$T_R = 0.7$							f grams/100 grai		t	
REFEREN	CES: 1-D	ow 2	-AP		Calc. from de	t. dat	ta 5-Calc. by for	mula		
SOURCE:	TION		AF		······································					
LITERATU		ERENC	AF CES							
		JI DIN	-	•						

No. 77 6-Methyl-1-heptene NAME STRUCTURAL FORMULA CH3CH (CH2)3CH=CH2 ĊH₃ Mole Ref Molecular Molecular C₈H₁₆ Weight 112, 208 % Pur Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C ۰<u>K</u> g 0.7511 5 B. P. ℃ h BP 0.0465 4 760 mm 113.2 2 t_e 0.0363 5 f† to 100 54.34 4 g' <u>°K</u> 30 28.36 0.6505 5 30 mm 10 8.37 5 h' ∆Hm cal/g 5 -24.8 to ∆Hv cal/g Pressure n ۰ĸ 83.00 25°C 5 mm 25°C 25, 25 o 5 30 mm 82.53 1041. 5 t_e ВP 70.73 5 m' to Density 5 69.19 te te (d, e) n' g/ml 20°C 0.7120 5 2 69.13 o١ d_4^t 25 0.7079 2 19.51 5 AHv/Te 30 0.7038 4 Surface tension d 20 to 86,48 5 0.7284 a 4 dynes/cm. 20°C 20.45 0.1391 å 1<u>25 °C</u> ь -0.0381 4 19.52 5 30 to 5 40 18,61 Ref. Index e' °C [P] $\mathbf{n}_{\mathbf{D}}$ 20°C 1.4070 Parachor 0.240 5 d_c g/ml 25 1.4045 2 20°C vc ml/g t °C 5 4.167 1.4018 30 4 30 t_c 280. 5 40 "C" 0.7610 4 5 18887. P_c mm Sugd. 335, 2 5 MR (Obs.) 38.79 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 1.0000 25°C 5 (nD-d/2)1.0510 2 30 mm 1.0000 5 Dispersion 2 117. 0.9500 Dielectric BP 5 Flash Point °C 0.9404 5 A 20 to 6.94874 1345.2 5 Fire Point 0.256 5 B 142 °C 5 M. Spec. С 217,49 5 AHc kcal/m Ultra V. A* 20 to ΔHf 1.39165 5 X-Ray Dif. ΔFf B*[135 °C 1263.4 Infrared ĸ Viscosity Solubility in centistokes Acetone to _k | Ն | Carbon tet. °C Benzene 10 to 7.37919 Ether B' į 20 °C 1554.37 5 n-Heptane B_v | 235.49 5 to Ethanol °C Water 10 to 5 1.78823 Water in B'* 20 °C (B^V) 1457.8 5 to Ac| 142 to 7.3591 5 (A^V)| °C Bc tc °C 1656. c_p liq. ۰ĸ Cc' 257. 5 Cryos. A. c_p vap. °K consts. B° c_v vap. te °C 5 124.7 $T_{\mathbf{R}} = 0.75 \, \mathbf{T_{\mathbf{c}}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

							No. 78
NAME	2-Methy	/1-2-l	eptene			STRUCTURAL	FORMULA
						СН ₃ (СН ₂) ₃ СН	=C CH ₂
Mole	Ref. M	olecul	ar C ₈ H ₁₆	Molecular		3 23	Ċнз
% Pur.	F	ormul	a C8H16	Weight 112.2	08		
		Ref.		,	Ref.		Rei
F.P. C F.P. 100%		—	dt/dP			f to	1
B. P. *C	<u>'</u>	╁	*C/mm 25°C	1.0857	· 5	g	
76 0 mm	122.6	2	BP	0.0474 0.0363	4 5	f' to	
100 30	62.51 35.98	4	t _e 30 mm	0.6646	5	g' K_	
10	15.72	5	ΔHm cal/g	0.0010	 	h'	
1	-18,21	5	ΔHv cal/g		 	m to	
Pressure mm 25°C	16.90	5	25°C	85.82	5	n •K_	1
t _e	1067.3	5	30 mm BP	84.91 72.71	5	<u> </u>	
Density g/ml 20°C	0.7341	١,	t _e	70.99	5	m' to	
at 25	0.7241 0.7200	2 2	te (d, e)	70.92	5	o' ' ====	
	0.7159	4	ΔHv/T _e	19.50 89.89	5	Surface tension	
a b	0.7405 -0.0 ₃ 82	4	e 135 °C	0.1408	5	dynes/cm. 20°C	21.88 5 20.90 5
Ref. Index		+-	d' 20 to	87.89 0.0829	5 5	40	19.95 5
n _D 20°C	1.4170	2	 	0,242	5	Parachor [P]	
25 30	1.4145	2 4	d g/ml vc ml/g	4.125	5	20°C 30	
"C"	0.7661	4	1 6	296. 19555.	5	40	
MR (Obs.)		2	P _c mm	17555.	-	Exp. L.1.%/wt.	335,2 5
MR (Calc. (nD-d/2)	38.677 1.0550	5 2	25°C	1.0000	5	u.	
Dielectric	1.0350	+-	30 mm BP	1.0000	5	Dispersion	124. 2
A 36 to	6.95927	5	te	0.9391	5	Flash Point °C Fire Point]]
B (154 °C	215.7	5	t _c	0,255	5	M Spec.	
A* 36 to		5	∆Hc kcal/m ∆Hf		ŀ	Ultra V. X-Ray Dif.	<u> </u>
B* 145 °C		5	ΔFf		<u> </u>	Infrared	
K — — -	i		Viscosity centistokes		1	Solubility in +	
t _k Tto		1	7 °℃	İ		Acetone Carbon tet.	1 1
A' 20 to		5				Benzene	
B' _ 36 °C		5				Ether n-Heptane	
C'	233.7	5	B ^V to A ^V °C		Ì	Ethanol Water	
A'* 20 to B'* 36 °C		5	(B ^V)to			Water in	
Ac 154 to	7.3692	5	(A ^V) °C				
Bc tc_°C	1	5	c _p liq. °K		<u> </u>		
Cryos, A°	256.	+ -	11				
consts. B°			ъ				
t _e °C	135.29	5	c _v vap.				
$T_R = 0.7$						grams/100 gran	ns solvent
	CES: 1-Dow			alc. from de	t. da	ta 5-Calc. by for	mula
SOURCE:			PI				
PURIFICAT			.PI				
₩TERATU	RE REFERE	NCES	5:				

								No. 79	
NAME	3-	Methy	l-cis	-2-heptene			STRUCTURAL	FORMULA	A.
							CH3(CH2)3C	= CH CH ₃	
Mole	D.	f. Mo	lecul		Molecular			:н ₃	
% Pur.		Fo	rmula	ar C ₈ H ₁₆	Weight 112.20	8 8		,	
			Ref.			Ref.			Ref.
F.P. °C	 		<u></u>	dt/dP			f to		
F.P. 100%	'		\vdash	°C/mm 25°C	1.0610	5	g '°I		
B. P. °C 760 mm	122.		2	BP	0.0473	4	h	-	
100 30	62.	5	4	t _e	0.0363	5	f' to		
10	35. 15.		5	30 mm	0.6637	5_	h' ;	-	
1	-18.	61	5	ΔHm cal/g ΔHv cal/g		h	m to		
Pressure mm 25°C	17.	33	5	25°C	85.63	5	n •I	2	
t _e	1066.	,,	5	30 mm BP	84.76 72.59	5			
Density		720		t_	70.89	5	m' to		
g/ml 20°C		729 725	2 2	t _e (d, e)	70.82	5	 	7	
d ₄ 25 30		721	4	d 35 to	19.50	5	Surface tension	1	
a b		745	4	d 35 to e 135 °C	89.76 0.1407	5	dynes/cm. 20°C	22.48	5
Ref. Index		038	4	d' 25 to e' 35 °C	87.69 0.0826	5 5	8 30 40	21.50	5 5
n _D 20°C	1.	419	2	d _c g/ml	0. 246	5	Parachor [P]		
25		417 414	2 4	v ml/g	4.072	5	20°C		
"C"		7643	4	۱_د	296.	5	40		i
MR (Obs.)			2	P _c mm	19812.	5		335.2	5
MR (Calc. (nD-d/2)) 38.	677	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt	·	
Dielectric	1.	055	2	30 mm BP	1.0000 0.9495	5	Dispersion	124.	2
A 35 to	6	95905	5		0.9391	5	Flash Point °C Fire Point		
B 154 °C	1377.	7	5	te tc	0. 255	5	M. Spec.	-	
C A* 35 to	215.	39354	5	∆Hc kcal/m ∆Hf	1		Ultra V.		
B* 145 °C			5	ΔFf			X-Ray Dif. Infrared		
K				Viscosity centistokes			Solubility in +		
t _k to				η °C			Acetone Carbon tet.		
t _x °C							Benzene		
A' 25 to B' 35 °C	1582.	35168 1	5 5				Ether n-Heptane		
C'	233.	82	5	B ^V to A ^V °C			Ethanol		
A'* 25 to B'* 35 °C		74680 4	5	1 4-V.			Water Water in		
Acl 154 to		3693	5	(B') to					
Bc tc °C	1698.		5	c _p liq. °K			1		
Cryos. A°	256.		•	1					
consts. B°				c _p vap. °K					
t _e °C	134.	61	5	c _w vap.					
$T_R = 0.7$	5 T _C				*		grams/100 gr	ams solvent	t
REFERENC	CES: 1	-Dow			Calc. from de	t. da	ata 5-Calc. by fo		
SOURCE:				PI					
PURIFICAT				PI			·····	******	
LITERATU	RE RE	FERE	NCES	5:					

							No. 80	
NAME	3-Methyl-	tran	s-2-heptene		_	STRUCȚURAL	FORMULA	
						$CH_3(CH_2)_3C =$	снсн3	
Mole % Pur,	Ref. Mo	lecul rmul	ar C ₈ H ₁₆	Molecular Weight 112.2	80:	Ċн _з	1	
,, <u>, , , , , , , , , , , , , , , , , ,</u>		Ref.			Ref.		P	Ref.
F.P. *C			dt/dP			f to		
F. P. 100%			*C/mm 25*C	1.0610	. 5	g <u>*K</u> _		
B. P. °C 760 mm	122.	2	BP	0.0473	4	h	<u> </u>	
100	62.5	4	t _e	0.0363	5	f' to		
30 10	35.50 15.27	4 5	30 mm	0.6637	5	g'	1	
1	-18.61	5	ΔHm cal/g			 		
Pressure			ΔHv cal/g 25°C	85.63	5	m to		
mm 25°C	17.33 1066.	5	30 mm	84.76	5	•		
Density	1000.	۲	BP	72.59 70.89	5	m' to		
g/ml 20°C	0.729	2	te (d, e)	70.82	5	n' °K		
d ₄ 25	0.725 0.721	2 4	AHv/T _e	19.50	5	<u> </u>	ļļ.	
1 30	0.745	4	d 35 to		5	Surface tension dynes/cm. 20°C	22,48	5
Ъ	-0.038	4	135 to		5	30	21.50	5
Ref. Index			e' 35 °C		5	40	20,55	5
ⁿ D 20°C	1.419	2 2	d _c g/ml	0.246	5	Parachor [P] 20°C		
30	1.414	4	t _c *C	4.072 296.	5	30		
"C"	0.7643	4	P _c mm	19812.	5	40 Sugd	335.2	5
MR (Obs.)	38.9	2	PV/RT	17012.	-	Exp. L.1.%/wt.	333.2	<u> </u>
MR (Calc.) (nD-d/2)	38.677 1.055	5 2	25°C	1.0000	5	u.		
Dielectric			30 mm BP	1.0000 0.9495	5	Dispersion	124.	2
A 35 to	6.95905	5	te	0.9391	5	Flash Point °C Fire Point	ļ į	
B <u> 154 °C</u> C		5	t _c	0.255	5	M Spec.		
A* 35 to	215, 82	5	ΔHc kcal/m ΔHf			Ultra V.		
B* 145 °C	1.39354 1294.5	5	ΔFf			X-Ray Dif. Infrared		
K — — —			Viscosity			Solubility in +		
1k 10	1		centistokes 7 °C			Acetone		
t _x ; *C			1			Carbon tet. Benzene		
A' 25 to B' 35 °C	7.35168 1582.1	5				Ether		
c, L 33 C	233.82	5	B ^V to			n-Heptane Ethanol		
A** 25 to	1.74680	5	AV C			Water		
B'* 35 °C		5	(B ^V) to			Water in	 -	
Ac 154 to Bc t °C	7.3693 1698.	5	(A ^V) °C					
Cc - c -	256.	5	c _p liq. °K					
Cryos, A° consts, B°			c _p vap. °K					
te °C	134.61	5	c _w vap.					
$T_{\mathbf{R}} = 0.75$	T _c					+ grams/100 gran	ns solvent	_
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from de	t, da	ta 5-Calc. by for	mula	_
SOURCE:			PI					
PURIFICAT			PI					
LITERATUE	RE REFEREN	NCES	:					

							No. 81		
NAME _	4-Methyl	-cis-	2-heptene			STRUCTURAL FORMULA			
						сн ₃ (сн ₂) ₂ сн с	:Н=СН СН.	_	
Mole % Pur.	Ref. Mo	lecul rmul	ar с ₈ н ₁₆	Molecular Weight 112,20	8	ĊH ₃		3	
		Ref.			Ref.			Ref.	
F, P, °C			dt/dP			f to			
F.P. 100%	 		°C/mm 25°C	0,7791	5	g '° <u>K</u>		İ	
B. P. °C 760 mm	114.	2	BP	0.0465	4	h			
100	55.05	4	t _e	0.0363	5	f' to			
30 10	29.03 9.17	4 5	30 mm	0.6516	-5-	h' <u> </u>			
1	-24.09	5	ΔHm cal/g	ļ	\vdash	m to		 -	
Pressure	24.22	_	ΔHv cal/g 25°C	83.06	5	n			
mm 25°C t _e	24.33 1043.	5	30 mm	82.76	5	0			
Density			BP t.	70.93 69.37	5	m' to			
g/ml 20°C		2	t _e (d, e)	69.32	5	n'		ĺ	
d ^t 25 4 30	0.712 0.708	2 4	AHv/T _e	19.52	5			<u> </u>	
a	0,732	4	d 29 to	86.80	5	Surface tension dynes/cm. 20°C	20.92	5	
ь	-0.038	4	126 °C	0.1392 84.95	5 5	8 30	19.99	5	
Ref. Index			e' 29 °C	0.0754	5	40	19.08	5	
ⁿ D 20°C	1.410 1.408	2	d _c g/ml	0, 242	5	Parachor [P]			
30	1.405	4	vc ml/g tc °C	4. 126 282.	5	30			
"C"	0.7624	4	P _c mm	19144.	5	40 Sugd	335.2	5	
MR (Obs.)	38.9	2	PV/RT	17,223	┽┥	Exp. L. 1. %/wt.	333.2	<u> </u>	
MR (Calc.) (nD-d/2)	38.677 1.052	5 2	25°C	1.0000	5	u.			
Dielectric			30 mm BP	1.0000 0.9500	5 5	Dispersion	119.	2	
A 29 to	6.95106	5	te	0.9403	5	Flash Point °C Fire Point			
B (144 °C)	1348.6 217.34	5 5	^t c	0.256	5	M. Spec.		-	
A* 29 to	1.39312	5	ΔHc kcal/m ΔHf]		Ultra V.			
B* 136 °C	1266.7	5	ΔFf			X-Ray Dif. Infrared			
K — — —			Viscosity centistokes		1	Solubility in +		_	
tk to	·}		η °C			Acetone			
<u> </u>	<u> </u>		•		} }	Carbon tet. Benzene			
A' 15 to B' 29 °C	7.35098 1552.9	5				Ether			
_c, '	235.34	5	B ^V to			n-Heptane Ethanol	}		
A'* 15 to	1.74916	5	AV °C	.		Water Water in			
B'* 29 °C	1453.3	5	(B ^V) to			water in	 	-	
Ac 144 to Bc tc °C	7.3615 1661.	5	(A ^V) °C	 	\vdash				
Cc	257.	5	c _p liq. °K						
Cryos, A° consts, B°			c _p vap. °K						
t _e °C	125.59	5	c _v vap.					L	
T _R = 0.75 T _C									
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for	mula		
SOURCE:		A	PI			_ ·			
PURIFICAT			PI						
LITERATU	RE REFERE	NCES	S:						

							No. 82		
NAME	4-Methyl	-tran	s-2-heptene			STRUCTURAL	FOR MUL	A.	
l_						СН ₃ (СН ₂) ₂ СН СН=СНСН ₃			
Mole		lecul		Molecular		сн ₃	_	•	
% Pur.	1 1 5	rmul		Weight 112.2	_			Ref	
	r	Ref.		T	Ref		Π	Ker	
F.P. *C F.P. 100%		+	dt/dP •C/mm		1	f to			
B. P. *C	<u> </u>	\vdash	25°C	0.7791	5	h .			
760 mm	114.	2	BP	0.0465 0.0363	5	f' to		+	
100 30	55.05 29.03	4	t _e 30 mm	0.6516	5	g' 'K_			
10	9.17	5		0.0310	┼┤	h'			
1	-24.09	5	ΔHm cal/g	 	\vdash	m l to		\vdash	
Pressure mm 25°C	24.33	5	ΔHv cal/g 25°C	83.06	5	n •K_	ļ	1	
t _e	1043.	5	30 mm	82.76	5	<u> </u>			
Density		1	BP t_	70.93	5 5	m' to		1	
g/ml 20°C	0.716	2	te te (d, e)	69.32	5	n' <u>°K</u> _	-	1	
d ^t 25 4 30	0.712 0.708	2 4	AHv/Te	19.52	5			—	
	0,732	4	d 29 to		5	Surface tension dynes/cm. 20°C	20.92	5	
ь	-0.038	4	$\frac{126}{4}$, $\frac{1}{15}$ $\frac{126}{to}$		5 5	30	19.99	5	
Ref. Index			e' 29 °C		5	40	19.08	5	
ⁿ D 20°C	1.410	2 2	d g/ml vc ml/g	0.242	5	Parachor [P]			
30	1.405	4	vc ml/g	4.126	5	20°C 30			
"C"	0.7624	4	ic C	282.	5	40		١.	
MR (Obs.)	38.9	2	P _c mm	19144.	5		335.2	5	
MR (Calc.)	38.677	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.			
(nD-d/2)	1.052	2	30 mm	1.0000	5	Dispersion	121.	2	
Dielectric	<u> </u>	<u> </u>	BP	0.9500 0.9403	5 5	Flash Point °C			
A 29 to B 1 144 °C	6.95106 1348.6	5	t _e t _c	0.256	5	Fire Point		 	
c TT S	217.34	5	ΔHc kcal/m	 	\vdash	M Spec.			
A* 29 to	1.39312	5	ΔHf ΔFf		i i	Ultra V. X-Ray Dif.		1	
B* 136 °C	1266.7	5	i 	 		Infrared			
c			Viscosity centistokes			Solubility in +			
tk to		}	η ∘c			Acetone Carbon tet.			
-x 1						Benzene			
A' 15 to B' 29 °C	7.35098 1552.9	5				Ether n-Heptane		1	
c,	235.34	5	B ^V to			Ethanol			
A** 15 to	1.74916	5	AV C			Water Water in	ļ		
B'* 29 °C	1453.3	5	(B ^V) to			water in		+	
Ac 144 to Bc t _c °C	7.3615 1661.	5	(A ^V) °C						
Cc	257.	5	c _p liq. ∘K						
Cryos. A° consts. B°			c _p vap. °K						
t _e °C	125.59	5	c _w vap.					}	
$T_{\mathbf{R}} = 0.75$	T _c					grams/100 grai	ms solven	ıt	
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from de	t. dat	ta 5-Calc. by for			
SOURCE:		AI							
PURIFICAT	ION:	AI	PI						
LITERATUR	E REFERE	NCES	S:						

							No. 83
NAME	5-Methyl	-cis-	-2-heptene			STRUCTURAL	FORMULA
						сн ₃ сн ₂ сн сн ₂	CH=CHCH3
Mole % Pur.		ecul		Molecular Veight 112.	208	Ċн ₃	_
		Ref.			Ref.		Re
F.P. °C F.P. 100%			dt/dP °C/mm			f to	
B. P. °C 760 mm	110	2	25°C BP	0.9078 0.0469	5 4	h	
100	118. 58.51	4	t _e	0.0363	5	f' to	
30 10	32.24 12.19	4 5	30 mm	0.6577	5_	h'	
1	-21.38	5	ΔHm cal/g		-	m to	
Pressure mm 25°C	20.57	5	ΔHv cal/g 25°C	84.31	5	n *K	
t _e	20.57 1055.	5	30 mm BP	83.73 71.76	5		
Density			t _e (d, e)	70.12	5	m' to	
g/ml 20°C	0.723	2 2		70.06	5	0,	
d ₄ 25 30	0.715	4	d 32 to	19.51	5	Surface tension	
a b	0.739 -0.0 ₃ 8	4	_e_ _130 °C	88.24 0.1397	5	dynes/cm. 20°C	21.75 5 20.79 5
Ref. Index	-0.030	1	d' 20 to e' 32 °C	86.30 0.0797	5	. 40	19.86 5
n _D 20°C	1	2	d _c g/ml	0.0171	5	Parachor [P]	
25 30	1.412	2	v _c ml/g	4.106	5	20 °C 30	
"C"	0.7620	4	-	289. 19479.	5	40	335.2 5
MR (Obs.)		2	P _c mm	17417.	-	Exp. L.1.%/wt.	335.2 5
MR (Calc. (nD-d/2)	38.677	5 2	25°C	1.0000	5	u.	
Dielectric			30 mm BP	1.0000 0.9500	5	Dispersion	119. 2
A 32 to	6.95360	5	t _e	0.9400	5	Flash Point C Fire Point	
B 1149 °C	1362.7 216.58	5	tc ΔHc kcal/m	0,256	-	M. Spec.	
A* 32 to	1,39140	5	∆Hf	:		Ultra V. X-Ray Dif.	
B*[135 °C K	1280.0	5	ΔFf Viscosites			Infrared	
c			Viscosity centistokes			Solubility in + Acetone	
t _k to			η °C			Carbon tet.	
A' 20 to	7,34977	5				Benzene Ether	
B' 1_32 °C	1567.0	5 5	BV I		 	n-Heptane	
A'* 20 to	1.74643	5	B ^V to C			Ethanol Water	
B'* 32 °C		5	(B ^V) to			Water in	
Acl 149 to	7.3640 1679.	5 5	(A ^V) °C		<u> </u>		
Bc t _c °C	- 257.	5	c _p liq. °K				
Cryos. A° consts. B°			c _p vap. °K				
t _e °C	130.12	5	c _v vap.	1			
$T_R = 0.7$	T _c					grams/100 gra	ms solvent
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for	mula
SOURCE:			PI				
PURIFICA			PI				
LITERATU	RE REFERE	NCE	5:				

							No. 84		
NAME	5- M e	thyl-tran	s-2-heptene		- 1	STRUCTURAL FORMULA			
Γ									
	T		·		-	CH ₃ CH ₂ CHCH ₂ C	CH=CH C	¹ 3	
Mole	Ref.	Molecul	ar C ₈ H ₁₆	Molecular		ĊН ₃			
% Pur,			a 08-16	Weight 112.2	_			-	
		Ref.		γ	Ref		, 	Ref	
F.P. °C	 		dt/dP	1		f to		1	
F.P. 100%			*C/mm 25*C	0.9078	5	g <u>K</u>			
B. P. *C 760 mm	118.	2	BP	0. 0469	4	h		<u> </u>	
100	58.51	4	t _e	0.0363	5	f' to			
30	32.24	4 5	30 mm	0.6577	5	g' K_	1	1	
10 1	12.19 -21.38	5	∆Hm cal/g			h'	ļ	ــــــــــــــــــــــــــــــــــــــ	
Pressure	1	-+-	ΔHv cal/g			m to			
mm 25°C	20.57	5	25°C	84.31	5	n •K	1		
t _e	1055.	5	30 mm BP	83.73 71.76	5 5	ļ		╄	
Density	T	\neg	t_	70.12	5	m' to			
g/ml 20°C			t _e (d, e)	70.06	5	n' ' •K	1		
d ₄ 25	0.71		AHv/T _e	19.51	5		<u> </u>	+-	
	0,73		d 32 to		5	Surface tension dynes/cm. 20°C	21.75	5	
b	-0.03		a, 1 130 °C		5 5	30	20.79	5	
Ref. Index			e' 32 °C		5	40	19.86	5	
n _D 20°C			d g/ml	0,244	5	Parachor [P]			
25 30	1.41		d g/ml vc ml/g	4, 106	5	20°C 30			
"C"	0.76		t _c •C	289.	5	40		1	
	+		P _c mm	19479.	5	Sugd.	335.2	5	
MR (Obs.) MR (Calc.	38.8 38.67	7 2 5	PV/RT		\vdash	Exp. L.1.%/wt.			
(nD-d/2)	1.05		25°C	1.0000	5	u.		1.	
Dielectric			30 mm BP	1.0000	5 5	Dispersion	121.	2	
A 32 to	6.95	360 5	te	0.9400	5	Flash Point °C Fire Point		İ	
B 1149 °C	1362.7	5	t _c	0.256	5			+	
С	216.58	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		1	
A* 32 to	1.39	140 5	ΔFf	İ		X-Ray Dif.		1	
B* 135 ℃	- 1280.0	"	Viscosity		+	Infrared			
c	_		centistokes			Solubility in + Acetone		1	
tk to			η •c			Carbon tet.		Ī	
		077			1 1	Benzene			
A' 20 to B' 32 °C		977 5				Ether n-Heptane			
č, – – –	234.58	5	B ^V to			Ethanol			
A!* 20 to			AV I _ °C	_		Water			
B'* 32 °C	+	5	(B ^V) to			Water in	 	+	
Ac 149 to	7.36		(A ^V) °C						
Bc tc_°C	1679. 257.	5 5	cp liq. °K						
Cryos, A°	123	- -	_	ļ.					
consts. B°	1		c _p vap. °K						
t _e °C	130.12	5	c _w vap.					1	
$T_{\mathbf{R}} = 0.7$	5 T _c		L		ــــــــــــــــــــــــــــــــــــــ	+ grams/100 gra	me es\		
REFERENC		ow 2-AF	PI 3-Lit. 4-0	Calc. from de	t d=0			16	
SOURCE:		AP		de	ual	- J-Care, by for			
PURIFICAT	TION:	AP							
LITERATU									
MIERAIU	RE REFI	ERENCES	:						

No. 85 6-Methyl-cis-2-heptene STRUCTURAL FORMULA NAME CH3CH (CH2)2CH=CHCH3 Molecular C8H16 ĊН3 Mole Ref. Molecular % Pur Weight 112, 208 Ref. Ref Ref. **F**, P, °C dt/dP f to F.P. 100% *C/mm 25*C °K g 0.8733 5 B. P. C h BP 0.0468 4 760 mm 117. 2 0.0363 5 f١ to 100 57.63 4 <u>°К</u> g' 30 30 mm 31.43 4 0.6563 5 10 5 11.43 h! ∆Hm cal/g 5 1 -22.07 to m AHv cal/g Pressure °K n 25°C 83.98 mm 25°C 21.47 5 o 30 mm 83.48 1052. te 5 BP 5 71.53 m' to Density te (d, e) 5 69.92 n' g/ml 20°C °K 0.718 2 69.86 5 01 25 0.714 d_4^t ΔHv/T_e 19.51 5 30 0.710 4 Surface tension 31 to 87,86 5 0.734 44 21.15 5 dynes/cm. 20°C 129 °C 0.1396 5 -0.0₃8 ь 20.22 5 to 30 d٦ 20 85.95 40 19.31 5 Ref. Index e' 31 °C 0.0788 5 20°C 1.412 1.410 2 [P] $\mathbf{n}_{\mathbf{D}}$ Parachor d_c g/ml 5 0.242 25 20°C vc ml/g 5 4.138 30 1.407 4 30 5 287. t_c 40 "C" 0.7638 4 P_c mm 19262. 5 335.2 5 Sugd. MR (Obs.) 38.9 38.677 2 PV/RT Exp. L. l. %/wt. MR (Calc.) (nD-d/2) 25°C 1,0000 u. 1.053 2 30 mm 5 119. 2 1.0000 Dispersion Dielectric BP 0.9500 Flash Point C 0.9401 6.95222 A 31 to Fire Point 5 1358.9 0.256B 147 °C M. Spec. С 216.77 5 AHc kcal/m Ultra V. A* 31 to B* 139 °C ΔHf 1.39108 5 X-Ray Dif. ΔFf 1276.4 Infrared Viscosity Solubility in centistokes Acetone to Carbon tet. ليكا °C Benzene 20 to 7.34929 5 Ether 1563. 2 В' 31 °C n-Heptane B_v C 234.77 5 to Ethanol °C A'* 20 to Water 1.74633 5 Water in B'* 31 °C (B^V) 1463.5 to Ac 147 to 7.3625 (A^V) 5 °C Bc tc °C °C c_p liq. ۰ĸ 257. 5 Cryos, Aº c_p vap. °K consts. B° te °C c_v vap. 128, 99 5 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** LITERATURE REFERENCES:

NAME							— т		No. 86		
Mole Ref. Molecular C8H16 Molecular Weight 112.208 CH3	NAME	6-Me	thyl-	tran	s-2-heptene			STRUCTURAL FORMULA			
Mole Ref. Molecular C8H16 Molecular Weight 112.208 CH3								CH_CH (CH_)_CH=CHCH_			
	N/-1-	2.4				Malazulan			,	3	
Ref		Rei.			ar C ₈ H ₁₆		208	3			
F. P. 'C 100%				Ref.						Ref	
F.P. 100% B.P. °C 1760 mm 177. 63	F. P. *C	T			dt/dP	1		f l to			
Too mm					*C/mm		١. ا	1 1 1 11		i	
100							1 - 1	h		1	
30		57.63			16			f' to			
1	30	31.43	;	4		0.6563	5	- :	-		
Pressure mm 25°C					ΔHm cal/g			h'		<u> </u>	
The color of the		+		Ť	ΔHv cal/g		1			1	
Nemathy Section Sect		21.47	,	5					ł	İ	
Density g/ml 20°C 0.718 2 2 4 25 0.714 4 4 30 0.710 4 4 4 30 0.710 4 4 4 4 4 4 4 5 6 9.92 5 6 1.9 5 5 6 1.9 5 5 6 1.9 5 5 6 1.9 6 5 6 1.9 6 5 6 1.9 6 5 6 1.9 6 5 6 1.9 6 5 6 1.9 6 5 6 1.9 6 5 6 1.9 6 5 6 1.9 6 5 6 1.9 6		1052.		5				 	<u> </u>	┼	
A 31 to 1.338.9 2 MR (Calc.) 33.6.77 4 31 to 33.6.77 5	Density				t _e ,	69.92	5			1	
A 30					e (a, e)	1			1		
a 0,734 4 e 129 °C 0,1396 5 5 5 5 20,22 Ref. Index	^a 4 30						-	Surface tension		+	
Ref. Index np 20°C 1.412 2 2 1.410 2 2 2.5 30 1.407 4 20 to 85.95 5 40 19.31				- 1				dynes/cm. 20°C		5	
No. 20°C 1.412 2 1.410 2 2.55 1.407 4 2.00 2.87 5 2.87 5 2.87 5 4.00 4 2.87 5 4.00 4 2.87 5 4.00 4 4.138 5 2.87 5 4.00 4			8	4	d' 20 to	85.95		J		5	
1,410		1	2	2		1	_		17.31	+-	
NR (Obs.) 38.9 2 P_c mm 19262. 5 Sugd. 335.2	45	1.41	0	2	d g/ml						
MR (Obs.) 38.9 38.67 19.262. 5 Sugd. 335.2		+		-	tc °C						
MR (Calc.) (nD-d/2) NR (Calc.) Calc. Calc. Calc. NR (Calc.) NR (Calc.) NR (Calc.) NR (Calc.) NR (Calc.) NR (Calc.) Calc. NR (Calc.) Solution Calc. Cal		+	38	\vdash	P _c mm	19262.	5	n	335.2	5	
Dispersion 121.			7		PV/RT			Exp. L.1.%/wt.			
BP								u.			
A 31 to 1358.9 5 tc 0.256 5 Fire Point A* 31 to 1.39108 5 AHc kcal/m AHf AFf A* 31 to 1.39108 5 AFf A* 20 to 7.34929 5 C 234.77 5 A* 20 to 1.74633 5 C 234.77 5 A* 31 °C 1563.2 234.77 5 B* 31 °C 1643.5 5 (B*) to A* 20 to 1.74633 5 (B*) to A* 20 to 1.74633 5 (B*) to A* 20 to 1.74633 5 (B*) to A* 20 to 1.74633 5 (B*) to A* 20 to 1.74633 5 (B*) to A* 4* 20 to 1.74633 5 (A*) °C Cc c c c C 257. 5 c Cc c c c C 257. 5 c Cryos. A* consts. B* c vap. °K Cryos. B* c vap. °K Cryos. A* consts. Consts. Cons	Dielectric								121.	2	
A 31 to			222		t _e						
A* 31 to 1.39108 5 AHf			,			0, 256	-	M Spec.		1	
B		+		_				Ultra V.			
Viscosity centistokes			108		ΔFf	1					
tk to tx **C Acetone Carbon tet. Benzene Ether n-Heptane Ether n-Heptane Ethanol Water A'* 20 to B'* 31 °C 1463.5 1.74633 5 B'* 31 °C 1463.5 BV to AV °C to (BV) to (AV) °C Water in Acetone Carbon tet. Benzene Ethanol Water B* 31 °C 1463.5 5 CC - C - 257. (BV) °C to (AV) °C Bc t _c °C 1674. 5 C ₂ C ₂ - C 257. C ₂ liq. °K Water in Cryos. A° consts. B° 128.99 5 C _V vap. T _R = 0.75 T _C **grams/100 grams solver REFERENCES: 1-Dow SOURCE: API PURIFICATION: API	K '	-						<u> </u>		+	
Carbon tet. Benzene		-1				:		Acetone			
A' 20 to 7.34929 5	tx c				1					l	
C'			929					Ether			
A** 20 to 1.74633 5 A* C C Water water in Ac 147 to 7.3625 5 C EV to Cc 257. 5 C p vap. o'K Cryos. A° consts. B° c 128.99 5 C vap. T _R = 0.75 T _C references: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API Archive API Archive API Water water in Water water in Water water in Water water in For ams/100 grams solver **grams/100 grams solver**			, '		B ^V to		1				
B * 31 °C 1463.5 5 (B V) to (A V) °C Ac 147 to 7.3625 5 (A V) °C Bc t_c °C 1674. 5 5 Cryos. A° consts. B° c _p vap. °K t_e °C 128.99 5 c _v vap. T _R = 0.75 T _c grams/100 grams solver REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API					AV C	:	1			İ	
Bc t_c °C 1674. 5 c_p liq. °K					(B ^V) to	7		Water in	<u> </u>	—	
Cryos. A° cp vap. °K consts. B°	Ac 147 to	7.36	25		(A ^V) •C	: [1	
Cryos. A° consts. B° cp vap. °K te °C 128.99 5 cv vap. TR = 0.75 Tc * grams/100 grams solver REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API	Bc Ltc_	7 1674. - 257			cp liq. °K						
consts. B° p p T		1-511			l					1	
T _R = 0.75 T _C + grams/100 grams solver REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API	consts. B°				<u>-</u>						
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API	t _e °C			5	c _v vap.		L			L	
SOURCE: API PURIFICATION: API	$T_{R} = 0.7$	5 T _C						grams/100 grai	ns solven	t	
PURIFICATION: API		CES: 1-D	ow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for	mula		
				AP	I						
LITERATURE REFERENCES:											
	LITERATU	RE REF	EREN	CES):						

No. 87 2-Methyl-cis-3-heptene STRUCTURAL FORMULA NAME CH3(CH2)2CH=CHCH CH3 ĊHą Mole Ref. Molecular Molecular C8H16 % Pur. Formula Weight 112, 208 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP to °C/mm ۰ĸ g 25°C 0.7214 B. P. ℃ h BP 0.0463 4 760 mm 112. 53.31 2 t_e 0.0363 5 ſ١ to 100 4 g' ۰ĸ 27.41 4 30 30 mm 0.6486 5 10 7.64 5 h' ∆Hm cal/g 5 1 -25.47 to m AHv cal/g Pressure n °K 25°C 82.42 mm 25°C 26.48 o 30 mm 82.25 5 1038. 5 te BP 70.49 5 m to Density 5 68.97 te (d, e) n' °K g/ml 20°C 5 0.706 2 68.91 0' $\mathbf{d_{4}^{t}}$ 25 0.702 2 AHv/Te 19.52 5 30 0.698 4 Surface tension 0.722 -0.0₃8 27 to 86.05 5 a 44 dynes/cm, 20°C 19.77 5 å <u> 123</u> 0.1390 b 30 5 18.88 5 to 84.19 15 18.01 40 5 Ref. Index e¹ 5 27 0.0708 1.407 ^{n}D 20°C 2 Parachor [P] 5 d_c g/ml 0.239 25 1.405 2 20°C vc ml/g 4.190 5 30 1.402 4 30 t_c 278. 5 40 "C" 0.7679 4 5 P_c mm 18716. Sugd. 335.2 5 MR (Obs.) 2 39.1 PV/RT Exp. L.1.%/wt. 38.677 MR (Calc.) 25°C 1.0000 5 (nD-d/2)1.054 2 30 mm 1.0000 2 Dispersion 119. Dielectric ВP 0.9500 5 Flash Point C t_e 5 A 27 to 0.9405 6.94826 5 Fire Point 5 0.256 B 140 °C 1341.1 M. Spec. С 217.72 5 AHc kcal/m Ultra V. ΔHf A* 27 to 1.39246 5 X-Ray Dif. ΔFf B* 130 °C 1259.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. $\mathbf{t}_{\underline{\mathbf{x}}}$ °C Benzene A' 15 to 7.35001 Ether В' 27 °C 1545.3 n-Heptane B_v | C' 235,72 to Ethanol °C Water 15 to 1.74894 A'* 5 Water in B'* 27 °C (BV) 1445.8 5 to Ac | 140 to 7.3585 5 (AV) °C Bc tc °C 1651. cp liq. °К Cc 5 257 Cryos. Aº c_p vap. °K consts. B° c vap. te °C 5 123.34 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

						1	No. 88
NAME	2-Methyl	-tran	s-3-heptene			STRUCTURAL FO	RMULA
						CH3(CH2)2CH=CH	сн сн.
Mole	Ref. Mo	11		Molecular			CH ₂
% Pur.		lecul rmul		Weight 112.2	08_		.
		Ref.			Ref		Ref
F, P, *C			dt/dP			f to	
F.P. 1007 B.P. *C	'	1	*C/mm 25*C	0.7214	5	g <u>°K</u>	
76 0 mm	112.	2	BP	0.0463	4	h f' to	
100 30	53.31 27.41	4	t _e 30 mm	0.0363	5	f' to to	
10	7.64	5	ΔHm cal/g	0.0480	٦	h'	
1	-25.47	5	ΔHv cal/g			m to	
Pressure mm 25°C	26.48	5	25°C	82.42	5	n •K	j
t _e	1038.	5	30 mm BP	82.25 70.49	5		
Density g/ml 20°C	0.706	2	to (d.e)	68.97 68.91	5	m' to	
at 25	0.702	2	te (d, e) AHv/Te	19.52	5	o'	
4 30	0.698	4	d 27 to	86, 05	5	Surface tension	
Ъ	0.722 -0.0 ₃ 8	4	_e _ _123_°C	0.1390	5		9.77 5 8.88 5
Ref. Index	:		d' 15 to e' 27 °C		5 5	40 1	8.01 5
ⁿ D 20°C	1.407	2 2	d _c g/ml	0.239	5	Parachor [P] 20°C	
30	1.402	4	tc *C	4.190 278.	5	30	
"C"	0.7679	4	P _c mm	18716.	5	40 Sugd. 33	5, 2 5
MR (Obs.) MR (Calc.		2	PV/RT		\vdash	Exp. L.1.%/wt.	
(nD-d/2)	1.054	2	25°C 30 mm	1.0000	5	u. Dispersion 12	1. 2
Dielectric	<u> </u>		BP	0.9500	5	Flash Point °C	
A 27 to B 140 °C	6.94826 2 1341.1	5	t _e t _c	0.9405 0.256	5	Fire Point	
c	217.72	5	∆Hc kcal/m	<u> </u>		M Spec. Ultra V.	
A* 27 to		5	ΔHf ΔFf			X-Ray Dif.	
B* 130 °C	2 1259.5	5	Viscosity		-	Infrared Solubility in +	
c t,t			centistokes 7 °C			Solubility in +	İ
t _k to		1	7 .c			Carbon tet. Bensene	
A' 15 to		5				Ether	
B' \(\sum_{27} \cdot \)	235.72	5	B ^V to	 		n-Heptane Ethanol	
A'* 15 to		5	A ^V •C	.]		Water	
B'* 27 °C		5	(B ^V) to	1		Water in	
Ac 140 to	7.3585 C 1651.	5	(A ^V) °C	ļ	<u> </u>		
Cc —	257.	5	c _p liq. •K				1
Cryos. As consts. B			c _p vap. °K				
t _e °C	123.34	5	c _w vap.				
$T_{R} = 0.7$						+ grams/100 grams	solvent
	CES: 1-Dow			Calc. from det	t. da	ta 5-Calc. by formu	la
SOURCE:	TION	AI					
PURIFICA	RE REFEREI	AI					
DIM I U	NEFERE		'•				

							N o. 89	
NAME _	3-Methyl-	cis-	3-heptene			STRUCTURAL	FORMULA	
						сн ₃ (сн ₂) ₂ сн=	C CH_CH_	
Mole % Pur.		lecul mul		Molecular Weight 112.2	08	3,52,2	ch ₃	
		Ref.			Ref.		Re	eſ.
F. P. °C			dt/dP			f to		
F.P. 100%	 	_	*C/mm 25*C	1.0203	5	g° <u>K</u>		
B.P. °C 760 mm	121.	2	BP	0.0472	4	h		_
100 30	61.12 34.68	4	t _e 30 mm	0.0363	5	f' to		
10	14.50	5	ΔHm cal/g	0.0022	+ ~	h'		
1	-19.31	5	ΔHv cal/g		1	m to		_
Pressure mm 25°C	18.09	5	25°C	85.30	5	n <u>*K</u>		
t _e	1062.7	5	30 mm BP	84.51 72.37	5			_
Density g/ml 20°C	0.728	2	t _e	70.69	5	m' to		
at 25	0.724	- 2	e (4, 6)	70,62	5	o'		
	0.720	4	ΔHv/T _e	19.50 89.38	5	Surface tension		_
a b	0.744 -0.0 ₃ 8	4	133 °C	0.1405	5	dynes/cm. 20°C		5 5
Ref. Index	1		d' 25 to	87.35 0.0819	5	40		5
ⁿ D 20°C	1.418	2 2	d g/ml	0.246	5	Parachor [P]		
30	1.416	4	v _c ml/g t _c °C	4.071	5	20°C 30		
"C"	0.7637	4	P _c mm	294. 19746.	5	40 Sund	225 2	_
MR (Obs.)	38.9	2	PV/RT	19740.	-	Exp. L.1.%/wt.	335.2	5_
MR (Calc.) (nD-d/2)	38.677 1.054	5 2	25°C	1.0000	5	u.		
Dielectric			30 mm BP	1.000 0 0.9495	5	Dispersion	124.	2
A 35 to	6.95770	5	t _e	0.9392	5	Flash Point C Fire Point		
B (153 °C	1374.0 216.	5	tc ΔHc kcal/m	0.255	,	M. Spec.		_
A* 35 to	1,39323	5	ΔHf			Ultra V. X-Ray Dif.		
B*[143 °C	1290.9	5	ΔFf			Infrared		
С			Viscosity centistokes		1	Solubility in +		
t _k to			η °C			Acetone Carbon tet.		
tx °C	7, 35121	5				Benzene Ether		
B' 35 °C	1578. 3	5	B _v to		ļ	n-Heptane		
C' A'* 25 to	234.	5.	B to C			Ethanol Water		
B' ≠ 35 °C	1.74671 1478.6	5	(B ^V) to	1		Water in		
Ac 153 to	7.3680	5	(A ^V) °C					
Bc t _c °C	1693. 257.	5	c _p liq. °K					
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	133.48	5	c _v vap.					
$T_{R} = 0.75$		L	u	1	ئــــــــــــــــــــــــــــــــــــــ	grams/100 gra	ms solvent	_
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		_
SOURCE:		A	PI.					_
PURIFICAT			PI					
LITERATUI	RE REFERE	NCES	S:				-	

- 							No. 90
NAME	3-Methyl	-tran	s-3-heptene			STRUCTURAL FO	RMULA
						CH3(CH2)2CH=C	сн,сн,
Mole	D-6 34-	11		Molecular		, г. с. с. с. с. с. с. с. с. с. с. с. с. с.	
% Pur.		lecul rmul		Weight 112.2	.08	3)
		Ref.			Ref		Ref
F.P. °C			dt/dP	1		f to	
F.P. 100%			°C/mm		_	g LK	ŀ
B. P. *C			25°C BP	1.0203 0.0472	5 4	h ;	
760 mm 100	121.	2 4	te	0.0363	5	f' to	
30	34.68	4	30 mm	0.6622	5	g' 'K_	
10 1	14.50 -19.31	5	ΔHm cal/g			h'	
Pressure	17.31	 	ΔHv cal/g			m to oK	
mm 25°C	18.09	5	25°C 30 mm	85.30 84.51	5 5	n ' *K	
t _e	1062.7	5	BP	72.37	5		
Density g/ml 20°C	0.720	١,	t _{e (3.3}	70.69	5 5	m' to	
at 25	0.728 0.724	2 2	t _e (d, e)	70.62	5	0'	
⁴ 4 30	0.720	4	ΔHv/T _e	19.50	5	Surface tension	
a b	0.744	4	d 35 to	89.38 0.1405	5	dynes/cm. 20°C	22.36 5
	-0.0 ₃ 8	+ +	d' 25 to	87.35	5		21.38 5
Ref. Index n _D 20°C	1.418	2	e' 35 °C		5	Parachor [P]	
45	1.416	2	a g/mi	0.246 4.071	5 5	20°C	
30 "C"	1.413	4	d g/ml vc ml/g tc °C	294.	5	30 40	
	0.7637	4	P _c mm	19746.	5	Sugd. 3	35.2 5
MR (Obs.) MR (Calc.)	38.9 38.677	2	PV/RT			Exp. L.1.%/wt.	
(nD-d/2)	1.054	2	25°C 30 mm	1.0000 1.0000	5 5	u. Dispersion	24. 2
Dielectric	<u> </u>		BP	0.9495	5	Flash Point °C	
A 35 to	6.95770	5	t _e t _c	0.9392 0.255	5	Fire Point	
B 1723.€	1374.0	5	ΔHc kcal/m		۲Ť	M Spec.	
A* 35 to	1.39323	5	ΔHf			Ultra V. X-Ray Dif.	
B* 143 °C	1290.9	5	ΔFf	ļ		Infrared	
K	1		Viscosity centistokes	1		Solubility in +	
t _e to			7 °c			Acetone Carbon tet.	
'x		<u> </u>				Benzene	
A' 25 to B' 35 °C	7.35121 1578.3	5				Ether n-Heptane	
ر. د عام	234.	5	B ^v to			Ethanol	
A'* 25 to	1.74671	5	AV C	.		Water Water in	
B'* 35 °C	 	5	(B ^V) to	ł		Marci III	
Ac 153 to	7.3680 1693.	5	(A ^V) °C		\sqcup		
Cc	257.	5	c _p liq. ∘K				
Cryos, A° consts, B°			c _p vap. °K				
t _e °C	133.48	5	c _w vap.				
$T_{R} = 0.75$						+ grams/100 grams	solvent
REFERENC	ES: 1-Dow	2-AI		alc. from de	t. dat	ta 5-Calc. by formu	ıla
SOURCE:			PI				
PURIFICAT			PI				
LITERATUR	E REFERE	NCES	:				

No. 91 4-Methyl-cis-3-heptene NAME STRUCTURAL FORMULA $CH_3(CH_2)_2C = CHCH_2CH_3$ ċн₃ Mole Molecular Molecular C8H16 Weight 112, 208 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g _°K_ 1.0610 5 B. P. °C h BP 0.0473 122. 62.00 760 mm 2 0.0363 5 ſ١ to 100 4 g' <u>°К</u> 30 35.50 4 30 mm 0.6637 5 10 15.27 5 h' ∆Hm cal/g 5 -18.61 to m AHv cal/g Pressure ۰ĸ n 25°C 85.63 5 mm 25°C 17.33 5 o 84.76 30 mm 1066. 5 te ВP 72.59 5 m' | to Density 5 5 70.88 n' ۰ĸ g/ml 20°C te (d, e) 0.725 2 70.82 o' $\mathbf{d_4^t}$ 25 0.721 2 5 AHV/Te 19.50 30 0.717 4 Surface tension d T 36 89.76 5 to 0.741 a 4 dynes/cm. 20°C 21.99 5 5 135 25 °C 0.1407 ь -0.038 4 30 21.03 5 ăΠ to 87.69 5 40 20.09 5 e' Ref. Index 36 0.0826 5 20°C [P] 1.417 ^{n}D Parachor 5 d_c g/ml 0.244 25 1.415 2 20°C vc ml/g t_ °C 4.096 5 30 1.412 4 30 ^tc 295. 5 40 "C" 0.7651 4 P_c mm 19660. 5 Sugd. 5 335.2 MR (Obs.) 38.9 2 PV/RT Exp. L. l. %/wt. 38.677 MR (Calc.) 5 25°C 1,0000 (nD-d/2) u. 1.055 2 30 mm 1,0000 5 Dispersion 124. 2 Dielectric BP 0.9495 0.9391 5 Flash Point C 5 6.95905 A 36 to Fire Point 0.255 5 B (153 ℃ 1377.7 5 M. Spec. С 5 AHc kcal/m 215.82 Ultra V. ΔHf A* 36 to B* 145 °C 1.39354 5 X-Ray Dif. ΔFf 1294.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. °C $^{t}_{\underline{\mathbf{x}}}$ Benzene 25 to 7.35168 Ether B١ 36 **°**C 1582.1 n-Heptane B_v 5 C' 233.82 to Ethanol °C A'* 25 to B'* 36 °C Water 1.74680 5 Water in 1482.4 (B^V) to Ac | 153 to 7,3692 5 (A^V)| °C Bc tc 1697. °C c_p liq. ۰ĸ Cc 256. 5 Cryos. Aº c vap. °K consts. B° c_v vap. te °C 134.61 5 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

						No. 92	
NAME _	4-Meth	yl-tran	s-3-heptene			STRUCTURAL FORMUL	A
						CH3(CH2)2C = CHCH2CH	H ₃
Mole	Ref. M	folecul	ar cu	Molecular	ł	Ċн ₃	
% Pur.		ormul	C ₈ H ₁₆	Weight 112.2	208		
		Ref.			Ref		Ref.
F.P. /C F.P. 100%		+	dt/dP	}		f to	1
B. P. *C	 	+-1	*C/mm 25*C	1.0610	5	g <u>*K</u>	1
760 mm	122.	2	BP	0.0473 0.0363	4 5	h to	┿
100 30	62.00 35.50	4 4	t _e 30 mm	l	5	f' to to	1
10	15.27	5		0.6637	-	h'	
1	-18, 61	5	ΔHm cal/g ΔHv cal/g	 	\vdash	m (to	\top
Pressure mm 25°C	17.33	5	25°C	85.63	5	n •K	1
t _e	1066.	5	30 mm BP	84.76	5	<u>°</u>	
Density	<u> </u>	+	t _e ,	72.59 70.88	5	m' to	
g/ml 20°C	0.725	2	t (d, e)	70.82	5	n' <u>*</u> K	
dt 25 4 30	0.721 0.717	2 4	AHv/T _e	19.50	5		₩.
	0.741	4	d 36 to		5	Surface tension dynes/cm. 20°C 21.99	5
ь	-0.038	4	a 135 to		5	30 21.03	5
Ref. Index			e' 36 °C		5	40 20.09	5
ⁿ D 20°C	1.417	2 2	d g/ml vc ml/g	0.244	5	Parachor [P]	
30	1.412	4	vc ml/g tc °C	4.096 295.	5	30	
"C"	0.7651	4	P _c mm	19660.	5	40 Sugd. 335, 2	5
MR (Obs.) MR (Calc.)	38.9	2	PV/RT	 	+	Exp. L. l. %/wt.	+-
(nD-d/2)	38.677 1.055	5 2	25°C	1.0000	5	u.	_
Dielectric			30 mm BP	1.0000 0.9495	5	Dispersion 124.	2
A 36 to	6.9590	5 5	t _e	0.9391	5	Flash Point °C Fire Point	
B 1123 C	1377.7	5	t _c	0, 255	5	M Spec.	
C 45 2/ 45	215.82	5	ΔHc kcal/m	1		Ultra V.	
A* 36 to B* 145 °C	1.39354 1294.5	4 5 5	ΔFf			X-Ray Dif. Infrared	l
K '			Viscosity			Solubility in +	+-
t _k	1	1 1	centistokes 7°C	1	}	Acetone	1
t c			'	1		Carbon tet. Benzene	1
A' 25 to	7.35168			1		Ether	
B' ∟36 °C	1582.1 233.82	5	B ^V to	<u> </u>	\vdash	n-Heptane Ethanol	1
A** 25 to	1,74680		AV °C	[Water	1
B'* 36 °C	1482.4	5	(BV) to	7		Water in	1
Ac 153 to	7.3692	5	(A ^V) °C				
Bc tc_C	1697. 256.	5 5	c _p liq. •K				1
Cryos, A*			c _p vap. *K				
consts. B°	<u> </u>	4-4	c, vap.				
t _e °C	134.61	5	~ ·=p.		لــــــــــــــــــــــــــــــــــــــ	l	
T _R = 0.75						grams/100 grams solven	1t
SOURCE:	20. 1-DOW	Z-AF		Caic. Irom de	t. dai	ta 5-Calc. by formula	
PURIFICAT	ION:	AF					
LITERATUE							

							No. 93
NAME _	5-Methyl-	cis-	3-heptene			STRUCTURAL	FORMULA
			-			сн ₃ сн ₂ сн сн=	сн сн ₂ сн ₃
Mole % Pur.		lecul mula	ar C ₈ H ₁₆	Molecular Veight 112,20	08	сн _{3.}	
		Ref.			Ref.		Res
F. P. °C			dt/dP			f to	
F.P. 100%			°C/mm		ا ۔ ا	g° <u>K</u>]
B. P. °C	١.,,		25°C BP	0.7214 0.0463	5 4	h	L
760 mm 100	112. 53.31	2 4	te	0.0363	5	f' to	
30	27.41	4	30 mm	0.6486	5	g' <u>*</u>	
10 1	7.64 -25.47	5	ΔHm cal/g			h'	
Pressure			ΔHv cal/g		_	m to	
mm 25°C	26.48	5	25°C 30 mm	82.42 82.25	5	0	
t _e	1038.	5	BP	70.49	5	m' to	
Density g/ml 20°C	0.713	2	t _e (d, e)	68.97 68.92	5	n'ı ek	
at 25	0.709	2	ΔHv/T _e	19.52	5	o'	1
4 3 0	0. 70 5	4			5	Surface tension	
a b	0.729	4	d 27 to	86.05 0.1389	5	dynes/cm. 20°C	20.57 5
Ref. Index	-0.038	4	d' 15 to	84.19	5	8 30 40	19.65 5
n _D 20°C	1.410	2	e! 27 °C	0.0709	5	Parachor [P]	
20	1.408	2	d g/ml vc ml/g	0.242 4.140	5	20°C	
30 "C"	1.405	4	tc °C	279.	5	30 40	
	0.7657	4	P _c mm	18976.	5	11	335.2 5
MR (Obs.) MR (Calc.)	39.0 38.677	5	PV/RT			Exp. L.1.%/wt.	
(nD-d/2)	1.054	2	25°C 30 mm	1.0000	5	u. Dispersion	119. 2
Dielectric			BP	0.9500	5	Flash Point °C	-
A 27 to	6.94826	5	t _e	0.9405 0.256	5	Fire Point	
B (141 °C	1341.1 217.72	5	tc ΔHc kcal/m	0.230	-	M. Spec.	
A* 27 to	1.39246	5	ΔHf	i		Ultra V. X-Ray Dif.	
B* 133 °C	1259.5	5	ΔFf			Infrared	
K ———			Viscosity centistokes			Solubility in +	
t _k Tto	İ		η °C			Acetone Carbon tet.	
* L			•			Benzene	
A' 15 to B' 27 °C	7.35001 1545.3	5				Ether n-Heptane	
č, '	235.72	5	B _v to			Ethanol	
A'* 15 to	1.74894	5	ĂV I C	}		Water	
B'* 27 °C	1445.8	5	(B ^V) to		1	Water in	+
Ac 141 to Bc t _c °C	7.3588 1652.	5	(A ^V) °C		<u> </u>		
Cere-	257.	5	c _p liq. °K				
Cryos. A° consts. B°			c _p vap. °K				
t _e °C	123,34	5	c vap.				
$T_{\mathbf{p}} = 0.75$			<u> </u>	L		grams/100 gra	ms solvent
		2-A	PI 3-Lit. 4-	Calc. from de	t. da	ita 5-Calc. by for	
SOURCE:			PI				
PURIFICAT	ION:		PI				
	RE REFERE	NCES	5:				
1							

							No. 94	
NAME	5-Methyl	-tran	ıs-3-heptene			STRUCTURAL F	ORMULA	4
					\neg	CII CII CII CII-C		
						CH ₃ CH ₂ CH CH=C	H CH ₂ CH	3
Mole		lecul		Molecular	۱ ۵۰	ĊH ₃		
% Pur.	Fo	rmul		Weight 112.2	-			Ref.
		Ref.		r	Ref	 		Kei.
F.P. °C F.P. 100%		├	dt/dP *C/mm			f to g °K		
B. P. °C		-	25°C	0.7214	5	g <u>*K</u>		
760 mm	112.	2	BP	0.0463	4 5	l		
100 30	53.31 27.41	4	t _e	0.0363	1 1	g' to		1
10	7.64	5	30 mm	0.6486	5	h' i		
1	-25.47	5	ΔHm cal/g		-	m to		
Pressure		_	ΔHv cal/g 25°C	82.42	5	n '°K		
mm 25°C	26.48 1038.	5	30 mm	82.25	5	•		ĺ
Density	1-050.	-	BP	70.49	5	m' to		
g/ml 20°C	0.713	2	te te (d, e)	68.97 68.92	5 5	n'		
dt 25	0.709	2	ΔHv/T	19.52	5	0'		L_
<u> </u>	0.705	4	d 27 to	86,05	5	Surface tension		
a b	0.729 -0.0 ₃ 8	4		0.1389	5	dynes/cm. 20°C	20.57	5
Ref. Index		Ť	d' 15 to		5	40	19.65 18.76	5
n _D 20°C		2			5	Parachor [P]		
25 30	1.408	2	d g/ml v ml/g	0.242 4.140	5	20°C		Į .
"C"	1.405	4	tc •C	279.	5	30 40		
	0.7657	4	P _c mm	18976.	5	Sugd.	335.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.054	2	25°C	1.0000	5	u.		
Dielectric			30 mm BP	1.0000	5	Dispersion	121.	2
A 27 to		5	t _e	0.9405	5	Flash Point °C Fire Point		ł
B [_141_*C	1341.1	5	tc	0,256	5	M Spec.		
C 27.	217.72	5	ΔHc kcal/m ΔHf	İ		Ultra V.		ł
A* 27 to B* 133 °C		5	ΔFf			X-Ray Dif. Infrared		ĺ
к — — -	-	1	Viscosity			 		├
tt	-		centistokes 7°C	ļ		Solubility in + Acetone		ļ
t _k t _x	1		7 ℃			Carbon tet.		Ì
A' 15 to	7. 35001	5				Benzene Ether		Ì
B' _ 27 °C	1545.3	5	B ^V to	<u> </u>	-	n-Heptane		ł
C'	235, 72	5	B' to	1		Ethanol Water		İ
A'* 15 to B'* 27 °C	1.74894	5				Water in		
Ac 141 to	+	5		1				
Bc tc °C	1652.	5			\vdash			
Cc -	257.	5	c _p liq. °K					
Cryos. A° consts. B°			c _p vap. °K					
te°C	123. 34	5	c _v vap.					
$T_R = 0.7$						f grams/100 gram		t
	CES: 1-Dow			Calc. from de	t. dat	ta 5-Calc. by forn	nula	
SOURCE:		Al						
PURIFICA'		AI						
LITERATU	RE REFERE	NCES	S:					
1								

No. 95 6-Methyl-cis-3-heptene NAME STRUCTURAL FORMULA CH3CH CH2CH=CHCH2CH3 ĊН3 Mole Ref. Molecular Molecular C8H16 % Pur Formula Weight 112.208 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 0.8098 5 B. P. ℃ h ВP 4 0.04662 760 mm 115. 55.93 2 t_e 0.03630 5 f١ to 100 4 g' <u>°К</u> 30 29.85 4 30 mm 0.6530 5 10 9.94 5 h' ∆Hm cal/g 5 -23.40 1 m to AHv cal/g Pressure n °K 25°C 83.39 5 5 mm 25°C 23.32 5 5 0 30 mm 83.02 te 1046. BP 71.15 5 m' Density to 69.57 5 n' °K g/ml 20°C 0.713 2 te (d, e) 5 69.51 o' $\mathbf{d_{4}^{t}}$ 25 0.709 2 ΔHv/T_e 5 19.52 30 0.705 4 Surface tension 30 d 87.18 5 a 0.729 4 dynes/cm. 20°C 20.57 127 °C 0.1394 5 ь 4 -0.038 30 19.65 5 ăΠ 20 to °C 85.31 5 40 18.76 5 e' | Ref. Index 30 0.0768 5 20°C 1.410 2 n_{D} Parachor [P] d_c g/ml 0.241 5 25 1.408 2o°C vc m1/g t °C 4.152 30 1.403 4 30 t_c 283. 5 "C" 40 0.7657 4 P_c mm 5 19060. Sugd. 335.2 5 MR (Obs.) 39.0 2 PV/RT Exp. L. l. %/wt. MR (Calc.) (nD-d/2) 38.677 5 25°C 1.0000 5 u. 1.054 2 30 mm 1.0000 5 Dispersion 2 119. Dielectric BP 0.9500 Flash Point C te tc 0.9402 A 30 to 6.95245 5 Fire Point 0.256 B 1144 °C 5 1352.4 M. Spec. С 217.15 5 AHc kcal/m Ultra V. ΔHf A* 30 to B* 134 °C 1.39344 5 X-Ray Dif. ΔFf 1270,2 5 Infra red ĸ Viscosity Solubility in c centistokes Acetone to tk tx ٠c Carbon tet. Benzene 20 to 7.35146 Ether B١ 3<u>0 °C</u> 5 1556.7 n-Heptane B^V | C 235.15 5 to Ethanol °C 1.74923 Water A1* 20 to 30 °C 5 Water in B'* (BV) 1457.1 5 to Ac | 144 to 7.3627 5 (A^V)| °C 1665. Bc tc °C 5 °C ۰ĸ c_p liq. 257. 5 Cryos. A° c_p vap. °K consts. B° c_v vap. te °C 126,72 5 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

]	No. 96
NAME	6-Methyl	-trar	s-3-heptene			STRUCTURAL FO	RMULA
						сн ₃ сн сн ₂ сн=сн	сн,сн,
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 112.2	.08	CH ₃	2 3
		Ref.			Ref.		Ref
F.P. °C F.P. 100%			dt/dP *C/mm			f to	
B. P. °C 760 mm	115.	2	25°C BP	0.8098 0.04662	5 4	h	
100 30	55.93	4	t _e 30 mm	0.0363 0.6530	5	f' to to	İ
10	29.85 9.94	5	ΔHm cal/g	0.0330	-	h'	
1	-23.40	5	ΔHv cal/g	 		m to	
Pressure mm 25°C	23.32 1046.	5 5	25°C 30 mm	83.39 83.02	5	n	
Density	2.50	_	te (d.e)	71.15 69.57	5 5	m' to	
g/ml 20°C dt 25 4 30	0.709	2 2 4	t _e (d, e) ΔHv/T _e	69.51	5	o' ' = -	
a 30	0,705	4	d 30 to	87.18	5	Surface tension dynes/cm. 20°C 2	20.57 5
Ъ	-0.038	4	d 1 27 to	85.31	5	ا 30 1	9.65 5 18.76 5
Ref. Index		2	e' 30 °C	 	5	Parachor [P]	3.10
D 25 30	1.408 1.403	2 4	d g/ml vc ml/g	0.241 4.152	5	20°C	İ
"C"	0.7657	4	1c C	283. 19 0 60.	5	40 Sugd. 33	35.2 5
MR (Obs.)	. 1 - /	2	P _c mm	19060.	-	Exp. L.1.%/wt.	15.2 5
MR (Calc. (nD-d/2)	38.677 1.054	5 2	25°C 30 mm	1.0000	5 5	u.	21. 2
Dielectric			BP	0.950 0	5	Flash Point °C	
A 30 to		5 5	t _e t _c	0.9402 0.256	5	Fire Point	
c	217.15	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.	
A* 30 to B* 134 °C		5	ΔFf			X-Ray Dif. Infrared	
K ———			Viscosity centistokes			Solubility in +	
t _k to			η •c			Acetone Carbon tet.	
A' 20 to		5				Benzene Ether	1
B' _ 30 °C	235.15	5	B ^V to			n-Heptane Ethanol	
A'* 20 to B'* 30 °C		5 5	AV C			Water Water in	
Ac 144 to	<u> </u>	5	(B ^V) to (A ^V) °C				
Bc tc_°C		5	c _p liq. °K				
Cryos. A° consts. B°			c _p vap. °K				
t _e °C	126.72	5	c _w vap.				
$T_R = 0.7$						grams/100 grams	solvent
	CES: 1-Dow			Calc, from de	t. da	ta 5-Calc, by formu	la
SOURCE:			PI				
PURIFICAT			PI 				
LIIERAIU	RE REFERE	NC ES);				

							No. 9	<u>'</u>
NAME _	2-Ethyl-1	-hex	ene			STRUCTURAL	FORMUI	LA
						CH (CH) C	- CH	·
•					\neg	CH ₃ (CH ₂) ₃ C		
Mole % Pur.	Ref. Mo	ecul		Molecular Veight 112.2	ne	C,	2 ^H 5	
W Fui.	1 1 1 1 1 1 1	Ref.	1 0 10 1	Weight 112.2	Ref.			Ref.
E D 86	1	Ver.	1./12	 	Kei.		T	Ter.
F.P. °C F.P. 100%		\vdash	dt/d₽ °C 1		1	f to		
B. P. °C		М	25℃	0.9813	5	g '° <u>K</u> h	1	
760 mm	120.	2	BP	0.0471 0.0363	5			+-
1 00 30	60.25	4 4	t _e		1	f' to		
10	13.73	5	30 mm	0.6607	_5_	h'		
1	-20.00	5	ΔHm cal/g	<u> </u>		m to		+
Pressure			ΔHv cal/g 25°C	84.97	5	n		
mm 25°C	18.88 1060.	5	30 mm	84.25	5	0		
t _e Density	1000.	۲	BP	72.15	. 5	m¹ to		
g/ml 20°C	0.7270	2	te te (d,e)	70.48 70.42	5	n' <u>°K</u>	1	
at 25	0.7228	2	AHv/Te	19.50	5	°'		
	0,7186	4	d 34 to	89.00	5	Surface tension		
a b	0.7438 -0.0 ₃ 83	4 4	_e 132 °C	0.1404	5	dynes/cm. 20°C	22.24	5
Ref. Index	-0.0303	1	d' 25 to	87.00	5	8 30 40	21.22	5
n _D 20°C	1.4157	2		0.0812	5	Parachor [P]	1	+
45.	1.4132	2	d g/ml vc ml/g	0.244 4.101	5	20°C		
30	1.4106	4	tc °C	292.	5	30 40	ĺ	
"C"	0.7608	4	P _c mm	19527.	5		335.2	5
MR (Obs.) MR (Calc.)	38.71 38.677	2 5	PV/RT	İ	1	Exp. L.1.%/wt.		
(nD'd/2)	1.0522	2	25°C	1.0000	5	u.		
Dielectric		\vdash	30 mm BP	1.0000 0.9495	5	Dispersion	121.	2
A 34 to	6,95634	5	t _e	0.9393	5	Flash Point C Fire Point		
B (_151 °C	1370.2	5	L t _c	0.255	5	M. Spec.		+
c	216. 2	5	ΔHc kcal/m ΔHf	ļ	1	Ultra V.		
A* 34 to B* 142 °C	1.39292 1287.3	5	ΔFf			X-Ray Dif. Infrared		
к С = =	120		Viscosity			l	-	+-
t, to			centistokes			Solubility in Acetone		-
t _k			7 °C			Carbon tet.		
A' 25 to	7, 35072	5				Benzene Ether		
B'34 °C	1575.	5	B _w to	 	-	n-Heptane		
C'	234, 2	5	B to C		ł	Ethanol Water		
A'* 25 to B'* 34 °C	1.74662 1474.8	5	$\frac{\mathbf{A}}{ \mathbf{B}^{\mathbf{V}} } - \frac{\mathbf{C}}{to}$	1		Water in		
Ac 151 to	7, 3666	5	(A ^V) °C					1
Bc t _c °C	1688.	5			+-	1		
Cc	257.	5	P					
Cryos, A°			c _p vap. °K	1				
consts, B°	 	\vdash	c vap.	1				
t _e °C	132.35	5	A -1.	L	<u> </u>	I	L	
$T_R = 0.79$			DT 2 T1:			grams/100 gra		nt
	ES: 1-Dow			Calc. from de	t. da	ata 5-Calc. by for	mula	
SOURCE:		AF		-				
PURIFICAT		AF						
LITERATU	RE REFERE	NCES	5:					

						1	No. 98
NAME	3-Ethyl-l	-hex	ene			STRUCTURAL FO	RMULA
						כם (כם) כם כו	u-Cn
Mala	7-6 14-	lecul		Molecular		CH ₃ (CH ₂) ₂ CH Cl	11-0112
Mole % Pur.		rmul		Molecular Weight 112.2	08	25	
		Ref.	E		Ref		Ref
F.P. *C			dt/dP			f to	
F.P. 100%			*C/mm 25*C	0.6758	5	g <u>*K</u> _	
B. P. °C 760 mm	110.3	2	BP	0.0462	4	h	
100	51.82	4	t _e	0.0363	5	f' to g' 'K	
30 10	26.02 6.32	4 5	30 mm	0.6461	5	g' 'K_	
1	-26.66	5	ΔHm cal/g			 	
Pressure			ΔHv cal/g 25°C	81.86	5	m to	
mm 25°C	28.46 1033.	5	30 mm	81.80	5	0	
t _e Density	1033.	-	BP	70.12 68.63	5 5	m' to	
g/ml 20°C	0.715	2	te te (d, e)	68.58	5	n' <u>*K</u>	
dt 25 4 30	0.711 0.707	2 4	AHv/Te	19.51	5		
4 JU	0.731	4	d 26 to	85.41	5	Surface tension dynes/cm, 20°C 2	0.80 5
Ъ	-0.038	4	e 121 °C d' 5 to		5 5	30 1	9.87 5
Ref. Index			e' 26 °C		5		8.97 5
n _D 20°C	1.407 1.405	2 2	d _c g/ml	0.243	5	Parachor [P]	
30	1.402	4	d g/ml vc ml/g tc °C	4.121 277.	5 5	30	
"C"	0.7582	4	P _c mm	18997.	5	40 Sugd. 33	5.2 5
MR (Obs.)	38.7	2	PV/RT	10771.	 	Exp. L.1.%/wt.	3.2
MR (Calc.) (nD-d/2)	38.677 1.050	5 2	25°C	1.0000	5	u.	İ
Dielectric		╅	30 mm BP	1.0000	5 5		7. 2
A 26 to	6.94556	5	te	0.9406	5	Flash Point °C Fire Point	
B 1140°C	1334.6 218.	5	t _c	0.256	5	M Spec.	
C A* 26 to	1.39159	5	ΔHc kcal/m ΔHf			Ultra V.	
B* 131°C		5	ΔFf			X-Ray Dif. Infrared	
к ———			Viscosity			Solubility in +	
t _k		1	centistokes 7 °C			Acetone	
t _x i *C						Carbon tet. Benzene	
A' 5 to B' <u>26 °C</u>	7.34887	5				Ether	
c, _ 20 _	1538.8 236.	5	B ^V to			n-Heptane Ethanol	
A'* 5 to	1.74852	5	AV °C			Water Water in	
B'* 26 °C		5	(B ^V) to		1 1	water in	
Ac 140 to	7.35634 1644.	5	(A ^V) °C		$oxed{oxed}$		
Cc	257.	5	c _p liq. ∘K	ļ			
Cryos, A° consts, B°			c _p vap. °K				
t _e °C	121.42	5	c _w vap.				
$T_R = 0.75$	Tc		-			+ grams/100 grams	solvent
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-C	alc. from de	t. dat	ta 5-Calc, by formu	la
SOURCE:		A.	PI				
PURIFICAT			PI				
LITERATUR	E REFERE	NCES	3:				

							No. 99		
NAME	4-Ethyl-	l-he	kene			STRUCTURAL FORMULA			
	-					сн ₃ сн ₂ сн сн	-CH=CH-		
Mole % Pur.		ecul:		Molecular Weight 112,20	08	c ₂ H ₅	22		
		Ref.			Ref.			Ref.	
F.P. °C F.P. 100%			dt/dP °C/mm			f to			
B.P. °C 760 mm 100 30 10	113. 54.18 28.22 8.40 -24.78	2 4 4 5 5	25°C BP te 30 mm ΔHm cal/g	0.7497 0.0464 0.0363 0.6501	5 4 5 5	g			
Pressure mm 25°C t _e	25.38 1040.	5 5	25°C 30 mm BP	82.74 82.50 70.74	5 5 5	n•K			
Density g/ml 20°C dt 25 d4 30	0.726 0.722 0.718	2 2 4	t _e t _e (d, e) ΔHv/T _e	69. 20 69. 14 14. 69	5 5 5	m' to n' K o' Surface tension			
a b	0.742 -0.0 ₃ 8	4	d 28 to e 110 °C d' 15 to	86.51 0.1396 84.58	5 5 5	dynes/cm. 20°C 30 40	22.11 21.14 20.19	5 5 5	
Ref. Index n _D 20°C 25 30	1.412 1.410 1.407	2 2 4	e' 28 °C d g/ml vc ml/g tc °C	0.0737 0.201 4.985 264.	5 5 5	Parachor [P] 20°C 30			
"C"	0.7554	4	P _c mm	15331.	5	40 Sugd.	335.2	5	
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	38.5 38.677 1.049	2 5 2	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9500	5 5 5	Exp. L.1.%/wt. u. Dispersion	117.	2	
A 28 to B 130 °C	6.94966 1344.9 217.53	5 5 5	t e t c	0. 9400 0. 256	5	Flash Point C Fire Point M. Spec.			
A* 28 to B* 120 °C K c tk to c tx °C	1.39289	5 5	ΔHc kcal/m ΔHf ΔFf Viscosity centistokes γ °C			Ultra V. X-Ray Dif. Infrared Solubility in + Acetone Carbon tet.			
A' 15 to B' 28 °C C'	7.35049 1549.1 235.53	5 5 5	B ^V to			Benzene Ether n-Heptane Ethanol			
A'* 15 to B'* 28 °C	1.74901 1450.0	5 5	$\frac{\mathbf{A}^{\mathbf{V}}}{(\mathbf{B}^{\mathbf{V}}) } - \frac{\mathbf{C}}{\mathbf{to}}$			Water Water in			
Ac 130 to Bc t _c °C Cc	7.3559 1642. 254.	5 5 5	c _p liq. °K						
Cryos. A° consts. B°			c _p vap. °K						
t _e °C	124.46	5	c _w vap.						
T _R = 0.75		2 **	DI 2 I': .	C-1- (:		grams/100 gra		t	
SOURCE:	ES: I-Dow		PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for	mula		
PURIFICAT	ION:		PI						
	RE REFERE								

							No. 10	0
NAME	2, 3-Dime	thyl-	l-hexene			STRUCTURAL 1	FORMULA	
							сн ₃	
24-1-	24 14			Malandan		CH ₃ (CH ₂) ₂ CH	Ċ = CH ₂	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	80	Ċн ₃		
		Ref.			Ref			Ref.
F.P. °C			dt/dP			f to		
F.P. 100%			*C/mm 25*C	0.6817	5	8		
B. P. °C 760 mm	110.5	2	BP	0.0462	4	h		
100	52.02	4	t _e	0.0363	5	f' to		
30 10	26, 21 6, 51	4 5	30 mm	0.6463	5	h'		
1	-26.48	5	ΔHm cal/g		L	m to		-
Pressure			ΔHv cal/g 25°C	81.96	5	n *K_		
mm 25°C	28.18 1033.	5	30 mm	81.89	5	° '		
Density		1	BP te	70.20 68.70	5 5	m' to		
g/ml 20°C	0.7214	2	le (a, e)	68.65	5	n' ' *K-		
dt 25 4 30	0.7172 0.713 0	2 4	ΔHv/T _e	19.52	5	<u> </u>		
	0.7382	4	d 26 to	85.52	5	Surface tension dynes/cm. 20°C	21.55	5
ъ	-0.0383	4	-• -122 °C -• -122 ℃	0.1387 83.50	5	30	20.56	5
Ref. Index	1.4113	2	e' 26 °C	0.0617	5	40	19.58	5
n _D 20°C 25	1.4089	2	d _c g/ml	0.244	5	Parachor [P] 20°C		ļ
30	1,4063	4	v _c ml/g t _c C	4.099 278.	5	30		
"C"	0.7590	4	P _c mm	19130.	5	40 Sugd.	335.2	5
MR (Obs.) MR (Calc.)	38.64 38.677	2 5	PV/RT		 	Exp. L.1.%/wt.		
(nD-d/2)	1.0506	2	25°C 30 mm	1.0000	5	u. Dispersion	122.	2
Dielectric			BP	0.9500	5	Flash Point °C	122.	-
A 26 to	6.94766	5	te t _c	0.9406 0.256	5 5	Fire Point		
B <u> </u>	1336.0 218.	5	ΔHc kcal/m		 	M Spec.		
A* 26to	1,39348	5	ΔHf		}	Ultra V. X-Ray Dif.		
B* 130°C	1254.6	5	ΔFf		├	Infrared		
c			Viscosity centistokes		Ì	Solubility in +		
tk to C			η •c			Acetone Carbon tet.		ĺ
A' 15to	7, 35087	5	ŀ			Bensene	1	
B' _ 26°C	1540.2	5			<u> </u>	Ether n-Heptane		
C'	236.	5	B ^V to A ^V *C			Ethanol Water		
A'* 15 to B'* 26°C	1.750 34 1440.7	5	17.V	1		Water in		L
Ac 140to	7.3585	5	(A ^v) •C	1				
Bc t C	1646.	5	c _p liq. °K	 	 			
Cc	257.	5		1	1			
Cryos. A° consts. B°			c _p vap. *K					
t _e °C	121.64	5	c _v vap.					
$T_R = 0.75$	T _c			<u> </u>		+ grams/100 gran	ns solven	
		2-AI	PI 3-Lit. 4-0	alc, from det		ta 5-Calc, by for		
SOURCE:			PI					
PURIFICAT			PI					
LITERATUR	E REFERE	CES	:					

No. 101 2, 4-Dimethyl-1-hexene STRUCTURAL FORMULA NAME $CH_3CH_2CH CH_2C = CH_2$ CH₃ ĊH₃ Mole **Molecular** Ref. Molecular C8H16 % Pur Formula Weight 112.208 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm ۰ĸ g 25°C 0.6991 5 B. P. °C BP 0.0463 **4** 5 760 mm 111.2 2 0.0363 f١ to 100 52.60 4 g' •<u>к</u> 30 26.74 4 30 mm 0.6475 5 10 7.00 **5** h' ∆Hm cal/g 1 -**26.** 05 m to ΔHv cal/g Pressure n ۰ĸ 25°C 82,13 27.42 mm 25°C 5 o 30 mm 82.02 5 1035. 5 t_e 70.30 5 BP m' to Density 68.80 5 te (d, e) n' <u>°K</u> g/ml 20°C 5 0.720 68.75 ۰, 25 ž 0.716 $d_{\mathbf{4}}^{\mathbf{t}}$ ΔHv/T_e 19.51 5 30 0.712 4 Surface tension 27 85.73 5 0.736 4 dynes/cm. 20°C 21.39 5 122 0.1387 5 •C ь -0.038 4 20.44 30 a 15 83.82 to 40 19.52 5 e¹ Ref. Index •c 0.0672 5 27 20°C 1.411 [P] nD Parachor d_c g/ml 0.244 5 25 1.409 2 20°C vc ml/g t °C 4.093 30 1.406 4 30 279. 5 40 "C" 4 0.7600 P_c mm 5 19195. 5 Sugd. 335.2 MR (Obs.) 38.7 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 25°C 1.0000 5 (nD-d/2) 1.051 2 30 mm 2 1.0000 5 Dispersion 122. Dielectric BP 0.9500 Flash Point C 0.9406 A 27 to 6.94593 Fire Point 0.256 B (141 °C 1337.7 M. Spec. C 217,87 AHc kcal/m Ultra V ΔHf A* 27 to 1.39098 X-Ray Dif. ΔFf B*[130 °C 1256.3 Infrared Viscosity Solubility in centistokes Acetone to Carbon tet. °C Benzene A' 15 to 7.34836 Ether B 27 °C 1541.8 5 n-Heptane B^V A C' 235.87 5 to Ethanol °C Water A'* 15 to 1.74767 Water in (BV) B'* 27 °C 1442.4 5 to Ac 141 to 7.3568 5 (A^V)| °C Bc Ltc 1649. 5 c_p liq. ۰ĸ Cc 5 257. Cryos, A° c_p vap. ۰ĸ consts, B° c_v vap. te °C 5 122.44 $T_R = 0.75 T_c$ grams/100 grams solvent 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula REFERENCES: 1-Dow SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 102
NAME	2,5-Dim	ethyl	-1-hexene			STRUCTURAL FO	
						сн ₃ сн (сн ₂) ₂ с	= CH
Mole % Pur.	Ref. Mo	lecul	ar C ₈ H ₁₆	Molecular Weight 112.	208		H ₃
	نضاسا	Ref.			Ref		Re
F.P. °C			dt/dP			f to	
F.P. 100%			*C/mm	0.7103	5	g <u>*K</u>	
B. P. *C 760 mm	111 4	2	25°C BP	0.7102 0.0463	4	h	
100	111.6 52.95	4	t _e	0.0363	5	f' to	
30 10	27.07	4 5	30 mm	0.6480	5	g'K_	
i	7.32 -25.76	5	ΔHm cal/g			h¹	
Pressure			ΔHv cal/g		5	m to	
mm 25°C	26.95	5	25°C 30 mm	82.28 82.13	5	•	
Donaltu	1036.	5	BP	70.40	5	m' to	
Density g/ml 20°C	0.7172	2	te (d, e)	68.88 68.83	5	n' °K_	
dt 25 d4 30	0.7129	2	ΔHv/T _e	19.51	5	o'	
	0.7086	4	d 27 to		5	Surface tension	
a b	0.7344 -0.0 ₃ 85	4	<u>• 1 _123 °C</u>	0.1388	5	dynes/cm. 20°C	21.06 5
Ref. Index		†	d' 10 to		5	40	19.08 5
n _D 20°C	1.4105	2		0.241	5	Parachor [P]	
25 30	1.4080 1.4057	2 4	d _c g/ml v _c ml/g t _c °C	4.148	5	20°C	
"C"	0.7620	4		278.	5	40	
MR (Obs.)	38,80	2	P _c mm	18908.	5		35.2 5
MR (Calc.)	38.677	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.	
(nD-d/2)	1.0519	2	30 mm	1.0000	5		22. 2
Dielectric A 27 to	,	-	BP t _e	0.9500 0.9405	5	Flash Point °C	
B 140°C	6.94709 1339.4	5	tc	0.256	5	Fire Point	
<u>c</u>	217.80	5	∆Hc kcal/m			M Spec. Ultra V.	
A* 27 to	1.39172	5	ΔHf ΔFf			X-Ray Dif.	
B* 130 ℃	1257.9	5	Viscosity		-	Infrared	
·			centistokes	1		Solubility in + Acetone	
t _k to	1		η •c	Ì		Carbon tet.	
A' 10 to	7,34919	5				Benzene Ether	
B' 27 °C	1543.6	5	B ^V to	<u> </u>		n-Heptane	
A'* 10 to	235.80	5	B to A C		l	Ethanol Water	
B'* 27 °C	1.74826 1444.1	5	(B ^V) to	-[Water in	
Ac 140 to	7.3576	5	(A ^V) °C	1			
Bc tc_C	1650.	5	c _p liq. °K			}	
Cryos. A*	257.	5	1				1
consts. B°		L	c _p vap. °K				
t _e °C	122.89	5	c _w vap.				
$T_{\mathbf{R}} = 0.75$						+ grams/100 gram	s solvent
	ES: 1-Dow			Calc, from de	t. da	ta 5-Calc, by form	ula
SOURCE:		A.	PI				
PURIFICAT			PI				
LITERATUE	RE REFERE	NCES	5:				

						No. 103	3
NAME	3, 3-Dimet	hyl-	-hexene			STRUCTURAL FORMULA	1
						CH ₃	
Mole % Pur.	Ref. Mo	lecul rmula		Molecular Weight 112.2	208	сн ₃ (сн ₂) ₂ с сн=сн ₂ сн ₃	
		Ref.			Ref.		Ref.
F.P. °C			dt/dP			f to	
F.P. 1009	•	ļ	°C/mm 25°C	0.5340	5	g '° <u>K</u>	
B. P. °C 760 mm	104.	2	BP	0.0455	4	h	
100 30	46.36 20.94	4	t _e 30 mm	0.0363	5	f' to to	
10	1.65	5	ΔHm cal/g	0,6365	13	n' ====	
1	-30.18	5	AHv cal/g		1	m to	
Pressure mm 25°C	36.97	5	25°C	79.70	5	n •K	
t _e	1016.	5	30 mm BP	80, 24 68, 88	5 5	m' to	
Density g/ml 20°0	0,7140	2	t _e	67.50	5	n' i °K	
at 25	0.7099	2	t _e (d, e) ΔHv/T _e	67.46	5	0'	
	0.7058	4	d 21 to	19.54 83.11	5	Surface tension	
a b	0.7303 -0.0 ₃ 81	4	_e_ 114 °C	0.1368	5	dynes/cm. 20°C 20.68 30 19.73	5 5
Ref. Index			d' to			40 18.81	5
ⁿ D 20°C	1.4070	2 2	d _c g/ml	0.243	5	Parachor [P]	
30	1.4021	4	v _c ml/g	4.115 268.	5	20°C 30	
"C"	0.7593	4	tc°C Pcmm	18786.	5	40 Sugd. 335. 2	5
MR (Obs.) MR (Calc.		2 5	PV/RT	10.00.	+-	Exp. L. l. %/wt.	
(nD-d/2)	38.677 1.0500	2	25°C 30 mm	0.9991 1.0000	5	u.	
Dielectric			BP	0.9510	5	Dispersion 117. Flash Point C	2
A 21 to		5	t e t c	0.9422 0.257	5	Fire Point	
B (132 °C	1312.1 219.24	5	ΔHc kcal/m	 	+-	M. Spec.	
A* 21 to		5	ΔHf ΔFf			Ultra V. X-Ray Dif.	
B*[124 °C	1231.5	5	Viscosity		1	Infrared	
°	_		centistokes			Solubility in T	
t _k to			η °c			Carbon tet.	
A' to						Benzene Ether	
B'°	-		B, to		1	n-Heptane Ethanol	
A¹* to			A C			Water	
B'* °C	+		(B ^V) to			Water in	
Ac 132 to Bc t _c °C		5	(A ^V) °C		\vdash	4	
Cc C	258.	5	c _p liq. °K				
Cryos. A° consts. B°			c _p vap. *K				
t _e °C	114.36	5	c _v vap.				
$T_R = 0.7$	'5 T _C					grams/100 grams solvent	
	CES: 1-Dow	2-A		Calc. from de	t. da	ata 5-Calc. by formula	
SOURCE:			PI				
PURIFICA			PI				
LITERATU	RE REFERE	NCES	5:				

NAME	2 / 5								4	
r	ME 3,4-Dimethyl-1-hexene					- 1	STRUCTURAL FORMULA			
i							CH ₃			
							сизсизси си	сн=сн,		
Mole	Ref.		ecul		Molecular	200	с́н,	•		
% Pur.			muk	8 16	Weight 112.	_			Ref	
	1		Ref.		<u> </u>	Ref			TKE!	
F.P. C F.P. 1007		\dashv	\dashv	dt/dP *C/mm	Ì	П	f to			
B. P. *C	+			25°C	0.7214	5	g <u>°K</u>			
76 0 mm	112.		2	BP	0.0463 0.0363	5	f! to		+	
100 30	53.31 27.41		4	t _e 30 mm	0.6486	5	g' '°K			
10	7.64	. 1	5	ΔHm cal/g	0.0100	╁	h'			
11	-25.47		5	ΔHv cal/g	 	\vdash	m to	····		
Pressure mm 25°C	26,48	. 1	5	25°C	82.42	5	n '*K_			
t _e	1038.	.	5	30 mm BP	82.25	5	<u> </u>		<u> </u>	
Density	1		_	t _e	70.50 68.98	5 5	m' to			
g/ml 20°C			2	te (d, e)	68.93	5	n' *K		1	
d ₄ 25	0.72 0.71		2	ΔHv/T _e	19.52	5			—	
	0,74	-	4	d 27 to		5	Surface tension dynes/cm, 20°C	21.87	5	
Ъ	-0.03		4	d 123 to		5 5	30	20.91	5	
Ref. Index	. 1	. 1	_	e' 27 °C		5	40	19.97	5	
n _D 20°C	1.41		2 2	d g/ml	0.246	5	Parachor [P] 20°C		1	
30	1.40		4	d g/ml vc ml/g tc °C	4.068 281.	5	30			
"C"	0.75	92	4	-	19384.	5	40 Sugal	225 2	5	
MR (Obs.)			2	P _c mm PV/RT	17304.	H	Exp. L.1.%/wt.	335.2	+-	
MR (Calc. (nD-d/2)	38.67		5 2	25°C	1.0000	5	u.		1	
Dielectric		-		30 mm BP	1.00 00 0.9500	5 5	Dispersion	117.	2	
A 27 to		826	5	te	0.9405	5	Flash Point °C			
B 142°C	1341.1	ŀ	5	tc	0.256	5	Fire Point		\vdash	
<u> </u>	217.72	$\overline{}$	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		1	
A* 27 to B* 133 °C		246	5	ΔFf		1	X-Ray Dif.		1	
K 133				Viscosity	† · · · · · · · · · · · · · · · · · · ·		Infrared Solubility in +		┼	
t to	_	1		centistokes			Solubility in +			
t _x to			- 1	η •c			Carbon tet.			
A' 15 to	7.35	00	5				Benzene Ether			
B' <u>27 °C</u>	1545.3 236.		5	B ^V to	 	-	n-Heptane		1	
		004	5	B' to	ł	1 1	Ethanol Water			
A'* 15 to B'* 27 °C		1074	5	(B ^V) to	1	1 1	Water in		L	
Ac 142 to	7, 35	92	5	(A ^V) ₁ °C	1					
Bc tc_	1653.		5	c _p liq. °K	 	\vdash			1	
Cc	257.	-+	5	_						
Cryos. A° consts. B°				c _p vap. *K						
t _e °C	123.34		5	c _v vap.	ļ)]				
$T_{R} = 0.7$	5 T _c						+ grams/100 gran	ns solven	t	
REFEREN	CES: 1-D	ow 2	2-AP	I 3-Lit. 4-0	Calc. from de	t. dat	ta 5-Calc. by form	nula		
SOURCE:			AF							
PURIFICA'			AF							
LITERATU	RE REFI	EREN	CES	:						

No. 105 3,5-Dimethyl-1-hexene NAME STRUCTURAL FORMULA CH3CH CH2CH CH=CH2 Molecular C8H16 Ċнз ĊН3 Mole Ref. Molecular % Pur. Formula Weight 112,208 Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 0.5340 5 B. P. °C h BP 0.0455 4 760 mm 104. 2 0.0363 5 ſ١ 100 46.36 to g' <u>°K</u> 30 20.94 4 30 mm 5 0.6365 10 1.65 5 h! ∆Hm cal/g 1 -30.18 5 m to ∆Hv cal/g Pressure n ۰ĸ 25°C 79.70 5 mm 25°C 36.97 0 30 mm 80.24 1016.0 5 t_e ВP 68.88 5 m' to Density te te (d, e) 67.50 5 °K g/ml 20°C 0.708 2 67.46 5 ٥' 25 0,704 2 $\mathbf{d_{4}^{t}}$ ΔHv/Te 5 19.54 30 0.700 4 Surface tension 21 83.11 5 0.724 4 8 dynes/cm. 20°C 19.99 _114 °C å 0.1369 5 Ъ -0.038 4 5 30 19.09 40 18.22 5 Ref. Index e¹ 20°C 1.404 2 [P] $\mathbf{n}_{\mathbf{D}}$ Parachor d_c g/ml 0.241 5 25 1.402 2 20°C vc ml/g t_ °C 4.142 30 1.399 4 30 267. 5 t_c 40 "C" 0.7604 4 P_c mm 18629. 5 Sugd. 335.2 5 MR (Obs.) 38.8 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38,677 25°C 0.9991 5 (nD-d/2)1.050 2 30 mm 1.0000 Dispersion 2 117. Dielectric ВP 0.9510 5 Flash Point C A 21 to B 132 °C 0.9422 6. 93989 5 Fire Point 1312.1 0.257 5 5 M. Spec. С 219.24 5 AHc kcal/m Ultra V. ΔHf A* 21 to 1.39099 5 X-Ray Dif. ΔFf B*[124 °C 1231.5 Infrared ĸ Viscosity Solubility in c centistokes Acetone t_k to Carbon tet. °C Benzene A' to Ether B١ °C n-Heptane B^V A C' to Ethanol °C Water A'* to (B^V) Water in B'* °C to Ac| 132 to 7.3510 5 (AV) °C Bc t_c °C 1617. c_p liq. ۰ĸ Cc 258, 5 Cryos. A° c vap. ۰ĸ consts. B° c vap. te °C 5 114.36 $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 106
NAME	4, 4 - Dim	ethyl-	l-hexene	STRUCTURAL FORMULA			
						CH ₃	
						сн ₃ сн ₂ с сн ₂ с	CH=CH ₂
Mole % Pur.		olecul ormul		Molecular Weight 112.2	08	CH ₃	
<u> </u>		Ref		Weight 112,2	Ref	<u> </u>	Re
	T	Wei.		T	1.01		
F.P. °C F.P. 100%	 	+-	dt/dP *C/mm	1		f to K	
B, P, *C	}	+	25°C	0.6016	5	g <u>*K</u>	
760 mm	107.2	2	BP	0.0458 0.0363	5		
100	49.14	4	t _e		1 1	f' to g'*K	
30 10	23.53 4.09	4 5	30 mm	0.6413	5	h'	1
ì	-27.99	5	∆Hm cal/g			ļ	
Pressure	1		ΔHv cal/g	7. 10	_	m to	
mm 25°C	32.37	5	25°C 30 mm	71.19 81.04	5		-
t _e	1025.	5	BP	69.56	5	m' to	
Density g/ml 20°C	0.7100	,	te te (d, e)	68.12 68.08	5	n' ' °K	
	0.7198 0.7157	2 2	(a, e)		1	0'	
d ₄ 25 30	0.7116	4	ΔHv/T _e	19.54	5	Surface tension	
8	0.7361	4	d 24 to		5	dynes/cm. 20°C	21.36 5
Ъ	-0.0381	4	-a, - 118 °C			30	20.40 5
Ref. Index			e' 'c			40	19.46 5
n _D 20°C		2	d _c g/ml	0.245	5	Parachor [P] 20°C	
30	1.4078 1.4053	2 4	II A . 1111/16	4.089	5	30	
"C"	0.7588	4	1 -	273.	5	40	
MR (Obs.)	38,64	2	P _c mm	19079.	5	Sugd.	335,2 5
MR (Calc.)		5	PV/RT 25°C	0.0007	_	Exp. L.1.%/wt.	
(nD-d/2)	1.0503	2	30 mm	0.8807 1.0000	5	u. Dispersion	117. 2
Dielectric			BP	0.9510	5	Flash Point °C	
A 24 to			ţe.	0.9419 0.257	5	Fire Point	1
B [136 °C	1323.7	5	t _c	0, 257	1 -	M Spec.	
	 		ΔHc kcal/m ΔHf			Ultra V.	
A* 24 to B* 128 °C	1.39088	5	ΔFf			X-Ray Dif. Infrared	
K	.		Viscosity			Solubility in +	
tto	-		centistokes	,		Acetone	ļ
t _x to			የ ℃	'		Carbon tet.	
A' to	 	+				Benzene Ether	
B'	.			ļ	<u> </u>	n-Heptane	
C'	1		B ^V to		1	Ethanol	
A'* to			A ^V - °C	4		Water Water in	
B'* °C	 	+	(B ^V) to	i			
Ac 136 to	7,3545	5	(A ^V) °C		<u> </u>		[
Cc	258.	5	c _p liq. °K				
Cryos. A°		1	c _p vap. °K				
consts. B°			<u>-</u>				
t _e °C	117.97	5	c _w vap.	1			
$T_{\mathbf{R}} = 0.79$						grams/100 gram	s solvent
	ES: 1-Dow	2-AI		Calc, from de	t. da	ta 5-Calc. by form	ula
SOURCE:		AF	PI				
PURIFICAT		AF					
LITERATU	RE REFERI	ENCES	5:				

							No. 107			
NAME _	4,5-Dimethyl-1-hexene					STRUCTURAL FORMULA				
					1	CH ₃				
			1.			си,си си си	I ₂ CH=CH ₂			
Mole % Pur.		lecul rmula		Molecular Weight 112.2	208	с́н ₃				
W Fui.	1 1 1 1	Ref.	,	Weight 115.1	Ref.	F	Re			
F. P. °C		IXEL.			Itel.					
F.P. 100%		1	dt/dP °C/mm			f to				
B. P. °C			25°C	0.6431	5	h i				
760 mm	109. 50.69	2 4	BP t _e	0.0460	5	f' to				
100 30	24.96	4	30 mm	0.6441	5	g'° <u>K</u>				
10	5.44	5	ΔHm cal/g		† –	h'				
1	-26.78	5	ΔHv cal/g		\dagger	m to				
Pressure mm 25°C	30.06	5	25°C	79.03	5	n <u>*K</u>				
t _e	1030.	5	30 mm BP	81.47 69.92	5					
Density			t _e	68.46	5	m' to	1			
g/ml 20°C ,t 25	0.728 0.724	2 2	'e (d, e)	68.41	5	", 				
d ₄ 30	0.720	4	ΔHv/T _e	19.54	5	Surface tension				
a	0.744	4	d 25 to	84.91	5	Surface tension dynes/cm. 20°C	22.35 5			
b	-0.038	4	-d +6	0.1374	ا ء	8 30 40	21.37 5 20.42 5			
Ref. Index	1.414	2	e' °C		<u> </u>		20.42			
ⁿ D 20°C	1.412	2	d _c g/ml	0.248 4.038	5	Parachor [P] 20°C				
30	1.409	4	tc °C	277.	5	30 40				
"C"	0.7568	4	P _c mm	19460.	5		335.2 5			
MR (Obs.) MR (Calc.)	38.5 38.677	2 5	PV/RT		<u> </u>	Exp. L.1.%/wt.				
(nD-d/2)	1.050	2	25°C 30 mm	0.9703	5	u.	117. 2			
Dielectric			BP	1.0000 0.9510	5	Dispersion Flash Point C	117.			
A 25 to	6.94402	5	te tc	0.9418	5	Fire Point				
B (_140 °C_ C	1329.8	5	ΔHc kcal/m	0, 257	-	M. Spec.				
A* 25 to	1.38967	5	ΔHf	1		Ultra V.				
B* 130 °C		5	ΔFf			X-Ray Dif. Infrared				
K — — —			Viscosity centistokes			Solubility in +				
t _k _ to			η °C			Acetone				
t _x			'			Carbon tet. Benzene				
A' to B' °C						Ether				
č.' =			B _v to			n-Heptane Ethanol				
A'* to			_A' _ °C	.]		Water				
B¹+ °C	<u> </u>		(B ^V) to			Water in	 			
Ac 140 to	7.3554 1641.	5 5	(A ^V) °C		ļ	4				
Bc tc °C	258.	5	cpliq. °K							
Cryos. A°			c _p vap. °K							
consts. B°		<u> </u>	41							
t _e °C	120.00	5	c _v vap.	L		L				
$T_{R} = 0.75$						† grams/100 gra				
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula			
SOURCE:		AF								
PURIFICAT		AF								
LITERATU	RE REFERE	NCE	5:							

								No. 10	8
NAME	5,5-	Dimetl	hyl-	l-hexene			STRUCTURAL		
							CH3		
						\neg	СН ₃ С (СН ₂)2	CH=CH ₂	
Mole % Pur.	Ref.	Mole	cula	г с ₈ н ₁₆	Molecular Weight 112.2	208	CH,	_	
_/U 1 U1.			lef.	<u> </u>	weight 112.	Ref	г		Ref
F.P. °C	T		``	dt/dP	1	1	() (Ţ	1
F.P. 100%	 	一十	ᅦ	*C/mm	}		f to	l	1
B. P. *C	1		$\neg 1$	25°C BP	0.5054 0.0453	5 4	h .		1
760 mm 100	102.5 45.07		2 4	t _e	0.0362	5	f' to		
30	19.74		4	30 mm	0.6341	5	g' ' <u>°</u> K	4	
10 1	0.52		5	AHm cal/g			h'		
Pressure	-31.17		-1	ΔHv cal/g			m to		
mm 25°C	39.31	- 1	5	25°C 30 mm	79.19	5 5	n <u>*K</u>	1	1
t _e	1013.		5	BP	79.89 68.65	5		 	┼
Density g/ml 20°C	0.70	.	2	te te (d, e)	67.30	5 5	m' to	ļ	
dt 25	0.70		2	ΔHv/T _e	67.26	5	o'	-	
4 30	0.70		4		19.57	5	Surface tension		T
a b	0.72 -0.0 ₃		4	d 20 to		5	dynes/cm. 20°C	20.10	5
Ref. Index	-0.03	+	긕	d' t		1 1	30 40	19.21 18.33	5
n _D 20°C	1.40		2			+	Parachor [P]		
25 30	1.40		2 4	d g/ml vc ml/g	0.244 4.101	5 5	20°C 30	ļ	
"C"	0.76	_	4	ic C	265.	5	40	İ	
MR (Obs.)	+		2	P _c mm	18670.	5	Sugd	335.2	5
MR (Calc.)	38.67	7	5	PV/RT 25°C	0.9991	5	Exp. L.1.%/wt.	ł	İ
(nD-d/2)	1.05	01	2	30 mm	1.0000	5	u. Dispersion	117.	2
Dielectric	↓		_	BP	0.9520 0.9434	5 5	Flash Point °C	 	-
A 20 to B 130 °C			5	te t _c	0.256	5	Fire Point		<u> </u>
c Live	219.52		5	AHc kcal/m	†	\vdash	M Spec. Ultra V.		
A* 20 to	1.39		5	ΔHf ΔFf			X-Ray Dif.	l	
B* 120 °C	1226.2	1	5	Viscosity	 	\vdash	Infrared	ļ	1_
c				centistokes			Solubility in + Acetone	ļ	1
tk to				η °C	;		Carbon tet.		
t o						1 1	Benzene Ether	1	
B' •C	<u>:</u>	1	ı	B ^V to		\vdash	n-Heptane	1	
C'	 			B ^V to		1 1	Ethanol Water	ŀ	
A'* to B'* *C		1		$\frac{1}{(B^{V})} - \frac{1}{t_0}$	-	1 1	Water in	l	
Ac 130 to	7,35		5	(A ^V) °C	1				
Bc t °C	1610.6	į.	5	c _p liq. °K		\vdash			
Cryos, A°	258.				1			1	
consts. B°				c _p vap. °K	·				
t _e °C	112.71		5	c _w vap.				1	
$T_R = 0.79$			#		<u> </u>		+ grams/100 gra	ms solven	t_
REFERENC		ow 2-	-AP	I 3-Lit. 4-	Calc. from de	t. dat	ta 5-Calc. by for		
SOURCE:			AF	PI					
PURIFICAT	ION:		AP	ı					
LITERATU	RE REFI	ERENC	ES	:					

							No. 10	9
NAME	3-Ethyl-ci	s-2-	hex e ne			STRUCTURAL	FORMUL.	A
						CH ₃ CH=C (C	CH ₂) ₂ CH ₃	
Mole % Pur.		lecul	ar C ₈ H ₁₆	Molecular Weight 112.2	208	c ₂ H ₅	2,203	
	+	Ref.			Ref.			Ref.
F.P. °C			dt/dP			f to		
F.P. 100% B.P. °C	 	-	°C/mm 25°C	1.020	5	g '° <u>K</u>		
760 mm	121.	2	BP	0.0472 0.0363	4 5	f' to		
100 30	61.12	4	t _e 30 mm	0.6622	5	g''K		
10	14.50	5	ΔHm cal/g	0.0022	+ ~	h'		
l Dragound	-19.31)	ΔHv cal/g		!	m to		
Pressure mm 25°C	18.09	5	25°C 30 mm	85.30	5 5	n <u>*K</u>		
t _e	1063.	5	BP BP	84.51 72.42	5	m' to		
Density g/ml 20°C	0,737	2	t _e (d, e)	70.73	5	n' °K		
_d t 25	0.733	2	ΔHv/T	19.52	5	0'		
	0.729	4	d 35 to	89.36	5	Surface tension	22.40	
a b	0.753 -0.0 ₃ 8	4	e 134 °C	0.1400	5	dynes/cm. 20°C	23.49 22.47	5
Ref. Index			e' 35 °C	87.35 0.0819	5	40	21.49	5
ⁿ D 20°C	1.424 1.422	2 2	d _c g/ml	0.248	5	Parachor [P] 20°C		
30	1.419	4	l v ml/g	4.030 296.	5	30		
"C"	0.7646	4	t _c °C P _c mm	20096.	5	40 Sugd.	335.2	5
MR (Obs.) MR (Calc.)	38.9	2 5	PV/RT	20070.	+	Exp. L.1.%/wt.		Ť
(nD-d/2)	38.677 1.056	2	25°C 30 mm	1.0000	5	u.	124.	2
Dielectric			BP	1.0000 0.9500	5	Dispersion Flash Point C	124.	-
A 35 to	6.95770	5	te tc	0.9397 0.256	5	Fire Point		
B <u> 154 °C</u> C	1374.0 216.	5	ΔHc kcal/m	+		M. Spec.		
A* 35 to	1.39234	5	ΔHf ΔFf		l i	Ultra V. X-Ray Dif.		
B*[144°C K	1290.7	5	Viscosity			Infrared		
·			centistokes			Solubility in TACetone		
t _k to t _x °C			η •c			Carbon tet.		
A' 25 to	7.35121	5				Benzene Ether		
B' 1_ 35 °C	1578.3 234.	5	B _v to			n-Heptane Ethanol		
A1* 25 to	1,74671	5	A C			Water		
B¹* 35 °C	1478.6	5	(B ^V) to			Water in		-
Ac 154 to Bc t _c °C	7.3684 1694.	5	(A ^V) °C		1			
Cc Cc	257.	5	c _p liq. °K					
Cryos. A° consts. B°			c _p vap. °K					
te °C	133.50	5	.c _v vap.					
$T_{\mathbf{R}} = 0.79$						grams/100 gra		t
	ES: 1-Dow	2-A		Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	TON.	AI						
PURIFICAT	RE REFERE	AI						
LITERATO	RE REFERE	NCE	:					

							N o. 110	<u>. </u>
NAME	3-Ethyl-t	rans-	-2-hexene		T	STRUCTURAL	FORMULA	1
					\neg	C ₂ H	5	
Mole	Ref. Mo	lecul	ar C ₈ H ₁₆	Molecular	,,	CH ₃ (CH ₂) ₂ c =	CH-CH ₃	
% Pur.	1 1 1 1 10		a ° 10	Weight 112.2	_			Ref.
F, P. *C	Т	Ref.	1, / 15	T	Ref.		· · · · · ·	Kei.
F.P. 1007	-	\vdash	dt/dP *C/mm	İ		f to	ŀ	
B. P. *C	1		25°C BP	1.0203 0.0472	5 4	h		
760 mm 100	121.	2	t _e	0.0363	5	f! to		
30	34.68	4	30 mm	0.6622	5	g' ' <u>*</u> K_	ł	
10	14.50 -19.31	5	ΔHm cal/g			h'	ļ	<u> </u>
Pressure	1-77.51	Ť	ΔHv cal/g			m to		
mm 25°C		5	25°C 30 mm	85.30 84.51	5 5			
t _e	1063.	5	BP	72.42	5	m' to		\vdash
Density g/ml 20°0	0.737	2	te te (d, e)	70.73 70.67	5	n' °K_		
dt 25	0.733	2	ΔHv/T	19.52	5	o'	ĺ	
<u> </u>	0.729	4	d 35 to		5	Surface tension		
a b	0.753 -0.0 ₃ 8	4	e 134 °C	0.1400	5	dynes/cm. 20°C	23.49	5
Ref. Index		<u> </u>	d' 25 to		5	40	21.49	5
n_ 20°0	1.424	2	<u> </u>	0,248	5	Parachor [P]		
25	1.422 1.419	2 4	d g/ml vc ml/g	4.030	5	20°C 30		
"C"	0.7646	4	, c	296.	5	40		
MR (Obs.	38.9	2	P _c mm	20096.	5	Sugd.	335.	2
MR (Calc.	38.677 1.056	5 2	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		
(nD-d/2) Dielectric		+-	30 mm	1.0000	5	Dispersion	124.	2
A 35 t		5	BP t _e	0.9500 0.9397	5	Flash Point °C		
B 154°	1374.0	5	tč	0.256	5	Fire Point	<u> </u>	├
С	216.	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		
A* 35 to B* 144 °C		5	ΔFf			X-Ray Dif.		
K L	= 1 - 70. 1	1	Viscosity			Infrared Solubility in +	ļ	┼—
t	_	l	centistokes 7 °C			Solubility in + Acetone		1
t _k to		l	∥ უ °⊂			Carbon tet. Benzene		
A' 25 to		5				Ether		
B' L 35 °	2 1578.3 234.	5	B ^V to	<u> </u>	1	n-Heptane Ethanol		ļ
A'* 25 to		5	A ^V °C			Water		
B'* 35 °		5	(B ^V) to	-		Water in		
Ac 154 to	7.3684	5	(A ^V) °C	1			1	
Bc tc_°	257.	5	c _p liq. °K					
Cryos. Acconsts. B			c _p vap. °K					
t _e °C	133.50	5	c _v vap.					
$T_R = 0.7$				<u> </u>		grams/100 grai	ns solven	t
	CES: 1-Dow	2-A1	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5-Calc. by for		
SOURCE:		A.	PI					
PURIFICA	TION:	A	PI					
LITERATU	RE REFERE	NCES	5:					
1								
}								
L								

г т				·········			No. 11	11		
NAME	4-Ethyl-	cis-2	-hexene			STRUCTURAL FORMULA				
						C ₂ H ₅				
Mole % Pur.	Ref. Mo	lecul mul		Molecular Weight 112,20	08	CH ₃ CH ₂ CH C	H=CH CH ₃	<u></u> .		
		Ref.			Ref.			Ref.		
F.P. °C F.P. 100%			dt/dP °C/mm 25°C	0.7500		f to				
B. P. °C 760 mm 100	113. 54.18	2 4	BP t _e	0.7500 0.0464 0.0363	5 4 5	h to g' c K		_		
30 10	28.22 8.40	5	30 mm	0.6501	5	h'				
Pressure mm 25°C	-24.78 25.38	5	ΔHm cal/g ΔHv cal/g 25°C	82.95	5	m to				
t _e	1040.	5	30 mm BP	82.50 70.72	5 5	m¹ to		├-		
Density g/ml 20°C t 25 d ₄ 30	0.725 0.721 0.717	2 2 4	t _e t _e (d, e) ΔHv/T _e	69.18 69.13 19.52	5 5 5	n' <u>°K</u>				
a b	0.741 -0.0 ₃ 8	4 4	d 20 to e 124 °C d' to	86.42 0.1390	5	Surface tension dynes/cm. 20°C 30 40	21.99 21.02 20.08	5 5 5		
Ref. Index n _D 20°C 25 30	1.412 1.410 1.407	2 2 4	e' °C dc g/ml vc ml/g tc °C	0.246 4.061	5 5	Parachor [P] 20°C 30	20.00			
"C"	0.7564	4	t _c °C P _c mm	282. 19451.	5	40 Sugd.	335, 2	5		
MR (Obs.) MR (Calc.) (nD-d/2)	38.5 38.677 1.050	2 5 2	PV/RT 25°C 30 mm	1.0000	5	Exp. L.1.%/wt. u. Dispersion	119.	2		
Dielectric A 20 to	6.94966	5	BP t _e	0.9500 0.9404	5	Flash Point °C		\vdash		
B 1144 °C		5	t _c ΔHc kcal/m	0.256	5	Fire Point M. Spec.		-		
A* 20 to B* 134 °C K	1.39279	5 5	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared				
c t _k — to t _x °C			Viscosity centistokes 7°C			Solubility in Acetone Carbon tet. Benzene				
A' 15 to B' 20 °C C'	7.3505 1549.1 235.53	5 5 5	B ^V to A ^V °C			Ether n-Heptane Ethanol				
A'* 15 to B'* 20 °C	1.74901 1450.	5 5	(B ^V) to			Water Water in		_		
Ac 144 to Bc t _c °C	7.3606 1658. 257.	5 5 5	(A ^V) °C c _p liq. °K							
Cryos. A° consts. B°			c _p vap. °K							
t _e °C	124.47	5	c _v vap.							
T _R = 0.7			DT 2 T-4			grams/100 gra		ıt		
SOURCE:	ES: 1-Dow		PI 3-Lit. 4-0 PI	Laic. from de	t. da	ta 5-Calc. by for	mula			
PURIFICAT	ION:		PI							
	RE REFERE									

							No. 11	,
NAME	4-Ethyl-	trans	-2-hexene			STRUCTURAL		
						сн _з сн _з сн сн	=СНСН.	
Mole % Pur.	Ref. Me	olecul ormul	ar C ₈ H ₁₆	Molecular Weight 112.	208	с ₂ н ₅	3	
,, <u>, , , , , , , , , , , , , , , , , , </u>		Ref.			Ref			Ref.
F.P. *C			dt/dP	1		f to		
F.P. 100%			°C/mm	0.750		gK_		1
B. P. *C	112		25°C BP	0.750 0.0464	5	h		
760 mm 100	113. 54.18	2 4	te	0.0363	5	f' to		
30	28.22	4	30 mm	0.6501	5	g'	-	ĺ
10 1	8.40 -24.78	5	∆Hm cal/g			h'		<u> </u>
Pressure		+	ΔHv cal/g			m to		}
mm 25°C	25.38	5	25°C 30 mm	82.95 82.50	5	" <u></u> -	1	
t _e	1040.	5	BP	70.72	5	m' to		+
Density g/ml 20°C	0.725	2	te te (d, e)	69.18 69.13	5	n'°K	j	
dt 25	0.721	2	ΔHv/T	19.52	5	o'	}	
4 30	0.717	4	· · · · · · · · · · · · · · · · · · ·		-	Surface tension		
a b	0.741	4	d 28 to		5	dynes/cm. 20°C	21.99	5
Ref. Index	-0.038	+	d' to	5		30 40	21.02	5
n _D 20°C		2	e' j •C	+	 _	Parachor [P]		+
25 30	1.410	2	d g/ml v ml/g	0.246 4.061	5	20°C	l	İ
"C"	1.407	4	tc °C	282.	5	30 40		
	0.7564	4	P _c mm	19451.	5	10	335.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.050	2	25°C 30 mm	1.0000	5	u. Dispersion	121.	2
Dielectric			BP	0.9500	5	Flash Point °C	121.	۴
A 20 to	6.94966	5	te	0.9404 0.256	5	Fire Point	1	1
B 144 °C	217.53	5	t _c	0.250	<u> </u>	M Spec.		1
A* 20 to	+	5	ΔHf		i	Ultra V. X-Ray Dif.	1	1
B* 134 °C		5	ΔFf	ļ	<u> </u>	Infrared		1
K — — —	1		Viscosity centistokes		ł	Solubility in +		\vdash
t _k to			7	:		Acetone Carbon tet.		1
'x '			,			Benzene		i
A' 15 to B' 20 °C		5				Ether		l
c, - = =	235.53	5	B ^V to			n-Heptane Ethanol		
A'* 15 to		5	AV - °C		l	Water	1	1
B'* 20 °C		5	(B ^V) to			Water in		+-
Ac 144 to	7.3606 1658.	5	(A ^V) °C		L			1
Cc - c -	257.	5	c _p liq. °K					
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	124.47	5	c _w wap.					
$T_{\mathbf{R}} = 0.7$						grams/100 grai		t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		AP	I					
PURIFICAT		AP						
LITERATU	RE REFERE	NCES	5:					

No. 113 2, 3-Dimethyl-2-hexene NAME STRUCTURAL FORMULA $CH_3(CH_2)_2C = C CH_3$ Molecular C8H16 ĊH₃ĊH₃ Mole Ref. Molecular Weight 112, 208 % Pur Ref. Ref. F.P. °C F.P. 100% -115.1 2 dt/dP f to °C/mm g °Κ 25°C 1.0500 5 B. P. °C h RP 0.0473 4 760 mm 121.77 2 t_e 0.0363 5 ſ١ 100 61.77 4 to 4 g' <u>• к</u> 30 35.28 30 mm 5 0.6635 15.05 5 10 h' ∆Hm cal/g 1 -18.81 to ΔHv cal/g Pressure n °K 25°C 85.51 mm 25°C 17.54 0 5 30 mm 84.67 t_e 1066. 5 ВP 72.55 5 m to Density 70.85 5 te (d, e) °K g/m1 20°C 0.7408 0.7366 2 5 70.78 o' 25 ž $\mathbf{d_{4}^{t}}$ AHv/Te 5 19.51 30 0.7324 4 Surface tension \top 35 to 89, 61 5 0.7576 4 dynes/cm. 20°C 23.97 °C 0.1401 5 134 ь -0.0384 4 5 30 22.90 dח 25 to 87.57 5 Ref. Index 40 21.85 0.0824 35 5 20°C 1.4268 nD [P] Parachor d_c g/ml 0.248 5 25 1.4244 2 20°C vc ml/g tc °C 4.040 5 30 1.4217 4 30 297. 5 40 "C" 0.7654 4 P_c mm 20079. 5 Sugd. 335.2 5 MR (Obs.) 38.87 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 25°C 1.0000 5 (nD-d/2) 1.0564 2 30 mm 1,0000 2 Dispersion 127. Dielectric BP 0.9500 5 Flash Point C A 35 to 0.9397 6.95670 5 Fire Point 0.256 B (154 °C 1376.2 5 M. Spec. 215,86 5 AHc kcal/m Ultra V. A* 35 to ΔHf 1.39053 5 X-Ray Dif. ΔFf B*[144_°C 1292.8 Infrared Viscosity Solubility in centistokes Acetone t_k to Carbon tet. °C Benzene A' | 25 to 7.34943 Ether B١ 35 °C 1580.5 5 n-Heptane B^V | 5 233.86 to Ethanol °C Water A'* 25 to 1.74467 5 Water in B'* 35 °C 5 (B^V) 1480.7 to Ac 154 to 7.36734 5 (A^{V}) °C $\underline{\underline{B}}c$ $\underline{\underline{t}}c$ •c 1697. c_p liq. °K Cc 5 257 Cryos. Aº ۰ĸ c_p vap. consts. B° c_v vap. te °C 134.38 = 0.75 T_c grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

No. 114 NAME 2, 4-Dimethyl-2-hexene STRUCTURAL FORMULA CH3CH2CH CH=C CH3 ĊН3 ĊHą ${\bf Molecular}$ Mole Ref. Molecular $C_{8}H_{16}$ % Pur. Formula Weight 112, 208 Ref Ref. Ref. °C dt/dP f to F.P. 100% °C/mm °K g 25°C 0.6835 5 B. P. °C h BP 0.0462 4 760 mm 110.6 2 ſ١ 0.0363 5 to 100 52.08 4 °K g' 30 4 26, 26 30 mm 5 0.6466 10 6.55 5 h' ∆Hm cal/g -26.45 1 to AHv cal/g Pressure °K n 25°C 81.95 28.11 mm 25°C 5 30 mm 81.87 5 te 1034. 5 BP 70.18 5 to m Density g/ml 20°C te (d, e) 5 68.69 °K 0.7213 2 5 68.64 o' 25 0.7171 $\mathbf{d_{4}^{t}}$ AHv/Te 5 19.51 30 0.7129 4 Surface tension 1 20 85.51 5 5 0.7381 -0.0₃83 dynes/cm. 20°C 21.54 5 5 0.1386 122 °C ь 4 30 20.54 ā٠ to 40 19.57 5 Ref. Index e' °C $\mathbf{n}_{\mathbf{D}}$ 20°C 1.4118 2 [P] Parachor d_c g/ml 0.244 5 25 1.4094 2 20°C vc ml/g 4.106 30 1.4067 4 30 $\mathbf{t_c}$ 278. 5 40 "C" 0.7600 4 P_c mm 19100. 5 5 Sugd. 335.2 MR (Obs.) 38.69 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 25°C 1.0000 11. (nD-d/2)1.0512 1.0000 30 mm 5 Dispersion 2 124. Dielectric BP 0.9500 5 Flash Point °C t_e 0.9406 A 20 to 6.94568 5 Fire Point tç 0.256 В 1335.7 M Spec. C 218. ΔHc kcal/m 5 Ultra V. ΔHf A* 20 to 1.39139 5 X-Ray Dif. ΔFf B* 130 °C 1254.3 Infrared ĸ Viscosity Viscozz, centistokes °C Solubility in Acetone to t_x Carbon tet. •c Benzene to Ether В' °C n-Heptane B^V A^V Ċ١ Ethanol °C Water A1* to Water in B'* °C (BV) to Ac | 140 to 7.3565 (AV) °C Bc tc °C 1646. 5 cp liq. ۰ĸ Сc 257. 5 Cryos. A° c_p vap. °K consts. B° c_v vap. te °C 121.76 5 $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME	2, 5-Dim	ethyl-	-2-hexene			STRUCTURAL FORMULA			
Mole	Ref. Mo	lecul	ar	Molecular	\dashv	сн ₃ сн сн ₂ с сн ₃	н=с сн _.	3	
% Pur.	Fo	rmul		Weight 112.2	08	3	3	-,	
		Ref.			Ref.			Ref.	
F.P. °C		<u> </u>	dt/dP	-		f to			
F.P. 100	%	-	°C/mm 25°C	0.7265	5	g <u>°K</u>			
B.P. °C 760 mm	112.2	2	BP	0.0464	4	h		-	
100	53, 47	4	t _e	0.0363	5	f' to			
30 10	27.55 7.77	4 5	30 mm	0.6490	5	B E			
1	-25.36	5	△Hm cal/g		1	m to		-	
Pressure			ΔHv cal/g 25°C	82.63	5	n			
mm 25°C	26.28	5	30 mm	82.27	5	°			
Density		╁┈╴	BP t.	70.52 69.00	5	m¹ to			
g/m1 20°		2	te te (d, e)	68.94	5	n' <u>*K</u> _		1	
d ₄ 25	0.716 0.712	2 4	AHv/T _e	19.51	5			+	
a	0.736	4	d 20 to	86.10	5	Surface tension dynes/cm. 20°C	21.39	5	
ь	-0.038	4	_e124 °C	0.1388	5	8 30	20.44	5	
Ref. Inde:			e' °C			40	19.52	5	
ⁿ D 20°0	1.4115	2	d _c g/ml	0.244 4.094	5	Parachor [P] 20°C			
30	1.4091	4	v _c ml/g t _c °C	280.	5	30			
"C"	0.7652	4	P _c mm	19225.	5	40 Sugd.	335.2	5	
MR (Obs. MR (Calc		2 5	PV/RT			Exp. L.1.%/wt.			
(nD-d/2)	1.054	2	25°C 30 mm	1.0000	5 5	u.	124		
Dielectric	:		BP	0.9500	5	Dispersion Flash Point C	124.	2	
A 20 to		5	t _e	0.9405 0.256	5	Fire Point			
B (142 °C	217.68	5	tc ΔHc kcal/m	0.250	-	M. Spec.			
A* 20 to		5	ΔHf			Ultra V. X-Ray Dif.			
B*[134 °C	1259.8	5	ΔFf		-	Infrared			
c	_[Viscosity centistokes			Solubility in +			
t _k to			η ° ⊂			Acetone Carbon tet.			
A' to		┼				Benzene Ether			
B' ~					-	n-Heptane			
C'		-	B ^V to C			Ethanol Water			
A'* to			$\frac{\mathbf{B}\mathbf{v}}{(\mathbf{B}\mathbf{v})} - \frac{\mathbf{v}}{\mathbf{to}}$			Water in			
Ac 142 to	7, 3581	5	(A ^V) °C						
Bc tc °C	1653.	5	c _p liq. °K						
Cc	257.	5	il						
Cryos. A consts. B	•	1	c _p vap. °K						
t _e °C	123.57	5	c _v vap.						
$T_R = 0.$	75 Т <mark>с</mark>					grams/100 gra	ms solve	nt	
REFEREN	CES: 1-Dow			Calc, from de	t. da	ta 5-Calc. by for	mula		
SOURCE:			PI						
PURIFICA			.PI						
LITERATI	JRE REFERE	NCE	S:						
								_	

							No. 11	6
NAME	3,4-1	Dimethy	-cis-2-hexene			STRUCTURAL		
Mole	Ref.	Molecu	lar C ₈ H ₁₆	Molecular Weight 112.	208	сн ₃ сн ₂ сн с сн ₃ сн		
A Pui.	ليبلب	Res		weight 112.	Ref			Ref.
F.P. °C	Ι	I.C.		T	IV61	-	1	1
F.P. 100%			dt/dP *C/mm			f to		
B. P. °C			25°C	0.8401	5	h .		1
760 mm 100	116.	2	BP t _e	0.0467	5	f' to		
30	56.76 30.61	4 4	30 mm	0.6548	5	g' '°K_		
10 1	10.66	5	ΔHm cal/g			h'		
Pressure	-22.77	5	ΔHv cal/g	<u> </u>	 	m to		
mm 25°C	22,40	5	25°C	83.66	5	n <u>*K</u> _	ł	l
t _e	1049.	5	30 mm BP	83.22 71.33	5			ـــ
Density] t	69.74	5	m' to		1
g/ml 20°C	0.73		te (a, e)	69.68	5	" -	1	
dt 25 4 30	0.72		AHv/T _e	19.51	5	Surface tension		╁
	0.75		d 31 to		5	dynes/cm. 20°C	23.48	5
ь	-0.03	8 4	d' 20 to	85.60	5	30 40	22.47	5
Ref. Index n _D 20°C	1.41	8 2	e' 31 °C		5	Parachor [P]		+-
25	1.41	6 2	d g/ml vc ml/g	0.249 4.010	5 5	20°C		
30	1.41		tc °C	289.	5	30 40		1
"C"	0.75	$-\!\!\!\!-\!\!\!\!\!-$	P _c mm	19946.	5		335.2	5
MR (Obs.) MR (Calc.)	38.4 38.67	7 2	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.05		25°C 30 mm	1.0000 1.0000	5 5	u. Dispersion	1,24	2
Dielectric			BP	0.9500	5	Flash Point °C	124.	+-
A 31 to	6.95		t _e	0.9401 0.256	5 5	Fire Point		
B <u> 148 °C</u> C	1355.2 216.96	5	t _c	0.250	Ť	M Spec.		
A* 31 to	1.39		ΔHf			Ultra V. X-Ray Dif.		
B* 138 °C	1272.8	5	ΔFf	_	ļ	Infrared	j	
K		ĺ	Viscosity centistokes			Solubility in +		
t _k to	1		7 .c			Acetone Carbon tet.		1
tÇ °C			∄ `		. '	Benzene		1
A' 20 to B' 31 ℃	7.34 1559.4	881 5				Ether n-Heptane		i
c' -	234.96	5	B ^v l to			Ethanol		1
A'* 20 to	1.74		A ^V - °C	_		Water Water in		1
B'* 31 °C		5	(B ^V) to	1				+
Ac 148 to Bc t C	7.36 1672.	20 5	(A ^V) °C					
Cc'	257.	5	c _p liq. °K				1	
Cryos, A° consts, B°			c _p vap. °K					
t _e °C	127.86	5	c _v vap.					
$T_{R} = 0.75$	T _c					grams/100 gram	ms solven	t
REFERENC		ow 2-A	PI 3-Lit. 4-	Calc, from det	t. da	ta 5-Calc. by for		
SOURCE:		A	PI					
PURIFICAT			PI					
LITERATUR	E REFE	ERENCE	S:					
		_						

		3 4-1	lime	thyl.	trans-2-hexene		STRUCTURAL FORMULA				
NAME		3, 4-1	711116	thy1-	trans-2-nexene	,		SI	RUCTURAL	FORMUI	LA
<u> </u>		ТТ			r			C	H ₃ CH ₂ CH C	CHCH3	
Mole % Pur.		Ref.		ecul.		Molecular Weight 112.2	08		Ċн ₃ Ċн	3	
				Ref.			Ref.				Ref.
F.P. ℃	L				dt/dP			f	to		
F.P. 1009	6				*C/mm 25*C	0 840,	5	g	' <u>•</u> K_		
B. P. °C 760 mm	İ	116.		2	BP	0.8401 0.0467	4	h	<u> </u>		۷
100	-	56.76		4	t _e	0.0363	5	f'	to		1
30	1	30.61		4	30 mm	0.6548	5	g'	i ' ° K		1
10		10.66 -22.77		5 ⁻	ΔHm cal/g			h'	<u> </u>		4
Pressure	+			<u> </u>	ΔHv cal/g			m n	to or		
mm 25°C	-	22.40		5	25°C 30 mm	83.66 83.22	5	ö	! ' -		
t _e	1	049.		5	BP	71.33	5	m'	 		+-
Density	.		_		t _e	69.74	5	n'	to or	İ	1
g/ml 20°0	1	0.73 0.73		2	te (a, e)	69.68	5	اه			
d ₄ 25	\perp	0.72		4	ΔHv/T _e	19.51	5	Sur	face tension		+
a	T	0.75		4	d 31 to	87.48 0.1393	5		es/cm. 20°C	23.48	5
ь	+	-0.03	8	4	d' 20 to	85.60	5	8	30 40	22.47 21.48	5 5
Ref. Index		1.41		2	e' 31 °C	0.0779	5	Base	rachor [P]	21.40	+-
25		1.41		2	d _c g/ml	0.249	5	Pa	20°C	ļ	
30	\bot	1.41	3	4	tc °C	4.010 289.	5		30		
"C"	\perp	0.75	43	4	P _c mm	19946.	5		40 Sugd.	335.2	5
MR (Obs.		38.4	_	2	PV/RT	17,733	1	Ext	L. 1. %/wt.	-	+
MR (Calc. (nD-d/2)	Ί	38.67 1.05		5 2	25°C	1.0000	5	1	u.		
Dielectric	\top				30 mm BP	1.0000 0.9500	5		persion	124.	2
A 31 to	_	6.95	084	5		0.9401	5		sh Point C e Point]	
B 148 °C	1	355.2		5	te tc	0.256	5		Spec.		+
С	$\overline{}$	216.96		5	ΔHc kcal/m ΔHf				ra V.	ŀ	-
A* 31 to B* 138 °C		1.39 27 2. 8	076	5	ΔFf				Ray Dif.		
K	- -			٠.	Viscosity				rared	ļ	
c t₁	-				centistokes	İ			ubility in T	}	
t _k to					η ·c				rbon tet.]	-
A' 20 to	+	7.34	881	5					nzene her		
B' _31_°C		559.4 234.96		5	B _v to	 	+-	n-	Heptane		
C'	+-			_	B to		ļ		:hanol ater		
A'* 20 to B'* 31 °C		1.74 459.8	023	5	$\frac{ \mathbf{B}^{\mathbf{v}} }{ \mathbf{B}^{\mathbf{v}} } - \frac{ \mathbf{v} }{ \mathbf{t} }$				ater in		
Ac 148 to	-	7.36	20	5	(A ^V) °C	1					
Bc tc °C	1	672.	-	5		† · · · · · · · · · · · · · · · · · · ·	 				
Ce	-	257.		5	P					1	
Cryos, A ^c consts, B ^c					c _p vap. *K						
t _e °C		127.86		5	c _w vap.			<u>L_</u>		l	
$T_R = 0$.									rams/10 0 gra		nt
REFEREN	CES	: 1-D	ow	2-A		Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:					PI						
PURIFICA					PI						
LITERATU	IRE	REFI	EREI	NCES	5:						

					···			No. 11	8
NAME	3,5-	Dimeth	hyl-	cis-2-hexene			STRUCTURAL	FORMULA	A.
							сн ₃ сн сн ₂ с =	CHCH.	
Mole	Ref.	Mole	au la	_	Molecular		ċн, ċн		
% Pur.	Rei.	Forn	nula	° С ₈ Н ₁₆	Weight 112.2	08		3	
		R	lef.			Ref			Ref.
F, P. *C				dt/dP			f to		
F.P. 100%				°C/mm		ا ۔ ا	g <u>*K</u>		
B. P. *C	T	T		25°C BP	0.7214 0.0463	5 4	h ;		
760 mm 100	112.		2 4	te	0.0363	5	f¹ to		
30	27.41		4	30 mm	0.6486	5	g'		
10 1	7.64		5 5	ΔHm cal/g			h'		
Pressure	-23.41		<u>-</u> 1	ΔHv cal/g			m to	İ	
mm 25°C	26.48	. 1	5	25°C	82.42	5 5	n •K_		
t _e	1038.		5	30 mm BP	82,25 70,50	5		ļ	-
Density				te (d. a)	68.98	5	m' to	l	
g/ml 20°C	0.72		2 2	ie (a, e)	68.93	5	0' '		
d ₄ 25 30	0.71		4	ΔHv/T _e	19.52	5	Surface tension		+-
	0.74	1 .	4	d 27 to		5	dynes/cm. 20°C	21.99	5
Ъ	-0.03	8 .	4	<u>d' 123 °C</u>		5	30	21.02	5
Ref. Index	_ 1	,	.	e' 27 °C	0.0710	5	40	20.08	5
ⁿ D 20°C	1.41		2 2	d _c g/ml	0.246	5	Parachor [P] 20°C		
30	1.41		4	v _c ml/g t _c °C	4.060 281.	5 5	30		
"C"	0.76	34	4	P _c mm	19421.	5	40 Sugd.	335. 2	5
MR (Obs.)			2	PV/RT	1-7	+	Exp. L.1.%/wt.	333.1	+
MR (Calc. (nD-d/2)	38.67		5 2	25°C	1.0000	5	u.		
Dielectric			-1	30 mm BP	1.0000 0.9500	5 5	Dispersion	124.	2
A 27 to	6.94	826	5	te	0.9405	5	Flash Point °C Fire Point		
B 142 °C	1341.1	1 !	5 L	t _c	0.256	5	M Spec.		+
С	217.72		5	ΔHc kcal/m ΔHf			Ultra V.		
A* 27 to B* 133 °C			5	ΔFf			X-Ray Dif.		
K	2	1	´	Viscosity			Infrared Solubility in +		ļ
c tto	_	ł		centistokes			Solubility in + Acetone		
t _k to		l l		η •			Carbon tet.		
A' 15 to	7, 35	001	5		İ	1 1	Benzene Ether		
B' 27 °C	1545.3	1	5	_ V	-		n-Heptane		
C'	235.72		5	B ^V to			Ethanol Water		
A'* 15 to B'* 27 °C			5	75 0 — — -	- }		Water in		
Ac 142 to			5		1				
Bc t °C	1653.		5					İ	
Ce	257.		5	c _p liq. •K					
Cryos, A° consts, B°				c _p vap. °K					
t _e °C	123.34		5	c _v vap.					
$T_R = 0.7$							grams/100 grai	ns solven	t
REFEREN	CES: 1-D	ow 2-	-AP	I 3-Lit. 4-	Calc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:			API						
PURIFICAT			API						
LITERATU	RE REF	ERENC	ES:						
1									

NAME	3	5,5-I	Dime	thyl	-trans-2-hexe	ne		STRUCTURAL FORMULA				
1	— [.	\Box		•				C	H ₃ CH CH ₂ C		,	
Mole % Pur.		Ref.		ecul: muls		Molecular Weight 112	. 208		с́н ₃ с́н	¹ 3		
				Ref.		T	Ref				Ref.	
F. P. °C					dt/dP			f	to			
F.P. 1009	•				°C/mm 25°C	0.731	. _	g		1		
B. P. °C 760 mm	112	,		2	BP	0.7214 0.0463		h	1			
1 0 0	53	3.31		4	t _e	0.0363		f'	to to			
30 10		7.41 7.64		4 5	30 mm	0.6486	5	g' h'	<u>*K</u>			
1		5.47		5	ΔHm cal/g			m	l to		+-	
Pressure					ΔHv cal/g 25°C	82.42	5	n	*K	}		
mm 25°C	1038	5.48 3.		5	30 mm	82,25	5	0	1 			
Density	1.050		-		BP	70.50 68.98	5	m'	to			
g/ml 20°C		725		2	t _e (d, e)	68.93	5	n' o'	<u> •K</u>	-		
d ₄ 25 30		0.721 0.717		2 4	AHv/Te	19.52	5				4_	
a		741		4	d 27 to		5		face tension es/cm. 20°C	21.99	5	
ь		0.038		4	d 123 %		5 5	∥ ชั′"	30	21.02	5	
Ref. Index	. 1		.	_	e' 27 °(<u> </u>	40	20.08	5	
ⁿ D 20°C	-	l.416 l.414		2 2	d _c g/ml	0.246	5	Par	rachor [P] 20°C			
30	1_1	1.41		4	v _c m1/g t _c °C	4.060 281.	5		30	1		
"C"		763	34	4	P _c mm	19421.	5		40 Sugd.	335.2	5	
MR (Obs.) MR (Calc.	. 1	3.9 3.677	,	2	PV/RT			Ext	o. L.1.%/wt.			
(nD-d/2)		1.054		2	25°C 30 mm	1.0000		1	u.			
Dielectric					BP	0.9500		1	persion sh Point °C	124.	2	
A 27 to		. 948	326	5	t _e	0.9405 0.256	5 5		e Point			
B 1142 °C		l.1 7.72		5	t _c ΔHc kcal/m		-		Spec.			
A* 27 to		392	246	5	∆Hf				ra V. Ray Dif.			
B*[133 °C				5	ΔFf				rared			
K c					Viscosity centistokes				ubility in +			
t _k to					η °⊂	:			etone arbon tet.			
A' 15 to		7.350	101	5					enzene			
B' 27 °C			~``	5	w 1				her Heptane			
C'	$\overline{}$	5.72		5	B ^V to				hanol ater			
A'* 15 to B'* 27 °C	1445	1.748 5.8	394	5	$\frac{1}{ \mathbf{B}^{\mathbf{v}} } - \frac{1}{t_0}$	-1			ater ater in		\perp	
Acl 142 to	7	7.359)3	5	(A ^V) °C							
Bc tc °C	1653	3.	-	5	c liq. °F		\neg	1			1	
Cryos A°	257	•		5	р -						1	
Cryos, A° consts, B°					c _p vap. °K	`						
t _e °C		3.34		5	c _v vap.							
$T_R = 0.7$	5 T _c							+ g	rams/100 gra	ms solve	nt	
REFEREN	CES:	1-De	w	2-A	PI 3-Lit. 4	-Calc, from	det. da	ata 5	-Calc. by for	mula		
SOURCE:					PI							
PURIFICA					PI							
LITERATU	RE R	EFE	REN	CES	S:							
											_	

									No. 17	20
NAME	4,4	Dimeth	yl-cis-2-he	xene			STR	UCTURAL 1	FORMUL	.A
-						\neg		СН3		
				$\neg \tau$			СН	зсн2с сн=	снсн.	
Mole	Ref.				Molecular	- 1	-	CH ₃	3	
% Pur.		Form		6	Weight 112.	_		3		
		Re	<u>'</u>		·	Ref.				Ref
F.P. *C F.P. 100%	_		dt/dP		ļ		f	to		
	'		*C/mn 25*C	n	0,5755	5	g	<u>*K</u>		1
B. P. °C 760 mm	106.	2	BP		0.0457	4	h	 		-
100	48.11	. 4	Te .		0.0363	5	f' g'	to *K		1
30 10	22.57 3.19				0.6395	5	h'	` ⁻ -		
ì	-28.80			1/g	<u> </u>		<u> </u>	1		+
Pressure			ΔHv cal 25°C	l/g	00.43	5	m n	to *K		1
mm 25°C	34.01 1020.	5	ll 20	1	80.42 80.76	5	0			1
t _e	1020.	3	II BP		69.25	5	m'	to	·	+
Density g/ml 20°C	0.72	2 2	t (d, e	.)	67.83 67.79	5	n'	•K_		1
at 25	0.71	8 2	A 13/7		19.53	5	٥'	! — — — — — — — — — — — — — — — — — — —		1
4 30	0.71		- a	e 22 to		5	Surf	ace tension		1
a b	0.73		1 - 1 1	30 °C		5	dyne	s/cm. 20°C	21.62	5
Ref. Index	-0.0	*	a'	_ tc	5		•	30 40	20.68 19.76	5
n _D 20°C	1.41	3 2	e' i	·••		+	Para	chor [P]		+
25	1.41	1 2	a _c g/m	ıl o	0.248 4.038	5		20°C		1
30	1.40		-11 to 10		272.	5	l	30 40		
"C"	0.76		⊸ P _a mm		19212.	5			335.2	5
MR (Obs.) MR (Calc.)	38.8	77 2	DV/DT			 	Exp.	L.1.%/wt.		+-
(nD-d/2)	1.05		25°C		1.0000	5	1	u.		1.
Dielectric			30 mm BP		1.0000 0.9500	5		ersion	119.	2
A 22 to	6.94	277 5	ا ب		0.9410	5		h Point °C Point		1
B [_136.°C		. 5			0.256	5	M SI			+
C A* 22 to	218.86		ΔHc kca ΔHf	ıl/m	İ		Ultra	a V.		
A* 22 to B* 136 °C	1.39	347 5	ΔFf				X-Ra Infra	ay Dif.		İ
к — — —			Viscosi				<u> </u>	bility in +		┿
t _k	-{	Ì	centisto	ke s °C	.]			tone		1
t _x C		ŀ	7		'			bon tet.		
A' to			┪				Eth			
B' •	<u>:</u>	ł	B ^V		 	├		leptane		
A'* to	 		╢╬┱┆	to •C	:]	1	Wat	anol ter		
B'* *C		- 1	(BV)	to	- l			ter in		
Ac 136 to	7.35		(A ^V)	°C	ł					
Bc t C	1627.	5		°K		\vdash	l			
Ce	258.	5	⊣ ।		1		1			
Cryos, A° consts, B°		- 1	c _p vap.	•K						
t _e °C	116.57	, 5	c _v vap.							
$T_R = 0.7$	5 T _C				<u> </u>		+ gra	ms/100 gran	ns solve	nt
REFERENC	ES: 1-D	ow 2-	PI 3-Lit.	4-	Calc. from de	t. da				
SOURCE:		A	PI							
PURIFICAT	ION:	A	PI							
LITERATU	RE REF	ERENCE	CS:							

							No. 12	21
NAME	4, 4-D	imethyl	-trans-2-hexen	e		STRUCTURAL	FORMUL	Α
						СH ₃		
· · ·						сн ₃ сн ₂ с сн	=CHCH ₃	
Mole % Pur.	Ref.	Molecul Formula	ar c_8H_{16}	Molecular Veight 112.20	۱۵	Ċн ₃		
- / · · · · · · ·		Ref.		l eight 112.20	Ref.	<u> </u>		Ref.
F.P. °C	1	1	dt/dP	1	IXEI.	4 1		101.
F.P. 100%			°C/mm			f to g*K_		ł
B. P. °C			25°C BP	0.5755 0.0457	5 4	h		1
760 mm 100	106. 48.11	2	te	0.0363	5	f¹ to		
30	22.57	4	30 mm	0.6395	5	g' <u>'°K</u>		
10	3. 19 -28. 80	5	∆Hm cal/g			h'		<u> </u>
Pressure		_	ΔHv cal/g			m to		
mm 25°C	34.01	5	25°C 30 mm	80.42 80.76	5			
Density	1020.	5	BP	69.25	5	m¹ to		\vdash
g/ml 20°C	0.722	2	t _e (d, e)	67.83	5	n' °K		
at 25	0.718	2	ΔHv/T	19.53	5	0'		
	0.714	4	d 22 to	83, 87	5	Surface tension		-
a b	-0.038	4	e 130 °C	0.1380	5	dynes/cm. 20°C	21.62 20.68	5
Ref. Index			d' to)		40	19.76	5
ⁿ D 20°C	1.413	2 2	d _c g/ml	0.248	5	Parachor [P]		
30	1.409	4	v _c ml/g	4.038 272.	5	30		ļ
"C"	0.761	3 4	t _c °C P _c mm	19212.	5	40 Sugd.	335.2	5
MR (Obs.)		2	PV/RT	17212.	<u> </u>	Exp. L.1, %/wt.	333.2	
MR (Calc. (nD-d/2)	38.677	5 2	25°C	1.0000	5	u.	_	
Dielectric		\pm	30 mm BP	1.0000 0.9500	5	Dispersion	121.	2
A 22 to	6. 942	77 5	t _e	0.9410 0.256	5	Flash Point C Fire Point		
B (_136 °C	_ 1319.6 218.86	5	tc ΔHc kcal/m	0.230	-	M. Spec.		t^-
A* 22 to			∆Hf			Ultra V. X-Ray Dif.		
B*[136 °C		5	ΔFf		ļ	Infrared		
K — — —		i	Viscosity centistokes		}	Solubility in +		
t _k to			η °C			Acetone Carbon tet.		
t _x °C						Benzene		
B' C					<u> </u>	Ether n-Heptane		
C'	+		B ^V to A ^V °C			Ethanol		
A'* to B'* °C		1	$\frac{1}{(\mathbf{B}^{\mathbf{v}}) } - \frac{\mathbf{c}}{\mathbf{t}_0}$			Water Water in		
Ac 136 to			(A ^V) °C					
Bc tc °C		5	c _p liq. °K					
Cryos. A°	250.	- 3	_					}
consts. B			Р					
t _e °C	116.57	5	c _v vap.					
$T_R = 0.7$	5 T _C					grams/100 gra	ms solven	ıt
REFEREN	CES: 1-Do	w 2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:			PI					
PURIFICA'			PI					
LITERATU	RE REFE	RENCES	5:					
Territoria (1971)								

							N o. 12	22
NAME	4,5	-Dimethy	l-cis-2-hexene			STRUCTURAL		
						כם כם כם כם	1-CU CU	
·			. T			сн ₃ сн сн сі сн ₃ сн ₃	1-0110113	į.
Mole % Pur.	Ref.	Molecu Formu	lar C ₈ H ₁₆	Molecular Weight 112.	208	0.1.30.1.3		
		Ref			Ref.			Ref.
F.P. °C			dt/dP	1.		f to		
F.P. 1009	·		*C/mm 25*C	0.6682	5	g <u> </u>		
B. P. *C 760 mm	110.	2	BP	0.0461	4	h	<u> </u>	-
100	51.56	4	t _e	0.0363	5	f' to		1
30 10	25.78 6.10		30 mm	0.6456	5	h'	1	
11	-26.86	5	ΔHm cal/g	 	-	m to	l	+-
Pressure mm 25°C	28.82	5	ΔHv cal/g 25°C	81.77	5	n ' °K_		
t _e	1032.	5	30 mm BP	81.73 70.07	5	0		—
Density			1 t.	68.58	5	m' to	İ	
g/ml 20°C	0.72		Te (a, e)	68.53	5	0, ' 2.	1	
dt 25 4 30	0.71		ΔHv/T _e	19.52	5	Surface tension		+-
8	0.74		d 20 to		5	dynes/cm. 20°C	21.99	5
b Ref. Index	-0.03	8 4	1 d' to	Ī,		30 40	21.02	5
n _D 20°0					5	Parachor [P]		_
25 30	1.41		d g/ml vc ml/g	0.247 4.051	5	20°C 30		
"C"	0.75		1 10	278.	5	40	1	
MR (Obs.)	+	2	P _c mm	19361.	5		335.2	5
MR (Calc. (nD-d/2)			PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.	1	l
Dielectric	1.05	1 2	30 mm BP	1.0000 0.9500	5	Dispersion	119.	2
A 20 to	-	544 5	t _e	0.9407	5	Flash Point °C Fire Point		
B 1140 °C	1333.6	5.	t _c	0.256	5	M Spec.	 	+
C	218.	5	ΔHc kcal/m			Ultra V.		
A* 20 to B* 131 °C		179 5	ΔFf	<u> </u>		X-Ray Dif. Infrared		
K	-		Viscosity			Solubility in +		+
ել			7 °C			Acetone Carbon tet.		
*x '			1			Benzene		
A' to					L.,	Ether n-Heptane		
C'			B ^V to	1		Ethanol		
A'* to B'* °($\frac{A^{\mathbf{v}}}{(B^{\mathbf{v}})_1} - \frac{{}^{\circ}C}{t_0}$	-		Water Water in		
Ac 140 to	7, 35	66 5	1					T
Bc t °	1645.	5	c _p liq. °K					
Ce	230.	5	-11 -					
Cryos. A° consts. B°			c _p vap. °K					1
t _e °C	121.08	5	c _v vap.					
$T_R = 0.7$						+ grams/100 gran	ms solver	nt
REFEREN	CES: 1-D			Calc, from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			PI					
PURIFICA			PI					
LITERATU	KE REF	ERENCE	5:					

<u></u>							No. 12	23
NAME	4,5-Dime	thyl-	trans-2-hexene			STRUCTURAL	FORMUL	,A
						си,си си с	H=CH CH.	
Mole	Ref. Mo	ecul		Molecular		ċн¸ċн¸	3	
% Pur.		mula		Weight 112.20	8	33		
	-	Ref.			Ref.			Ref.
F. P. °C	ļ		dt/dP			f to		
F.P. 100%			°C/mm 25°C	0.6682	5	g '° <u>K</u>	ł	1
B. P. °C 760 mm	110.	2	B P	0.0461	4	h		-
100 30	51.56	4	t _e	0.0363	5	f' to		
10	25.78 6.10	4 5	30 mm	0.6456	5	h'		
1	-26.86	5	ΔHm cal/g ΔHv cal/g		\vdash	m to		
Pressure mm 25°C	28, 82	5	25°C	81.77	5	n <u>*K</u>		
t _e	1032.	5	30 mm B P	81.73 70.07	5 5			ļ
Density g/ml 20°C	0.735	,		68.58	5	m' to		
dt 25 4 30	0.725 0.721	2	t _e t _e (d, e)	68.53	5	o'		
	0.717	4	ΔHv/T _e	19.52 85.30	5	Surface tension		
a b	0.741 -0.0 ₃ 8	4	_e_ _121 <u>°C</u>	0.1385	5	dynes/cm. 20°C	21.99 21.02	5
Ref. Index	1	<u> </u>	d' to			40	20.08	5
ⁿ D 20°C	1.413	2	d _c g/ml	0,247	5	Parachor [P]		
30	1.411	2	v ml/g	4.051	5	20°C 30		
"C"	0.7582	4	•	278. 19361.	5	40 Suad	225 2	5
MR (Obs.)	38.6	2	P _c mm	17301.	-	Sugd. Exp. L.1.%/wt.	335.2	1-
MR (Calc.) (nD-d/2)	38.677 1.051	5 2	25°C	1.0000	5	u.		
Dielectric		_	30 mm BP	1.0000 0.9500	5	Dispersion	121.	2
A 20 to	6.94544	5	t _e	0.9407 0.256	5	Flash Point °C Fire Point		
B (<u>140</u> ℃ C	1333.6 218.	5 5	tc ΔHc kcal/m	0.250	-	M. Spec.		T-
A* 20 to	1.39179	5	∆Hf			Ultra V. X-Ray Dif.		
B* 131 °C	1252.4	5	ΔFf		-	Infrared		
с	}		Viscosity centistokes			Solubility in +		
t _k to	1		η •⊂			Acetone Carbon tet.		
t _x °C						Benzene Ether		
B'°C			B _w to		\vdash	n-Heptane		
A¹* to			B to			Ethanol Water		
B'* °C			(B ^V) to			Water in		<u> </u>
Acl 140 to	7.3566	5	(A ^V) °C					
Bc t _c °C	1645. 258.	5 5	c _p liq. °K					
Cryos. A°			c _p vap. °K					
consts. B°	-	L_	_					
t _e °C	121.08	5	c vap.	L	L	L	L	<u> </u>
$T_R = 0.79$	ES: 1-Dow	2. 4	PI 3-Lit. 4-0	Calc from do	+ 4-	ta 5-Calc. by for		ıt
SOURCE:	PO: I-DOM		PI 3-Lit. 4-0	Jaic, irom de	ı. ua	ia 3-Caic, by for		
PURIFICAT	ION:		.PI					
	RE REFERE							
L								

							No. 12	4
NAME	5,5-Dim	ethyl	-cis-2-hexene			STRUCTURAL I	FORMUL	A.
						CH ₃		
Mole	Ref. Mo	lecul		Molecular		CH ₃ C CH ₂ CH	=СН СН 3	
% Pur.		rmul		Weight 112.2	08	ĊH ₃		
		Ref.			Ref.			Ref
F.P. C		L.	dt/dP			f to		ļ
F. P. 100%	<u>'</u>	<u> </u>	*C/mm 25*C	0.5950	.5	g <u>*K</u>		
B. P. *C 760 mm	106.9	2	BP	0.0458	4	h		-
100 30	48.88 23.29	4	t _e 30 mm	0.0363	5	f' to		
10	3.86	5	ΔHm cal/g	0.0407	۲	h'		ļ
1	-28.19	5	ΔHv cal/g	ļ	-	m to		
Pressure mm 25°C	32, 77	5	25°C	80.74	5	n •K		l
t _e	1023.	5	30 mm BP	80.97 69.42	5	 		<u> </u>
Density	0.71/0		t_	67.99	5	m' to		
g/ml 20°C dt 25 d4 30	0.7169 0.7125	2 2	te (d, e)	67.94	5	o' '		1
^d 4 30	0.7081	4	ΔHv/T _e	19.52	5	Surface tension		\vdash
a b	0.7344 -0.0 ₃ 87	4 4	e 118 °C		5	dynes/cm. 20°C	21.02	5
Ref. Index		+-	d' to	}		30 40	20.00 19.00	5
n _D 20°C	1.4113	2		0, 242	5	Parachor [P]		
25 30	1.4088 1.4060	2 4	d g/ml vc ml/g tc °C	4.128	5	20°C		
"C"	0.7638	4	11	271.	5	40		_
MR (Obs.)		2	P _c mm	18758.	5		335.2	5
MR (Calc. (nD-d/2)) 38.677 1.0528	5 2	25°C	1.0000	5	Exp. L.1.%/wt. u.		
Dielectric	1	+-	30 mm BP	1.0000 0.9500	5	Dispersion	119.	2
A 23 to	6.94315	5	te	0.9409	5	Flash Point °C Fire Point		
B (135 °C		5	t _c	0.256	5	M Spec.		
A* 23 to		5	ΔHc kcal/m ΔHf		1	Ultra V.		
B* 126 °C		5	ΔFf			X-Ray Dif. Infrared		
K — — —			Viscosity centistokes			Solubility in +		
t _e to			7 °c			Acetone Carbon tet.		
'x	1	<u> </u>				Benzene		
A' to B' L *C						Ether n-Heptane	i	
C'			B ^V to A ^V C			Ethanol		
A'* to B'* *C				-{		Water Water in		
Ac 135 to	+	5	(A ^V) to					
Bc t °C	1629.	5	c _p liq. °K		-			
Cc	258.	5	1)					
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	117.58	5	c _v vap.					1
$T_{R} = 0.7$						+ grams/100 gran	ns sol ve n	t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5-Calc. by form		
SOURCE:			PI					
PURIFICAT			PI					
∟ ITERATU	RE REFERE	NCES	5:					

							No. 1	25
NAME _	5,5-Dim	ethyl-	trans-2-hexene			STRUCTURAL	FORMUI	LA
1						CH ₃		
	T 1					CH ₃ C CH ₂ CH	=CHCH2	
Mole	Ref. M	olecul		Molecular		с́н,	,	
% Pur.	F			Weight 112.20				-
		Ref.			Ref.			Ref.
F. P. °C	ļ	-	dt/dP			f to		
F.P. 100%	+	┼	*C/mm 25*C	0.5363	5	g <u>'°K</u>		
B. P. °C 760 mm	104.1	2	BP	0.0455	4	h		+
100	46.46	4	t _e	0.0363	5	f' to		
30	21.04	4	30 mm	0.6366	5	g'° <u>K</u>		1
10 1	1.74	5	∆Hm cal/g			h'		
Pressure	1	1	ΔHv cal/g	ļ		m to		
mm 25°C	36.79	5	25°C 30 mm	79.74 80.29	5			
t _e	1015.	5	BP	68.84	5	m¹ to		+-
Density		_	t _e ,	67.45	5	m' to		
g/ml 20°C	0.7066 0.7023	2 2	te (d, e)	67.41	5	0'		
d ₄ 25	0.6980	4	ΔHv/T _e	19.52	5	Surface tension		+-
a	0,7237	4	d 20 to	83.19	5	dynes/cm. 20°C	19.83	5
ъ	-0.0385	4	-å- 120 °C	0.1379	5	8 30	18.87	5
Ref. Index	1 4055	,	e' °C			40	17.94	5
ⁿ D 20°C	1.4055 1.4030	2 2	d _c g/ml	0.24	5	Parachor [P] 20°C	İ	
30	1.4002	4	v _c ml/g t _c °C	4.17 266.	5	30		
"C"	0.7646	4	tc°C Pc mm	18389.	5	40 Sugd	335.2	5
MR (Obs.)		2	PV/RT	10307.	-	Exp. L.1.%/wt.	333.2	+-
MR (Calc. (nD-d/2)	38.677	5 2	25°C	1.0000	5	u.		
Dielectric	1.0322	+-	30 mm BP	1.0000 0. 9500	5	Dispersion	121.	2
A 20 to	6. 94095	5		0.9412	5	Flash Point C		
B 131 °C		5	t e t c	0.256	.5	Fire Point		
с — — —	219.22	5	AHc kcal/m			M. Spec. Ultra V.		
A* 20 to	1.39372		ΔHf ΔFf			X-Ray Dif.		
B*[122 °C K	- 1232.6	5	Viscosity		 -	Infrared		+
°		-	centistokes			Solubility in Acetone		
t _k to			η °C		l	Carbon tet.		
A' to		+			l	Benzene Ether		
B'	_				ļ	n-Heptane		
			B ^V to A ^V °C		l	Ethanol		
A'* to B'* °C			⊩ ⊸	.		Water Water in		
Ac 131 to	7 2510	-	11					1
Bc tc °C		5		 	├	1	-	
Cc	258.	5	c _p liq. °K	1				
Cryos. A°			c _p vap. °K					
consts. B°	1	+	c _v vap.					i
t _e °C	114.43	5	1 -v F.	L	<u>L</u>	L	1	
$T_{\mathbf{R}} = 0.7$						grams/100 gra		nt
	ES: 1-Dow			Calc. from de	t. da	ata 5-Calc. by for	mula	
SOURCE:			PI					
PURIFICAT	TION:	A	PI					
LITERATU	RE REFER	ENCE	S:					
]								

							N o. 12	6
NAME	3-Ethyl-	3-hex	cene			STRUCTURAL	FORMULA	A.
						CH CH C - C	ucu cu	
	2614	••		M-11		$CH_3CH_2C = C$ C_2H_5	n Cn ₂ Cn ₃	3
Mole % Pur.	Ref. Mo	rmul	ar C ₈ H ₁₆	Molecular Weight 112.2	208	25		
		Ref.			Ref			Ref.
F.P. °C			dt/dP			f to		1
F. P. 1007	'	ļ	*C/mm 25*C	0.8401	5	g <u>*K</u>		
B. P. °C 760 mm	116.	2	BP	0.0467	4	h		┼
100 30	56.76 30.61	4	t _e 30 mm	0.0363 0.6548	5 5	f' to		1
10	10.66	5	ΔHm cal/g	0.0340	╁┤	h'		
1	-22.77	5	ΔHv cal/g	<u> </u>	\vdash	m to		
Pressure mm 25°C	22.40	5	25°C	83.66 83.22	5	n •K	İ	
t _e	1049.	5	30 mm BP	71.32	5	m' to		├
Density g/ml 20°0	0.729	2	te te (d, e)	69.73 69.67	5 5	n' °K_		
dt 25	0.725	2	ΔHv/T	19.51	5	o' '		
4 30	0.721	4	d 31 to	87.48	5	Surface tension dynes/cm, 20°C	22.48	5
ь	-0.038	4	d' 1 128 ℃		5 5	30	21.50	5
Ref. Index		,	e' 15 to		5	40	20.54	5
D 25	1.416	2	d g/ml	0.246	5	Parachor [P] 20°C		
30	1.413	4	tc *C	4.057 287.	5	30 40		1
"C"	0.7626	4	P _c mm	19646.	5		335.2	5
MR (Obs. MR (Calc.		5	PV/RT	1,0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.054	2	25°C 30 mm	1.0000	5	u. Dispersion	124.	2
Dielectric			BP t _e	0.9500 0.9401	5 5	Flash Point °C		
B 147°		5	tc	0.256	5	Fire Point		┼
C	217.	5	ΔHc kcal/m ΔHf	1		M Spec. Ultra V.		
A* 31 to B* 138 °C		5	ΔFf			X-Ray Dif. Infrared		
K C	_		Viscosity			Solubility in +		\vdash
t _k Te			centistokes 7 °C	1		Acetone Carbon tet.		1
t			•			Benzene		
B' _ 31 °		5			\perp	Ether n-Heptane		
C'	235.	5	B ^V to			Ethanol Water		1
A'* 15 to B'* 31 °c		5 5	$\frac{1}{(\mathbf{B}^{\mathbf{V}})} - \frac{1}{\mathbf{t}_0}$	-	1 1	Water in		
Ac 147 to		5	(A ^V) •C	1				
Bc tc_	C 1671. - 257.	5	c _p liq. °K					
Cryos. A	,	Ė	c _p vap. °K					
consts. B			c _v vap.					
t _e °C T _R = 0.7	127.86	5	~ ·-P·	<u> </u>	\perp	L		J
	CES: 1-Dow	2-AI	PI 3-Lit, 4-0	Calc from de	t de	grams/100 grar ta 5-Calc. by for		<u>.t</u>
SOURCE:			PI	oute. Hom de	t. da	ia 3-Caic, by lori	IIuia	
PURIFICA	TION:	Al	PI					
LITERATU	RE REFERE	NCES	5:			-		

No. 127 2, 2-Dimethyl-cis-3-hexene NAME STRUCTURAL FORMULA CH3C CH=CHCH2CH3 Molecular C8H16 Mole Ref. Molecular Weight 112,208 Ċнз % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% 2 dt/dP -137.350 f to °C/mm g °K 25°C 0.5627 5 B. P. ℃ h ВP 0.0457 4 760 mm 105.43 2 0.0363 5 f١ to 100 47.59 4 <u>°к</u> g' 30 mm 30 22.07 4 0.6388 5 10 2,71 5 h' ∆Hm cal/g 5 1 -29.23 to m ∆Hv cal/g Pressure °K n 25°C 80.17 5 mm 25°C 34.89 o 30 mm 80.57 5 1019. te 5 ВP 69.08 5 m' to Density 5 67.67 n' g/ml 20°C te (d, e) °K 0.7128 67.63 5 ٥' d_4^t 25 0.7086 2 ΔHv/Te 19.51 5 30 0.7044 4 Surface tension ď 22 83,62 to 5 0.7295 20.54 5 dynes/cm. 20°C ᇷᅴ 125 °C 0.1379 5 ь -0.0383 19.58 18.64 5 30 40 5 Ref. Index e' ^{n}D 20°C 1.4099 2 (P) Parachor d_c g/ml 0.242 5 25 1.4074 2 20°C vc ml/g t_°C 4,130 30 1.4049 4 30 ^tc 269. 5 "C" 40 0.7657 4 P_c mm 18677. 5 5 Sugd 335.2 MR (Obs.) 38.99 2 PV/RT 38.677 Exp. L. l. %/wt. MR (Calc.) 5 25°C 1,0000 5 (nD-d/2)1.0535 2 30 mm 1.0000 5 Dispersion 2 119. Dielectric BP 0.9500 5 Flash Point C 0.9411 A 22 to 6.93979 Fire Point 0.256 B | 134 °C 1316.7 M. Spec. Ultra V. 218.97 5 AHc kcal/m ΔHf A* 22 to 5 1.39110 X-Ray Dif. ΔFf B*[126 °C 1236.3 Infrared ĸ Viscosity Solubility in centistokes Acetone to t_k | Carbon tet. °C Benzene A'I Ether B' ۰c n-Heptane Bv | Av | C' Ethanol °C Water A'* to Water in (B^V) °C to Ac | 134 to (A^V)| 7,3508 5 5 5 °C Bc tc °C 1622. cp liq. ۰ĸ Cc 258. Cryos. Aº cp vap. °K consts. B° t_e °C c_w vap. 115.93 5 $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

· · · · · · · · · · · · · · · · · · ·						No.	128
NAME	2, 2-Di	imethyl	-trans-3-hexen	e		STRUCTURAL FORMU	LA
l	***					CH ₃	
						сн ₃ с сн=сн сн ₂ сн	
Mole	Ref. 1	Molecul		Molecular		CH ₃	•
% Pur.				Weight 112.2	_	- 3	- E -
		Ref.		r	Ref.		Ref
F.P. °C F.P. 100%			dt/dP	1	i i	f to	ŀ
B. P. *C	<u> </u>	\dashv	*C/mm 25*C	0.4755	5	g <u>*K</u>	1
760 mm	100.85	2	BP	0.0452	4	h	
100 30	43.64	4	t _e	0.0363	5	f' to to	
10	18.40	4 5	30 mm	0.6317	5	h'	
1	-32.33	5	ΔHm cal/g	 	-	m to	
Pressure			∆Hv cal/g 25°C	78.56	5	n <u>*K</u>]	
mm 25°C	42.09 1006.	5 5	30 mm	79.47	5	0	ł
Density	1.000.	+-	BP	68.14 66.81	5	m¹ to	
g/ml 20°C	0.7039	2	te (d, e)	66.78	5	n' <u>*</u> K_	
dt 25 4 30	0.6995		AHV/Te	19.53	5	0'	
	0.6949		d 15 to		5	Surface tension	
a b	0.7218 -0.0 ₃ 88			0.1374	5	dynes/cm. 20°C 19.53 30 18.54	
Ref. Index		\vdash	d' to		1 1	40 17.58	
n _D 20°C				0, 239	5	Parachor [P]	
25 30	1.4037		d g/ml vc ml/g tc °C	4.189	5	20°C	-
"C"	0,7689			260.	5	40	1
MR (Obs.)	+	2	P _c mm	18108.	5	Sugd. 335.2	5
MR (Calc.	38.677	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.	-
(nD-d/2)	1.0544	2	30 mm	1.0000	5	u. Dispersion 121.	2
Dielectric		\perp	BP	0.9500	5	Flash Point °C	
A 15 to B 127 °C		9 5	t _e t _c	0.9415 0.256	5	Fire Point	
c E	219.84	5	AHc kcal/m	 	 	M Spec.	1
A* 15 to	1,3933	4 5	ΔHf	ŀ		Ultra V. X-Ray Dif.	1
B* 120 °C	1221.2	5	ΔFf	}	-	Infrared	
c		1 1	Viscosity centistokes			Solubility in +	
tk			η ∘c	}	ļ .	Acetone Carbon tet.	
'x						Benzene	ĺ
A' to				L		Ether n-Heptane	
C'			B ^V to			Ethanol	
A'* to			A ^V °C	_		Water Water in	
B'* °C			(B ^V) to				+
Ac 127 to		5 5	(A ^V) °C	L	ļ!		
Cc - c-	258.	5	c _p liq. °K				
Cryos. A			c _p vap. °K				
consts. B°	 	\dashv	c _v vap.				
t _e *C	110.77	5	-v	L	<u> </u>		
$T_R = 0.7$		•				grams/100 grams solve	ent
REFERENC	ES: 1-Dow			Calc. from det	t. da	ta 5-Calc, by formula	
SOURCE:			PI				
PURIFICAT			PI	·			
LITERATU	RE REFER	ENCES	:				

No. 129 2, 3-Dimethyl-cis-3-hexene NAME STRUCTURAL FORMULA CH3CH C = CHCH2CH3 ҁҥӡҁҥ Mole Ref. Molecular Molecular C8H16 . % Pur. Weight 112.208 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 0.7791 5 B. P. °C h BP 0.0465 760 mm 114. 2 t_e 0.0363 5 f١ 55.05 to 100 4 g† •<u>K</u> 30 29.03 4 30 mm 0.6516 5 10 9.17 5 h١ ∆Hm cal/g 5 -24.09 to m ΔHv cal/g 25°C Pressure n °K 83.06 mm 25°C 24, 33 5 o 30 mm 82.76 5 1043. t_e BP 70.94 5 m' to Density 5 69.38 te (d, e) n' °K g/ml 20°C 0,728 2 69.33 ۰, d_4^t 25 0.724 2 AHv/T 5 19.52 30 0.720 4 Surface tension 29 86, 80 5 to 0.744 4 dynes/cm. 20°C 22.25 5 <u>.с</u> å, 135 0.1391 5 h -0.038 4 30 21.28 5 to 84.95 40 20.41 5 Ref. Index 29 0.0755 [P] nD 20°C 1.416 Parachor d_c g/ml 0.247 5 25 2 1.414 20°C vc ml/g 4.05 30 1.402 4 30 t_c 284. 5 40 "C" 0.7602 4 P_c mm 19592. 5 Sugd 335.2 5 MR (Obs.) 38.7 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38,677 25°C 1.0000 5 (nD-d/2) 1.052 2 30 mm 1,0000 5 Dispersion 124. 2 0.9500 Dielectric BP 5 Flash Point C 0.9403 A 29 to 6.95106 1**348**.6 Fire Point 0.256 5 B 145 °C M. Spec. C 5 AHc kcal/m 217.34 Ultra V. **AHf** A* 29 to 1.39312 5 X-Ray Dif. ΔFf B* 140 °C 1266.7 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. t_x °C Benzene 15 to 7.35098 Ether В' 29 °C 1552.9 n-Heptane B^V A 235.34 5 C to Ethanol °C Water A'* 15 to 1.74916 5 Water in B'* 29 °C (B^V) 1453.3 5 to Ac 145 to (AV) 7.3620 5 °C Bc tc °C 1663. 5 c_p liq. ۰ĸ Cc 257. 5 Cryos. A c_p vap. ۰ĸ consts. B° c, vap. te °C 125.59 5 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

								No. 13	0
NAME	2,3-	Dime	thyl	-trans-3-hexen	e	_	STRUCTURAL	FORMULA	4
							CH ₃ CH C = CH	сн,сн,	
Mole	Ref.	Mol	ecul	ar C ₈ H ₁₆	Molecular		¸ċн₃ċн₃	2 3	
% Pur.					Weight 112.2	-			
	T		Ref.	ļ		Ref			Ref
F.P. *C F.P. 100%	<u> </u>			dt/dP *C/mm	1		f to		
B. P. *C	 		\vdash	25°C	0.7791	5	g <u>*K</u>		ì
760 mm	114.		2	BP	0.0465	4	h		┼
100 30	55.05		4	t _e	0.0363	5	f' to		
10	29.03 9.17		4 5	30 mm	0.6516	5	h'		l
1	-24.09		5	ΔHm cal/g	_		m to		
Pressure				ΔHv cal/g 25°C	83.06	5	n K		
mm 25°C	24.33		5	30 mm	82.76	5	•		
Te Te	1043.		5	BP	70.94	5	m' to		
Density g/ml 20°C	0.72	8	2	te (d, e)	69. 38 69. 33	5 5	n' °K_		
at 25	0.72	4	2	ΔHv/T	1 .	5	0'		į
4 30	0.72		4		19.52	5	Surface tension		Г
a b	0.74		4	d 29 to		5	dynes/cm. 20°C	22.25	5
	-0.03	<u>- </u>	4	d' 20 to	84.95	5	30 40	21.28 20.41	5
Ref. Index	1.41	6	2	e' i 29 °C		5	Parachor [P]		+-
25	1.41	4	2	d g/ml	0. 247 4. 05	5 5	20°C		
30	1.40		4	t _c *C	284.	5	30 40		1
"C"	0.76	02	4	P _c mm	19592.	5		335.2	5
MR (Obs.)	38.7	,	2 5	PV/RT	<u> </u>	\vdash	Exp. L.1.%/wt.		
MR (Calc.) (nD-d/2)	38.67 1.05		2	25°C	1.0000	5	u.		_
Dielectric				30 mm BP	1.0000 0.9500	5 5	Dispersion	124.	2
A 29 to	6.95	106	5	t _e	0.9403	5	Flash Point °C Fire Point		
B 1145 °C	1348.6		5	t _c	0,256	5	M Spec.		-
C	217.34	$\overline{}$	5	ΔHc kcal/m ΔHf	1		Ultra V.		
A* 29 to B* 140 °C	1.39 1266.7	312	5	ΔFf			X-Ray Dif. Infrared		
K				Viscosity					╁
t.	ļ			centistokes	ł		Solubility in + Acetone		
t _x to t _x *C	}			η •c			Carbon tet.		1
A' 15 to	7,35	098	5			1 1	Benzene Ether		
B' _ 29 °C	1552.9		5			\vdash	n-Heptane		
C'	235.34	$\overline{}$	5	B ^V to	1		Ethanol Water		1
A'* 15 to B'* 29 °C	1.74 1453.3	916	5		-Í		Water in	İ	
Ac 145 to	7.36	20	5						
Bc t C	1663.		5		ļ	 			
Cc	257.		5	c _p liq. °K					
Cryos, A° consts, B°				c _p vap. °K					
t _e °C	125.59		5	c _w vap.					
$T_{R} = 0.75$							grams/100 gran	ns sol ve n	t
REFERENC	ES: 1-D	ow :	2-AF		Calc. from de	t. dat	ta 5-Calc, by for	nula	
SOURCE:			AP	PI .				-	
PURIFICAT			AP						
LITERATUI	E REFI	EREN	CES	:					

Γ							No. 1	31
NAME _	2,4-Dim	ethyl-	-cis-3-hexene			STRUCTURAL	FORMUI	LA.
						CH3CH CH=C	CH,CH	,
Mole	Ref. Mo	1 1.		(a) l			H ₃	•
% Pur.		lecula mula		Molecular Veight 112.2	08	3	3	
		Ref.		T	Ref.	I	*********	Ref.
F.P. °C			dt/dP			f to	i i	
F.P. 100%			°C/mm			g °K	_	ŀ
B. P. ℃			25°C BP	0.6431	5	h	1	1
760 mm 100	109.0	2 4	te	0.0363	5	f' to		
30	24.96	4	30 mm	0.6441	5	g' ' <u>°K</u>	.	
10 1	5.44 -26.78	5	ΔHm cal/g		Γ	h' i		ļ
Pressure	20.10	-	ΔHv cal/g			m to	1	1
mm 25°C	30.06	5	25°C 30 mm	81.47	5	, , <u>-</u>	-	
t _e	1029.	5	BP	81.47 69.84	5	m¹ to	 	+
Density g/ml 20°C	0 7170	2	t _e	68.38	5	n'ı ek		
dt 25	0.7178 0.7135	2	te (d, e)	68, 32	5	0'	1	
^a 4 30	0.7095	4	ΔHv/T _e	19.51	5	Surface tension	<u> </u>	+-
a b	0.7343	4	d 20 to	84.93 0.1384	5	dynes/cm. 20°C	21.12	5
Ref. Index	-0.0382	4	d' to			8 30 40	20.16	5
n _D 20°C	1.4140	2	e' °C		<u> </u>	Parachor [P]	1	+
25	1.4114	2	d _c g/ml v _c ml/g	0.244 4.109	5	20°C		1
30	1.4088	4	tc°°C °	275.	5	30 40		
MR (Obs.)	0.7675	4	P _c mm	18980.	5		335.2	5
MR (Calc.)	39.06 38.677	2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0551	2	25°C 30 mm	1.0000	5	u. Dispersion	124.	2
Dielectric			BP	0.9500	5	Flash Point °C	121.	+ <u>-</u> -
A 20 to	6.94402	5	te tc	0.9407 0.256	5	Fire Point		
B <u>138 °C</u> C	1329.8 218.29	5 5	ΔHc kcal/m	<u> </u>	+-	M. Spec.		
A* 20 to	1.39145	5	ΔHf			Ultra V. X-Ray Dif.		
B* 135 °C	1248.8	5	ΔFf		ļ	Infrared		
K — — —			Viscosity centistokes			Solubility in +	T	
t _k to		}	η °C			Acetone Carbon tet.		
X	·				ļ	Benzene	1	
A' to B' °C					<u> </u>	Ether n-Heptane		
C'			B ^v		1	Ethanol		
A!* to B!* °C	1					Water Water in		
Acl 138 to	7. 3549	5	v					+
Bc tc °C	1639.	5			<u> </u>	1		
Cc — —	258.	5	c _p liq. °K					1
Cryos. A° consts. B°			c _p vap. °K					-
t _e °C	119.96	5	c _v vap.					
$T_{\mathbf{R}} = 0.75$		ا د		L	<u> </u>	+ ~~~~ - /100	1	
	ES: 1-Dow	2-4	PI 3-Lit. 4-0	Calc from do	+ 4-	grams/100 grants for the grants of the grant		nt
SOURCE:	10. 1-DOW		PI 3-Lit. 4-0	Calc. Ifom de	a	ita 3-Caic, by 10		
PURIFICAT	ION:		PI			<u> </u>		
	RE REFERE							
	NOFERE	W ES	,.					

							No. 13	2
NAME	2, 4-D	imethyl-	trans-3-hexen	•		STRUCTURAL 1	FORMULA	A.
						כח כח כח-כ	CH CH	
						сн ₃ сн сн=с сн ₃ сн	_ ,	
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 112.2	:08	03	3	
		Ref.	<u> </u>		Ref		**************************************	Ref.
F. P. °C			dt/dP			f to		
F.P. 100%			*C/mm 25*C	0,6109	5	g <u>*K</u> _		
B. P. °C 760 mm	107.6	2	BP	0.0459	4	h		↓
100	49.50	4	t _e	0.0363	5	f' to		1
30 10	23.86	4 5	30 mm	0.6419	5	h' '^		
1	-27.70	5	ΔHm cal/g		\vdash	m to		+-
Pressure mm 25°C	31.82	5	ΔHv cal/g 25°C	81.00	5	n ºK		
t _e	1025.	5	30 mm BP	81.16	5 5	0		
Density				69.57 68.13	5	m' to		l
g/ml 20°C	0.714		te (d, e)	68.08	5	n' '		
d 25 4 30	0.705		ΔHv/T _e	19.52	5	Surface tension		+
a	0.732		d 23 to e 130 °C		5 5	dynes/cm. 20°C	20.74	5
b Ref. Index	-0.038	8 4	a' 1 0 0	1	ا ً ا	30 40	19.71 18.71	5
n _D 20°C	1.412			0,240	5	Parachor [P]		T
25 30	1.410		d g/ml vc ml/g tc °C	4.161	5	20°C 30		
"C"	0,768	_		272.	5	40		
MR (Obs.)		2	P _c mm	18641.	5		335.2	5
MR (Calc. (nD-d/2)	38.677		PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.		
Dielectric	1.055	7 + -	30 mm BP	1.0000	5	Dispersion		
A 23 to	6.944	45 5	te	0.9500 0.9409	5	Flash Point °C Fire Point		İ
B (135 °C	1325.4	5	tc	0.256	5	M Spec.		+
C A* 23 to	218.56	5 41 5	ΔHc kcal/m ΔHf			Ultra V.		
B* 130 °C		5	ΔFf	<u> </u>	L	X-Ray Dif. Infrared		
K C			Viscosity centistokes		1	Solubility in +		\top
t _k Tto			7 °C			Acetone Carbon tet.		
t _x ; •0						Benzene		
A' to						Ether n-Heptane		
C'			B ^V to A ^V °C			Ethanol		
A'* to B'* *C				-		Water Water in		
Ac 135 to	+	1 5	(A ^V) to					Π
	1632.	5	c _p liq. °K		\vdash			
Cryos. A°	257.	5	_					
consts. B°			c _p vap. °K					
t _e °C	118.37	5	c _v vap.					
$T_{\mathbf{R}} = 0.7$						+ grams/100 gran	ns solven	t
REFEREN	CES: 1-Do			Calc, from de	t. dat	ta 5-Calc. by for	nula	
SOURCE:		A	PI					
PURIFICA:			PI					
LITERATU	RE REFE	RENCES	3:					

Nt. 122

									No. 1	33
NAME	2,5-	Dime	ethyl	-cis-3-hexene			STI	RUCTURAL	FORMUL	A
							С	H ₃ CH CH=C	нсн сн.	
14-1-	72-6						·	сн ₃	ċн,	3
Mole % Pur.	Ref.	For	mul	C ₈ H ₁₆	Molecular Veight 112.20	08		3	3	
			Ref.		I	Ref.				Ref.
F. P. °C				dt/dP			f T	to	-	
F.P. 100	6			*C/mm 25°C	0.4958	5	gl	<u>*K</u>		1
B. P. °C	102.		2	BP	0.0453	4	h			
100	44.62		4	t _e	0.0363	5	f' g'	to •K		
30 10	19.31		4 5	30 mm	0,6335	5	h'			1
1	-31.56	5	5	ΔHm cal/g ΔHv cal/g	-	-	m	to		1
Pressure mm 25°C	40.18		5	25°C	78.96	5	n i	<u>*K</u>		1
te	1009.		5	.30 mm BP	79.73 68.37	5				ļ
Density				t _e (d, e)	67.03	5	m'	to °K		
g/ml 20°0 dt 25 4 30	0.71		2		66.99	5	۰' '			
	0.70)2	4	d 15 to	19.52	5		ace tension		
a b	0.72		4	<u>• 120 °C</u>	82.38 0.1374	5		s/cm. 20°C 30	20.22 19.31	5
Ref. Inde		,	H	d' to			Ľ	40	18.42	5
ⁿ D 20°0	1.40		2	d _c g/ml	0,244	5	Para	chor [P]		
30	1.40		2	V mi/g	4, 103	5		2 0°C 30		
"C"	0.76		4	t _c °C P _c mm	264. 18625.	5		40 Sugd	335, 2	5
MR (Obs.			2	PV/RT	10025.	 	Exp	L.1.%/wt.	335.2	+-
MR (Calc. (nD-d/2)	38.67		5 2	25°C	0.9989	5	_	u.		
Dielectric	:			30 mm BP	1.0000 0.9500	5		ersion h Point C	119.	2
A 15 to		698	5	te t_	0.9414 0.256	5		Point		
B (130 °C	219.62	:	5 5	ΔHc kcal/m	0.230			Spec.		
A* 15 to		205	5	ΔHf ΔFf			Ultr X-R	a V. ay Dif.		
B*[120 °C	1224.8		5	Viscosity	 		Infra			
c	_			centistokes				bility in †		1
t _k ~°C				η °C			Car	rbon tet.		
A' to							Eth	nzene ner		
B' °	2			B ^V to				leptane anol		
A'* to	, -			AV °C			Wa	ter		
B'* °(3			(B ^V) to			Wa	ter in		+-
Ac 130 to	7.34	183	5 5	(A ^V) °C						
Cc	258.		5	c _p liq. °K						
Cryos. A'				c _p vap. °K						
te °C	112.07		5	c _v vap.						1
$T_R = 0$.				L	L	L	+ gra	ams/100 gra	ms solve:	nt
		Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:			A	PI						
PURIFICA				PI						
LITERATI	JRE REF	EREI	NCES	S:						

							No. 134	4
NAME	2,5-Dime	thyl-	trans-3-hexene			STRUCTURAL F	ORMULA	4
						Сн3Сн Сн=Сн	сн сн.	
Mole	Ref. Mo	10001		Molecular			Ċнз	
% Pur.		lecul rmul		Weight 112.2	208	3	<u> </u>	
		Ref.			Ref.			Ref.
F.P. C			dt/dP			f to		
F.P. 1009	·		*C/mm 25*C	0.4958	5	g <u>*K</u> _		
B. P. °C 760 mm	102.	2	BP	0.0453	4	h		
100 30	44.62 19.31	4	t _e	0.0363	5	f' to g' *K		
10	0.1120	5	30 mm	0.6335	5	h'		
1	-31.56	5	ΔHm cal/g ΔHv cal/g		-	m to		┢
Pressure mm 25°C	40,18	5	25°C	78.96	5	n °K_		
t _e	1009.	5	30 mm BP	79.73 68.37	5	0		_
Density			t	67.03	5	m' to		
g/ml 20°C	0.710 0.706	2 2	t _e (a, e)	66.99	5	n' *K		
d ₄ 25	0.702	4	ΔHv/T _e	19.52	5	Surface tension		╁
	0.726	4	d 15 to e 120 ℃	82.38 0.1374	5	Surface tension dynes/cm. 20°C	20.22	5
<u>b</u>	-0.038	4	_a'	0.1374		30 40	19.31 18.42	5
Ref. Index		2	e' °C			Parachor [P]	10.42	۲
D 25	1.404	2	d g/ml vc ml/g tc °C	0.244 4.103	5	20°C		
"C"	1.402	4	tc °C	264.	5	30 40		
MR (Obs.)	0.7618	2	P _c mm	18625.	5	Sugd.	335.2	5
MR (Calc.		5	PV/RT 25°C	0.0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.051	2	30 mm	0.9989 1.0000	5	u. Dispersion	119.	2
Dielectric			BP	0.9500 0.9414	5	Flash Point °C		\vdash
A 15 to B 1130 °C		5 5	te t _c	0.256	5	Fire Point		<u> </u>
c	219.62	5	AHc kcal/m			M Spec. Ultra V.		1
A* 15 to		5	ΔHf ΔFf			X-Ray Dif.		
B* ₁₂₀ •0	1224.8	٦	Viscosity			Infrared Solubility in +		-
t - to	_		centistokes 7°C			Solubility in + Acetone		
t _k to			η ℃			Carbon tet. Benzene		
A' to						Ether		
B' 'C	-		B ^V to			n-Heptane Ethanol		
A'* to	,		AV C			Water		
B'* °(7		(B ^V) to			Water in		+
Ac 130 to	7.3483 1608.	5	(A ^V) °C					
Cc c_	258.	5	c _p liq. °K					
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	112.07	5	c _v vap.					
$T_R = 0.7$	5 T _c			<u> </u>	 _	+ grams/100 gran	ns solven	t
		2-AI	PI 3-Lit. 4-C	alc, from det	t. da	ta 5-Calc. by forn	nula	
SOURCE:			PI					
PURIFICA'			PI					
LITERATU	RE REFERE	NCES	:					

							No. 13	5
NAME	3, 4-Dime	thyl-	cis-3-hexene			STRUCTURAL FORMULA		
						CH ₃ CH ₂ C = C	CH_CH_	
Mole	Ref. Mo	lecul		Molecular		с́н _з с	н.	
% Pur.		mula		Meight 112.20	8	33-	- 3	
		Ref.			Ref.			Ref.
F, P. ℃			dt/dP			f to		
F.P. 100%			°C/mm 25°C	1.0610	5	g '° <u>K</u>	- 1	
B. P. °C 760 mm	122.	2	BP	0.0473	4	h		
100	62.00	4	t _e	0.0363	5	f' to to <u>"K</u>		1
30 10	35.50 15.27	4 5	30 mm	0.6637	5_	h'		
1	-18.61	5	ΔHm cal/g	 		m to		
Pressure mm 25°C	17.33	5	ΔHv cal/g 25°C	85.63	5	n <u>*K</u>	.] [
t _e	1066.	5	30 mm BP	84.76 72.64	5		L	
Density			te te (d, e)	70.94	5	m' to		
g/m1 20°C	0.747 0.743	2		70.87	5	ö, ¦	1	
d ₄ 25 30	0.739	4	ΔHv/T _e	19.52	5	Surface tension		
a b	0.763	4	d 36 to	89.74 0.1401	5	dynes/cm. 20°C	24.79	5
Ref. Index	-0.038	-	d' 20 to	87.70	5	30 40	23.73	5
n _D 20°C	1.430	2	e' 36 °C	0.0826	5	Parachor [P]		
25 30	1.428 1.425	2	d g/ml vc ml/g	0.252 3.970	5	20°C 30		
"C"	0.7644	4	te °C	299.	5	40		
MR (Obs.)	38.8	2	P _c mm	20507.	5	-	335.2	5
MR (Calc.	38.677	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		l
(nD-d/2) Dielectric	1,057	2	30 mm	1.0000	5	Dispersion	127.	2
A 36 to	6. 95905	5	BP t_	0.9500	5	Flash Point C		
B 156 °C	1377.7	5	te tc	0.256	5	Fire Point M. Spec.		
C	215.82	5	ΔHc kcal/m			Ultra V.		ĺ
A* 36 to B* 145 °C	1.39265 1294.3	5	ΔFf			X-Ray Dif. Infrared		ĺ
K	-		Viscosity			Solubility in +		 -
t _k T to	•		centistokes 7°C			Acetone		
x l						Carbon tet. Benzene		
A' 20 to B' 36 °C	7.35168 1582.1	5				Ether n-Heptane		ĺ
C' '	233.82	5	B ^V to			Ethanol		
A'* 20 to B'* 36 °C	1.74680	5	A - °C -			Water Water in		ĺ
Acl 156 to	7, 3700	5	(B ^V) to					-
Bc tc °C	1700.	5		 	-			
Cc — —	257.	5	P					
Cryos, A° consts. B°			c _p vap. °K					
t _e °C	134.63	5	c _v vap.					
$T_R = 0.7$				1	لــــــــــــــــــــــــــــــــــــــ	grams/100 gra	ms solven	t
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc. by for		
SOURCE:		А	PI					
PURIFICAT			PI					
LITERATU	RE REFERE	NCES	5:					
L								-

							No. 13	6		
NAME	3, 4-Dim	ethyl	-trans-3-hexen	e		STRUCTURAL 1	FORMUL	A.		
						CH3CH2C = C CH2CH3				
Mole	Ref. Mo	lecul		Molecular		ch3ch3				
% Pur.	Fo	rmul		Weight 112.2	08		<i></i>			
		Ref.			Ref			Ref.		
F.P. *C			dt/dP			f to				
F.P. 100%			°C/mm	1 , ,,,,	.5	g		ł		
B. P. *C 760 mm	122.	2	25°C BP	1.0610 0.0473	4	h	L			
100	62.00	4	t _e	0.0363	5	f' to	l			
30	35.50	4	30 mm	0.6637	5	g' K_	1			
10 1	15.27 -18.61	5	ΔHm cal/g			h'	<u> </u>	↓		
Pressure		H	ΔHv cal/g			m to		ì		
mm 25°C	17.33	5	25°C 30 mm	85.63 84.76	5 5	"	1	j		
t _e	1066.	5	BP	72.64	5		 	┼		
Density g/ml 20°C	0.747	2	t _e	70.94	5	m' to		1		
dt 25 d4 30	0.743	2	te (d, e)	70.87	5	o'	1	1		
⁴ 4 30	0.739	4	ΔHv/T _e	19.52	5	Surface tension		+-		
a	0.763	4	d 36 to _e 135 ℃	89.74 0.1401	5 5	dynes/cm. 20°C	24.79	5		
ь	-0.038	4	d' 20 to	87.70	5	30 40	23.73 22.71	5		
Ref. Index	1.430	2	e' j 36 °C	0.0826	5	Parachor [P]	22.11	+-		
25	1.428	2	d g/ml v ml/g	0.252	5	20°C				
30	1.425	4	tc °C	3.970	5	30 40	ł			
"C"	0.7644	4	P _c mm	20507.	5		335.2	5		
MR (Obs.) MR (Calc.)	38.8 38.677	2 5	PV/RT		\vdash	Exp. L.1.%/wt.	l	 		
(nD-d/2)	1.057	2	25°C	1.0000	5	u.	l	1.		
Dielectric			30 mm BP	1.0000	5 5	Dispersion	127.	2		
A 36 to	6.95905	5	t _e	0.9396	5	Flash Point °C Fire Point	}	1		
B (156 °C		5	t _c	0.256	5	M Spec.		+		
	215.82		ΔHc kcal/m ΔHf		1 1	Ultra V.		İ		
A* 36 to B* 145 °C	1.39265	5	ΔFf			X-Ray Dif. Infrared				
к — — —			Viscosity			Solubility in +		+-		
t _k	1	1 1	centistokes 7°C		1 1	Acetone				
tx ¦ °C		1	7			Carbon tet, Benzene				
A' 20 to	7, 351 68	5		1		Ether	1			
B' ∟36 °C	1582.1 233.82	5	B ^V to		\vdash	n-Heptane Ethanol				
A'* 20 to	1.74680	5	A ^V C	[Water		Í		
B'* 36 °C		5	(BV) to	1		Water in		—		
Ac 156 to	7.3700	5	(A ^V) °C				ļ]		
Bc tc °C	1700. 257.	5	cp liq. °K							
Cryos, A°	† 	-	· -	1		1	ł	İ		
consts. B°			р -				Ì			
t _e °C	134.63	5	c _w vap.	1]		}	j		
$T_{R} = 0.79$			·	<u> </u>		+ grams/100 gran	ns solven	ıt		
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	alc. from de	t. dat					
SOURCE:		A	PI							
PURIFICAT	ION:	A	PI							
LITERATU	RE REFERE	CES	:							

NAME	2-n-Prop	yl-1-	pentene		STRUCTURAL FORMULA					
Γ										
Mole % Pur.		ecul:		Molecular Weight 112.2	08	сн ₃ (сн ₂) ₂ с с	:СН ₂ .Н ₇			
		Ref.			Ref.			Re		
F, P. °C			dt/dP			f to				
F.P. 100%			°C/mm		_	g °K				
B. P. °C		١. ا	25°C BP	0.8975 0.0469	5 4	h				
760 mm 100	117.7 58.25	2 4	te	0.0363	5	f' to		I		
30	32.00	4	30 mm	0.6573	5	g' <u>°K</u>		1		
10 1	-11.97 -21.58	5	∆Hm cal/g			h'				
Pressure	-21.30	-	ΔHv cal/g			m to		1		
mm 25°C	20.83	5	25°C 30 mm	84.22	5	<u>"</u>	-			
t _e	1054.0	5	BP	83.66 71.69	5	m' to	 	+-		
Density		١. ا	t _e	7 0 . 07	5	m' to				
g/ml 20°C	0.7240 0.7198	2 2	le (d, e)	70.01	5	0'	1			
dt 25 4 30	0.7156	4	ΔHv/T _e	19.51	5	Surface tension		+-		
a	0.7408	4	d 32 to e 130 °C	88.13	5 5	dynes/cm. 20°C	21.87	5		
<u>ь</u>	-0.03834	4	d 20 to	86.20	5	30 40	20.86	5 5		
Ref. Index	1.4136	2	e' 32 °C		5		19.89	+*		
ⁿ D 20°C	1.4111	2	d _c g/ml	0. 243	5	Parachor [P] 20°C				
30	1.4085	4	v _c ml/g t _c °C	4.123 288.	5	30				
"C"	0.7603	4	P _c mm	19365.	5	40 Sugal	335.2	5		
MR (Obs.)	38.69	2	PV/RT	17,303.	\dashv	Exp. L.1.%/wt.	333.2	+-		
MR (Calc.) (nD-d/2)	38.677 1.0516	5 2	25°C	1.0000	5	u.				
Dielectric	 		30 mm BP	1.000 0 0.9500	5 5	Dispersion	123.	2		
A 32 to	6,9535	5		0.9400	5	Flash Point C Fire Point		1		
B 148 °C	1361.6	5	te t _C	0.256	5		ļ	+		
С	216.64	5	ΔHc kcal/m ΔHf			M. Spec. Ultra V.		Ì		
A* 32 to	1.3916	5	ΔFf			X-Ray Dif.	1			
B*[13 <u>8 °C</u> K	1279.0	. 5	Viscosity		t	Infrared		+		
د حــــ	.		centistokes			Solubility in T				
tk to			ŋ •°C			Carbon tet.				
A' 20 to	7.350	5	Ì			Benzene Ether				
B' _ 32 °C	1566.0	5	B ^v to	 	\vdash	n-Heptane				
C'	234.64	5	B ^V to C			Ethanol		ı		
A'* 20 to B'* 32 °C	1.7467 1466.27	5 5	⊢.=v. 	-		Water Water in				
Ac 148 to	7, 3638	5	. v.							
Bc tc °C	1677.4	5		+	+-					
Cc	256.71	5	c _p liq. °K							
Cryos. A° consts. B°			c _p vap. °K							
t _e °C	129.78	5	c _w vap.							
$T_R = 0.7$	5 T _c					grams/100 gra	ms solve	nt		
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc, by for	rmula			
SOURCE:		A	PI							
PURIFICAT	ION:	A	PI							
LITERATU	RE REFERE	NCES	3:							

										No. 1	38	
NAME		2-Isc	prop	yl-l	-pentene			STRU	CTURAL 1	FORMUL	A	
Mole % Pur.	Ref. Molecul			lecul	ular C ₈ H ₁₆ Molecular weight 112.208			сн ₃ (сн ₂) ₂ с=сн ₂ сн(сн ₃) ₂				
/ Pur.			FU	Ref.		weight 112.	Ref				Ref	
F. P. *C	T				dt/dP	T	1	f	T	i	+	
F.P. 100%	1				°C/mm		1. 1	g	to K			
B. P. °C					25°C BP	0.7497 0.0464	5 4	h ¦			1	
760 mm 100		l 3. 54. 18		2 4	t	0.0363	5	f'	to			
30		28.22		4	30 mm	0.6501	5	g'	_ K			
10 1	_2	8.40 24.78		5 5	ΔHm cal/g		$I \cap I$	h'			 	
Pressure	+-				ΔHv cal/g			m	to oK		1	
mm 25°C		25.38		5	25°C 30 mm	82.95 82.50	5 5	0			-	
t _e	104	10.		5	BP	70.72	5	m' i	to		+	
Density g/ml 20°C	:	0.72	5	2	te te (d, e)	69.18 69.13	5 5	n'	<u>•</u> K_		1	
dt 25 4 30	ļ	0.72	1	2	ΔHv/T _e	19.52	5	0'				
4 JU	+	0.71		4	d 25 t		5		ce tension	31 00	\ _	
b	.	0.74 0.03-		4 4	e 124 °	C 0.1390	5	dynes	30°C 30°C	21.99 21.02	5	
Ref. Index					d' t				40	20.08	5	
ⁿ D 20°C	-	1.41		2 2	d g/ml	0.246	5	Para			İ	
30		1.40		4	d g/ml v ml/g t °C	4.061 282.	5 5		20°C		1	
"C"		0.75	99	4	, `	19451.	5		40 Su a d	225 2	5	
MR (Obs.)		38.7		2	P _c mm	19451.	┝╩┤	F-n	L. 1. %/wt.	335.2	+-	
MR (Calc. (nD-d/2)	"	38.67 1.05		5 2	25°C	1.0000	5	_	u.			
Dielectric	+-	1.03		\vdash	30 mm BP	1.0000 0.9500	5 5		rsion	122.	2	
A 25 to	.+	6. 94	966	5	te	0.9404	5	Flash Fire	Point °C		1	
B 1144 °C	134	14.9		5	t _c	0.256	5	M Spe			+	
C	+	7.53		5	ΔHc kcal/m ΔHf			Ultra	v.		ł	
A* 25 to B* 134 °C		1.39 53.1	219	5 5	ΔFf	_		X-Ra				
к — — -	1				Viscosity		1		ility in +		+-	
t _k -t ₀					centistokes	;		Acet	one			
t _x *C	;				•		1 1	Benz	oon tet. zene			
A' to						1		Ethe				
č, – – –	-				B ^V to			Etha	eptane nol		-	
A'* to					AV C	<u>니</u>		Wate				
B'* °(+			<u> </u>	(B ^V) to	I	1 1		,, ,,,		+	
Ac 144 to	165	7.36 58.	06	5 5	(A ^V) •c		1					
Cc	25	57.		5	c _p liq. ∘k		1 1	l				
Cryos, A° consts, B°					c _p vap. *k							
t _e °C		24.47		5	c _w vap.							
$T_R = 0.7$								+ gran	ns/100 gran	ns solver	at	
REFEREN	ES:	1-D	ow			Calc. from de	t. dat	a 5-C	alc. by form	nula		
SOURCE:				AP								
PURIFICA:				AP					-			
LITERATU	RE .	KEF I	ek ei	NCES	:							

							No. 13	39		
NAME _	2-Ethyl-	3-me	thyl-l-pentene		STRUCTURAL FORMULA					
						сн,сн,сн с	= CH,			
Mole	Ref. Mo	lecul		Molecular		с́н ₃ с				
% Pur.	For	mula		Weight 112.20	08	3 2 5				
		Ref.			Ref.		,	Ref.		
F.P. °C F.P. 100%	-	_	dt/dP °C/mm			f to				
B. P. °C		-	25°C	0.7348	5	g '° <u>K</u> h				
760 mm	112.5	2	BP t _e	0.0464	4 5	f' to		\vdash		
100 30	53.73 27.79	4	30 mm	0.6495	5	g' '° <u>K</u>				
10 1	7.99 -25.16	5	ΔHm cal/g			h'		<u> </u>		
Pressure	23.10	-	ΔHv cal/g			m to				
mm 25°C	25.96	5 5	25°C 30 mm	82.73 82.35	5					
Density	1039.)	BP	70.59 69.06	5	m¹ to				
g/ml 20°C	0.729	2	t _e (d, e)	69.00	5	n' <u>•K</u>	ļ			
d ₄ 30	0.725 0.721	2 4	AHv/T _e	19.51	5		ļ	 		
a	0.745	4	d 25 to e 124 °C	86, 20	5 5	Surface tension dynes/cm. 20°C	22.48	5		
ь	-0.038	4	-d1 10	0.1388	"	30 40	21.50 20.54	5		
Ref. Index	1.4142	2	e' °C		_	Parachor [P]	20.51	+-		
25 30	1.4118	2	d _c g/ml v _c ml/g	0.248 4.038	5	20°C				
"C"	1.4094 0.7561	4	v _c m1/g t _c °C	282.	5	30 40				
MR (Obs.)	38.5	2	P _c mm	19563.	5	Sugd.	335, 2	5		
MR (Calc.) (nD-d/2)	38.677	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.				
Dielectric	1.050	2	30 mm BP	1.0000	5	Dispersion	122.	2		
A 25 to	6.94746	5	t.	0.9500 0.9404	5	Flash Point C Fire Point				
B (143 ℃ C	1342.5 217.62	5 5	tc ΔHc kcal/m	0.256	5	M. Spec.		╁		
A* 25 to	1.39112	5	∆Hf			Ultra V. X-Ray Dif.				
B* 134 °C	1260.8	5	ΔFf		<u> </u>	Infrared				
K — — —			Viscosity centistokes			Solubility in +				
t _k to			η ℃			Acetone Carbon tet.	ŀ			
A'I to		_				Benzene Ether				
B'°C			BV I		-	n-Heptane				
A¹* to		-	B ^V			Ethanol Water				
B'* °C			(B ^V) to			Water in		 		
Ac 143 to Bc to °C	7.3586	5	(A ^V) °C							
Bc t _c °C	1656. 258.	5	c _p liq. °K							
Cryos. A° consts. B°			c _p vap. °K							
te °C	123.91	5	c _w vap.							
$T_{\mathbf{R}} = 0.75$						grams/100 gra	ms solver	nt		
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for				
SOURCE:			PI							
PURIFICAT			PI							
LITERATUR	RE REFERE	NCES	5:							

							No. 140
NAME	2-E	thyl-4-m	ethyl-l-pentenc	<u> </u>		STRUCTURAL FO	ORMULA
						CH CH CH C	- CU
						CH ₃ CH CH ₂ C	
Mole % Pur.	Ref.	Molecul	ar C ₈ H ₁₆	Molecular Weight 112.2	08	CH ₃ C ₂	~ 5
<u> </u>		Ref.		weight 112.2	Ref		Ref
F.P. °C	T	1.61.	dt/dP	T	1		
F.P. 100%	 		*C/mm			f to g *K	-
B. P. *C			25°C	0.6758	5	h	l
760 mm 100	110.3	2 4	BP t _e	0.0462 0.0363	5	f' to	
30	26.02		30 mm	0.6461	5	g' 'K_	ļ
10 1	6.32		ΔHm cal/g			h'	
	-26.66		ΔHv cal/g	 		m to	
Pressure mm 25°C	28,46	. 5	25°C	81.95	5	n <u>*K</u>	
. t _e	1033.	5	30 mm BP	81.80 70.12	5		
Density			11 +	68,63	5	m' to	
g/ml 20°C	0.71		t_ (a,e)	68.58	5		ŀ
d ₄ 25 30	0.71		ΔHv/T _e	19.51	5	Surface tension	
	0.73		d 20 to		5	dynes/cm. 20°C	21.33 5
Ъ	-0.03	85 4		7		30	20.31 5
Ref. Index	1.41	05 2	e' 'c		\sqcup	Parachor [P]	19.33 5
25	1.40		d _c g/ml	0.242 4.128	5	20°C	l
30	1.40	53 4	d g/ml vc ml/g tc °C	277.	5	30	
"C"	0.75		P _c mm	18963.	5	40 Sugd. 3	35.2 5
MR (Obs.) MR (Calc.)	38.68 38.67		PV/RT	 	╁	Exp. L.1.%/wt.	
(nD-d/2)	1.05		25°C 30 mm	1.0000	5	u.	.
Dielectric			BP BP	1.0000 0.9500	5		22. 2
A 20 to			t _e	0.9406	5	Flash Point °C Fire Point	i
B (139 °C	1334.6	5 5	t _c	0, 256	-	M Spec.	
A* 20 to			ΔHc kcal/m ΔHf			Ultra V.	1
B* 130 °C	1.39 1253.3	159 5	ΔFf	L		X-Ray Dif. Infrared	
K – – –	1	- 1	Viscosity			Solubility in +	
1k 10	1		centistokes 7 °C	1		Acetone	ì
t⊊i •c	İ		•	1		Carbon tet. Bensene	
A' to B' C				1		Ether	
B' •C	-1		B ^V to			n-Heptane Ethanol	
A'* to			A ^V C	_[Water	
B'* *C			(BV) to	1		Water in	
Ac 139 to	7.35	63 5	(A ^V) °C	1			
Bc tc C	257.	5	c _p liq. •K				1
Cryos. A			c _p vap. °K	1			
te °C	121,42	5	c _w vap.				
$T_{\mathbf{R}} = 0.7$			1	<u> </u>	<u> </u>	+ grams/100 gram	
REFERENC		ow 2-AI	PI 3-Lit. 4-0	Calc from de	t da	ta 5-Calc. by form	
SOURCE:		AF		40			
PURIFICAT	ION:	AF					
LITERATU							

No. 141 3-Ethyl-2-methyl-1-pentene NAME STRUCTURAL FORMULA CH_3CH_2CH $C = CH_2$ Ċ₂H₅ĊH₃ Mole Ref. Molecular Molecular C8H16 % Pur. Formula Weight 112, 208 Ref. F.P. °C F.P. 100% dt/dP f to °C/mm <u>°K</u> g 25°C 0.6682 5 B. P. °C h BP 0.0461 760 mm 110. ^te 0.0363 5 ſ١ 51.56 to 100 4 g' <u>°к</u> 4 30 25.78 30 mm 0.6456 5 10 6.10 5 h' ∆Hm cal/g -26,86 m to AHv cal/g Pressure n ۰ĸ 25°C 81.84 mm 25°C 28.82 0 30 mm 81.73 te 1032. 5 ΒP 70.07 5 m' to Density te (d, e) 68.59 5 n' °K g/ml 20°C 0.730 0.726 2 68.54 5 ۰, 25 ž $\mathbf{d_{4}^{t}}$ ΔHv/Te 5 19.52 30 0.722 4 Surface tension 20 to 85, 30 5 0.746 4 dynes/cm. 20°C 22,60 ٠Ć 0.1385 121 Ъ -0.038 4 30 5 21.61 ďΠ to 40 20.65 Ref. Index 5 e' °C 20°C 1.415 2 Parachor [P] ^{n}D d_c g/ml 0.249 25 1.413 2 20°C vc ml/g t °C 4.020 30 1,410 4 30 t_c 279. 5 "C" 40 0.7564 4 P_c mm 19545. 5 Sugd. 335.2 5 MR (Obs.) 38.5 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 25°C 1.0000 5 (nD-d/2) 1.050 2 30 mm 1.0000 2 Dispersion 122. Dielectric ВP 5 0.9500 Flash Point C 0.9407 5 A 20 to 6.94544 5 Fire Point 0.256 В _141 °C 1333.6 M. Spec. С 218. 5 ΔHc kcal/m Ultra V. ΔHf A* 20 to 1.39179 5 X-Ray Dif. ΔFf B*[131°C 1252.4 Infrared ĸ Viscosity Solubility in c centistokes Acetone to Carbon tet. t_x °C Benzene A'I to Ether В' °C n-Heptane B^V | C to Ethanol A'* °C Water to (B^V) Water in B'* °C to Ac | 141 to 7.3568 5 (AV) °C Bc tc °C 1645. c_p liq. °K 258. 5 Cryos, A° consts, B° cp vap. °K c_w vap. te °C 121.08 $T_{\mathbf{R}} = 0.75 \, \mathbf{T_{\mathbf{c}}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: **PURIFICATION:** LITERATURE REFERENCES:

NAME	3-E	thy1-3-	-me	thyl-1-pentene			STRUCTURAL	No. 14 FORMUL	
						CH ₃			
	7.6				Malala	\neg	CH ₃ CH ₂ C	CH=CH ₂	
Mole % Pur.	Ref.		mula		Molecular Weight 112.2	208	с ₂ н	5	
			Ref.			Ref	T		Ref
F.P. °C	T			1. (15	T	1		1	-
F.P. 100%	+		\dashv	dt/dP *C/mm			f to		
B. P. *C	 	+		25°C	0.7214	5	h .		İ
76 0 mm	112.	- 1	2	BP	0. 0463 0. 0363	5	f' to		\vdash
100 30	53.31		4 4	t _e 30 mm	1	5	g' K		
10	7.64		5		0.6486	-	h'	1	1
1	-25.47		5	ΔHm cal/g		\vdash	m to		+
Pressure				ΔHv cal/g 25°C	82.45	5	n '°K		1
mm 25°C	26.48 1038.	3	5	30 mm	82.25	5	•]	
Te te	1038.		-	BP	70.50	5	m' to		t^-
Density g/ml 20°C	0.73	305	2	te te (d, e).	68.99	5 5	n' K_	1	
at 25	0.72	264	2	ΔHv/T _e	19.52	5	o' '		Ì
⁴ 4 30	0.72	223	4				Surface tension		T
	0.74		4	d 20 to		5 5	dynes/cm. 20°C	22.66	5
ь	-0.0	381	4	a' - to	1	`	30 40	21.65 20.67	5
Ref. Index		.	2	e' i °C	<u> </u>	L_	Parachor [P]	20.01	+-
ⁿ D 20°C	1.4		2	d _c g/ml	0.248	5	20°C		
30	1.41	13	4	v _c ml/g t _c °C	4. 039 282.	5	30		1
"C"	0.76	511	4	P _c mm	19557.	5	40 Sugd	335.2	5
MR (Obs.)			2	PV/RT	17,331.	<u> </u>	Exp. L. 1, %/wt.	333.2	1
MR (Calc. (nD-d/2)			5	25°C	1.0000	5	u.		
	1.05	23	2	30 mm	1.0000	5	Dispersion	117.	2
Dielectric	1			BP t _e	0.9500	5	Flash Point °C		
A 20 to		1826	5	t c	0. 256	5	Fire Point		<u> </u>
č	217.72	2	5	ΔHc kcal/m			M Spec.	1	
A* 20 to	1.39	246	5	ΔHf		1 1	Ultra V. X-Ray Dif.		
B* 133 °C		- 1	5	ΔFf		├ ─┤	Infrared		
K	1	1		Viscosity centistokes			Solubility in +		
ել		İ		η °c	}		Acetone Carbon tet.		
t _x l	<u> </u>				İ		Benzene		l
A' to B' c							Ether		
c,	4	-		B ^v to			n-Heptane Ethanol		ļ
A'* to	,		\neg	AY C			Water		1
B'* °C		ľ		(B ^V) to	1		Water in		
Ac 143 to		94	5	(A ^V) °C					1
Bc tc_°C	1 - 0	1	5	c _p liq. °K	+			1	1
	258.							1	
Cryos. A° consts. B°				c _p vap. °K					
t _e °C	123.34	1	5	c _v vap.	L		<u> </u>		<u> </u>
$T_R = 0.7$							grams/100 gra		ıt
REFEREN	.೯೦: 1-L	70W 2			Jaic, from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			AI						
PURIFICAT			AI						
LITERATU	RE REF	EREN	CES	:					

NAME	3-Ethyl-	4-met	thyl-1-pentene			STRUCTURAL	No. 14	
NAME	•							
	T					сн ₃ сн сн сн ₃ с ₂ н		
Mole % Pur.		lecula rmula		Molecular Weight 112.2	208	01130211	,	_
		Ref.			Ref.			Ref.
F.P. °C F.P. 100%		1-1	dt/dP			f to		
B. P. °C	<u> </u>	\vdash	*C/mm 25*C	0.6083	5	g '° <u>K</u>		
760 mm	107.5	2	BP t _e	0.0459 0.0363	5	f' to		\dagger
100 30	49.40 23.77	4 4	30 mm	0.6418	5	g' <u>*K</u>		
10 1	4.31 -27.79	5 5	ΔHm cal/g		† –	h'		-
Pressure		+	ΔHv cal/g		T	m to		
mm 25°C	31.97 1025.	5 5	25°C 30 mm	80.95 81.12	5	0		
t _e Density	1025.		BP	69.54 68.10	5 5	m¹ to		1
g/ml 20°C		2	t _e (d, e)	68.06	5	n'°K_		
d ^t 25 4 30	0.7158 0.7116	2 4	ΔHv/T _e	19.52	5	<u> </u>		+
a	0.7367	4	d 20 to e 130 °C	84.40 0.1382	5	Surface tension dynes/cm. 20°C	21.38	5
b Def Index	-0.0383	4	d' to	0.1302		30 40	20.40 19.43	5
Ref. Index	1.4097	2	e' °C	0, 245	5	Parachor [P]	17.13	+ -
25 30	1.4072 1.4047	2 4	d g/ml vc ml/g tc°C	4.088	5	20°C 30		
"C"	0.7577	4	1 -	273.	5	40		1_
MR (Obs.)	38.59	2	P _c mm	19008.	5	Sugd.	335.2	5
MR (Calc.) (nD-d/2)	38.677 1.0497	5 2	25°C	1.0000	5	Exp. L.1.%/wt.		
Dielectric	1.01/1	╁	30 mm BP	1.0000 0.9500	5	Dispersion	117.	2
A 20to	6.94340	5	t _e	0.9409 0.256	5	Flash Point C Fire Point		
B _137°C C	1324.7 218.57	5	tc ΔHc kcal/m	0.250	-	M. Spec.		
A* 20to	1.39246	5	ΔHf ΔFf			Ultra V. X-Ray Dif.		
B*[_1 <u>37°C</u> K	1243.9	5	Viscosity		-	Infrared		_
С			centistokes			Solubility in *Acetone		
t _k to t _x °C			η °C	}		Carbon tet.		
A' to				1		Benzene Ether		
B' ° <u>C</u>			B ^v to A ^v °C			n-Heptane Ethanol		
A!* to				.]		Water		
B'* °C	7	+	(B ^V) to	-		Water in		-
Ac 137 to Bc t _c °C	7.3545 1633.	5	(A ^V) °C	-	-	1		
Cc — —	258.	5	P -					
Cryos, A° consts, B°			c _p vap. °K					
t _e °C	118.26	5	c _v vap.					
$T_R = 0.7$						grams/100 gra	ms solve	nt
	ES: 1-Dow			Calc, from de	et. da	ata 5-Calc. by for	mula	
SOURCE:	TON.		PI PI					
	RE REFERE							
	NOFERE	E	·.					

No. 144 NAME 2, 3, 3-Trimethyl-1-pentene STRUCTURAL FORMULA CH₃ сн3сн2с C = CH, Mole Ref. Molecular Molecular CaH16 ĊH₂ĊH₂ % Pur Formula Weight 112, 208 Ref Ref Ref F.P. °C F.P. 100% -69 2 dt/dP to *C/mm 25*C g 0.6276 ٠5 B. P. *C h BP 0.0459 4 760 mm 108.31 2 0.0363 5 f' to 100 50.12 4 g* ۰ĸ 30 5 30 mm 0.6429 24.45 4 10 4.96 5 h' ∆Hm cal/g 1 5 -27, 20 to AHv cal/g m Pressure °K 25°C n 81.27 mm 25°C 30.87 5 0 30 mm 81.35 5 t_e 1027. 5 ВP 69.75 5 Density g/ml 20°C m to te (d, e) 68.30 5 •ĸ 0.7352 2 68.25 o١ 0.7308 2 dt AHV/Te 5 19.53 30 0,7263 4 Surface tension 1 20 84.73 5 0.7529 4 dynes/cm. 20°C 23.25 125 •c 0.1383 5 Ъ -0.0388 4 30 22.14 5 ď١ to 40 21.06 5 Ref. Index e' •c 20°C 1.4174 [P]2 n_D Parachor d_c g/ml 0.25 5 25 20°C 1.4151 2 ml/g 4.0 5 30 1.4128 4 c 30 •c 5 ŧč 276. 40 "C" 0.7552 4 Pc 19451. 5 5 mm Sugd. 335.2 MR (Obs.) 38.41 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 38.677 25°C 1.0000 5 (nD-d/2)1.0498 30 mm 1.0000 Dispersion 122. 2 Dielectric RP 0.9500 Flash Point °C 0.9408 A 20 to 6.94586 5 Fire Point tç 0.256 1132 °C 1328.2 M Spec. c 218.42 AHc kcal/m Ultra V ΔHf A* | 20 to 1.39405 5 X-Ray Dif. ΔFf B* 130 °C 1247.2 Infrared ĸ Viscosity Viscom, centistokes °C Solubility in Acetone to °C Carbon tet. Benzene A' to Ether В' •c n-Heptane Ċ Ethanol to °C A'* Water to •C B'* (BV) Water in to Ac | 139 to 7.3573 (AV) °C Bc _tc_ °C 1638. cp liq. °K Cc 5 258. Cryos. Aº ۰ĸ c_p vap. consts. B° t_e °C vap. 119.17 5 $T_{\mathbf{R}} = 0.75 T_{\mathbf{c}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 145 2, 3, 4-Trimethyl-1-pentene NAME STRUCTURAL FORMULA CH3CH CH C = CH, Molecular C8H16 снзснзснз Mole Ref. Molecular Weight 112, 208 % Pur Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g <u>°K</u> 0.6194 5 B. P. °C h BP 0.0459 4 108. 760 mm 2 t_e 0.0363 5 ſ١ 49.82 4 to 100 g¹ °K 24.15 4 30 30 mm 0.6427 5 5 10 4.67 h' ∆Hm cal/g -27.47 5 m to AHv cal/g Pressure n ۰ĸ 25°C 81.10 5 mm 25°C 31.35 0 30 mm 81.22 5 te 1026. 5 ВP 69.63 5 mĨ to Density 5 te (d, e) 68.19 n' °K g/ml 20°C 0.729 0.725 2 68.14 5 ۰, 25 30 2 d_4^t ΔHv/T_e 5 19.52 0.721 4 Surface tension 20 to 84, 55 5 0.745 4 8 dynes/cm. 20°C 22,47 5 125 °C 0.1381 5 21.50 Ъ -0.038 4 5 30 ăח to 40 20.54 5 Ref. Index e' °C ⁿD 20°C 1.415 2 [P] Parachor d_c g/ml 0.249 25 1.413 2 20°C vc ml/g t °C 4.015 5 30 1.411 4 30 t_c 276. 5 40 "C" 0.7575 4 P_c mm 19460. 5 Sugd. 335.2 5 MR (Obs.) MR (Calc.) 38.6 2 PV/RT Exp. L. l. %/wt. 38.677 25°C 1,0000 5 (nD-d/2)1.051 2 30 mm Dispersion 2 1,0000 122 Dielectric ВP 0.9500 5 Flash Point C 0.9408 5 A 20 to 6.9426 Fire Point 0.256 5 B (139 °C 1326.1 5 M. Spec. C 218.48 5 AHc kcal/m Ultra V. ΔHf A* 20 to 1.39111 5 X-Ray Dif. ΔFf B*[129 °C 1245.2 Infrared Viscosity Solubility in centistokes Acetone to Carbon tet. t_x °C Benzene A' to Ether B <u>°С</u> n-Heptane B^V | C to Ethanol °C Water A'* to Water in B'* °C (B^V) to Ac | 139 to 7.3541 5 (A^V) °C Bc te °C 1636. cp liq. °К 5 Cryos, A° consts, B° cp vap. ۰ĸ c_v vap. te °C 118.83 5 $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

								No. 14	6
NAME	2,4	, 4-T	rimet	hyl-l-pentene			STRUCTURAL	FORMUL	Α.
						\neg	CH ₃		
		T -					сн ₃ с сн ₂ с	- CH	
Mole	Ref	. Mo	lecul	ar C ₈ H ₁₆	Molecular		- 1		
% Pur.		Fo	rmul	a 08116	Weight 112.20	08	CH ₃ CH	·3	
			Ref.			Ref.			Ref.
F.P. °C F.P. 1007	-93.4	80	2	dt/dP			f to		
	<u>'</u>		-	*C/mm 25°C	0,4858	· ₅	g <u>*K</u> _	i	
B. P. *C 760 mm	101.4	4	2	BP	0.0452	4	h + -		
100	44.1	4	4	t _e	0.0363	5	f' to		
30 10	18.8		4 5	30 mm	0.6326	5	h' '		
1	-31.9		5	AHm cal/g			 		+-
Pressure				ΔHv cal/g 25°C	70 75	5	m to		-
mm 25°C	41.1 1007.	0	5	30 mm	78.75 79.60	5		1	1
Denelte	1007.		1 3	BP	68.26	5	m' to		1
Density g/ml 20°C	0.7	150	2	te te (d, e)	66. 93 66. 89	5	n' °K	1	
_a t 25	0.7	108	2	ΔHv/T _e	19.52	5	o'		
		065	4	d 18 to		5	Surface tension		
a b	-0.0	319	4	e 120 °C		5	dynes/cm. 20°C	20.79 19.81	5
Ref. Index		301	+ -	d' to			40	18.85	5
n _D 20°C		086	2			5	Parachor [P]		\top
25 30		060	2	d g/ml v ml/g	0.244 4.10	5	20°C		
"C"		034	4	tc °C	264.	5	30 40		
		611		P _c mm	18658.	5		335.2	5
MR (Obs.) MR (Calc.			2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)		511	2	25°C 30 mm	0.9984 1.0000	5	u. Dispersion	122.	2
Dielectric				BP	0.9500	5	Flash Point °C	122.	+-
A 15 to		3714	5	te .	0.9414	5	Fire Point		1
B 1129 °C	219.7		5	t _c	0.230	-	M Spec.		1
A* 15 to		9284	5	ΔHc Kca1/m		1	Ultra V.		
B* 120 °C			5	ΔFf		<u></u>	X-Ray Dif. Infrared		
к ——	-			Viscosity			Solubility in +	-	†
t _k	-			centistokes 7 °C		'	Acetone		
t _x i •c	7			'		1	Carbon tet. Benzene		
A' to							Ether		
B', L _ º	4			B ^V to		<u> </u>	n-Heptane Ethanol		
A'* to	;			A ^V C			Water		
B'* °((B ^V) to	7)		Water in	<u> </u>	—
Ac 129 to		486	5	(A ^V) •C	1			1	
Bc tc_°C	258.		5	c _p liq. °K					
Cryos, A°			+-						
consts. B°			L	c _p vap. °K	1	:			
t _e °C	111.4	3	5	c _w vap.					
$T_R = 0.7$	5 T _c		1		.1		grams/100 gra	ms solver	nt.
REFEREN		Dow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da			
SOURCE:				PI					
PURIFICA'	TION:			PI					
LITERATU	RE REI	ERE							
				-					

NAME	3, 3, 4-Tı	imet	hyl-l-pentene		STRUCTURAL	FORMUL		
Mole % Pur.		lecul		Molecular Weight 112,20	08	сн ₃ сн ₃ сн с сн ₃ сн ₃	CH=CH ₂	
		Ref.		T	Ref.			Ref.
F.P. °C	1	1			1			1
F.P. 100%	. †	┼──	dt/dP °C/mm		1	f to		1
B.P. °C		 	25°C	0.5543	5			
760 mm	105.	2	BP	0.0456	4	h		┼─
100	47.24	4	t _e	0.0363	5	f' to		İ
30 10	21.76	4	30 mm	0,6380	_5_	g' <u>*K</u>		1
1	2.42 -29.49	5	ΔHm cal/g		<u> </u>	h'		∔
Pressure			ΔHv cal/g	1	1	m to		1
mm 25°C	35.46	5	25°C 30 mm	80.06	5	ö 		
t _e	1018.	5	BP BP	80.50 69.03	5	<u> </u>		↓ —
Density			t_	67.64	5	m' to		
g/ml 20°C		2 2	t _e (d, e)	67.59	5			
dt 25 4 30	0.725 0.721	4	ΔHv/T _e	19.53	5			+
a	0,745	4	d 15 to	83.50	5	Surface tension dynes/cm. 20°C	22.47	5
ь	-0.038	4	_e125 °C	0.1378	5	8 30	21.50	5
Ref. Index	:		e' °C			40	20,55	5
n _D 20°C		2	d _c g/ml	0, 25	5	Parachor [P]		
25 30	1.4120 1.4096	2 4	v _c ml/g t _c °C	3.99	5	20°C		ļ
"C"			t _c °C	271.	5	40		
	0.7564	4	P _c mm	19425.	5	Sugd.	335,2	5
MR (Obs.) MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0499	2	25°C 30 mm	1.0000	5	u.	117	
Dielectric			BP	1.0000 0.9500	5	Dispersion	117.	2
A 15 to	6.94133	5	te	0.9411	5	Flash Point C Fire Point		
B 135 °C	1315.8	5	t _c	0.256	5	M. Spec.		+
С	219.	5	ΔHc kcal/m ΔHf	1		Ultra V.		
A* 15 to B* 120 °C		5	ΔFf			X-Ray Dif.		
к 120 С	- 1235.5	,	Viscosity			Infrared		+-
c	_		centistokes		1	Solubility in TACetone		
t _k to		1	η °C			Carbon tet.		
t _x °C						Benzene		
B' °C						Ether n-Heptane		
C' '	-	ŀ	B ^V to C			Ethanol		
A¹* to						Water		
B'* °C			(B ^V) to			Water in		+
Acl 135 to	7.3531	5	(A ^V) °C]		
Bc tc °C	1624. - 258.	5	c _p liq. °K					
Cryos. A°		۲	{	1				1
consts. B°			P					1
t _e °C	115.44	5	c _v vap.	1	1			
$T_{\mathbf{R}} = 0.7$	'5 T _c					grams/100 gran	ms solver	nt
_	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da			
SOURCE:			PI					
PURIFICA	TION:		.PI					
	RE REFERE						*	

						1	lo. 148
NAME	3, 4, 4	-Trime	thyl-1-pentene			STRUCTURAL FO	RMULA
Γ						CH ₃	
					\neg	сн ₃ с сн сн=с	CH ₂
Mole % Pur.	Ref. N	Aoleculi Formuli		Molecular Weight 112.2	08	് ċн₃ċн₃	_
70 Pur.		Ref		weight 110.2	Ref		Ref
F, P. *C	Τ	- Kei	14/17	Τ	I Ver	· · · · · · · · · · · · · · · · · · ·	
F.P. 100%	 	+	dt/dP *C/mm			f to g *K	
B. P. °C	†		25°C BP	0.534 0.0455	5 4	h	ĺ
760 mm 100	104. 46.36	2 4	t.	0.0363	5	f ¹ to	
30	20.94	4	30 mm	0.6365	5	g' <u>•</u> K	ļ
10 1	1.65 -30.18	5	ΔHm cal/g			h'	
Pressure	1	╅┪	ΔHv cal/g			m to	
mm 25°C	36.97	5	25°C 30 mm	79.69 80.24	5 5	 	
t _e	1015.	5	BP	68.81	5	m' to	
Density g/ml 20°C	0,719	2	te te (d, e)	67.43 67.39	5	n' •K_	
at 25	0.715	2	ΔHv/T	19.53	5	0'	
	0,711	4	d 15 to		5	Surface tension	
a b	0.735 -0.038	4 4	<u>e 120 °C</u>	0.1377	5		1.26 5 0.31 5
Ref. Index			d' to		1 1		9.38 5
ⁿ D 20°C	1.412 1.410	2 2	d_g/ml	0,247	5	Parachor [P]	
30	1.408	4	d g/ml v ml/g t °C	4.055	5	20°C	
"C"	0.7628	4	1 _c -C	268.	5	40	
MR (Obs.)	38.9	2	P _c mm PV/RT	18987.	-	Sugd. 33 Exp. L.1.%/wt.	5.2 5
MR (Calc.) (nD-d/2)	38.677 1.053	5 2	25°C	1.1161	5	u.	
Dielectric	1	+-	30 mm BP	1.0000 0.9500	5 5	Dispersion 11	7. 2
A 15 to	6,9398	9 5	te	0.9412	5	Flash Point °C Fire Point	
B 1_133 ℃		5	tc	0.256	5	M Spec.	
C A* 15 to	219.24	5 7 5	ΔHc kcal/m ΔHf			Ultra V.	
B* 120 °C	1.3927	' 5	ΔFf			X-Ray Dif. Infrared	İ
K ———			Viscosity			Solubility in +	
t _e to	1		centistokes 7°C		1 1	Acetone	ł
<u>'x </u>			•			Carbon tet. Benzene	
A' to B' °C						Ether n-Heptane	- 1
c, =	1		B ^v to			Ethanol	
A¹* to			A ^V C			Water Water in	
B'* °C	7 2514	+-	(B ^V) to	1			
Ac 133 to	7.3514 1618.	5	(A ^V) °C	 	\vdash	1	
Cc	258.	5	c _p liq. °K			1	
Cryos, A° consts, B°			c _p vap. °K				
t _e °C	114.32	5	c _v vap.				
$T_{R} = 0.75$		لتـــــــــــــــــــــــــــــــــــــ	<u> </u>	L	L	+ grams/100 grams	anlunt
REFERENC		2-AP	I 3-Lit, 4-0	Calc, from de	t. dat		
SOURCE:		AP					
PURIFICAT	ION:	AP	rI				
LITERATU	RE REFER	ENCES	:				

	2 54		1-1 34				No. 14	
NAME _	3-Ethy	l-2-met	hyl-2-pentene			STRUCTURAL	FORMUL	J.A.
Mole	Ref.	Molecul	,	Molecular		CH ₃ CH ₂ C =		
% Pur.	1	Formula		Weight 112.2	:08	Ċ ₂ н ₅	CH ₃	
		Ref.			Ref.			Ref.
F, P, *C			dt/dP			f to		
F.P. 100%	ļ	\perp	°C/mm 25°C	0.0722	_	g '° <u>K</u> _		1
B.P. °C 760 mm	117.0	2	BP	0.8733 0.0468	5 4	h		—
100	57.63	4	t _e	0.0363	5	f' to		
30 10	31.43 11.43	4 5	30 mm	0.6563	_5_	g' <u>*°K</u>		1
ĭ	-22.07	5	ΔHm cal/g			}		+
Pressure			ΔHv cal/g 25°C	83.98	-	m to		
mm 25°C	21.47 1052.	5 5	30 mm	83.48	5	0		
t _e Density	1052.	 	BP	71.55	5 5	m¹ to		
g/ml 20°C	0.739	2	te te (d, e)	69.94 69.88	5	n' K_		
dt 25 4 30	0.735 0.731	2 4	AHv/Te	19.51	5	<u> </u>		
	0.755	14	d 31 to	87.86	5	Surface tension	22 74	_
a b	-0.038	4	e 129 °C	0.1394	5	dynes/cm. 20°C	23.74 22.72	5
Ref. Index			e' 31 °C	85.95 0.0789	5	40	21.72	5
ⁿ D 20°C	1.4247		d _c g/ml	0.250	5	Parachor [P]		
30	1.4198		V mi/g	3.996	5	20°C 30		
"C"	0.7637	7 4		290.	5	40	225 2	_
MR (Obs.)	39.0	2	P _c mm	20052.	-	Sugd.	335.2	5
MR (Calc.) (nD-d/2)	38.677 1.0552	5 2	25°C	1.0000	5	Exp. L.1.%/wt. u.		
Dielectric	1.0552		30 mm BP	1.0000	5	Dispersion	127.	2
A 31 to	6.9522	2 5	t.	0.9500 0.9401	5	Flash Point C Fire Point		
B 149 °C	1358.9	5	t _c	0.256	5			┼
С	216.77	_ 5	ΔHc kcal/m ΔHf			M. Spec. Ultra V.		
A* 31 to B* 139 °C	1.3910 1276.4)8 5 5	ΔFf			X-Ray Dif. Infrared		ŀ
K C32 S			Viscosity			Solubility in +		+-
t ₁ to			centistokes 7 °C		}	Acetone		
t _k to t _x °C			η °C			Carbon tet. Benzene		
A' 15 to	7.3492					Ether		
B' 1_31_°C	1563.2 234.77	5 5	B ^v to			n-Heptane Ethanol		
A'* 15 to	1.7463		B ^V		ł	Water		
B'* 31 °C	1463.5	5	(B ^V) to			Water in		
Ac 149 to	7. 3633		(A ^V) °C					
Bc_tc_°C	1677. 257.	5 5	c _p liq. °K					
Cryos, A°	<u> </u>	+-	c _p vap. °K					
consts. B°			l -	1				
t _e °C	128.99	5	c _v vap.					
$T_{\mathbf{R}} = 0.7$	5 T _c					† grams/100 gra	ms solve	nt
REFERENC	ES: 1-Do			Calc, from de	t. da	ta 5-Calc. by for	mula	·
SOURCE:			PI					
PURIFICAT	ION:	A	PI					
LITERATUI	RE REFEI	RENCES	5:					

							No. 150	0
NAME	3-Ethyl-4	-me	thyl-cis-2-pente	ene		STRUCTURAL 1	FORMULA	4
						CH ₃		
	7.4			Malala		CH3CH C = C	:нсн ₃	
Mole % Pur.		lecul rmul	a C ₈ H ₁₆	Molecular Weight 112.2	08	С ₂ Н ₅		
		Ref.	r		Ref			Ref.
F.P. °C			dt/dP			f to		
F.P. 1007	<u>'</u>		*C/mm 25*C	0.8401	5	g <u>°K</u> _		1
B. P. °C 760 mm	116.	2	BP	0.0467	4	h -		-
100 30	56.76	4	t _e	0.0363 0.6548	5	f' to		
10	30.61 10.66	5	30 mm	0.6546	-	h'		
1	-22.77	5	ΔHv cal/g		\vdash	m to		
Pressure mm 25°C	22,40	5	25°C	83.66	5	n •K_		
t _e	1049.	5	30 mm BP	83.22 71.33	5 5	<u> </u>		-
Density g/ml 20°C	0.739	2	te (d. a)	69.74	5	m' to		
dt 25	0.735	2	te (d, e) ΔHv/Te	69.68	5	0'		
	0.731	4	d 31 to	87.48	5	Surface tension	22	_
a b	0.755 -0.0 ₃ 8	4	1128_°C	0.1392	5	dynes/cm. 20°C	23.74 22.72	5
Ref. Index			d' 15 to	85.60 0.0779	5 5	40	21.72	5
ⁿ D 20°C	1.424	2 2	d g/ml v ml/g	0.250	5	Parachor [P] 20°C		
30	1.419	4	t _c *C	3.995 289.	5	30		
"C"	0.7625	4	P _c mm	20021.	5	40 Sugd.	335.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT		H	Exp. L.1.%/wt.		
(nD-d/2)	1.054	2	25°C 30 mm	1.0000	5	u. Dispersion	124.	2
Dielectric			BP	0.9500	5	Flash Point °C		<u> </u>
A 31 to B 1_148 °C		5	t _e t _c	0.9401 0.256	5	Fire Point		
c - 10 3	217.	5	ΔHc kcal/m			M Spec. Ultra V.		
A* 31 to	1.39076	5	ΔHf ΔFf			X-Ray Dif.		
B* 138°C	1272.8	5	Viscosity			Infrared Solubility in +		├
k	-]	centistokes 7°C			Acetone		
t _x •('			Carbon tet. Benzene		
A' 15 to B' <u>31</u> °C		5				Ether		
Tr.	235.	5	B ^V to			n-Heptane Ethanol		
A'* 15 to		5	AV - °C		i	Water Water in		
B'* 31 °C		5	(B ^V) to	1				†
Ac 148 to	1672.	5 5	(A ^V) °C		\vdash			
Cc	257.	5	c _p liq. °K					
Cryos. A° consts. B°			c _p vap. °K					1
t _e °C	127.86	5	c _v vap.					
$T_R = 0.7$						grams/100 gran	ns solvent	t
	CES: 1-Dow			alc. from de	t. da	ta 5-Calc. by for	nula	
SOURCE:	TION.		PI					
	RE REFEREI		PI S•					
	NOFERE	·OES	••					

No. 151 3-Ethyl-4-methyl-trans-2-pentene NAME STRUCTURAL FORMULA CH3 $CH_3\dot{C}HC = CHCH_3$ Mole Ref. Molecular Molecular ċ₂н₅ C8H16 Weight 112, 208 % Pur. Formula Ref. Ref. Ref. F. P. °C dt/dP f to F.P. 100% °C/mm 25°C g <u>°K</u> 0.7881 5 B. P. ℃ h BP 0.0466 **4** 5 760 mm 114.3 2 t_e 0.0363 ſ١ to 100 55.31 4 29.27 g' <u>°K</u> 30 4 30 mm 5 0.6520 10 9.40 5 h! AHm cal/g -23.89 5 m to ∆Hv cal/g Pressure °K n 25°C 83.16 mm 25°C 24.03 5 o 30 mm 82.83 5 te 1044. 5 ВP 71.00 5 m' to Density ^te 69.44 n' ۰ĸ g/ml 20°C te (d, e) 0.7350 0.7308 2 69.38 5 ٥' 25 $\mathbf{d_4^t}$ 2 AHv/T 5 19.52 30 0.7266 4 Surface tension d 29 to 86.90 5 0.7518 dynes/cm. 20°C a 4 23.23 ᇸᅱ ٠c 0.1391 5 b -0.0383 4 22.17 5 30 to 85.05 5 5 Ref. Index 40 21.15 e' 29 0.0760 5 ⁿD 20°C 1.4210 [P] Parachor d_c g/ml 0.248 5 25 1.4183 2 20°C vc ml/g t_°C 5 4,028 30 1.4158 4 30 ^tc 285. 5 40 "C" 0.7615 4 P_c mm 19717. 5 Sugd. 335.2 5 MR (Obs.) 38.71 2 PV/RT Exp. L.l.%/wt. MR (Calc.) 38.677 5 25°C 1.0000 5 (nD-d/2)1.0535 2 u. 30 mm 1.0000 Dispersion 2 5 5 124 Dielectric BP 0.9500 Flash Point C 0.9403 5 A 29 to 6.95117 5 Fire Point 0.256 5 B 1146 °C 1349.7 5 M. Spec. C 217.28 5 AHc kcal/m Ultra V. ΔHf A* 29 to 1.39291 5 X-Ray Dif. ΔFf B* 136 °C 1267.6 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. $\mathbf{t_x}_{oldsymbol{\perp}}$ °C Benzene A' 15 to 7.35081 Ether B' 1_29 °C 1553.9 5 n-Heptane B_v | C' 235.28 5 to Ethanol ۰c Water A** 15 to 1.74885 5 B'* 29 °C (B^V) Water in 5 1454.3 to Ac 146 to (A^V) 7.3622 5 °C Bc tc °C 5 1665. cp liq. °К 257. 5 Cryos. A° cp vap. °K consts. B° c_v vap. t_e °C 125.93 5 $T_{\mathbf{R}} = 0.75 T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 152	2
NAME	2, 3, 4-Tri	meth	yl-2-pentene			STRUCTURAL I	FORMULA	L
Ī						сн ₃ сн с = 0	C + CH -	
			T		\neg	Ċн ₃ ċн ₃ ċн ₃	Эн.	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	08	303	3	
		Ref.			Ref			Ref.
F.P. *C	-113.3	2	dt/dP	1		f to		
F.P. 100%			°C/mm	0.0400	ا ۔ ا	g <u>*K</u>		
B. P. °C 760 mm	116.26	2	25°C BP	0.8499 0.0467	5 4	h		
100	57.02	4	t.	0.0363	5	f' to		
30 10	30.86	5	30 mm	0.6550	5	g' 'K_ h'		
1	-22.54	5	ΔHm cal/g			 		
Pressure			ΔHv cal/g 25°C	83.79	5	m to		
mm 25°C	22.11 1050.	5	30 mm	83.33	5	•		ĺ
Density		i -	BP	71.43 69.83	5	m¹ to		
g/ml 20°C		2	te (d, e)	69.77	5	n' •K_		
dt 25 4 30	0.7391 0.7348	2 4	AHv/T	19.52	5	0'		
	0.7606	4	d 31 to		5	Surface tension dynes/cm. 20°C	24, 31	5
ъ	-0.0385	4	a, - 135 °C		5	30	23.19	5
Ref. Index			e' 31 °C		5	40	22.11	5
ⁿ D 20°C	1.4275	2	d g/ml	0.250	5	Parachor [P] 20°C		
30	1.4223	4	tc *C	3.997	5	30		İ
"C"	0.7639	4	P _c mm	20011.	5	40 Sugd	335.2	5
MR (Obs.) MR (Calc.		2	PV/RT			Exp. L.1.%/wt.	333.2	Ť
(nD-d/2)	38.677 1.0558	5 2	25°C 30 mm	1.0000	5	u.		
Dielectric			BP BP	1.0000 0.9500	5	Dispersion	127.	2
A 31 to		5	t _e	0.9401 0.256	5	Flash Point °C Fire Point		•
B 1.149.℃	1356.9 217.	5	t _c	0.230	 	M Spec.		
A* 31 to	+	5	ΔHf		1 1	Ultra V. X-Ray Dif.		Ì
B* 138 °C		5	ΔFf	ļ	1	Infrared		1
K c			Viscosity centistokes			Solubility in +		
tk to			η •c	}		Acetone Carbon tet.]
t _x *C		_				Benzene		1
B' _ 31 °C		5			$oxed{oxed}$	Ether n-Heptane		l
C'	235.	5	B ^V to			Ethanol		
A'* 15 to B'* 31 °C		5 5		-		Water Water in		
Ac 149 to	7.3646	5	1	1				
Bc t °C	1674.	5	c _p liq. °K		-			
Cc	257.	5	1 -					
Cryos. A° consts. B°	1		c _p vap. °K					
t _e °C	128.15	5	c _v vap.					
$T_R = 0.7$			4		·	+ grams/100 gran	ns solven	 t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for		
SOURCE:		AI	PI					
PURIFICAT		AF						
LITERATU	RE REFERE	VC ES	3:					

Mole Ref. Molecular C8H16 Molecular CH3 CH3 CH3 CH3	NAME	2, 4, 4-T	imet	thyl-2-pentene		STRUCTURAL FORMULA	A	
Mole Ref. Molecular Ref. Molecular Ref.	<u> </u>			· · · · · · · · · · · · · · · · · · ·			CH ₃	
F. P. C						8		
F.P. 100%			Ref.			Ref.		Ref.
B. P. °C 104.91 104.91 104.91 100 47.15 4 30 21.67 100 2.33 5 10 2.23 5 10 2.23 5 10 2.23 5 10 2.23 5 10 10 2.23 5 10 10 2.23 5 10 10 2.23 5 10 10 2.23 5 10 10 2.23 5 10 10 10 10 10 10 10			2					
Description		•			0.5523	5	g '° <u>K</u>	
100		104 91	,				h	
10	100		4	t _e	0.0363	5	1	
1				30 mm	0,6379	5_	1 1	
Pressure				ΔHm cal/g	<u> </u>		}	
mm 25°C 35,62 5 80,01 80,46 5 6 70 1017. 5 80,01 80,46 5 6 70 1017. 6 80,99 6 70 1017. 6 80,99 6 70,60 5 6 70,60 5 6 70,60 5 70 70 70 70 70 70 70	Pressure							
Density g/ml 20°C 0,7218 2 t t t t t t t t t	mm 25°C							
Seminary Seminary		1017.	5	BP	68.99	5	m' to	\neg
dt 25		0 7218	,	te (d. a)			n' i ok	
A					1	1 1	0'	
Ref. Index	4 30	0.7133	4				Surface tension	
Ref. Index nD 20°C				e 120 °C	83.45 0.1378		dynes/cm. 20°C 21.60	
No			4	d' to	1		1 2 1 20.07	
D 25			2					┪
MR (Obs.) 39, 01 2 MR (Calc.) 38, 677 5 25°C 1.0000 5 1.0000 5 25°C 30 mm 1.0000 5 5 25°C 30 mm 1.0000 5 5 5 25°C 30 mm 1.0000 5 5 5 25°C 30 mm 1.0000 5 5 5 5 5 5 5 5 5	25	1.4135	2	d _c g/ml				
MR (Obs.) 39, 01 2 MR (Calc.) 38, 677 5 25°C 1.0000 5 1.0000 5 25°C 30 mm 1.0000 5 5 25°C 30 mm 1.0000 5 5 5 25°C 30 mm 1.0000 5 5 5 25°C 30 mm 1.0000 5 5 5 5 5 5 5 5 5		1.4110	4	tc •C				1
MR (Obs.) MR (Calc.) (nD-d/2) 38.677 1.0551 PV/RT 25°C 1.0000 BP 0.9500 BP 0.9500 Flash Point °C Fire Point M. Spec. Ultra V. X-Ray Dif. Infrared Wiscosity centistokes N A' 15 to B' 234.9 A' 15 to B' 25°C C' - C 15.34.9 A' 15 to B' 25°C C' - C 15.34.9 A' 15 to B' 25°C C' - C 15.34.9 A' 15 to B' 25°C C' - C 15.34.9 A' 15 to B' 25°C C' - C 15.34.9 A' 15 to B' 25°C C' - C C C C C C C A' 15 to B' 25°C C' - C C C C C A' 15 to B' 25°C C' - C C C C C A' 15 to B' 25°C C' - C C C C C A' 15 to B' 25°C C' - C C C C C A' 15 to B' 25°C C' - C C C C C A' 15 to B' 25°C C' - C C C C C A' 15 to B' 25°C C' - C C C C C A' 15 to B' 25°C C' - C 1622. C' - C 258. C' - Vap. TR = 0.75 T _C REFERENCES: 1-Dow 2-API 3-Lit, 4-Calc, from det, data 5-Calc, by formula SOURCE: API	"C"	0.7668	4		1	1 1		5
No. 1.051 2 25°C 1.0000 5 1.0000 1.0000 5 1.0000 5 1.0000 5 1.0000 5 1.0000 1.0000 5 1.0000 5 1.0000 1.0000 5 1.0000 1.0000 1.0000 5 1.0000 1.0000 1.0000 5 1.0000 1.0000 1.0000 5 1.0000 1.0000 1.0000 5 1.0000 1.0000 1.0000 5 1.00000 1.00000 1.00000 1.						H		
Dielectric				25°C			u.	
A 15 to 6.94038 5 1315.2 5 C 1315.2 5 C 1315.2 5 AHc kcal/m AFf	Dielectric							2
No. Spec. Spec.		6.94038	5	t_	0.9411	5		
A 15 to	B (134 °C	1315.2	5	°c	0.256	5		
A				∆Hc kcal/m	1			
Viscosity centistokes Visc								
t _k to Acetone A' to B' c' C B' -°C B' to B' to Bi* °C B' to B' to Bc t _c °C 1622. 5 cc 1622. 5 cc 1622. 5 consts. B° C Cp vap. °K Cryos. A° consts. B° c _p vap. °K c _p vap. °K c _p vap. T _R = 0.75 T _c † grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API		- 1234. /	,	Viscosity				
Carbon tet. Benzene Ether n-Heptane Ethanol Water in	I . — — -	-						
A' to B' - °C B' to A' to A' C Ether n-Heptane Ethanol Water Water in Ach 134 to 7, 3517 5 C (B') to (A') °C Bc tc °C 1622, 5 5 C C C Diq. °K Cryos. A' consts. B' c vap. °K te °C 115.34 5 c vap. °K TR = 0.75 Tc T grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API				"			Carbon tet.	
B' °C	A' to		-					
A'* to B'* °C		<u>: </u>				\vdash	n-Heptane	
B * °C (B') to Water in		 		B to				
Acl 134 to 7.3517 5 (A ^V) °C			i I		-			
Bc tc °C 1622. 5 cp liq. °K	Ac 134 to	7, 3517	5					
Cryos. A*	Bc tc °C	1622.	5		 	\vdash		
te °C	Cc	258.	5	_				
te °C 115.34 5 C _v vap. TR = 0.75 T _C ** grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit, 4-Calc, from det, data 5-Calc, by formula SOURCE: API PURIFICATION: API				c _p vap. °K				
TR = 0.75 Tc			5	c _v vap.				
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API			L		1		grams/100 grams solvent	
SOURCE: API PURIFICATION: API			2-A	PI 3-Lit. 4-	Calc. from de	t. da		
	SOURCE:	<u> </u>	A					
LITERATURE REFERENCES:	PURIFICA	TION:	A	PI			100	
	LITERATU	RE REFERE	NCES	3:				
·								

						No.	154
NAME	3, 4, 4	-Trimeth	yl-cis-2-pente	ne		STRUCTURAL FORM	
			,			CH ₃	
1_						сн, с с = снсн	•
Mole	Ref.	Molecul		Molecular		ĊH3ĊH3	3
% Pur.	ليل		a 98-16	Weight 112.	$\overline{}$	3 3	5.
	T	Ref.			Ref		Ref.
F.P. °C F.P. 100%	 		dt/dP *C/mm			f to to	1
B. P. °C	 		25°C	0.7214	5	g <u>*K</u>	ı
76 0 mm	112.	2	BP	0.0463	4 5		
100 30	53.31 27.41		t _e 30 mm	0.6486	5	f' to g' K	- 1
10	7.64	5	ΔHm cal/g	0.0400	ť	h'	
1	-25.47	5	ΔHin cal/g		-	m to	
Pressure mm 25°C	26,48	5	25°C	82.58	5	n •K	
t _e	1038.	5	30 mm	82.25	5	0	
Density			BP te	70.51 68.99	5	m' to	- 1
g/ml 20°C	0.73		te (a, e)	68.94	5	n' °K	1
dt 25 4 30	0.73	- 1 - 1	AHv/T _e	19.52	5		
a	0.75	5 4	d 20 to	86.05	5	Surface tension dynes/cm. 20°C 23.7	4 5
ь	-0.03	8 4	_e,		5	30 22.7	
Ref. Index n _n 20°C	1.42	3 2	e'			40 21.7	2 5
ⁿ D 20°C	1.42		d g/ml	0.252	5	Parachor [P] 20°C	ì
30	1.41	8 4	v _c ml/g t _c °C	3.969	5	30	
"C"	0.76		P _c mm	19940.	5	40 Sugd. 335.2	. 5
MR (Obs.) MR (Calc.)	38.7 38.67	7 2	PV/RT		-	Exp. L.1.%/wt.	
(nD-d/2)	1.05		25°C 30 mm	1.0000	5	u.	2
Dielectric			BP	0.9500	5	Dispersion 124. Flash Point °C	
A 20 to			t _e	0.9405 0.256	5	Fire Point	İ
B <u> 144 °C</u> C	1341.1	5 5	t _c AHc kcal/m	0,230	-	M Spec.	
A* 20 to	1.39	+-	ΔHf	j		Ultra V. X-Ray Dif.	
B* 133 ℃		5	ΔFf		_	Infrared	ľ
K — — —			Viscosity centistokes			Solubility in +	
t _k to	1		η °c			Acetone Carbon tet.	
t c						Benzene	
A' to B' *C						Ether n-Heptane	- 1
c,			B ^V to	Ţ.		Ethanol	
A'* to B'* °C			$\frac{A^{\mathbf{v}}}{(B^{\mathbf{v}})} - \frac{{}^{\mathbf{c}}C}{to}$	-[Water Water in	
Ac 144 to		98 5					
Bc t °C	1655.	5			\vdash		
Cc	258.	5	c _p liq. °K				1
Cryos. A° consts. B°			c _p vap. °K				
t _e °C	123.34	5	c _v vap.				
$T_R = 0.7$						+ grams/100 grams sol	vent
REFERENC	ES: 1-De	ow 2-AF	PI 3-Lit, 4-0	Calc. from de	t. da	ta 5-Calc, by formula	
SOURCE:	···-	AF					
PURIFICAT		AF					
LITERATU	RE REFE	ERENCES	:				

							No. 1	55
NAME	3, 4, 4-Tı	imet	hyl-trans-2-per	ntene		STRUCTURAL	FORMU	LA
1 [CH ₃		
							C11 C11	
Mole	Ref. Mo	lecul	ar	Molecular	- 1	СН ₃ Ċ С =	CHCH ₃	
% Pur.		rmula		Weight 112.2	08	ĊH ₃ ĊH ₃		
		Ref.			Ref.	T		Ref.
F.P. °C		1	dt/dP					
F.P. 100%	, 	 	°C/mm			f to		
B.P. °C			25°C	0.7214	5			
760 mm	112.	2	BP	0.0463	4	h		+-
100	53, 31	4	t _e	0.0363	5	f' to	İ	
30 10	27.41 7.64	4 5	30 mm	0.6486	5_	h		
1 1	-25, 47	5	∆Hm cal/g			h' !	 	-
Pressure		1	∆Hv cal/g			m to		
mm 25°C	26.48	5	25°C	82.58	5			
t _e	1038.	5	30 mm BP	82.25 70.51	5 5			+
Density			t_	68. 99	5	m' to		
g/ml 20°C		2	te (a, e)	68.94	5	n' <u>•K</u>	1	
d ₄ 30	0.735 0.731	2 4	AHv/Te	19.52	5	<u> </u>	ļ	
		<u> </u>	d 20 to	86.05	5	Surface tension		_
a b	0.755 -0.0 ₃ 8	4 4	_e <u> 123 °C</u>	0.1387	5	dynes/cm. 20°C	23.74	5
Ref. Index		+	d' to e' °C			40	21.72	5
n _D 20°C		2			+-	Parachor [P]		
25	1.421	2	d g/ml	0.252 3.969	5	20°C		
30	1.418	4	vc ml/g tc °C	283.	5	30		
"C"	0.7608	4	P _c mm	19940.	5	40 Sugd.	335.2	5
MR (Obs.)		2	PV/RT		\vdash	Exp. L. l. %/wt.	-	+
MR (Calc. (nD-d/2)		5 2	25°C	1.0000	5	u.		
Dielectric	1.054	+ -	30 mm	1.0000	5	Dispersion	124.	2
A 20 to	(0402(_	BP t _e	0.9500 0.9405	5	Flash Point C		
B 144 °C		5	tc	0.256	5	Fire Point		
c	217.72	5	AHc kcal/m	 		M. Spec.		
A* 20 to	1.39246	5	ΔHf	i		Ultra V. X-Ray Dif.		
B* _133 °C	1259.5	5	ΔFf		-	Infrared		
K			Viscosity centistokes		1	Solubility in +		
t _k to			η °C		1	Acetone		1
'x					1	Carbon tet. Benzene		
A'I to					1	Ether		
B'°			B ^v to			n-Heptane Ethanol		
	+	1	B ^V to			Water		
A'* to			(B ^V) to	1		Water in		
Acl 144 to		5	(A ^V) °C					
Bc tc °C	1655.	5		 	+	1		
Cc	258.	5	c _p liq. °K				1	
Cryos. A°			c _p vap. °K		-		1	1
consts. B°	 	1	c _v vap.			1		
t _e °C	123.34	5	v vap.				L	1
$T_R = 0.7$	5 T _c					† grams/100 gra	ms solve	nt
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		A	.PI					
PURIFICA	TION:	A	PI					
LITERATU	RE REFERE	NCES	S:					

								No. 15	6
NAME	2-	Isopre	opyl-3	-methyl-1-bute	ene		STRUCTURAL	FORMUL	A.
ľ							CH ₃		
1		_		т			, , , ,	= CH ₂	
Mole	Re	f. M	olecul	ar Cu	Molecular		ċн(с		
% Pur.	L	F	ormul	аг С ₈ Н ₁₆	Weight 112.2	208		**3′2	
			Ref.			Ref			Ref.
F, P. °C	<u> </u>		_il	dt/dP			f to		1
F.P. 100%				*C/mm	0.5340	ا ۽ ا	g		
B. P. *C	1			25°C BP	0.5340 0.0455	5 4	h '		
760 mm 100	104.		2 4	t _e	0.0363	5	f' to		
30	20.		4	30 mm	0.6365	5	g'	j	
10	1.	65	5	ΔHm cal/g	†	1	h'		
1	-30.	18	5	ΔHv cal/g		+	m to		
Pressure mm 25°C	36.	97	5	25°C	79.69	5	n ' <u>*K</u>	4	
t _e	1015.	71	5	30 mm	80.24	5	<u> </u>		
Density	+		+	BP t	68.81 67.44	5	m¹ to		
g/ml 20°C		722	2	t _e (d, e)	67.39	5	n' •K	4	
dt 25 4 30		718	2	ΔHv/T	19.53	5	° 1		
	+	714	4	d 15 to	83,13	5	Surface tension		
a b		738 0 ₃ 8	4	<u> </u>		5	dynes/cm. 20°C	21.62	5
Ref. Index		-1-	+-	d' to		1	40	19.72	5
n _D 20°C		4085	2			5	Parachor [P]		
25 30		4061	2	d g/ml vc ml/g tc °C	0.247 4.04	5	20°C		
		4037	4	tc °C	269.	5	30 40	1	
"C"		7535	4	P _c mm	19096.	5		335.2	5
MR (Obs.) MR (Calc.			2 5	PV/RT		1	Exp. L.1.%/wt.		
(nD-d/2)		677 04 75	2	25°C	1.0000	5	u.		ł
Dielectric	—		\top	30 mm BP	1.0000 0.9500	5	Dispersion	122.	2
A 15 to	6.	9399	5	te	0.9412	5	Flash Point °C Fire Point		
B (133 °C	1312.	1	5	t _c	0.256	5	M Spec.	 	+-
C	219.	24	5	ΔHc kcal/m ΔHf	1		Ultra V.	l	
A* 15 to		39277		ΔFf	i		X-Ray Dif.		1
B* 125 °C	1231.	9	5	Viscosity	1	1	Infrared	ļ	<u> </u>
c	_			centistokes	1		Solubility in + Acetone		
tk to				η ° ⊂	;		Carbon tet.		
t _x i °C							Benzene	!	1
B' °C							Ether n-Heptane	į	-
c'	_			B ^V to			Ethanol	į	1
A1* to				A ^V O	<u>.</u>		Water	ļ	
B'* *(;		\bot	(B ^V) to	-		Water in	<u> </u>	+
Ac 133 to	7.	3515	5	(A ^V) •c	;			ŀ	
Bc tc_C	1618. 258.		5	c _p liq. °K					
Cryos, A°	1		+ -	_				ŀ	
consts. B°	İ			c _p vap. °K					
te °C	114.	32	5	c _v vap.				j	1
$T_{R} = 0.7$	5 T.			L	<u> </u>		grams/100 gra	1	
REFEREN		- Dow	2-AF	PI 3-Lit. 4-	Calc from de	• 4=	ta 5-Calc. by for		i t
SOURCE:			AP		<u> </u>	t. da	us 3-Carc. by for	III UIA	
PURIFICA'	TION:		AP						
LITERATU		FERI							
			DINO EQ	'•					

	2 Esh 2	2 4	ima thad I bases				57	
NAME	2-Etny1-3	, 3-a	imethyl-l-buter	ie		STRUCTURAL	. FORMU	LA
L	2-tert-Bu	tyl-l	-butene			CH ₃		1
Mole	Ref. Mo	1 1		Molecular		сн ₃ с с		
% Pur.	For	ecul mula	C ₈ H ₁₆	Weight 112.20	08	ĊH ₃ Ċ	2 ^H 5	
		Ref.		T	Ref.			Ref.
F. P. °C			dt/dP			f to	T	
F.P. 1009	6		°C/mm		}	g °i		
B. P. ℃			25°C BP	0.6682 0.0461	5	h	1	
760 mm 100	110. 51.56	2	t _e	0.0363	5	f' to		
30	25.78	4	30 mm	0.6456	5	g' ¦º1	<u>s</u>	
10	6.10	5	ΔHm cal/g		Τ-	h'		
Pressure	-20.00	<u> </u>	ΔHv cal/g			m to		
mm 25°C	28.82	5	25°C	81.84	5	n º1	<u> </u>	
te	1032.	5	30 mm BP	81.73 70.07	5		-	+
Density			t_	68.59	5	m' to		
g/ml 20°0	0.728	2 2	t _e (d, e)	68.53	5		7	
dt 25 4 30	0.720	4	ΔHv/T _e	19.52	5	Surface tension	+	+-
a	0.744	4	d 20 to	85.30 0.1385	5	dynes/cm. 20°C		5
b	-0.038	4	d' to	0.1309	_	30 40	21.37	5
Ref. Index		١, ١	e' °C		ļ	Parachor [P]	20.42	+-
25	1.4135	2 2	d _c g/ml	0.248	5	20°C	;	
30	1.4111	4	vc ml/g tc °C	4.035 279.	5	30		
"C"	0.7601	4	P _c mm	19472.	5	40 Suga	1. 335. 2	5
MR (Obs.) MR (Calc.		2 5	PV/RT		<u> </u>	Exp. L.1.%/wt		
(nD-d/2)) 38.677 1.0519	2	25°C	1.0000	5	u.	İ	
Dielectric			30 mm BP	1.0000 0.9500	5	Dispersion	122.	2
A 20 to	6.94544	5	t_	0.9407	5	Flash Point C Fire Point		
B 1141 °C		5	¹c	0.256	5	M. Spec.		+-
C	218.	5	ΔHc kcal/m ΔHf	!		Ultra V.		
A* 20 to B* 131 °C		5	ΔFf			X-Ray Dif. Infrared		
к — — -	_		Viscosity			Solubility in +		+
t _k - to	-		centistokes り °C			Acetone		
t _x °C			'			Carbon tet. Benzene		
A' to						Ether	1	
B'°	-		B ^V to			n-Heptane Ethanol		
A'* to		-	B ^V to A ^V C	İ		Water		
B'* °C			(B ^V) to	1		Water in		
Ac 141 to		5	(A ^V) °C					
Bc tc °C	7 1645. - 258.	5	c _p liq. °K					
Cryos. A		<u> </u>	II -					
consts. B			Р .	ĺ				
te °C	121.08	5	c _v vap.	1				
$T_R = 0.7$	75. T c					grams/100 gr	ams solve	nt
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ata 5-Calc. by fo		
SOURCE:		А	PI					
PURIFICA	TION:	A	PI					
LITERATU	JRE REFERE	NCE	S:					

			_					No.158	
NAME	l-Nor	nene					STRUCTURAL	FORMULA	
							617 4677 \ 61		
Mole % Pur.	Ref.	Mole	ecul mul		Molecular Weight 126.	234	CH ₃ (CH ₂) ₆ C	i=CH ₂	
			Ref.	Ī —		Ref			Ref.
F.P. °C	-81.37		2	dt/dP	T		f to	1	
F.P. 100%				°C/mm	2 010	ا ۔ ا	g °K		
B. P. *C	1 146 066	\Box		25°C BP	3.018 0.04944	5 4	h ¦		
760 mm 100	146.86 84.21		2 2	t	0.0362	5	f to		
30	56.55		2	30 mm	0.6928	5	g' K		
10 1	35.44	1	2 5	ΔHm cal/g			h'		
Pressure	+		-	ΔHv cal/g			m 300 to	0.0247	4
mm 25°C	5, 33	9	5	25°C 30 mm	86.83 82.32	5 5	n 600 °K	0.0013 -0.0 ₆ 49	
t _e	1132.		5	BP BP	69.58	5		 	
Density g/ml 20°C	0.72	222	2	t _e	67.48 67.36	5 5	m' 700 to n' 1000 °K	0.1080 0.0011	4
	0.72		2	t (d, e)			0'	-0.0638	
d ₄ 25	0.72	140	4	ΔHv/T _e	19.55	5	Surface tension		\vdash
	0.74		4	d 25 to e 170 °C	90.42 0.1419	5 5	dynes/cm. 20°C	21.83	5
ъ	-0.03	780	4		1		30 40	20.91	5
Ref. Index		572	2	e' i °C			Parachor [P]		Ě
25	1.41	333	2	d g/ml	0.25 4.0	5 5	20°C		
30	1.41		4	vc ml/g tc °C	321.	5	30 40		
"C"	0, 75		4	P _c mm	18000.	5		374.2	5
MR (Obs.) MR (Calc.			2 5	PV/RT		\vdash	Exp. L.1.%/wt.		
(nD-d/2)	1.05		2	25°C 30 mm	1.0000 1.0000	5 5	u.	 ,,,,,	١,
Dielectric				BP	0.9463	5	Dispersion Flash Point °C	114.4	2
A 25 to		387	2	t _e	0.9338 0.245	5	Fire Point		
B <u>173</u> °C C	1435.4 205.53	5	2 2	t _c	1341.90	2	M Spec.		<u> </u>
A* 25 to	1.42		-5	ΔHC KCal/m	1341.70	ا ۲	Ultra V.		
B* 165 °C			5	ΔFf			X-Ray Dif. Infrared		ŀ
к — — —	-	-		Viscosity			Solubility in +		
ξ _k − το	-	1		7 40 °C	0.688	2	Acetone		
t _x '°C		1	j	60	0.568	2	Carbon tet. Benzene	İ .	
A' to				80 100	0,488 0,426	2 2	Ether	·	
B' ∟ _ °	-			BV 30 to	434, 21	4	n-Heptane Ethanol		
A'* to		-+		A 70 °C	2.4510	4	Water		
B'* °C				(BV) 70 to	388.81	4	Water in	<u></u>	<u> </u>
Ac 173 to		08	5	(A ^V) 110 °C	2.58744	4			Ì
Bc tc_°C	1518. 211.		5	cp liq. °K					
Cryos, A°	+	\dashv		1 -	0.38231	2			1
consts. B°				P 400	0.48204	2			
t _e °C	162.52		5	c _v vap.					
$T_R = 0.75$							grams/100 gra	ms solveni	<u> </u>
REFERENC	ES: 1-D	ow 2	-AF		alc. from det	da	ta 5-Calc. by for	mula	
SOURCE:			AP						
PURIFICAT			AP						
LITERATU:	KE REFI	LKEN	CES	:					

								No. 159	
NAME _	l-Dece	ne				ST	RUCTURAL	FORMUL	A
							CH (CH) C	·u-cu	
Mole % Pur.	Ref. Moi	ecul:		Molecular Veight 140.26	0		сн ₃ (сн ₂) ₇ с	л-сн ₂	
		Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-66; 310	2	dt/dP °C/mm	0 770		f g	to °K		
B. P. °C 760 mm 100 30 10	170.570 105.198 76.33 54.29	2 2 2 2	25°C BP te 30 mm	8.778 0.05157 0.03615 0.7231	5 2 5 5	h f' g' h'	to °K		
Pressure mm 25°C t _e	18.2 1.632 1197.	5 5 5	ΔHv cal/g 25°C 30 mm	86.70 79.78	5	m n o	900 •K	0.0246 0.0013 -0.0 ₆ 49	4
Density g/ml 20°C dt 25 d4 30	0.74081 0.73693 0.73305	2 2 4	BP te te (d,e) ΔHv/Te	66.84 64.46 64.27 19.55	5 5 5	m' n' o'	700 to 1000 °K	0.1096 0.0011 -0.0 ₆ 38	4 4 4
a b	0.75632 -0.0 ₃ 775	4	d 25 to e 189 °C d' to	90.27 0.1374	5 5		face tension es/cm. 20°C 30 40	22.68 21.75 20.84	5 5 5
Ref. Index n _D 20°C 25 30	1.42146 1.41913 1.41675	2 2 4	e' °C d _c g/ml v _c ml/g t _c °C	0.243 4.11 343.	5 5 5	Par	achor [P] 20°C 30	20.01	
"C"	0.7563	4	t _c °C P _c mm	16209.	5		40 Sugd.	413.2	5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	48.059 47.913 1.05106	2 5 2	PV/RT 25°C 30 mm BP	1.0000	5 5 5	_	u. persion	112.9	2
A 25 to B 233 °C	6.96034 1501.872	2 2	t e t	0.9450 0.9303 0.243	5 5	Fir	sh Point °C e Point Spec.		
A* 76 to B* 199 °C	197.58 1.45987 1417.2	5 5	ΔHc kcal/m ΔHf ΔFf	1488.82	2	Ult:	ra V. Ray Dif. ared		
K c to to to to to to to to to to to to to			Viscosity centistokes 7 40°C 60 80 100	0.855 0.692 0.583 0.501	2 2 2 2 2	Ac Ca Be Et	ubility in + etone rbon tet. nzene her		
C' - to			B ^V 30 to A ^V 70 °C	479.3 2.40181	4	Et Wa	Heptane hanol ater ater in		
B'* °C Ac 233 to Bc t _c °C	7.8050 2317.	5	(B ^V) 70 to (A ^V) 110°C c _p liq. °K	433.9 2.53743	4				
Cryos. A° consts. B°	302.	5	c vap.300K p 400 c vap.	0.38329 0.48332	2 2				
t _e °C	189.28	5	vr.	l		L		L	<u> </u>
TR = 0.82		2-A	PI 3-Lit. 4-0	Calo from 3-			cams/100 gra		.t
SOURCE:	ES: 1-Dow	API		Calc, from de	ı. da	ua 5	-Carc. by for	IIIUIA	
PURIFICAT	TION:	API							
	RE REFERE								

									No.160	
NAME	1 -	Und	ecen	•		_	STR	UCTURAL	FORMULA	L
								CH (CH) C	л-сн	
Mole % Pur.	Ref.	Mo	lecul	ar C ₁₁ H ₂₂	Molecular Weight 154.2	86		СН ₃ (СН ₂)8С	,,,,_,,,,	
// 1 41 /		10	Ref.		Weight 100.0	Ref				Ref.
F.P. °C	-49.18	5	2	34/35	T	1.001	<u> </u>	<u> </u>		-
F.P. 100%			-	dt/dP *C/mm			f g	to K		l
B. P. *C				25°C	26.16	5	h			1
760 mm	192.67		2	BP	0.05348 0.03615	2 5		+ to		
100 30	124.85 94.89		2 4	t _e 30 mm	0.7506	5	g'	*K		
10	72.02		5	ΔHm cal/g	1	+-	h'			
1	34.54		5	ΔHv cal/g	 	\vdash	m] 300 to	0,0233	4
Pressure mm 25°C	0.49	0	5	25°C	86.85	5	n	60 0 ° K	0.0014	4
t _e	1255.	U	5	30 mm	77.50	5	<u> </u>	i	-0.0 ₆ 50	4
Density	 			BP t	64.33 61.70	5	m'	700 to	0.1143	4
g/ml 20°C	0.75	032	2	te te (d, e)	61.44	5	n' o'	1000 °K	0.0011	4
d ^t 25 4 30	0.74		2	AHv/Te	19.53	5	0.	<u>i </u>	-0.0 ₆ 38	4
	0.74		4	d 25 to	90, 27	5		ace tension		1_
a b	0.76		4	<u> </u>	0.1346	5	dyne	ss/cm, 20°C	23.39 22.46	5
Ref. Index				d' to				40	21.56	5
n _D 20°C			2		-	Н	Para	chor [P]		
25 30	1.42		2 4	d g/ml vc ml/g tc *C				20°C 30		1
"C"	0.75		4	l -	364.	5		40		l
MR (Obs.)	+		2	P _c mm	14948.	5		Sugd.	452.2	5
MR (Calc.)			5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	1.05	093	2	25°C 30 mm	1.00 00 1.00 00	5	Dist	u. ersion	112.1	2
Dielectric			L	BP	0.9420	5		h Point °C		<u> </u>
A 25 to		662	2	t _e	0.9253	5		Point		l
B (_257 °C	189.74		2 2	t _c	1635.75	2	M S	pec.		Г
A* 94 to	1.49		5	ΔHf	1035.13		Ultr			
B* 225 °C	1477.5	712	5	ΔFf			Infra	ay Dif. ared		
K C				Viscosity			Solu	bility in +		
t _k to	-			centistokes 7 40 °C	1.054	2	Ace	etone	·	1
tî °C	:			60	0.835	2		rbon tet. nzene		
A' to				80 100	0.691 0. 586	2 2	Eth	er		
B', ∟ _ °	-			B ^V 30 to	527.6	4		leptane anol		
A¹* to	 		\vdash	AV 70 °C	2.33787	4	Wa			
B'* °C				(BV) 70 to	471.6	4	Wa	ter in		<u> </u>
Ac 257 to	7.92	72	5	(AV) 110 °C	2. 50 3 97	4				1
Bc tc_°C	2539. 312.		5	c _p liq. °K	† · · · · · · · · · · · · · · · · · · ·					
Cryos, A°	312.		3	-	0.38403	١, ١				
consts. B°	ĺ			c _p vap.300°K 400	0.38403 0.48443	2			1	İ
t _e °C	214.20		5	c _w vap.						
$T_{\mathbf{R}} = 0.83$							+ gra	ms/1 0 0 grai	ns solven	t
REFERENC	ES: 1-D	ow	2-AF	PI 3-Lit. 4-0	Calc. from det	t, dat	ta 5-	Calc. by for	mula	
SOURCE:			AP	PI						
PURIFICAT	ION:		AF	PI						
LITERATU	RE REF	ERE	ICES	:				·		

								No. 161	١
NAME	l -Dod	ecene				ST	RUCTURAL	FORMUL	A
					$\neg \neg$		сн ₃ (сн ₂₎₉ с	H=CH ₂	
Mole % Pur.	Ref. M	olecul		Molecular	,,				
# Tu:-		Ref.	·	Veight 168.3	Ref.	T			Ref.
F.P. ℃	-35.230	2	dt/dP		Kei.	-			Kei.
F.P. 100%		+	°C/mm			f g	to °K		
B. P. °C			25°C	71.96	5	h			
760 mm	213.357	2	BP t _e	0.05522			to		
30	143.315	2 4	30 mm	0.7757	5	g'	°K		
10	88.72	5	ΔHm cal/g			h'	·		
1	49.20	5	ΔHv cal/g			m	300 to	0.0231	4
Pressure mm 25°C	0.1446	5	25°C	86.74	5	n o	600 ∘ K	0.0014 -0.0 ₆ 50	
t _e	1309.	5	30 mm BP	75.41 62.15	5				
Density	0.7503/		l t	59.30	5	m'	700 to	0.1145 0.0011	4
g/ml 20°C	0.75836 0.75474		'e (u, e,	58.97 19.54	5	0'	1	-0.0638	
d ₄ 30	0.75112	4	ΔHv/T _e		السّا	C	face tension		
a	0.77284		d 25 to	90.16 0.1313	5		es/cm. 20°C	23.99	5
b	-0.03724	4	d' to	0.1313	,	×	30 40	23.09 22.21	∽ 5
Ref. Index		2	e' °C			Pa.	achor [P]		Ť
25	1.42782	2	d _c g/ml	[Fai	20°C		ĺ
30	1.42562	+	tc *C	384.	5		30 40		ĺ
"C"	0.7530	4	P _c mm	13911.	5			491.2	5
MR (Obs.) MR (Calc.		5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	1.05084		25°C 30 mm	1.0000 1.0000	5	Dia	u. persion	110.7	2
Dielectric			BP	0.9400	5		sh Point °C	110.1	-
A 112 to	6.97522		: •	0.9214	5		e Point		
B _280 °C	_ 1619.86 182.27	2	tc ΔHc kcal/m	1782.68	2		Spec.		
A* 112 to	1.52979	+	ΔHf	1102.00	-		ra V. Ray Dif.		
B* 248 °C		5	ΔFf		_		ared		
K			Viscosity centistokes	1		Sol	ability in +		
t _k T to	-		η 40°C	1.286	2		etone rbon tet.		
t _x °C			60 80	0.995 0.811	2	Be	nzene		
A' to	1		ll 100	0.676	2		her Heptane		
c,'=			B _v to	581.2	4		hanol		
A¹* to			_A' °C	2.25329	4		ater ater in		
B'* °C		+-	(B ^V)	521.0 2.43370	4				
Acl 280 to Bc t _c °C		5	(A ^V)	2. 13310	-				
Cc C	327.	5	c _p liq. °K						
Cryos, A°			c vap.300°K	0.38464					
consts. B°	 	4	P 400 c, vap.	0.48529	2			1	Ì
te °C	237.59	5	,	L		L	4	L	L
$T_R = 0.84$							rams/100 gra		t
	CES: 1-Dow		PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		AI					***************************************		
PURIFICA		Al							
LITERATU	RE REFERI	ENCE	5:						
L									

									No. 162	
NAME	1-7	ride	cene				STF	UCTURAL	FORMULA	
T								CTT / CTT \ .		
				T		\neg		СН ₃ (СН ₂) ₁₀ (CH=CH ₂	
Mole % Pur.	Ref.	Mol For	lecul rmul	ar C ₁₃ H ₂₆	Molecular Weight 182.3	38				
			Ref.	ŀ		Ref				Ref.
F.P. °C	-23.07	0	2	dt/dP			f	l to		
F.P. 100%				*C/mm 25*C	207.11	5	g	°K		
B. P. °C 760 mm	232.78		2	BP	0.05680		<u>h</u> _	-		
100	160.7		2	t _e	0.0361	5	f	to		
30 10	128.84		4 5	30 mm	0.7980	5	g' h'	°K		
1	63.9		5	ΔHm cal/g				1 300 to	0,0233	4
Pressure				ΔHv cal/g 25°C	86.76	5	m n	600°K	0.0014	4
mm 25°C	0.05	4//	5	30 mm	73.58	5	۰	i	-0.0650	4
Density	1.350.		H	BP	60.05 56.96	5	m'	700to	0.1116	4
g/ml 20°C			2	te te (d, e)	56.57	5	n' o'	1000°K	0.0011	
dt 25 4 30	0.76		2 4	AHV/T	19.50	5	 	1	-0.0639	<u> </u>
8	0.77		4	d 25 to		5		ace tension s/cm. 20°C	24.52	5
Ъ	-0.03	72	4			5	3,	30	23.61	5
Ref. Index				e' °C			<u> </u>	40	22.73	5
n _D 20°C	1.43		2	d g/ml vc ml/g			Par	achor [P] 20°C	1	
30	1.42		4	t _c °C	401.	5		30		
"C"	0.75	18	4	P _c mm	12776.	5		40 Sugd	530.2	5
MR (Obs.)	61.96		2	PV/RT	 	-	Exp	L.1.%/wt.		H
MR (Calc.) (nD-d/2)	61.76		5 2	25°C	1.0000	5	∥ -	u.		2
Dielectric				30 mm BP	0.9370	5	<u> </u>	h Point °C	110.	-
A 128 to			2	t _e	0.9165	5		Point C		
B 1_300°C	1662.68		2 2	t _c AHc kcal/m	1929,60	2	M S	pec.		
A* 128 to	1.55	-	5	ΔHf	1727.00	-	Ultr	a V. ay Dif.		ŀ
B* 270°C	1577.9		5	ΔFf	<u> </u>			ay Dil.		
K — —	1			Viscosity centistokes			Solu	bility in +		
t _k to	1			7 40 °C		2		etone rbon tet.		
×	ļ			60 80	1.184 0.945	2 2	Be	nzene		
A' to B' °C				100	0.780	2	Eth n-I	ier Ieptane	1	
c'	1			B ^V to		1	Etl	anol		
A¹* to B¹* °C	ł			AV I °C	-		Wa Wa	ter ter in		
Ac 300 to	8,20	20	5	(B ^V) (A ^V)						
Bc t °C	3037.	20	5		-	<u> </u>				
Cc	341.		5	c _p liq. ∘K						
Cryos. A° consts. B°				c _p vap.300°K 400	0.38522 0.48602					
t _e °C			5	c _v vap.	<u></u>		<u> </u>			<u>L</u> .
TR = 0.85				N. 2 T.				ms/100 gran		<u> </u>
SOURCE:			API	PI 3-Lit. 4-0	aic. irom det	. da	ta 5-	Calc. by for	mula	
PURIFICAT			API							
LITERATU		CR EN								
	REFI	JIV EJN	-CES	·•						

									No.163	
NAME	1-	Tetr	adec	ene			ST	RUCTURAL	FORMUL	A.
								СН ₃ (СН ₂) ₁₁ С	H=CH ₂	
Mole % Pur.	Ref.	Mo!	ecul		Molecular Weight 196.30	64				
70 1 411		10.	Ref.	1 17 20 1	Weight 170.3	Ref.	Γ			Ref.
F.P. °C	-12,85		2	dt/dP			f	4-		
F.P. 100%				°C/mm			g	to °K		
B.P. °C				25°C BP	497.07 0.05820	5 2	h			
760 mm	251.10 177.1	0	2	te	0.0361	5	f'	to		
30	144.40		4	30 mm	0.8181	5	g'	°K		
10	119.0 74.5		5	ΔHm cal/g			h'			_
Pressure				ΔHv cal/g			m n	300 to	0.0231 0.0014	
mm 25°C	0.01	815	5	25°C 30 mm	86.67 71.91	5	0	••	-0.0650	4
Density	1406.		5	BP	58.18	5	m'	700 to	0.1155	-
g/ml 20°C	0.77	13	2	t _e (d, e)	54.89 54.42	5	n'	1000 °K	0.0011	4
dt 25 4 30	0.76		2	AHv/Te	19.48	5	ο'		-0.0 ₆ 38	4
a 30	0.76		4	d 25 to	90,50	5		face tension	24.00	_
b	-0.03		4	_e_ 290°C	0.1288	5	dyn	es/cm. 20°C 30	24.99 24.07	5
Ref. Index				d' to				40	23.17	5
ⁿ D 20°C	1.43		2	d _c g/ml			Par	achor [P]		ĺ
30	1.43		4	v ml/g	416.	5		30		ĺ
"C"	0.75	07	4	tc °C P _c mm	11740.	5		40 Suad	569.2	5
MR (Obs.)			2	PV/RT	11110	-	Evr	. L.1.%/wt.	307.2	- -
MR (Calc. (nD-d/2)) 66.38 1.05		5 2	25°C	1.0000	5	_	u.		ĺ
Dielectric	†			30 mm BP	1.0000 0.9350	5		persion	109.	2
A 144 to	6.96	15	2	t _e	0.9129	5		sh Point °C e Point		
B 319 °C	- 1699.76 165.53		2	tc ΔHc kcal/m	2067,52	2	М.	Spec.		
A* 144 to			5	ΔHf	2007.52	-		ra V. Ray Dif.		
B* 290 °C			5	ΔFf				ared		
K — —				Viscosity centistokes				bility in +		
t _k to				η 40°C	1.85	2		etone rbon tet.		
t _x °C		0.41	-	60 80	1.40	2 2	Be	nzene		
B' 144 °C		041	5	100	0.891	2		her Heptane		ĺ
	203.3		5	B ^V to A ^V °C			Et	hanol		
A'* 20 to B'* 144 °C		444	5	$\frac{1}{ \mathbf{B}^{\mathbf{v}} } - \frac{1}{ \mathbf{B}^{\mathbf{v}} }$	ŀ			iter iter in		
Acl 319 to		16	5	(A ^V)						
Bc tc °C	3298.	-	5	c _p liq. °K		\vdash				
Cryos. A°	356.		5	P	0.305//					1
consts. B°				c _p vap.300°K	0.38566 0.48665	2				
te °C	280.17		5	c _w vap.						
$T_R = 0.86$	T _c						+ g1	ams/100 gra	ms solven	t
REFEREN)ow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			AP	I						
PURIFICAT			AP							
LITERATU	RE REF	EREI	NCES	5 :						

·							<u>-</u>		No. 164	
NAME		l-Per	ntad	ecene			STR	UCTURAL 1	FORMULA	
Mole % Pur.	Ref.	Mole For	ecul:	arC ₁₅ H ₃₀	Molecular Weight 210.3	90		СН ₃ (СН ₂)12	сн≖сн₂	
			Ref.			Ref				Ref.
F.P. °C	-3.73	0	2	dt/dP			f	to		
F. P. 100%				*C/mm	1629.5	5	g	°K		
B. P. °C	3/0 17			25°C BP	0.0596	4	h			
760 mm 100	268.17 192.5	1	2 2	t.	0.0361	5		to		
30	159.1	- 1	4	30 mm	0,8362	5	g'	°K		
10 1	133.7		5	ΔHm cal/g			h'			
Pressure	1 /1.2		<u>-</u>	ΔHv cal/g	1		m	300 to	0.0232	
mm 25°C	0.00	609	5	25°C	86.60 70.37	5 5	n o	600 °K	0.0014 -0.0 ₆ 50	
t _e	1448.0		5	30 mm BP	56.41	5	ļ			-
Density	0.77			t	52.92	5	m' n'	700 to	0.1155 0.0011	4
g/ml 20°C	0.77		2 2	t _e (d, e)	52.39	5	0'	1000 K	-0.0638	
dt 25 4 30	0.76		4	ΔHv/T _e	19.44	5	S			
a	0.79	04	4	d 159to		5		s/cm. 20°C	25.39	5
Ъ	-0.03	70	4			5 5	3	30	24.49	5
Ref. Index			١ ,	e' 159°C		5	<u> </u>	40	23.61	,
n _D 20°C	1.43		2 2	d g/ml vc ml/g			Para	chor [P] 20°C		!
30	1.43		4	tc °C	431.	5		30		
"C"	0.74	97	4	1 -	10918.	5	}	40 50 ad	608.2	5
MR (Obs.)			2	P _c mm PV/RT	10718.	1	F		608,2	-
MR (Calc. (nD-d/2)	71.00 1.05		5 2	25°C	1.0000	5	Exp.	L. 1. %/wt. u.		
Dielectric	1.03	-	-	30 mm	1.0000 0.9325	5	Disp	ersion	108.	2
A 159 to	6, 95	02	2	BP t _e	0.9086	5		h Point °C		
B 337°C			2	tc				Point		ļ
c ——	157.		2	ΔHc kcal/m	2223.44	2	M S _I			
A* 159 to		900	5	ΔHf ΔFf				y Dif.		
B* ∟305 °C	1647.3	i	5	Viscosity	 	-	Infra			
c	_	ŀ		centistokes				bility in +		
tk to				7 40 °C	2.21	2 2		tone bon tet.		
t _x i *C	<u> </u>			60 80	1.64 1.265	2	Ber	zene		
B' °C				10 0	1.016	2	Eth	er Ieptane		
<u>c'</u>		i		B _v l to			Eth	anol		
A¹* to				AV I _ °C			Wat	ter ter in		
B'* °C				(B ^V)			- " a	er m		1
Ac 337 to Bc t _c °C		72	5	(A ^V)	L					ĺ
Cc Cc	376.	- 1	5	c _p liq. ∘K						1
Cryos. A°				cp vap.300 °K	0.38609	2				
consts. B°	_			400	0.48719	2				
t _e °C	299.58		5	c _v vap.						
$T_{R} = 0.87$	T _c						+ gra	ms/100 grai	ns solven	t
REFERENC	ES: 1-D	ow 2	-AF		Calc. from det	t. dat	a 5-	Calc. by for	mula	
SOURCE:			AF	PI						
PURIFICAT	ION:		AF	PI						
LITERATU	RE REF	EREN	CES	:						

No. 165 l-Hexadecene NAME STRUCTURAL FORMULA CH3(CH2)13CH=CH2 Molecular C16H32 Mole Ref. Molecula r 224.416 % Pur. Weight Ref. Ref. Ref. 4.120 2 F. P. ℃ dt/dP f to F.P. 100% °C/mm g ۰c 25°C 4630. 5 B. P. °C h BP 0.0610 4 284.4 2 760 mm t_e 0.0362 5 ſ١ 207.2 to 100 2 °C g' 173.13 4 30 30 mm 0.8527 5 10 147.2 5 h! ∆Hm cal/g 103.9 5 0.0231 300 to m AHv cal/g Pressure ۰ĸ n 600 0.0014 25°C 86.45 mm 25°C 0.0352 68.96 30 mm 5 1487. 5 o -0.0650 t. BP 54.75 5 m' | to 700 0.1158 4 Density 5 51.07 te (d, e) g/ml 20°C 0.78112 n' 1000 °K 0.0011 50.46 5 d_4^t 25 0.77759 ۰, -0.0638 4 AHv/T 5 19.39 30 0.77406 Surface tension d 173 91.08 5 0.79524 to dynes/cm. 20°C 25.75 5 1310 ᇷᅴ •c 0.1277 5 Ъ -**0.0**3706 30 24.83 5 to 40 23.94 5 Ref. Index e! °C 20°C [P] 1.44120 $\mathbf{n}_{\mathbf{D}}$ Parachor d_c g/ml 25 1.43907 2 20°C vc ml/g t °C 30 1.43694 4 30 $^{\mathbf{t}}_{\mathbf{c}}$ 5 444. 40 "C" 0.7489 4 P_c mm 10008. 5 Sugd. 647. 2 5 MR (Obs.) 75.898 PV/RT Exp. L.1.%/wt. MR (Calc.) 75.621 25°C 1.0000 (nD-d/2) 1.05064 2 30 mm 1.0000 Dispersion 107.1 2 0.9300 Dielectric BP 5 Flash Point °C 0.9043 5 A 173 to 6.936 2 Fire Point B _352°C 1755.2 2 M. Spec. Ultra V. С 148. 2 2370.37 AHc kcal/m 2 ΔHf A# 173 to 1.59011 5 X-Ray Dif. ΔFf B*[325 °C 1674.1 Infrared ĸ Viscosity Solubility in centistokes Acetone to 40 °C t_k [t_{x [} 2.62 2 Carbon tet. °C 60 1.89 2 Benzene 80 1.46 2 A' I to Ether 100 1.149 B' °C n-Heptane Т C 35 Ethanol to 740.1 5 **v** | 70 •c **Z.** 05527 Water AI* to Water in B'* °C (BV) 70 to 685.6 5 Ac 352 to (AV) 105 .c 8.6611 5 5 ₹. 22324 3905. Bc _tc_ 5 °C c_p liq. Cc 5 394. c_p vap.300K Cryos. A 0.38643 2 consts. B° 0.48767400 c vap. t, °C 5 317.97 = 0.87T grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

									No. 166	
NAME	l-Hept	adec	ene			\dashv	STRUCTU	RAL 1	FORMULA	L
Mole % Pur.	Ref.		lecul rmul		Molecular Weight 238.4	142	СН3(С	H ₂) ₁₄ C	H=CH ₂	
			Ref	,	9	Ref			· · · ·	Ref
F.P. °C	11.2		2	dt/dP	T		f	to		
F.P. 100%				°C/mm	1		g	°K		İ
B. P. °C				25°C BP	12449. 0.0622	5 4	h ¦			
760 mm 100	299.7 221.0		2	t _e	0.0362	5	<u></u>	- to		
30	186.3	4	4	30 mm	0.8672	5	g'	°K		l
10 1	160. 116.		2 5	ΔHm cal/g			h'			
Pressure	110.			ΔHv cal/g	<u> </u>	\Box	m 300		0.0229	
mm 25°C	0.0	3708	5	25°C	86.40	5	n 600	°K	0.0014 -0.0650	
t _e	1522.	•	5	30 mm BP	67.65 53.06	5 5				
Density				t _e	49.31	5	m' 700		0.1160 0.0011	
g/ml 20°C	0.7		2	t _e (d, e)	48.51	5	0 1000	,	-0.0638	
d ^t 25 4 30	0.7		4	ΔHv/T _e	19.33	5	Surface te	ngion		
a	0.7	992	4	d 186 to		5	dynes/cm		26.07	5
ь	-0.0	370	4	a, 325 c		5	y.	30	25.16	5
Ref. Index	1.4	432	,	e' ; °C	;	5	P	40 [D]	24.26	5
ⁿ D 20°C	1.4		2	d g/ml vc ml/g		1 1	Parachor	[P] 20°C		ĺ
30	1.4	390	4	t _c ml/g	455.	5		30		l
"C"	0.74	482	4	P _c mm	9252.	5		40 Sugd.	686.2	5
MR (Obs.)	80.5		2	PV/RT	1,200	\vdash	Exp. L.l.			<u> </u>
MR (Calc.) (nD-d/2)	80.2		5 2	25°C	1.0000	5	u.	,0, wc.		
Dielectric	1.0	-	-	30 mm BP	1.0000 0.9250	5 5	Dispersion	n.	107.	2
A 186 to	6.9	20	2	t _e	0.8995	5	Flash Poin			i
B 1366 °C	1774.6	20	2	tc			Fire Point	•		
с	139.7		2	∆Hc kcal/m	2517.30	2	M Spec. Ultra V.			
A* 186 to B*, 340 °C		9829	5 5	ΔHf ΔFf			X-Ray Dif	•		
B* 340 °C	1695.7		٦	Viscosity	<u> </u>	\Box	Infrared			<u> </u>
·				centistokes	1		Solubility Acetone	in +		
t _x to t _x *C	ļ			7 40 °C	3.08	2 2	Carbon to	et.		
A' to	ļ		\vdash	80	1.66	2	Benzene Ether			1
B'	:			100	1,30	2	n-Heptan	е	1	
C'				B ^V 35 to A ^V 70 °C	783.1	5	Ethanol			
A'* to B'* °C				<u> </u>	3.98823 699.7	5	Water Water in			
Ac 366 to	8.8	342	5	(B ^V) 70 to		1 1				
Bc t °C	4239.	- -	5	c _p liq. °K	2.23908	5				
Cryos. A°	416.		5			2				
consts. B°				c _p vap.300°K 400	0.38672	2				
t _e °C	335.1	1	5	c _v vap.						
$T_R = 0.88$							f. grams/l	00 grai	ms solven	<u>t</u>
REFERENC	ES: 1-D	ow			Calc. from de	t. da	ta 5-Calc.	by for	mula	
SOURCE:			AF	*						
PURIFICAT		op me	AF							
LITERATUE	CE KEFI	EKEN	ICES	:						

No. 167 1-Octadecene NAME STRUCTURAL FORMULA CH3(CH2)15CH=CH2 Molecular C18H36 Mole Molecular Weight 252.468 % Pur. Ref. Ref. Ref. F. P. °C 17.6 2 dt/dP f to F.P. 100% °C/mm g <u>°K</u> B. P. °C h BP 0.0633 760 mm 314.2 2 t_e 0.03596 5 ſ١ 2 to 100 234.2 198.91 g' <u>• K</u> 4 30 30 mm 0.8805 5 172. 2 10 h١ ∆Hm cal/g 1 128. 5 0.0230 300 m to ∆Hv cal/g Pressure 600 °K 0.0014 n 25°C mm 93.77 °C 0.1000 o -0.0650 4 30 mm 66.44 5 1575. 5 te BP 52,04 5 m' 700 to 0.1164 4 Density te (d, e) 48.18 5 1000 °K 0.0011 n' g/ml 20°C 0.7888 2 47.31 5 ٥' -0.0638 4 25 0.7853 dt4 2 AHv/Te 19.45 5 30 0.7818 4 Surface tension ₹ 199 to 91.20 5 0.8028 dynes/cm. 20°C 26.36 e 360 a 1 5 °C 0.1246 ь -0.0370 4 25.43 30 to 40 24.53 5 Ref. Index e' °C ⁿD 1.4449 2 [P] 20°C Parachor d_c g/ml 25 1.4428 2 20°C vc ml/g t_°C 30 1.4408 4 30 ^tc 466. 5 40 "C" 0.7476 4 P_c mm 8609. 5 Sugd. 725.2 5 MR (Obs.) 85.185 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 84.851 5 1.0000 5 25°C (nD-d/2)1.0506 2 u. 30 mm 1.0000 Dispersion 107. 2 Dielectric BP 0.9303 5 Flash Point C 0.9053 A 199 to 6.901 2 Fire Point B 1380 °C 1789.4 2 M. Spec. С 131. 2 AHc kcal/m 2664.22 2 Ultra V. ΔHf A* 199 to 1.58563 5 X-Ray Dif. ΔFf B* 361 °C 1707.4 Infrared ĸ Viscosity Solubility in centistoke Acetone to 40 °C 3.59 Carbon tet. °C t^ 2 60 2.51 Benzene 1.87 80 2 A'I to Ether 100 1.46 B' °C n-Heptane B^V | 35 to A^V | 70 °C 810.4 5 Ethanol AI* 3.96761 Water to (BV) 70 to Water in B'* °C 5 560.8 Acl 380 to 9.04000 5 (A^V)| 105 °C 2.69007 5 Bc tc °C 4643. c_p liq. Cc 5 443.1 c_p vap.300K Cryos. Aº 0.38702 2 consts. Bº 400 0.48846 2 c_v vap. te °C 352, 14 $T_R = 0.88 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

								No. 168	
NAME	l-Nona	lecen	e				STRUCTURAL		
I						\neg			
ļ						\neg	СH ₃ (СH ₂) ₁₆	CH=CH ₂	
Mole % Pur.	Ref.		ecul mul		Molecular Weight 266.4	.94			
			Ref.	1 1/ 30 1		Ref			Ref.
F. P. °C	23.4		2	dt/dP	1		f to	1	
F.P. 100%				*C/mm	l		f to g °K	ł	ļ
B. P. °C	1			106.43 °C BP	125.89	5 2	h	ł	l
760 mm 100	328.0 247.		2 2	t	0.03604	5	f' to		i
30	211.15	.	4	30 mm	0.8925	5	g' K	į	
10 1	184.		2 5	ΔHm cal/g			h'	ļ <u>.</u>	<u> </u>
Pressure	-	\dashv	\neg	ΔHv cal/g			m 300 to n 600 °K	0.0230	
mm 106.43°			5	25°C 30 mm	86.30 65.33	5. 5	0 000 1	-0.0650	
t _e	1606.06	<u>'</u>	5	BP	50.79	5	m' 700 to	 	
Density g/ml 20°C	0.79	20#	2	te te (d, e)	46.63 45.83	5	n' 1000 °K	0.1187	
at 25	0.78	86	2	ΔHv/T _e	19.38	5	o' ¦	-0.0638	
	0.78		4	d 211 to		5	Surface tension		
a b	0.80		4 4	<u> </u>		5	dynes/cm. 20°C	26.60 25.70	5
Ref. Index	1 3,53	-	-	d' to			40	24.83	5
n _D 20°C	1.44		2			\vdash	Parachor [P]		
25 30	1.44		2 4	d g/ml v ml/g	ł		20°C 30	1	l
"C"	0.74		4	16 °C	477.	5	40	1	
MR (Obs.)	89.83		2	P _c mm	8079.4	5		764.2	5
MR (Calc.)	89 47	5	5	PV/RT 106.43 °C	1.0000	5	Exp. L.1.%/wt. u.	1	İ
(nD-d/2)	1.05	06"	2	30 mm	1.0000	5	Dispersion	106.	2
Dielectric	ļ			BP	0.9300	5	Flash Point °C		
A 211 to B 394 °C		1	2 2	te t _c	0.7005		Fire Point		
c	122.		2	ΔHc kcal/m	2811.14	2	M Spec. Ultra V.	İ	
A* 211 to	1.58		5	ΔHf ΔFf			X-Ray Dif.	İ	
B* _L 37 <u>7</u> °C	1720.93	1	5	Viscosity	 	\vdash	Infrared	ļ	<u> </u>
۔	4	l		centistokes			Solubility in + Acetone		ŀ
t _k to		1		7 40 ℃	4.17 2.87	2 2	Carbon tet.	İ	
A' to	 			80	2.12	2	Benzene Ether		
B' ∟ _ °C	.]	l		B ^V 35 to	1.63	2	n-Heptane		1
C' to	<u> </u>			B' 35 to A' 70 °C	846.6 3.91708	4	Ethanol Water	1	
B'* °C				(BV) 70 to	752.3	4	Water in		
Ac 394 to	9. 29		5	(AV) 105 °C	2.19638	4			
Bc tc_°C	5158.13		5	c _p liq. °K	 				
Cryos. A°	479.41			l ⁻	0 20725			1	
consts. B°		i		c _p vap.300°K	0.38725 0.48879	2 2		1	
t _e °C	367.84	Ť	5	c _w vap.					
$T_{\mathbf{R}} = 0.8$			unde	rcooled liquid	<u> </u>		grams/100 gra	ms solven	<u></u>
					Calc. from det	da:	ta 5-Calc. by for	mula	
SOURCE:			AP						
PURIFICAT	ION:		AP						
LITERATUI	RE REFI	EREN							
1									

No. 169 l-Eicosene NAME STRUCTURAL FORMULA $CH_3(CH_2)_{17}CH=CH_2$ Mole Molecular Molecular C20H40 Weight 280.520 % Pur. Formula Ref. Ref. F.P. °C F.P. 100% 28.6 2 dt/dP f to °C/mm g °<u>K</u> 117.0°C 127.12 5 B. P. *C h BP 0.06527 4 760 mm 341,2 2 te 0.03634 5 ſ١ 100 to 259. 2 g' <u>°К</u> 30 222.92 4 0.9036 5 30 mm 10 195. 2 h' ∆Hm cal/g 1 149. 300 to 0.0230 m AHv cal/g Pressure n 600 °K 0.0014 25°C 86.27 5 mm 117.0°C 0.1000 o -0.0650 4 30 mm 64.32 5 te 5 1622. ΒP 49.34 5 700 to m 0.1166 4 Density 5 44.87 te (d, e) n' 1000 °K 0.0011 g/ml 20°C 0.7950 2 44.12 0.7916 0.7882 ۰' -0.0638 4 25 $\mathbf{d_{4}^{t}}$ 2 AHv/T 19.19 5 30 4 Surface tension 223 to 92.55 5 0.8086 a dynes/cm, 20°C 26.85 <u> 1 383</u> 5 •C 0.1266 h -0.0368 30 25.94 ď٦ to 40 25.06 Ref. Index 5 e' •c ⁿD 20°C 1.4480[‡] 1.4459[‡] [P] Parachor dc g/ml 25 2 20°C vc ml/g t °C 1.4439[‡] 30 4 30 486. 5 t_c 40 "C" 0.7466 4 5 7425. P_c mm Sugd. 803.2 5 94.47[‡] MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 94.093 25°C 1,0000 5 1.0506 (nD-d/2)2 106.[≠] 30 mm 1.0000 ž Dispersion Dielectric BP 0.9250 5 Flash Point C 5 A 223 to 6.859 te t 0.8894 2 Fire Point B 405 °C 1807.9 2 M. Spec. Ċ 113. 2 AHc kcal/m 2958.06 Ultra V. ΔHf A* 223 to 1.59840 5 X-Ray Dif. ΔFf B*[393 °C 1734.6 Infrared ĸ Viscosity Solubility in centistokes Acetone to 40 °C 4.81 2 °C Carbon tet. 60 3.27 2 Benzene 80 2.38 2 Α' to Ether 100 1.81 2 B <u>•с</u> n-Heptane B^V | 35 A^V | 70 C' 5 874.5 to Ethanol 3.89001 5 A'* °C Water to (BV) 70 Water in B'* °C 5 783.6 to Ac 405 to 9.50720 (A^V)| 105 Z. 15801 5 5 °C 5591. Bc tc C c_p liq. Cc 5 505. c_p vap 300°K 0.38749 Cryos, Aº 2 consts. B 400 0.48909 c_v vap. te °C 5 382.48 $T_{R} = 0.89T$ for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: PURIFICATION: LITERATURE REFERENCES:

										No. 170)
NAME		1 - F	lene	icose	ne			STRUCTU	RAL I	FORMUL	A.
								сн ₃ (сі	1.) CI	H=CH_	
Mole % Pur.	R	lef.	Mo Fo	lecul	ar C ₂₁ H ₄₂	Molecular Weight 294,5	46	03(0.	-2/18	2	
				Ref.			Ref				Ref
F. P. *C	33	3.3		2	dt/dP	T		f	to		
F.P. 1007					*C/mm			g	°K		1
B.P. °C		_			0.1 mm BP	146.20	5 4	ъ ;			1
760 mm 100	355	5. 1.27		5	te	0.0354	5	<u>-</u>	to		
30	233	3.06		5	30 mm	0.9666	5	g'	°K		1
10 1		3.53 3.7 5		5	ΔHm cal/g			h'			
0.1		5.62		5	ΔHv cal/g			m (to		1
Press. mm				\vdash	0.1 mm	69.75	5	n	°К		
t _e	1602	2.		5	30 mm BP	59.60 47.98	5 5	<u> </u>			
Density	. 🗆	. 70	#		l t	44.61	5	m' n'	to °K		1
g/ml 20°C	, ,	0.79 0.79	43 [‡]	2 2	t _e (d, e)	44.22	5	6 !			1
d ₄ 25	6	7.79	09 [‡]	4	ΔHv/T _e	19.39	5	Sunface to			┼
a		81	13	4	d 233 to		5	Surface te dynes/cm.		27.06	5
ь	$\overline{}$	0.03	68	4	d' 393 - °C		5) y	30	26.15	5
Ref. Index	. 1 1	1.44	0.4 ‡	١, ١	e' 233 °C		5		40	25.27	5
ⁿ D 25	1 1	44	マット	2 2	d _c g/ml			Parachor	[P] 20°C		Ì
30	1	. 44	5 2 ₹	4	vc ml/g tc °C	1	1 1		30		1
"C"		. 74		4	P _c mm		1 1		40 Sugd.	842.2	5
MR (Obs.)		.09		4	PV/RT	 	\vdash	Exp. L.1.		012.2	13
MR (Calc. $(nD-d/2)$) 98	3.71 .05	1 05≠	5 2	0.1 mm	1.0000	5	u.	,0, w.c.	,	1
Dielectric		2.10		5	30 mm BP	1.0000	5	Dispersion		105.≠	2
A 233 to		. 54		4	t _e	0.8948 0.8627	5	Flash Point			
B 443°C	2457	. 3		4	t _c	<u> </u>					╁
С	172	<u> </u>		5	ΔHc kcal/m			M Spec. Ultra V.			
A* 2 33 to B* 403 °C		31	717	5	ΔFf			X-Ray Dif	•		1
K Lion		. 0			Viscosity			Infrared	in +	l	-
t	_				centistokes	1		Solubility Acetone	ın T		Į
t _x t ₀					η •ο			Carbon to	et.		
A' to				1	1			Benzene Ether			
B' •	<u>:</u>]	B ^V to	 	1	n-Heptan	e]
C'	+			1	B' to	i		Ethanol Water			i
A'* 115 to B'* 233 *(. 23	287	5	(BV)	-		Water in			
Ac to		<u> </u>		H	(A ^V)						
Bcit 0				1	c _p liq. °K	+	+				
	1-			\vdash	_	1					
Cryos, A° consts. B°				i	c _p vap. °K	1					Ì
t _e °C	394	. 33		5	c _v vap.						
≠ for und								f grams/l	00 grai	ns solver	ıt
REFEREN	CES:	1 - Do	w	2-AF	PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc.	by for	mula	
SOURCE:				API							
PURIFICA				API							
LITERATU	RE R	EFE	CREI	NCES	::						

								N o. 17	1
NAME	1-Docose	ne				STRU	CTURAL	FORMUI	-A
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 308.5	72	СН3	(СН ₂) ₁₉ СІ	H=CH ₂	
		Ref		T	Ref.				Ref
F. P. °C	37.8	2	dt/dP			f	to		
F.P. 100%			°C/mm			gl	°K		1
B. P. °C	2/2	١.	0,1 mm BP	148.82 0.0656	5 4	h			
760 mm 100	367. 281.87	2 5	t _e	0.0354	5	f'	to		
30	243.02	5	30 mm	0.9831	5	g'	°K		
10 1	212.98 162.33	5	ΔHm cal/g			h'			
0.1	123.52	5	ΔHv cal/g			m	to		1
Press. mm			0.1 mm 30 mm	68.09	5	n	°K		ļ
t _e	1632.3	5	BP	58.15 47.39	5	m'			+-
Density g/ml 20°C	0.8002	2	te (d.e)	43.34	5	n' i	to °K		ŀ
dt 25		2	'e (-, c,	43.0	5	0'			1
4 30	0.7934	4	ΔHv/T _e	19.63	1	Surfac	e tension		+-
a L	0.8138	4	d 243 to e 408 °C	80.49 0.0919	5 5		cm. 20°C	27.27	5
b Def 7.1	-0.0368	*	d' 122 to	78.37	5	8	30 40	26.35 25.46	5
Ref. Index) , ,=o=F	2	e' 243 °C	0.0832	5	Paracl		23.40	13
25		2	d g/ml			Faraci	20°C		
30	1, 1104	4	vc ml/g tc°C			ļ	30 40		
"C"	0.7457	4	P _c mm					881.2	5
MR (Obs.) MR (Calc.)	103 329	2 5	PV/RT			Exp. I	J. 1. %/wt.		1
(nD-d/2)	1.0505	2	0.1 mm 30 mm	1.0000 1.0000	5			105.≠	2
Dielectric	2.10	5	BP	0.8943	5	Disper		105.	<u> -</u>
A 243 to	7.55128	4	t _e	0.8612	5	Flash Fire P	Point °C		
B 458 °C	2505.7 169.5	4 5	t _c ΔHc kcal/m		1	M. Spe			+
A* 243 to	2.33832	5	ΔHC RCa1/H	1		Ultra '	٧.		
B*[418 °C		5	ΔFf			X-Ray Infrare			
K — — —		1	Viscosity	1		Solubil			+
t _k – tō	·	l	rentistokes 7°C			Aceto	ne		1
tx C		1	'			Benze	on tet.	Ì	1
A' to B' °C					1	Ether	•		1
G'	-	ĺ	B ^V to			n-Hej Ethan		,	
A** 122 to	2,25251	5	AV I ℃			Wate	r		
B'* 243 °C	2390, 03	5	(B ^V)	-[Water	r in		
Acl to			(A ^V)[İ	
Bc tc °C	-	1	c _p liq. °K						
Cryos. A°	 	 	-11	İ		1		1	1
consts. B			P -						
t _e °C	408.05	5	c _v vap.					<u> </u>	\perp
# for unde	rcooled liquid					† gran	s/100 gra	ms solve	nt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da	ta 5-Ca	lc. by for	mula	
SOURCE:		API							
PURIFICAT	TION:	API							
LITERATU	RE REFERE	NCE	S:						

							No. 172	
NAME	1-Tricos	ene				STRUCTURAL 1	FORMULA	
14.1.	2 / 1/					CH ₃ (CH ₂) ₂₀ C	H=CH,	
Mole % Pur.	Ref. Mo	rmu]	ar C ₂₃ H ₄₆	Molecular Weight 322.	598	3 2 20	•	
		Ref.		T	Ref.			Ref.
F.P. *C	41.6	2	dt/dP			f to		
F. P. 1009	6		°C/mm		ا ۔ ا	g °K		
B.P. °C	270	,	0.1 mm BP	151.55 0.0667	5 4	h		
760 mm 100	379. 292.47	5	t.	0.0355	5	f' to		ŀ
30	252.95	5	30 mm	1.0001	5	g' °K		
10	222.39 170.83	5	ΔHm cal/g			h'		
0. 1	131.32	5	ΔHv cal/g			m to		ŀ
Press. mn		_	0.1 mm 30 mm	66.50 56.80	5	ö		1
t _e	1662.0	5	BP	45.63	5	m' to		
Density g/ml 20°0	0.8023	2	te te (d, e)	42.20	5	n' 'K		
dt 25	0.79905	2	ΔHv/Te	41.85 19.59	5	0'		
	0.7958	4	d 254 to		5	Surface tension		
a b	0.8156 -0.0 ₃ 66	4	e 419 °C		5	dynes/cm. 20°C	27.44 26.55	
Ref. Index		┝∸	d' 132 to		5	40	25.68	
n _D 20°0	~ 1 4 5 1 7 7	2		0.0797	1	Parachor [P]		
25	1.4516, 1.4496, 1.4475	2	d g/ml vc ml/g		i	20°C		
"C"	0,7453	4	I'c C			40		l
MR (Obs.		2	P _c mm			Sugd.	920.2	5
MR (Calc.	1 107 047	5	PV/RT	1 0000	_	Exp. L.1.%/wt.		
(nD-d/2)	1.0505*	2	0.1 mm 30 mm	1.0000	5	u. Dispersion	105.≠	2
Dielectric		5	BP t _e	0.8936 0.8596	5	Flash Point °C		
A 254 t		4	tc	0.0370		Fire Point		
c	167.5	5	ΔHc kcal/m			M Spec. Ultra V.	1	
A* 254 to			ΔHf ΔFf			X-Ray Dif.		
B* 429 °C	2477.9	5	Viscosity	+	 	Infrared		
\$.— —.	_		centistokes			Solubility in + Acetone		ĺ
t _k t			η ° ℃	•		Carbon tet.		
A' i to		-				Benzene Ether		
B' º	의	1	B ^V to	<u> </u>	-	n-Heptane		
C'	2 2252	-	B' to			Ethanol Water		1
A'* 132 to B'* 254 °C		5	(BV)	-		Water in		<u> </u>
Ac te		<u> </u>	(A ^V)					
Bc tc_'	의	ĺ	c _p liq. °K					
Cc	+		11 -					
Cryos. A'			c _p vap. °K					İ
t _e °C	421.77	5	c _v vap.					
# for und	ercooled liquid			•		+ grams/100 gran	ns solven	t
	CES: 1-Dow	2-A1	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		AP	I					
PURIFICA		AP						
LITERATU	RE REFERE	1CES	S:					
l								
1								
i .								

								No. 17	3
NAME _	l-Tetra	osen	e			ST	RUCTURAL	FORMUL	.A.
						_	н (сн.) сч	і=СН	ł
Mole % Pur.	Ref. Mo	lecula mula	C ₂₄ H ₄₈	Molecular Weight 336.	624		:н ₃ (сн ₂) ₂₁ сн	2	
	_	Ref.			Ref.				Ref.
F.P. °C	45.3	2	dt/dP			f	to		
F.P. 100%		\vdash	*C/mm 0.1 mm	153.90	5	g	°K		
B.P. °C 760 mm	390.	2	BP	0. 0 677	4	_h _	<u> </u>		
100	302.20	5	te	0.0355 1.0151	5	f' g'	to °K		
30 10	262.09 231.06	5	30 mm	1.0151	-	h'	•		1
1	178.73	5	ΔHm cal/g	 		m	to		1
0,1 Press.mm	138, 61	5	ΔHv cal/g 0,1 mm	65.04	5	n	*K		
t e	1690.	5	30 mm BP	55.51	5	°			
Density			t.	44.52 41.05	5	m'	to		
g/ml 20°C	0.8045	2	·e (4, 6)	40.70	5	n' o'	*K		
dt 25 4 30	0.8011 0.7977	2 4	ΔHv/T _e	19.69	5		(n n n n n n n n n n n n n n n n n n n		+
a	70,8181	4	d 263 to e 431 °C	78.01	5		face tension es/cm. 20°C	27.62	5
ь	-0.0 ₃ 68	4	d' 140 to	0.0859 75.73	5	8	30 40	26.69	5
Ref. Index n _D 20°C	1.4527	2	e' 263 °C	0.0772	5	Par	achor [P]	25.80	+3
25	1.4506 1.4485	2	d _c g/ml				20°C		
30		4	vc ml/g tc °C				30 40		
"C"	0.7449	4	P _c mm				Sugd.	959.2	5
MR (Obs.) MR (Calc.	113.03	2 5	PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	1.0505	2	0.1 mm 30 mm	1.0000	5	Dis	u. persion	104.≠	2
Dielectric	2.11	5	BP	0.8933	5		sh Point °C		+
A 263 to B 481 °C	7.56767 2601.2	4	te tc	0.8585	٦	Fir	e Point		
c	165.	5	ΔHc kcal/m	 	<u> </u>		Spec. ra V.		
A* 263 to	2.38017	5	ΔHf ΔFf			X-I	Ray Dif.		
B*[441_°C K	2519.5	5	Viscosity				ared		↓
t ₁ – to	.		centistokes				ubility in ^T eton e		1
t _k to t _x °C			η °C			Ca	rbon tet.		
A' to		\vdash					nzene her		
B'°C	-		B _v to				Heptane hanol		
A'* 140 to	2.29129	5	A I C			W.	ater		
B'* 263 °C	2481.6	5	(B ^V)			W	ater in	ļ	+-
Ac to			(A ^V)						
Bc tc °C	-		c _p liq. *K						
Cryos, A° consts, B°			c _p vap. 'K						
te °C	434.4	5	c vap.						
	rcooled liquid			•		+ g1	rams/100 gra	ms solve	nt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:		API							
PURIFICAT	TION:	API							
LITERATU	RE REFERE	NCES	::						

							No. 174	
NAME	1-Pentacosene					STRUCTURAL FORMULA		
						СН ₃ (СН ₂) ₂₂ СН	I=CH.	
Mole	Ref. Mo	lecul	lar C ₂₅ H ₅₀	Molecular		3,011,2,2201	2	
% Pur.		Ref.		Weight 350.6	Ref			Ref.
F.P. *C	48.7	2	1	T	Kei		T	Kei.
F.P. 100%	10.1	ᡰᢆ	dt/dP *C/mm	ļ		f to g 'K		1
B.P. °C	 	 	0.1 mm	156.49	5	h .	İ	Ì
760 mm	401.	2	BP	0.0686 0.0355	4 5	$\left \frac{\ddot{\mathbf{f}}}{\mathbf{f}'} + \frac{\ddot{\mathbf{b}}}{\mathbf{t}_0} \right $		
100 30	311.91 271.18	5	t _e 30 mm	1.0309	5	g' °K	Ì	į
10	239.66	5	ΔHm cal/g	1	+-	h'	ĺ	
1 0, 1	186.48 145.70	5	ΔHv cal/g		\vdash	m to		
Press. mm		 ' -	0.1 mm	63.53	5	n °K		
t _e	1717.3	5	30 mm	54.27	5 5	<u> </u>		
Density	4	1	BP t_	43.51 40.02	5	m' to		
g/ml 20°C	0.8063 0.8030	2 2	t _e (d, e)	39.69	5	n' K	1	
dt 25 4 30	0.7998	4	ΔHv/T _e	19.48	5	<u> </u>		<u> </u>
	0.8196	4	d 272 to		5	Surface tension dynes/cm, 20°C	27,77	5
b	-0.0366	4	e 443 °C		5	30	26.87	5
Ref. Index			e' 272 °C		5	40	26.00	5
n _D 20°C	1.4536 1.4515	2 2	d g/ml			Parachor [P]	}	1
30	1.4495	4	d g/ml vc ml/g tc °C	İ		20°C 30	ļ	
"C"	0.7447	4	11 -			40		
MR (Obs.)	117.68	2	P _c mm			Sugd.	998.2	5
MR (Calc.)	117.183 ₄	5	PV/RT 0.1 mm	1.0000	5	Exp. L.1.%/wt.		}
(nD-d/2)	1.0505		30 mm	1.0000	5	Dispersion	104.	2
Dielectric	2.11	5	BP	0.8927	5	Flash Point °C		T
A 272 to B 493 °C		4	te t _c	0.8571	5	Fire Point		
č	163.5	5	ΔHc kcal/m		\vdash	M Spec.		
A* 272 to	2.40531	5	ΔHf			Ultra V. X-Ray Dif.		
B* 453 °C	2570.2	5	ΔFf			Infrared		1
c V			Viscosity centistokes	}		Solubility in +		1
tk to			7 °C	: [Acetone Carbon tet.		1
*x 1		_		1		Benzene		1
A' to B' °C		1				Ether n-Heptane		1
_c,		1	B ^V to			Ethanol		ļ
A'* 147 to	2.31448		A ^V O			Water Water in		
B'* 272 °C	2531.2	5	(B ^V)			Water III		†
Ac to		1	(A ^V)	<u> </u>			1	1
Cc L-c-		1	c _p liq. ∘K	:				ì
Cryos, A° consts, B°			c _p vap. °K	:				
t _e °C	446.99	5	c _v vap.					
<pre># for unde:</pre>	cooled liquid	Ĺ				grams/100 gran	ms solven	t
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da			
SOURCE:		AP					_	
PURIFICAT	ION:	AP	I					
LITERATU	RE REFERE	NCES	5:					

							No. 17	5	
NAME	l-Hexacosene					STRUCTURAL FORMULA			
l [
					\dashv	CH ₃ (CH ₂) ₂₃ (CH=CH ₂		
Mole % Pur.	Ref. Mol	lecul	ar C ₂₆ H ₅₂	Molecular Veight 364.6	74				
	1 1 1 1 1 1 1 1 1	Ref.		veight 304.0	Ref.	r		Ref.	
F. P. °C	51.8	2		 	Ver.	<u> </u>	r	Kei.	
F.P. 1007		ا ا	dt/dP *C/mm			f to g %			
B.P. °C			0.1 mm	158.69	5	h i	1		
760 mm	411.	2	BP t _e	0.0695 0.0356	5	f' to	1		
100 30	320.74 279.47	5 5	30 mm	1.0448	5	g' °K		1	
10	247.53	5	ΔHm cal/g			h'			
0.1	193.62 152.26	5 5	ΔHv cal/g		$\vdash \vdash$	m to			
Press. mn		5	0.1 mm	62.15	5	n oK			
t e	1742.7	5	30 mm BP	53.07 42.50	5 5			<u> </u>	
Density			t _e (d, e)	38.98	5	m' to rK			
g/ml 20°0	0.8082 0.8048	2		38.66	5	o' .			
dt 25 4 30	0.8048 [‡] 0.8014 [‡]	4	ΔHv/T _e	19.43	5	Sumfa = a Association	ļ	 	
a	0.8218	4	d 280 to e 454 °C	75.52	5	Surface tension dynes/cm. 20°C	27.92	5	
ь	-0.0368	4	d' 154 °C	0.0804 73.02	5 5	8 30	26.99	5	
Ref. Index	- 1 45457	2	e' 280 °C	0.0714	5	40	26.09	5	
ⁿ D 20°C	1.4545 1.45247 1.4503 [#]	2	d _c g/ml			Parachor [P] 20°C	ì		
30	1.4503	4	vc ml/g tc °C			30			
"C"	0.7443	4	P _c mm			40 Sugd	1037.2	5	
MR (Obs.)		2	PV/RT		\vdash	Exp. L.1.%/wt.		H	
MR (Calc. (nD-d/2)	1.0504 [#]	5 2	0.1 mm	1.0000	5	u.	4		
Dielectric			30 mm BP	1.0000 0.8924	5	Dispersion	104. ≠	2	
A 280 to		4	t	0.8561	5	Flash Point °C Fire Point			
B 504 °C		4	-c			M. Spec.		├	
C	161.5	5	ΔHc kcal/m ΔHf			Ultra V.		1	
A* 280 to B* 464 °C		5	ΔFf			X-Ray Dif. Infrared			
к — — -			Viscosity			Solubility in +		+	
t _k [-tō	-		centistokes			Acetone			
t _x °C			ຸກ °⊂			Carbon tet.			
A' to						Benzene Ether	İ		
B' '°	2		B ^V to		+-+	n-Heptane	1		
A'* 154 to	2,33132	5	B' to A' C			Ethanol Water			
B'* 280 °C		5	(B ^V)			Water in		<u></u>	
Acl to			(A ^V)						
Bc tc °C	<u> </u>		c liq. °K		\Box				
	. +	<u> </u>	[]	1					
Cryos, Accounts, Bo			c _p vap. °K						
t _e °C	458.48	5	c _w vap.	1					
	rcooled liquid	L	и		لـــــا	grams/100 gra	ms solver	ıt.	
	CES: 1-Dow		PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc, by for			
SOURCE:		AP							
PURIFICA	TION:								
PURIFICATION: API LITERATURE REFERENCES:									
	NOT BRE								

							No. 176	
NAME	l-Heptacosene					STRUCTURAL :	FORMULA	
					g., (g.,) g., g.,			
Mole	Ref. Mo		1	Molecular		CH ₃ (CH ₂) ₂₄ (CH=CH ₂	
% Pur.	Fo Fo	rmul		Weight 378.7	02			
	_	Ref.			Ref.			Ref.
F.P. C F.P. 100%	54.7	2	dt/dP			f to		
B. P. °C	<u> </u>	-	*C/mm 0.1 mm	160.89	5	g °K		
760 mm	421.	2	BP t	0.0704	4			
100 30	329.58 287.77	5	t _e 30 mm	0.0356 1.0587	5	g' to		1
10	255.40	5	ΔHm cal/g		1	h'		
1 0.1	200.76 158.83	5 5	ΔHv cal/g			m to		
Press. mm			0.1 mm 30 mm	60.86	5	n °K		
t _e	1767.5	5	BP	51.96 41.57	5 5	m' to		\vdash
Density g/ml 20°C	0.8097	2	te te (d, e)	38.04 37.74	5	n' 'K		
dt 25 4 30	0.8064 0.8032	2	ΔHv/T	19.11	5	o' ¦		
a 30	0.8032	4	d 289 to	74.39	5	Surface tension dynes/cm. 20°C	39.05	_
ь	-0.0366	4	e 465 °C d' 161 to		5 5	30 gynes/cm. 20-C	28.05 27.14	5
Ref. Index	1 45537	2	e' 289 °C		5	40	26.26	5
D 25	1 1 4532	,	d _c g/ml v _c ml/g			Parachor [P] 20°C		
30	1,4513		tc °C			30 40		
"C"	0.7441	4	P _c mm	ļ		!]	1076.2	5
MR (Obs.) MR (Calc.	1 126 210	2 5	PV/RT		T .	Exp. L.1.%/wt.		
(nD-d/2)	1.0504	2	0.1 mm 30 mm	1.000 0 1.00 00	5 5	u. Dispersion	104.≠	2
Dielectric	2.12	5	BP t _e	0.8919 0.8549	5	Flash Point °C		
A 289 to B <u> 515 °C</u>		4	tc	0.031)		Fire Point		
С	159.5	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		
A* 289 to B* 475 °C		5	ΔFf			X-Ray Dif. Infrared		
к — — –	1		Viscosity			Solubility in +	 	├
t _k			rentistokes °C			Acetone		1
'x			∄ '			Carbon tet. Benzene		ļ
A' to B' °C						Ether n-Heptane		
C'	<u> </u>		B ^V to A ^V C			Ethanol		
A'*161 to B'*289 °C		5 5	$\frac{A^{\mathbf{v}}}{(B^{\mathbf{v}})} - \frac{{}^{\mathbf{c}}}{-}$	-[Water Water in		
Ac to	+	-	(A ^V) ₁					
Bc tc_C	<u>:</u>		c _p liq. °K	 	 			1
Cryos, A*	 		11					İ
consts. B°			P					
t _e °C	469.86	5	c _v vap.	I				
	rcooled liquid					grams/100 gra		<u>t</u>
SOURCE:	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
PURIFICAT	rion.	AP AP						
	RE REFEREN							

No. 177 1-Octacosene NAME STRUCTURAL FORMULA CH3(CH2)25CH=CH2 Mole Ref. Molecular Molecular $C_{28}H_{56}$ % Pur Weight 392.728 Formula Ref. Ref. F.P. °C F.P. 100% 57.5 2 dt/dP f to °C/mm ۰ĸ g 0.1 mm 163.06 B. P. 'C h BP 0.0712 2 760 mm 430. t_e 0.0356 5 ſ١ to 100 337.52 5 5 5 g¹ ۰ĸ 30 mm 1.0718 5 295.19 30 h' 10 262.42 ∆Hm cal/g 1 207.07 5 to AHv cal/g 0.1 164.59 n °К 0.1 mm 30 mm 59.46 5 Press. mm o 1789.2 5 50.81 t_e BP 40.61 5 m to Density 5 te (d, e) 37.06 n' ۰ĸ 0.8114 g/ml 20°C 36.82 5 0.8080 0.8046 o' 25 $\mathbf{d_{4}^{t}}$ AHv/Te 19.32 5 30 4 1297 Surface tension d to 73.13 0.0756 0.8250 dynes/cm. 20°C 28, 18 5 1475 ٠<u>c</u> 475 ما 168 ما -0.0368 4 ь 30 27.25 to 70.37 0.0663 26.34 40 Ref. Index e' | 297 1.4560 1.4540 20°C 2 [P] $\mathbf{n}_{\mathbf{D}}$ Parachor d_c g/ml 25 2 20°C 1.4519 vc ml/g 30 30 ťč °C 0.7438 40 "C" 4 P_c mm 1115.2 5 Sugd MR (Obs.) 131.57 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 131.337 5 1.0504 0.1 mm 1.0000 (nD-d/2)103.[≠] 30 mm 1.0000 ż Dispersion Dielectric 2.12 5 ВP 0.8912 5 Flash Point °C A 297 to t_e 0.8535 5 7.60499 4 Fire Point 2780.18 B _525 °C 4 M. Spec. C 158.5 AHc kcal/m Ultra V ΔHf 2.46441 A* 297 to 5 X-Ray Dif. B* 485 °C ΔFf 2 696, 18 Infrared ĸ Viscosity Solubility in centistokes Acetone tō °C Carbon tet. °C Benzene Ā to Ether B١ °C n-Heptane B_v | C' to Ethanol °C Water 2.36904 2652,91 168 to 5 Water in (BV)I B'* 297 °C Acl (A^V)| to Bc °C c_p liq. Cc Cryos. A° consts. B° °K c_p vap. c vap. te °C 480,28 5 ≠ for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

								No. 178	
NAME	1-Nonacosene						STRUCTURAL FORMULA		
							СН ₃ (СН ₂) ₂₆ С	H=CH,	
Mole % Pur.	Ref.	Mo:	lecul rmul	ar C ₂₉ H ₅₈	Molecular Weight 406.7	54	3 2 20	2	
			Ref.			Ref			Ref.
F. P. *C	60.0		2	dt/dP	T		f to		
F.P. 100%	1			*C/mm			g °K		
B. P. °C				0.1 mm BP	165.14 0.0721	5 4	h ¦		1
760 mm 100	440. 346.3	7	5	t _e	0.0357	5	f' + to		
30	303.5	1	5	30 mm	1.0853	5	g' K		Í
10 1	270.3		5	ΔHm cal/g			h'		
0. 1	171.2		5	AHv cal/g	1		m to		
Press. mm				0.1 mm	58.43	5	n °K		
t _e	1814.7		5	30 mm BP	49.87 39.82	5	<u> </u>		L
Density		, , , , #		t_	36.26	5	m' to	ļ	ļ
g/ml 20°C	1 08	127 ‡ 094 ‡	2 2	e (=, =,	36.0	5	o' l		1
dt 25 4 30	0.8	062	4	ΔHv/T _e	19.28	5			-
a	0.8		4	d 305 to		5	Surface tension dynes/cm. 20°C	28,30	5
Ъ	-0.0	366	4	d' 485 °C		5	y 30	27.39	5
Ref. Index	1	-/-#		e' 305 °C		5	40	26.50	5
n _D 20°C	1.4	567 547 ±	2 2	d g/ml			Parachor [P] 20°C	1	İ
30	1.4	526 [‡]	4	∥ v _c mı/g			30		
"C"	0.7	436	4	H C			40	1154 2	5
MR (Obs.)	136.2	3	2	P _c mm	<u> </u>			1154.2	-
MR (Calc.) (nD-d/2)	135.9	55 504 [≠]	5	0.1 mm	1.0000	5	Exp. L.1.%/wt. u.	١.	
Dielectric			2	-30 mm	1.0000	5	Dispersion	103.	2
	2.17		5	BP t _e	0.8910 0.8526	5	Flash Point °C		
A 305 to		0654	4	tc	0.0520		Fire Point		<u> </u>
c	156.		5	AHc kcal/m		m	M Spec.		
A* 305 to		7604	5	ΔHf ΔFf			Ultra V. X-Ray Dif.		
B* 1495 °C	2732.0		5		 	├	Infrared		
·	1			Viscosity centistokes	1		Solubility in +		
tk to				7 °C	·]		Acetone Carbon tet.		
t _x c							Benzene		1
B' C	}				L		Ether n-Heptane		
C'	<u> </u>			B ^v to			Ethanol	Ĭ	
A!#174 to		3008	5	AV - °C	_		Water Water in		
B'*305 °C	2687.9		5	(B ^V)	1	Í	water in	 	+
Ac to			ŀ	(A ^V)	1	<u> </u>			
Cc L-c-	1			c _p liq. °K		1			
Cryos, A° consts, B°				c _p vap. *K					
t _e °C	491.82	:	5	c _w vap.					
# for under	cooled lie	quid		<u> </u>	<u> </u>	1	grams/100 gra	ms solven	t
REFERENC	ES: 1-D	ow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:			AP						
PURIFICAT	ION:		AP	I .					
LITERATUI	RE REF	ERE							

No. 179 l-Triacontene NAME STRUCTURAL FORMULA $CH_3(CH_2)_{27}CH=CH_2$ Molecular C 30 H 60 Mole Ref. Molecular Weight 420.780 % Pur Ref. Ref. Ref. F.P. °C 62.4 2 dt/dP f to °C/mm g ۰ĸ 0.1 mm 166.92 B. P. °C h BP 0.0728 4 760 mm 448. 2 ^te 5 ſ١ 0.0357 to °K 100 353,43 5 g' 310.14 5 1.0964 5 30 mm 30 10 276.61 5 h' ∆Hm cal/g 219.97 5 1 m to ∆Hv cal/g 0.1 176.49 5 ٥ĸ n 57,20 0.1 mm Press. mm 0 30 mm 48.82 5 1834.4 ^te BP 38.95 5 m' Density to te te (d, e) 35.37 5 n' g/ml 20°C ۰ĸ 0.8141 2 5 35.17 01 0.8107 $\mathbf{d_{4}^{t}}$ 25 2 19.22 5 0.8075 AHV/T 30 4 Surface tension T313 71.04 to 5 0.8273 a 4 28,40 dynes/cm. 20°C <u> 149</u>5 °C 0.0716 ь -0.0366 4 27.49 5 30 to 68.27 26.60 40 5 Ref. Index e' 313 °C 0.0627 5 1.4573 20°C $\mathbf{n}_{\mathbf{D}}$ [P] Parachor dc g/ml 25 1.4553 2 20°C 1.4534 vc ml/g t °C 30 30 t_c 40 "C" 0.7433 4 P_c mm 1193.2 5 Sugd. MR (Obs.) 140.85 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 140.373 0.1 mm 1.0000 5 1.0503[≠] (nD-d/2) 2 103. [#] 30 mm 2 1.0000 5 Dispersion Dielectric 2.12 5 BP 0.8906 Flash Point °C 0.8517 A 313 to 7.61184 4 Fire Point В _545 °C 2850.4 M. Spec. Ultra V. C 154.5 5 AHc kcal/m A* 313 to B* 505 °C ΔHf 2.49231 5 X-Ray Dif. ΔFf 2765.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to tk tx Carbon tet. °C Benzene A' to Ether В' °C n-Heptane ВŸ Av A to Ethanol °C Water A!*180 to 2.39519 5 B'*313 °C Water in (B^V) 2720.4 Acl to (A^V) Bc °C c_p liq. ۰ĸ Cc Cryos. A c_p vap. °K consts. B° c, vap. te °C 501.03 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							······································	No. 18	0
NAME	1 - F	lentr	iacor	nten e			STRUCTURAL	FORMUL	A
Mole % Pur,	Ref.	Mo	lecul	ar C ₃₁ H ₆₂	Molecular Weight 434.	806	СH ₃ (СH ₂) ₂₈ С	H=CH ₂	
70 Fur.		FO	Ref	a 31 62	weight 454.	Ref			Ref
F. P. *C	64.6		2	44/45	1	I.C.	1 .	1	-
F.P. 100%			۲	dt/dP *C/mm			f to		l
B. P. °C	1			0.1 mm	168.97	5	h	1	
760 mm	457.		2 5	BP t _e	0.0736 0.0357	4 5	<u>f'</u> + <u>to</u>		
100 30	361.30		5	30 mm	1.1092	5	g' K		
10	283.6		5	ΔHm cal/g			h'		ì
1 0. 1	182.3		5	ΔHv cal/g	 	 	m to		
Press. mm			-	0.1 mm	56.12	5	n °K		1
t _e	1857.6		5	30 mm BP	47.91 38.20	5	<u> </u>	L	<u> </u>
Density		#		t_	34.61	5	m' to		ĺ
g/ml 20°C	0.81	153 [‡] , 120 ‡	2 2	e (a, e,	34.43	5	n' K		
d ^t 25 4 30	0.80	153° _≠ 120 [‡] 088 [‡]	4	ΔHv/T _e	19.17	5			├
a	0.8		4	d 320 to		5	Surface tension dynes/cm. 20°C	28.51	5
<u>b</u>	-0.0	366	4	d' 504 °C		5 5	30	27,60	5
Ref. Index	1.4	580 [#]	2	e' 320 °C		5	40	26.71	5
n _D 20°C			2	d _c g/ml			Parachor [P] 20°C		
30	1.45	540 [±]	4	vc ml/g tc °C	1	l	30		l
"C"	0.74		4	P _c mm		l	40 Sund	1232,2	5
MR (Obs.)			2	PV/RT	 	₩	Exp. L.1.%/wt.	1232.2	13
MR (Calc.) (nD-d/2)	144.99	91 503	5 2	0.1 mm	1.0000	5	u.	١,	
Dielectric	2.12		5	30 mm BP	1.0 000 0.8905	5	Dispersion	103. [‡]	2
A 320 to			4	te	0.8510	5	Flash Point °C	1	1
B [554 °C	2890.4		4	t _c			Fire Point		╁
C	153.		5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		1
A* 320 to B* 514 °C		0893	5	ΔFf		l	X-Ray Dif.		1
к 🗀 =	-			Viscosity		1	Infrared Solubility in +		<u> </u>
£ to	-			centistokes		ŀ	Solubility in +		
t _k to	1			η •ο	1		Carbon tet.		
A' to	İ		\vdash				Benzene Ether		
B' °	.[B ^V to	 	-	n-Heptane		
	 		 	B to		1	Ethanol Water		1
A'* 187 to B'* 320 °C		1119	5	(BV)	-		Water in		
Acl to	<u> </u>		\vdash	(A ^V)					
Bcit C	_			c _p liq. °K	-	+	1		
Ce	 -			_		1			
Cryos. A° consts. B°	ļ			c _p vap. *K					
t _e °C	511.43		5	c _v vap.		<u> </u>	I .		
# for unde							f grams/100 gra	ms solven	t
REFERENC	ES: 1-D	OW	2-AF		Calc, from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			AP						
PURIFICAT			AP						
LITERATU	RE REFI	EREN	ICES	:					

								No.181	
NAME	l-Dotri	acon	tene			ST	RUCTURAL	FORMUL	A
							СН ₃ (СН ₂) ₂₉ С	I=CH_	
Mole % Pur.	Ref. Mo	lecul mula	ar C ₃₂ H ₆₄	Molecular Weight 448,83	32		3, 5 <u>2</u> , 29	2	
		Ref.			Ref.				Ref.
F. P. °C	66.7	2	dt/dP	ł		f	to		
F.P. 100%			*C/mm 0.1 mm	170.75	5	g	•к		i
B.P. °C 760 mm	465.	2	BP	0.0743	4	h_	!		l
100	368.45	5	t _e	0.0357 1.1203	5	f' g'	to °K		
30 10	324.22 289.96	5 5	30 mm	1.1203		h'			
1	232.06	5	ΔHm cal/g		┼╌┤	m	l to		
0.1	187.58	5	ΔHv cal/g 0,1 mm	55.04	5	n	°K		
Press. mm	1879.8	5	30 mm	46.98	5	°	i		
Density			BP t _e	37.47 33.87	5	m'	to		
g/ml 20°C	0.8165 0.8132	2	t _e (d, e)	33.70	5	n' o'	•K		
dt 25	0.8132 ₇	2 4	ΔHv/T _e	19.14	5		<u> </u>		
a	0,8298	4	d 327 to		5		face tension es/cm. 20°C	28.61	5
ь	-0. 0366	4	d 192 to		5	8,	30	27.70	5
Ref. Index	1 4505#	١. ا	e' 327 °C		5	<u> </u>	40	26, 81	5
ⁿ D 20°C	1.4585	2 2	d _c g/ml			Par	rachor [P] 20°C		
30	1.4546	4	vc ml/g tc °C				30		
"C"	0.7429	4	P _c mm				40 Sugd.	1271.2	5
MR (Obs.) MR (Calc.)	150.13 149.609	2 5	PV/RT	·	\vdash	Ext	b. L.1.%/wt.		-
(nD-d/2)	1.0503	2	0.1 mm	1.0000	5		u.	10 3 .≠	
Dielectric	2.13	5	30 mm BP	1.0000 0.8910	5		persion	103.	2
A 327 to	7.62424	4	t e t c	0.8511	5		sh Point °C e Point		
B 1563 ℃ C	2924.3 151.5	4 5	ΔHc kcal/m		\vdash	M.	Spec.		
A* 327 to	2,52222	5	ΔHc Kcal/Hi			Ult	ra V.		
B*(523 °C	2837.3	5	ΔFf				Ray Dif. rared		l
K c			Viscosity centistokes		1	Sol	ubility in +		T
t _k Tto			η °C				etone rbon tet.		
1x			•				enzene		
A' to B' °C	1	١.,					her		į
č, ' =			B ^V to A ^V °C				Heptane hanol		ļ
A!* 192 to	2.42526	5		_			ater ater in		
B'* 327 °C	2791.2	5	(B ^V)				ater in		-
Ac to Bc tc °C	1		(A ^V)	_	\sqcup				
Co			c _p liq. °K						
Cryos, A° consts, B°			c _p vap. °K						
t _e °C	520.74	5	c _w vap.						
	rcooled liquid	<u> </u>	ш		نــــــا	+ g:	rams/100 gra	ms solver	ıt
REFERENC		2-A	PI 3-Lit. 4	-Calc. from de	t. da				
SOURCE:		AP							
PURIFICAT	ION:	AP							
	RE REFERE								

	 						No. 182	
NAME	l - Tritria	conte	ene			STRUCTURAL	FORMULA	
						СН ₃ (СН ₂) ₃₀ СІ	H=CH_	
Mole % Pur.	Ref. Mo	lecul rmul	ar C ₃₃ H ₆₆	Molecular Weight 462.8	358	2/30	2 2	
		Ref.			Ref.			Ref.
F.P. *C F.P. 100%	68.7	2	dt/dP *C/mm			f to g *K		
B. P. °C 760 mm 100 30 10 1 0.1	473. 375. 52 330. 85 296. 24 237. 75 192. 82	2 5 5 5 5 5	0.1 mm BP t _e 30 mm AHm cal/g AHv cal/g 0.1 mm 30 mm	172.53 0.0750 0.0356 1.1315 54.03 46.12	5 2 5 5 5	h to g' to o'K h' to n c'K o'		
t _e Density g/ml 20°C dt 25 4 30	0.81437 0.8111 [#]	2 2 4	BP te (d, e) ΔHv/Te d 334 to	36.84 33.23 33.1 20.49	5 5 5 5	m' to n' *K o' Surface tension		
a b Ref. Index	0.83 0 9 -0.0 ₃ 66	4	a 1 522 - C	0.0653 65.09	5 5	dynes/cm. 20°C 30 40	28.69 27.77 26.88	5 5 5
n _D 20°C 25 30	1.4591 1.4571 1.4550 0.7428	2 2 4	d g/ml vc ml/g tc °C	0.0573	5	Parachor [P] 20°C 30 40		
MR (Obs.) MR (Calc. (nD-d/2)	154.79 154.227 1.0503 [≠]	2 5 2	P _c mm PV/RT 0.1 mm 30 mm	1.0000	5	Exp. L.1.%/wt. u. Dispersion	1310.2	2
Dielectric A 334 to B 572°C		5 4 4	BP t _e t _c	0.8930 0.8529	5	Flash Point °C Fire Point	100.	
C A* 334 to B* 532 °C K	150. 2.53183	5 5	ΔHc kcal/m ΔHf ΔFf Viscosity			M Spec. Ultra V. X-Ray Dif. Infrared Solubility in +		
c t _k to t _x *C A' to B' *C	<u> </u>		centistokes η °C B ^V to			Acetone Carbon tet. Benzene Ether n-Heptane Ethanol		
A'* 198 to B'* 334 °C	2.43886 2823.7	5 5	(B ^V) - °C			Water Water in		
Ac to Bc tc Cc	-		(A ^v) c _p liq. °K		-			
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	530.21	5	c _v vap.	L	L			
	cooled liquid					f grams/100 gra	ms solvent	
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		AP						
PURIFICAT		AP						
LATERATU	RE REFEREI	NC ES	:					

								No. 183	3
NAME	1-Teti	atri	acontene	<u> </u>	_	ST	RUCTURAL	FORMUL.	A
							CH (CH) C	u-cu	
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 476.88	34		СН ₃ (СН ₂) ₃₁ С	n=Cn2	
		Ref.			Ref.				Ref
F.P. °C F.P. 100%	70.5	2	dt/dP			f	to		
B. P. °C	<u> </u>	-	°C/mm 0.1 mm	174.32	5	g	°K		
760 mm	481.	2	BP t _e	0.0757 0.0358	4 5	_h _f'			
100 30	382.59 337.48	5	30 mm	1.1427	5	g'	to °K		
10	302.53	5	ΔHm cal/g			h'			
1 0.1	243.45 198.06	5 5	ΔHv cal/g		\Box	m n	to °K		
Press. mm	1017 3	_	0.1 mm 30 mm	53.08 45.30	5	0	, a		
t _e	1917.3	5	BP	36.03	5	m'	to		-
Density g/ml 20°C	0.8186 0.8153	2	te te (d, e)	32.40 32.29	5	n' o'	°K		
dt 25 4 30	0.81537 0.8121	2	AHv/T _e	19.02	5	L			
a	0,8319	4	d 341 to	67.11	5		face tension es/cm. 20°C	28.78	5
b	-0.0366	4	e 531 °C to	0.0646 64.13	5	8,	30	27.87	5
Ref. Index	1.4596	2	e' 341 °C	0.0558	5		40	26. 97	5
D 25		2	d _c g/ml			Par	achor [P] 20°C		l
30	1,4550	4	t _c °C				30 40		
"C" MR (Obs.)	0.7425	4	P _c mm				Sugd.	1349.2	5
MR (Calc.)	159.44 158.845	2 5	PV/RT 0.1 mm	1 0000		Exp	L.1.%/wt.		
(nD-d/2)	1.0503	2	30 mm	1.0000 1.0000	5	Dis	u. persion	103. [‡]	2
Dielectric A 341 to	2.13	5	BP t _e	0.8895 0.8484	5		sh Point °C		
B 1581 °C	7.63409 2992.2	4	t _c		Ū		e Point		-
C A* 341 to	148.5	5	ΔHc kcal/m ΔHf			Ult	ra V.		
B*[541 °C	2552.8 2905.2	5	ΔFf				Ray Dif. ared		
K — — —			Viscosity centistokes				ubility in +		
t _k to	1		η °C				etone rbon tet.		
t _x °C	<u> </u>		•	1		Be	nzene		
B'							her Heptane		
C'	ļ	<u> </u>	B ^V to A ^V C			Et	hanol ater		
A'* 204 to B'* 341 °C	2.45201 2856.2	5. 5	-(B _V)				ater ater in		
Acl to	T	Ť	(A ^V)						
Bc tc °C	-		c _p liq. °K						
Cryos. A*		 	c _p vap. °K						
consts. B°] -						
t _e °C	539.14	5	c _v vap.	L		L	41.55	L	<u> </u>
	rcooled liquid		PI 3-14+ 4	Calc. from de	+ 4-		rams/100 gra -Calc. by for		t
SOURCE:	,10. 1-10W	API		Carc. Hom de	. ua	J	-Care, by for	******	
PURIFICAT	ION:	API							
	RE REFERE						1.1		

,							No. 184	
NAME	1-Pentat	riac	ontene			STRUCTURAL	FORMULA	
Mole	Ref. Mo	lecul	a.r	Molecular		CH ₃ (CH ₂) ₃₂	CH=CH ₂	
% Pur,	Fo	rmul	ar C ₃₅ H ₇₀	Weight 490.9	10			
		Ref.			Ref			Ref.
F.P. °C F.P. 1007	72.3	2	dt/dP		1	f to		İ
B. P. °C	<u> </u>	├	*C/mm 0.1 mm	176.10	5	g °K		İ
760 mm	489.	2	BP	0.0764	4 5	$\frac{1}{h} + \frac{1}{h}$		İ
100 30	389.65 344.11	5	t _e 30 mm	0.0358	5	g' to		İ
10	308.82	5	ΔHm cal/g		\vdash	h'		İ
1 0.1	249.15 203.29	5	ΔHv cal/g	 	╁╌╴	m to		
Press. mm			0.1 mm 30 mm	52.18	5	n °K		İ
te	1937.2	5	BP	44.53 35.38	5	m' l to		-
Density g/ml 20°C	0.8196	2	te (d, e)	31.74 31.65	5	n' 'K		
dt 25 4 30		2	ΔHv/T	18.89	5	o' ¦		
4 30 a	0.8163 0.8131 0.8329	4	d 347 to	66, 25	5	Surface tension	30.07	_
b	-0.0366	4	e 539 to		5	dynes/cm. 20°C	28.87 27.95	5
Ref. Index	1 4/01#	Ι.	e' 347 °C		5	40	27.05	5
n _D 20°0	1 45917	2 2	d _c g/ml			Parachor [P] 20°C		
30	1.4560	4	v _c m1/g t _c °C			30		
"C"	0.7425	4	P _c mm		İ	40 Sugd.	1388.2	5
MR (Obs.) MR (Calc.	1 163 463	2 5	PV/RT		†	Exp. L.1.%/wt.		
(nD-d/2)	1.0503	2	0,1 mm 30 mm	1.0000	5	u. Dispersion	102.≠	2
Dielectric		5	BP	0.8892	5	Flash Point °C	1	├
A 347 to		4	te t _c	0,8475	5	Fire Point		
c C	147.	5	ΔHc kcal/m	<u> </u>	1	M Spec. Ultra V.		Ì
A* 347 to B* 549 *(5	ΔHf ΔFf			X-Ray Dif.		
B* ∟ ⁵⁴⁹ °C	- 2730.1	"	Viscosity			Infrared Solubility in +		<u> </u>
t _k	-	ŀ	centistokes	.]		Solubility in + Acetone		
t _x + ·c		ŀ	"			Carbon tet. Benzene		
A' to			1			Ether		
B' '	<u>- </u>		B ^V to	 	1	n-Heptane Ethanol		
A1# 209 to		5	AV •C	<u>: </u>		Water Water in		
B'* 347 °(5	(B ^V)	1	1	warer III		+
Ac to			(A ^V)	<u> </u>	<u> </u>	4		
Ce	1	<u> </u>	c _p liq. ∘K					
Cryos. A' consts. B'			c _p vap. °K				1	
t _e °C	548.39	5	c _v vap.					
# for unde	rcooled liquid		•			grams/100 gra	ms solven	t
	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da			
SOURCE:		API					<u>-</u>	
PURIFICA		API						
LITERATU	RE REFERE	NCE	S:					
L								

No.185 l-Hexatriacontene NAME STRUCTURAL FORMULA $CH_3(CH_2)_{33}CH=CH_2$ Molecular C36H72 Mole Ref Molecular Weight 504.936 % Pur Ref. Ref Ref. F.P. *C F.P. 100% 73.9 2 dt/dP f to °C/mm °K g 177.73 0.1 mm 5 B. P. °C h BP 0.077 2 496. 760 mm 2 t_e 0.0358 5 f 100 395.83 5 ۰ĸ 1.1638 5 g' 349.90 30 mm 5 30 10 314.30 5 h' ∆Hm cal/g 227, 27 5 1 m to AHv cal/g 0.1 207.81 5 n ۰ĸ 51.22 43.73 0.1 mm 5 5 Press. mm 0 30 mm 1957.8 5 34.77 BP 5 m' Density to 31.12 5 0.8205[†] n' g/ml 20°C te (d, e) 2 31.06 5 ٥' d_4^t 25 0.8172 2 ΔHv/T_e 20.46 5 0.8140[#] 30 Surface tension d 353 65.18 5 a 0.8337 4 28.93 1 546 dynes/cm, 20°C °C 0.0613 b -0.0366 5 28.01 ă 214 30 62.18 5 40 27.11 5 Ref. Index e' | 353 0.0527 5 1.4605 $\mathbf{n}_{\mathbf{D}}$ 20°C [P] Parachor d_c g/ml 25 1.4585 2 20°C 1.4566 vc ml/g t_°C 30 4 30 ^tc "C" 40 0.7423 4 P_c mm Sugd. 1427.2 5 MR (Obs.) 168.71 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 168.081 5 0.1 mm 1,0000 1.0503[#] (nD-d/2)2 30 mm 1,0000 5 102. Dispersion 2 Dielectric 2.13 5 BP 0.8900 5 Flash Point °C A 353 to 0.8480 7.64540 4 Fire Point 3058.9 B 596 °C M. Spec. C 146. 5 AHc kcal/m Ultra V. ΔHf A* 353 to 2.57957 5 X-Ray Dif. ΔFf B*| 556 °C 2969.8 Infrared K Viscosity Solubility in c centistokes Acetone to Carbon tet. t_x °C Benzene A' to Ether B' °C n-Heptane B_v | C to Ethanol ۰c Water A'* 214 to 2.47928 5 Water in B'* 353 °C (B^V) 2920.1 Acl to (A^V)| Bc °C c_p liq. °К Cc Cryos. A. cp vap. ٩K consts. B c, vap. te °C 556.60 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

							No. 186	
NAME	l -Hepta	tria c	ontene			STRUCTURAL	FORMULA	L
			-			: ·		
Mole	Back Ma	11		Molecular		сн ₃ (сн ₂) ₃₄ (CH=CH ₂	
% Pur.	Ref. Mo	rmul	ar C 37H74	Weight 518.	962			
		Ref.			Ref			Ref.
F.P. °C	75.5	2	dt/dP			f to		
F. P. 1009			*C/mm 0.1 mm	179.12	5	g °K		
B.P. °C 760 mm	503.	2	BP	0.0776	4	<u>+ </u>	1	
100	402.03	5	t _e	0.0359	5	f' to		
30 10	355.74 319.85	5	30 mm	1,1731	5	h'	1	
1	259.17	5	ΔHm cal/g		<u> </u>	m to	 	
0.1 Press. mn	212.53	5	ΔHv cal/g 0.1 mm	50,43	5	n °K	1	
t _e	1972.4	5	30 mm	43.01	5.	°	Ĺ	<u> </u>
Density		┢	BP t _e	34.11 30.45	5	m' to		
g/ml 20°0	0.8214	2	t _e (d, e)	30.4	5	n' °K	1	1
dt 25 4 30	0.8181 0.8189 0.8149	2	AHv/T _e	20.46	5		<u> </u>	├
8	0.8347	4	d 359 to		5	Surface tension dynes/cm, 20°C	29.02	5
<u>ь</u>	-0.0366	4	a 554 °C		5	30	28.10	5
Ref. Index	1 4610	2	e' 359 °C		5	40	27.20	5
ⁿ D 20°C	1 1 45007	2	d _c g/ml			Parachor [P] 20°C		1
30	1.4570	4	tc °C			30	İ	
"C"	0.7421	4_	P _c mm			40 Sugd	1466.2	5
MR (Obs.) MR (Calc.	1 172 400	2	PV/RT		1	Exp. L.1.%/wt.	1	
(nD-d/2)	1.0502	2	0.1 mm 30 mm	1.0000	5	u.	102. #	
Dielectric		5	BP	1.000 0 0.8888	5	Dispersion Flash Point °C	102.	2
A 359 to B 604 °C	7.64418 3081.9	4	t _e t _c	0.8462	5	Fire Point		
C POT	144.	4 5	ΔHc kcal/m	 	\vdash	M Spec.		
A* 359 to		5	ΔHf			Ultra V. X-Ray Dif.]	
B* 564 °C	2993.7	5	ΔFf	 	-	Infrared		
c			Viscosity centistokes	1		Solubility in +		
tk to			η °c			Acetone Carbon tet.		İ
t'x i *C		-		}		Benzene Ether		ļ
B' _ °C			ļ	ļ	-	n-Heptane	İ	l
C'		<u> </u>	B ^V to A ^V °C			Ethanol Water		
A'* 219 to B'* 359 *(5	(BV)	-		Water in		
Acl to		Ť	(A ^V)	1				
Bc tc *C			c _p liq. °K	 				
Cryos. A		\vdash	1) -					
consts. B		L	c _p vap. °K					
t _e °C	564.62	5	c _v vap.					L
	ercooled liquid					f grams/100 gra	ms solven	
	CES: 1-Dow	2-AI		Calc, from det	t. da	ta 5-Calc. by for	mula	
SOURCE:		AP						
PURIFICA		AP						
LITERATU	RE REFERE	VCES	5:					

No.187 l-Octatriacontene NAME STRUCTURAL FORMULA CH3(CH2)35CH=CH2 Molecular C38H76 Ref. Mole Molecular Weight 532,988 % Pur. Ref. Ref. Ref. F.P. °C 77.0 2 dt/dP f to F.P. 100% °C/mm g °K 0.1 mm 180.76 B. P. °C h 0.0782 BP 4 510. 760 mm 2 t_e 0.0359 5 ſ١ to °K 100 408.21 5 5 5 g' 1.1831 30 mm 30 361.52 h١ 10 325.33 5 ∆Hm cal/g 1 264.11 5 m to ∆Hv cal/g 0.1 217.05 5 n ۰ĸ 0.1 mm 49.56 5 Press. mm o 30 mm 42.29 ^te 1990.9 5 ВP 33.54 5 m' to Density 5 te (d, e) 29.86 n' g/ml 20°C 0.8223 5 29.81 0.8190 o' $\mathbf{d_{4}^{t}}$ 25 2 AHV/T 20.36 5 0.8158 30 Surface tension 365 to 63.59 5 0.8356 4 dynes/cm. 20°C 29.10 5 ٠c e 562 d' 224 0.0589 60.50 -0.0366 ь 4 30 28.17 to 40 27.27 5 Ref. Index 365 0.0504 5 e' °C 1.4614 20°C [P] nD 1.4594 Parachor d_c g/ml 25 2 20°C 1.4574 vc ml/g 30 4 30 t_c ٠c 0.7419 40 "C" 4 1505.2 5 P_c mm Sugd MR (Obs.) 177.99 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 177.317 5 1.0502 0.1 mm 1.0000 5 (nD-d/2)2 102. 30 mm 1.0000 2 Dispersion Dielectric 2.13 5 ΒP 0.8889 Flash Point °C A 365 to B 612 °C 0.8459 7,65055 4 ţe. Fire Point 3114.6 4 M. Spec. C 143. 5 AHc kcal/m Ultra V. ΔHf A*| 365 to 2.60324 5 X-Ray Dif. ΔFf B* 572 °C 3025.6 Infrared Viscosity Solubility in centistokes Acetone to °C Carbon tet. °C Benzene A۱ to Ether B١ °C n-Heptane Bv | Av | C' to Ethanol °C Water 2,50025 A1 * 224 to 5 Water in (B^V)I B'* 365 °C 2973.6 Acl (A^V)| to Bc °C c_p liq. °K Cc Cryos. A° °K c_p vap. consts. B° c_v vap. te °C 572.76 5 # for undercooled liquid grams/100 grams solvent 5-Calc. by formula REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

NAME		l-No	natr	iacon	itene			STRUCTURAL	FORMUL	A
Mole % Pur.		Ref.	Mo	lecul	ar C ₃₉ H ₇₈	Molecular Weight 547.	014	СН ₃ (СН ₂) ₃	6CH=CH ₂	
/ <u>/</u>				Ref			Ref			Re
F. P. *C	T	78.4		2	1, (15	Т		· · · · · · · · · · · · · · · · · · ·	т	1
F.P. 100	4	10.4		۲	dt/dP *C/mm			f to		
B.P. °C	\top				0.1 mm BP	182.40	5	h .	1	1
760 mm		517.		2	t _e	0.0788 0.0359	2 5	f + to		
100 30		41 4. 39 367. 31		5 5	30 mm	1.1931	5	g' K		1
10		330.8	l	5	ΔHm cal/g	 		h'		1
1		269.00		5		 	-	m to		\top
0.1		221.5		5	ΔHv cal/g 0.1 mm	48.75	5	n °K		1
Press. mr t		006.7		5	30 mm	41.60	5	•		1
t _e Density	1			1-1	BP	32.96	5	m' l to	1	_
g/ml 20°	s	0.82	231 [‡]	2	te te (d, e)	29.27 29.25	5	n' 'K		1
dt 25		0.82	98	2	e (u, u,	20, 35	5	0'		1
4 30		0.81	66*	4	ΔHv/Te			Surface tension	 	+
a .		0.83		4	d 370 to		5	dynes/cm, 20°C	29.16	5
<u>ь</u>		-0.03	166	4	a 7 228 to		5	30	28.24	
Ref. Inde:		1.46	,,,≠	2	le!i370 °C	0.0490	5	40	27.34	5
ⁿ D 20°	٦	1.46	98	2	d g/ml vc ml/g			Parachor [P]	.	1
30		1.46	18	4	vc ml/g tc °C	i		30		1
"C"		0.74		4	, -			40	1	1_
MR (Obs.)	82.63	1	2	P _c mm			Suga	. 1544.2	5
MR (Calc	.) 1	181.93 1.05	5 22 #	5	PV/RT 0.1 mm	1,0000	5	Exp. L.1.%/wt.		1
(nD-d/2)	+-			2	30 mm	1.0000	5	Dispersion	102.	2
Dielectric		2.14		5	BP	0.8881	5	Flash Point °C	<u> </u>	T
A 370 1 B 619 •	<u>c</u> 31	7.65 47.4 42.	683	4 4 5	t _c	0.8446	3	Fire Point M Spec.	 	╀
A* 370 t	,	2,61	874	5	ΔHc kcal/m ΔHf ΔFf			Ultra V. X-Ray Dif.		
B* 579_° K c	30	58.4		5	Viscosity	<u> </u>		Infrared Solubility in +	<u> </u>	╁
եւ	0				remtistokes °C			Acetone Carbon tet.		Ì
A' t	c -			\Box				Benzene Ether		
č, 🗀 — -	<u>-</u>			1 1	B ^V to			n-Heptane Ethanol		
A'+228 t	•	2.51	369	5	AV I °C	1		Water		-
B'*370 •	C 30	04.9		5	(B ^V)			Water in	+	+
	Ĉ.				c _p liq. °K		-			
Cryos. A					c _p vap. *K					
consts. B	-	80,82		5	c _v vap.					
for und	erco	oled H	avid		L	<u> </u>	Ь	+ grams/100 gra	me ealwar	<u></u>
REFEREN					PI 3-Lit. 4-0	Calc. from de	t. de			
OURCE:				API						
PURIFICA	TION			API						
JTERATI			ERE							

No. 189 1-Tetracontene NAME STRUCTURAL FORMULA CH3(CH2)37CH=CH2 Molecula r Mole Ref Molecular $C_{40}^{H_{80}}$ % Pur. Weight 561.040 Formula Ref. Ref. Ref. F. P. °C 79.8 2 dt/dP f to F.P. 100% °C/mm g °К 0.1 mm 183,76 B. P. °C h BP 0.0794 4 523. 760 mm 2 ^te 0.0359 5 f to °K 100 419.69 5 5 g' 1,2016 372.28 30 mm 30 5 h' 10 335.51 5 ∆Hm cal/g 273.31 5 1 m to ∆Hv cal/g 0.1 225,48 5 ۰ĸ n 0.1 mm 47.92 40.90 5 5 Press. mm 0 30 mm ^te 2022,3 5 BP 32, 39 5 m Density to 5 28.70 te (d, e) n' g/ml 20°C 0.8238 2 28,73 0.8205 ۰' 25 dt4 AHv/T 5 0.8173[‡] 20.33 30 4 Surface tension 376 to 61.94 5 0.8371 4 a dynes/cm, 20°C 29.22 ٠Ĉ 0.0565 -0.0366 5 Ъ 30 28.29 5 58.**70** to 5 40 5 Ref. Index 27.39 376 0.0478 5 1.4622[‡] 1.4602[‡] n_D 2 [P] 20°C Parachor d_c g/ml 25 2 20°C 1.4582 vc ml/g 30 4 30 t_c 40 "C" 0.7417 4 P_c mm Sugd 1583.2 5 MR (Obs.) 187.30 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 186.553 5 1.0000 0.1 mm 5 1.0502[≠] (nD-d/2)2 102. # 30 mm 1.0000 5 Dispersion 2 Dielectric 2.14 5 ВP 0.8881 5 Flash Point °C A 376 to B 625 °C 0.8443 te 7.66107 4 Fire Point 3174.1 M. Spec. C 141. 5 AHc kcal/m Ultra V ΔHf A* 376 to 2.63081 5 X-Ray Dif. ΔFf B*| 585 °C 3084.5 Infrared ĸ Viscosity Solubility in c centistokes Acetone to Carbon tet. °C ţ Benzene A١ to Ether B °C n-Heptane B_v | C' to Ethanol °C Water A'* 232 to 2.52552 3030.5 5 Water in (B^V)[B'* 376 °C Acl (AV) Bc °C c_p liq. ۰ĸ Cc Cryos, Aº •ĸ c_p vap. consts. B° c_v vap. te °C 587.80 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API API **PURIFICATION:** LITERATURE REFERENCES:

							No. 1	
NAME	Vinyl bromi	.de				STRUCTURAL	FORMULA	
	Bromoethen							
						СН В г=СН ₂		
Mole % Pur. 100	Ref. Mo	lecul rmul		Molecular Weight 106.9	60			
		Ref.			Ref.			Ref.
F.P. ℃	-139.54	1	dt/dP			f to		
F.P. 100%	-139.54	1	*C/mm 25*C	0.0300	5	g '		1
B. P. °C 760 mm	+15.80	2	BP	0.038	2	<u>+</u> +	·	1
100	-31.7	5	t _e	0.03736	5	f' to		
30 10	-52.3	5	30 mm	0.5224	5	h'		
1	ļ		ΔHm cal/g ΔHv cal/g	11.44	4	m to		
Pressure mm 25°C	1033.	5	25°C	50.5	5	n °K		
t _e	777.0	5	30 mm BP	80.8 52.4	5	0 1		
Density	1.4933		t_	51.6	5	m' to		
g/ml 20°C dt 25 d4 30	1.47387	2 2	t (d, e)	51.2	5	0'		
^a 4 30	1.4542	5	ΔHv/T _e	19.12	5 5	Surface tension		\vdash
a b	1.5730	4	20 0		5	dynes/cm, 20°C	20.04 18.97 18.97	4
Ref. Index	 	1	d' to			25 30	18.97 ₄ 17.92 [‡]	4
n _D 20°C		2	d _c g/ml		1	Parachor [P]		
25 30	1.435	2	II V mil/g			20°C 30		
"C"	0.5847	5	ic C			40		١.
MR (Obs.)		4	P _c mm		 	Exp. L.1.%/wt.	152.1	5
MR (Calc. (nD-d/2)	18.734 0.6944	5	25°C	0.9512	5	u,		
Dielectric	5.628#	1	30 mm BP	1.0000 0.9650	5	Dispersion		
A -60 to	<u> </u>	5	t.	0.9642	5	Flash Point °C Fire Point		
B 1_60.°⊆	953.4 236.0	5	t _c AHc kcal/m		-	M Spec,		
A* -60 to		5	ΔHf			Ultra V. X-Ray Dif.		
B* _ 20 °C		5	ΔFf			Infrared		
K C	I		Viscosity centistokes			Solubility in +		
t _k to			7 -20 °C	0.2759 0.2528	1	Acetone Carbon tet.		
A' to	1	-	-10	0.2393	ì	Benzene Ether		1
B'						n-Heptane	ļ	
A'* to	 		B ^V -30 to A ^V 10 °C	213.8 2.59636	4	Ethanol Water		1
A'* to B'* °C			(BV)	1		Water in		
Ac to			(A ^V)					
Bc tc °C	<u>-</u>		c _p liq. °					
Cryos. A°	0.03453	1	c _p vap. °K					
t _e °C	16.44	5	c _v vap.					
	quid at satura		ll	t 5°C	L	+ ~~~~ (100	l	<u></u>
	CES: 1-Dow	2-AI			t. da	grams/100 graints 'grams/100 graints' grams/100 graints' grams/100 graints' grants' grants' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grams/100 graints' grain		<u> </u>
SOURCE:			, 2					
PURIFICAT	TION:		, 2					
LITERATU	RE REFERE							
1								
1								
L								

							No. 2	
NAME	Vinyl chl	oride				STRUCTURAL	FORMULA	
	Chloroet	hene				CH ₂ =CH	CI	
Mole % Pur. 100	Ref. Mo	lecul		Molecular Veight 62.501				
		Ref.			Ref.		R	ef.
F.P. °C F.P. 1009	-153.79	1	dt/dP °C/mm	0.01055		f to g° <u>K</u>		
B. P. °C 760 mm	-13, 37	1	25°C BP	0.01355 0.03423	4	h		
100 30	-55.8 -73.95	4	te	0.03675	5	f' to		
10	-87.5	4	30 mm	0.4500 18.14	4	h'		
1	-109.42	4	ΔHm cal/g ΔHv cal/g	10.14	-	m to		_
Pressure mm 25°C	2660.0 692.3	4 5	25°C 30 mm BP	71.26 93.50 79.5 3	5 5 5	n		
Density				79.84	5	m! to		
g/ml -30°	0.99986 0.99176	1	te te (d, e)	80.08	5	",		
d ₄ -25 -20	0.98343	1	ΔHv/T _e	19.38	5	Surface tension		_
a b	0.95421 -0.00158	4	e 20 °C to	76.45 0.2306	5	dynes/cm-30°C -20 -10	22, 27	5 5 5
Ref. Index	1.4046	5	e¹	0.370	_	Parachor [P]	20.00	_
25	1		d _c g/ml v _c ml/g t _c °C	0.370 2.70	5	20°C 30	ŀ	
"C"	0,54	5		156.5	5	40		
MR (Obs.)			P _c mm	42000.	5			5
MR (Calc.	15.836 10°C 0.9213	5	PV/RT 25°C	0.9178	5	Exp. L.1.%/wt.		1
Dielectric		1	30 mm BP	1.0000 0.9640	5	Dispersion		
A -100 to		4	t_	0.9652	5.	Flash Point °C Fire Point	-78.0	1
B _ 50 °C	783.4 230.0	4	tc ΔHc kcal/m	0. 264	5	M. Spec.		
A* -80 to		5	ΔHf	-144.0	5	Ultra V. X-Ray Dif.		
B* _ 0°C		5	ΔFf			Infrared		
c	_		Viscosity centistokes			Solubility in +		
t _k to			7 -40 ℃ -30	0.3339 0.3026	1	Acetone Carbon tet.		
A' to		-	-20	0.2780	1	Benzene Ether		
B'°	<u>-</u>		Bv -50 to	0,2563	5	n-Heptane		
A'* to			A	234.945 7.51616	4	Ethanol Water		
B'* °C			(B ^V) to			Water in	0.11	1
Acl 50 to		5	(A ^V) °C					
Bc t _c °C	652.	5	c _p liq. 293°K	0.38	1			
Cryos. A° consts. B°		1	c vap.298°K	0.205	1			
t _e °C	-15.76	5	c _v vap.			L	L	
T _R = 0.7			# 17.2°C	2-1- (grams/100 gra		
SOURCE:	CES: 1-Dow	Z-A Do		Jaic, irom de	t. da	ta 5-Calc. by for	mula	
PURIFICA	TION:		stillation					_
	RE REFERE							

							No. 3	
NAME	cis-1-Chlor	o-1- ₁	propene			STRUCTURAL :	FOR MULA	
						CHCI-CH	~U	
Mole	Ref. Mo	lecul	ar	Molecular		CHC1=CH-C	[^] 3	
% Pur.	² Fo	rmu		Weight 76.	527			_
		Ref.			Ref		IR IR	lef.
F.P. °C F.P. 100%	-134.8	2	dt/dP *C/mm		1	f to		
B. P. *C	-	-	25°C	0.0472	5	g K		
760 mm 100	+32.8	2 5	BP t _e	0.0379 0.03578	5 5	$\frac{1}{1} + \frac{1}{10}$		
30	-36.2	5	30 mm	0.5266	5	g' 'K	l	
10 1	-52, 1	5	ΔHm cal/g			h'		
Pressure	<u> </u>		ΔHv cal/g 25°C	81.72	5	m to		
mm 25°C	575.0 817.9	5	30 mm	92.30	5	•		
Density	+ 0111.7	-	BP	80.42 80.38	5 5	m¹ tó		
g/ml 20°C		2	te te (d, e)	80.06	5	n' K		
d ₄ 25	0.9271 0.9193	5	ΔHv/T _e	19.96	5	O'	-	
a	0.9656	4	d -40 to		5	Surface tension dynes/cm. 20°C	21.84	5
b Pot Indon	-0.00144	4		1		25 30	21.10	5
Ref. Index		2	e' j •c	0,337	5	Parachor [P]	-	_
25 30	1.400	2	d g/ml v ml/g tc °C	2.97	5	20°C 30		
"C"	0.5780	4	11 -	217.8	5	40		
MR (Obs.)	20.828	4	P _c mm PV/RT	B5700.	5		177.4	5
MR (Calc. (nD-d/2)) 20.454 0.9382	5	25°C	0.9631	5	Exp. L.1.%/wt. u.		
Dielectric	0.7502	i	30 mm BP	1.0000 0.9560	5	Dispersion		
A -40 to	6.88871	5	t _e	0.9540 0.265	5	Flash Point °C Fire Point		
B ∟9 <u>5</u> ° <u>C</u>	2065.3	5 5	t _c ΔHc kcal/m	0.265	3	M Spec.		
A* -40 to		5	ΔHf			Ultra V. X-Ray Dif.		
B* ∟45 °C	994.8	5	ΔFf Viscosity	 	-	Infrared		
c	_		centistokes			Solubility in + Acetone		
t _x to			η •c		1 1	Carbon tet.		
A' to						Benzene Ether		
B' L _ 2	2		B ^V to	 	1	n-Heptane Ethanol		
A'* to			A ^V °C			Water		
B'* °C		<u> </u>	(B ^V)			Water in	 	
Ac 95 to	7.26279	5	(A ^V)		Ш			
Ce	270.	5	op Mq.					
Cryos, A° consts. B°			c _p vap. °K					
t _e °C	34, 93	5	c _v vap.					
				<u> </u>		grams/100 gra	ns solvent	_
	CES: 1-Dow	2-A1	PI 3-Lit. 4-0	Calc. from det	t. da			
SOURCE:		AF						
PURIFICA?		AF						
LITERATU	RE REFERE	NCES	S:					

							No. 4
NAME _	trans-1-Chlo	oro-	l-propene			STRUCTURAL	FORMULA
L			r		-	CHC1=CH-C	СН3
Mole % Pur.	Ref. Moi 2 For	ecul	ar C ₃ H ₅ C1	Molecular Veight 76.5	27		
		Ref.			Ref.		Re
F.P. °C F.P. 100%	-99.0	2	dt/dP			f to	
B. P. °C			*C/mm 25*C	0.05405	5	g c	
760 mm	37.4	2	BP	0.0385	5	h	
100 30	-11.2 -32.6	5	t _e	0.03594	5	f¹ to cC].
10	-48.7	5	30 mm	0.5341	5	h'	1 1
Pressure		-	ΔHv cal/g			m to	
mm 25°C	487.2	5	25°C	83.51	5	n °K	
t _e	830.8	5	30 mm BP	93.81 81.57	5	•	
Density g/ml 20°C	0.035		t _e (d, e)	81.03	5	m' to	
dt 25	0.935 0.927	2 2	t _e (a, e)	81.11	5	ا 'ه	
⁴ 4 30	0.9185	4		19.82	5	Surface tension	
a b	0.9696 -0.00162	5	_e_!50 °C	88.11 0.1748	5 5	dynes/cm. 20°C	21.77 5 21.13 5
Ref. Index	-0.00102	5	d' to			8 25 30	20.33
n _D 20°C	1.4054	2	d _c g/ml	0,333	5	Parachor [P]	
25 30	1.400	2	v _c mi/g	3.000	5	20°C	j
"C"	0.578	5		225.0 35900.	5	40	
MR (Obs.)	20.828	4	P _c mm	55900.	3		177.4 5
MR (Calc.) (nD-d/2)	20.454 0.9379	5 4	25°C	0.9654	5	Exp. L.1.%/wt. u.	1
Dielectric	0.7317	Ė	30 mm BP	1.0000 0.9560	. 5 5	Dispersion	
A -40 to	6,88340	5	t te t	0.9525	5	Flash Point °C Fire Point	
B 1100 °C	1078.3	5	t _c	0.265	5	M. Spec.	
A* -40 to	1,24083	5	ΔHc kcal/m ΔHf			Ultra V.	
B* 50 °C	1007.1	5	ΔFf			X-Ray Dif. Infrared	
K — — —			Viscosity centistokes			Solubility in +	
t _k to			η °c			Acetone Carbon tet.	
A' to		_				Benzene	1
B' C					_	Ether n-Heptane	
C'			B ^V to A ^V °C			Ethanol Water	
A'* to B'* °C			(BV)			Water in	
Acl 100 to	7.28034	5	(A ^V)				
Bc t _c °C	1349. 270.	5	c _p liq. °C				
Cryos, A° consts, B°			c _p vap. •K				
t _e °C	40.03	5	c _w vap.				
						grams/100 gra	ms solvent
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-6	Calc. from de	t. da	ta 5-Calc. by for	:mula
SOURCE:		AP	I				
PURIFICAT		AP					
LITERATUR	RE REFERE	NCES	5:				

*****	1 1	Twie	ablas	maathulama		T	STRUCTURAL	No. 5	
NAME	1, 1,	2 - 1 F10	emoi	roethylene		\dashv			n
Mole % Pur.	Ref.	Mole	eculi muli	C2HC13	Molecular Weight 131.3	99	CC12=CH	Cl	
//			Ref.	<u></u>		Ref			Rei
F,P. °C	T			34/35		H		Ι	+
F. P. 1009			\neg	dt/dP *C/mm			f to	1	
B. P. *C	+	\rightarrow	\dashv	25°C	0.2890	5	h .		
760 mm	87.0		1	BP	0.04369	5			+
100 30	31.5 6.9		4 5	t _e	0.0359	1 1	g' to		
10	-11.8		5	30 mm	0.6178	5	b'	1	
1	-42.8	9	. 5	ΔHm cal/g		\sqcup		<u></u>	+
Pressure				ΔHv cal/g 25°C	62,33	5	m to		1
mm 25°C			5	30 mm	64.02	5	0	1	
t _e	975.5	4	5	BP	56.43	5	m¹ to		+
Density g/ml 20°0	. 1 1 4	6422	1	te (d.e)	55.61	5	n' °K		
		5541	i	e (4, 0)	55.62	5	0'	1	1
dt 25 4 60		9501	1	ΔHv/T _e	19.81	5	Surface tension	 	+-
a		9948	4	d 7 to	64.67	5	dynes/cm. 20°C	29.28	1
ь	-0.0	0174	4	├ ぷ ,-┤-96_ ° S	0.0946	5	30	27.94	1
Ref. Index			.	e' *C			40	26.76	1
ⁿ D 20°0		7734	1	d _c g/ml	0.513	5	Parachor [P] 20°C	208.7	4
50		6056	î	V/g	1.950	5	30	208.8	4
"C"	0.4	302	4	tc •C	298.	5	40	209.0	4
MR (Obs.	25, 3	, 	4	P _c mm	36876.	5	Sugd.	212.8	5
MR (Calc.) 25.5	7	5	PV/RT 25°C	0.9957	5	Exp. L.1.%/wt.	ĺ	
(nD-d/2)	0.7	4523	4	30 mm	1.0000	5	u. Dispersion		
Dielectric		1	- 1	BP	0.9578	5	Flash Point °C		+
A 7 t		2808	4	te	0.9507 0.265	5 5	Fire Point	l	
B (_155 •9	2 1315.0 230.		4	t _c	0.203	-	M Spec.		
A* 7 to		1642	5	ΔHC KCal/H			Ultra V.		
B+ 106 °C		1042	5	ΔFf			X-Ray Dif. Infrared		
к — — -	-			Viscosity			Solubility in +		+
c	-1	1	l	centistokes 7 20 °C	0 2044	1	Acetone	1	1
tk to				7 20 °C	0.3844 0.3291	i	Carbon tet.		
A' to	,	\rightarrow	-	60	0.2867	1	Benzene Ether		
B' •		1		80	0.2553	1	n-Heptane	ł	1
C'				B ^V 10 to A ^V 70 °C	311.03	4	Ethanol		
A'* to					Z. 52395	4	Water Water in		1
B1# *(`	75	<u>_</u> -	(B ^V) to	1				+
Ac 155 to		075	5	(A ^V) °C		\sqcup			
Bctc_°	280.		5	c _p liq. ∘K	1				
Cryos, A'				c _p vap. °K					
t _e °C	95.60	,	5	c _v vap.					
$T_{R} = 0.75$					1		+ grams/100 gran	ne entre-	
REFEREN		ow 2	-AP	PI 3-Lit. 4-0	Calc. from det	teh :			
SOURCE:			Do			. 441	_ J-Care, by for		
PURIFICA	TION			stillation					
LITERATU		EREN							
				-					

No. 6 Perchloroethylene STRUCTURAL FORMULA NAME CC12=CC12 Molecular C2Cl4 Mole Molecular Weight 165.848 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g _°K 25°C 1.0102 5 B. P. ℃ h BP 0.04721 5 760 mm 120.97 1 t_e 0.0357 5 ſ١ to 100 60.97 4 g' <u>°К</u> 0.6670 5 30 34.37 4 30 mm 10 14.02 5 h! ∆Hm cal/g 5 -20.12 m to AHv cal/g Pressure °K n 25°C 57.09 5 mm 25°C 18.47 5 o 30 mm 56.65 5 5 1086. te ВP 50.0 5 m' to 48.96 Density 5 5 t_e (d, e) n' °K g/ml 20°C 1,62272 1 48.98 ۰, d_4^t 25 1.61446 1 19.93 5 AHV/Te 30 1.60590 4 Surface tension d 34 to 59.29 5 a 1.65599 4 dynes/cm. 20°C 31.33 1 0.0768 5 134 °C 25 to ь -0.00166 4 30.81 30 1 d٦ 58.27 5 40 30.20 1 e ¹ Ref. Index 34 °C 0.0472 5 20°C 1.50534 Parachor [P] n_D 0.573 d_c g/ml 4 25 1.50284 1 20°C 241.8 4 vc ml/g t °C 1.745 4 1.50004 30 4 30 243.3 4 ^tc 340. 5 4 40 244.6 "C" 0.4095 4 P_c mm 33687. 5 Sugd. 250.0 5 MR (Obs.) 30.34 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30.437 25°C 1,0000 (nD-d/2) 0.69417 4 30 mm 1.0000 5 Dispersion Dielectric BP 0.9672 Flash Point C 0.9580 5 A 34 to 7.02003 Fire Point 0.255 В 1415.5 187°C 4 M. Spec. Ultra V. С 221. 4 AHc kcal/m ΔHf A* 34 to 1.58865 X-Ray Dif. ΔFf B* 144°C 1321.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to 20 0.54976 1 t_x Carbon tet. 40 0.45702 1 Benzene 60 0.39347 1 25 to 7.41073 Ether 0.34061 80 B' 1622.1 34 °C 5 n-Heptane $\mathbf{B}^{\mathbf{v} \top}$ C' 239. 5 10 $\frac{353.2}{2.53223}$ Ethanol A | 90 °C A1* Water 25 to 1.96901 5 34 °C (BV) Water in 1518.6 5 to Ac | 187 to 7.4489 5 (A^V)| °C 1787. Bc tc °C c_p liq. ۰ĸ Cc 271. 5 c_p vap. Cryos. A °K consts. B° c vap. t_e °C 134.28 5 $T_{R} = 0.75 T$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula Dow SOURCE: Distillation PURIFICATION: LITERATURE REFERENCES:

· · · · · · · · · · · · · · · · · · ·							No. 7		
NAME	Hexachlor	opro	pene			STRUCTURAL	FORMULA		
	Hexachlor	opro	pylene			C Cl ₃ C Cl=C Cl ₂			
Mole % Pur.	Ref. Mo	lecul rmul	ar C ₃ Cl ₆	Molecular Weight 248.7	72				
		Ref.			Ref			Ref.	
F. P. °C	-57.12	1	dt/dP			f to			
F.P. 100%			*C/mm 25*C	48,562	5	g <u> </u>		ĺ	
B.P. °C 760 mm	214.14	1	BP	0.05767	4	h			
100	141.13	i	t _e	0.03718	5	f' to		İ	
30	108.97 84.5	4 5	30 mm	0.805 0	4	g'K_	1	ĺ	
10 1	43.5	5	ΔHm cal/g			h' (ļ		
Pressure	†		ΔHv cal/g	50.40	_	m to		İ	
mm 25°C	0.2717	5	25°C 30 mm	53.40 47.91	5 5		1	ĺ	
t _e	1336.0	5	BP	40.7	5	m' to	-	_	
Density g/ml 20°C	1.76376	1	te te (d, e)	39.04 38.89	5	n' K	}	ĺ	
dt 25	1.75656	1	ΔHv/T	19.06	5	0'			
	1.7494	4	d 100 to		5	Surface tension			
a b	1.79256 -0.00144	4	e 245 °C	0 .0687	5	dynes/cm. 20°C	38.12 36.95	1 1	
Ref. Index	1	i	d' 20 to		5	40	35.87	l î	
n _D 20°C		1		0.58	5	Parachor [P]		\Box	
25 30	1.54720 1.53570	1	d g/ml v ml/g	1.72	5	20°C	350.5 349.2	4	
"C"	0.40748	4	t _c *C	447.	5	40	350.9	4	
MR (Obs.)	44.903	4	P _c mm	23966.	5	Sugd.	363.4	5	
MR (Calc.)		5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.	1		
(nD-d/2)	0.66768	4	30 mm	1.0000	5	u. Dispersion		İ	
Dielectric	2.682	1	BP	0.9522 0.9348	5	Flash Point °C			
A 109 to B <u>267</u> °C		1	te tc	0.229	5	Fire Point		<u> </u>	
c	193.87	i	AHc kcal/m	1	 	M Spec.		l	
A* 110 to	1.61793	5	ΔHf ΔFf			Ultra V. X-Ray Dif.	1	İ	
B* _ 250°C	1550.51	5	Viscosity	<u> </u>	-	Infrared		L_	
c	_		centistokes	1		Solubility in +	1		
5k -00	· ·		7 20 °C	1.9698	1 1	Carbon tet.	1 1	İ	
t *C	4	5	60	1.0652	1	Benzene Ether			
B' _109 °C	1863.7	5	80	0.8515	1	n-Heptane		ĺ	
C' 20	212.95	5	B ^V 35 to A ^V 90 °C	601.4 7.22746	4	Ethanol Water	0.015	1	
A'* 20 to B'* 109 °C	1.9832 1765.4	5 5			-	Water in	0.014	ī	
Ac 269 to	7.3256	5	(A ^V) to						
Bc tc_C	2041.	5	c _p liq. °K	 	\vdash	1			
Cryon A*	245.7	5	, F						
Cryos, A° consts, B°			c _p vap. *K						
t _e °C	240.45	5	c _v vap.	<u> </u>	L				
$T_{R} = 0.75$						f grams/100 gran	ms solvent	t	
	ES: 1-Dow			Calc, from det	da	ta 5-Calc, by for	mula		
SOURCE:			DW						
PURIFICAT			st. chromat.						
LITERATU	RE REFEREI	1CES	i:						

								N o. 1	
NAME	Propadi	ene				ST	RUCTURAL	FORMUL	A
	Allene								
Mole % Pur	Ref.	Molecula Formula		Molecular Weight 40.	062		CH ₂ =C=0	CH ₂	
		Ref.		T	Ref.	Ī			Ref
F, P, °C	-136.	2	dt/dP			f	to		
F.P. 1009			°C/mm 25°C	0.0110	_	g	•c		
B. P. ℃ 760 mm	-34.5	2	BP	0.0110 0.033	5 4	_h_	l		
100	-73.0	4	t _e	0.0372	5	f!	to °C		
30 10	-88. 2 -99.	4	30 mm	0.3674	5	g' h'	! [
ì			ΔHm cal/g				to		-
Pressure			ΔHv cal/g 25°C			m	*K		
mm 25°C	4286. 637.1	5 5	30 mm	153.8	5	0	i		
Density			BP t _e	109.0 112.0	5	m'			
g/ml 20°C	;		te (d, e)	112.5	5	n'	•K		
dt 25 4 30			AHv/Te	19.23	5	0'	<u> </u>		<u> </u>
a. 30			d -90 to		5		face tension es/cm, 20°C		
b			-30 °C		5	8,,,	30		
Ref. Index			e' °C			<u> </u>	4.0		-
ⁿ D 20°C	1		d _c g/ml			Par	achor [P]		
30			vc ml/g tc °C	120.0	2		30		
"C"			P _c mm		-	ł	40 Sugd.		ļ
MR (Obs.) MR (Calc.			PV/RT		\vdash	Ext	L.1.%/wt.		f^{-}
(nD-d/2)	'		25°C 30 mm	0.8750	5 5		u.		}
Dielectric			BP	1.0000 0.9700	5		persion sh Point °C		
A -95 to			t _e	0.9769	5		e Point		
B25 °C	441.0	2 2	t _c ΔHc kcal/m	·	+-	M.	Spec.		T
A* -90 to			ΔHf				ra V. Ray Dif.		İ
B*[_15 °C		5	ΔFf		ļ		ared		
K			Viscosity centistokes		1		ubility in +		
t _k to			η °c				etone rbon tet.		
t _x °C						Be	nzene		
B' °C			ļ				her Heptane		
C'			B ^V to A ^V °C			Et	hanol		
A'* to B'* °C			$\frac{A'}{(B') } - {^{\circ}C}$	-			ater ater in		
Acl to			(A ^V)						1
Bc tc °C				+	+				
Cc			Р -						
Cryos, A° consts, B°			c _p vap. *K						
te °C	-38.8	5	c _v vap.						
						+ g:	rams/100 gra	ms solver	nt
REFEREN	CES: 1-D	ow 2-A	PI 3-Lit. 4	-Calc. from de	et. da	ta 5	-Calc. by for	mula	
SOURCE:		AP	I						
PURIFICA		AP							
LITERATU	RE REF	ERENCES	S:						
<u> </u>									

							No. 2	
NAME	1,2-B	utadiene				STRUCTURAL	FORMUL!	4
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 54.08	8	CH ₂ =C=CH	сн ₃	
		Ref.			Ref			Ref.
F.P. °C	-136, 190		dt/dP	Τ		4 1 1 4		\vdash
F.P. 100%			*C/mm	i	1 1	f to		ĺ
B. P. *C	†	\rightarrow	25°C	0.0234	5	h .	·	
760 mm	10.85		BP	0.03495	4		 	⊢
100	-33.83	2	t _e	0.03500	5	f' to		l
30 10	-54. -69.07	2 2	30 mm	0.5022	5	h'	1	1
ì	-94.48	5	ΔHm cal/g				0.0/54	-
Pressure			ΔHv cal/g		_	m 300 to n 600 °K	0.0654	
mm 25°C	1259.8	5	25°C 30 mm	104.13 117.39	5 5	-020	-0.0642	
t _e	758.60	5	BP BP	106.74	5			-
Density		#	t _e ,	106.76	5	m' 700 to n' , 1000 °K	0.1346	4
g/ml 20°C	0.652 0.646	ž 2	t _e (a, e)	106.75	5	" 1 <u>1000 F</u>	0.0 ₃ 90 -0.0 ₆ 30	4
d ₄ 25	0.640	2 4	ΔHv/T _e	20.33	5		6	<u> </u>
a .	0,677		d -54 to	108,53	5	Surface tension	1,757	_
ь	-0.001		<u></u>		5	dynes/cm. 20°C	16.56 15.24	5
Ref. Index	†		d' to		1	40	13.95	5
n _D 20°C	:	} }		0,2468	3	Parachor [P]		
- 25			d g/ml vc ml/g tc °C	4.051	3	20°C		l
30			tc °C °	170.6	3	30	ŀ	
"C"	<u></u>		Pcmm	33770.	3	40 Sugd	168.2	5
MR (Obs.)			PV/RT		-	Exp. L.1.%/wt.	100.2	
MR (Calc.	19.738	5 5	25°C	0.9461	5	Exp. L.1.76/Wt.		
(nD-d/2)	ļ	\rightarrow	30 mm	1.0000	5	Dispersion		
Dielectric	<u> </u>		BP	0.9600	5	Flash Point °C	 	
A -54 to			te .	0.9601		Fire Point	ŀ	
B _68_•0	251.0	2 2	t _c AHc kcal/m	588.37	2	M Spec.		
A* -54 to			ΔHc Real/H ΔHf gas	38.77	2	Ultra V.		ł
B* 21 °C	1045.57	5	∆Ff gas	47.43	2	X-Ray Dif. Infrared		
ĸ	•	1 1	Viscosity			<u> </u>		
·	_		centistokes			Solubility in + Acetone	1	Ì
tk to		1 1	ነ የ ℃	ļ		Carbon tet.		
A' to					1	Benzene		l
B' °C						Ether n-Heptane		
c, – – –	-		B ^V to			Ethanol		
A'* to			AV °C	_		Water		1
B'* °C			(B ^V) to			Water in	ļ	₩-
Ac 68 to			(A ^V)i °C				}	
	1438.26	5 5	c _p liq. °K					1
Ce	273.48	- "	-	1				1
Cryos. A° consts. B°			c _p vap.300°K 400	0.35553 0.43522	2 2			
t _e °C	10.80	5	c _v vap.					L
$T_R = 0.7$	5 Т _с		≠ at saturation	pressure		f grams/100 gra	ms solven	t
REFERENC	ES: 1-Do	w 2-AF	PI 3-Lit. 4-0	Calc, from det	t. da	ta 5-Calc. by for		
SOURCE:		AP	1					
PURIFICAT	ION:	AP	ī					
		RENCES	: 3 ASTM Sp.	Tech. Pub. 1	109			

								No. 3		
NAME	1,3-Butad	iene				STRUCTURAL FORMULA				
							כם -כםכו	1-CU		
Mole % Pur.	Ref. Mo	lecul mula		Molecular Weight 54.08	8		сн ₂ = с нс	1-CH ₂		
		Ref.			Ref.				Ref.	
F.P. °C	-108.915	2	dt/dP			f	to			
F.P. 100%	·	-	°C/mm 25°C	0.0154	5	g	<u>'°K</u>			
B. P. °C 760 mm	-4.413	2	BP	0.03377	4	h	<u> </u>		-	
100 30	-47.035 -65.74	2 2	t _e	0.03542 0.4675	5	f' g'	to K			
10	-79.89	2	30 mm	0.4075		h'				
1	-103.17	5	ΔHm cal/g	 	\vdash	m	300 to	-0.0514		
Pressure mm 25°C	2104.5	5	25°C	92.25	5	n	_600 <u>•</u> K	0.0016		
t _e	714.96	5	30 mm BP	112.75 98.89	5 5			-0.0 ₆ 98		
Density			t_	99.26	5	m' n'	700 to	0.1766	4	
g/ml 20°C	0.6211 [‡] 0.6149 [‡]	2 2	ie (4, 6)	99.25	5	٥'	1 1000 K	0.0 ₃ 87 -0.0 ₆ 30	4	
d ₄ 25 30	0.6086	4	ΔHv/T _e	20.09	5	S., .	face tension		-	
a	0.6480	4	d	97.90 0.2259	5 5		es/cm. 20°C	13.41	5	
Ref. Index	-0.0011	4	d' to	1,111,		8	30 4 0	12.20 11.01	5	
n _D 20°C			e' °C	0, 245	2	Par	achor [P]			
30	1		d _c g/ml v _c ml/g t _c °C	4. 082	2		20°C			
"C"		-	tc°°C	152.	2		30 4 0			
MR (Obs.)		\vdash	P _c mm	32452.	2			168.2	5	
MR (Calc.		5	PV/RT 25°C	0, 9269	5	Exp	u. L.1.%/wt.			
(nD-d/2)			30 mm	1.0000	5	Dis	persion			
Dielectric	6, 85941	2	BP t _e	0.9600 0.9616	5 5		sh Point C			
B 1_46 °C	935.53	2	t ^e c	0.271	2		e Point			
C	239.55	2	ΔHc kcal/m ΔHf	575.93 21.21	2 2		Spec. ra V.			
A* -66 to B* 5 °C	1.11738 872.39	5	ΔFf	21.21	-		Ray Dif. ared			
к — — –	-		Viscosity				ubility in +			
t _k -tō	-		centistokes り °C		1		etone			
t _x °C			7				rbon tet.			
A' to						Et	her			
B'°	-		B ^V to A ^V °C				Heptane hanol			
A'* to					1 1	Ŵ	ater			
B'* °C			(B ^V) to			- w	ater in		-	
Ac 46 to	7.29710 1202,54	5	(A ^V) °C		\sqcup	1				
Cc	277.80	5	c _p liq. °K							
Cryos. A° consts. B°			c vap 300°K 400	0.35331 0.44908	2 2					
t _e °C	-5.97	5	c _v vap.			L				
$T_R = 0.7$			≠ at saturation				rams/100 gra		t	
	CES: 1-Dow	2-A		Calc. from de	t. da	ta 5	-Calc. by for	mula		
SOURCE:	TION.		PI							
PURIFICA	FION: RE REFERE		.PI							
LITERATO	RE REFERE	NCES	:							

									No. 4	
NAME	1,2-F	entad	liene				STR	UCTURAL	FORMULA	
Mole % Pur.	Ref.	Mol-	ecul	ar C ₅ H ₈	Molecular Weight 68.11	4		CH _Z =C=CHC	H ₂ CH ₃	
			Ref.			Ref				Ref.
F, P. °C	-137.26	5	2	dt/dP			f	l to		
F.P. 100%				°C/mm			g	L_ K	1	ł
B. P. °C				25°C BP	0.0691 0.03867	5	h	•		
760 mm 100	44.85		2 2	t.	0.03528		f'	to		
30	-26.04		2	30 mm	0.5457	5	g'			
10 1	-42.6 -69.99	.	2 5	ΔHm cal/g		П	h'	<u> </u>		<u> </u>
Pressure	-07.7	'		ΔHv cal/g	<u> </u>	М	m	300 to	0.0148	
mm 25°C	367.26	,	5	25°C	99.47	5	n o	_600 °K	0.0014 -0.0 ₆ 73	
t _e	853.68	3	5	30 mm BP	108.86	5 5				
Density				t_	95.12	5	m' n'	700 to	0.1189 0.0010	
g/ml 20°C	0.69		2 2	1 te (4, 6)	95.12	5	61	17000 7	-0.0637	
d ^t 25 4 30	0.68		4	ΔHv/T _e	20.15	5		1		
8	0.71	265	4	d -26 to		5		s/cm. 20°C	19.57	5
Ъ	-0.0	947	4	-a, -48 to		5	3,	30	18.41	5
Ref. Index	1 , 4:		٠, ١	e' i		Ш	<u> </u>	40	17.27	5
ⁿ D 20°C	1,42		2 2	d _c g/ml			Para	chor [P] 20°C		
30	1.41		4	V mi/g	221.	5		30		
"C"	0,80	080	4	1 -	32089.	5		40 Su = 4	207.2	_
MR (Obs.)	24.93		2	P _c mm	32007.	H	F		207.2	5
MR (Calc.) (nD-d/2)	24.35		5 2	25°C	0.9744	5	Exp.	L.1.%/wt. u.		
Dielectric	2,02	$\overline{}$	5	30 mm	1.0000	5	Disp	ersion	164.6	2
A -26 to	+		2	BP t _e	0.9575 0.9543	5 5		h Point °C		
B97 °C			2	tc		1 1		Point		<u> </u>
c ——-	234.65		2	∆Hc kcal/m	736.25	2	M S _l			ļ
A* -26 to	1.29		5	ΔHf gas ΔFf gas	34.80	2 2		ay Dif.		ļ
B* ∟ 58 °C	. 1079.85	·	5	Viscosity	50.29	-	Infra	red		
c		-	1	centistokes		1 1		bility in +		
tk to	1	1		η •c				tone bon tet.	80	l
A' to					1		Bei	zene	œ	
B' C]	1			1		Eth	er Ieptane	80	
				B ^v l to				anol	· ·	
A¹* to				A ^V I C	_		Wa	ter ter in		
B'* °C	 			(B ^V) to			- " a	61 111		<u> </u>
Ac 97 to	7.44		5	(A ^V) °C						İ
Cc	276.31		5	c _p liq. ∘K						1
Cryos, A*			1	cp vap.300°K	0.37144	2				ľ
consts. B°	-			1 400	0.46099	2				
t _e °C	48.31		5	c _v vap.						
$T_R = 0.7$							+ gra	ms/100 gran	ns solvent	<u> </u>
REFERENC	ES: 1-D	ow 2	-AF		Calc. from det	dat	ta 5-	Calc, by form	nula	
SOURCE:			AF	PI						
PURIFICAT			AF							
LITERATU	RE REFI	CREN	CES							

								No. 5	
NAME _	1, cis-3-P	entac	li e ne			ST	RUCTURAL	FORMUL	A.
Mole % Pur.		ecul:		Molecular Veight 68,11	4		сн ₂ =сн сн	-сн с н ₃	
		Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-140,820	2	dt/dP °C/mm 25°C	0.0673	5	f g	to°K		
B.P. °C 760 mm 100 30	44.068 -5.02 -26.67	2 2 2	BP t _e 30 mm	0. 03875 0. 03544 0. 5421	5 5	f' g'	to c		
10	-43.1 -70.22	2 5	ΔHm cal/g			_h'			
Pressure mm 25°C t _e	379.07 851.42	5	ΔHv cal/g 25°C 30 mm BP	98.84 109.02 95.09	5 5 . 5	m o	300 to	-0.0544 0.0015 -0.0 ₆ 73	4 4
Density g/ml 20°C dt 25 4 30	0.69102 0.68592 0.68078	2 2 4	te te (d,e) AHv/Te	94.44 94.42 20.06	5 5	m' n' o'	700 to 1000 °K	0.1498 0.0 ₃ 91 -0.0 ₆ 29	4 4
a b	0.7116 -0.0 ₃ 97	4 4	d -27 to e 47 °C d' to	103.76 0.1969	5		face tension es/cm. 20°C 30	19.39 18.21	5
Ref. Index n _D 20°C 25 30	1.43634 1.43291 1.42943	2 2 4	d g/ml vc ml/g tc °C	219.	5	Par	40 rachor [P] 20°C 30	17.05	5
"C"	0.8377	4	P _c mm	30901.	5		40 Sund	207.2	5
MR (Obs.) MR (Calc.) (nD'-d/2)	25.790 25.626 1.09083	2 5 2	PV/RT 25°C 30 mm	0.9739 1.0000	5 5	_	L.l.%/wt. u. persion	243.8	ž
Dielectric A -27 to	2. 06 6. 94178	5 2	BP te	0.9575 0.95 4 3	5		sh Point °C e Point		
B _ 96 °C _ C _ A* -27 to	1118.37 231.33 1.23515	2 2 5	tc ΔHc kcal/m ΔHf gas	720.15 18.70	2 2	Ult	Spec.		
B* 57 °C	1046.11	5	ΔFf gas Viscosity centistokes	34.88	2	Infr Solu	lay Dif. ared bility in +		
t _k to °C			η •c			Ca Be	etone rbon tet. nzene		
A' to B' °C C' - °C			BV to			n- Et	her Heptane hanol		
A'* to B'* °C			$\frac{\mathbf{A}}{ \mathbf{B}^{\mathbf{v}} } - \frac{\mathbf{v}}{\mathbf{to}} -$				iter iter in		
Acl 96 to Bc t _c °C	7.37139 1417.55 272.55	5 5 5	(A ^V) °C c _p liq. °K						
Cryos, A° consts, B°			c _p vap300°K 400	0.33326 0.43310	2 2				
te °C	47.45	5	c _w vap.			<u> </u>			
T _R = 0.75							ams/100 gra		t
	ES: 1-Dow	2-A		Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE: PURIFICAT	ION·	AF							
	RE REFERE								

							No. 6	
NAME	1, trans-3	-Pen	tadiene			STRUCTURAL	FORMULA	
				·····				
Mole % Pur.		lecul rmul	ar C ₅ H ₈	Molecular Weight 68.114	4	CH ₂ =CHCH=	ECH CH ₃	
		Ref.			Ref.			Ref
F.P. °C	-87.470	2	dt/dP			f to		
F.P. 100%			°C/mm	1	_	g K		
B. P. *C			25°C BP	0.0632	5 4	h		
760 mm 100	42.032 -7.06	2	t	0.03567	5	f' to		
30	-28.7	2	30 mm	0.5414	5	g'	-	
10 1	-45.1 -72.15	2 5	ΔHm cal/g			h'		
Pressure	+	 	ΔHv cal/g			m 300 to	0.0087 0.0014	4
mm 25°C	410.5	5	25°C	97.06 107.39	5	n 600 °K	-0.0 ₆ 66	4
t _e	845.6	5	30 mm BP	93.76	5	1 700		
Density	0 (7/02		te (d. a)	93.17	5	m' 700 to n' 1000 K	0.1303	4
g/ml 20°C	0.67603 0.67102	2 2	'e (u, c)	93.15	5	0'	-0.0637	4
dt 25 4 30	0.66597	4	ΔHv/T _e	19.93	5	Surface tension	 	-
a	0.6963	4	d -29 to		5	dynes/cm. 20°C	17.75	5
ь	-0.0395	4		1	ا آ	30 40	16.66 15.58	5
Ref. Index	1,43008	2	e' i °C			Parachor [P]	13.30	<u> </u>
D 25	1.42669	2	d g/ml			20°C		
30	1.42302	4	v _c ml/g t _c °C	214.	5	30		
"C"	0.8448	4	Pcmm	29456.	5	40 Sugd	207.2	5
MR (Obs.)	26. 032	2	PV/RT			Exp. L.1.%/wt.		
MR (Calc.) (nD-d/2)	1.09206	5 2	25°C	0.9724	5-	u.	[]	_
Dielectric	2,04	5	30 mm BP	1.0000 0.9575	5	Dispersion	245.7	2
A -29 to		2	te	0.9545	5	Flash Point °C Fire Point		
B92 °C	1108.94	2	t _c	<u> </u>	L		 	
С	232, 34	2	ΔHc kcal/m ΔHf gas	720.05 18.60	2 2	M Spec. Ultra V.		
A* -29 to B*55 °C	1.21772	5	ΔFf gas	35.07	2	X-Ray Dif.		
K L==	-		Viscosity			Infrared Solubility in +		
t.	-{	į	centistokes 7 °C	1		Solubility in + Acetone	oc	
t _k to t _x to			7 °⊂			Carbon tet.	00	
A' to	 	_		1		Benzene Ether	80	Ì
B' °	4		B ^V to		-	n-Heptane	œ	
	 	-	B' to	İ		Ethanol Water	∞	
A'* to B'* °C		1	(B ^V) to	-		Water in		
Ac 92 to	7.35341	5	(A ^V) °C					
Bc t C		5	c _p liq. °K	 				
Cc	273.32	5	41 -	1				
Cryos. A° consts. B°		_	c _p vap.300°K	0.36556 0.45806	2	E		
t _e °C	45.22	5	c _v vap.	l		L,	<u> </u>	<u> </u>
$T_{R} = 0.7$						f grams/100 gra		<u> </u>
	ES: 1-Dow			Calc. from det	. da	ta 5-Calc, by for	mula	
SOURCE:		AF						
PURIFICAT		AI						
LITERATU	RE REFERE	NCES	5:					

								No. 7	
NAME	1,4-Penta	dien	•			ST	RUCTURAL	FORMUL	A
Mole % Pur.	Ref. Mo	lecul	ar C ₅ H ₈	Molecular Weight 68.114	1		CH ₂ =CHCH ₂	CH=CH ₂	
		Ref.			Ref.				Ref.
F.P. °C F.P. 100°	-148.275	2	dt/dP °C/mm			f g	to °K		
B. P. °C			25°C BP	0.0382	5 4	h	I		
760 mm	25.967 -20.96	2 2	t _e	0.03573	5	f'	to		
30	-41.5	2	30 mm	0.5142	5	g'	° <u>K</u>		
10	-57.1 -82.69	5	ΔHm cal/g			h'	i		
Pressure	-02.09	ا	ΔHv cal/g			m	300 to	0.0221 0.0013	4
mm 25°C	734.3	5	25°C	88.28	5	n o	_600 •K	-0.0666	
t _e	800.1	5	30 mm BP	101.49 88.09	5		1 700 .		├
Density	2 ((22)		t _e	87.82	5	m' n'	700 to	0.1175 0.0010	4
g/ml 20°0	0.66076	2 2	te (d, e)	87.80	5	0'	=======	-0.0 ₆ 37	4
d ₄ 25 30	0.6506	4	ΔHv/T _e	19.90	5		face tension	<u> </u>	
a	0.6816	4	d -41 to		5 5		es/cm. 20°C	16.09	5
Ь	-0.0392	4	d' to			8	30 40	15.03 13.98	5
Ref. Index		2	e' °C			- Bar		13.90	3
ⁿ D 20°0	1.38542	2	d _c g/ml	}		Par	achor [P]		İ
30	1.38247	4	tc °C	187.	5		30		
"C"	0.7857	4	P _c mm	27238.	5		40 Sugd.	207.2	5
MR (Obs.		2	PV/RT		\vdash	Ext	L. 1. %/wt.		<u> </u>
MR (Calc. (nD-d/2)	24.356 1.05838	5 2	25°C	0.9589	5	-	u.		
Dielectric		5	30 mm BP	1.0000 0.9580	5 5		persion	153.1	2
A -41 to		2	t _e	0.9565	5		sh Point °C e Point	ĺ	
B _ 72 °C	1025.016	2	<u>c</u>				Spec.		-
C	232, 354	2	ΔHc kcal/m ΔHf gas	726.65 25.20	2 2	Ult	ra V.		1
A* -41 to B* _37 °C		5	∆Ff gas	40.69	2		Ray Dif. ared		
к ——	- 701.30	*	Viscosity						\vdash
t, to	-1		centistokes n °C				ubility in ^T etone	ļ }	
t _k to		1	∥າ °c				rbon tet.		
A' to		†					enzene her		
B'°	길	ł	B _v to	 	\vdash		Heptane		
A'* to		+-	B to A °C				hanol ater		
B'* °C		1	(B ^V) to	-	1		ater in		
Ac 72 to		5	(A ^V) °C						
Bc_tc_°C		5	c _p liq. °K						
Cc	270.91	5	(1						1
Cryos, A' consts. B'		<u> </u>	c _p vap.300°K	0.36997 0.45952	2 2				
t _e °C	27.43	5	c _v vap.	<u></u>		L,		L	<u> </u>
$T_R = 0.7$							rams/100 gra		ıt
	CES: 1-Dow			-Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		AF							
PURIFICA		AF							
LITERAT	JRE R e fere	NCE	5:						

								No. 8	
NAME	2,3-Penta	diene	•			STR	UCTURAL I	FORMULA	
Mole % Pur.	Ref. Mo	lecul rmul	ar C ₅ H ₈	Molecular Weight 68.114	4		CH ₃ CH=C=C	нсн ₃	
		Ref.			Ref.				Ref
F.P. *C	-125.650	2	dt/dP			f	to		
F. P. 100% B. P. *C	_	<u> </u>	*C/mm 25*C	0.0768	5	g	<u>•K</u> _		
760 mm	48.265	2	BP	0.03870	4	h	 		
100 30	-0.64 -22.1	2 2	t _e	0.03505	· 5	f' g'	to cK		
10	-38.4	2	30 mm	0.5377	3	h'			
1	-65.24	5	ΔHm cal/g		-	m	300 to	0.0660	4
Pressure	220.0	_	ΔHv cal/g 25°C	102.91	5	n	_600 °K_	0.0011	4
mm 25°C	320.0 862.8	5	30 mm	114.00	5	٥		-0.0 ₆ 37	4
Density		\vdash	BP	97.70 96.86	5	m¹	700 to	0.0790	4
g/ml 20°C		2	t (d, e)	96.82	5	n' o'	1 <u>1000 •K</u>	0.0010 -0.0 ₆ 37	4
dt 25 4 30	0.69000 0.68494	2 4	ΔHv/T _e	20.28	5		l	-0.0631	-
	0,7152	4	d -22 to	108.87	5		s/cm. 20°C	19.87	5
ь	-0.0 ₃ 96	4		0.2315	5	3,	30	18.69	5
Ref. Index			e' °C			<u> </u>	40	17.53	5
n _D 20°C	1.42842	2 2	d _c g/ml			Para	chor [P] 20°C		
30	1.42170	4	tc °C	226.	5		30		
"C"	0.8187	4	P _c mm	31482.	5		40 Sugd	2 0 7.2	5
MR (Obs.)		2 5	PV/RT	31102.	-	Exp	L. 1. %/wt.		ř
MR (Calc. (nD-d/2)	1.08091	2	25°C	0.9770	5		u.		
Dielectric	2.04	5	30 mm BP	1. 0 000 0.9570	5			174.5	2
A -22 to	6,88603	2	te	0.9534	5		h Point °C Point		
B [101 °C	1086.64	2	t _c	734.55		M S			
A* -22 to		5	ΔHc kcal/m ΔHf gas	734.55 33.10	2 2	Ultra	a V.		
B* 62 °C		5	ΔFf gas	49.22	2	Infra	ay Dif. red		
к — —	7		Viscosity				bility in +		 -
t _k to		l	centistokes 7°C			Ace	tone	œ	
t _x ;			'				bon tet. zene	8 0	
A' to B' •C						Eth	er	œ	
B' •	-1		B ^V to	-			leptane anol	ος ος	
A'* to			AV °C			Wat	er	~	
B'* °C	4	L_	(BV) to			Wat	ter in		<u> </u>
Ac 101 to Bc t °C		5	(A ^V) °C						
Cc c	263.23	5	c _p liq. ∘K						
Cryos. A° consts. B°			c _p vap.300°K 400	0.35675 0.43897	2 2				
t _e °C	52.05	5	c _v vap.	1, 255 /.	-	ľ			
$T_R = 0.7$		<u> </u>	u	l	L	+ 072	ms/100 gran	ne solven	<u> </u>
		2-A1	PI 3-Lit. 4-C	alc. from det	da:	ta 5-	Calc. by for	nula	
SOURCE:			API			<u>_</u>			
PURIFICAT	TION:	1	\PI						
LITERATU	RE REFERE	VCES	5:						

								No. 9	
NAME	3-Methyl-	1,2-1	outadiene			ST	RUCTURAL	FORMUL	A
							CH ₂ =C=C	CH,	
Mole % Pur.		ecul	ar C ₅ H ₈	Molecular Weight 68.11	4				
	1	Ref.			Ref.				Ref.
F.P. °C F.P. 100%			dt/dP °C/mm 25°C	0.0588	5	f g	to		
B. P. °C 760 mm 100 30	40. -8.	2 2	BP t _e	0.03796 0.035 07	4 5	f' g'	to *K		
10	-30. -46.	2	30 mm ΔHm cal/g	0, 5353	_5_	h'			
Pressure	-72, 69	5	ΔHv cal/g	07.50	_	m	300 to	0.0588 0.0012	4
mm 25°C	438.6 840.4	5 5	25°C 30 mm BP	97.50 107.81 94.63	5 5 5	0		-0.0651	4
Density g/ml 20°C dt 25 d4 30	0.680 0.675	2 2	te te (d, e) AHv/Te	94.08 94.08 20.27	5	m' n' o'	700 to 1000 °K	0.1525 0.0 ₃ 91 -0.0 ₆ 29	4 4 4
a b	0.670 0.7002 -0.0 ₃ 94	4 4	d -30 to e 43 °C d to	102.21	5		face tension es/cm. 20°C 30	18.16 17.05	5
Ref. Index n _D 20°C 25 30	1.410 1.407 1.404	2 2 4	e' °C	212.	5	Par	40 Pachor [P] 20°C 30	15.95	5
"C"	0.8028	4	tc°C P _c mm	31519.	5		40 Sugd	207.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	24.818 24.356 1.070	4 5 2	PV/RT 25°C 30 mm	0.9713 1.0000	5	1	L.l.%/wt. u. persion	201,2	-
Dielectric	1.99	5	BP	0.9580 0.9555	5		sh Point °C		-
A -30 to B 90 °C	7.005 1130. 234.	2 2 2	te tc AHc kcal/m		2	├ ──	e Point		_
A* -30 to B* 53 °C	1.29994 1057.09	5	ΔHf gas ΔFf gas	732,45 31,00° 47,47	2 2	X-F	ra V. Ray Dif. ared		
K c t _k to c t _x °C			Viscosity centistokes n °C			Ac Ca Be	bility in tetone rbon tet.	& &	
B' °C C' to			B ^V to A ^V °C			n- Et Wa	her Heptane hanol ater	88	
B'* °C Acl 90 to Bc t _c °C	7.43814 1428.51	5 5	(B ^V) to (A ^V) °C c liq. °K			-wa	ater in		
Cryos. A° consts. B°	274.34	5	c _p liq. °K c _p vap.300°K 400	0.37144 0.45512	2 2				
t _e °C	42.93	5	c vap.						
$T_R = 0.75$	T _c					+ gr	ams/100 gra	ms solven	t
REFERENC	ES: 1-Dow			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			PI						
PURIFICAT			PI						
LITERATUF	RE REFERE	NCES	5:						

						No.	. 10
NAME	2-Methyl-	1,3-1	outadiene			STRUCTURAL FOR	MULA
	Isoprene					CH2=C CH=CH	2
Mole	Ref. Mo	lecul	ar	Molecular		CH ₂ =C CH=CH	2
% Pur.		rmul	г с ₅ н ₈	Weight 68.11	4		
		Ref.			Ref.		Ref.
F.P. C F.P. 1009	-145.950	2	dt/dP			f to	
B. P. *C	•	\vdash	*C/mm 25*C	0.0492	5	g <u>•K</u>	
760 mm	34.067	2	BP	0.03817 0.03582	4	h -	
100 30	-14.21 -35.5	2 2	t _e	0.03582	5	f' to g' 'K	
10	-51.6	2	30 mm	0.3313	-	h'	
1	-78.08	5	ΔHm cal/g ΔHv cal/g	-	├	m 300 to -0	.0351 4
Pressure mm 25°C	550.0	5	25°C	92.34	5	n _600 °K 0	.0016 4
t _e	824.2	5	30 mm BP	103.42 90.67	5	l	.0695 4
Density			t.	90.23	5		. 1880 4 . 0 . 88 4
g/ml 20°0	0.68095 0.67587	2 2	'e (u, e)	90.24	5	n' 1000 °K 0	.0 ₃ 88 4 .0 ₆ 29 4
d ^t 25	0.67074	4	ΔHv/T _e	19.85	5	Surface tension	-
	0.7017	4	d -35 to e 36 °C	96.92 0.1834	5	dynes/cm. 20°C 18	. 22 5
b	-0.0395	4			_		.08 5 .95 5
Ref. Index		2	e' °C		 	Parachor [P]	.,,,
D 25	1.41852	2	d g/ml v ml/g			20°C	
30	1.41524	4	v _c ml/g t _c °C	202.	5	30 40	
	0.8237	4	P _c mm	28860.	5	Sugd. 207	.2 5
MR (Obs.) MR (Calc.		2 5	PV/RT	2 2/51	_	Exp. L.1.%/wt.	
(nD-d/2)	1.08146	2	25°C 30 mm	0.9671 1.0000	5	u. Dispersion 224	.8 2
Dielectric		5	BP	0.9590 0.9568	5	Flash Point °C	-
A -35 t B84 •		2 2	te t _c	0.7508		Fire Point	
c Lot	234.67	2	ΔHc kcal/m	719.55	2	M Spec.	
A* -35 to		5	ΔHf ΔFf	11.80	2	Ultra V. X-Ray Dif.	
B* <u> </u>	<u>C 1009. 21</u>	5	Viscosity	-		Infrared	
c	_}		centistokes	1		Solubility in + Acetone	
tk to			η •c			Carbon tet.	
t _x i °C		\vdash			1	Benzene Ether	
B' 4			B ^V to	_		n-Heptane	
	_	\vdash	B' to			Ethanol Water	
A'* to B'* *			(B ^V) to	•	1	Water in	
Ac 84 to	7, 33735	5	(A ^V) °C	1	}		
Bc tc_°	C 1374.92 275.34	5	c _p liq. °K			1	
Cryos. A		1	c _p vap.300°K	0.36997	2		
consts. B		Ш	400	0.46686	2		
t _e °C	36.44	5	c _w vap.		L		
$T_R = 0.7$						† grams/100 grams s	olvent
	CES: 1-Dow	2-AF		Calc. from det	. da	ta 5-Calc. by formula	1
SOURCE:		AF					
PURIFICA		AF	~ ~~~				
LIIEKATU	RE REFERE	NCES	:				

						No. 11
NAME	1,2-Hexa	liene				STRUCTURAL FORMULA
Mole % Pur.		lecul mula		Molecular Weight 82.14	.0	CH ₂ =C=CH(CH ₂) ₂ CH ₃
		Ref.			Ref.	Re
F.P. °C F.P. 100%			dt/dP °C/mm			f to g °K
B. P. °C 760 mm 100	76. 23.40	2	25°C BP t _e	0.2012 0.04136 0.03497	5 4 5	h f¹ to
30	0.064	5	30 mm	0,5853	5_	g' '° <u>K</u>
10	-17.70 -47.11	5	ΔHm cal/g			h'
Pressure mm 25°C t _e	107.7 940.1	5	ΔHv cal/g 25°C 30 mm	98.35 102.88	5	m to o
Density g/ml 20°C	0.7149 0.7102	2 2	BP t _e t _e (d, e)	89.22 88.03 87.99	5 5 5	m' to n' K
dt 25 4 30	0.7055	4	ΔHv/T _e	20.31	5	S
a b	0.7337 -0.0 ₃ 92	4 4	d 0 to e 83 °C d to	0.1799	5	Surface tension dynes/cm. 20°C 21.04 5 5 5 5 5 5 5 5 5
Ref. Index n _D 20°C 25 30	1.4282 1.4252	2	d g/ml v ml/g			Parachor [P] 20°C
"C"	1.4222 0.7955	4	vc ml/g tc °C Pc mm			30 40 Sugd. 246. 2 5
MR (Obs.) MR (Calc. (nD-d/2)		5 2	PV/RT 25°C 30 mm	0.9915 1.0000	5	Exp. L.1.%/wt.
Dielectric	2.04	5	BP	0.9550	5	Dispersion Flash Point ⁶ C
A 0 to B 1 123 °C C	7.03380 1248.39 224.6	4 4 5	te tc ΔHc kcal/m	0.9490	5	Fire Point M. Spec.
A* 0 to B* 93 °C	1.37445	5	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared
K c t _k to			Viscosity centistokes 7°C			Solubility in + Acetone Carbon tet.
A' to B' _ °C			B _v to			Benzene Ether n-Heptane
A¹* to B¹* °C			$ \begin{vmatrix} \mathbf{B} \\ \mathbf{A}^{\mathbf{V}} \\ \mathbf{B}^{\mathbf{V}} \mathbf{O} \\ $	-		Ethanol Water Water in
Ac to Bc t _c °C			(A ^V) °C c _p liq. °K		-	
Cryos, A° consts, B°			c _p vap. °K			
t _e °C	82.84	5	c _w vap.	<u> </u>		
B D D D D D D D D D D D D D D D D D D D		-	.			† grams/100 grams solvent
	CES: 1-Dow	2-A		Calc. from de	t. da	ita 5-Calc. by formula
SOURCE:	TION:	AP				
	RE REFERE					
			•			

						No. 12			
NAME	1, cis-3-H	exadi	ene			STRUCTURAL FORMULA			
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 82.14	10	CH ₂ =CHCH=CHCH ₂ CH ₃			
		Ref.			Ref	R	Ref.		
F, P. *C			dt/dP			f to			
F.P. 100%			*C/mm			g K			
B. P. *C			25°C BP	0.1807 0.04106	5 4	h			
760 mm 100	73. 20.78	2 5	t	0.03498	5	f¹ to			
30	-2.37	5	30 mm	0.5808	5	g' K			
10 1	-20.00 -49.18	5	ΔHm cal/g			h¹			
Pressure	17.10	 	ΔHv cal/g	İ		m to			
mm 25°C	121.55	5	25°C	96.88	5	n °K			
t _e	931.49	5	30 mm BP	101.84 88.32	5	 			
Density			t_	87.17	5	m' to *K			
g/ml 20°C	0.705 0.700	2 2	e (4, 6)	87.16	5	 			
dt 25 4 30	0.6950	4	ΔHv/T _e	20.30	5	Surface Associated			
8	0.7250	4	d -2 to	101.41	5	Surface tension dynes/cm. 20°C 19.89	5		
ъ	-0.0398	4	-a,¦79- ° C	0.1793	5	30 18.76	5		
Ref. Index	1 420		e' C		<u> </u>	40 17.65	5		
n _D 20°C	1.438	2 2	d _c g/ml			Parachor [P] 20°C			
30	1.432	4	li V mil∕o			30			
"C"	0.8241	4	li -			40	5		
MR (Obs.)	30.585	4	P _c mm			Sugd. 246.2 Exp. L.1, %/wt.			
MR (Calc.) (nD-d/2)		5 2	25°C	0.9901	5	u.			
Dielectric	1.086	+	30 mm	1.0000	5	Dispersion 225.	2		
A -2 to	2.07	5	BP t _e	0.9550 0.9493	5	Flash Point °C			
B 118 °C	7.02864 1236.47	4	tc			Fire Point			
<u> </u>	225.1	5	ΔHc kcal/m			M Spec. Ultra V.			
A* -2 to	1.37294	5	ΔHf ΔFf			X-Ray Dif.			
B* ∟ 89 °C	1160.15	5	Viscosity		-	Infrared			
c			centistokes			Solubility in +			
tk To	}		η •c	ĺ		Acetone Carbon tet.			
t _x *C		ļ	 	j		Benzene			
B' C						Ether n-Heptane			
			B ^V to			Ethanol			
A'* to	!	}	AV _ °C	}		Water Water in			
B'* °C		-	(B ^V) to)					
Ac to			(A ^V) °C						
Cc			c _p liq. ∘K						
Cryos. A° consts. B°			c _p vap. °K						
t _e •C	79.49	5	c _w vap.			+			
REFERENC	ES: 1-Dow	2 - AT	PI 3-Lit. 4-C	ale from de		grams/100 grams solvent ta 5-Calc. by formula			
SOURCE:		Al		a.c. irom de	. aa	J-Care, by formula			
PURIFICAT	ION:	Al							
	E REFERE								

No. 13 1, trans-3-Hexadiene NAME STRUCTURAL FORMULA CH,=CHCH=CHCH,CH, Mole Ref. Molecular Molecular C6H10 % Pur Formula Weight 82.140 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm ٩ĸ g 25°C 0.1807 5 B. P. °C h BP 0.04106 4 760 mm 73. 2 t_e 0.03498 5 f 100 20.78 5 to g¹ 5 <u>°К</u> 30 -2.370.5808 5 30 mm 10 -20.00 5 h! ∆Hm cal/g 1 -49.18 5 to AHv cal/g Pressure n ۰ĸ 25°C 96.88 5 mm 25°C 121.55 5 o 5 101.84 30 mm te 931.49 5 BP 88.32 5 m to Density 5 te (d, e) 87.17 n' °K g/ml 20°C 0.705 87.16 2 5 o' 25 0.700 2 d_4^t AHv/T 20.30 5 30 0.6950 4 Surface tension -2 d to 101.41 0.1793 5 0.7250 4 19.89 5 dynes/cm. 20°C <u>79 ℃</u> 5 Ъ 18.76 -0.0398 4 30 ď٦ to 5 17.65 40 Ref. Index e¹ ⁿD 20°C 1.438 [P] Parachor d_c g/ml 25 1.435 2 20°C ,с .С ml/g 30 1.432 4 30 $\mathbf{t_c}$ 40 "C" 4 0.8241 5 P_c mm 246.2 Sugd. MR (Obs.) 30.585 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30, 244 5 25°C 0.9901 (nD-d/2)1.086 2 30 mm 1.0000 5 225. 2 Dispersion 5 Dielectric 2.07 BP 0.9550 5 Flash Point °C 0.9493 A -2 to B 118 °C 7.02864 1236.47 te tc 44 Fire Point M. Spec. 225.1 5 C AHc kcal/m Ultra V ΔHf 1.37294 -2 to A*| 5 X-Ray Dif. B*| 89 °C ΔFf 1160.15 Infrared ĸ Viscosity Solubility in centistokes Acetone t_k to °C Carbon tet. t_x °C Benzene A to Ether В °C n-Heptane B_V C to Ethanol °C A'* Water to Water in B'* $(\mathbf{B}^{\mathbf{V}})$ °C to Acl to (AV) °C Bc °C c_p liq. °K Cc Cryos, Aº °K c_p vap. consts. B° c, vap. te °C 79.49 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 4-Calc, from det. data 3-Lit. 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 14		
NAME	1, cis-4-H	lexad	iene			STRUCTURAL FORMULA			
Mole % Pur.	Ref. Mo	lecul	ar C ₆ H ₁₀	Molecular Weight 82,14	0	сн ₂ =снсн ₂ сн	=CHCH3		
		Ref.			Ref			Ref	
F, P. *C			dt/dP		П	f to			
F.P. 100%			°C/mm		1 _ 1	g <u>*K</u>			
B. P. *C	4.5		25°C BP	0.1363 0.04028	5 4	h ;			
760 mm 100	65.	5	te	0.03499	5	f¹ to			
30	-8.88	5	30 mm	0.5689	5	g'			
10 1	-26.15 -54.71	5	ΔHm cal/g			h¹			
Pressure	-51.11	-	ΔHv cal/g			m to		1	
mm 25°C	167.20	5	25°C	92.97	5 5	n ' °K_			
t _e	908.62	5	30 mm BP	99.03 85.91	5	!		├	
Density	0.700		te (d.e)	84.95	5	m' to			
g/ml 20°C	0.700 0.695	2 2	e (4, 6)	84.93	5	0, =			
d 25 4 30	0.690	4	ΔHv/T _e	20.30	5	Surface tension		-	
a	0.7200	4	d -9 to		5	dynes/cm. 20°C	19.31	5	
ь	-0.0397	4				30	18.19	5	
Ref. Index	1 415	١,	e' j *(<u> </u>		40	17.10	5	
n _D 20°C	1.415	2 2	d g/ml vc ml/g			Parachor [P] 20°C			
30	1.409	4	v _c ml/g t _c °C			30		1	
"C"	0.7888	4	P mm			40 Sugal	246.2	5	
MR (Obs.)	29.385	4	P _c mm PV/RT	+		Exp. L.1.%/wt.	240.2	+-	
MR (Calc.) (nD-d/2)	28.974 1.065	5 2	25°C	0.9860	5	u.			
Dielectric	2,00	5	30 mm BP	1.0000	5 5	Dispersion			
A -9 to	7.01701	4	t _e	0.9550 0.9500	5	Flash Point °C			
B 1 108 °C		4	tc	l		Fire Point		┼	
<u>c </u>	226.6	5	ΔHc kcal/m			M Spec. Ultra V.			
A* -9 to	1.37095	5	ΔHf ΔFf			X-Ray Dif.			
B*	1131.31	5	Viscosity		\vdash	Infrared		<u> </u>	
c			centistokes			Solubility in + Acetone			
tk to	İ		η ∘ο			Carbon tet.			
A' to	 	-				Benzene Ether			
B'		1	<u> </u>			n-Heptane		1	
C'	ļ	ļ	B ^V to			Ethanol Water			
A'* to B'* °C		1		-1		Water in		1	
Ac to	 		, v.'	1					
Bc t C	j	1		+	\vdash				
Cc	·!	L	c _p liq. °K						
Cryos. A° consts. B°			c _p vap. °K						
t _e *C	70.57	5	c _v vap.			+ ~~~~ (100		<u></u>	
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc, from de	t det	grams/100 grants 'grams/100 grants 'grams/100 grants		<u> </u>	
SOURCE:		Al		ue	., ual	- J-Carc. by for			
PURIFICAT	ION:	AJ	***						
	RE REFERE								
			•						

							No. 15	5
NAME	l, trans-	4-Hex	adiene		STRUCTURAL FORMULA			
Mole % Pur.	Ref. M	olecul	ar C ₆ H ₁₀		CH ₂ =CHCH ₂ CH=CHCH ₃			
76 Pur.	F		10 10 1	Weight 82.140				Ref.
F. P. °C	1	Ref.	1.715		Ref.	L	T	Ker.
F.P. 100%	 	+-	dt/dP °C/mm	1		f to g*K_		
B. P. °C			25°C	0.1363	5	h	1	
760 mm 100	65. 13.80	2 5	BP t _e	0.04028 0.03496	5	f' to	†	
30	-8.88	5	90 mm	0.5689	5	g'° <u>K</u>		
10 1	-26.15 -54.71	5	ΔHm cal/g			h'		
Pressure		+-	AHv cal/g			m to		1
mm 25°C	167.20	5	25°C 30 mm	93.02 99.03	5 5	" ' <u>~</u>		
t _e	909.69	5	BP	86.00	5	m' to	 	-
Density g/ml 20°C	0,700	2	te (d. a)	85.03 85.02	5	n' °K		
dt 25	0.695	2	te te (d, e) AHv/Te	20.31	5	0'		
	0.6900	4	d -9 to	97.46	5	Surface tension		
a b	0.7200 -0.0 ₃ 97	4 4	_e_ 7 <u>1 °C</u>	0.1762	5	dynes/cm. 20°C	19.31 18.19	5
Ref. Index	†	1	d' to			40	17.10	5
ⁿ D 20°C	1.415	2	d _c g/ml		H	Parachor [P]		
30	1.412	2 4	II V mila			20°C 3 0		l
"C"	0.7888	4	tc°C			40	246 2	5
MR (Obs.)	29.385	4	P _c mm PV/RT		\vdash		246.2	13
MR (Calc.) (nD-d/2)	28.974 1.065	5 2	25°C	0.9866	5	Exp. L.1.%/wt. u.		1
Dielectric	2.00	5	30 mm BP	1.0000	5 5	Dispersion		
A -9 to	7,01701		t	0.9510	5	Flash Point C Fire Point		1
B 108 °C _ C	1206.11 226.6	4 5	ΔHc kcal/m			M. Spec. Ultra V.		<u> </u>
A* -9 to B* _81 °C K	1.36916 1130.92	5	ΔHf ΔFf			X-Ray Dif. Infrared		
t _k -to			Viscosity centistokes			Solubility in + Acetone		
t _x °C	ļ	4				Carbon tet. Benzene Ether		
B'∟ °C			B _v to		\vdash	n-Heptane		
C1		+	B to A C			Ethanol Water		
A'* to B'* °C			$\frac{ \mathbf{B}^{\mathbf{v}} }{ \mathbf{B}^{\mathbf{v}} } - \frac{ \mathbf{b}^{\mathbf{v}} }{ \mathbf{b}^{\mathbf{v}} }$			Water in		
Acl to			(A ^V) °C					
Bc tc °C	.		c _p liq. °K					
Cryos. A°		+	1					
consts. B°	70, 61	5	c _p vap. °K c _v vap.					
• ~	10.61	13	L •	l	لــــــا	+ arams/100 ===	me ester-	<u> </u>
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da	grams/100 gra ta 5-Calc. by for		
SOURCE:			PI					
PURIFICAT	ION:	A	PI					
LITERATU		ENCES	S:					
į								

						No. 16			
NAME	1,5-Hexadiene					STRUCTURAL FORMULA			
Mole	Ref. Mo	lecul	ar	Molecular		CH ₂ =CH(CH ₂) ₂ CH=CH ₂	2		
% Pur.	Fo	rmul	ar C ₆ H ₁₀	Weight 82.14	10				
		Ref.		_	Ref.		Ref		
F.P. °C	, 	-	dt/dP *C/mm	1		f to oK	ļ		
B. P. *C	-	\vdash	25°C	0.1125	5	g <u>*K</u>			
760 mm 100	59.46 8.95	2 5	BP t _e	0.03976 0.03499	4 5	f' to	\vdash		
30	-13.43	5	30 mm	0.5609	5	g!K			
10 1	-30.44 -58.5 9	5	ΔHm cal/g			h'	L.		
Pressure			ΔHv cal/g	90,27	5	m to K			
mm 25°C	208.01 893.84	5	25°C 30 mm	97.02	5	0			
Density	+ 075.01	<u> </u>	BP	84.29 83.44	5 5	m¹ to			
g/ml 20°C		2	t _e (d, e)	83.42	5	n' <u>*K</u>			
dt 25 4 30	0.6878 0.6833	2 4	ΔHv/T _e	20.30	5		├-		
	0.7104	4	d -13 to		5	Surface tension dynes/cm. 20°C 18.46	5		
b D. C. T. I.	-0.0386	4		5		30 17.47 40 16.50	5		
Ref. Index		2	e' •c		-	Parachor [P]	Ť		
25 30	1.4010 1.3978	2	d g/ml vc ml/g			20°C			
"C"	0.7780	4	ic o			40			
MR (Obs.)		4	P _c mm			Sugd. 246, 2	5		
MR (Calc. (nD-d/2)	28.974 1.0580	5 2	PV/RT 25°C	0.9834	5	Exp. L.1.%/wt.			
Dielectric	1.97	5	30 mm BP	1.0000 0.9560	5	Dispersion	_		
A -13 to	7.00740	4	te	0.9515	5	Flash Point °C Fire Point	ŀ		
B 1 102°C	2 1184.99 227.7	4 5	t _c	 	-	M Spec.			
A* -13 to		5	ΔHf	1		Ultra V. X-Ray Dif.			
B*	1110.81	5	ΔFf Viscosity			Infrared			
¢ .— —.:	_	l	centistokes	1		Solubility in + Acetone			
t _k to			η °C			Carbon tet. Benzene	1		
A' to						Ether			
B' °	4		B ^V to			n-Heptane Ethanol			
A'* to			A ^v C	_]		Water Water in	l		
B'* °C		-	(BV) to	1		4042 711	 		
Bc t °C			(A ^V) °C						
Cryos, A	}	-	P -						
consts. B°			c _p vap. °K						
t _e °C	64.45	5	c _v vap.				L		
D. D. D. D. D. D. D. D. D. D. D. D. D. D						grams/100 grams solven	t		
SOURCE:	CES: 1-Dow	Z-AI	PI 3-Lit, 4-0 PI	Calc. from det	t. da	ta 5-Calc, by formula			
PURIFICA?	TION:		PI						
	RE REFERE								

No. 17 2, 3-Hexadiene STRUCTURAL FORMULA NAME CH2CH=C=CHCH2CH3 Molecular C6H10 Mole Ref. **Molecular** % Pur Weight 82.140 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C °K g 0.1514 5 B. P. °C h ВP 0.04057 4 760 mm 68.0 2 t_e 0.03495 5 f١ to 100 16.42 5 g <u>«к</u> 30 -6.45 5 30 mm 5 0.5734 10 -23.85 -52.65 5 h! AHm cal/g 5 1 to m ∆Hv cal/g Pressure ۰ĸ n 25°C 94.47 mm 25°C 148.45 5 0 30 mm 100.07 5 918, 28 5 t_e BP 86.90 5 m' to Density te te (d, e) 5 85.86 n' °K g/ml 20°C 0.680 2 85.85 5 01 $\mathbf{d_4^t}$ 25 0.675 2 ΔHv/T_e 5 20,32 30 0.670 4 Surface tension -6 98.93 5 d to 0,7000 dynes/cm. 20°C 17.20 5 74 •C 0.1769 5 ь -0.03974 30 16.18 5 a 5 40 15.19 Ref. Index e' [P] 20°C 1.395 2 Parachor ⁿD d_c g/ml 25 2 20°C 1.392 vc ml/g t °C 30 1.389 4 30 t_c 40 "C" 0,7750 4 P_c mm Sugd. 246.2 5 28.95 28.974 MR (Obs.) 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 5 25°C 0.9881 5 5 5 (nD-d/2) 1.055 2 30 mm 1.0000 Dispersion Dielectric 1.95 5 ВP 0.9560 Flash Point C 0.9507 5 A -6 to 7.02235 4 Fire Point 110 °C В 1218.02 M. Spec. С 226.1 5 AHc kcal/m Ultra V. A* -6 to B* 84 °C ΔHf 1.37077 5 X-Ray Dif. ΔFf 1142.22 Infrared ĸ Viscosity Solubility in centistokes Acetone Carbon tet. °C $\mathbf{t_x}$ Benzene ΑΊ to Ether B' °C n-Heptane B_v C' to Ethanol °C Water A'* to Water in (BV) B'* °C to Acl to (A^V) °C Βc °C cp liq. ۰ĸ Cc' Cryos. A° c_p vap. •ĸ consts. B° c_v vap. te °C 5 73.95 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: API **PURIFICATION:** LITERATURE REFERENCES:

NAME	cis-2	, cis	-4-H	exadiene			STRUCTURAL FORMULA			
Mole % Pur.	Ref.		ecul		Molecular Weight 82.14	10	сн ₃ сн=снсн	- с нсн ₃		
			Ref.			Ref	 		Ref	
·	Τ	\neg	V. B.		T	1	-	T	-	
F.P. °C F.P. 100%	 		-	dt/dP *C/mm	1		f to			
	 	-	-	25°C	0, 2325	5	g <u>*K</u>	İ	1	
B. P. °C 760 mm	80.		2	BP	0.04175	4	h		ـــــ	
100	26.89	1	5	t _e	0.03492	5	f' to	ļ		
30	3.32		5	30 mm	0.5912	5	g' ' <u>*</u> K_	4		
10 1	-14.62 -44.34		5	ΔHm cal/g			h'	<u></u>		
	-44, 34			ΔHv cal/g	†	\vdash	m to	ł	ł	
Pressure mm 25°C	91.55		5	25°C	100.38	5	n •K	4	1	
t _e	953.22		5	30 mm	104.30	5	<u> </u>		1	
Density	 		\vdash	BP t	90.57 89.28	5	m¹ to			
g/ml 20°C	0.72		2	te (d, e)	89.25	5	n' °K	1		
at 25	0.71		2	ΔHv/T	20.34	5	o'	1		
4 30	0.71		4			5	Surface tension			
	0.74		4	d 3 to		5	dynes/cm. 20°C	21.65	5	
ь	-0.03	98	4	a, to	-1		30	20.45	5	
Ref. Index		,		e' i °C	:[40	17.20	13	
ⁿ D 20°C	1.45		2 2	d_ g/ml			Parachor [P] 20°C		1	
30	1.44		4	d g/ml vc ml/g	1	1	30	1		
"C"	0,82		4	, c			40	1	1	
	30,65		4	P _c mm	1	1 1	Sugd	246.2	5	
MR (Obs.) MR (Calc.)			5	PV/RT			Exp. L.1.%/wt.			
(nD-d/2)	1.09		2	25°C	0.9937	5	u.	225		
Dielectric	2.10		5	30 mm BP	1.0000 0.9565	5 5	Dispersion	225.	2	
A 3 to	7.03	864	4	te	0.9502	5	Flash Point °C Fire Point			
B 1128 °C		- 1	4	tc	<u> </u>	\sqcup	M Spec.	 	+	
<u>C</u>	223.8		5	ΔHc kcal/m ΔHf	1	1 1	Ultra V.	İ		
A* 3 to	1.37		5	ΔFf	1		X-Ray Dif.	ļ	1	
B*	1184.88		5	Viscosity	 	$\vdash \dashv$	Infrared		ــــــ	
c	1			centistokes			Solubility in +	ļ	1	
tk to	1			η °c		1 1	Acetone Carbon tet.		ĺ	
t _x i °C				'		1 1	Benzene	ľ		
A' to						1 1	Ether	ļ		
B', ∟ _ <u>°</u> ⊆	ł			B ^V to	 	$\vdash \dashv$	n-Heptane			
	 		-	B' to			Ethanol Water	ľ		
A'* to B'* °C	1	j			-		Water in	l	1	
Ac to	 		\vdash	l v. '						
Bci t C	!				+	\vdash		1		
Cc - c-	1			c _p liq. °K	1					
Cryos. A°				c _p vap. °K						
te °C	87.36	-	5	c, vap.						
<u>-e </u>	1 67.36		_ع_ا		L		+ grams/100 gra	me solve-	<u></u>	
REFERENC	ES: 1-D	ow :	2-AF	PI 3-Lit. 4-0	Calc. from de	t. dat	a 5-Calc. by for			
SOURCE:			AI	PI						
PURIFICAT	ION:		AI	PI						
LITERATU	RE REFI	EREN	ICES	:						

No. 19 cis-2, trans-4-Hexadiene NAME STRUCTURAL FORMULA CH3CH=CHCH=CHCH3 Molecular C6H10 Mole Ref. Molecular % Pur. Weight 82.140 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C °<u>K</u> g 0.2325 0.04175 5 4 B. P. °C h ВP 760 mm 80. 2 0.03498 5 ſ١ 100 26.89 5 to <u>°К</u> 3.32 5 5 g' 30 0.5912 30 mm 10 -14.62 5 h' ∆Hm cal/g 5 1 -44.34 to AHv cal/g Pressure n ۰ĸ 100.31 25°C 5 91.55 mm 25°C o 30 mm 104.30 5 5 950.95 5 te ΒP 90.38 m to Density 89.10 5 te te (d, e) n' °K g/ml 20°C 89.06 5 0.720 2 o' $\mathbf{d_{4}^{t}}$ 25 0.715 2 ΔHv/T_e 20,30 5 30 0.710 4 Surface tension d 3 104.90 5 0.7400 a b dynes/cm. 20°C 21.65 5 °C 0.1816 5 ᇷᅥ 87 -0.0398 4 30 20.45 5 to 19.28 5 40 Ref. Index e' °C 20°C 1.450 2 Parachor [P] n_D d_c g/ml 25 1.447 2 20°C vc ml/g t °C 30 1.444 4 30 $\mathbf{t_c}$ "C" 40 0.8277 4 P_c mm 5 246.2 Sugd. MR (Obs.) 30,658 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30,244 5 25°C 0.9930 (nD-d/2) 1.090 2 1.0000 30 mm 5 Dispersion 225. 2 Dielectric 2.10 5 BP 0.9545 5 Flash Point C 0.9481 A 3 to 7,03864 t_e Fire Point B | 127 °C 1263.15 M. Spec. c 223.8 5 AHc kcal/m Ultra V ΔHf A*Ī 3 to 1.37555 X-Ray Dif. ΔFf B*| 97 °C 1185.70 Infrared K Viscosity Solubility in centistokes to Acetone Carbon tet. ٠c $\mathbf{t_x}$ Benzene A۱ to Ether B١ °C n-Heptane B^v | C' to Ethanol °C A'* Water to ٠c Water in B'* (B^V)| to Acl (A^V) to °C Bc °C c_p liq. ۰ĸ Cc Cryos. Aº c_p vap. °K consts. B° te °C c, vap. 87.28 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API API **PURIFICATION:** LITERATURE REFERENCES:

							No. 20	
NAME	trans	s-2, trans	-4-Hexadiene		_	STRUCTURAL		
Mole % Pur.	Ref.	Molecul Formul	ar C ₆ H ₁₀	Molecular Weight 82.14	10	сн ₃ сн=снсн	=CHCH ₃	
		Ref.			Ref			Ref
F.P. °C			dt/dP			f to		T
F.P. 100%			°C/mm			g K		1
B. P. °C	T		25°C BP	0.2325 0.04175	5 4	h ¦		1
760 mm 100	80. 26.89	2 5	t.	0.03498	5	f' to		
30	3.32	5	30 mm	0.5912	5	g'	1	
10 1	-14.62 -44.34		AHm cal/g		П	h'		
	+	<u> </u>	ΔHv cal/g	 	\vdash	m to		
Pressure mm 25°C	91.55	5	25°C	100.31	5	n <u>*K</u>	-	İ
te	950.95		30 mm BP	104.30 90.38	5 5	<u> </u>	ļ	
Density			t	89.10	5	m' to		
g/ml 20°C			'e (u, u,	89.06	5	n' ' <u>°</u> K		
dt 25 4 30	0.71		AHv/Te	20.30	5			
	0.74		d 3 to	104.90	5	Surface tension	21.65	5
b	-0.03		-å, ⁸⁷ to		5	dynes/cm. 20°C	20.45	5
Ref. Index			d' to			40	19.28	5
n _D 20°C			d _c g/ml			Parachor [P]		
25 30	1.44		II V. mi/g			20°C 30		1
"C"	0, 82		tc °C			40		
MR (Obs.)			P _c mm			Sugd.	246.2	5
MR (Calc.			PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.09		25°C 30 mm	0.9930 1.0000	5 5	u. Dispersion	225.	2
Dielectric	2.10	5	BP	0.9545	5	Flash Point °C	225.	+-
A 3 to			t _e	0.9481	5	Fire Point		İ
B (_127°C			t _c	ļ	\vdash	M Spec.		+-
	223.8	5	ΔHc kcal/m	1		Ultra V.		
A* 3 to B* 97 °C			ΔFf	ļ	1. 1	X-Ray Dif. Infrared		1
K	-		Viscosity					+
t. to		į	centistokes	.		Solubility in + Acetone		ļ
t _x to			7 •	'	1 1	Carbon tet.		
A' to	,		ġ.		1	Benzene Ether		
B' '	2		B ^V to		\vdash	n-Heptane		
C'			B ^V to			Ethanol Water		1
A'* to B'* *C				-		Water in		
Ac to			[[v. '	1				
Bc t *C					╀──┤			
Cc — —	-		c _p liq. ∘K					
Cryos. A° consts. B°			c _p vap. °K					
t _e °C	87, 28	5	c _v vap.	<u> </u>		L ₊	<u> </u>	
REFEREN	TEC. 1 D	2 AT	DF 2 F 1			grams/100 gra		nt
SOURCE:	J-10. 1-D	A1		Calc. from de	t. dat	ta 5-Calc. by for	mula	
	PION:		PI					
PURIFICA:								
Z. Z. Z. Z. Z. Z. Z. Z. Z. Z. Z. Z. Z. Z	KE KEF	BRENCES	·					

No. 21 3-Methyl-1, 2-pentadiene NAME STRUCTURAL FORMULA CH2=C=CCH2CH3 Molecular C6H10 ĊH Mole Ref. Molecular Weight 82,140 % Pur Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm °K g 25°C 0.1625 5 B.P. °C h ВP 0.04077 4 760 mm 70. 2 0.0350 5 ſ١ 100 18.17 5 °<u>K</u> g' 30 -4.82 30 mm 0.5764 5 10 -22.31 5 h' ∆Hm cal/g 1 -51.26 5 m to ∆Hv cal/g Pressure n °K 25°C 95.38 mm 25°C 137.1 5 o 30 mm 100.78 5 t. 922.4 5 BP 5 87.38 m¹ Density to 86.30 5 n' g/ml 20°C °K_ te (d, e) 0.715 2 5 86,28 ۰' d_4^t 25 0.710 2 ΔHv/T 20.29 5 30 • 0.705 4 Surface tension d -5 99.91 0.7350 4 4 dynes/cm. 20°C 21.03 °C 0.1791 76 -0.0398 ь 30 19.85 5 to 40 18.69 5 Ref. Index e¹ °C 20°C 1.425 2 ^{n}D [P] Parachor d_c g/ml 25 1.422 2 ۍ. °C 20°C ml/g 30 4 1.419 30 $\mathbf{t_c}$ 40 "C" 0.7898 4 P_c mm 5 Sugd. 246.2 MR (Obs.) 29.375 4 PV/RT Exp. L. 1. %/wt. MR (Calc.) 5 28,974 25°C 0.9884 1.0000 5 (nD-d/2)1.068 2 30 mm Dispersion Dielectric 2.03 5 BP 0.9545 Flash Point C 0.9490 A -5 to 7.02488 4 Fire Point B [116 °C] 1225.40 4 M. Spec. C 225.7 5 ∆Hc kcal/m Ultra V ΔHf A* -5 to 1.37362 5 X-Ray Dif. ΔFf B* 86 °C 1149.83 Infrared ĸ Viscosity Solubility in centistokes Acetone tō tk tx Carbon tet. °C Benzene A١ to Ether В' °C n-Heptane $\overline{B_{\mathbf{v}}^{\mathbf{v}}}$ C to Ethanol °C Water A'* to Water in B'* °C (B^V)| to Acl (AV) to °C Βc °C cp liq. ۰ĸ Cc Cryos. A° c_p vap. °K consts. B° c vap. te °C 76.12 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

						N	o. 22
NAME	4-Methyl-	1,2-	pentadiene			STRUCTURAL FOR	MULA
						CH2=C=CHCH C	н.
Mole	Ref. Mo	lecul		Molecular		Ċн ₃	3
% Pur.		rmul		Weight 82.14	0	J	
		Ref.			Ref.		Ref
F.P. °C F.P. 1007			dt/dP			f to	ŀ
B. P. *C		-	*C/mm 25°C	0.1625	5	g <u>*K</u>	l
760 mm	70.	2	BP	0.04077 0.03498	4 5	f' to	
100 30	18.17 -4.82	5	t _e 30 mm	0.5764	5	g' 'K_	
10 1	-22.31 -51.26	5	ΔHm cal/g		\Box	h'	
Pressure	-51.20	-	ΔHv cal/g	1		m to	
mm 25°C	137.08	5	25°C 30 mm	95.40 100.78	5 5	ö	
Deneity	922.91	5	BP	87.42	5	m' to	
Density g/ml 20°0	0.708	2	te (d, e)	86.35 86.32	5	n' K_	
dt 25	0.703 0.698	2 4	AHV/T	20.30	5	0'	
	0.7280	4	d -5 to		5	Surface tension dynes/cm. 20°C 20	0.22 5
ь	-0.0398	4	-a, -76 %	5	5	30 19	9.07 5 7.95 5
Ref. Index		2	e' i °(<u> </u>	\vdash	Parachor [P]	1.73
25	1.421	2	d _c g/ml v _c ml/g			20°C	
"C"	0.7959	4	tc °C	1		30 40	l
MR (Obs.)	+	4	P _c mm	<u> </u>		Sugd. 246	5, 2 5
MR (Calc.) 28.974	5 2	PV/RT 25°C	0.9886	5	Exp. L.1.%/wt.	
(nD-d/2) Dielectric	2,03	5	30 mm BP	1.0000	5	Dispersion	
A -5 to		4	t _e	0.9550 0.9495	5	Flash Point °C Fire Point	1
B L115*9	1225.40	4 5	t _c	<u> </u>		M Spec.	
A* -5 to	+	5	ΔHc kcal/m ΔHf			Ultra V.	1
B* 86°0	1149.63	5	ΔFf			X-Ray Dif. Infrared	
K	1		Viscosity centistokes			Solubility in +	
t _k tc			η •	;		Acetone Carbon tet.	
t _x ' °C						Benzene Ether	
B' °	2		B ^V to	 		n-Heptane	Ì
A'* to			B' to			Ethanol Water	
B'* °((B ^V) to	7		Water in	
Ac to			(A ^V) •c	:			
Bc t _c °C			c _p liq. °K	·			
Cryos, A° consts, B°			с _р va p. °К				
t _e °C	76.14	5	c _v vap.				
REFEREN	CES: 1-Dow	2 . 41	PI 3-Lit. 4-	Cala (==== 1 :		grams/100 grams	solvent
SOURCE:			PI 3-Lit. 4-	Carc. from de	. aai	ta 5-Calc. by formula	1
PURIFICA	TION:		PI				
	RE REFERE						

No. 23 2-Methyl-1, cis-3-pentadiene NAME STRUCTURAL FORMULA CH2=CCH=CHCH3 Molecular C6H10 ċн₃ Ref. Mole Molecular % Pur. Weight 82,140 Formula Ref Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm 25°C g <u>.°K</u> 0.2012 5 B. P. °C h ВP 0.04136 4 760 mm 76. 2 ^te 0.03497 5 ſ١ 100 to 23.40 5 g' °<u>K</u> 30 5 0.5853 5 0.64 30 mm 10 -17.70 5 h' AHm cal/g 5 -47.11 m to AHv cal/g Pressure °K n 25°C 98.35 5 mm 25°C 107.71 5 o 30 mm 102.88 5 940.07 5 te ΒP 89.22 5 m' to Density te (d, e) 5 5 88 03 °K g/ml 20°C 0.719 2 87.99 o' 25 0.714 2 $\mathbf{d_{4}^{t}}$ AHv/T 20.31 5 30 0.709 4 Surface tension ī to 102.89 5 0,7390 dynes/cm. 20°C a 21.52 5 <u>°С</u> 0.1799 -0.0398 ь 4 20.32 5 30 ď٠ to 5 40 19.16 Ref. Index e' ⁿD 20°C 1.446 2 [P] Parachor dc g/mi 25 1.443 2 20°C vc ml/g t °C 30 1.440 4 30 tc "C" 40 4 0.8219 P_c mm Sugd 246.2 5 30, 465 MR (Obs.) 4 PV/RT Exp. L. 1. %/wt. MR (Calc.) 30.244 25°C 0.9915 (nD-d/2)1.087 2 30 mm 1.0000 5 2 Dispersion 225. Dielectric 2.09 5 BP 0.9550 5 Flash Point °C 0.9490 te tc 7.03380 l to 4 Fire Point B (123 °C) 1248.39 M. Spec. 224.6 AHc kcal/m Ultra V. ΔHf A* 1 to 1.37445 5 X-Ray Dif. ΔFf 9<u>3</u> ℃ 1171.46 B* Infrared ĸ Viscosity Solubility in c centistokes Acetone to °C ٠c Carbon tet. Benzene A' to Ether B °C n-Heptane B_V | C' to Ethanol °C Water A'* to Water in (BV) B!* °C to Acl to (AV) °C Bc_1 °C c_p liq. ۰ĸ Cc Cryos, A° ٩K cp vap. consts. B° c_v vap. te °C 82.84 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

No. 24 NAME 2-Methyl-1, trans-3-pentadiene STRUCTURAL FORMULA CH2=CCH=CHCH3 Molecular C6H10 ĊH3 Mole Ref. Molecular Weight 82.140 % Pur. Formula Ref. Ref. Ref F.P. °C F.P. 100% dt/dP to °C/mm <u>°</u>K g 25°C 0.2012 5 B. P. °C h BP 0.04136 4 760 mm 76. 2 0.03499 5 ſ١ to ŧ. 100 23, 40 5 °K g' 30 0.06 5 30 mm 5 0.5853 10 -17.70 5 h' ΔHm cal/g 1 -47.11 5 to m ΔHv cal/g Pressure °K n 25°C 98.33 5 mm 25°C 107.71 5 o 30 mm 102.88 939.51 5 t_e BP 5 89.17 m١ to Density 87.98 5 te (d, e) ٩ĸ g/ml 20°C 0.719 2 5 87.94 01 25 0.714 2 ď4 ΔHv/Te 20.30 5 30 0.709 4 Surface tension 102.89 0 to 5 0.7390 4 21.52 dynes/cm. 20°C 83 •C 0.1805 ь 4 -0.0398 5 30 20.32 ď to 40 19.16 5 Ref. Index •C 20°C 1.446 [P] 2 $\mathbf{n}_{\mathbf{D}}$ Parachor d_c g/ml 25 2 4 1.443 20°C ml/g 1.440 t_c 30 30 •C 40 "C" 0.8219 4 Sugd. 5 mm 246.2 MR (Obs.) 30.465 4 PV/RT Exp. L.1. %/wt. MR (Calc.) 30,244 5 25°C 0.9913 5 ź (nD-d/2)1.087 30 mm 1.0000 Dispersion 2 5 225. Dielectric 2.09 5 BP 0.9545 5 Flash Point °C 0.9485 7.03380 0 to 4 Fire Point tc В 123°C 1248.39 4 M Spec. C 224.6 5 AHc kcal/m Ultra V ΔHf A*| 0 to 1.37535 5 X-Ray Dif. ΔFf B* 93 °C 1171.66 Infrared ĸ Viscosity Viscos.., centistokes °C Solubility in Acetone t_x to Carbon tet. •c Benzene A' to Ether B١ •c n-Heptane Ċ١ В Ethanol ·c A'* Water to Water in (BV) B'* •c to Acl to (AV) °C Bc ٠c cp liq. °K Cc Cryos. A° consts. B° c_p vap. °K t_e °C c, vap. 82.82 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 25

NAME	3-Methyl	-l,ci	s-3-pentadiene		STRUCTURAL	STRUCTURAL FORMULA				
Mole	Ref. Mo	lecul		Molecular		сн ₂ =сн с=сі сн ₃				
% Pur.	Fo		6-10	Weight 82.140	-			1		
	·	Ref.		 	Ref.	<u> </u>		Ref.		
F.P. °C F.P. 100%			dt/dP			f to				
	 	+	°C/mm 25°C	0.2086	5	g '° <u>K</u>				
B. P. °C 760 mm	77.	2	BP	0.04145	4	h		-		
100	24.27	5	t _e	0.03499	5	f' to				
30	0.88	5 5	30 mm	0.5868	5	g'° <u>K</u>				
10	-46.42	5	ΔHm cal/g			h'		 		
Pressure		1-1	ΔHv cal/g			m to				
mm 25°C	103.43	5	25°C 30 mm	98.83 103.24	5	o				
t _e	942. 37	5	BP	89.50	5	m' to		+		
Density	0.725	1,1	t _e , ,	88.26	5	m' to				
g/ml 20°C	0.735 0.730	2 2	t _e (d, e)	88, 25	5	0'				
d ₄ 30	0.725	4	ΔHv/T _e	20, 30	5	Surface tension		\vdash		
a	0.7550	4	d 1 to e 84 °C	103.39 0.1804	5	dynes/cm. 20°C	23.51	5		
ь	-0.0 ₃ 98	4	-84 °C to	0.1504		8 30	22.23 20.98	5		
Ref. Index	1 453	,	e' °C			40	20.76	1 3		
n _D 20°C	1.452 1.449	2 2	d g/ml v ml/g	1		Parachor [P] 20°C				
30	1.446	4				30				
"C"	0.8142	4	tc°C Pcmm			40 Sugd.	246.2	5		
MR (Obs.) MR (Calc.)	30.148 30.244	4 5	PV/RT		\vdash	Exp. L.1.%/wt.				
(nD-d/2)	1.085	2	25°C	0.9917	5	u.				
Dielectric	2,11	5	30 mm BP	1.0000 0.9545	5 5	Dispersion	225.	2		
A 1 to B 126 °C	7.03502 1252.08	4 4	te tc	0.9484	5	Flash Point °C Fire Point				
_ c	224.4	5	ΔHc kcal/m			M. Spec. Ultra V.		1		
A* 1 to	1.37541	5	ΔHf ΔFf			X-Ray Dif.				
B* <u>94</u> °C K	1175.17	5	Viscosity	·	\vdash	Infrared		ļ		
c			centistokes			Solubility in *Acetone				
t _x °C			າ ℃			Carbon tet. Benzene				
A' to B' °C				1		Ether				
c,'			B ^V to A ^V °C			n-Heptane Ethanol				
A'* to		1	A I °C			Water				
B'* °C			(B ^V) to			Water in		┼		
Acl to			(A ^V) °C							
Bc_t _{c_} °C			c _p liq. *K							
Cryos, A° consts, B°			c _p vap. *K							
t _e °C	83.93	5	c _v vap.							
						f grams/100 gra	ms solve	nt		
REFERENC	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula			
SOURCE:			PI							
PURIFICAT			PI							
LITERATUI	RE REFER <u>e</u>	NCES	: -							

No. 26 NAME 3-Methyl-1, trans-3-pentadiene STRUCTURAL FORMULA CH2=CHC=CHCH3 Molecular C6H10 ĊH, Ref. Mole Molecular % Pur. Weight 82,140 Formula Ref Ref. Ref F.P. °C F.P. 100% dt/dP to °C/mm °K g 25°C 0.2086 5 B. P. °C h 4 BP 0.04145 77. 760 mm 2 t_e 0.03499 5 24.27 ſ١ to 5 100 g' °К 30 0.88 5 30 mm 0.5868 5 10 -16.93 h' ∆Hm cal/g -46.42 5 1 to m AHv cal/g Pressure °K n 25°C 98.83 mm 25°C 103.43 5 0 30 mm 103.24 5 te 942.37 5 ВP 89.50 5 Density g/ml 20°C m to 88.26 5 te (d, e) ۰ĸ 0.735 2 88, 25 5 ۰, 0.730 2 $\mathbf{d_{4}^{t}}$ ΔHv/T 20.30 5 30 4 0.725 Surface tension 103, 39 ı to 0.7550 4 23.51 dynes/cm. 20°C *C 0.1804 -0.0398 84 ь 4 5 30 22.23 40 20.98 5 Ref. Index °C $\mathbf{n}_{\mathbf{D}}$ 20°C 1.452 [P] d vc tc Parachor g/ml 25 1.449 2 ml/g 30 1.446 4 30 °C 40 "C" 0.8142 4 5 Sugd. 246. 2 mm MR (Obs.) 30.148 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30, 244 25°C 0.9917 5 2 1.085 (nD-d/2) 1.0000 30 mm 5 2 Dispersion 225. 2.11 5 Dielectric 0.9545 BP 5 Flash Point °C 0.9484 1 to 7.03502 4 Fire Point tc <u> 126 °C</u> 1252.08 С 224.4 5 M Spec. AHc kcal/m Ultra V ΔHf 1.37541 A* l to 5 X-Ray Dif. ΔFf B* <u>94</u> ℃ 1175.17 Infrared ĸ Viscosity Viscos.., centistokes °C Solubility in Acetone to t_x Carbon tet. •c Benzene A' to Ether B١ •c n-Heptane Ċ١ В to Ethanol Á °C A'* Water to Water in B'* ٠c (BV) to Acl to (AV) °C Bc °C cp liq. °K Cc Cryos. A° consts. B° c_p vap. °K t_e °C c, vap. 83.93 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 27 4-Methyl-1, 3-pentadiene NAME STRUCTURAL FORMULA CH2=CHCH=CCH3 Molecular C6H10 Ċнз Molecular Weight 82.140 Mole Ref. % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g <u>°K</u> 0.2034 5 B. P. °C h ВP 0.04139 760 mm 76.3 2 5 t_e 0.03499 5 f 100 23.66 to 5 g' <u>°К</u> 30 0.31 0.5857 5 30 mm 10 -17.47 5 h' ∆Hm cal/g 1 -46.91 5 m to ∆Hv cal/g Pressure n ۰ĸ 25°C 98.47 mm 25°C 106.42 o 30 mm 102.99 940.37 5 t_e ВP 5 89, 26 m' to Density te te (d, e) 5 88.06 'n °K g/ml 20°C 0.719 2 88.02 5 ٥' 25 0.714 2 $\mathbf{d_{4}^{t}}$ ΔHv/Te 20.30 5 30 0.709 4 Surface tension 0 to 103.04 5 0.7390 a dynes/cm. 20°C 21,52 ٠Ç <u>83</u> 0.1806 -0.0398 ь 4 20.32 5 30 ď٠ to 40 5 Ref. Index 19.16 e ¹ 20°C 1.451 2 [P] ^{n}D d g/ml vc ml/g tc °C Parachor 25 1.448 2 Z0°C 30 1.445 4 30 t_c 40 "C" 0.8306 4 P_c mm Sugd 246.2 5 MR (Obs.) 30.760 4 PV/RT Exp. L. l. %/wt. MR (Calc.) 30, 244 25°C 0.9914 (nD-d/2)1.092 2 30 mm 1.0000 Dispersion 2 225 Dielectric 5 2.10 BP 0.9545 Flash Point C 0.9485 0 to te t 7.03324 4 Fire Point 1249.05 B | 123 °C 4 M. Spec. С 224.5 5 AHc kcal/m Ultra V ΔHf A*I 0 to 1.37449 5 X-Ray Dif. ΔFf B* 9<u>3 °C</u> 1172.30 Infrared ĸ Viscosity Solubility in centistokes $\mathbf{t_k}$ Acetone to °C Carbon tet. °C Benzene A to Ether B ٠c n-Heptane B^V | C' to Ethanol °C Water to B'* Water in ۰c (BV) to Acl to (A^V) °C Bc ۰c c_p liq. ۰ĸ Cc Cryos. A° ۰ĸ c_p vap. consts. B° c vap. t_e °C 5 83.15 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 28	
NAME	2 - Met	hyl-1,4-1	pentadiene			STRUCTURAL	FORMUL	A
						CH2=C CH2C	н=Сн	
Mole	D-6	M-11		Molecular		Ċн ₃	2	
% Pur.	Ref.	Formul	C ₆ H ₁₀	Weight 82.14	10	3		
		Ref			Ref			Ref.
F. P. °C			dt/dP			f to		T
F.P. 100%			°C/mm	į.		g L°K_		
B. P. °C			25°C BP	0.1000 0.03940	5 4	h ;	l	1
760 mm 100	56. 5.95	2 5	t _e	0.03496	5	f' to		T
30	-16.22	5	30 mm	0.5557	5	g'	ļ	
10 1	-33.07 -60.96	5	ΔHm cal/g			h'		<u> </u>
Pressure	-00.70		ΔHv cal/g	1		m to		
mm 25°C	237.83	5	25°C	88.68	5	n <u>*K</u> -	ł	
t _e	884.46	5	30 mm BP	95.84 83.34	5	<u> </u>		┼
Density	0 (0		to (d.e)	82.56	5	m' to		1
g/ml 20°C ,t 25	0.69		e (4, 5)	82.54	5	0'	1	1
d 25 4 30	0.68		ΔHv/T _e	20, 31	5	Surface tension	 	+
a	0.71		d -16 to e 61 °C		5	dynes/cm. 20°C	18.63	5
b	-0.03	96 4	_d'to	1		30 40	17.53 16.45	5
Ref. Index	1.40	5 2	e' i •c		L	Parachor [P]	10.43	+-
D 25	1.40	2 2	d g/ml v ml/g			20°C	ļ	1
30	1.39		v _c ml/g t _c °C			30		ŀ
"C"	0.77	75 4	P _c mm	ļ		40 Sugd.	246.2	5
MR (Obs.) MR (Calc.)	29.00		PV/RT	<u> </u>	\vdash	Exp. L.1.%/wt.	<u> </u>	+
(nD-d/2)	28.97		25°C	0.9816	5	ū.		
Dielectric	1.97	5	30 mm BP	1.0000 0.9565	5	Dispersion		┼
A -16 to	7.00	507 4	t _e	0.9523	5	Flash Point °C Fire Point		
B		4	tc	<u> </u>		M Spec.		+-
C 1/1	228.4	5	ΔHc kcal/m ΔHf			Ultra V.		
A* -16 to B* 71 °C	1.36	730 5	ΔFf	l		X-Ray Dif. Infrared		1
к ———			Viscosity			Solubility in +		+-
tk to			centistokes 7°C			Acetone		1
t _x c			'	1		Carbon tet. Benzene		
A' to						Ether	j	
B', ∟ _ <u>°</u> C			B ^V to			n-Heptane Ethanol		
A'* to	 		AV I C			Water		
B'* °C			(B ^V) to	1		Water in		┿
Ac to			(A ^V) °C	1				
Bc tc C			c _p liq. °K					
Cryos, A°	†		3			1		
consts. B°	L		р					
t _e °C	60.61	5	c _v vap.	L		+ (122	<u></u>	<u></u>
REFERENC	ES: 1-De	w 2-AF	PI 3-Lit. 4-0	Calc. from de	t. de	grams/100 granta 5-Calc. by for:		IT.
SOURCE:		API						
PURIFICAT	ION:	API						
LITERATUR								

							No. 29	·
NAME _	3-Methyl-	1,4-	pentadiene			STRUCTURAL	FORMUL	A
						сн,=снсн	сн=сн,	
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 82.14	0	Сн ₃		
		Ref.			Ref.			Ref.
F.P. °C F.P. 100%		<u> </u>	dt/dP °C/mm 25°C	0.0966	5	f to		
B. P. °C 760 mm 100	55. 5.08	2 5	BP t _e	0. 03930 0. 03498	4 5	f' to		-
30 10	-17.02 -33.83	5 5	30 mm	0,5541	5	g'° <u>K</u>		
1	-61.64	5	ΔHm cal/g	 		m to		
Pressure mm 25°C t _e	247.21 881.08	5	ΔHv cal/g 25°C 30 mm	88.17 95.51	5	n *K		
Density g/ml 20°C	0.695	2	t _e (d, e)	83.00 82.24 82.23	5 5 5	m' to		
d ₄ 25 30	0.690 0.685	2	ΔHv/T _e	20.30	5			
a b	0.7151 -0.0 ₃ 96	4	d -17 to e 59 °C d to	92.55 0.1736	5 5	Surface tension dynes/cm. 20°C 30	18.73 17.63	5 5
Ref. Index n _D 20°C 25	1.405 1.402	2 2	e' °C	-	-	Parachor [P]	16.54	5
30	1.399 0.7764	4	t _c °C			30 40	244.2	_
MR (Obs.)	28.967	4	P _c mm	 		Sugd. Exp. L.1.%/wt.	246.2	5
MR (Calc.) (nD-d/2) Dielectric	1.058	5 2	25°C 30 mm BP	0.9806 1.0000	5	u. Dispersion		
A -17 to	7,00219	5 4	l t	0.9560 0.9519	5	Flash Point C Fire Point		
B _97 °C_ C	1168.41 228.5	4 5	t _c ΔHc kcal/m			M. Spec. Ultra V.		\vdash
A* -17 to B* 69 °C K	1.36668 1095.11	5 5	ΔHf ΔFf Viscosity			X-Ray Dif. Infrared		
t _k -tō			centistokes 7 °C			Solubility in *Acetone Carbon tet. Benzene		
A' to B' C						Ether n-Heptane		
A ¹ * to			B ^V to A ^V *C			Ethanol Water		
B'* °C	 		(B ^V) to (A ^V) °C			Water in		\vdash
Bc t _c °C			c _p liq. °K					
Cryos. A° consts. B°			c _p vap. •K					
t _e °C	59.49	5	c _w vap.		<u> </u>	L		
REFERENC	ES: 1-Dow	2-A	PI 3-Lit, 4-	Calo from 3:		grams/100 gra		it
SOURCE:	EG. 1-10W	AP		Caic. Irom de	t. U	ata 5-Calc, by for	111418	
PURIFICAT	ION:	AP						
	RE REFERE							

												No. 30)
NAME	2-M	ethyl-	-2,3-	pentad	iene				ST	RUCTURA	L I	FORMU L	A
										CH ₃ C = C	=CH	ICH.	
Mole	Ref	Mo	lecul	9.7		Molecula				ċн,	-	3	
% Pur.	Ker.		rmul		5 ^H 10	Weight		0					
			Ref.					Ref					Ref
F.P. °C F.P. 1007				dt/d					f	1	to		
B. P. *C	-		-	25*			744	5	g	'	<u>•K</u>		
760 mm	72.	_	2	BP			4097 3498	4 5	f'	 	to		+
100 30	19.9		5	t _e	mm	0.5		5	g'	l '	_K_		1
10 1	-20.7 -49.8	7	5		cal/g				h'	<u> </u>			
Pressure	-47.8	<u>'</u>	3		cal/g				m	1	to °K		İ
mm 25°C	126.5		5	25°	C mm	96.3		5 5	n o	! '	<u>_r</u> _		
t _e	928.6	3	5	BP		88.0	1	5	m'		to		+-
Density g/ml 20°C	0.7	11	2	te ((d, e)	86.9		5	n'	i '	<u>"K</u>		
dt 25 4 30	0.7 0.7	06	2 4	ΔH	v/T _e	20.3		5	0'	<u>i</u>			<u> </u>
a 30	0.7		4	d !	-3 to	100.9	1	5		ace tensies/cm. 20		20.57	5
ь	-0.0		4	-읍,	_78_ ° C		792	5	3,11	30)	19.41	5
Ref. Index		25	2	6'	•(-	40 achor [P]		18,28	5
D 25	1.4		2	d _c g	g/ml				Par)°C		1
30	1.4		4	t _c	C C	1				30 40			
"C"	0.7		4	P _c r	nm						igd.	246.2	5
MR (Obs.) MR (Calc.			5	PV/1		0.9	904	5	Exp	. L.1.%/v	vt.		
(nD-d/2)	1.0		2	30 1	mm	1.0	000	5	Dis	u. persion			
Dielectric		3 2740	5 4	BP t _e		0.9		5		h Point °	С		\top
A -3 to	1232.7	8	4	tc						Point			- }
C	225.3		5	ΔHc ΔHf	kcal/m				M S Ultr	pec. a V.			
A* -3 to B* 88 °C		7287 4	5	ΔFf						ay Dif. ared			i.
к ——-	-				osity					bility in	+		╁
k				η	istokes °C	:			Ac	etoné rbon tet.			1
x '				'					Be	nzene			
A' to									Etl n-l	ner Heptane			
C'			1	B ^V	to °C			1 1	Etl	anol			1
A'* to				(BV)		-				ter ter in			1
Ac to	+			(A ^V)		į.							Г
Bc Ltc_°	<u> </u>			c _p li				\vdash					
Cryos. A	- 		\vdash	c _p v		1							
consts. B			_	c _v v									
t _e °C	78.3	1	5		·	<u> </u>			+	/: 00		<u> </u>	<u> </u>
REFEREN	CES: 1-I	Dow	2-AI	PI 3-1	Lit. 4-	Calc. fro	m dei	t. dat		Calc. by			ıt
SOURCE:				PI									
PURIFICA	TION:		A1	PI									
LITERATU	RE REF	ERE	NCES	i:									

							No. 31	
NAME	2 - Ethyl - 1	3-bı	utadiene			STRUCTURAL	FORMULA	
			·			CH ₂ =C CH=	CH,	
Mole % Pur.		ecul mula		Molecular Veight 82,140		с́ ₂ н ₅		
	1	Ref.			Ref.		R	ef.
F.P. °C F.P. 100%			dt/dP °C/mm 25°C	0, 1941	5	f to		
B. P. °C 760 mm 100 30	75. 22.53 -0.74	2 5 5	BP t _e 30 mm	0.04126 0.03499 0.5837	4 5	f' to g' °K		
1 0 1	-18.46 -47.79	5	AHm cal/g			h'		
Pressure mm 25°C t _e	112.14 936.65	5	ΔHv cal/g 25°C 30 mm BP	97.84 102.55 88.88	5 5 5	m to		
Density g/ml 20°C dt 25 d4 30	0.717 0.712 0.707	2 2 4	te te (d, e) AHv/Te	87.70 87.67 20.30	5 5	m' to *K		
a b Ref. Index	0.7370 -0.0 ₃ 98	4	d 0 to e 82 °C d' to	102.41 0.1805	5 5	Surface tension dynes/cm. 20°C 30 40	20.09	5 5 5
ⁿ D 20°C 25 30	1.445 1.442 1.439	2 2 4	d _c g/ml v _c ml/g t _c °C			Parachor [P] 20°C 30 40		
"C"	0,8225	4	P _c mm			Sugd.	246.2	5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	30.490 30.244 1.086	4 5 2	PV/RT 25°C 30 mm BP	0.9908 1.0000	5	Exp. L.1.%/wt. u. Dispersion	225.	2
A 0 to B 121 °C	7.03112 1243.85	4 4	te t	0.9545 0.9486	5	Flash Point C		
C A* 0 to B* 92 °C	1.37396 1167.37	5 5 5	ΔHc kcal/m ΔHf ΔFf			M. Spec. Ultra V. X-Ray Dif. Infrared		
K c t _k to c t _x °C			Viscosity centistokes n °C			Solubility in + Acetone Carbon tet. Benzene		
A' to B' °C C' - C			B ^V to A ^V *C			Ether n-Heptane Ethanol Water		
B¹* °C			(B ^V) to (A ^V) °C			Water in		
Bc t _c °C			c _p liq. °K					
Cryos, A° consts, B°			c vap. °K					
t _e °C	81.70	5	c vap.	1	<u></u>	+ grams/100 gra	ms solvent	
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da			
SOURCE:		A	PI					_
PURIFICAT			PI					
LITERATUE	RE REFEREI	NCES	5:					

							No. 32	
NAME	2, 3-Dime	thyl-	1,3-butadiene			STRUCTURAL	FORMUL	A.
Ī						CH.	3	
			T			CH ₂ =C C =	CH ₂	
Mole % Pur.	Ref. Mo	lecul rmul	ar C ₆ H ₁₀	Molecular Weight 82.14	ю	с́н ₃		
		Ref.			Ref			Ref.
F. P. *C	-76,005	2	dt/dP	T		f to		
F.P. 100%			*C/mm	0.1557	ا ۔ ا	g°K		
B. P. *C 760 mm	68,78	2	25°C BP	0.1557 0. 04 064	5 4	h		ــــ
100	17.11	5	t.	0.03494	5	f' to g'K		1
30 10	-5.79 -23.23	5	30 mm	0.5745	5	h' '		
1	-52.08	5	ΔHm cal/g	-		m to		
Pressure mm 25°C	143.84	5	ΔHv cal/g 25°C	94.89	5	n •K		
t _e	920.51	5	30 mm BP	100,38 87,18	5 5	°i		<u> </u>
Density			1	86.14	5	m' to		
g/ml 20°C	0.7267 0.7222	2 2	te (d, e)	86.11	5	". ' - -		
d ₄ 30	0.7177	4	ΔHv/T _e	20.33	5	Surface tension		\vdash
a	0.7447	4	d -6 to		5	dynes/cm. 20°C	22,44	5
b Ref. Index	-0.0387	4	a' ' ' ' ' '	Ī.		30 40	21.32 20.21	5
n _D 20°C	1.4394	2			-	Parachor [P]		
25 30	1.4362 1.4330	2 4	d _c g/ml v _c ml/g t _c °C			20°C 30		
"C"	0.8019	4	1 -	Î		40		
MR (Obs.)		4	P _c mm	<u> </u>			246.2	5
MR (Calc. (nD-d/2)		5 2	PV/RT 25°C	0.9885	5	Exp. L.1.%/wt.		İ
Dielectric	1.0760 2.07	5	30 mm BP	1.0000 0.9560	5	Dispersion	225.	2
A -6 to		4	te	0.9507	5	Flash Point °C Fire Point		
B 1116°C	1220.88	4	t _c			M Spec.		╁
A* -6to	225.9	5	ΔHc kcal/m ΔHf			Ultra V.		1
B* 85°C		5	ΔFf			X-Ray Dif. Infrared		1
K — — –	-		Viscosity centistokes			Solubility in +		T
t _k to		1	7 °C	1		Acetone Carbon tet.		1
x		<u> </u>				Benzene		1
A' to						Ether n-Heptane		1
C'			B ^V to	1		Ethanol Water		}
A'* to B'* *C				-		Water in		
Acl to			(A ^V) to					
Bc tc C			c _p liq. °K	 	\vdash			
Cryos, A		\vdash	·					
consts. B°			P -					
t _e °C	74.82	5	c _v vap.					
						+ grams/100 gran	ns solven	t
	CES: 1-Dow	2-AI		Calc. from de	t, da	ta 5-Calc. by for	nula	
SOURCE:		AI	·					
PURIFICAT		AI	·					
LITERATU	RE REFERE	NCES	i:					

No. 1 Ethyne NAME STRUCTURAL FORMULA Acetylene CH=CH Molecular 26.036 Molecular C2H2 Ref. Mole % Pur Formula Ref. Ref. Ref. -80.8[₹] <u>F.P.</u> ℃ 2 dt/dP f to F.P. 100% °C/mm °K g 25°C B. P. °C h ВP 0.018 2 -84.0[#] 760 mm 2 0.02582 5 f t_e 100 -108.3 2 to g' ۰ĸ 0.3036 5 -120.1 2 30 mm 30 10 -129.5 2 h! ∆Hm cal/g 1 -153.35 0, 1355 300 to m AHv cal/g(a) 0,0012 Pressure ņ 600 25°C mm 25°C ò -0.0₆86 4 30 mm 196.37 5 494.3 5 BP 193.46 5 5 700 0. 3444 m' 4 to Density te te (d, e) 195.04 0. 0₃37 -0. 0₇98 n' 1000 4 g/ml 20°C 4 dt4 25 AHv/T 30 Surface tension d to a dynes/cm. 20°C <u>с</u> Ъ 30 ai to 40 Ref. Index °C ⁿD 20°C Parachor [P] d_c g/ml 0.231 25 20°C 4.329 vc ml/g t °C 2 30 30 t_c 2 36.3 40 "C" 46816. 2 5 P_c mm Sugd. 90.4 MR (Obs.) PV/RT Exp. L.1. %/wt. MR (Calc.) 9.434 25°C u. (nD-d/2) 1.0000 30 mm 5 Dispersion Dielectric 0.9700 5 BP Flash Point °C 0.9778 5 A -81 to 7,0949 2 Fire Point 5 0.273-60 °C 709.1 M. Spec. 253.2 C 2 AHc kcal/m 2 300,10 Ultra V. ΔHf 54.194 2 A* -81 to 1.60062 5 X-Ray Dif. ΔFf 50,000 2 -60 °C B* 676.6 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. t_x •c Benzene ۸ to Ether В' °C n-Heptane C' $\mathbf{B}^{\mathbf{v}}$ Ethanol B' I °C Water A1* to Water in B'* (B^V) °C A^S
B^S
C^S
A*S 9.1363 2 Acl (A^V)| to 1230,3 Βc °C cp liq. 300°K 280.7 10.532 2 11.973 3, 1388 5 400 2 A*s cp vap 300°K 1160.71 Cryos. Aº 0.40452 2 consts. B° 400 0.45986 2 c_w vap. te °C -89.6 Saturation press (triple pt.) #Sublimation point grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: PURIFICATION: LITERATURE REFERENCES: (a) Heat of sublimation

							No. 2	
NAME	Propyne					STRUCTURAL I	FORMULA	
	Methylacet	ylene				СН₃С≣СН		
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 40.0	062			
		Ref.			Ref.			Ref.
F.P. °C F.P. 100%	-102.7	2	dt/dP *C/mm			f to g °K		
B. P. °C 760 mm 100 30 10	-23.22 -61.1 -78.2 -90.2	2 2 2 2	25°C BP te 30 mm	0.0301 0.0334 0.413	2 5 5	h to g' °K		
Pressure mm 25°C	666.5	5	ΔHv cal/g 25°C 30 mm	152.2	5	m 300 to n 600 °K	0.0996 0.0010 -0.0 ₆ 46	4
Density g/ml 20°C dt 25 4 30			BP te te (d,e) ΔHv/Te	131. 0 131. 9 132. 15 21. 39	5 5 5	m' 700 to n' 1000 °K	0.1830 0.0 ₃ 74 -0.0 ₆ 24	4 4 4
a b Ref. Index			d -78 to e -30 °C d' to e' °C	0. 3856	5	Surface tension dynes/cm. 20°C 30 40		
n _D 20°C 25 30			d g/ml vc ml/g tc °C	0, 229 4, 367 121, 6	5 5 2	Parachor [P] 20°C 30 40		
MR (Obs.) MR (Calc. (nD-d/2)		5	P _C mm PV/RT 25°C 30 mm	1,0000	5	Sugd. Exp. L.1.%/wt. u. Dispersion	129.4	5
Dielectric A -78 to B 24 °C		2 2	BP t _e t _c	0.9700 0.9699 0.270	5 5 5	Flash Point °C Fire Point		
A* -78 to B* -16 °C K	228. 93 0. 89250 738. 2	5 5	ΔHc kcal/m ΔHf ΔFf	442.07 44.319 46.313	2 2 2	M Spec. Ultra V. X-Ray Dif. Infrared		
c to to two A' to			Viscosity centistokes 7°C			Solubility in + Acetone Carbon tet. Benzene		
B' °C C' to	2		B ^V to A ^V °C			Ether n-Heptane Ethanol Water Water in		
Ac to			(B ^V) (A ^V) c _p liq. °K					
Cryos. A° consts. B°	-26, 2	5	c _p vap.300°K 400 c _v vap.	0.36319 0.43258	2 2			
t _e °C	1-20.2			L	<u> </u>	+ ===== (100 ===		<u> </u>
REFEREN	CES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc. from det	da	grams/100 granta 5-Calc. by form		
SOURCE:		AF						
PURIFICA?	TION:	AP	I					
LITERATU	RE REFERE	NCES	:					

No. 3

									No. 3	
NAME	1 - Bu	tyne					ST	RUCTURAL	FORMUI	LA
[[Ethy	lacetyl	lene							
Mole % Pur.	Re		lecul		Molecular Weight 54.	088		C ₂ H ₅ C≡CH		
		1.0.	Ref.		l Ji.	Ref.	<u> </u>			Ref.
	1 125	720	1		 	Ker.			T	1.61.
F.P. °C F.P. 100%	-125.	120	2	dt/dP °C/mm			f	to		
	<u>'</u>		-	25°C	0. 0208	5	g	•к		1
B. P. °C 760 mm	8.	09	2	BP	0.0336	5	_h_	<u> </u>	l	
100	-34.		2	t _e	0.0337	5	f'.	to		1
30	-53.		2	30 mm	0.4724	5	g'	°K		
10 1	-67. -91.		2 5	∆Hm cal/g			h'			
Pressure	+		1	ΔHv cal/g			m	to	1	
mm 25°C				25°C	105.61	5	n o	°K		
te	758.	4	5	30 mm BP	125.13 109.82	5				
Density				t.	109.82	5	m'	to		1
g/ml 20°C	0.	65 [‡] 65 [‡]	2	te (d, e)			n' o'	°K		1
dt 25	0.	65′ 65	2	AHv/T _e	21.12	5				
			4	d -60 to	111.83	5		face tension		
a b		6511 03129	5	e 30 °C	0.2489	5	dyn	es/cm. 20°C	16.40	5
Ref. Index		03127	-	d' to			•	30 40	16.25 16.03	5
n _D 20°C				e' °C		-	Par	achor [P]	10.05	+_
_ 25				d _c g/ml	1		,	20°C		
30				vc ml/g tc °C				30		
"C"				P _c mm	l	1		40 Sugd.		
MR (Obs.)				PV/RT		\vdash	F	L.1.%/wt.	 	+-
MR (Calc. (nD-d/2)	וי			25°C			Ext	u.		
				30 mm	ŀ		Dis	persion		1
Dielectric	-		\vdash	BP		1	Fla	sh Point °C		
A -67 to B 63 °C		97497 46	2 2	t e t c			Fir	e Point		
c	232.		2	ΔHc kcal/m	589.08	2		Spec.		
A* -60 to	1	20441	5	ΔHf	39.48	2		ra V. Ray Dif.		
B* 18 °C	920.		5	ΔFf	48.30	2		ared		
к — — –				Viscosity			Sol	ubility in +		
t _k – to	-			centistokes り °C			Ac	etone		
t <u>x</u>				,				rbon tet. enzene		
A' to								her	ļ	
B'°	-1			B. to		+		Heptane	1	
C'	-			B' to A' I °C				hanol ater		
A'* to B'* °C				(BV)	•			ter in		
Acl to	+		\vdash	(A ^V)					T	\top
Bc tc °C						\vdash	1		1	
Cc C				c _p liq. °K						
Cryos. A°				c _p vap. *K		1	1			
consts. B°	ļ			() -	Į.					1
t _e °C			L	c _v vap.	1	L	<u></u>		<u> </u>	
$T_{R} = 0.75$	T _c		≠ a	t saturation pre	ssure		+ g	rams/100 gra	ms solve	nt
REFEREN	CES: 1	-Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da	ta 5	-Calc. by for	mula	
SOURCE:										
PURIFICA'	TION:									
LITERATU	RE RE	FERE	NCE	 5:						
1										
1										

····							No. 4	
NAME	2-Butyne					STRUCTURAL	FORMULA	
	Dimethyla	cetyl	ene			CH CEC CH		
Mole % Pur.		lecul rmul		Molecular Weight 54.0	88	CH ₃ C≡C CH ₃		
		Ref.			Ref			Ref
F.P. °C F.P. 100%	-32,260	2	dt/dP °C/mm			f to cC		
B. P. °C 760 mm 100	26.97 -18.7	2 2	25°C BP t	0.0379 0.0358 0.0341	5 5 5	$\frac{h}{f'} + - \frac{1}{to}$		
30 10	-38.98 -54.46	2	30 mm	0.5097	5	g' °C h'		
1	-80.0	5	ΔHv cal/g		1	m 300 to	0.1082	
Pressure mm 25°C t _e	706.6 808.6	5	25°C 30 mm BP	117.43 131.8 117.03	5 5 5	n 600 °K	0.0 ₃ 84 -0.0 ₆ 17	4
Density g/ml 20°C	0.6910 0.6856	2 2	t _e t _e (d, e)	116.5 116.65	5 5	m' 700 to n' 1000 °K	0.1020 0.0 ₃ 93 -0.0 ₆ 30	4 4 4
d ₄ 25 30 a	0.6801	4	ΔHv/T _e	20, 87 123, 07	5	Surface tension dynes/cm. 20°C		
b Ref. Index	-0.00101	4	e 35 °C d' to e' °C		5	30 40	21.19 19.80 18.42	5 5 5
n _D 20°C 25 30	1.3921 1.3893 1.3865	2 2 4	d g/ml v ml/g		_	Parachor [P] 20°C 30		
"C"	0.7574	4	t _c °C P _c mm	212. 38478.	5 5	40 Sugd.	168.4	5
MR (Obs.) MR (Calc.) (nD-d/2)	18.64 18.670 1.0466	2 5 2	PV/RT 25°C 30 mm	0.9653 1.0000	5	Exp. L.1.%/wt. u. Dispersion		
Dielectric			BP	0.9650	5	Flash Point °C		
A -32 to B -91°C C		2 2 2	t _e t _c AHc kcal/m	0.9628 584.57	5	Fire Point M Spec.		
A* -32to B*_ 39°C	1.28120	5 5	ΔHf ΔFf	34.97 44.32	2 2	Ultra V. X-Ray Dif. Infrared		
c t _k t _o °C			Viscosity centistokes 7°C			Solubility in + Acetone Carbon tet.		
A' to B' °C			B ^v to			Benzene Ether n-Heptane		
A'* to B'* °C			B' to °C			Ethanol Water Water in		
Ac 91 to Bc t _c °C	1354.8	5	(A ^V) c _p liq. °C		\sqcup			
Cryos. A° consts. B°	277.8	5	c _p vap.300°K	0.34573	2			
t _e °C	28.65	5	c _v vap.	0.41821	2			
$T_{\mathbf{R}} = 0.75$		olid				grams/100 grai	ms solveni	t
	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	calc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:		API	·					
PURIFICAT		API						
- LAKIU.	RE REFERE	NOES	;					

Mole % Pur.		l-Pen	tyne									
								\dashv	ST	RUCTURAL	FORMUL	A
		Ref.		ecula mula		Molecular Weight	8.1	14		CH≡C(CH ₂) ₂	сн ₃	
				Ref.		T	-	Ref.				Ref
F. P. ℃		05.7		2	dt/dP				f	to		
F.P. 100	%				°C/mm	0.050		_	g	°K		
B. P. °C 760 mm	İ	40.18		2	25°C BP	0.058		5	_h _			
100 mm		-7.		2	t _e	0.034		5	f'	to		
30		28. 44.		2	30 mm	0.525	1	5	g'	°K		ŀ
10 1		70.		5	∆Hm cal/g				h'			<u> </u>
Pressure					ΔHv cal/g			_]	m n	300 to 600 °K	0.0650 0.0012	
mm 25°0	; 4	31.4		4	25°C 30 mm	100.63		5	o	000 1	-0.0648	
t _e	°	51.4		5	BP	97.28		5	m'	70 0 to	0.1410	4
Density g/ml 20°	ر ا	0.69	ا ۱	2	t _e (d, e)	96. 62 96. 63	- 1	5	n'	1000 °K		
d ₄ 25	Ŭ	0.68		2	L _e (d, e) ΔHv/T _e	20.78		5	∞01		0.0 ₃ 94 -0.0 ₆ 32	4
4 30	_				d -50 to	_		5	Sur	face tension		
a b					e 50 °C		4	5		es/cm. 20°C	19.34	5
Ref. Inde	_				d to		- 1		•	30 40	18.13 16.94	5
n _D 20°		1.38		2	e' °C		\dashv		Par	achor [P]		Ť
25 30		1.38	26	2	d g/ml vc ml/g tc °C					20°C		
"C"	+				tc °C °					30 40		İ
	.				P _c mm						207.4	5
MR (Obs. MR (Calc		23. 14		2	PV/RT				Exp	. L.1.%/wt.		
(nD-d/2)	``	1.04	02	2	25°C 30 mm	0.981 1.000		5	Die	u. persion	119.	2
Dielectri	c				B P	0.969	3	5		sh Point °C	,.	<u>-</u> -
A -50 to B 70°		6, 97; 95, 42	263	2 2	t _e t _c	0.966	6	5	Fir	e Point		-
С		27.53		2	ΔHc kcal/m ΔHf	735.95 34.50		2 2		Spec. a V .		
A* -50 to B* 50° K		1.25 22.54	625	5	ΔFf	50.16		2		lay Dif. ared		
c - t					Viscosity centistokes 7°C				Αc	ibility in + etone		
t _x					,		İ		Be	rbon tet. nzene		
A' t	c									her Heptane		
C'					B _v to			l l	Et	hanol		
A1* t				1	A C	_	- 1			iter iter in		
B'* °	_			<u></u> -	(B ^V)							+
Ac t Bc t _c *	c C				(A ^V) c _p liq. °K		\dashv					
Cryos. A					c vap. 300K 400	0.371 0.454		2				
t _e °C		43.45		5	c _v vap.	0.434	U 7	۲				
									† gı	ams/100 gra	ms solven	ıt
REFEREN	ICES:	: 1-D	ow	2-A1	PI 3-Lit. 4	Calc, from	det	. da	ta 5	-Calc. by for	mula	
SOURCE:												
PURIFICA												
LITERAT	URE	KEFI	∴REÌ	NCES	:							

							No. 6	
NAME	2-1	Pentyne				STRUCTURAL	FORMULA	`
						C ₂ H ₅ C≣CC	H ₂	
Mole % Pur.	Ref.	Molecul Formul		Molecular Weight 68.	114	2 3	•	
		Ref.			Ref			Ref.
F.P. °C	-109.3	2	dt/dP	T	\vdash	f to		<u> </u>
F.P. 1009			°C/mm		1	g °K		
B. P. °C			25°C BP	0.1003 0.0392	5	ъ ;		
760 mm 100	56.07 6.20		t	0.0343	5	f' + to		
30	-16.0	2	30 mm	0.556	5	g' °K		ł
10 1	-32.8	2 5	ΔHm cal/g			h'		
Pressure	+		ΔHv cal/g			m 300 to n 600 °K	0.0691 0.0010	
mm 25°C		5	25°C 30 mm	108.5 115.58	5	n 600 °K	-0.0 ₆ 30	4
t _e	898.4	5	BP	102.6	5	m' 700 to		
Density g/ml 20°0	0.71	07 2	te (3 a)	101.66	5	m' 700 to n' 1000 °K	0.1628	4
t 25	0.70		te (d, e)	20.71	5	0' {	0.0 ₃ 81 -0.0 ₆ 22	4
	0.70		ΔHv/T _e		5	Surface tension	†	
a b	0.73		d -19 to		5	dynes/cm, 20°C	21.84	5
Ref. Index		101 4	d' - to			30 40	20.54 19.27	5 5
n _D 20°0	1.40			 	-	Parachor [P]		
25 30	1.40		d g/ml vc ml/g			20°C	ì	
"C"	0.75		tc °C	241.	5	30 40	İ	
MR (Obs.			P _c mm	28675.	5	Sugd	. 207.4	5
MR (Calc.			PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.04	85 2	25°C 30 mm	0.9901	5	u. Dispersion		
Dielectric	1.97	. 5	BP	0.9700	5	Flash Point °C	 	-
A -19t B 113°	2 1193.05	2	t _e t _c	0.9663	5	Fire Point M Spec.	-	-
A* -19to	229.96		ΔHc kcal/m ΔHf	732.25	2 2	Ultra V.	1	1
B* 71°	1113.17		ΔFf	46.41	2	X-Ray Dif. Infrared		ļ
K — — -	_		Viscosity	İ		Solubility in +		<u> </u>
the Total	5		remtistokes °C			Acetone		İ
t _x ' 'C			•			Carbon tet. Benzene		
B' _ °		i				Ether n-Heptane		į
C'			B ^V to C			Ethanol		
A'* to B'* °			(BV)	-		Water Water in	ļ	
Ac 113 to	C 1424.6	5	(A ^V)		_			
Cc	212.6	5	c _p liq. •K					
Cryos. Acconsts. B	`		c _p vap.300°K 400	0.34780 0.42869	2			
t _e °C		5	c _v vap.	1		L		L
TR = 0.75						f grams/100 gra		t
REFEREN	CES: 1-D		PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	TION	AF						
PURIFICA		AF ERENCES						

No. 7 NAME 3-Methyl-1-butyne STRUCTURAL FORMULA CH3(CH2)2C≣CH Mole Ref. Molecular **M**olecul**a** r C₅H₈ % Pur Weight 68.114 Formula Ref Ref. Ref. F. P. °C -89.7 2 dt/dP f to F. P. 100% °C/mm g °K 0.0376 25°C 5 B. P. °C 5 h ВP 0.0362 760 mm 26.35 2 t_e 0.0348 5 f 100 -19.4 4 to g' -39.5 °K 30 4 0.4974 5 30 mm 10 -55. 5 h! ∆Hm cal/g 300 0.0293 m to AHv cal/g Pressure 600 0.0013 n ۰ĸ 25°C 91.0 mm 25°C 723.4 -0.064 4 4 o 30 mm 106.8 5 799.8 te BP 90.63 5 70**0** m' 0.1145 to Density 5 90.26 t_e (d, e) 0.0010 n' 1000 4 g/ml 20°C 0.666 ٩K 2 90.29 5 ۰' -0.0637 4 0.660 $\mathbf{d_4^t}$ 25 2 AHv/T 20.43 5 30 0.654 4 Surface tension -40 to 97.10 5 a 0.691 4 16.67 dynes/cm. 20°C 28 °C 0,2455 ь -0.001 4 15.41 5 30 d٠ to 40 14.17 5 Ref. Index e¹ °C ⁿD 20°C 1.3723 2 [P] Parachor d_c g/ml 25 1.3695 2 ZO°C vc ml/g 30 1.3667 4 30 ^tc 188. "C" 0.7482 4 40 28529. 5 Sugd. 207. 4 5 P_c mm MR (Obs.) 23, 261 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 23, 288 5 25°C 0.9578 5 (nD-d/2)1.039 2 30 mm 1.0000 5 Dispersion Dielectric 1.88 5 BP 0.9566 Flash Point °C 0.9549 A -40 to te tc 6.88971 5 Fire Point B 1_ 73 °C 1017.50 5 M. Spec. C 227,46 5 AHc kcal/m 734.05 Ultra V. 32.60 49.12 ž ΔHf A≠ -40 to 1.21594 5 X-Ray Dif. ΔFf B* _ <u>38</u> °C 952.8 Infrared ĸ Viscosity Solubility in c centistokes Acetone to t_k ٠c Carbon tet. $\mathbf{t_x}$ Benzene AI to Ether В' °C n-Heptane B_v | C' to Ethanol °C Water A'* to Water in (B^V) B'* °C Acl 73 to 7.34037 5 (A^V) Bc °C 1329.4 c_p liq. ۰ĸ Cc 272.6 5 Cryos. A° cp vap.300K 0.36894 2 consts. Bo 0.45659 c, vap. te °C 27.76 5 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 8	
NAME	l-Hexyn	е				STRUCTURAL I	FORMULA	
N-1-	P. 6 No.	1		M-11		C ₄ H ₉ C≣CH		
Mole % Pur.	Ref. Mo	rmul		Molecular Weight 82.1	40			
		Ref.			Ref.			Ref.
F.P. °C	-131.9	2	dt/dP			f to		
F. P. 1007 B. P. °C	•	-	*C/mm 25*C	0.1670	5	g °K		
760 mm	71.33	2	BP	0.042 0.0360	2 5	$\left \frac{\mathbf{h}}{\mathbf{f'}} + \frac{1}{\mathbf{f'}}\right $		
100 3 0	18.2 -5.2	4 4	t _e 30 mm	0.5854	5	g' to		
10 1	-22.9 -52.2	5	ΔHm cal/g			h'		
Pressure	+ 32.2	 	ΔHv cal/g	/	_	m 300 to n 600 °K	0.0576 0.0012	4
mm 25°C		5	25°C 30 mm	93.6 98.94	5	o l	-0.0648	4
t _e Density	925.6	5	BP	85.40 84.3	5	m' 700 to	0.1451	4
g/ml 20°0		2	te te (d, e)	84. 27	5	n' 1000 °K	0.0 ₃ 96 -0.0 ₆ 32	4 4
dt 25 4 30	0.7106 0.7057	2 4	AHV/Te	19.73	5		-0.0632	4
a	0,7351	4	d -8 to	98. 0 2 0. 1769	5	Surface tension dynes/cm. 20°C	21.16	5
ь	-0.0396	4	$\frac{\mathbf{e}}{\mathbf{d}} - \frac{1}{1} - \frac{78}{10} \cdot \frac{\mathbf{c}}{\mathbf{c}}$]]	30 40	19.99	5 5
Ref. Index		2	e' i °C		<u> </u>	Parachor [P]	18.85	5
25 30	1.3960	2 4	d g/ml v ml/g			20°C 30		
"C"	0,7435	4	ic C	248.	5	40		_
MR (Obs.)		2	P _c mm	22734.	5		246.4	5
MR (Calc. (nD-d/2)) 27.906 1.0412	5 2	25°C	0.9913	5	Exp. L.1.%/wt. u.		
Dielectric		5	30 mm BP	1.0000 0.9550	5	Dispersion	115.	2
A -8 t	6.91212	5	t _e	0.9479	5	Flash Point °C Fire Point		
B 118 °C	2 1194.6 225.	5	t _c ΔHc kcal/m	882.85	2	M Spec.		
A* -8 to		5	∆Hf	29.55	2	Ultra V. X-Ray Dif.		
B* L 88 °C	1119.4	5	ΔFf Viscosity	52.17	2	Infrared		
c			centistokes			Solubility in + Acetone		
t _k to			η •c			Carbon tet.		
A' to		\vdash		1		Benzene Ether		
B' º	2		B ^V to			n-Heptane Ethanol		
A1* to	,		AV °C			Water		
B'* °(<u> </u>	(B ^V)			Water in		
Ac 118 to	7. 16827 C 1453. 8	5	(A ^V)		_			
Cc	268.5	5	c _p liq. °K					
Cryos. Acconsts. B			c _p vap.300°K 400	0.37460 0.46104				
t _e °C	77.76	5	c _v vap.	0.40104				
$T_{R} = 0.75$	T _c	<u> </u>				grams/100 gran	ns solvent	
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from det	. da			
SOURCE:		AF	PI					
PURIFICA		AF						
LATERATU	RE REFERE	NCES	5:					

No. 9 l-Heptyne STRUCTURAL FORMULA NAME C5H11C≣CH Molecular C7H12 Mole Ref. Molecular Weight 96.166 % Pur. Formula Ref. Ref. Ref. F.P. °C -80.9 2 dt/dP f to F.P. 100% °C/mm g °K 25°C 0.4081 5 B. P. °C h ВP 2 0.0450 760 mm 99.74 2 ^te 0.03836 5 f١ to 100 39.63 4 g† <u>• к</u> 30 13.56 4 0.6491 5 30 mm 10 -6.0 5 h! AHm cal/g 1 -38.0 5 300 to 0.0526 m AHv cal/g Pressure 600 °K 0.0012 4 n 25°C 85.69 5 mm 25°C 52.51 -0.0₆48 4 5 0 30 mm 87.26 5 te 1017.9 5 75.40 5 BP 700 to m' 0.1346 4 Density te (d, e) 5 73.85 0.0010 -0.0₆34 n' 1000 °K g/ml 20°C 0.7328 2 73.88 5 ٥' 4 25 d_4^t 0.7283 2 AHV/Te 18,50 5 30 0.7238 4 Surface tension 14 to 89.13 5 0.7508 4 dynes/cm. 20°C 22,34 a ١ 0.1377 5 111 °C ь 21.25 -0.0389 5 4 30 to 5 40 20.19 Ref. Index e† ٠c 1.4087 20°C [P] nD 2 Parachor d_c g/ml 25 20°C 1.4061 2 vc ml/g t °C 30 4 1.4033 30 280. 5 40 "C" 0.7427 4 P_c mm 5 Sugd. 285.4 5 19163. MR (Obs.) 32.42 2 PV/RT 5 Exp. L.l.%/wt. 32.524 MR (Calc.) 25°C 0.9996 5 (nD-d/2)1.0423 2 30 mm 1.0000 5 2 Dispersion 113. Dielectric 1.98 5 0.9622 BP Flash Point C 0.9529 5 6.68593 A 14 to 5 Fire Point В 142 °C 1216.6 5 M. Spec. C 1029.77 220. 5 AHc kcal/m Ultra V. 24.62 2 ΔHf A*| 14 to 1.05646 5 X-Ray Dif. 54.18 2 ΔFf B* 121 °C 1132.5 Infrared ĸ Viscosity Solubility in c centistokes Acetone t_k to Carbon tet. t_x °C Benzene A' I to Ether В' ۰c n-Heptane B_v C to Ethanol °C Water Water in (B^V)| B'* °C to Acl 142 to 7.10459 (A^V)| °C 1538.4 °C Bc tc 5 c_p liq. ۰ĸ Cc 265.0 5 Cryos. A° consts. B° c_p vap.300K 0.37716 2 0.46597 2 c_v vap. te ℃ 110.76 $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

_								No. 10	
NAME	1-Octyne					STRUCT	URAL :	FORMULA	1
						C,H.	с≡сн		
Mole % Pur.	Ref. Mo	lecul rmul	ar C ₈ H ₁₄	Molecular Weight 110.1	92	-613	,		
		Ref.			Ref	T			Ref.
F.P. °C	-79.3	2	dt/dP			f	to		
F.P. 1009	•		°C/mm 25°C	1.3009	5	g	°K		
B. P. °C 760 mm	126. 2	2	BP	0.0471	2	h + -			
100	66.39	5	t _e	0.0358	5	f' .	to °K		l
30 10	39.86 19.68	5	30 mm	0.6651	5	g' h'	·K	l	
1	-13.73	5	AHm cal/g		1	<u>-</u>	300 to	0.0487	4
Pressure			ΔHv cal/g 25°C	90.74	5		600 °K	0.0012	4
mm 25°C t _e	13.60	5	30 mm	88.54	5	۰ ;		-0.0 ₆ 48	4
Density	+	╁	BP te (d.e)	75.99 74.18	5		700 to	0.1326	4
g/ml 20°0		2	te (d, e)	74.11	5	n' 1	000 °K	0.0010	
dt 25 4 30	0.7419 0.7377	2 4	ΔHv/T _e	19.82	5			-0.0634	-
a	0.7629	4	d 25 to	94.33	5	Surface t dynes/cr		23.27	5
ь	0.000836		_a,_ 150_ °C		5	*	30	22,23	5
Ref. Index	1.4159	2	e' °C				40	21.23	5
ⁿ D 20°C	1.4134	2	d _c g/ml			Paracho	r [P] 20°C	1	
30	1.4109	4	t _c °C				30		
"C"	0.7146	5	P _c mm				40 Sugd.	324.4	5
MR (Obs.) MR (Calc.		2	PV/RT		\vdash	Exp. L.			_
(nD-d/2)	37.142 1.0428	5 2	25°C 30 mm	1.0016	5 5	u	١.	111.	2
Dielectric			BP	0.9500	5	Dispersi Flash Po		111.	-
A 25 to		4	t _e	0.9394	5	Fire Poi			
B 1 170 °C	215.0	5	t _c ΔHc kcal/m	1176, 70	2	M Spec.			
A* 33 to		5	∆Hf	19.70	2	Ultra V. X-Ray D	if		
B* 160 °C		5	ΔFf	56.19	2	Infrared			
K — — -			Viscosity centistokes		1	Solubility			
ել tc			η °c			Acetone Carbon			
t _x °C		-				Benzene			
B' c					_	Ether n-Hepta	ne		
C'	1		B ^V to C			Ethanol Water			
A'* to B'* **			(BV)			Water i	n		ļ
Ac to			(A ^V)						
Bc t °C			c _p liq. •K		-				1
Cc — -		 	1						
Cryos. A° consts. B°			c _p vap.300°K 400	0.3790 0.4696	2				
t _e °C	139.16	5	c _w vap.			L		<u> </u>	
DEFER	CEC. I D.							ms solven	t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	aic, from de	t. da	ta 5-Calc	. by for	mula	
SOURCE:	TION								
	RE REFERE	NCEC							
	NEFERE.	.10E3	••						

No.11 1-Nonyne NAME STRUCTURAL FORMULA C7H15C≣CH Mole Ref. Molecular Molecular Weight 124.218 C9H16 % Pur. Formula Ref. Ref. Ref. dt/dP f to F.P. 100% -50. 2 °C/mm g ۰ĸ 25°C 2.727 5 B. P. °C h ВP 0.0530 5 760 mm 150.8 2 t_e 0.0385 5 f١ to 100 84.23 5 g' ۰ĸ 5 30 55.19 5 0.7247 30 mm 10 33.28 h' ∆Hm cal/g 5 1 2.64 300 0.0459 m to ∆Hv cal/g 600 0.0013 Pressure n °K 25°C 83.15 mm 25°C -0.0₆49 6.26 5 4 o 5 30 mm 79.30 1148.0 5 ^te 5 ΒP 67.35 m' 700 0.1312 Density te te (d, e) 65,28 5 n' 1000 °K 0.0010 g/ml 20°C 0.7568 2 64.52 5 ď -0.0635 4 ď4 25 0.7527 2 5 ΔHv/Te 18.40 30 0.7486 4 Surface tension 50 to 86.20 5 d 0.7732 -0.0₃82 5 5 dynes/cm. 20°C 24.02 1<u>60 °C</u> 0.1250 ь 30 23.00 5 d٦ to 40 22,00 5 e¹ Ref. Index 1.4217 2 20°C [P] nD Parachor d_c g/ml 1.4193 2 25 20°C **v**c ... ml/g 4 30 1.4169 30 tc 334. 5 40 "C" P_c mm 5 14300. 5 Sugd. 363.4 MR (Obs.) 41.64 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 41.760 5 25°C 1.0000 5 (nD-d/2)1.0430 2 30 mm 1.0000 5 Dispersion 110. 2 Dielectric BP 0.9482 5 Flash Point °C 0.9346 A 50 to te t 6.77410 5 Fire Point B (223 ℃ C 1404.7 5 M. Spec. 210. 5 ∆Hc kcal/m Ultra V ΔHf A* 51 to 1.22540 5 X-Ray Dif. ΔFf B* 180 °C 1316.75 5 Infrared ĸ Viscosity Solubility in centistokes Acetone to °C Carbon tet. $\mathbf{t_{x}}$ °C Benzene A١ to Ether В' °C n-Heptane B_v | C' to Ethanol °C Water A'* to Water in (B^V) B'* °C Acl (AV) to Bc °C cp liq. °K Cc cp vap.300°K Cryos. A° 0.38046 2 consts. B° 0.47248 2 c_w vap. te °C 168.2 5 $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 12	
NAME	1-Decyne					STRUCTURAL	FORMULA	1
Mole	Ref. Mo	lecul	27	Molecular		C ₈ H ₁₇ C≡CH		
% Pur.	Kei. Mo	rmul	ar C ₁₀ H ₁₈	Weight 138.	244			
		Ref.			Ref.			Ref.
F.P. °C	-44.	2	dt/dP			f to		
F.P. 100%			°C/mm	10.1246	5	g °K		
B. P. °C 760 mm	174.	2	25°C BP	0.0514	5	<u> </u>		
100	108.48	5	t _e	0.0358	5	f' to		ŀ
30	79.28 57.00	5	30 mm	0.7333	5	g' K		
10 1	20.00	5	∆Hm cal/g			h'		
Pressure	 	1-	ΔHv cal/g			m 300 to n 600 °K	0.0424	
mm 25°C	1.42	5	25°C 30 mm	88.69 81.14	5	0 1 000 K	0.0013 -0.0 ₆ 49	4
t _e	1198.8	5	BP	68.78	5	}		
Density g/ml 20°C	0 7/55	,	t _e ,	66.49	5	m' 700 to n' 1000 °K	0.1345 0.0010	
	0.7655	2 2	t _e (d, e)	66.38	5	0'	-0.0635	
d ₄ 25 30	0.7577	4	ΔHv/T _e	19.63	5	Surface tension		-
a	0.7811	4	d 78 to		5	dynes/cm. 20°C	24.65	5
ь	-0.0378	4				30 40	23.66	5
Ref. Index		,	e'				22.70	-
ⁿ D 20°C	1.4242	2 2	d g/ml vc ml/g		1	Parachor [P] 20°C		
30	1.4219	4	t _c *C	354.	5	30		1
"C"	0.7402	4	P _c mm	17454.	5	40 Sugd	402.4	5
MR (Obs.)		2	PV/RT	1	-	Exp. L. 1, %/wt.	102.1	-
MR (Calc. (nD-d/2)	1.0437	5 2	25°C	1.0000	5	u.		ŀ
Dielectric	1	+-	30 mm BP	1.0000 0.9400	5	Dispersion	108.	2
A 78 to	7,10870	5	t _e	0.9250	5	Flash Point °C Fire Point		
B 246 °C	160 6	5	t _c	L				
С	206.0	5	ΔHc kcal/m ΔHf	1470.54	2	M Spec. Ultra V.		İ
A* 78 to B* 204 °C	1.60046	5	ΔFf	ļ		X-Ray Dif.		
K L	2 131 7. 03		Viscosity	<u> </u>		Infrared Solubility in +		-
·	_		centistokes	1		Solubility in + Acetone		1
t _k to t _x °C			η •c			Carbon tet.		
A' to	 	 				Benzene Ether		1
B' ∟ _ °	<u>:</u>	1 1	B ^v to	 	-	n-Heptane		1
C'		-	B' to A' ℃	}		Ethanol Water		
A'* to B'* °C			(BV)	-		Water in		l
Acl to		\vdash	(A ^V)	Į				
Bcit c°C								l
	-	\vdash	-			1		
Cryos. A° consts. B°			c _p vap.300°K 400	0.38157 0.47481	2			
t _e °C	192.7	5	c _v vap.					L
$T_{\mathbf{R}} = 0.83$						fgrams/100 gra	ms solven	t
REFERENC	CES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc. from det	t. da	ta 5-Calc. by for	mula	
SOURCE:		AP	PI .					
PURIFICAT	rion:	AP	1					
LITERATU	RE REFERE	NCES):					

No. 13 1-Undecyne NAME STRUCTURAL FORMULA C9H19C≡CH Molecular C11H20 Mole Ref. Molecular % Pur. Weight 157, 270 Ref. Ref. Ref. F. P. ℃ -25 2 dt/dP f to F.P. 100% °C/mm ۰ĸ g 110.84 0.1 mm 5 B.P. °C h BP 0.0534 5 760 mm 195. 2 t_e 0.0361 5 ſ١ to 100 126.84 4 4 ۰ĸ g' 0.7642 5 30 mm 30 96.43 h' 73.20 10 5 ∆Hm cal/g 34.61 5 m 300 to 0.0405 0.1 5.51 5 AHv cal/g 0,0013 ۰ĸ n 600 88.55 0.1 mm Press. mm -0.0₆49 o 4 30 mm 75.22 5 1248. t_e 5 BP 63, 31 5 0.1329 0.0010 m' 700 to Density te (d, e) 60.92 5 1000 n' 4 g/ml 20°C 0.7728 2 60.74 5 ٥' -0.0₆35 4 0.769**0** $\mathbf{d_{4}^{t}}$ 25 2 ΔHv/T_e 5 19.57 30 0.7652 4 Surface tension 7 97 86.88 5 a 0.7880 4 dynes/cm. 20°C 22.13 5 217 °C 0.1209 ь -0.0376 4 21.27 20.44 30 to 5 40 Ref. Index e¹ 1.4306 2 ⁿD 20°C [P] Parachor dc g/ml 25 1,4284 2 Z0°C vc ml/g t °C 1.4262 4 30 30 t_c 373 5 40 0.7398 4 "C" P_c mm 16891. 5 5 Sugd. 441.4 MR (Obs.) 50.97 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 50.996 5 0.1 mm 1,0000 5 1.0442 2 11. (nD-d/2)1.0000 30 mm Dispersion 2 107. 0.9339 Dielectric 2.05 5 BP Flash Point °C 0.9169 A 80 to 7.13064 4 Fire Point B 1_267 °C 1687.2 4 M. Spec. C 202. 5 AHc kcal/m 1617.48 Ultra V. ΔHf 4.92 ž 1.67084 A* 80 to 5 X-Ray Dif. ΔFf 62.21 2 B* 2 67°C 1599.5 5 Infrared K Viscosity Solubility in centistokes Acetone to Carbon tet. t_x °C Benzene A'ı to Ether В' ۰c n-Heptane B^v C' to Ethanol B' ' °C Water A1* to B'* (B^V) Water in °C Acl 267 to 8.22204 (A^V) Bc _t_c °C 2858.81 5 c_p liq. 342.89 5 c_p vap.300°K Cryos. Aº 0.38248 2 consts. B° 400 0.47665 c, vap. te C 216.2 5 $T_R = 0.84 T_C$ grams/100 grams solvent 5-Calc. by formula REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No.14	
NAME	l-Dodec	yne				STRUCTURAL	FORMULA	1
						C ₁₀ H ₂₁ C≣C	Н	
Mole % Pur.	Ref. Mo	lecul	ar C ₁₂ H ₂₂	Molecular Weight 166.	296	10 21		
		Ref.		weight to	Ref			Re
F. P. *C	-19.	2		γ	1			-
F.P. 100%		+-	dt/dP *C/mm	1		f to]
B. P. 'C	 	╁┈┤	0.1 mm	115.26	5	g l °K	ļ	ļ
760 mm	215.	2	BP	0.0550	5			l
100	144.63 113.15	4	t _e	0. 035 7 0. 79 13	5	g' to		ĺ
30 10	89.09	5	30 mm	0.1713	1-1	h'	!	1
1	49.0 5	5	ΔHm cal/g		\perp		0.0391	4
0.1	18.8	5	ΔHv cal/g	0. 75	ا ۔ ا	m 300 to	0.0013	4
Press. mm	1313.0	5	0.1 mm 30 mm	81.75 75.06	5 5	0	-0.0 ₆ 50	4
te	1313.0	13	BP	63.49	5	m' 700to	0, 1278	4
Density g/ml 20°C	0.7788	2	t _e (d, e) AHv/T	60.89	5	n' 1000•K	0.0011	
	0.7751	2	t _e (a, e)	60.74	5	0'	-0. 0 ₆ 36	4
d 25 4 30	0.7714	4	е	19.76	5	Surface tension	<u> </u>	┢
	0.7936	4	d 100 to		5	dynes/cm. 20°C	25, 62	5
<u>b</u>	-0.0374	4	-e,- 250_ : €	0.1136	5	30	24.66	5
Ref. Index			e' C	l		40	23.73	5
ⁿ D 20°C	1.4340 1.4318	2 2	d g/ml v ml/g			Parachor [P]		1
30	1.4296	4				20°C 30		l
"C"	0,7396	4	16 TO	3 91.	5	40		1
MR (Obs.)		2	P _c mm	15719.	5	Sugd.	480.4	5
MR (Calc.		5	PV/RT	0.0351		Exp. L.1.%/wt.		
(nD-d/2)	1.0446	2	0.1 mm 30 mm	0.9251 1.0000	5 5	u. Dispersion	107.	2
Dielectric	2.06	5	BP	0.9391	5	Flash Point °C	101.	۰
A 1113 to		4	t _e	0.9211	5	Fire Point	 	ł
B [288_•C		4	tc	17/1 10	+-	M Spec.		
A* 100 to	198.	5	ΔHc kcal/m ΔHf	1764.40	2 2	Ultra V.	1	
B* 288 °C	1.70703	5	ΔFf	64. 22	2	X-Ray Dif. Infrared	1	
K	1.0		Viscosity					┢
t to	-		centistokes	İ		Solubility in + Acetone		1
t _x t ₀			η •c	}		Carbon tet.		
A' to	 				1 1	Benzene Ether		
B' *C			L		\sqcup	n-Heptane	l	1
C'			B ^V to A ^V °C	İ	1 1	Ethanol		1
A'* to			<u> </u>	l		Water Water in	i	
B'* °C		+	(B ^V)	1			 	+
Ac 288 to		5	(A ^V)				1	
Collection ?	362.54	5	c _p liq. °K		1 1			}
Cryos, A*	†		c _p vap.300°K	0.38329	2			1
consts. B.	Í		P 400	0.47818	2		ļ	l
t _e °C	239.2	5	c _v vap.]
$T_{R} = 0.84$	Tc	لــــا	L	L	——	grams/100 gran	ma aalman	_
	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	alc. from de	+ 4=1			<u> </u>
SOURCE:		API		110111 de	-, ual	- J-Care, by for		
PURIFICA?	ION:	API						
I EKA I'U	RE REFERE	NCES):					

								No . 15	
NAME	l-Tridecy	ne				ST	RUCTURAL	FORMUL	A.
							C ₁₁ H ₂₃ C≣CI	н	
Mole % Pur.	Ref. Mo	ecul mul		Molecular Weight 180.3	22		11230-0.	•	
		Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-5.	2	dt/dP °C/mm 0.1 mm	119.74	5	f g	to °K		
B. P. °C 760 mm 100 30 10	234. 160.72 127.97 102.95	2 4 4 5	BP t _e 30 mm ΔHm cal/g	0.0573 0.0364 0.8232	5 5	h f' g' h'	to °K		
0.1	61.31 29.89	5 5	ΔHv cal/g 0,1 mm	84, 53	5	m n	900 °K	0.0376 0.0013	4
Press. mm	1350.7	5	30 mm BP	71.78 60.04	5 5	o m'	700 to	-0.0 ₆ 50	4
Density g/ml 20°C dt 25 d4 30	0.7842 0.7806 0.7770	2 2 4	t _e (d, e) ΔHv/T _e	57.33 57.10 21.5	5 5 5	n' o'	1000 •K	0.0011 -0.0 ₆ 36	4
a b Ref. Index	0.7986 -0.0 ₃ 72	4 4	d 120 to e 260 °C d' to e' °C	85.87 0.1104	5 5		face tension es/cm. 20°C 30 40	26.03 25.09 24.17	5 5 5
n _D 20°C 25 30	1.4371 1.4349 1.4327 0.7394	2 2 4	d _c g/ml v _c ml/g t _c °C	408. 14663.	5	Par	20°C 30 40	510.4	
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	60.25 60.232 1.0449 2.06	2 5 2	P _c mm PV/RT 0.1 mm 30 mm BP	1.0000 1.0000 0.9299	5 5 5	_	L.1.%/wt. u.	519. 4 106.	2
A 120 to B 308 °C	7.15674 1834.4	4	te t _C	0.9095	5	Fir	sh Point °C e Point Spec.		
A* 120 to B* 270 °C K	195. 1.73019 1742.5	5 5 5	ΔHc kcal/m ΔHf ΔFf Viscosity	1911.32 -4.93 66.23	2 2	Ult:	ra V. Ray Dif.		
c t _k			centistokes 7°C			Ac Ca Be Et	ubility in † etone rbon tet. nzene her		
B' _ °C C'			B ^v to A ^v °C (B ^v)	-		Et Wa	Heptane hanol ater ater in		
Acl 308 to Bc t _c °C Cc	8.66116 3575.90 387.79	5 5 5	(A ^V) c _p liq. °K						
Cryos. A° consts. B°			c vap 300°K	0.38392 0.47948	2				
t _e °C T _R = 0.85	260.6	5	c _v vap.	<u> </u>					
REFERENC		2. 4	DI 3-1:4 4	Cala from 3-	- نہ ہ		Cala by for		t
SOURCE:	1-DOM	2-A AP		Calc. from de	ι, α a	. La 5	-Calc. by for	muia	
PURIFICAT	ION:	AP							
LITERATUE	RE REFERE				-		4, 10		

							No. 16	
NAME	l-Tetra	dec yn				STRUCTURAL	FORMULA	`
Mole	Ref. M	lolecul	ar C ₁₄ H ₂₆	Molecular	.,,	C ₁₂ H ₂₅ C≡C	Н	
% Pur.	_ _	Ref.		Weight 194.3	Ref			Ref.
F, P. *C		2			Kei,	· · · · · · · · · · · · · · · · · · ·	T	Ker.
F.P. 100%	0.		dt/dP *C/mm			f to		l
B. P. 'C	†	+-	0.1 mm	123.53	5	h .	1	
$760~\mathrm{mm}$	252.	2	BP t _e	0.0582 0.0358	5	$\frac{1}{f'} + \frac{1}{to}$]
100 30	177.42 143.95	4	30 mm	0.8425	5	g' °K		1
10	118.31	5	ΔHm cal/g	<u> </u>		h'	1	1
1 0, 1	75.54 43.16	5	ΔHv cal/g	 	-	m 300 to		
Press, mm		+-	0,1 mm	82.84	5	n 600 °K		
t _e	1401.3	5	30 mm BP	70.33 58.89	5	<u> </u>	-0.0 ₆ 50	1
Density			l -	56. 05	5	m' 700 to		
g/ml 20°C	0.7888	2 2	te (d, e)	55, 83	5	n' 1000 °K	0.0011 -0.0 ₆ 36	
d ^t 25 4 30	0.7816		ΔHv/T _e	19.66	5	<u></u>	11.6	<u> </u>
a .	0, 8032	4	d 145 to	85.52	5 5	Surface tension dynes/cm. 20°C	26.38	5
ъ	-0.0372	4	-å,-¦ ²⁸¹ - to	0.1059	"	30	25.43	5
Ref. Index	1.4396	2	e'			40	24.51	5
n _D 20°C	1.4375	2	d g/ml v ml/g			Parachor [P]	.	İ
30	1.4354	4	t _c °C	422.	5	30		1
"C"	0.7391	4	P _c mm	13515.	5	40 Suga	. 558. 4	5
MR (Obs.)	64.89	2	PV/RT		<u> </u>	Exp. L. 1. %/wt.		 _
MR (Calc.) (nD-d/2)	64.850 1.0452	5 2	0.1 mm	1.0000	5	u.		į .
Dielectric	2,07	5	30 mm BP	1.00 0 0 0 .9 30 6	5	Dispersion	105.	2
A 1145 to	+		te	0.9092	5	Flash Point °C Fire Point		
B (326 °C		4	t _c		<u> </u>	M Spec.	+	┼─
C	191.	5	ΔHc kcal/m ΔHf	2058.24 -9.86	2 2	Ultra V.		·
A* 145 to B* 291 °C	1.8236 1833.2	7 5	ΔFf	68.24	2	X-Ray Dif.		1
к — — —	1		Viscosity		1	Infrared Solubility in +	 	├ ──
t _k	-{		centistokes 7°C		ł	Solubility in +		l
tx C	1		7 °⊂			Carbon tet.		}
A' to	†	+				Benzene Ether		1
B', ∟ °⊆	.[B ^V to	 	├	n-Heptane		
A'* to	 	+	B' to		i	Ethanol Water		ł
B'* *C	ł		(B ^V)	1	ŀ	Water in		<u> </u>
Ac 326 to		7 5	(A ^V)					
Bc tc_C		5	c _p liq. °K	<u> </u>	-	1	Ì	
Cryos, A*	413.26	 } -	li .					1
consts. B°			c _p vap.300°K	0.38452 0.48058	2			
t _e °C	280.8	5	c _w vap.	1	<u> </u>	L		<u> </u>
$T_R = 0.86$						grams/100 gra	ıms solven	t
REFERENC	ES: 1-Dow		PI 3-Lit. 4-(calc, from de	t. da	ta 5-Calc. by fo	rmula	
SOURCE:		API						
PURIFICAT		API						
LITERATU	RE REFER	ENCES	3:					

								No. 17	
NAME	1-Pentade	cyne				ST	RUCTURAL	FORMUL	A
					_		C ₁₃ H ₂₇ C≡CH	-I	
Mole % Pur.		lecul:		Molecular Weight 208.3	74		13 27		
		Ref.			Ref.				Ref
F.P. °C F.P. 100%	10.	2	dt/dP °C/mm			f g	to °K		
B.P. °C 760 mm 100	268. 190.84	2	0.1 mm BP t _e	127.20 0.06 03 0.0366	5 5	h f' g'	to *K		
30 10	156.26 129.79	4 5	30 mm	0.8700	5	h'	K		ĺ
1 0, 1	85.69 52.33	5	ΔHm cal/g ΔHv cal/g			m	300 to 600 °K	0.0357	4 4
Press. mm	1427.3	5	0,1 mm 30 mm BP	79.33 67.33	5 5	٥		0.0013 -0.0 ₆ 50	
g/ml 20°C	0.7928 0.7893	2 2	te te (d, e) AHv/Te	55.69 52.77 52.48	5 5 5	m' n' o'	700 to 1000 °K	0.1294 0.0011 -0.0 ₆ 36	4 4 4
a b	0.7858 0.8068 -0.0370	4 4 4	d 150 to e 305 °C	19. 22 83. 61 0. 1042	5 5		face tension es/cm. 20°C 30	26.69 25.76	5
Ref. Index n _D 20°C 25 30	1.4419 1.4398 1.4377	2 2 4	e' °C d _c g/ml v _c ml/g			Par	40 achor [P] 20°C	24.85	5
"C"	0.7390	4	v _c ml/g t _c °C P _c mm				30 40 Sugd	597.4	5
MR (Obs.) MR (Çalc.) (nD-d/2)	69.53 69.468 1.0455	2 5 2	PV/RT 0.1 mm 30 mm	1.0000	5	_	L.1.%/wt. u. persion	105.	2
Dielectric A 140 to	2.08 7.20553	5 4	BP t_	0.9208 0.8969	5	Fla	sh Point °C e Point	103.	-
B 349°C C A* 150 to	1972.07 188.	4 5	tc AHc kcal/m AHf	2205.17 -14.78	2 2 2	M. Ult:	Spec. ra V. Ray Dif.		
B* 310°C K c t _k c t _x °C A' to B' °C	1881.34	5	ΔFf Viscosity centistokes γ °C	70.25	2	Solu Ac Ca Be Et	ared ubility in + etone rbon tet. nzene her		
A ¹ * to B ¹ * °C			BV to AV C (BV)			Et Wa	Heptane hanol ater ater in		
Acl to Bc t _c °C			(A ^V) c _p liq. °K						
Cryos. A° consts. B°			c vap.300°K 400	0.38498 0.48154					
t _e °C	298.8	5	c _v vap.			L.	ame/100	me solus-	<u></u>
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da		ams/100 gra -Calc. by for		•
SOURCE:		AF							
PURIFICAT	ION:	AF							
LITERATUI	RE REFERE	NCES	5:						

							No. 18	
NAME	1-Hexad	ecyn	е			STRUCTURAL	FORMU LA	L
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 222.4	400	C ₁₄ H ₂₉ C≡CH		
		Ref.			Ref	ľ		Re
F. P. *C	15.	2	dt/dP					
F.P. 100%	1	† <u>-</u>	*C/mm	1		f to g °K	1	
B.P. °C			0.1 mm	130.28	5	h	Ì	
760 mm	284.	2	BP	0.0583 0.0344	5	$\frac{1}{f'}$ + $\frac{1}{to}$		l
100 30	208.55 174.24	4	t _e 30 mm	0.8670	5	g' K		
10	147.8	5	ΔHm cal/g	0.0010	Ť	h'	ł	l
1 0. 1	103.3	5	ΔHv cal/g		-	m 300 to	0.0349	4
Press. mm	1 07.2	 	0, 1 mm			n 600 °K	0.0013	
t _e	1465.5	5	30 mm BP	68.73	5	<u> </u>	-0.0650	4
Density			t_	57.04 53.99	5	m' 700 to	0.1290	
g/ml 20°C ,t 25	0.7965	2 2	te (d, e)	53.7	5	n' 1000 °K	0.0011 -0.0 ₆ 36	
d ^t 25 4 30	0.7930 0.7895	4	ΔHv/T _e	21.7	5		-0.0650	Ŀ
a	0.8105	4	d 172 to	87.30	5	Surface tension dynes/cm, 20°C	26.99	5
<u> </u>	-0.0370	4	_e, - 317 _ €	0.1066	5	y 30	26.05	5
Ref. Index	1.4440	2	e' j •C			40	25.14	5
ⁿ D 20°C	1.4419	2	d g/ml	ĺ		Parachor [P] 20°C	l	1
30	1.4398	4	t _c °C			30		
"C"	0.7388	4	Pcmm	1		40 Sugd.	636.4	5
MR (Obs.) MR (Calc.)	74.17	2 5	PV/RT		1	Exp. L.1.%/wt.		H
(nD-d/2)	74.086 1.0457	2	0.1 mm	1.0000	5	u.		ĺ.
Dielectric	2.08	5	30 mm BP	1.0000 0.9197	5	Dispersion	104.	2
A 172 to	7.47511	4	t _e	0.8955	5	Flash Point °C Fire Point		
B 1_367°C C	2154.7 185.	4 5	t _c		1	M Spec.	<u> </u>	\vdash
A* 172 to	2,12483	5	ΔHc kcal/m ΔHf	2352.09 -19.71	2	Ultra V.	1	l
B* _ 327 °C		5	ΔFf	72.26	2	X-Ray Dif. Infrared		1
K			Viscosity centistokes	Í		Solubility in +	<u> </u>	
te Tto	1		7 °C	1		Acetone		
t _{x l} *C		<u> </u>	•			Carbon tet. Benzene		
A' to B' C	ł					Ether		
č, 🗀 💴 🛎		i '	B ^v l to			n-Heptane Ethanol		
A¹* to			AV C			Water		
B'* °C	ļ	<u> </u>	(B ^V)			Water in	 	
Ac to			(A ^V)			1		
Cc		<u> </u>	c _p liq. °K	1				
Cryos. A° consts. B°			c _p vap.300°K 400	0.38539 0.48237	2 2			
t _e °C	316.2	5	l'I					
8						grams/100 gra	ms solven	<u>t_</u>
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		Al						
PURIFICAT			PI					
LITERATUR	RE REFERE	NCES	::					

Mole % Pur. F.P. °C F.P. 100% B.P. °C 760 mm 100 30 10 1 0,1	22.		C ₁₇ H ₃₂	Molecular Weight 236,42		ST	RUCTURAL C ₁₅ H ₃₁ C≡		A
% Pur. F.P. °C F.P. 100% B.P. °C 760 mm 100 30 10	22.	Ref.					C, EH, , C≡	СН	
% Pur. F.P. °C F.P. 100% B.P. °C 760 mm 100 30 10	22.	Ref.							
B. P. °C 760 mm 100 30 10	22.	Ref.			6		15 31		
B. P. °C 760 mm 100 30 10	299.	2			Ref.				Ref.
B. P. °C 760 mm 100 30 10	299.		dt/dP			f	to		
760 mm 100 30 10		-	°C/mm 0.1 mm	133,58	5	g	°K		
100 30 10 1		2	BP	0.0597	5	_h_			
10 1	221.80 186.68	4	t _e 30 mm	0.0345 0.8879	5	f' g'	to °K		
	159.6	5	ΔHm cal/g	0.007	-	h'		·	
	114.0 79.1	5	ΔHv cal/g	 	\vdash	m	300 to	0.0342	
Press. mm		+-	0.1 mm			n o	600 ∘K	0.0013 -0.0 ₆ 50	
t _e	1502.1	5	30 mm BP	66.70 55.18	5 5		700 .		<u> </u>
Density g/ml 20°C	0.7996	, # 2	te te (d, e)	52.06	5	m' n'	700 to	0.1278 0.0011	
dt 25	0.7961	2	ΔHv/T _e	51.5 21.5	5	٥'		-0.0 ₆ 37	4
	0.7926		d 184 to	85.85	5		face tension		
a b	0.8136		_e <u> 334 °C</u>	0.1026	5	dyn	es/cm. 20°C 30	27.22 26.28	5
Ref. Index	1		d' to				40	25.37	5
ⁿ D 20°C	1.4457 1.4437	* 2 2	d _c g/ml	1		Par	achor [P] 20°C		
30	1.4417		V mi/g				30		
"C"	0.7386		tc°C P _c mm				40 Sugd	675 .4	5
MR (Obs.) MR (Calc.		, 5	PV/RT		\vdash	Ext	L.1.%/wt.		-
(nD-d/2)	1.0459	≠ 3 2	0.1 mm 30 mm	1.0000	5	-	u.	104	
Dielectric	2.09	5	RP	1.0000 0.9176	5 5		persion sh Point °C	104.	2
A 184 to B 384 °C	7,4882		te t _c	0.8921	5		e Point		
c 1204 C	182.	5	ΔHc kcal/m	2499.01	2		Spec.		
A* 184 to			ΔHf ΔFf	-24.64 74.27	2 2		ra V. Ray Dif.		
B* 344 °C	- 2125.3	5	Viscosity	14.21	-		ared		ــــ
°	-		centistokes				ubility in Tetone		
t _k to			ŋ °C			Ca	rbon tet.		
A' to							nzene her	į	
B'°	-	1 1	B _v to		1-1		Heptane hanol		
A'* to	+	_	AV I °C			Wa	ater		
B'* °C	+		(B ^V)			W ₄	ter in		₩
Ac to			(A ^V)		1				
Cc Cc			c _p liq. °K						
Cryos. A° consts. B°			c _p vap300°K	0.38579 0.48311	2 2				
t _e °C	333.7	5	c _v vap.						
	rcooled liqu	id				+ g1	ams/100 gra	ms solven	ıt
REFEREN	CES: 1-Dov	w 2-A	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:		AP	I						
PURIFICA		A P							
LITERATU	RE REFER	(ENCES	5 :						

							No. 20	
NAME	1-Octade	c yne			_	STRUCTURAL 1	FORMULA	
					_	C ₁₆ H ₃₃ C≡	СН	
Mole		olecul		Molecular		16 33		
% Pur.	F	ormul		Weight 250.45	_			Ref
E D 4C	27.	Ref.			Ref		1	Kei
F.P. C F.P. 100%	 	+-	dt/dP *C/mm		1 1	f to		1
B.P. °C	†	+	0.1 mm	136.86	5	h .		
760 mm	313.	2	BP t _e	0.0609 0.0345	5 5	$\frac{1}{f'} + \frac{1}{to}$		
100 30	234.16 198.25	4	30 mm	0.9080	5	g' °K		
10	170.5	5	ΔHm cal/g			h'		
1 0. 1	123.8 88.1	5	ΔHv cal/g	 		m 300 to	0.0334	
Press. mm		+-	0.1 mm		ا ا	n 600 °K	0.0013 -0.0 ₆ 50	
t _e	1540.6	5	30 mm BP	64.71 53.55	5	ļ.—————	-0.0630	Ľ
Density	4		t	50.39	5	m' 700 to n' 1000 °K	0.1301	
g/ml 20°C	0.8025 0.7990	2 2	te (d, e)	50.2	5	n' 1000 °K	0.0011 -0.0 ₆ 36	
d ₄ 25	0.7955 [#]	4	ΔHv/T _e	21.4	5	<u> </u>		H
a .	0.8165	4	d 196 to		5 5	Surface tension dynes/cm, 20°C	27.46	5
<u>b</u>	-0.0370	4	<u>-a,</u> - 350 €			30	26.51	5
Ref. Index	1 4474#		e' •C			40	25.59	5
n _D 20°C	1.4474 1.4453	2 2	d g/ml vc ml/g			Parachor [P] 20°C	}	
30	1.4432 [#]	4	v _c ml/g t _c °C	1		30	ł	
"C"	0.7386	4	1 -		1	40 Sund	714 4	5
MR (Obs.)	83.451 [#]	2	P _c mm	ļ		<u> </u>	714.4	-
MR (Calc.) (nD-d/2)	83.322 1.0461	5 2	0.1 mm	1.0000	5	Exp. L.1.%/wt.	,	
Dielectric	2.09	5	30 mm	1.0000	5	Dispersion	10 4 . [≠]	2
A 196 to		+	BP t _e	0.9178 0.8914	5	Flash Point °C		
B 1400 °C	7.50787 2281.1	4	tc	0.0711		Fire Point	ļ	-
c	180.	5	∆Hc kcal/m	2645.94	2	M Spec. Ultra V.		
A* 196 to	2.19183		ΔHf ΔFf	-29.56 76.28	2 2	X-Ray Dif.		
B* ∟360 °C	2187.9	5	Viscosity	10.20	Ī	Infrared		<u> </u>
·	İ		centistokes	1		Solubility in + Acetone		1
tk to	İ		η •c	1		Carbon tet.		
tx C		+		1		Benzene Ether		
B' ∟ °C	1		 	 	_	n-Heptane		
C'			B ^V to	1		Ethanol	1	
A'* to B'* °C			(BV)	-		Water Water in		l
Ac to	 	+-	(A ^V))	1				Π
Bc t C		1		 	-		1	1
Cc			c _p liq. ∘K	1				ł
Cryos. A° consts. B°			c _p vap.300°K 400	0.38610 0.48377				
t _e °C	348.0	5	c _v vap.					
for under	cooled liqui	d				grams/100 gran	ms solven	t
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from det	da	ta 5-Calc. by for	mula	
SOURCE:		AF						
PURIFICAT		AP						
LITERATU	RE REFERI	NCES	3:					

								No. 21	
NAME	l-Nonad	ecyne				ST	RUCTURAL	FORMUL.	A
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 264.4	78		C ₁₇ H ₃₅ C≡C	Н	
		Ref.			Ref.				Ref.
F. P. *C	33.	2	dt/dP			f	to		
F.P. 1009			°C/mm			g	∣ °K		
B.P. °C			0.1 mm BP	139.89	5	h	I		
760 mm	327.	2	t.	0.0621 0.0344	5	_f,_	to		
100 30	246.54 209.87	4	30 mm	0.9272	5	g'	°K		
10	181.6	4	∆Hm cal/g		\vdash	h'			
1 0. 1	133.9 97.4	5	ΔHv cal/g			m	300 to	0.0329	
Press. mn		+	0.1 mm			n o	600 °K	0.0013	
te	1580.0	5	30 mm BP	63.00	5			-0.0 ₆ 50	
Density			t.	52.12 48.92	5	m' n'	700 to	0.1267	
g/ml 20°C	0.8050	2	te (d, e)	48.68	5	0'	1000 °K	0.0011 -0.0 ₆ 37	
d ^t 25 4 30	0.8016 0.7982	2 4	AHv/T _e	21.2	5		<u> </u>	0.065	<u> </u>
	0.8186	4	d 208 to		5		face tension es/cm. 20°C	27.65	5
Ъ	- 0. 0368	4	_e366_ °C		5	8,"	30	26.73	5
Ref. Index			e' °C				40	25.83	5
ⁿ D 20°C	1 4467	2 2	d g/ml			Par	achor [P]		
30	1.4446	4	v _c mi/g		1 1		20°C		İ
"C"	0.7384	4	t _c °C				40		l
MR (Obs.)		2	P _c mm	<u> </u>	\sqcup			753.4	5
MR (Calc.	1 87 040	5	PV/RT 0.1 mm	1 0000	5	Exp	o. L.1.%/wt. u.		
(nD-d/2)	1.0463	2	30 mm	1.0000	5	Dis	persion	103. [‡]	2
Dielectric			BP	0.9184	5	Fla	sh Point °C		ऻ
A 208 to B 416 °C		4	te t _c	0.8912	5	Fir	e Point		
c -3	177.	5	ΔHc kcal/m	2792.86	2		Spec.		l
A* 208 to		5	ΔHf ΔFf	-34.49	2		ra V. Ray Dif.	İ	}
B*[376 °C	2241.3	5	Viscosity	78. 29	2	Infz	ared		<u> </u>
c			centistokes				ubility in +		l
t _k			η °c				etone rbon tet.		ļ
X		-				Be	nzene		
A' to B' °C							her Heptane		İ
c'	_		B _v to				hanol		
A¹* to			AV I °C	_			ater ater in		
B'* °C		1	(B ^V)			— "	B 6 C I A II		+
Acl to Bc tc °C			(A ^V)		\sqcup				-
Cc C	-		c _p liq. °K						
Cryos. A°			c_ vap.300°K	0.38642	2				
t _e °C	i	5	j - 400	0.48435	2				
	rcooled liqui		ш	1		+ 91	rams/100 gra	ms solven	ıt.
			PI 3-Lit. 4	-Calc, from de	t. da				
SOURCE:		API							
PURIFICA	TION:	API							
	RE REFERE								

								No. 22	
NAME	l-Eicosyne						STRUCTURAL FORMULA		
Mole Ref. Molec % Pur. Form				ular C ₂₀ H ₃₈ Molecular ula C ₂₀ H ₃₈ Weight 278.504			C ₁₈ H ₃₇ C≡C	н	
			Ref.		***************************************	Ref			Ref.
F. P. *C	36.		2	dt/dP			f to		
F. P. 100%	T			°C/mm			g °K		1
B. P. °C				0.1 mm BP	142.65	5	h .		
760 mm	340.	.	2	t	0.0633 0.0347	5 5	$\frac{1}{\mathbf{f}} + \frac{1}{\mathbf{to}}$		
100 30	258.03		4	30 mm	0.9540	5	g' K		l
10	191.8		5	ΔHm cal/g	····		h'		Ì
1 0. 1	143.2	- 1	5	ΔHv cal/g	 	\vdash	m 300 to	0.0323	4
Press. mm	100.0	-		0, 1 mm			n 600 °K	0.0013	
t _e .	1601.5		5	30 mm	60.40	5	° ;	-0.0650	4
Density				BP t	50.36 47.09	5 5	m' 700 to	0.1264	
g/ml 20°C	0.80	73 7	2	te te (d, e)	46.82	5	n' 1000 °K	0.0011	
d ₄ 25	0.80 0.80	397 05	2 4	ΔHv/T _e	21.4	5	l	-0.0637	4
a 30				d 219 to	81.68	5	Surface tension	27.04	1_
b	0.82		4	<u>e 381 °C</u>	0.0921	5	dynes/cm. 20°C	27.84	5
Ref. Index	1	-	\dashv	d' to	1		40	26.01	5
n _D 20°C	1.45 1.44	01 🗯	2			$\vdash \dashv$	Parachor [P]		
25 30	1.44	81 4	2	d g/ml v ml/g			20°C		1
"C"	1.44		4	tc °C			30 40		1
	0.73		4	P _c mm			li .	792.4	5
MR (Obs.) MR (Calc.)	92.72 92.55		2 5	PV/RT	 	1	Exp. L.1.%/wt.		
(nD-d/2)	1.04	65#	2	0.1 mm	1.0000	5	u.	#	1_
Dielectric	2.10		5	30 mm BP	1.0000 0.9122	5 5	Dispersion	103.≠	2
A 219 to	7,52	336	4	te	0.8833	5	Flash Point °C Fire Point		İ
B 431 °C	2386.3		4	tc				 	
	174.		5	ΔHc kcal/m ΔHf	2939. 78	2	M Spec. Ultra V.	1	İ
A* 219 to	2.24	673	5	ΔFf	-39.41 81.00	2 2	X-Ray Dif.	l	
B* ∟391 °C	2274.3		"	Viscosity		\vdash	Infrared		├
°			- 1	centistokes			Solubility in + Acetone		1
tk to	ľ			ነ የ • ℃			Carbon tet.	ļ	
A' to	 						Benzene		1
B' C						\perp	Ether n-Heptane		1
C'				B ^V to			Ethanol		
A'* to			1	A ^V C			Water Water in	1	1
B'* °C	 		-	(B ^V)	1			†	†
Ac to				(A ^V)		1		1	
Cc				c _p liq. •K	1				
Cryos, A° consts, B°				c _p vap.300°K	0.38677 0.48488	2			
	379.8		5						
<pre># for under</pre>	cooled li	quid					+ grams/100 gra	ms solven	t
REFERENC	ES: 1-D	ow 2	2 - AF	PI 3-Lit. 4-0	alc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:			AP						
PURIFICATION: API									
LITERATU	RE REF	EREN	CES	:					

No. 23 l-Heneicosyne NAME STRUCTURAL FORMULA C₁₉H₃₉C≡CH Molecular C21H40 Mole Ref Molecular Weight 292.530 % Pur. Ref. Ref. Ref. F.P. °C F.P. 100% 2 41. dt/dP f to °C/mm g °K 0.1 mm 145.39 B. P. °C h BP 0.0643 5 760 mm t_e 352. 2 0.0354 5 f١ to 100 268.63 4 g' °K **3**0 mm 0.9700 5 30 230.60 4 10 201.2 h١ 5 ∆Hm cal/g 151.7 5 to m 0.1 AHv cal/g 113.8 ٩K n 0.1 mm Press, mm 0 30 mm 58,90 ^te 1594.2 5 BP 48.06 m' Density to te (d, e) 44.72 5 0.8094_{\pm}^{\ddagger} 0.8060_{\pm}^{\ddagger} 0.8026^{\ddagger} n' °K g/ml 20°C 2 44.32 5 01 d_4^t 25 ΔHv/T_e 20.9 5 30 4 Surface tension 229 81.85 d to 5 0.8230 4 28.00 5 dynes/cm, 20°C <u> 394</u> °C 0.0960 ь -0.0368 27.07 5 4 30 ď 40 26.17 5 Ref. Index e' °C 20°C 1.45137 2 [P] a a Parachor 1.4493 1.4473 dc g/ml 25 2 20°C vc ml/g 30 4 30 t_c 40 "C" 0.7383 4 P_c mm 831.4 Sugd. 5 MR (Obs.) 97.33 * 2 Exp. L.1. %/wt. PV/RT MR (Calc.) 97.176 0.1 mm 1.0466[‡] 1.0000 u. (nD-d/2)2 30 mm 103.[≠] 1.0000 5 Dispersion 2 Dielectric 2.11 5 BP 0.8948 5 Flash Point °C A 229 to 0.8628 7.53602 4 Fire Point B 1444 °C 2439.3 M. Spec. C 172. 5 ∆Hc kcal/m Ultra V ΔHf A* 229 to B* 404 °C 2.30870 5 X-Ray Dif. ΔFf 2360.3 Infrared ĸ Viscosity Solubility in c centistokes Acetone t_k to Carbon tet. °C Benzene A' | to Ether ٠<u>c</u> B١ n-Heptane Bv | Av | C١ Ethanol to °C A!* Water to Water in (BV)I B'* °C Acl (A^V)| Bc_1 °C c_p liq. Cc Cryos, A° cp vap. °K consts. B° c_v vap. te °C 393.3 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

						,	No. 24	
NAME	1-Doc	osyne	•			STRUCTURAL I	FORMUL	A
						C ₂₀ H ₄₁ C≣C	н	
Mole	Ref. Mo	lecul		Molecular		20.41	•	
% Pur.	Fo	rmul	ar C ₂₂ H ₄₂	Weight 306.5	56			
		Ref.			Ref			Ref.
F.P. °C	45.	2	dt/dP			f to		
F. P. 1009	•		*C/mm 0.1 mm	147.86	5	g °K		
B.P. °C 760 mm	363.	2	BP	0.0653	5	<u>h</u> +		ì
100 30	278.35 239.72	4 4	t _e 30 mm	0. 0354 0.9774	5	f' to		
10	209.9	5	ΔHm cal/g	0.7.1.2	 -	h'		
1 0. 1	159.5 121.0	5 5	ΔHm cal/g	 	-	m to		+
Press. mn		-	0.1 mm			n °K		ł
t _e	1621.8	5	30 mm BP	58.13 46.74	5 5	<u> </u>		<u> </u>
Density	2 21.4±		t_	43.38	5	m' to		
g/ml 20°0	0.8114 [‡] 0.8080 [‡] 0.8046 [‡]	2 2	le (4, 6)	43.0	5	o' .		1
dt 25 4 30	0.8046	4	ΔHv/T _e	20.9	1	Surface tension		+-
a b	0.8250 -0.0368	4	d 238 to		5	dynes/cm. 20°C	28,17	5
Ref. Index	,	-	d' to	1		30 40	27.24 26.33	5
n _D 20°0	- 1 1 15 3 1 5	2	d _c g/ml	'	-	Parachor [P]		+-
25 30	1.4524 1.4504 1.4484	2	V mi/g			20°C 30		1
"C"	0.7381	4	t _c ·C			40		1
MR (Obs.	101,97 ±	2	P _c mm		<u> </u>	ļ	870,4	5
MR (Calc. $(nD-d/2)$	1.0468 [‡]	5 2	PV/RT 0.1 mm	1,0000	5	Exp. L.1.%/wt.		
Dielectric		5	30 mm BP	1.0000	5	Dispersion	10 3. [‡]	2
A 238 t		4	t _e	0.8943 0.8615	5	Flash Point °C		
B 457 °C	2486.4	4	t _c	L	<u> </u>	Fire Point M Spec.		-
C	170.	5	∆Hc kcal/m ∆Hf			Ultra V.		
A* 238 to B* 417 °C		5	ΔFf		<u> </u>	X-Ray Dif. Infrared		
к ——	-1		Viscosity			Solubility in +		+
t _k			centistokes 7°C			Acetone		•
t _x *('	ł		Carbon tet. Benzene		
A' to				[{	Ether		
c, '	<u> </u>		B ^v l to			n-Heptane Ethanol		
A!* to			_A'_ °C	_		Water Water in		
B'* 'C		+	(B ^V)					+
Bc t			(A ^V)	 		1		
		<u> </u>	c _p liq. °K					
Cryos, Acconsts, B			с _р vaр. °К					
t _e °C	405.7	5	c _v vap.					
	ercooled liquid					grams/100 gran	ns solver	at
	CES: 1-Dow	2-AI		Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		AP						
PURIFICA		AP						
MIEKATU	RE REFERE	NCES	i:					

No. 25 NAME 1-Tricosyne STRUCTURAL FORMULA C₂₁H₄₃C≡CH Molecular C23H44 Mole Ref Molecular Weight 320.582 % Pur Formula Ref. Ref. Ref. F. P. °C 49. dt/dP f to F.P. 100% °C/mm g °K 0.1 mm 150.33 5 B.P. °C h 5 BP 0.0663 374. 760 mm 2 ^te 0.0355 ſ١ 288.06 to 100 4 ٩ĸ g' 30 mm 0.9928 5 248.83 4 30 218.5 5 h' 10 ∆Hm cal/g 167.3 5 1 m to ∆Hv cal/g 0.1 128.1 5 ۰ĸ n 0.1 mm Press. mm o 30 mm 56.68 5 ^te 1648.9 5 ВP 45.54 5 m' | to Density te (d, e) 42.18 5 0.8131 \(\frac{\pm}{4} \)
0.8100 \(\frac{\pm}{4} \)
0.8069 \(\frac{\pm}{4} \) n' °K g/ml 20°C 2 41.82 5 ٥' 25 30 2 $\mathbf{d_{4}^{t}}$ ΔHv/T_e 20.9 5 4 Surface tension 248 78,84 to 0.8255 4 a 28.30 dynes/cm. 20°C 419 0.0890 ᇷᅴ °C ъ -0.0362 4 27.45 26.62 5 30 to 5 Ref. Index 40 e' 1.4534 2 n D 20°C [P] Parachor d_c g/ml 25 1.4514 2 20°C 1.4494 vc ml/g tc °C 30 4 30 40 "C" 0.7381 4 P_c mm 909.4 5 Sugd 106.62 # MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 106.412 1.0469 0.1 mm 1.0000 (nD-d/2) 11. 2 102. 1.0000 30 mm 2 5 Dispersion Dielectric 2.11 5 BP 0.8936 5 Flash Point °C 0.8600 A 248 to 7.55521 4 Fire Point B _469°C 2533.5 M. Spec. C AHc kcal/m Ultra V. ΔHf A* 248 to 2.35591 X-Ray Dif. ΔFf B* 429°C 2453.2 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. $\mathbf{t}_{\mathbf{x}}$ °C Benzene to Ether B' °C n-Heptane B^v | c' Ethanol ۰c A'* Water to B'* °C (BV) Water in (A^V)| Aci to Вс ۰c cp liq. ۰ĸ Cc Cryos. A c_p vap. °K consts. B° te °C c, vap. 418.1 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

									No. 26	
NAME	1-1	Tetra c	osy	ne			STRUCT	URAL :	FORMUL	A
							C. H	45 C≣CH	I	
Mole % Pur.	Ref.	Mole	cula	ar C ₂₄ H ₄₆	Molecular Weight 334.6	.08	- 22	45		
	<u> </u>		Ref.			Ref				Ref
F.P. °C	52.		2	dt/dP	1	1	4 1		1	1
F.P. 100%			ᅴ	*C/mm			f g	to °K		1
B. P. 'C	†		\exists	0.1 mm	152.80	5	h		1	
760 mm	385.		2	BP t _e	0.0672	5 5	<u>-</u> + -	<u>-</u>	i	
100 30	297.78		4	30 mm	1.0082	5	g'	°K		
10	227.1		5	ΔHm cal/g		\dagger	h'		l	1
1 0. 1	175.2	- 1	5 5	ΔHv cal/g	†	+	m (to		1
Press mm	133.3		-1	0.1 mm		1 1	n o	°K		1
t _e	1677.		5	30 mm BP	55.36 44.42	5 5	<u> </u>			<u> </u>
Density		4	\neg	ŧ	41.01	5	m'	to		
g/ml 20°C	0.81	48' ₄	2 2	le (a, e)	40.66	5	n'	°K		
dt 25 4 30	0.81 0.81 0.80	80#	4	AHv/T _e	20.8	5	<u> </u>		ļ	+
a .	0.82		4	d 257 to		5	Surface to dynes/cm		28, 45	5
ь	-0.03		4	432 °C		5	3,	30	27.51	5
Ref. Index		≠		• 6				40	26.59	5
n _D 20°C	1.45 1.45	23+	2 2	d _c g/ml			Parachor	[P] 20°C		
30	1.45	02*	4	V mi/g				30		
"C"	0.73		4	, -				40	040 4	1_
MR (Obs.)	111.26	#	2	P _c mm	<u> </u>	1	<u> </u>		948.4	5
MR (Calc.)	111.03	0 #	5	PV/RT 0.1 mm	1.0000	5	Exp. L.1			1
(nD-d/2)	1	70.	2	30 mm	1.0000	5	Dispersion		102. #	2
Dielectric	2.11		5	BP	0.8933	5	Flash Po			
A 257 to B 482 °C		438	4	t _e t _c	0.8589	5	Fire Poir	nt		<u> </u>
c —— =	166.		5	∆Hc kcal/m		1	M Spec. Ultra V.			
A* 257 to	2, 37	736	5	ΔHf ΔFf			X-Ray Di	f.		ļ
B* <u>L442</u> °C	2499.5		5	Viscosity	 	+-1	Infrared		<u> </u>	
c	j			centistokes			Solubility	in +		ļ
tk to				η •c			Acetone Carbon	tet.		
X '	.					1 1	Benzene			
A' to B' °C	ł						Ether n-Hepta:	ne		
C'				B ^V to			Ethanol			
A!* to				AV I - °C	_	1	Water Water in			İ
B'* °C	-			(B ^V)	j		Water II		 	+
Ac to Bc t °C	ļ			(A ^V)					1	
Cc	-			c _p liq. ∘K						1
Cryos. A° consts. B°				c _p vap. °K						
t _e °C	430.4		5	c _v vap.						
# for under							† grams/	100 gra	ms solve	nt
	ES: 1-D	ow 2	-AF	PI 3-Lit. 4-0	Calc, from de	t. dat	a 5-Calc	by for	mula	
SOURCE:										
PURIFICAT										
LITERATU	RE REF	ER EN (CES	: -						

No. 27 NAME 1-Pentacosyne STRUCTURAL FORMULA C₂₃H₄₇C≡CH Molecular C₂₅H₄₈ Mole Ref Molecular Weight 348.634 % Pur Formula Ref Ref. Ref. F.P. °C F.P. 100% 55. 2 dt/dP f to °C/mm ۰ĸ g 0.1 mm 155.00 5 B. P. °C h ВP 0.0681 5 395. 306.62 760 mm 2 t_e 0.0355 5 f١ 100 4 ۰ĸ g' 1.0221 5 266.24 30 mm 30 4 10 235.0 5 h' ∆Hm cal/g 182.3 5 1 to 0.1 141.9 5 ∆Hv cal/g n ٩K 0.1 mm Press. mm 0 30 mm 54.06 1701.6 5 ^te 43.32 5 BP m' Density to 39.89 5 te te (d, e) n' g/ml 20°C 0.8163 0.8129 4 °К 2 39.56 5 ۰, $\mathbf{d_4^t}$ 25 2 AHv/Te 0.8095 5 20.8 30 4 Surface tension 265 76.26 d 5 0.8299 4 28.57 5 dynes/cm. 20°C 1 443 0.0834 5 °C ь -0.0368 4 27.63 30 to 40 26.71 5 Ref. Index e' °C 1.4552 20°C n_D [P] Parachor dc g/ml 25 1.4532 2 Z0°C 1.4512# vc ml/g 30 4 30 ŧċ 40 "C" 0.7380 4 P_c mm 5 987.4 Sugd. MR (Obs.) 115.89 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 115.648 5 1.0000 0.1 mm 1.0471 (nD-d/2)5 2 30 mm 102. 1.0000 5 Dispersion 2 Dielectric 2.12 5 BP 0.8927 5 Flash Point °C 265 to 0.8575 7.57091 4 Fire Point 2621.8 В _493 °C 4 M. Spec. C 164. 5 AHc kcal/m Ultra V. ΔHf A* 265 to 2.39684 5 X-Ray Dif. ΔFf B*|_453 °C 2540.2 Infrared ĸ Viscosity Solubility in c centistokes Acetone t_k | to $\mathbf{t_x}$ Carbon tet. °C Benzene A' | to Ether B <u>°с</u> n-Heptane B^V A^V C' to Ethanol °C A'* Water to Water in B'* (BV) °C Acl to (AV) Bc °C cp liq. °К Cryos. Aº ٩K c_p vap. consts. B° t_e °C c_v vap. 441.7 for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc, by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

								No. 28	
NAME	l-Hexa	osyn	ie			STRU	JCTURAL 1	FORMULA	1
						С	; ₂₄ H ₄₉ C≡CH		
Mole		lecul		Molecular	,		24-49		
% Pur.	_ Fo	rmul	a 26 50	Weight 362.6	_				5 (
	T 57	Ref.		·	Ref	ļ <u>-</u>			Ref.
F.P. °C F.P. 100%	57.	2	dt/dP *C/mm	1		f	to °K		
B. P. °C	 		0.1 mm	157.19	5	g h	K		
760 mm	405.	2	BP	0.0690 0.0356	5	<u>-</u> ; +			
100 30	315.45 274.53	4	t _e 30 mm	1.0359	5	g'	to °K		
10	242.9	5	ΔHm cal/g	1.0337	Ť	h'			
1 0. 1	189.4	5		+	├	m I	to		
Press. mm	148.5	3	ΔHv cal/g 0.1 mm			n ;	°K		
t _e	1727.	5	30 mm	52.86	5	°			
Density	1		BP te	42.32 38.87	5	m'	to		
g/ml 20°C	0.8177 [‡] 0.8143 [‡]	2 2	t (d, e)	38.55	5	n' o'	°K		
dt 25 4 30	0.8109	4	AHv/T _e	20.8	5	<u> </u>			
	0,8313	4	d 274 to		5		ce tension s/cm, 20°C	28,68	5
ь	-0.0368	4	d, 454 °C		5	gyne.	30	27.74	5
Ref. Index	#		e' *C				40	26.82	5
n _D 20°C	1.4560 [‡] 1.4540 [‡]	2	d g/ml			Para	chor [P] 20°C		
30	1.4540 ⁷ 1.4520 [‡]	4	d g/ml vc ml/g tc °C			l	30		
"C"	0.7379	4	*c ~				40	102/ 4	_ ا
MR (Obs.)	120.53 ≠	2	P _c mm		-	<u> </u>		1026.4	5
MR (Calc.)	120,266 1.0472 [‡]	5 2	PV/RT 0,1 mm	1,0000	5	Exp.	L.1.%/wt. u.		
(nD-d/2)	2,12	-	30 mm	1.0000	5	Dispe	rsion	102.	2
Dielectric A 274 to	+ 	5	BP t _e	0.8924 0.8565	5		Point °C		
B 1_504°C		4	tc			i 	Point	ļ	_
С	162.	5	ΔHc kcal/m			M Sp Ultra		İ	
A* 274to	2,41490	5	ΔHf ΔFf	1		X-Ra	y Dif.		
B* ∟464°C	2580.6	5	Viscosity	 		Infra			L
·	.]		centistokes	1		Solub Ace	ility in +	}	
t _x t _c			7 ° ℃			Carl	bon tet.	}	
A' l to	ļ					Ben: Ethe	zene		
₿! Ĺ _ °C			ļ.,	 	-		eptane	ĺ	
C'		ļ	B ^V to A ^V °C			Etha Wat			
A'* to B'* °C			(BV)	-{			er in	ĺ	l
Acl to		-	(A ^V)						
Bc t C	J			 		ll .		i	
	 	<u> </u>	1 *			1		1	
Cryos. A° consts. B°			c _p vap. °K						
t _e °C	452.9	5	c _v vap.	<u></u>	<u> </u>	L			<u> </u>
	rcooled liquid		N. 2				ms/100 gran		<u> </u>
REFERENC	E3: 1-D0M	2-AI		Calc. from de	t. da	ta 5-0	Calc. by for	mula	
SOURCE:	TON.		PI PI						
PURIFICAT	RE REFEREI								
LI I EKA I U	KE KEFEKEI	NCES) :						
									

No. 29 NAME 1-Heptacosyne STRUCTURAL FORMULA C25H51CECH Molecular C27H52 Mole Molecular Weight 376.686 % Pur. Ref Ref. Ref. F.P. °C F.P. 100% 60. 2 dt/dP f to °C/mm °K g 159.39 0.1 mm 5 B. P. °C h BP 0.0699 5 760 mm 415. 2 t_e 0.0356 5 ſ١ to 100 324.29 4 g¹ ۰ĸ 1.0498 5 30 mm 30 282.82 4 250.7 10 5 h! ∆Hm cal/g 196.6 5 1 m to ∆Hv cal/g 0.1 155.0 5 ٥ĸ n 0.1 mm Press. mm 0 30 mm 51.75 5 t_e 1751.8 5 BP 5 41.38 m' | Density to 37.91 t_e (d, e) n' g/m1 20°C ۰ĸ 0.81907 2 37.64 5 0.8156 0.8122 ٥' dt 4 25 2 AHv/Te 20.6 5 30 4 Surface tension 282 73.94 5 to a 0.8326 28.79 5 dynes/cm. 20°C <u>| 465</u> ٠C 0.0784 -0.0368 ь 27.85 5 4 ăĦ 30 to 26.93 5 40 Ref. Index e¹ ۰c 20°C 1.4568 2 [P] $\mathbf{n}_{\mathbf{D}}$ d_c g/ml v_c ml/g t_c °C Parachor 25 1.4548 2 20°C 1.4528 30 4 30 $\mathbf{t}_{\mathbf{c}}$ 40 "C" 0.7380 4 P_c mm Sugd. 1065.4 5 MR (Obs.) 125.17 ≠ 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 124.884 0.1 mm 30 mm 1,0000 5 1.0473 u. (nD-d/2) 2 1.0000 102. 5 Dispersion 2 Dielectric 2.12 5 BP 0.8919 Flash Point °C 0.8553 A 282 to 7.58347 4 Fire Point 2704.0 B |_515°C 4 M. Spec. C 160. 5 ΔHc kcal/m Ultra V. ΔHf A* 282 to 2.43266 5 X-Ray Dif. ΔFf B*|_475°C 2621.3 Infrared K Viscosity Solubility in c centistokes Acetone tk tx to Carbon tet. °C Benzene A to Ether B °C n-Heptane B^V | C to Ethanol °C Water A1* to Water in (B^V) B'* °C Acl (AV) to Bc_1 °C cp liq. Cc Cryos. Aº cp vap. ۰ĸ consts, B° c vap. te °C 464.2 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME	1-C	ctacosyne	•			STRUCTURAL	FORMUL.	
						C ₂₆ H ₅₃ C≣C	н	
Mole % Pur.	Ref.	Molecul Formul		Molecular Weight 390.	712	261153		
		Ref			Ref			Re
F.P. °C	62.	2	dt/dP	1	 	6 1	<u> </u>	+
F.P. 100%			*C/mm		1 1	f to g °K		
B.P. °C	 		0.1 mm	161.32	5	h		ļ
760 mm	424.	2	BP	0.0707 0.0356	5			l
100 30	332.25	4 4	t _e 30 mm	1,0621	5	g' to		
10	257.8	5		1	+	h'	}	
1	203.0	5 5	ΔHm cal/g	 	-	m to	 	十
0.1 Press. mm	161.0	- 3	ΔHv cal/g 0.1 mm			n °K		
t _e	1773.5	5	30 mm	50.65	5	°		1
e Density			BP	40.44 36.94	5	m' to		T
g/ml 20°C	0.82	02 2	t _e (d, e)	36.64	5	n' °K	l	Ì
dt 25 4 30	0.81 0.81	687 2 34	ΔHv/T	20.7	5	0'		
			d 290 to	72.82	5	Surface tension		
a b	0.83		_e <u> 4</u> 76 °C		5	dynes/cm. 20°C	28.89 27.94	5
Ref. Index			d' to		i i	40	27.02	5
n _D 20°C	1 45	75 2		1	1-	Parachor [P]		\top
25 30	1.45	55 [‡] 2	d g/ml vc ml/g			20°C		
"C"	1.45	35 4	tc° °C°		1 1	30 40		
	0.73		P _c mm				1104.4	5
MR (Obs.) MR (Calc.	1 120 50	2 1 5 1	PV/RT			Exp. L.1.%/wt.		\top
(nD-d/2)	1.04	74 [‡] ž	0.1 mm 30 mm	1.0000	5	u.	102.≠	2
Dielectric	2.12		BP	1.0000 0.8912	5	Dispersion Flash Point °C	102.	1-
A 290 to		724 4	t _e	0.8539	5	Fire Point		İ
B 1 526 °C		4 5	t _c	 	\vdash	M Spec.		†
A* 290 to	158. 2.44		ΔHc kcal/m ΔHf			Ultra V.	İ	1
B* 486 °C		5	ΔFf			X-Ray Dif. Infrared		1
к	1		Viscosity			Solubility in +	 	+
t to	-		centistokes			Acetone		İ
t _x to			7 °⊂		1	Carbon tet.		1
A' to	 	-				Benzene Ether		
B' *	<u>: </u>		B ^V to	 	+	n-Heptane		
	ļ		B' to			Ethanol Water		
A'* to B'* *C			(BV)	-		Water in		
Acl to	+	-+-	(A ^V)					T
Bci t C				 	+			
Cc	1		c _p liq. °K			ş-		1
Cryos. A° consts. B°			c _p vap. °K					
te °C	474.3	5	c _v vap.	<u> </u>		<u> </u>		L
for unde						grams/100 gra		nt
REFEREN	JES: 1-D			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	NOV:	AP						
PURIFICAT		AP						
LITERATU	RE REFI	ERENCES	:					

No. 31 l-Nonacosyne NAME STRUCTURAL FORMULA C₂₇H₅₅CH≣CH Molecular C29H56 Mole Ref Molecular Weight 404.738 % Pur. Ref. Ref. Ref. F. P. °C 65. 2 dt/dP f to F.P. 100% °C/mm ۰ĸ g 0.1 mm 163.23 5 B.P. °C h BP 0.0714 5 760 mm 432. 2 ſ١ 0.0357 5 100 to 339, 31 4 ۰ĸ g1 30 mm 1.0737 5 30 296.90 4 10 264.1 5 h! ∆Hm cal/g 5 208.7 m to AHv cal/g 0.1 166.1 5 ۰ĸ n 0.1 mm 30 mm Press. mm o 49.51 1793.8 ^te 5 BP 5 39.52 m Density te te (d, e) 36,05 5 g/ml 20°C n† ۰ĸ 0.8213 2 5 0.8180 35.78 ۰, d_4^t 25 2 ΔHv/Te 0.8147[#] 20.5 5 30 4 7 297 Surface tension 71.45 5 0.8345 4 dynes/cm. 20°C 28.98 5 l <u>485</u> °C 0.0739 ь -0.0366 4 30 28.06 5 to 27.16 5 40 Ref. Index e' °C 20°C 1.4581 n_D Parachor [P] d_c g/ml 1.4561[‡] 1.4541[‡] 25 2 20°C vc ml/g 30 4 30 "C" 40 0.7379 4 P_c mm 5 1143.4 Sugd. MR (Obs.) 134.45 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 134.120 5 1.0475 1.0000 0.1 mm 30 mm (nD-d/2) 2 102. 1.0000 5 Dispersion 2 Dielectric 5 2.13 BP 0.8910 Flash Point °C ^te 0.8531 A 297 to 7.59699 Fire Point B 535°C 2777.8 4 M. Spec. 157. C 5 AHc kcal/m Ultra V. ΔHf A* 297 to 2.46916 5 X-Ray Dif. ΔFf B*|_495°C 2694.2 Infrared ĸ Viscosity Solubility in centistokes Acetone to °C Carbon tet. °C Benzene A۱ to Ether B١ °C n-Heptane B^V | C' to Ethanol °C Water A1* tο (B^V)| Water in B'* °C Acl (AV) to Bc ۰c c_p liq. °К Cc Cryos. Aº cp vap. ۰ĸ consts. B° c_v vap. te °C 483.3 for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 32	
NAME	l-Tria	conty	ne			STRUCTURAL I	FORMULA	١.
L1						С ₂₈ Н ₅₇ С≇СН		
Mole % Pur.	Ref. Mo	lecul	ar C ₃₀ H ₅₈	Molecular Weight 418.	764	28 57		
		Ref.			Ref			Ref.
F.P. °C	67.	2	dt/dP			f to		
F.P. 100%	<u>'</u>	-	*C/mm 0.1 mm	165.2	5	g °K		
B. P. °C 760 mm	441.	2	BP	0.0722	5	<u>+</u> +		
100 30	347.27 304.38	4 4	t _e 30 mm	0.0357 1.086	5	f' to		
10	271.2	5	ΔHm cal/g	1.000	Ť	h'		
1 0. 1	215.1 172.1	5	ΔHv cal/g	<u> </u>	-	m to		
Press. mm			0.1 mm 30 mm	40 55	ا ۔ ا	n °K		
t _e	1816.2	5	BP	48.55 38.71	5 5	<u> </u>		-
Density g/ml 20°C	0.8224	2	t _e (d, e)	35.19 34.99	5 5	m' to n' 'K		ĺ
dt 25	1 0 8100₹	2	ΔHv/T _e	20.6	5	o'		
4 30	0.8156	4	d 304 to		5	Surface tension		_
Ъ	0.8360 -0.0 ₃ 68	4	d' 1 495 to	0.0721	5	dynes/cm. 20°C	29.07 28.12	5
Ref. Index	1 4507#		e' •C			40	27.20	5
n _D 20°C	1 4567	2	d g/ml v ml/g			Parachor [P] 20°C		İ
30	1.4547	4	tc °C			30		
"C"	0,7378	4	P _c mm			40 Sugd.	1182.4	5
MR (Obs.) MR (Calc.	1 120 720	2 5	PV/RT		1	Exp. L.1.%/wt.		
(nD-d/2)	1.0475	2	0.1 mm 30 mm	1.0000 1.0000	5	u. Dispersion	101.≠	2
Dielectric	2,13	5	BP	0.8906	5	Flash Point °C		-
A 304 to		4	t _e t _c	0.8521	5	Fire Point		
С	155.	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		
A* 304 to B* 505 °C		5	ΔFf		1	X-Ray Dif. Infrared		
к — — —			Viscosity			Solubility in +		_
k			centistokes 7 °C	Ì	į	Acetone		
'x			'		ļ	Carbon tet. Benzene		
A' to					1	Ether n-Heptane		
C'			B ^V to			Ethanol		
A'* to B'* *C			A ^V C	-		Water Water in		
Ac l to	+		(A ^V)					
Bc tc_C			c _p liq. °K	†	 	1		
Cryos. A	1	 	c _p vap. °K					
consts. B	ļ	_	· -		ĺ			
t _e °C	493.4	5	c _v vap.	<u> </u>	<u> </u>	L		L
	rcooled liquid		PI 3-Lit. 4-0	Cala (mam 3-		grams/100 gram		<u> </u>
SOURCE:		AP		Jaic, irom de	L. QA	ta 5-Calc. by form	nula	
PURIFICA"	TION:	AP						
	RE REFERE							
<u> </u>								

								No. 33	
NAME	l-Hentr	iaco	ntyne			STRU	CTURAL	FORMUL	A
					\dashv	c	C ₂₉ H ₅₉ C≡C	СН	
Mole % Pur.	Ref. Mol	ecul.		Molecular Weight 432.79	0		-, -,		
		Ref.			Ref.				Ref.
F.P. °C	69.	2	dt/dP			f	to		
F.P. 100%	<u> </u>		°C/mm 0.1 mm	167. σ6	5	gl	°K		1
B.P. °C 760 mm	449.	2	BP	0.0729	5	_ <u>h</u>			
100	354.32	4	t _e	0.0357	5	f' g'	to °K		
30 10	310.98 277.4	4 5	30 mm	1.0976	3.	h'	-		
1	220.7	5	ΔHm cal/g			m	to		_
0.1 Press. mm	177.2	5	ΔHv cal/g 0.1 mm			n i	°K		
t _e	1836.8	5	30 mm BP	47.55	5				<u> </u>
Density	4		l t_	37.90 34.39	5	ma'l	to •K		
g/ml 20°C	0.8234 0.8200 0.8200	2	'e (u, e)	34.20	5	6'	- K		
d ^t 25	0.8260 _≠	2	ΔHv/T _e	20.6	5			ļ	
	0,8370	4	d 311 to	69.30	5.		e tension cm. 20°C	29.16	5
_ <u>b</u>	-0.0368	4	d' 504 °C	0.0699	5	8	30	28.21	5
Ref. Index	1 4593#	2	e' i °C				40	27.28	5
25	1.4573	2	d _c g/ml			Paraci	20°C		
30	1.4553	4	vc ml/g tc °C		1 1		30		Ì
"C"	0.7378	4	Pcmm				40 Sugd.	1221.4	5
MR (Obs.) MR (Calc.)	143,71 * 143.356 _	2 5	PV/RT			Exp. I	. 1. %/wt.		I^-
(nD-d/2)	1.0476	2	0.1 mm 30 mm	1.0000	5	Dianes	u.	101.≠	١,
Dielectric	2.13	5	BP	1.00 0 0 0.8905	5 5	Disper	Point °C	101.	2
A 311 to	7.60994	4	t _e	0.8515	5	Fire P			
B _554°C	2851.7 154.	4 5	tc ΔHc kcal/m	1	-	M. Spe			
A* 311to	2.50239	5	∆Hf		1 1	Ultra V			
B*[_5 <u>14</u> °C K	2766.9	5	ΔFf		├	Infrare			
C C			Viscosity centistokes			Solubil			
t _k			η °C			Aceto Carbo	ne on tet.	1	
A' to	-				1 1	Benze	ene		
B' °C	_]			ļ	1	Ether n-He		ļ	
C'			B ^V to A ^V C			Ethan	ol		
A'* to B'* °C			$\begin{vmatrix} \mathbf{A} \\ -\mathbf{B} \\ \mathbf{V} \end{vmatrix} = \mathbf{C}$	-		Wate:			
Acl to	 		(B) (A ^V)						\vdash
Bc tc C					+			}	
Ce	1	L_	Р -						
Cryos. A° consts. B°			c _p vap. *K						
te °C	502.4	5	c _v vap.			<u> </u>			
	rcooled liquid						s/100 gra		ıt
	ES: 1-Dow	2-A		Calc. from de	t. da	ta 5-Ca	lc. by for	mula	
SOURCE:	PTON.	AI							
PURIFICAT	RE REFEREI	AI							
			•						

									No. 34	
NAME	l - Do	triac	onty	ne			STRU	CTURAL	FORMULA	4.
	_							C ₃₀ H ₆₁ C≡C	I	
Mole	Ref.	Mol	lecul	ar C ₃₂ H ₆₂	Molecular			30 61		
% Pur.		For	rmul	a C ₃₂ H ₆₂	Weight 446.8	16				
			Ref.			Ref.				Ref
F.P. °C	71.		2	dt/dP	İ		f	to		
F.P. 100%	 			*C/mm 0,1 mm	168,72	5	g	°K		
B.P. *C 760 mm	457.		2	BP	0.0736	5	<u> </u>			l
100	361.40)	4	t _e	0.0357	5	f'	to		1
30	317.64	·	4	30 mm	1,1084	5	g'	°K		
10 1	283.7 226.5		5	∆Hm cal/g	1		h'			<u> </u>
ō . 1	182.5	1	5	ΔHv cal/g			m	to °K		l
Press. mm	1			0.1 mm	1444		n	-K		ł
t _e	1858.9		5	30 mm BP	46.66 37.18	5				-
Density		£		t _e	33.67	5	m'	to •K		İ
g/ml 20°C	0.82 0.82	43' ₊	2	t (d, e)	33.47	5	0,	K		
dt 25 4 30	0.82	77#	4	ΔHv/T _e	20.6	5				├
	0.83		4	d 318 to		5		ce tension /cm. 20°C	29.23	5
ъ	-0.03		4	-a, -513- * 5		5	3,200	30	28.31	5
Ref. Index		4		e' •c				40	27.40	5
n _D 20°C	1.45	987	2	d g/ml			Para			
25 30	1.45	58≠	2	d g/ml vc ml/g				20°C 30		1
"C"	0,73		4	ic C				40		
MR (Obs.)			2	P _c mm				Sugd.	1260.4	5
MR (Calc.)	147 97	4 . 1	5	PV/RT			Exp.	L. 1. %/wt.		
(nD-d/2)	1.04	77*	2	0.1 mm 30 mm	1.0000	5	Diene	u.	101.#	2
Dielectric	2.13		5	BP	1.0000 0.8910	5		rsion Point °C	101.	12
A 318 to		111	4	te	0.8515	5	Fire			
B 1_563 ℃ C	2880.8 152.		4 5	t _c	-	-	M Spe	c.		
A* 318 to	2.51	100	5	ΔHc kcal/m ΔHf		1	Ultra	v.		
B* 523 °C		170	5	ΔFf	}		X-Ra Infra			1
к — — —				Viscosity						├
t to	-			centistokes			Acet	,		1
t _k to t _x C				ን ℃		1		on tet.		
A' l to	 						Benz Ethe			
B' •C				L		-		ptane		1
<u>c' </u>	<u> </u>			B ^V to			Etha			1
A'* to B'* °C	1			<u> </u>	-[Wate			1
	 			(B ^V)]					\dagger
Ac to				(A ^V)		<u> </u>				1
Cc	-			c _p liq. °K	1					ļ
Cryos. A*				c _p vap. °K						ł
consts, B°	.			1 -	!					ľ
te °C # for under	511.4		5	CV TOP.		<u> </u>	L		L	<u>L</u>
			2 4 5	27 2 7 4 4 4	2-1- (1			ns/100 gran		t
SOURCE:		-		PI 3-Lit. 4-0	Jaic. Irom de	t. da	III 5-C	aic. by for	mula	
PURIFICAT	TON:		AF							
LITERATU		ED Er	AF							
1 EKA 10	REF.	er en	CES) .						

No. 35 1-Tritriacontyne NAME STRUCTURAL FORMULA C31H63C≡CH Molecular C 33 H64 Mole Molecular % Pur Weight 460.842 Ref Ref. Ref. F.P. °C F.P. 100% 73, 2 dt/dP f to °C/mm °K R 170,36 0.1 mm B. P. °C h BP 0.0742 5 760 mm 464. 2 0.0358 5 f t, to 100 367.58 4 g' ۰ĸ 30 mm 1.1184 5 30 323.42 4 h' 10 289.2 5 ∆Hm cal/g 231.4 5 m to AHv cal/g 0.1 187.1 ۰ĸ n 0.1 mm Press. mm o 30 mm 45.71 5 1873.0 t_e 5 ВP 36.35 32.82 m to Density g/ml 20°C te te (d, e) 5 n' 0.8252 0.8218 2 32.66 5 ٥, 2 dt 25 0.8184 AHv/T 20.5 5 30 4 Surface tension 324 ď to 67.25 0.8388 29.31 2 4 dynes/cm. 20°C 5 å. 521 °C 0.0666 Ъ -0.0368 4 30 28.35 5 40 27.42 5 Ref. Index e' °C 1.4603 ⁿD 20°C [P] 2 Parachor d_c g/ml 25 1.4583 2 20°C 1.4563 vc ml/g 30 4 30 t_c 40 "C" 0.7377 4 P_c mm 1299.4 Sugd. 5 MR (Obs.) 153.01 [#] 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 152.592 0.1 mm 1.0477 1.0000 5 (nD-d/2)2 30 mm 101. [≠] 1.0000 5 Dispersion 2 Dielectric ВP 2.13 5 0.8895 5 Flash Point °C 0.8494 A 324 to 7.61812 4 Fire Point 2913.4 B _571 °C 4 M. Spec. C 151. 5 AHc kcal/m Ultra V ΔHf A* 324 to 2.53156 5 X-Ray Dif. ΔFf B* 531 °C 2828.3 Infrared K Viscosity Solubility in centistokes Acetone tk tx to °C Carbon tet Benzene A'ı to Ether B١ °C n-Heptane B^V A c' to Ethanol °C A'* Water Water in B'* (B^V)[°C Acl (A^V) Bc c_p liq. °K Сc Cryos. A* •ĸ c_p vap. consts, B° c_v vap. te °C 519.3 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 36	
NAME	1-Tetrat	riacon	yne			STRUCTURAL I	FORMULA	4
						C 11 C-C11		
Mole	Ref. N	(alasul		Malagulan		C ₃₂ H ₆₅ C≣CH		
% Pur.	Rei. N	Formul	ar C ₃₄ H ₆₆	Molecular Weight 474,	868			
		Ref.			Ref.			Ref.
F. P. *C	74.	2	dt/dP			f to		
F. P. 1007			*C/mm	172 02	5	g °K		1
B.P. °C 760 mm	472.	2	0.1 mm BP	172.02 0.0749	5	<u> </u>		
100 11111	374.66	4	t _e	0,0358	5	f' to		l
30 10	330.08 295.6	4	30 mm	1.1292	5	g' K h'		1
1	237.2	5	ΔHm cal/g			ļ <u>-</u>		₩
0, 1	192.4	5	ΔHv cal/g			m to r	!	1
Press. mn	1893.8	5	0.1 mm 30 mm	44.93	5	•		
e Density		-	BP t	35.70 32.17	5	m' to		Г
g/ml 20°0	0.8260 0.8227	ž 2	te (d, e)	32.02	5	n' *K		
dt 25 4 30	0.8227 0.8194	¥ 2 4	ΔHv/T _e	18.81	5	0'		<u> </u>
a	0.8392		d 330 to	66.38	5	Surface tension	29.37	5
Ď.	-0.0366				5	dynes/cm. 20°C	28.45	5
Ref. Index		4	d' to			40	27.54	5
n _D 20°0	1 4500		d g/ml v ml/g			Parachor [P] 20°C		
30	1.4568	≠ 4	vc ml/g	ł		30		
"C"	0.7377					40	1220 4	5
MR (Obs.)			P _c mm	 		Exp. L.1.%/wt.	1338.4	13
MR (Calc. (nD-d/2)) 157.210 1.0478	≠ 5 2	0.1 mm	1.0000	5	u.		ļ
Dielectric		5	30 mm BP	1.0000 0.8895	5 5	Dispersion	101. [≠]	2
A 330 to		9 4	te	0.8489	5	Flash Point °C Fire Point		İ
B _ 580 °C		4	t _c			M Spec.		+-
C	149.	7 5	ΔHc kcal/m ΔHf			Ultra V.		1
A* 330 to B* 540 °C		' 5	ΔFf			X-Ray Dif. Infrared		ĺ
к ——-	-	1	Viscosity			Solubility in +		\vdash
* to	, 		centistokes 7 °C			Acetone		1
t _x ; •(7		'	į		Carbon tet. Benzene		
A' to						Ether		ł
B', L _ •	-	1 1	B ^V to	<u> </u>	\vdash	n-Heptane Ethanol		1
A1* , to	,		A ^V °C			Water		
B'* ' 'C	3		(B ^V)			Water in		+-
Ac to			(A ^V)					
Bc tc_*C			c _p liq. °K			Ĭ		1
Cryos, A			c _p vap. *K					
consts. B	+	_	c, vap.					
t _e °C	528.3	5	-vp.	<u> </u>		L <u>.</u>	L	<u> </u>
	rcooled liqu		7 2 7 1			grams/100 gran		t
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:	TION:	AI						
	RE REFER	AI						
- LEKAIU	ne refek	ENCES) 					

No. 37 l-Pentatria contyne NAME STRUCTURAL FORMULA C₃₃H₆₇C≡CH Molecular C35H68 Mole Ref. Molecular % Pur Weight 488.894 Ref. Ref Ref. F.P. °C F.P. 100% 76. 2 dt/dP f to °C/mm 1 ۰ĸ g 0.1 mm 177,23 5 B. P. 'C h 0.0755 BP 5 479. 760 mm 0.0358 2 5 f to 100 380.84 ۰ĸ g' 30 mm 1.1392 5 30 335,87 4 10 301.0 h' 5 ∆Hm cal/g 242.1 1 5 m AHv cal/g 0. 1 196.9 5 n °К 0.1 mm 30 mm Press. mm 0 44.09 ^te 1911.2 5 35.02 5 ΒP 31.48 m to Density 5 te te (d, e) 0.8268 0.8235 n' ۰ĸ g/ml 20°C 2 31.34 5 ۰, 25 dt4 ΔHv/Te 18.77 5 0.8202 30 4 Surface tension d 336 65.37 5 0.8400 4 29.44 5 dynes/cm. 20°C <u>| 537</u> ٠Ç 0.0634 -å--0.0366 Ъ 28.51 5 4 30 to 5 27.61 40 Ref. Index e¹ °C 1.4612 20°C $\mathbf{n}_{\mathbf{D}}$ Parachor [P] d_c g/ml 1.4593 2 25 20°C 1.4574 vc ml/g 30 4 t_c 30 40 "C" 0.7376 4 5 P_c mm 1377.4 Sugd. 162. Z8 # MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 161.828 1.0479 1.0000 0.1 mm 5 (nD-d/2) 2 101. [#] 30 mm 1.0000 5 2 Dispersion Dielectric 2,13 5 BP 0.8892 5 Flash Point °C 0.8481 A 336 to 7.62600 4 Fire Point 2975.2 B 1587 °C M. Spec. C 148. 5 AHc kcal/m Ultra V. A*|336 to B*|547 °C AHf 2.55750 5 X-Ray Dif. ΔFf 2889.0 Infrared Viscosity Solubility in centistokes Acetone tk tx to Carbon tet. °C Benzene A' to Ether B <u>°с</u> B^V | n-Heptane C to Ethanol °C Water to Water in B'* °C (B^V) Acl to (A^V) Bc °C c_p liq. ۰ĸ Cryos. A cp vap. •ĸ consts. B° c vap. te .C 536.2 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 38	
NAME	l-Hexatriac	ontyr	ie			STRUCTURAL		4
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 502.	920	С ₃₄ Н ₆₉ С <u>≡</u> €	СН	
		Ref.			Ref			Ref.
F.P. °C	77.	2	dt/dP			f to		
F.P. 100%			°C/mm		_	g °K		
B. P. *C	1		200.97°C BP	175.29	5	h i		
760 mm 100	486. 387.53	2	te	0.03519	5	f' - to		
30	342.34	4	30 mm	1.1453	5	g' °K		
10	307.2 247.4	5	ΔHm cal/g			h'		
Pressure	241.4	1 3	ΔHv cal/g			m to		
mm 200.97°	0.1000	5	200.97°C	50.77	5	n °K		
te	1954.3	5	30 mm BP	43. 29 34. 37	5	<u> </u>		
Density	ŧ		t_	30.84	5	m' to		
g/ml 20°C	0.8275 [‡] 0.8242 [‡]	2	t _e (d, e)	30.79	5	0'		
dt 25 4 30	0.8209	4	ΔHv/T _e	18.98	5	Surface tension		
a	0.8407	4	d 342 to		5	dynes/cm. 20°C	29.50	5
ь	-0.0366	4	e 545 °C		5	y 30	28.57	5
Ref. Index		2	e'			40	27.66	5
n _D 20°C	1 4507	1 2	d g/ml v ml/g			Parachor [P] 20°C		
30	1.4577	4	d g/ml vc ml/g tc °C			30		
"C"	0.7378	4	P _c mm			40 Sugd	1416.4	5
MR (Obs.)		2	PV/RT		_	Exp. L.1.%/wt.	1410.4	
MR (Calc. (nD-d/2)	166.446 1.0479 [‡]	5 2	200.97°C	1.0000	5	u.	1 4	1
Dielectric	2,14	5	30 mm BP	1.0000	5	Dispersion	101. ≠	2
A 342 to	+	-	t _e	0.8475	5	Flash Point °C		
B 595 °C		4	tc			Fire Point		├
С	147.	5	AHc kcal/m			M Spec. Ultra V.	1	1
A* 342 to			ΔHf ΔFf		1	X-Ray Dif.		1
B* ∟ ⁵⁵⁵ °C	2931.19	5	Viscosity		 	Infrared		ـــ
° .—	_		centistokes			Solubility in + Acetone		
t _k to			η ∘c			Carbon tet.		
A' to	1	\vdash		ļ		Benzene Ether		
Bii °C			<u> </u>		<u> </u>	n-Heptane		
C'		 	B ^V to	į.		Ethanol		
A'* to B'* °C			(BV)	-		Water Water in		1
Ac to		+	(B') (A')					
Bc t °C								
Cc	1	 	c _p liq. ∘K					
Cryos. A° consts. B°			c _p vap. *K	1				
t _e °C	545.40		c _v vap.	L		L		
for under	cooled liquid					fgrams/100 gran	ms solven	t
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from det	t. da	ta 5-Calc, by for	mula	
SOURCE:	API							
	TION: API							
LITERATU	RE REFERE	NCES	:					
1								

								No. 39	
NAME	1-Hept	atria	contyne			ST	RUCTURAL	FORMUL	A
							C35H71 C≣CH	ī	
Mole % Pur.	Ref. Mo	leculi mula		Molecular Weight 516.9	946				
		Ref.			Ref.				Ref
F.P. °C	79.	2	dt/dP			f	to		
F.P. 100%		$\vdash \dashv$	*C/mm 0.1 mm	176.68	5	g	°K		
B. P. °C 760 mm	493.	2	BP	0.0767	5	_h _f'			
100 30	393.21	4	t _e 30 mm	0.0359	"	g'	to •K		
10	347.49 312.1	4 5	ΔHm cal/g		+	h'			i
1 0, 1	252.2 206.2	5	ΔHv cal/g		+	m	to		
Press. mm	200.2	اثا	0.1 mm	42.50	_	n o	°К		
t _e	1946.4	5	30 mm BP	42.58 33.76	5	m'			<u> </u>
Density g/ml 20°C	0.8282	2	t _e (d, e)	30, 21	5 5	n'	to °K		
dt 25	0 02407	2	ΔHv/T _e	30.13 18.90	5	0'			
	0.8216	4	d 348 to	63, 64	5		face tension	20.51	
a b	0.8414	4	_e 553 °C	0.0606	5	dyn	es/cm. 20°C 30	29.56 28.63	5
Ref. Index			d' to				40	27.72	5
ⁿ D 20°C	1.4621 ± 1.4601 ± 1.4	2	d _c g/ml	<u> </u>	1	Par	achor [P]		
30	1.4581 ≠	2 4	v _c ml/g		1 1		20°C 30		
"C"	0.7377	4	t _c °C P _c mm				40 Sugd.	1455.4	5
MR (Obs.)	171.56 ≠	2	PV/RT		\vdash	Evr	L.1.%/wt.	1133.1	<u> </u>
MR (Calc.) (nD-d/2)	171.034 1.0480≠	5 2	0.1 mm 30 mm	1.0000	5	-	u.	≢	_
Dielectric	2.14	5	BP	1.0000	5		persion sh Point °C	101.	2
A 348 to	7.63157	4	t _e	0.8468	5		e Point		Ì
B 1603 °C	3031.0 145.	4 5	tc ΔHc kcal/m	-	+		Spec.		
A* 348 to	2,58066	5	∆Hf				ra V. Ray Dif.		
B*[563 °C K	2944.0	5	ΔFf Vicesites		\vdash		ared		
c			Viscosity centistokes				ubility in +		
t _k to t _c °C	1		ļη °C				etone rbon tet.		
t _x °C							enzene her		
B'°C			B _v to		+	n-	Heptane		
A¹* to		-	B to C	1	1 1		hanol ater		
B'* °C			(B ^V)				ater in		<u> </u>
Acl to			(A ^V)						
Bc tc °C	-		c _p liq. °K						
Cryos, A° consts, B°			c _p vap. *K						
t _e °C	553.0	5	c _w vap.						
	rcooled liquid						rams/100 gra		ıt
	ES: 1-Dow			Calc. from de	et. da	ta 5	-Calc. by for	mula	
SOURCE:	701	API					-		
PURIFICAT	ION: RE REFEREI	API							

								No. 40	
NAME	1 -Oc	tatriac	ontyne			STRUCT	URAL I	FORMULA	
						C. H.	73С≣СН		
Mole	Ref. M	(alaan)		Molecular		36	73		
% Pur.	Kei. M	ormul	ar C ₃₈ H ₇₄	Weight 530.	972				
		Ref.			Ref.				Ref.
F.P. *C	80.	2	dt/dP			f	to		
F.P. 1007	<u>'</u>	-	*C/mm 0.1 mm	178.05	5	g	•K		
B.P. *C 760 mm	499.	2	BP	0.0773	5	- + -		ĺ	
100 30	398.51 352.45	4	t _e 30 mm	0.0359	5 5	f' g'	to •K		l
10	316.8	5	ΔHm cal/g	1.1007	+-	h' i			ł
1 0, 1	256.4 210.1	5	ΔHr cal/g	+		m l	to	l	\vdash
Press. mm		+-	0.1 mm			n i	•K		l
t _e	1962.3	5	30 mm BP	41.81 33.14	5				<u> </u>
Density g/ml 20°C	0.8289	<i>į</i>	l t	29.59	5	m'	to •K		l
dt 25	1 00054		e (4, 5,	29.52 18.86	5	0'		ĺ	l
	0.0223	4	ΔHv/T _e	 	5	Surface t	ension		
a b	0.8421 -0.0366	4	e 560 °C		5	dynes/cm	a. 20°C	29.62 28.69	5
Ref. Index			d' to			1	40	27.78	5
ⁿ D 20°C	1		d _c g/ml		\vdash	Parachor			
30	1.4605	4 4	vc ml/g tc °C				20°C 30		l
"C"	0.7377	4					40		_
MR (Obs.)			P _c mm	 	-	Exp. L.1		1494.4	5
MR (Calc. (nD-d/2)) 175.652 1.0480 ⁷	4 5 2	0.1 mm	1.0000	5	u		1	
Dielectric		5	30 mm BP	1.0000	5	Dispersion		101.#	2
A 353 to			t _e	0.8466	5	Flash Poir			
B <u>[610</u> •0	3057.7 144.	4 5	t _c ΔHc kcal/m	 	-	M Spec.			
A* 353 to			ΔHf			Ultra V. X-Ray Di	e .		
B* 570 °C	2970.0	5	ΔFf	 	 	Infrared			
С			Viscosity centistokes			Solubility			
tk to			η •c	1		Acetone Carbon			
t c		+	ŀ		1	Bensene Ether	t	l	Ì
B' •	2		B ^V to		-	n-Hepta:	ne	İ	İ
A¹* to		+	B' to			Ethanol Water			
B'* *C			(BV)	-[Water in	·		<u> </u>
Ac to			(A ^V)	1					
Bc tc_*C	4		c _p liq. •K					1	}
Cryos, A			c _p vap. *K					1	
consts. B	 	-	l -					1	
t _e °C	560.0	5	c _v vap.		L	Ļ		L	<u>L</u>
	cooled liquid		OT 3 744 4	Cala (==== 1				ms solven	<u>t </u>
SOURCE:		2-AF		Calc, from de	t. Q8	ua 5-Calc.	. by ior	mula	
PURIFICA"	TION:	AF							
	RE REFER								

Mole % Pur.	1 - N	onatr	iaco	ntuna		- 1				
				ityne			ST	RUCTURAL	FORMUL	.A
								C ₃₇ H ₇₅ C≡CH	I	
	Ref.	Mol For	ecula mula		Molecular Weight 544.9	98		37 75		
			Ref.	r	T	Ref.				Ref
F.P. °C	82.		2	dt/dP			f	to		T
F.P. 100	%			°C/mm			g	l °K		1
B.P. °C				0.1 mm BP	179.41	5 5	h	l		1
-760 mm 100	505. 403.8	,	2	t _e	0.0359	5	f'	to		
30	357.4		4 4	30 mm	1.1754	5	g'	•к		
10 1	321.5 260.7		5	∆Hm cal/g			h'			↓_
0.1	214.0		5	ΔHv cal/g			m n	to or to		
Press. mr				0.1 mm 30 mm	41.00	_	0			Ī
^t e	1975.5		5	BP	41.09 32.53	5	m'			╁
Density g/ml 20°	റിരം	295#	2	te (d. a)	28.97	5	n'	to °K		
at 25	0.8	2627	2	t _e (d, e) ΔHv/T _e	28.95	5	٥'			
4 30	0.8	229	4			1	Sur	face tension		\top
a b	-0.0		4	d 358 to e 567 °C	61.83	5		es/cm. 20°C	29.67	5
Ref. Inde		300	-	d'			•	30 40	28.74 27.83	5
ⁿ D 20°	ā 1 1 1.	628‡	2	e'		+	Par	achor [P]		+-
25 30		608 588	2 4	dc g/ml vc ml/g				20°C		
"C"	0.7		4	v _c ml/g t _c °C				30 40		
MR (Obs.			2	P _c mm					1533.4	5
MR (Calc	1 180 2	70	5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	1.04	181 [≠]	2	0.1 mm 30 mm	1.0000	5	Dis	u. persion	101.≠	2
Dielectric		4	5	BP	0.8881	5		sh Point °C		┿
A 358 to		1056	4	te t _C	0.8453	5		e Point		
C 1010 7	143.		4 5	ΔHc kcal/m	 	+-1		Spec.		
A*1358 to	2.60	759	5	ΔHf ΔFf				ra V. Ray Dif.		
B* 576 °C	2996.9		5	·	 	+		ared		
c				Viscosity centistokes				ubility in +		
t _k				" ° C				etone rbon tet.		
'x							Be	nzene		
A' to				ļ		1		he <i>r</i> Heptane		
C'				BV to			Et	hanol		
A'* to					_			ater ater in		
B'* °C				(B ^V)			— <u> </u>			+
Acl to				(A ^V)	 	+				
Cc L				c _p liq. °K						
Cryos. A consts. B				c _p vap. °K						İ
t _e °C	566.9		5	c _w vap.						
≠ for und	ercooled l	i q uid					+ g:	rams/100 gra	ms solve	nt
REFEREN	CES: 1-I	Dow .	2-A	PI 3-Lit. 4-	Calc. from de	et. da				
SOURCE:		A	API							
PURIFICA			API							
LITERAT	URE REF	EREI	NCES	5 :						

								No. 42	
NAME	l-Tetracontyne				_	STRUCTURAL FORMULA			
Mole % Pur.	Ref.	Mole	ecula mula	C ₄₀ H ₇₈	Molecular Weight 559.	024	C ₃₈ H ₇₇ C≡C	Н	
			Ref.			Ref.			Ref
F. P. °C	83.		2	dt/dP			f to		
F.P. 100%				°C/mm			g °K		1
B. P. 'C	513			0.1 mm BP	181.05 0.0784	5	h '		
760 mm 100	512. 409.99	,	2	t _e	0.0359	5	f' to		
30	363, 21		4	30 mm	1.1854	5	g' c'K	İ	1
10 1	327.0 265.6	- 1	5	ΔHm cal/g		ПП	h!		
0. 1	218.5	- 1	5	ΔHv cal/g		\Box	m to		ļ
Press. mm	1993.7		5	0.1 mm 30 mm BP	40.45 32.01	5	n o K		
Density		4		t_	28.45	5	m' to		l
g/ml 20°C dt 25 d4 30	0.83 0.82 0.82	:687	2 4	te (d, e) AHv/Te	28.45 18.75	5 5	n' o'		<u> </u>
	0.84		4	d 364 to		5	Surface tension dynes/cm, 20°C	29.72	5
Ъ	-0.03		4	-a,, 575_ * €		5	30	28.79	5
Ref. Index		#	٦	e' •C		\perp	40	27.87	5
ⁿ D 25	1.46	12#	2	d g/ml vc ml/g			Parachor [P] 20°C		
30	1.45	92*	4	v _c ml/g t _c °C	1		30		
"C"	0.73		4	P _c mm			40 Suad	1572.4	5
MR (Obs.)		#	2	PV/RT	 	╁─┤	Exp. L.1.%/wt.	1572.4	13
MR (Calc. (nD-d/2)	184.88	81 [#]	5 2	0.1 mm	1.0000	5	u.	4	1
Dielectric	2.14		5	30 mm BP	1.0000 0.8881	5	Dispersion	101. ≠	2
A 364 to		695	4	t _e t _c	0.8449	5	Flash Point °C Fire Point		
B [625 °C	142.		4 5	∆Hc kcal/m	 		M Spec. Ultra V.		
A* 364 to B* 585 °C		122	5	ΔHf ΔFf			X-Ray Dif. Infrared		
K — — —]		}	Viscosity centistokes			Solubility in +		†
t _k to		ļ		7 00			Acetone Carbon tet.		
A' to							Benzene		
B' ∟ °C			ŀ	ļ.,		4	Ether n-Heptane		
C'	ļ			B ^V to A ^V °C			Ethanol Water		1
A'* to B'* °C		1	l	(BV)	-		Water in		
Ac to Bc tc °C	1			(A ^V)					
Cc				c _p liq. °K	1				1
Cryos, A° consts, B°				c _p vap. °K					
t _e °C	575.0		5	c _v vap.	<u> </u>				
# for unde							grams/100 gra	ms solven	t
REFERENC	ES: 1-D	ow 2	-AF	PI 3-Lit. 4-0	Calc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:			AP						
PURIFICAT LITERATU		FD FX	AP						
				•					

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